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Article

Extraction of Major Groundwater Ions from Total Dissolved Solids and Mineralization Using Artificial Neural Networks: A Case Study of the Aflou_Syncline Region, Algeria

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Abstract: Global water demand due to population growth and agricultural development, has led to widespread overexploitation of groundwater, particularly in semi-arid regions. Traditional hydrochemistry monitoring system still suffers from limited laboratory accessibility and high costs. This study aims to predict major ions of groundwater, including Ca^{2+} , Mg^{2+} , Na^+ , $SO4^{2-}$, Cl^- , K^+ , $HCO3^-$, and $NO3^-$, utilizing two field measurable parameters (i.e., total dissolved solids (TDS) and mineralization (MIN)) in Aflou_Syncline region, Algeria. A multilayer perceptron (MLP) model optimized with the Levenberg-Marquardt backpropagation (LMBP) provided the most predictive accuracy for the different ions of $SO4^{2-}$, Mg^{2+} , Na^+ , Ca^{2+} , and Cl^- with $R^2 = (0.842, 0.980, 0.759, 0.945, 0.895)$ and RMSE = (53.660, 12.840, 14.960, 36.460, 30.530) (mg/L) in the testing phase, respectively. However, the predictive accuracy for the remaining ions of K^+ , $HCO3^-$, and $NO3^-$ was supplied as $R^2 = (0.045, 0.366, 0.004)$ and RMSE = (6.480, 41.720, 40.460) (mg/L), respectively. The performance of our model (LMBP-MLP) was validated in similar geological areas in the adjacent area, including Aflou, Madna, and Ain Madhi. In addition, LMBP-MLP showed very promising results, with performance similar to the original research area.

Keywords: groundwater ions; artificial neural networks; groundwater overexploitation; semi-arid regions; hydrochemical monitoring

1. Introduction

Global demand for water is expected to surge by 2050 due to population growth, economic growth, and changing consumption patterns. Estimates suggest that as many as 6 billion people could face water shortages if demand increases by 20 to 30 percent from current levels [1]. Agriculture, which accounts for 70% of global freshwater withdrawals, will increase competition for resources, especially in dry areas [2,3]. These challenges were illustrated in Algeria obviously, where groundwater quality, a major source of irrigation in arid regions, is declining due to salinity and agricultural pollution [4,5]. For this reason, groundwater quality monitoring is very important for sustainable water resource management [6,7]. Extensive sampling campaigns and extensive water chemistry scans are required to monitor the amount of degradation. These scans should include the

important chemical ions in water such as Ca²⁺, Mg²⁺, Na⁺, K⁺, HCO₃⁻, Cl⁻, and SO₄²⁻ [8,9]. Effective monitoring, however, still requires significant resources and ongoing sampling and analysis. This highlights the urgency of innovative solutions such as artificial neural networks to streamline water quality assessment [10].

Artificial intelligence (AI) has emerged as an innovative tool to simplify water quality assessment. Early application includes [11], who predicted the drinking water quality index (WQI) of Baghdad using artificial neural networks (ANN), identified pH and chloride as the main factors (R²=0.973). Subsequent work by [12] optimized the ANN architecture for prediction, showing that a simpler MLP-4-5-4 model performed better accuracy (R²=0.989) than a deeper network. Based on these foundations, [13] accomplished nitrate concentration predictions by integrating land use data with pH, conductivity, and temperature, highlighting the adaptability of ANN to multivariable systems. More recently, [9] developed an ANN to predict ion concentrations (Ca²+, Mg²+, Na+, K+, HCO₃-, Cl-, and SO₄-) directly from electrical conductivity (EC), achieving high accuracy within the trained EC range. These developments are consistent with a broader trend in ANN-based environmental modeling, with hybrid approaches combining physical and data-driven models gaining popularity.

These methodological innovations have been applied to address regional challenges. [14] compared radial basis function neural networks (RBF-NN) and probabilistic neural networks (PNN) in Iraq's Alnekheeb Basin. They found that PNN was superior in assessing irrigation suitability through salinity and sodium uptake ratios. Similarly, [15] utilized ANN to predict groundwater salinity, outperforming conventional regression models and enabling tailored irrigation strategies for salinity-sensitive crops in Spain's Campo de Cartagena. Furthermore, [16] demonstrated the scalability of ANN in stressed groundwater layers, achieving perfect TDS prediction (R²=0.984) in the Babylonian region of Iraq. However, there are still gaps in applying these technologies to regions with complex evaporite geology, such as the semi-arid regions of North Africa.

In this research, the authors focused on the application of ANN techniques in the Aflou_syncline region of Algeria, a region with distinct geological and climatic features and dependent on groundwater stored in sandstone strata influenced by Aptian gypsum and Triassic evaporite [17]. Here, the increase in the number of wells and the intensive exploitation of groundwater resources have accelerated evaporation and dissolution, increasing the risk of salinity [4]. This research attempted to estimate the ions including Ca²⁺, Mg²⁺, Na⁺, K⁺, SO₄²⁻, Cl⁻, NO₃⁻, and HCO₃⁻ employing ANN optimized with various learning algorithms based on two field measured parameters including total dissolved solids (TDS) and mineralization (MIN) values.

The progress of this research is structured as follows. Chapter 2 explains research material including study area and data collection. Chapter 3 presents model and evaluation including artificial neural networks, optimization algorithm, and measures of accuracy. Chapter 4 provides methodology including model development and hyperparameters selection. Chapter 5 organizes the results and discussion including testing the developed model in adjacent areas. Finally, the main conclusions are addressed in Chapter 6.

2. Research Material

2.1. Subsection Study Area and Data Collection

The objective of this study was to predict major ion concentrations, including Ca2+, Mg²⁺, Na⁺, K⁺, SO₄²⁻, Cl⁻, NO₃⁻, and HCO₃⁻, utilizing two field-measured parameters, total dissolved solids (TDS) and mineralization (MIN) in the Barremian-Aptian-Albian groundwater system of Aflou_Syncline region. Situated in the Central Sahara Atlas, about 300 km southwest of Algiers, Aflou_Syncline region is located north of Djebel Amour, at 1,400 m above sea level (Figure 1).

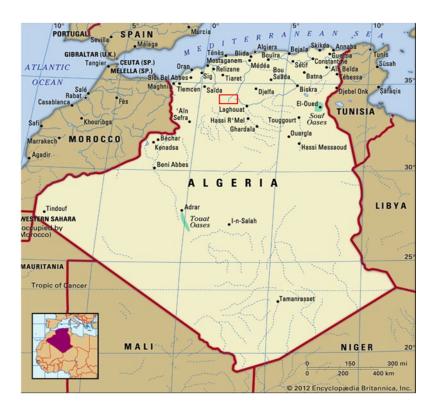


Figure 1. Geographic map of the research area.

From its geographical coordinates (i.e., 34.11°N and 2.10°E), it is located in a mountainous area that acts as a natural barrier between the Sahara Atlas and the Sahara Plateau. This high terrain further exacerbates the climatic contrasts, protecting the region from Mediterranean influences and creating a semi-arid climate with relatively cool temperatures and limited rainfall. Geologically, the area is part of the Saharan Atlas Fold Belt, composing of Mesozoic sediments that date from the Triassic to the Cretaceous. These deposits reflect alternating marine and continental deposits, with limestone, limestone-rich beds, and sandstone-dominated strata. To accomplish this research, 153 groundwater samples were collected from wells distributed throughout the research area and analyzed at the National Office of Water Resources (NAWR) Hydrology Laboratory. These datasets form the basis for modeling correlation among total dissolved solids (TDS), mineralization (MIN), and major ion concentrations employing ANN.

For this purpose, the dataset was split into three subsets: training (75%), validation (15%), and test (10%). The training subset was utilized to adjust the model parameters, while the validation subset was utilized for fine-tuning hyperparameters and mitigate overfitting. Finally, the test subset was utilized to assess the model's generalization capability, evaluating its performance on new data.

3. Model and Evaluation

3.1. Artificial Neural Networks and Optimization Algorithms

A multilayer perceptron (MLP), also known as a feedforward connected neural networks (FCNN), is a fundamental architecture in deep learning in which every neuron in one layer is connected to all neurons in the next layer, allowing the network to learn nonlinear and complicated relationships in the data [18].

The training concept of MLP is the process of optimizing weights and biases to minimize the loss function, and is usually accomplished utilizing the backpropagation method, a gradient-based optimization algorithm [19]. Backpropagation applies the chain rule to compute the gradient of the loss function for each weight, allowing the network to iteratively adjust its parameters [20]. However,

standard backpropagation can be slow to converge or unstable, which has led to the development of advanced optimization algorithms.

These optimizing algorithms include 1) Levenberg-Marquardt (trainlm), which combines gradient descent and Gauss-Newton methods for fast convergence, but requires significant memory; 2) Conjugate Gradient with Polak-Ribière Updates (traincgp), which is memory efficient and suitable for large networks; 3) Gradient Descent with Momentum and Adaptive Learning Rate (traingdx), which utilizes momentum to accelerate convergence and adapts the learning rate dynamically; 4) One-Step Secant (trainoss), which approximates the Hessian matrix to reduce computational complexity; 5) BFGS Quasi-Newton (trainbfg), a second-order optimization method that approximates the inverse Hessian for faster convergence; 6) Conjugate Gradient with Powell-Beale Restarts (traincgb), which periodically resets the search direction to avoid stagnation; 7) Gradient Descent with Adaptive Learning Rate (traingda), which adjusts the learning rate based on gradient behavior; 8) Resilient Backpropagation (trainrp), which updates weights based on the sign of the gradient rather than its magnitude, making it robust to gradient vanishing; and 9) Conjugate Gradient with Fletcher-Reeves updates (traincgf), another conjugate gradient method, ensures efficient optimization [21–25].

The addressed optimization algorithms are implemented in various machine learning frameworks, such as MATLAB's neural network toolbox, and are chosen based on problem requirements, including network size, data complexity, and computational constraints. For example, Levenberg-Marquardt (trainlm) algorithm is often utilized for small to medium-sized networks because of its speed, whereas Conjugate Gradient with Polak-Ribière Updates (traincgp) and Conjugate Gradient with Powell-Beale Restarts (traincgb) algorithms are preferred for large networks because of their memory efficiency. Also, the choice of optimization algorithm depends on the characteristics of loss surface. In addition, BFGS Quasi-Newton (trainbfg), one of second-order methods is effective on smooth, convex surfaces, whereas Gradient Descent with Momentum and Adaptive Learning Rate (traingdx), a first-order method, is more versatile on non-convex terrain [23,26,27].

3.2. Measures of Accuracy

To evaluate the performance of developed model, the authors employed two main statistical measures of accuracy, namely the coefficient of determination (R2) (Eq. 1) and the root mean square error (RMSE) (Eq. 2). R² quantifies the proportion of variance in the dependent variable that can be predicted by the independent variables, providing insight into the explanatory power of developed model. Also, it assesses the predictive accuracy by comparing the model's performance to the mean of the measured data, with values closer to one indicating a better fit [7,28]. In addition, RMSE provides a simple way to interpret a predictive accuracy by measuring the average size of the error between the predicted and measured values [29,30].

$$R^{2} = \frac{\sum_{i=1}^{n} [(Z_{i^{*}} - \overline{Z_{i^{*}}})(Z_{i} - \overline{Z_{i}})}{\sum_{i=1}^{n} [(Z_{i^{*}} - \overline{Z_{i^{*}}})^{2} \sum_{i=1}^{n} (Z_{i} - \overline{Z_{i}})^{2}} (1)$$

$$RMSE = \frac{1}{n} \sum_{i=1}^{n} (Z_{i} - \overline{Z_{i}})^{2} (2)$$

Where, Z_{i^*} = The predicted values, Z_i = The measured values, \overline{Z}_i = The mean of the measured values, $\overline{Z_{i^*}}$ = The mean of the predicted values, n = The number of data available.

In addition, the authors incorporated ion balance (a chemical index) to assess the model's predictive ability to maintain chemical balance, which is especially important for applications involving water quality or environmental chemistry. Also, the addressed measurements provide a comprehensive assessment of the model's predictive accuracy and reliability.

Ionic equilibrium is often evaluated via the charge balance (CB) index (Eq. 3), which is an important metric for assessing chemical consistency of a solution, especially in water quality researches [31]. The interpretation of charge balance values depends on specific thresholds defined for the analysis context such as Eq.4 and Eq. 5. $CB = \frac{(\sum c - \sum A)}{(\sum c + \sum A)} \times 100 \text{ (3)}$

$$CB = \frac{(\sum c - \sum A)}{(\sum c + \sum A)} \times 100 (3)$$

$$\sum C = \frac{Mg^{2+}}{12.15} + \frac{Ca^{2+}}{20.04} + \frac{K^{+}}{39.01} + \frac{Na^{+}}{22.99}$$
(4)
$$\sum A = \frac{Cl^{-}}{35.45} + \frac{So_{4}^{2-}}{48.03} + \frac{HCO_{3}^{-}}{61.02} + \frac{NO_{3}^{-}}{62}$$
(5)

Where, ΣC = sum of cations (mg/L) and ΣA = sum of anions (mg/L).

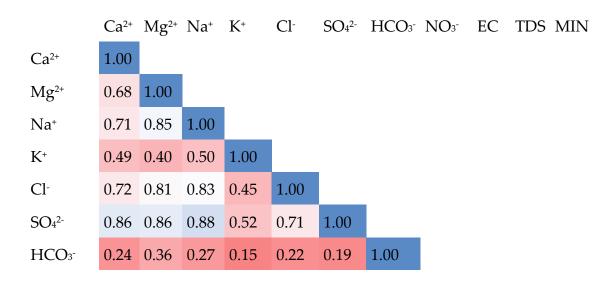
For instance, a |CB| value of less than 5 indicates a good ionic balance, reflecting a high degree of chemical consistency. A |CB| between 5 and 8 suggests a moderate ionic balance, while a |CB| greater than 8 signifies a poor ionic balance, indicating potential issues with the chemical composition. However, these thresholds can vary depending on the research's requirements. In other cases, |CB| < 6 may be considered good, $6 \le |CB| \le 12$ is moderate, and |CB| > 12 is poor. For more lenient assessments, thresholds such as |CB| < 10 (good), $10 \le |CB| \le 20$ (moderate), and |CB| > 20 (poor) might be applied. These ranges help to classify the reliability of ion balances, ensuring the accuracy and validity of chemical data in environmental or analytical studies [32].

4. Methodology

4.1. Model Development

In this research, a Levenberg-Marquardt backpropagation multilayer perceptron (LMBP-MLP) was trained to predict the concentrations of important ions (i.e., Ca^{2+} , Mg^{2+} , Na^+ , SO_4^{2-} , Cl^- , K^+ , HCO_3^- , and NO_3^-) in water utilizing measurements of MIN, TDS, and some ions (i.e., Mg^{2+} , Na^+ , and SO_4^{2-}). The selection of appropriate feature variables and hyperparameters, including the number of neurons in the hidden layer, the type of activation function, the type of learning function, and the learning rate of ANN model, played a critical role in the model development.

The selection of appropriate features for each ANN model (LMBP-MLP) was guided by a correlation heatmap of Pearson correlation coefficients, ensuring that the most relevant variables were utilized for each ion prediction (see Figure (2)). A correlation matrix helps to identify factors that exhibit statistical association based on the Pearson correlation coefficient, which quantifies the linear relationship between two variables. Figure 2 demonstrated that MIN and TDS showed strong correlations with most ions, especially SO₄²⁻, Na⁺, Mg²⁺, Ca²⁺, and Cl⁻, with correlation coefficients exceeding 0.85. This suggested that these elements originated primarily from evaporite deposits, such as gypsum and saltpeter, found in the Triassic and Aptian formations, and from limestone layers embedded within the Baremian-Aptian sandstone formations, which constitute the most important aquifer system in the region. In contrast, the remaining factors showed weak correlations with coefficients less than 0.53. In particular, NO₃⁻ and HCO₃⁻ displayed low correlation values, reflecting that the two substances have different origins. In addition, Nitrates (NO₃⁻) mainly comes from agricultural fertilizers, while bicarbonates (HCO₃⁻) is produced by dissolution of calcite, the main mineral matrix of the Barem-Aptian sandstone. Potassium ions (K⁺), derived from the dissolution of the rare mineral sylvin, is present in low concentrations and contribute minimally to MIN and TDS.



NO_{3}	0.23	0.30	0.18	-0.01	0.25	0.11	0.12	1.00			
EC	0.89	0.92	0.92	0.52	0.86	0.95	0.32	0.28	1.00		
TDS	0.88	0.92	0.93	0.52	0.86	0.95	0.33	0.29	0.99	1.00	
MIN	0.88	0.92	0.92	0.52	0.85	0.95	0.33	0.28	0.99	0.99	1.00

Figure 2. Heatmap of statistical associations based on the Pearson correlation coefficient (PCC).

The heatmap also highlighted a very strong correlation for (MIN, SO_4^{2-}) and (TDS, SO_4^{2-}) (PCC>0.95), followed by Na⁺ and Mg²⁺, both of which exhibit significant correlations (PCC>0.92) with MIN and TDS. Both Na⁺ and Mg²⁺ displayed important correlations with SO_4^{2-} (PCC>0.86). Also, a notable correlation was observed between Ca²⁺ and Cl⁻ with MIN and TDS (PCC>0.85). The moderate correlation was provided between SO_4^{2-} and Cl⁻ (PCC>0.71). In contrast, the remaining ions (i.e., K^+ , HCO₃⁻, and NO₃⁻) did not display any significant correlations with the other studied elements. Depending on results of this analysis, feature variables suitable for each ANN model were selected as shown in Table 1.

Table 1. Features and ou	put variables for the develo	ped ANN models of all major ions.
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ANN Model	Features	Output
LMBP-MLP1	TDS, MIN	SO ₄ ² -
LMBP-MLP2	TDS, MIN, SO ₄ ²⁻	Mg^{2+}
LMBP-MLP3	TDS, MIN, SO ₄ ²⁻	Na ⁺
LMBP-MLP4	TDS, MIN, SO ₄ 2-, Na+, Mg ²⁺	Ca ²⁺
LMBP-MLP5	TDS, MIN, SO ₄ ²⁻ , Na ⁺	Cl-
LMBP-MLP6	TDS, MIN, SO ₄ ²⁻ , Na ⁺	K ⁺
LMBP-MLP7	TDS, MIN, Mg ²⁺	HCO ₃ -
LMBP-MLP8	TDS, MIN, Mg ²⁺	NO ₃ -

4.2. Hyperparameters Selection

The activation functions utilized in all developed models include the sigmoid activation function in the hidden layer and the linear transfer function in the output layer. The learning rate was set to 0.001 to ensure stable and efficient training. To select the most suitable training algorithm, the authors trained a model with two neurons utilizing the first three variables (i.e., SO42-, Na+, and Mg2) listed in Table 1. The training algorithm illustrating the highest performance criterion was selected for optimization process. In this research, the evaluated training algorithms included trainlm, traincgp, traingdx, trainoss, trainbfg, traincgb, traingda, trainrp, and traincgf, respectively. Figure 3 shows the performance of training algorithms in this research. Results showed that Levenberg-Marquardt (trainlm) algorithm was superior to remaining algorithms and was the recommended choice for this research.

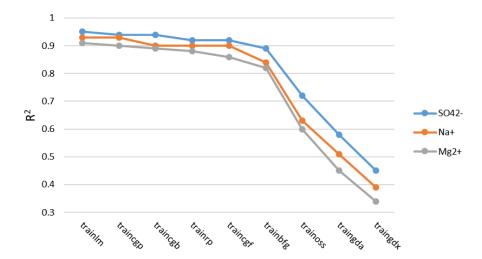


Figure 3. The performance of training algorithms.

Choosing the number of neurons in the hidden layer is a critical factor in determining the accuracy of model training process. An excessive number of neurons can lead to overfitting, where the model memorizes noise instead of learning meaningful patterns. Conversely, if there are too few neurons, the model's capacity to capture complex relationships is limited, which can lead to underfitting [33–35].

To enhance the performance of each model, the optimal number of neurons in the hidden layer was determined using a trial-and-error approach, with the number of neurons ranging from 1 to 30. Figure 4 illustrates the effect of difference in the number of hidden neurons on model accuracy in the validation phase. This showed how changing the number of hidden neurons affected the model accuracy. Based on the results, the optimal number of hidden neurons for each model was identified as follows: five neurons for LMBP-MLP1, LMBP-MLP2, and LMBP-MLP3; three neurons for LMBP-MLP6; and eighteen neurons for LMBP-MLP8.

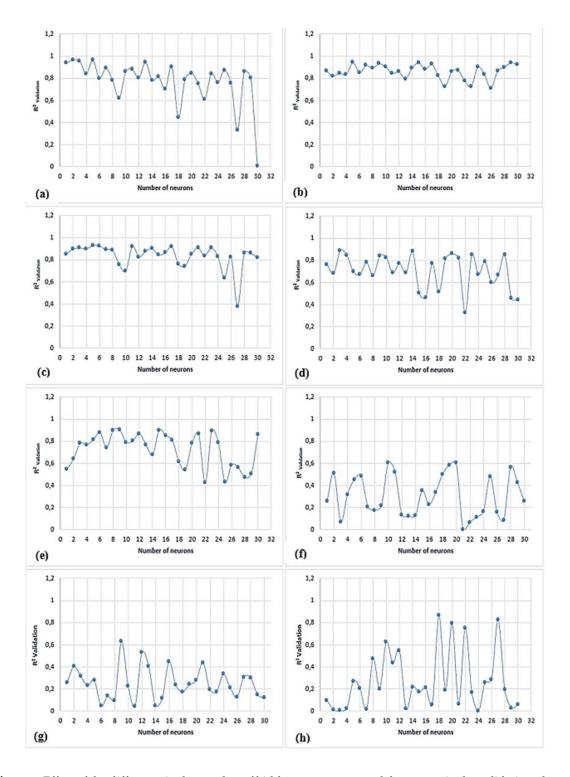


Figure 4. Effect of the difference in the number of hidden neurons on model accuracy in the validation phase (a) SO42-, (b) Mg2+, (c) Na+, (d) Ca2+, (e) Cl-, (f) K+, (g) HCO3-, and (h) NO3-.

Figure 5 illustrates the flowchart of modeling process. Initially, SO_4^{2-} concentrations were predicted based on measured TDS and MIN values. Afterwards, Na^+ and Mg^{2+} concentrations were computed utilizing measured (i.e., TDS and MIN) and predicted (SO_4^{2-}) values. Also, the prediction of Ca^{2+} concentrations incorporated measured (i.e., TDS and MIN) and predicted (i.e., SO_4^{2-} , Na^+ , and Mg^{2+}) values. Similarly, Cl^- and K^+ concentrations were predicted utilizing measured TDS and MIN, along with predicted SO_4^{2-} and Na^+ values. Finally, NO_3^- and HCO_3^- concentrations were calculated based on measured TDS and MIN, integrating with the predicted Mg^{2+} values.

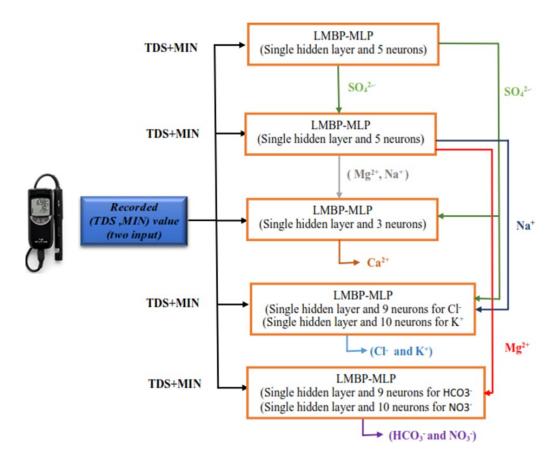


Figure 5. Flowchart of modeling process.

5. Results and Discussion

Table 2 presents the results of developed ANN models for all major ions in the Aflou_Syncline region of Algeria, utilizing the coefficient of determination (R²) and root mean square error (RMSE) as evaluation metrics in training, validation, and test datasets. The predictive accuracy of model performance varied greatly based on different ANN models.

Table 2. Results of the developed ANN models for all major ions.

	Training		Vali	dation	7	Γest	All		
ANN Model	\mathbb{R}^2	RMSE (mg/L)	\mathbb{R}^2	RMSE (mg/L)	\mathbb{R}^2	RMSE (mg/L)	\mathbb{R}^2	RMSE (mg/L)	
LMBP-MLP1	0.923	65.730	0.964	56.970	0.842	53.660	0.936	63.368	
LMBP-MLP2	0.921	14.890	0.943	11.800	0.980	12.840	0.924	14.274	
LMBP-MLP3	0.916	20.230	0.927	17.270	0.759	14.960	0.916	19.346	
LMBP-MLP4	0.867	21.990	0.887	23.510	0.945	36.460	0.892	24.034	
LMBP-MLP5	0.865	44.640	0.902	43.600	0.895	30.530	0.872	43.296	
LMBP-MLP6	0.533	2.990	0.601	2.850	0.045	6.480	0.441	3.482	
LMBP-MLP7	0.300	64.250	0.630	37.760	0.366	41.720	0.330	59.029	
LMBP-MLP8	0.325	43.400	0.865	40.870	0.004	40.460	0.523	41.886	

The following ANN models, including LMBP-MLP1, LMBP-MLP2, LMBP-MLP3, LMBP-MLP4, and LMBP-MLP5, presented the accurate performance, as indicated by high R² values and relatively low RMSE values across all subsets. This implies that addressed models can accurately predict the concentrations of these ions (SO4²-, Na+, Mg²+, Cl⁻-, and Ca²+) in the Aflou_Syncline region. However, the addressed models, including LMBP-MLP6, LMBP-MLP7, and LMBP-MLP8, provided poorer

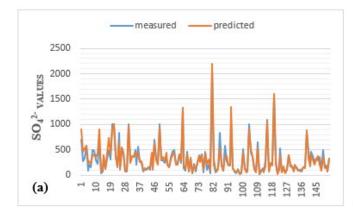
performance. That is, low R^2 values and high RMSE values indicated that these models have difficulty capturing the variability of K+, HCO3-, and NO3- ions.

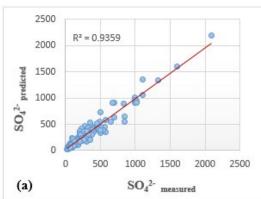
Figure 6 illustrates the comparison between predicted and measured values of all major ions in the testing phase. Utilizing line plots and scatter plots. It provides a comparative assessment of measured and predicted ion concentrations, providing important insights into the hydrochemical dynamics of the region.

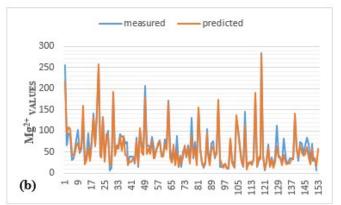
For sulfate (SO_4^{2-}), magnesium (Mg^{2+}), and sodium (Na^+) in the testing phase, the line plots for individual ANN models (i.e., LMBP-MLP1, LMBP-MLP2, and LMBP-MLP3) supplied the strong visual agreement between observed and predicted values, highlighting the reliability of LMBP-MLP1, LMBP-MLP2, and LMBP-MLP3. Scatterplots further quantified this alignment, showing strong predictive accuracy with high coefficients of determination (R^2 =0.940 for SO_4^{2-} , R^2 =0.920 for Mg^{2+} , and R^2 =0.910 for Na^+).

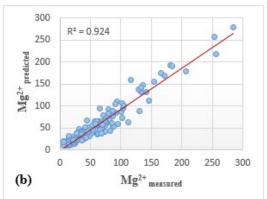
For calcium (Ca²⁺) and chloride (Cl⁻), The analysis results revealed partial model efficacy. The Ca²⁺ line plot is broadly consistent with the measured trend, but the deviations in 70–80 of samples show that it has limited ability to capture regional differences. The Ca²⁺ scatterplot (R²=0.890) confirmed this, suggesting that high concentrations of outliers led to lower accuracy in the higher ranges. Similarly, the Cl⁻ scatterplot tracks the trend adequately, but underestimates the sharp peak at 90–100 of samples, as evidenced by the clustering of scatterplot outliers (R²=0.870) above the regression line. These discrepancies may arise from extreme values or incomplete representation of location-specific geochemical interactions, highlighting the need for targeted fine-tuning to improve prediction performance for these ions.

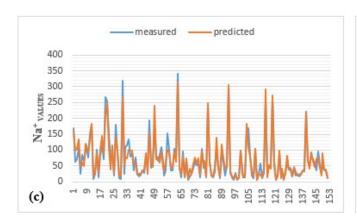
In contrast to the strong correlations for SO₄²⁻, Mg²⁺, and Na⁺, some models (i.e., LMBP-MLP6, LMBP-MLP7, and LMBP-MLP8) provided poor performance for the scatterplots of potassium (K⁺), bicarbonate (HCO₃⁻), and nitrate (NO₃⁻). The K⁺ scatterplot (R²=0.440) showed minimal agreement with the observed data and failed to reproduce the variability. Also, The HCO₃⁻ scatterplot (R²=0.330) provided a nearly random dispersion, indicating a fundamental flaw in either variable selection or mechanical assumptions. Although NO₃⁻ scatterplot (R²=0.520) supplied a marginal improvement, LMBP-MLP8 systematically underestimated maximum concentrations, due to unaccounted for anthropogenic or biogeochemical influences.

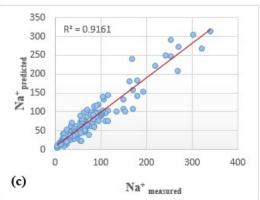


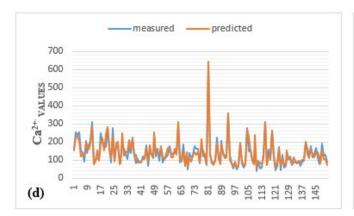


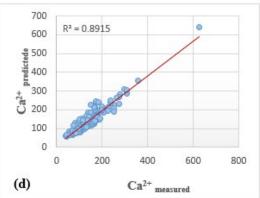


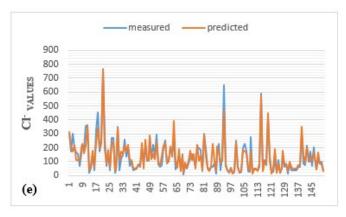


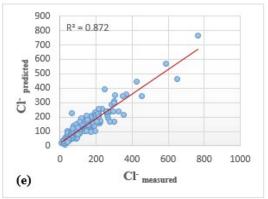












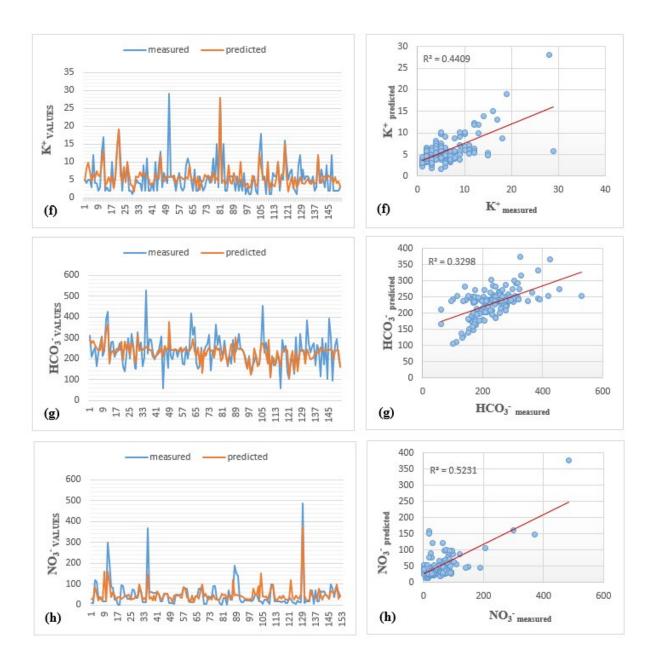


Figure 6. Comparison between predicted and measured values of all major ions for the all dataset (a) SO42-, (b) Mg2+, (c) Na+, (d) Ca2+, (e) Cl-, (f) K+, (g) HCO3-, and (h) NO3-.

Figure 7 presents the sorted charge balance (CB) values of all samples utilizing the predicted ion concentrations. It displays aligned equilibrium ion values for 153 water samples from the Aflou_Syncline region employing the predicted ion concentrations. This ion balance serves as an indicator of data quality and the accuracy of ion estimation, with values closer to 0 indicating a better match between positive and negative charges. Samples are classified into three groups: "Good" (green), "Moderate" (yellow), and "Poor" (red) based on their deviation from ion balance. Most of the samples (84%) were classified as "Good", suggesting the overall reliability of data and satisfactory performance of the developed models. However, 11% of the samples fell into the "Moderate" category, indicating potential problems such as measurement errors, the presence of uncounted ions, or sample degradation. A small number of samples (5%) were classified as "Poor," indicating serious errors that required reevaluation.

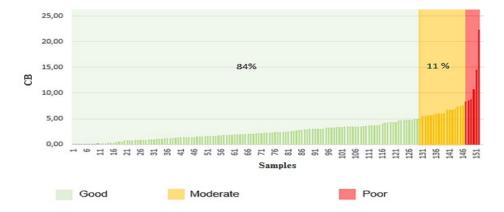


Figure 7. The sorted charge balance (CB) values of all samples utilizing the predicted ions.

5.1. Testing the Developed Model in Adjacent Areas

To evaluate the generalized ability of developed models, their performance was tested utilizing 20 water samples collected from three external locations within the research area. These locations share the same geological structure as the main research area, but show some petrological differences. The selected locations include Madna (6 samples), Aflou (4 samples), situated southwest of the Aflou_Syncline, and Ain Madhi (10 samples), situated further south. The predictive accuracy of developed models was assessed utilizing the coefficient of determination (R²), and the results were presented in Figure 8.

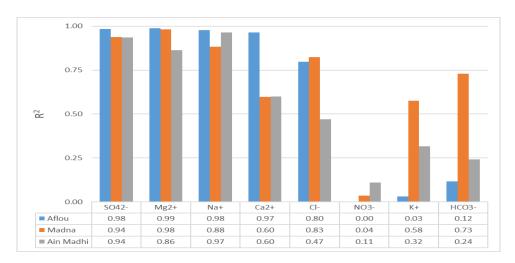


Figure 8. Comparison of developed models performance in adjacent areas (Aflou, Madena, and Ain Madhi).

Our results presented significant differences in model performance across different ions and locations, highlighting that both ion-specific behavior and location-specific characteristics have an impact. The applied models showed high prediction accuracies for SO_4^{2-} , Mg^{2+} , and Na^+ , and consistently high R^2 values at all points, indicating that the relationships between these ions and the feature variables were stable.

For Ca^{2+} and Cl^- , the applied models performed well in Aflou and Madna, but supplied poor accuracy in Ain Madhi, suggesting location-specific hydro-geochemical factors that influence ion concentrations. In contrast, the applied models performed poorly for NO_3^- and K^+ , with R^2 values close to 0 at three locations. This is likely due to external influences such as agricultural activities (NO_3^-) and local mineral dissolution (K^+) .

The prediction of HCO₃⁻ varied significantly, supplying moderate performance at Madna but low accuracy at Aflou and Ain Madhi, indicating that hydro-geochemical control may be possible depending on the location. The differences of performance in applied models can be attributed to

differences in rock composition, groundwater flow dynamics, and local environmental factors affecting ion concentrations.

The geological characteristics at Ain Madhi may provide more pronounced variations than those of Aflou and Madna, which may lead to inconsistencies that the applied models may not fully capture. In addition, location-specific geochemical processes, anthropogenic influences (e.g., fertilizer use affecting NO_3^-), and varying mineral dissolution rates (e.g., sylvite dissolution for K^+) may contribute to the observed discrepancies.

Overall, the applied models demonstrated strong predictive ability for SO₄²⁻, Mg²⁺, and Na⁺, whereas it performed poorly for Ca²⁺, Cl⁻, HCO₃⁻, NO₃⁻, and K⁺, especially at Ain Madhi. These results highlighted the need for additional location-specific calibrations to improve model accuracy for specific ions and account for local hydro-geochemical variations.

To estimate the charge balance (CB) in the adjacent areas of Aflou, Madna, and Ain Madhi, the authors utilized Figure 8 to identify ions with high predictive performance (R²>0.600) and replaced ions with low performance utilizing measured data to ensure accuracy.

The values of selected ions utilized for Ionic Balance calculations varied depending on locations. In Aflou, the selected ions were SO_4^{2-} , Cl^- , Ca^{2+} , Mg^{2+} , Na^+ , and K^+ . For Ain Madhi, only SO_4^{2-} , Ca^{2+} , Mg^{2+} , and Na^+ satisfied the selection criteria, while in Madna, the chosen ions included SO_4^{2-} , HCO_3^- , Cl^- , Ca^{2+} , Mg^{2+} , and Na^+ . The results of the Ionic Balance and their evaluation are presented in Tables 3, 4, and 5.

Table 3. The value of charge balance (CB) and their evaluation of samples (Aflou).

A #10.0	SO ₄ 2-	NO ³⁻	HCO ₃ -	Cl-	Ca ²⁺	Mg ²⁺	Na⁺	K +	CB	Evaluatio
Area	Pred.	Meas.	Meas.	Pred.	Pred.	Pred.	Pred.	Pred.	%	n
- A (1	105	5	240	45	88	23	23	7	0.03	Good
	410	30	326	135	152	68	82	7	3.54	Good
Aflou -	906	15	273	210	292	146	168	14	7.45	Moderate
_	393	14	239	190	153	61	86	7	3.23	Good

Table 4. The value of charge balance (CB) and their evaluation of samples (Aflou).

A #0.0	SO ₄ 2-	NO ³⁻	HCO ³⁻	C1-	Ca ²⁺	Mg ²⁺	Na⁺	K +	СВ	Evaluati
Area	Pred.	Meas.	Meas.	Meas.	Pred.	Pred.	Pred.	Meas.	%	on
	434	9	237	145	206	83	102	5	11.60	Poor
	426	10	232	145	206	80	104	5	11.90	Poor
	123	13	212	70	95	27	28	2	0.07	Good
	124	16	185	93	95	27	28	2	1.57	Good
Ain	1677	4	237	400	177	298	270	15	4.87	Good
Madhi	1227	10	217	370	576	173	247	12	15.28	Poor
Mauili	291	13	247	240	187	38	140	6	4.51	Good
	281	2	241	205	183	38	131	6	7.40	Moderat
	201		241	203	163	36	131	0	7.40	e
	352	5	162	220	163	50	104	15	2.65	Good
	257	34	144	155	128	41	58	6	0.77	Good

Table 5. The value of charge balance (CB) and their evaluation of samples (Madna).

Area										Evaluati
7 II Cu	Pred.	Meas.	Pred.	Meas.	Pred.	Pred.	Pred.	Meas.	%	on
	888	7	230	354	199	142	221	12	1.28	Good
Madna	903	84	281	250	236	148	209	14	2.44	Good
	631	54	263	257	185	106	155	12	1.11	Good

	437	17	247	198	213	82	107	7	7.77	Moderat e
	629	71	266	250	181	104	158	14	1.64	Good
_	65	3	167	29	73	16	14	4	6.72	Moderat e

Each table (Tables 3, 4, and 5) provides predicted and measured ion concentrations, the calculated charge balance, and their evaluation of samples. In Aflou, 75% of the samples were rated as "Good", indicating reliable data and accurate ion estimates, 25% were rated as "Moderate" and 0% were rated as "Poor". In contrast, Ain Madhi featured a wider range of sample quality, featuring 50% "Good" samples, along with 30% "Poor" and 20% "Moderate" ratings. This suggests potential data quality issues specific to Ain Madhi, which could arise from sample contamination or measurement errors. Madna, similar to Aflou, gave mostly "Good" ratings (66.67%), with 33.33% being "Moderate" samples, suggesting generally reliable data with localized discrepancies. The variability in charge balance across all regions highlights the importance of ion balance analysis as a tool to assess data quality and validate ion estimates. The presence of "Poor" and "Moderate" samples highlights the need for further investigation to identify and correct potential problems to ensure the accuracy and reliability of hydrochemical data.

6. Conclusion

In Algeria, groundwater remains critical for irrigation, but groundwater quality in areas such as the Aflou region is increasingly compromised by salinity and agricultural contamination, threatening agricultural sustainability. Conventional monitoring methods that rely on expensive sampling campaigns and laboratory analyses highlight the urgent need for innovative and cost-effective solutions to ensure water security. Artificial intelligence (AI), especially artificial neural networks (ANN), has emerged as a revolutionary tool in the field of hydrochemistry.

In this research, the authors introduced a novel algorithm to predict eight major ions (Ca^{2+} , Mg^{2+} , Na^+ , K^+ , HCO_3^- , SO_4^{2-} , Cl^- , and NO_3^-) utilizing only two accessible parameters (i.e., total dissolved solids (TDS) and mineralization (MIN). The Levenberg-Marquardt backpropagation multilayer perceptron (LMBP-MLP) model with ion-specific customized architecture achieved robust predictive accuracies for SO_4^{2-} , Mg^{2+} , Na^+ , Ca^{2+} , and Cl^- (R^2 and $NSE \ge 0.87$), proving its usefulness in real-time monitoring. However, predictive accuracies for K^+ , HCO_3^- , and NO_3^- were less reliable ($R^2 \le 0.50$), most likely due to complex environmental interactions and low concentrations leading to little statistical significance. The validation via charge balance analysis confirmed strong ionic balance in 95% of predictions, but 5% showed discrepancies requiring improvement.

Spatial tests across three locations (i.e., Aflou, Madena, and Ain Madhi) showed consistent accuracy for SO_4^{2-} , Mg2+, and Na^+ , moderate performance for Ca^{2+} and Cl^- , variable results for K^+ and HCO_3^- , and overall poor prediction of NO_3^- . These results highlighted the model's adaptability to regions such as Aflou and Madna, while also emphasizing the need for expanded geographic data to improve generalization. Despite these limitations, this algorithm has made great strides in water resource management in salinity-affected areas. Direct TDS and MIN measurements enable early detection of important ions (Ca^{2+} , Mg^{2+} , Na^+ , SO_4^{2-} , and Cl^-), providing three key benefits: cost savings, adaptability, and efficiency.

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Abbreviations

The following abbreviations are used in this manuscript:

TDS Total dissolved solids
MIN Mineralization

MLP Multilayer perceptron

LMBP Levenberg-Marquardt backpropagation

AI Artificial intelligence
WQI Water quality index
ANN Artificial neural networks
EC Electrical conductivity

RBF-NN Radial basis function neural networks

PNN Probabilistic neural networks

FCNN Feedforward connected neural networks

R² Coefficient of determination RMSE Root mean square error CB Charge balance

LMBP-MLP Levenberg-Marquardt backpropagation multilayer perceptron

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