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[Joanna Rosik](#) * and [Sylvia Stegenta-Dąbrowska](#)

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Article

Optimizing the Composting Process Emissions—Process Kinetics and Artificial Intelligence Approach

Joanna Rosik * and Sylwia Stegenta-Dąbrowska ¹

Department of Applied Bioeconomy, Wrocław University of Environmental and Life Sciences, Chelmońskiego Street 37a, 51-630 Wrocław, Poland; sylwia.stegenta-dabrowska@upwr.edu.pl (S.S.-D.)

* Correspondence: 118318@student.upwr.edu.pl (J.R.)

Abstract: Although composting has many advantages in the treatment of organic waste, there are still many problems and challenges associated with emissions, like NH_3 , VOCs, and H_2S , as well as greenhouse gases such as CO_2 , CH_4 , and N_2O . One promising approach to enhancing composting conditions is used of novel analytical methods based on artificial intelligence. To predict and optimize the emissions (CO , CO_2 , H_2S , NH_3) during composting process kinetics thought mathematical models (MM) and machine learning (ML) models were utilized. Data about everyday emissions from laboratory composting with compost's biochar with different incubation (50, 60, 70 °C) and biochar doses (0, 3, 6, 9, 12, 15% d.m.) were used for MM and ML models selections and training. MM has not been very effective in predicting emissions, (R^2 0.1 - 0.9), while ML models such as acritical neural network (ANN, Bayesian Regularized Neural Network; R^2 accuracy CO :0,71, CO_2 :0,81, NH_3 :0,95, H_2S :0,72) and decision tree (DT, RPART; R^2 accuracy CO :0,693, CO_2 :0,80, NH_3 :0,93, H_2S :0,65) have demonstrated satisfactory results. For the first time, the ML models to predict CO and H_2S during composting were demonstrated. Further research in a semi-scale and field study composting with biochar is needed to improve the accuracy of development models.

Keywords: Machine learning; biochar application; greenhouse gases; composting optimizing

1. Introduction

Composting process is one of the most popular ways to manage biodegradable wastes because it is highly effective, low risk and environmentally beneficial. The mechanism is compounded and involves various interrelated processes, including microbiological, physicochemical, and thermodynamic processes [1]. During the process of composting, microorganisms release heat and energy as they break down organic materials. A series of transformations that occur during aerobic stabilization results in the formation of carbon dioxide and stable forms of carbon, which facilitate the decomposition and mineralization of organic matter leading to the formation of stable humic substances [2]. Throughout the composting process, a notable amount of heat is produced, effectively sustaining a temperature above 50°C for an extended duration. As a result, any harmful bacteria, diseases, or insect eggs that may be present in the composting material are thoroughly eliminated, yielding a final product that is entirely safe and innocuous [3].

In the composting process, various factors such as initial moisture content, C:N ratio, bacterial agents, particle size of composting materials, composting duration, and other indicators play a critical role in determining the success or failure, efficiency, and quality of compost products [4–6]. Oxygen stabilization is performed on an industrial scale under controlled conditions to maintain proper levels of relevant technological parameters mentioned before.

Despite many advantages, composting process may cause emissions of hazardous odors and greenhouse gases like NH_3 , H_2S , CO and CO_2 which is especially environmentally disadvantageous [7–9]. Therefore, it is particularly important to determine what composting process conditions are the most optimal from the point of view of reducing gaseous emissions. Currently, a popular solution used to reduce emissions of greenhouse gases and volatile organic compounds is biochar, which can retain gaseous substances on its surface due to its physicochemical properties [10–12]. At this point, there is still a lack of research to determine the ideal parameters for biochar production, dosage and

incubation temperature of the composted material. In addition, the relationships between these parameters are very complex, making mathematical models ineffective in solving such problems.

The use of artificial intelligence (AI) to optimize various processes is becoming increasingly common. With AI, it is possible to assess and improve response conditions and maximize operational efficiency by optimizing necessary parameters, especially in agricultural and environmental sciences [13–16]. In recent years, artificial intelligence has become a popular tool for predicting various processes in waste management. In literature, various AI models have been utilized for the prediction and categorization of solid waste, composting processes, and anaerobic fermentation. Artificial intelligence methods include models such as artificial neural networks (ANN), support vector machines (SVM), decision trees (DT), K Nearest Neighbor (kNN), radial basis function (RBF) and various other ensemble learning techniques.

Lin et al. investigated the application of ANNs to forecast significant composting process variables like composting temperature and pH. The authors developed two prediction models using ANNs and traditional multiple-linear regression (MLR) models and compared their effectiveness. The results showed that 1-day before forecasting were more accurate than 2-days and 3-days before predictions, which shows that ANNs are useful tools for short-term predictions in composting process [17]. Boniecki et al. used neural prediction of heat loss in the pig manure composting process. The models used included kNN, DT, MLP, AdaBoost, bagging and Gradient Boost. The models created by the researchers estimated the heat lost during exothermic reactions occurring during the composting process. The input data were temperature, dry organic matter, oxygen content, stream volume, carbon dioxide content and time. The most optimal results were observed for MLP with 9-5-1 structure taught with the use of optimization algorithms Back Propagation and Conjugate Gradients [18]. Ding et al. examined the possibilities of using machine learning models to optimize the kitchen waste composting maturity. Measurable parameters such as daily temperature pH, moisture content, total nitrogen, C/N, ammonia, total organic carbon and seen germination index were used to build models. The study revealed that different stages of the composting process should be modeled using different parameters and the model-based system exhibited better maturity of the final material [19]. In addition, predictions related to the optimization of the composting process have been widely reported in many studies, but there is still a lack of information on the possibility of using artificial intelligence to determine the kinetic parameters of emitted gas from compost.

The present study aims to compare process kinetic thought mathematical models (MM) and machine learning (ML) models to predict the emissions (CO , CO_2 , H_2S , NH_3) during the first 10 days of composting with compost's biochar addition. Data about everyday emissions for modeling were collected during laboratory composting with compost's biochar with different incubation (50, 60, 70 °C) and biochar doses (0, 3, 6, 9, 12, 15% d.m.). This study confirms that the use of AI for optimizations and limitations of the emissions during composting has good potential and can be used to improve the safety of the process.

2. Materials and Methods

2.1. The Experiment Design and Procedure

The research on kinetics prediction (section 2.4) and machine learning model training (section 2.5) relied on data from published sources [20]. The study centers on the influence of compost's biochar addition to feedstock, and how it impacts CO , CO_2 , H_2S , and NH_3 emissions during the early stages of laboratory composting. The composting experiments used aa feedstock mix of 90% green waste and 10% sewage sludge acquired from a composting plant (Best-Eko, Rybnik, Poland). Various biochars (B550; B600; B650), produced at different pyrolysis temperatures, were applied at doses of 0, 3, 6, 9, 12 and 15% d.m., as seen in Figure 1. The appropriate biochar variant was added to the feedstock, placed in 1L reactors, and kept at 50, 60, or 70°C in a thermostatic cabinet for 10 days. The concentrations of CO , CO_2 , H_2S , and NH_3 were measured daily throughout the composting process and then used to calculate emissions.

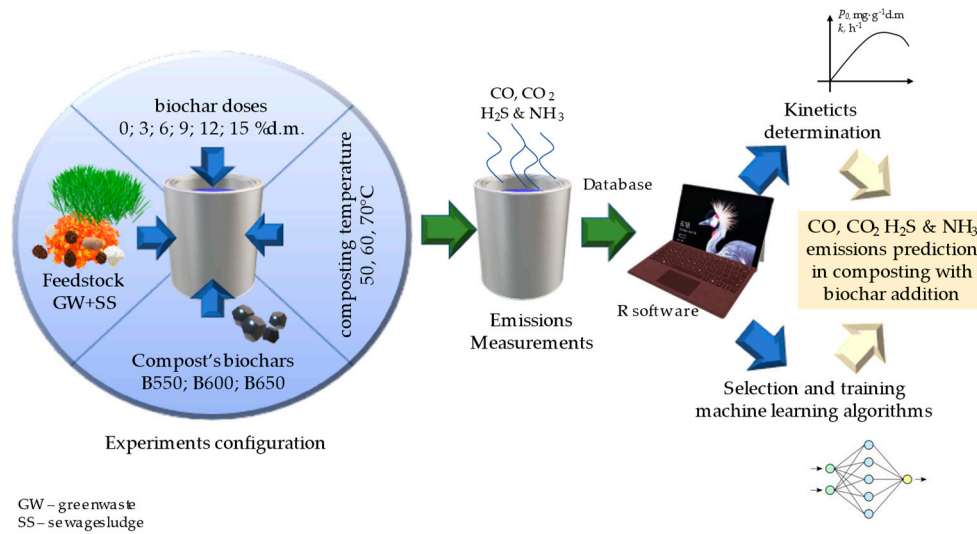


Figure 1. Experiments configurations.

2.3. Gas Production Monitoring

During the laboratory composting, everyday gas concentrations of CO, CO₂, H₂S and NH₃ were done. For gas concentration measurements the *electrochemical* gas portable analyzer was used (Nanosens DP-28 BIO; Wysogotowo, Poland). Concentrations of CO, H₂S, and NH₃ were determined in ppm in the following ranges: CO 0–2000 ppm (±20 ppm), H₂S, NH₃ 0–1000 ppm (±10 ppm), and CO₂ 0–100% (±2%). Each measurement lasted 45 s, followed by automatic cleaning of the analyzer.

2.4. Gas Production Kinetics Determination

Data for kinetic analysis were analyzed by excluding the lag-phase [21]. Nonlinear least squares regression was used to determine the kinetic parameters of CO, CO₂, H₂S & NH₃ production. The 1st-order reaction models were used. Prior research has established that the gathered data aligns well with the proposed model [22,23].

The 1st-order reaction equation for CO, CO₂, H₂S or NH₃ production is:

$$P = P_0 \cdot (1 - e^{-k \cdot t}) \quad (1)$$

where:

P – total production (CO₂, mg·g⁻¹d.m; H₂S or NH₃ μg·g⁻¹d.m),

P_0 – maximum production (CO₂, mg·g⁻¹d.m; CO, H₂S or NH₃ μg·g⁻¹d.m),

k – production (CO, CO₂, H₂S or NH₃) constant rate, (h⁻¹),

t – time, (h).

The k and P_0 , calculated from nonlinear regression, were used to calculate the average production or consumption rate (r) of CO, CO₂, H₂S or NH₃ according to:

$$r = k \cdot P_0 \quad (2)$$

where:

r – average production rate (r) of (CO₂, mg·g⁻¹d.m; CO, H₂S or NH₃ μg·g⁻¹d.m),

2.5. Data Pre-Processing

Figure 2 depicts the data processing steps. Initially, 66,048 datasets without missing data were extracted from the selected references. Subsequently, the collected data was normalized from 0 to 1 using Z-Score normalization. Finally, the dataset was randomly divided into training and testing datasets to enhance prediction accuracy, as previously reported [24]. The data was divided into training/validation/test groups in a 70%/15%/15% proportion. For the fine-tuning process, k-fold cross-validation with grid search was employed. The training dataset assisted in adjusting the hyperparameters and enhancing the prediction abilities of the model, while the testing dataset was

used to evaluate the performance of the model and select the appropriate model by comparing the RMSE and R^2 values [25].

$$R^2 = 1 - \left[\frac{\sum_{t=1}^T (y^*_t - y_t)^2}{\sum_{t=1}^T y^*_t - y_t^2} \right] \quad (3)$$

$$RMSE = \sqrt{\frac{\sum_{t=1}^T (y^*_t - y_t)^2}{T}} \quad (4)$$

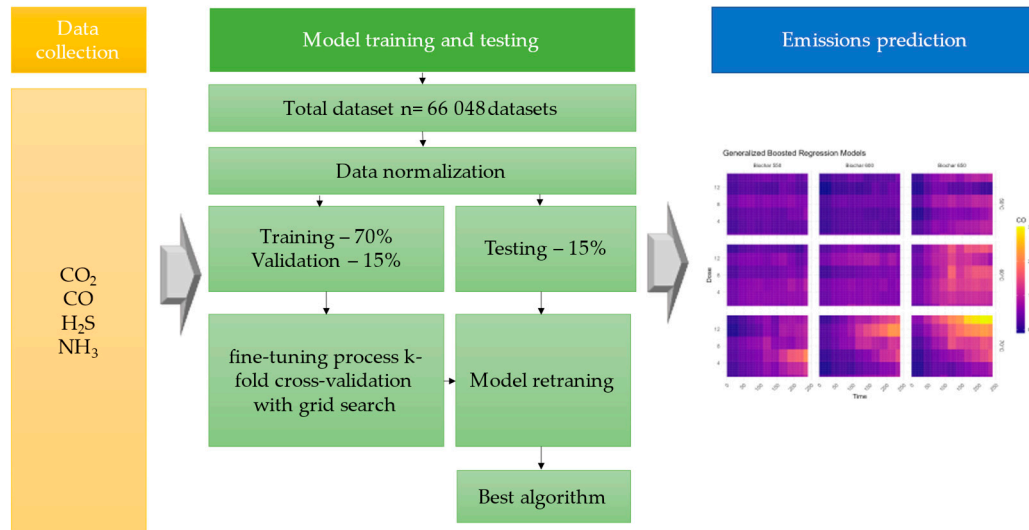


Figure 2. Machine learning flowchart for predicting emissions from composting with biochar addition.

2.6. Selection ML Model Selection and Training Machine Learning Algorithms Evaluation

In this study, ten learning algorithms were evaluated, including both machine set learning and non-set learning. To assess the viability of machine learning methods in the prediction of CO, CO₂, H₂S and NH₃ emissions during the first stage of composting various classes of methods were compared: Linear Models, Tree-Based Models (also part of Ensemble Methods), Support Vector Machines (SVM) and Neural Networks. Calculations were performed using R for Windows [26] (ver 4.3.2, Vienna, Austria) with caret [27] and h2o [28] libraries. The data used for model training related to CO, CO₂, H₂S, and NH₃ emissions from composting, which were obtained from published studies. To predict each gas emission (CO, CO₂, H₂S, and NH₃) individually, principal component analysis (PCA) was conducted to exclude irrelevant parameters. The PCA analysis indicated that observed emissions have a significant correlation and the. The use of other parameters is not justified. PCA (which is a linear dimensionality reduction algorithm) facilitated dimensions standardization and reduction of the initial complexity of the model. Moreover, it will be easier to apply the model in practice if the variables are limited to those that can be easily and cheaply implemented in composting i.e., gas emissions (Supplementary Materials Figure S1). In model training and prediction, the output and input of the model were the data about CO, CO₂, H₂S, and NH₃ emissions. During the training, when one gas emission was used as an output, the data about the other emissions were utilized as input.

The top four models (Generalized Boosted Regression Models (GBM); SVM with Radial Basis Function (RBF) Kernel Nearest Neighbor Models; Bayesian Regularized Neural Network; Recursive Partitioning and Regression Trees) were depicted as heatmaps, revealing the impact of the four variables: biochar dose, biochar type, incubation temperature, and time on gas emission. Finally, the predicted emissions were compared to the actual emissions to determine the models' accuracy.

2.7. VOS Viewer Network Map

The VOS viewer software was used to create a network map to analyze the co-occurrence of important for this study keywords. The map's occurrence and link are determined by taking into consideration the relative abundance of each keyword. Notepad was used to prepare a tab-delimited file containing all keywords, including those of low abundance, for bibliographic data of Web of Science. Further, the type of analysis employed was based on the cooccurrence of keywords that the software read as coauthors. Each keyword is depicted by a circle, with the size of the circle proportional to the frequency of occurrence of the corresponding keyword. The relative abundance of the keyword is determined by the radius of the circle, thereby enabling the visualization of the most frequently occurring keywords. The link between the keyword and the relative size is indicated by a curved line.

3. Results

3.1. Kinetics of Gas Production during Composting with Compost's Biochar (Mathematical Models)

3.1.1. Kinetics of CO Production

The results show a reverse trend in maximum CO production, with the highest production occurring at 70 °C (5830.9 $\mu\text{g CO}\cdot\text{g}^{-1}\text{d.m.}$) and the lowest at 50 °C (176.2 $\mu\text{g CO}\cdot\text{g}^{-1}\text{d.m.}$, Supplementary material Table S1). This is a typical pattern for CO production during composting - at 70 °C, the thermochemical mechanism may dominate due to the lower activity of microorganisms [22]. The observed emissions were much higher ($>1000 \mu\text{g CO}\cdot\text{g}^{-1}\text{d.m.}$) compared to the previous observation ($\sim 160 \mu\text{g CO}\cdot\text{g}^{-1}\text{d.m.}$), and the trend is similar to that shown in Figure 3. Additionally the predictions for the maximum CO production were higher than the actual emissions (Supplementary Figure 2, [20]). The differences between these parameters were much larger at 50 and 60 °C, which could be attributed to low $R^2 < 0.5$. At 70°C, a better fit with an $R^2 > 0.95$ was observed, resulting in much better predictions for the maximum CO production. The addition of B550 increased the average CO production rate by at least 20%, particularly in lower doses up to 6% d.m. (Supplementary Materials S1). The constant rate of CO production ranged from 0.001-0.031 h^{-1} , with higher values observed at 50 and 60 °C and lower values at 70 °C, consistent with previous findings on composting of bio-waste at different temperatures [29].

3.1.2. Kinetics of CO₂ Production

The use of kinetic models allowed for the determination of basic data on CO₂ production. Based on the data illustrated in Figure 4, it was observed that the maximum CO₂ emissions were highest at a temperature of 50°C (884,6 $\text{mg}\cdot\text{g}^{-1}\text{d.m.}$), while the lowest was recorded at 70°C (14,9 $\text{mg}\cdot\text{g}^{-1}\text{d.m.}$). This could be attributed to temperatures that are outside the optimal temperature range ($>59^\circ\text{C}$) for composting microorganisms [30]. While temperatures of this nature may be experienced during the few days of the composting process, usually it does not appear to have a significant impact on the work of these microorganisms. However, when the higher-temperature composting process takes longer, such as the 10 days observed in this study, there could be a notable reduction in the effectiveness of microorganisms in the composting process [23]. A similar trend was observed before when the maximum predicted CO₂ was ~ 200 at 40-50°C in composting grass with dairy cattle manure and sawdust [22]. The exception at 60°C was observed - biochar BC650 addition resulted in the greatest CO₂ emissions, but doses of 12% and 15% d.m. (Figure 4b) reduced the emissions observed. Generally, lower maximum CO₂ emissions were expected with lower biochar addition levels (3-6% d.m.). Some studies suggest that one of the mechanisms that explains the effect of CO₂ reduction in composting with biochar is that biochar can reduce C cycle activity, which in turn suppresses CO₂ emissions [31].

The addition of biochar BC600 and BC650 significantly reduced ($>50\%$) the constant rate k value of CO₂ production (Figure 4f) and the average production rate r (Supplementary material Table S1) when compared to BC550. This could indicate that biochar BC550 enhances microorganism activity better than the other variants tested and accelerates the decomposition of organic matter, even at

unborable composting temperatures, as was observed before with use of biochar from woodchips with poultry manure composting [32].

3.1.3. Kinetics of H₂S Production

The prediction of maximum H₂S production (Supplementary Material Table S1) was comparable to the actual values of 50-100 µg H₂S·g⁻¹d.m. at 50 °C and ~200 µg H₂S·g⁻¹d.m. at 60 °C with BC650, and 70 °C (Supplementary Figure 3, [20]), although the R² values observed were in the wide range of 0.1-0.8. The range of observed H₂S emissions was similar (150-300 µg H₂S·g⁻¹d.m.) to the emissions observed by Liu et al. during composting of mixed pig manure, kitchen waste, and corn stalk as raw materials [33]. The average H₂S production rate increased with the incubation temperature from ~2-4 µg H₂S·h⁻¹ at 50 °C to >10 µg H₂S·h⁻¹ at 70 °C (Supplementary Materials S1). It was observed that the addition of BC 650 had a negative impact on the maximum H₂S production of >50% (Figure 5), particularly at 60°C. The constant rate of H₂S production ranged from 0.006-0.065 h⁻¹. However, there is insufficient literature on the kinetics of H₂S production in composting, making it challenging to assess the obtained kinetic parameters effectively. This suggests that the development of bacteria that decompose sulfur compounds may have been promoted

The maximum H₂S production prediction (Supplementary Material Table S1) was similar to the real value 50-100 µg H₂S·g⁻¹d.m. in 50 °C and ~200 µg H₂S·g⁻¹d.m. in 60 °C with BC650 and 70 °C. (Supplementary Figure 3, [20]) although the observed R² were wide range 0,1-0,8. The observed emissions range was similar (150-300 µg H₂S·g⁻¹d.m.) to those observed by Liu et. al, during composting of mixed pig manure, kitchen waste, and corn stalk as raw materials [33].

The average production rate of H₂S production increased with incubation temperature ~2-4 µg H₂S·h⁻¹ at 50 °C to > 10 µg H₂S·h⁻¹ at 70 °C (Supplementary Materials S1). It was noticed that the addition of BC 650 had a negative effect on the > 50% increase in H₂S maximum H₂S production (Figure 5), especially at a temperature of 60 °C. This may indicate the promotion of the development of bacteria that decompose sulfur compounds. The H₂S production constant rate ranged from 0.006-0.065 h⁻¹. However, there is a lack of literature on the kinetics of H₂S production in composting, which makes the effective assessment of the obtained kinetic parameters difficult.

3.1.4. Kinetics of NH₃ Production

The highest NH₃ production observed was in the range of 0-350 µg NH₃·g⁻¹d.m (Figure 6; Supplementary Material Table S1), slightly exceeding the actual value of 0-200 µg NH₃·g⁻¹d.m (Supplementary Material Figure 3, [20]). This could be due to the short lag-phase and a cyclic increase in the trend, indicating the potential for ammonia production in the composting process. This effect could be the result of the observation the short lag-phase and then the cyclical increase in the trend. This could indicate the developing potential of ammonia production in the composting process. High levels of ammonia emissions were observed during the composting of chicken manure, which prevented the trend from being visible since day one [34]. However, in other studies where chicken feces were composted with biochar, a lag phase was observed [35]. The primary reason for the NH₃ loss was the acidity (pH); in the first study, the pH was 8, and in the second study, it was 9, similar to our results [20]. Despite a relatively low R² value (0.1-0.9), the predicted potential for ammonium seems to be useful for composting estimations.

The addition of biochar to the composting matrix typically increases the observed maximum NH₃ production (Supplementary material Table S1). The highest NH₃ production was observed at 60°C with the addition of BC550 (Supplementary Materials S1). At 70°C, the addition also promoted the average production rate - the more biochar added, the higher the NH₃ emissions. As observed previously [20], during the first 10 days of composting, the addition of biochar increases total ammonia emission, but after 15 days, ammonia production ceases, and only the control variant without biochar continues to produce ammonia. As a result, after 30 days, cumulative ammonia emissions with biochar addition were much lower than the control.

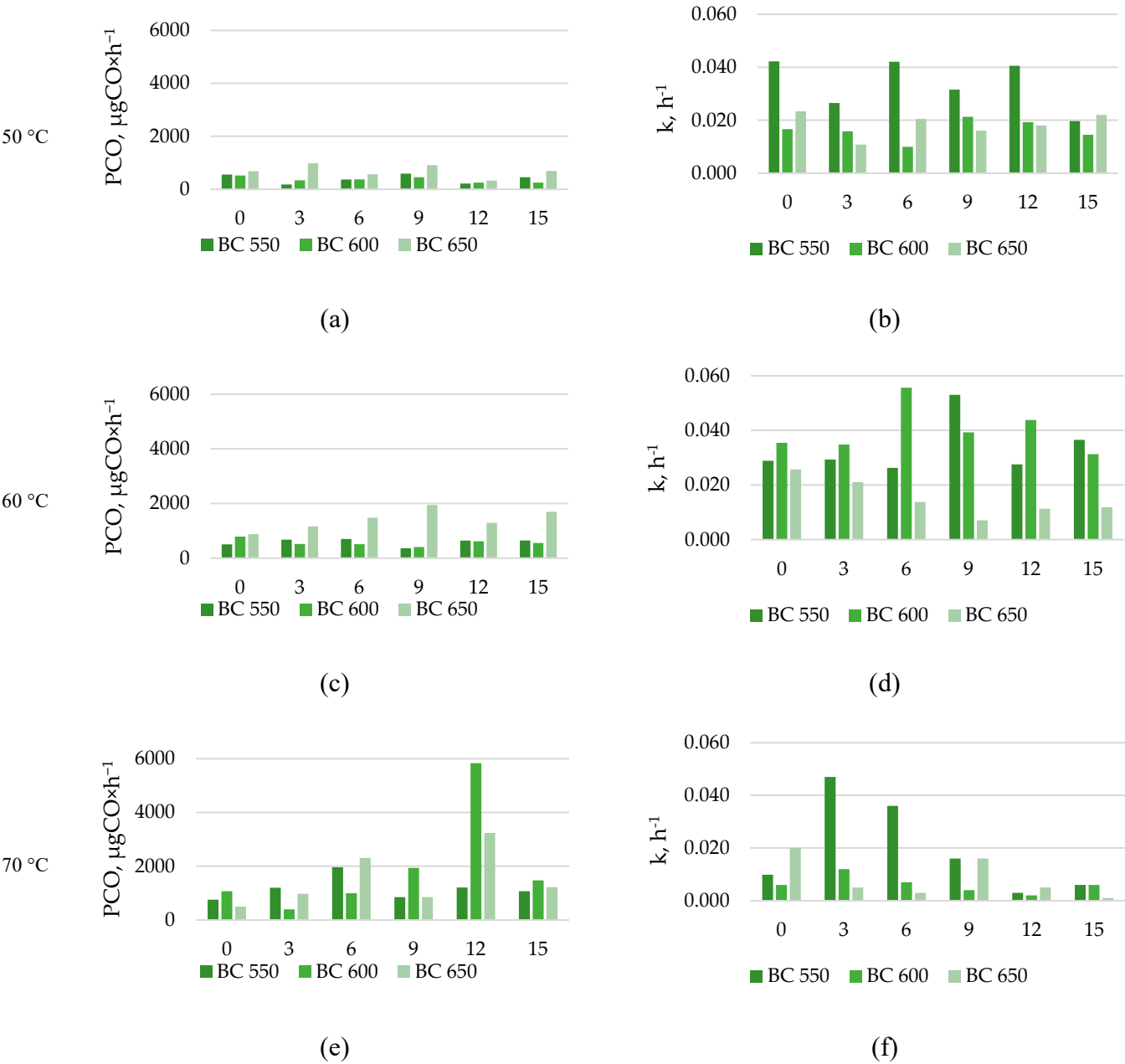


Figure 3. Estimated maximum CO production ($\mu\text{g}\cdot\text{g}^{-1}\text{ d.m.}$), and production constant rate (h^{-1}), during the different temperature incubations, biochar type, and biochar dose, a) maximum CO production at 50 °C, b) CO production constant rate at 50 °C, c) maximum CO production at 60 °C, d) CO production constant rate at 60 °C, e) maximum CO production at 70 °C, f) CO production constant rate at 70 °C.

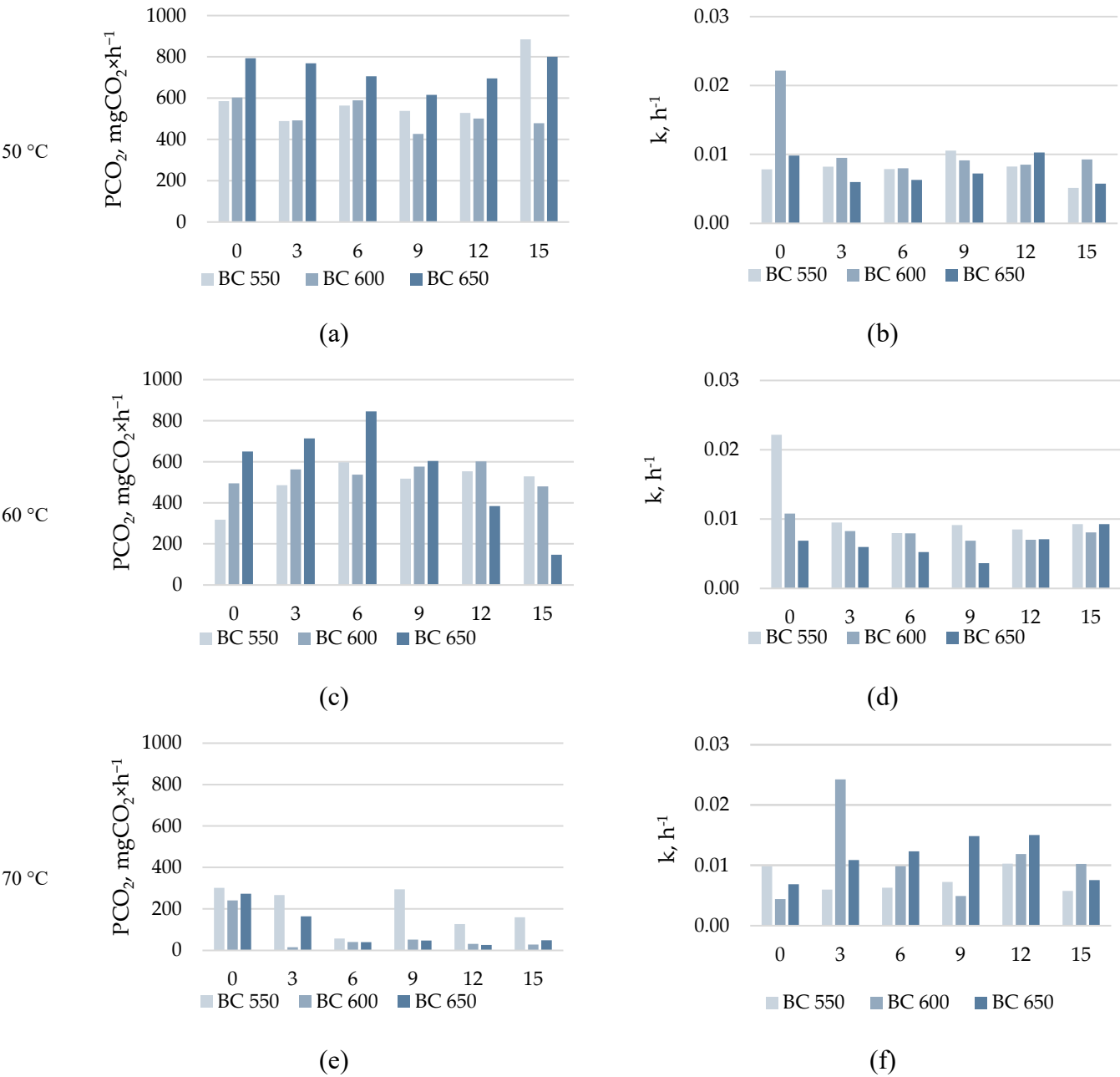


Figure 4. Estimated maximum CO₂ production (mg·g⁻¹ d.m.), and production constant rate (h⁻¹), during the different temperature incubations, biochar type, and biochar dose a) maximum CO₂ production at 50 °C, b) CO₂ production constant rate at 50 °C, c) maximum CO₂ production at 60 °C, d) CO₂ production constant rate at 60 °C, e) maximum CO₂ production at 70 °C, f) CO₂ production constant rate at 70 °C.

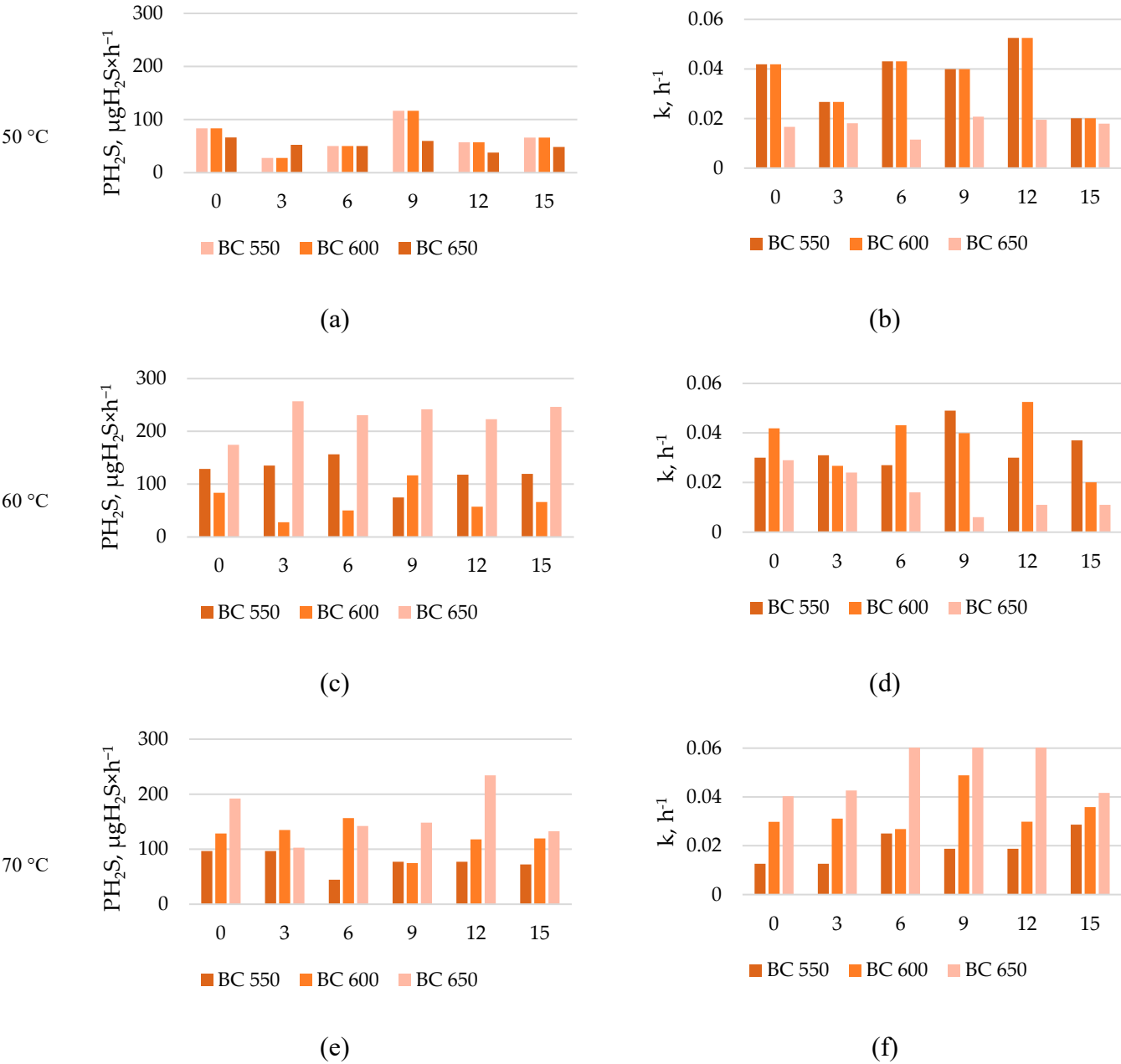
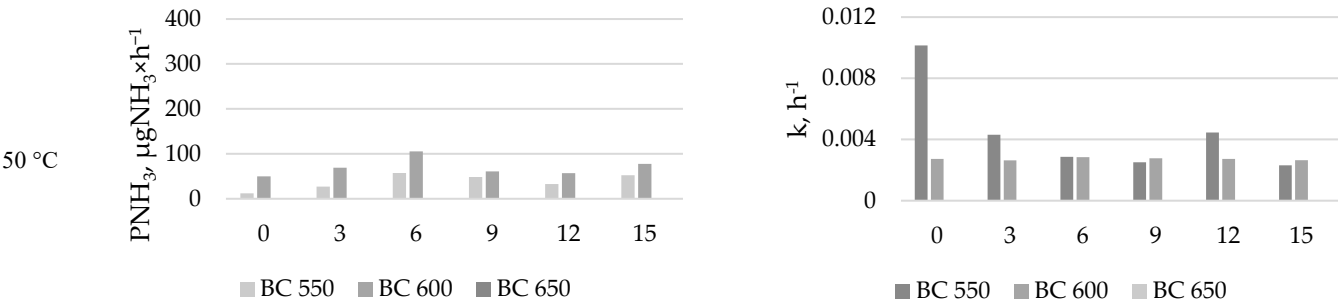


Figure 5. Estimated maximum H₂S production (µg·g⁻¹ d.m.), and production constant rate (h⁻¹), during the different temperature incubations, biochar type, and biochar dose, a) maximum H₂S production at 50 °C, b) H₂S production constant rate at 50 °C, c) maximum H₂S production at 60 °C, d) H₂S production constant rate at 60 °C, e) maximum H₂S production at 70 °C, f) H₂S production constant rate at 70 °C.



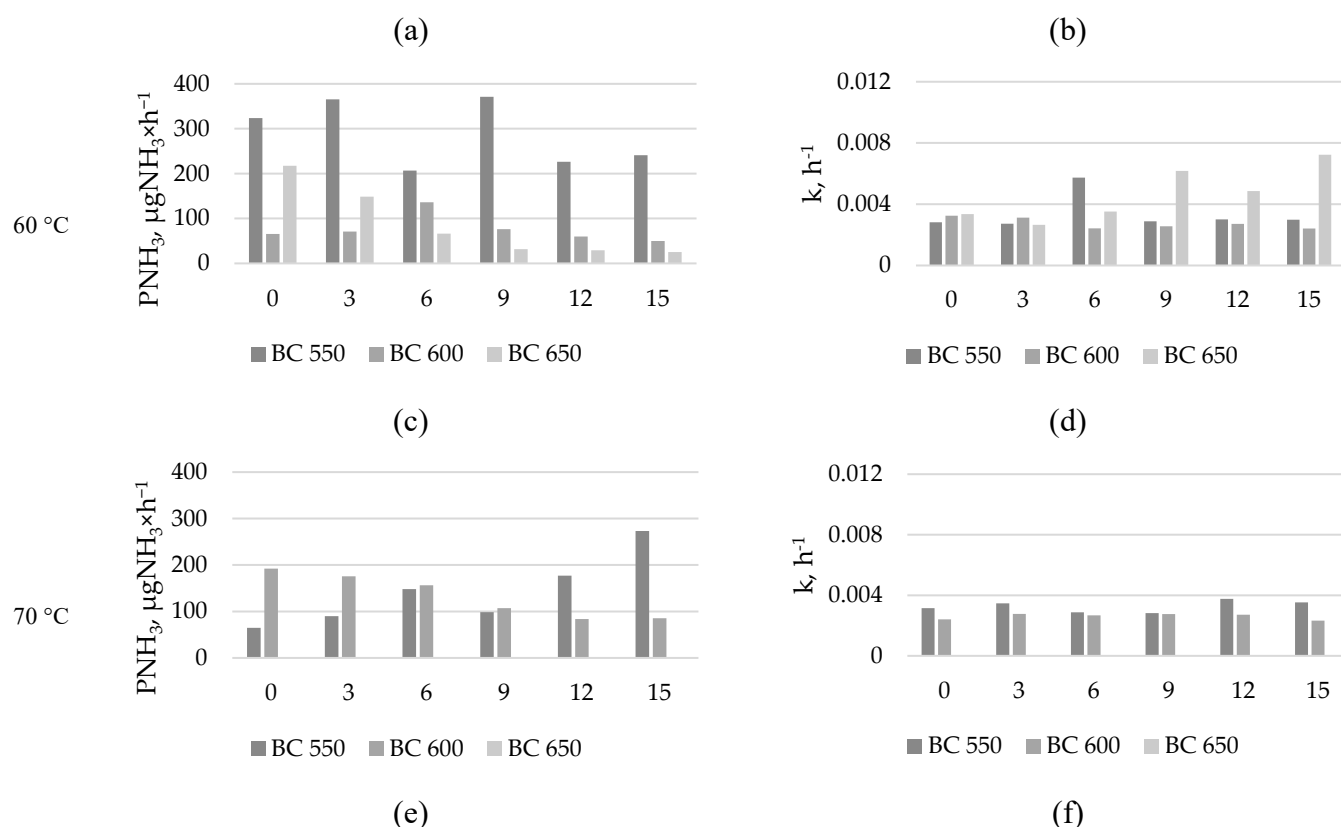


Figure 6. Estimated maximum NH_3 production ($\mu\text{g}\cdot\text{g}^{-1}\cdot\text{d.m.}$), and production constant rate (h^{-1}), during the different temperature incubations, biochar type, and biochar dose, a) maximum NH_3 production at 50 °C, b) NH_3 production constant rate at 50 °C, c) maximum NH_3 production at 60 °C, d) NH_3 production constant rate at 60 °C, e) maximum NH_3 production at 70 °C, f) NH_3 production constant rate at 70 °C.

3.2. Prediction of the Gaseous Emissions during Composting with Composts' Biochar (Machine Learning)

Ten kinds of classifiers, Linear Regression, Generalized Linear model, Random Forest, SVM with Linear Kernel, SVM with Radial Basis Function Kernel, k-Nearest Neighbours, Bayesian Regularized Neural Network, RPART, Generalized Boosted Regression Models and Extreme Gradient Boosting Tree were trained using collected data to evaluate the practicality of the classification model in predicting gaseous emissions output. Determination coefficients R^2 and RMSE were used to determine the effectiveness of the model, the results are shown in Table 1. For each emission, the best results ($R^2 \geq 0.6$) were observed for the Bayesian Regularized Neural Network, comparably good performance was also characteristic of RPART. These models were also characterized by a low RMSE ($\text{CO} < 380$; $\text{CO}_2 < 120$; $\text{H}_2\text{S} < 40$ $\text{NH}_3 < 80$), while its values are dependent on the measured emission value of the gas, hence the large discrepancies between the observed results. In addition, the best accuracy was observed NH_3 emission, where $R^2 > 0.9$. This demonstrates not only the good fit of the model to the results obtained during the tests but also the high potential for predicting emissions of this gas from the remaining input data. A high potential for predicting NH_3 has also been observed in the literature. Xie et al. used models based on artificial neural networks, the Adaptive Neuro Fuzzy Inference System (ANFIS), to predict ammonia emissions from pig-fattening houses using various inputs. He contrasted the results with other models such as the Multiple Linear Regression Model and Backpropagation. With ANFIS, it was possible to obtain high R^2 values (above 0.6) during both summertime and wintertime [36]. In turn, Küçüktopcu et al. used ANFIS and Multilayer Perception (MLP) models to model NH_3 emissions located on poultry farms. Modeling was performed using input data such as indoor air temperature, air humidity, air flow, NH_3 emission concentrations, litter moisture, litter pH and litter surface temperatures. Input data were used for modeling in different configurations, while the best results were obtained for the ANFIS model with subtractive clustering

(R²=0.910; RMSE=0.919) in the input data configuration using litter moisture, air temperature and airflow [37]. High potential is also shown by models, for CO₂ prediction. Li et al. used the AdaBoost, Bagging, Gradient Boost, Random Forest, k-Nearest Neighbors and Decision Tree models. The k-Nearest Neighbors model achieved the highest prediction accuracy, with an RMSE of 54.9. However, the authors have pointed out that the regression model's prediction granularity is too sensitive to changes in data distribution, resulting in less-than-ideal prediction performance [38]. It needs to be underlined that use of ML models to predict CO and H₂S during composting was demonstrated for the first time with sufficient accuracy with use Bayesian Regularized Neural Network (CO R²:0.71, RMSE: 243.3; H₂S R²:0.75, RMSE: 48.1)

Table 1. Comparisons between particular models by values of R squared and RMSE.

Model	CO		CO ₂		NH ₃		H ₂ S	
	R ²	RMSE	R ²	RMSE	R ²	RMSE	R ²	RMSE
Linear Regression	0.304	376.870	0.538	120.130	0.350	36.010	0.141	83.533
Random Forest	0.463	331.256	0.741	89.841	0.918	12.791	0.567	59.277
SVM with Linear Kernel	0.255	389.928	0.503	124.443	0.212	39.644	0.072	86.811
SVM with RBF Kernel	0.636	272.579	0.776	83.699	0.900	14.125	0.602	56.888
k-Nearest Neighbors	0.466	330.187	0.730	91.852	0.895	14.453	0.261	77.461
Bayesian Regularized Neural Network	0.710	243.318	0.808	77.465	0.948	10.159	0.715	48.111
RPART	0.693	250.324	0.802	78.562	0.930	11.796	0.648	53.459
Generalized Boosted Regression Models	0.595	287.527	0.764	79.493	0.899	14.163	0.584	58.104
Extreme Gradient Boosting Tree	0.309	375.754	0.798	85.764	0.793	20.326	0.486	64.608
Partial Least Squares Regression	-	-	0.544	119.348	0.360	35.737	0.149	83.131

3.2.1. Prediction of CO Emission

Figure 7 presents the simulation performed with the chosen models, which were Generalized Boosted Regression Models, SVM with RBF Kernel, Recursive Partitioning and RPART and Bayesian Regularized Neural Network. These models were compared to empirical data, in that case, it was possible to specify individual models. The characteristic of each model was an increase in CO emissions relative to empirical data. For the empirical data (Figure 7e), CO emissions were observed to be in the range of 0 to 2126.51 $\mu\text{g}\times\text{g}^{-1}$ d.m. (Supplementary Materials Table S2). The lowest gas emission values of less than 1000 $\mu\text{g}\times\text{g}^{-1}$ d.m. were for materials incubated at 50 °C, in which case the type of biochar did not significantly affect the increase in emissions. Equally low values were seen for material enriched with BC550, while incubated at 60 °C. The highest values, exceeding 2000 $\mu\text{g}\times\text{g}^{-1}$ d.m., were recorded for material with 6% BC550, incubated at 70 °C. High values also characterized the materials with 15 and 9% BC650, incubated at 60 °C and 70 °C, respectively. A characteristic of the models obtained was an overestimation of emissions in areas of missing data present in the empirical data, caused by device failure. The highest emission values were observed for the Bayesian Regularized Neural Network (Figure 7d) for the material with 15% BC650 incubated at 70 °C (3104.68 $\mu\text{g}\times\text{g}^{-1}$ d.m.) (Supplementary Materials Table S2); additionally, this was the model that predicted emission with the highest accuracy. For this model, the addition of 3 and 6% biochar incubated at 50 °C was the most effective for reducing emissions, irrespective of the pyrolysis process temperature. The least accurate tool for predicting CO emissions was found to be the Generalized Boosted Regression Model (Figure 7a). With this model, particularly for BC650 incubated at 60 °C and 70 °C and for BC600 incubated at 60 °C, a significant overestimation of emissions was observed that was not present in the empirical data. Furthermore, it has been observed that certain models exhibited varying degrees of accuracy in predicting emissions. Additionally, it is important to note that the

selected models may not be suitable for extrapolating data beyond the time range during which measurements were taken.

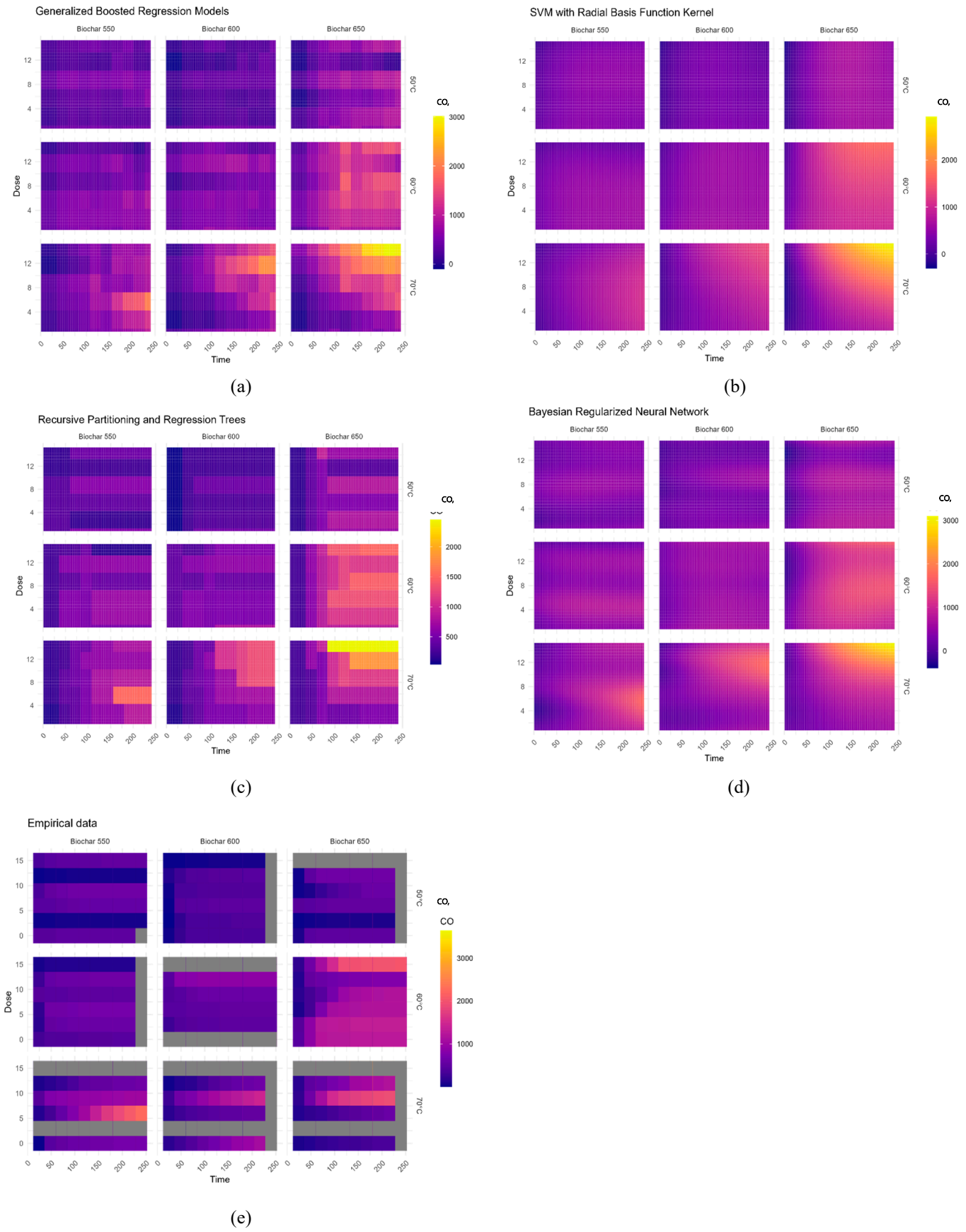
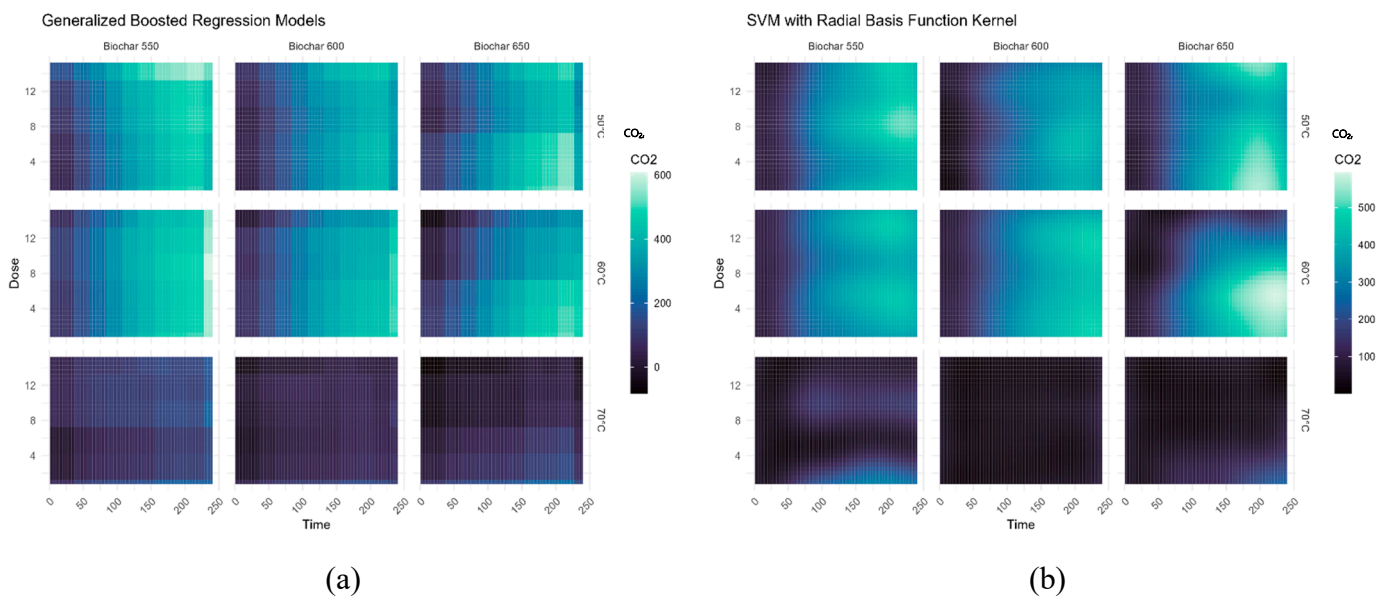


Figure 7. Predicted CO production ($\mu\text{g}\cdot\text{g}^{-1}\cdot\text{d.m.}$) based on biochar temperature production, incubation temperature and dose of biochar, using a) Generalized Boosted Regression Models °C, b) SVM with Radial Basis Function Kernel, c) Recursive Partitioning and Regression Trees, d) Bayesian Regularized Neural Network, e) Empirical data.

3.2.2. Prediction of CO₂ Emission

Figure 8 shows approximated CO₂ emissions (Figure 8a–d) and empirical data (Figure 8e) collected during the laboratory research part. In the case of the empirical data, especially for materials with BC600 and BC650 and stored at 70 °C, there was a significant reduction in emissions relative to temperatures of 50 °C and 60 °C; in these cases, emissions reached values close to zero. This indicates that the higher temperatures of the pyrolysis process and the higher storage temperature of the material have a positive effect on the adsorption of CO₂ emissions. BC550 also showed a significant reduction in gas emissions, but only at doses of 6% and 12%. A dose of 15% BC650 incubated at 60 °C also effectively reduced CO₂ emissions. The results of the empirical data were in the range 4.58–888.84 $\text{mg}\times\text{g}^{-1}\cdot\text{d.m.}$ (Supplementary Materials Table S3), while the highest modeled emission fell for the Bayesian Regularized Neural Network (Figure 8d) and was embedded in the range 0–620.61 $\text{mg}\times\text{g}^{-1}\cdot\text{d.m.}$ (Supplementary Materials Table S3), at the same time it was the model that performed best in approximating the results from the input data. Equally accurate results were obtained from the RPART model, while in this case there was also a significant underestimation of the predicted final values, as the maximum CO₂ emission was 592.54 $\text{mg}\times\text{g}^{-1}\cdot\text{d.m.}$ (Supplementary Materials Table S3) and was observed for material with a 6% dose of BC650 incubated at 60 °C. Despite a relatively high R^2 (>0.7), the tool with the lowest modeling efficiency for gas emissions was the Generalized Boosted Regression Models model, which underestimated the actual CO₂ production the most of all the models presented graphically.



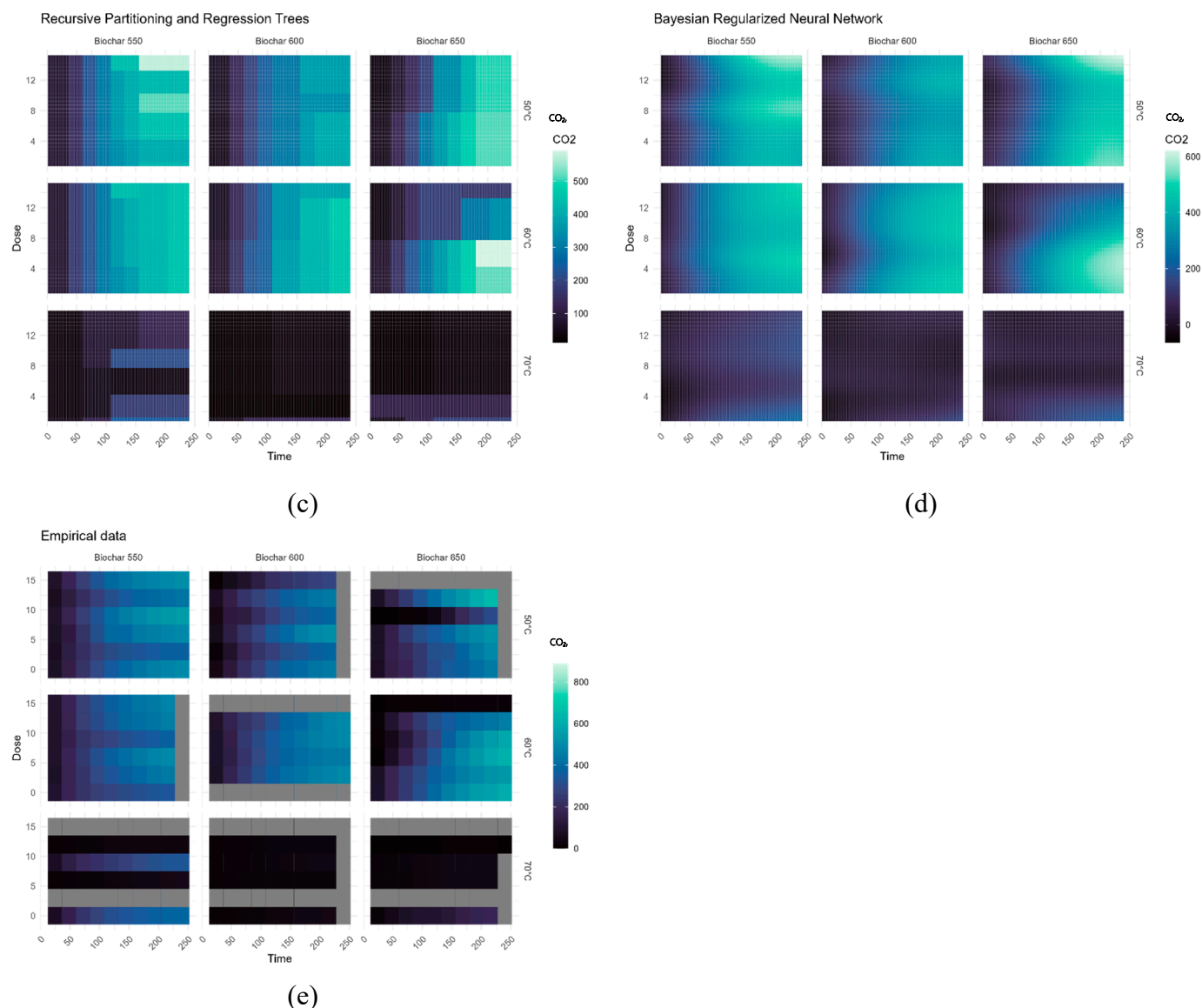
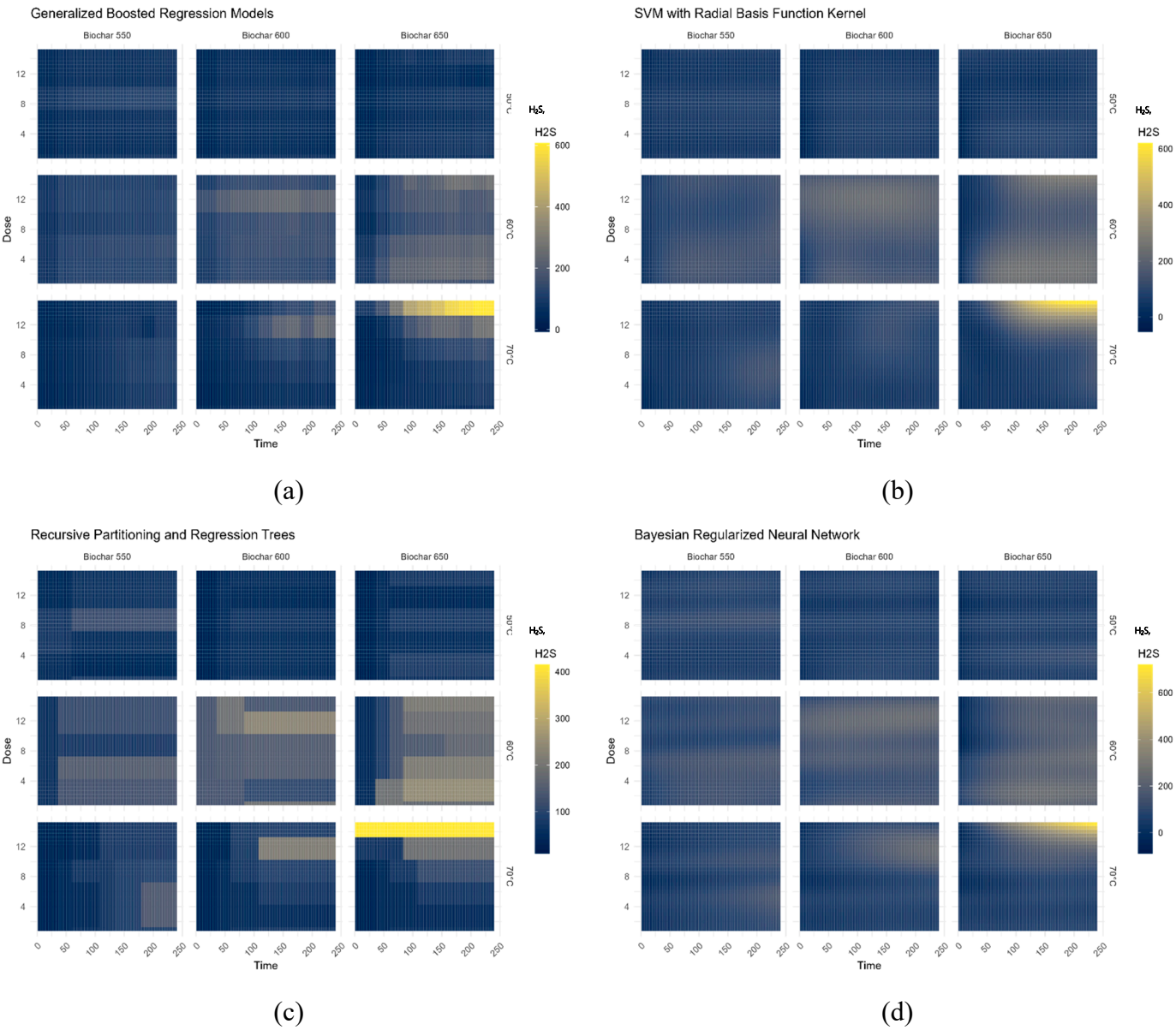


Figure 8. Predicted CO₂ production (mg·g⁻¹ d.m.) based on biochar temperature production, incubation temperature and dose of biochar, using a) Generalized Boosted Regression Models °C, b) SVM with Radial Basis Function Kernel, c) Recursive Partitioning and Regression Trees, d) Bayesian Regularized Neural Network, e) Empirical data.

3.2.3. Prediction of H₂S Emission

Figure 9a-d shows a graphical representation of the models predicting the average concentration of H₂S emissions in the test material and contrasts them with the empirical data shown in Figure 9e. The range of results within which the empirical data fell was from 0.03 µg·g⁻¹ d.m. to 659.44 µg·g⁻¹ d.m. (Supplementary Materials Table S4). The lowest emissions were observed for material incubated at 50 °C; in addition, the type of biochar (depending on the temperature of the pyrolysis process) did not have a particularly significant effect on H₂S production. This suggests that H₂S emissions reduction is influenced only by storage conditions, such as lower temperatures, and not by the dose or type of biochar used. The area on the heatmap with the highest gas emissions fell for the 15% BC650 additive stored at 70 °C. Again, the model with the highest performance was the Bayesian Regularized Neural Network, the nature of the prediction in this case was very close to the empirical data, as the range of results obtained was from 0.65 µg·g⁻¹ d.m. to 719.20 µg·g⁻¹ d.m. (Supplementary Materials Table S4), In addition, it was the only model for which R²>0.7 was observed. A model with a similar level of fit was RPART, while for the final approximation results a significant under-

estimation of emissions was observed relative to the control sample and settled in the range from 11.14 $\mu\text{g}\times\text{g}^{-1}$ d.m. to 414.96 $\mu\text{g}\times\text{g}^{-1}$ d.m. (Supplementary Materials Table S4). The lowest level of fit of the data to the model was observed for Generalized Boosted Regression Models. A possible reason for the low levels of model fit was the failure to include factors in the input data that directly affect H_2S emissions.



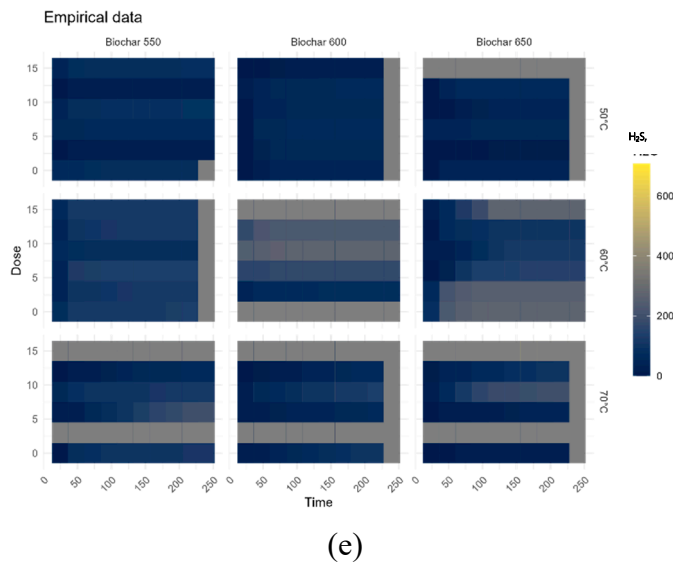


Figure 9. Predicted H_2S production ($\mu\text{g}\cdot\text{g}^{-1}$ d.m.) based on biochar temperature production, incubation temperature and dose of biochar, using a) Generalized Boosted Regression Models $^\circ\text{C}$, b) SVM with Radial Basis Function Kernel, c) Recursive Partitioning and Regression Trees, d) Bayesian Regularized Neural Network, e) Empirical data.

3.2.4. Prediction of NH_3 Emission

The experimental data shown in Figure 10e had a range from $0.04 \mu\text{g}\cdot\text{g}^{-1}$ d.m. to $215.58 \mu\text{g}\cdot\text{g}^{-1}$ d.m. (Supplementary Materials Table S5). The highest emissions were observed for the control sample and the material with 9% BC550 incubated at 60°C . High emission values were also achieved by material with 15% BC550 addition stored at 50°C . The lowest NH_3 emission values were recorded for BC650, as this type of biochar reduced the measured emissions in the material regardless of dose and storage temperature. Similarly to the other gas emissions, the Bayesian Regularized Neural Network was the most successful model, generating results ranging from $0.04 \mu\text{g}\cdot\text{g}^{-1}$ d.m. to $252.27 \mu\text{g}\cdot\text{g}^{-1}$ d.m. (Supplementary Materials Table S5). Despite a slight over-prediction, the model had the highest fit, as evidenced by high R^2 values and low RMSE. RPART (Figure 10c) was a model with a similar degree of fit; additionally, the approximated values were not as over-predicted as those of the Bayesian Regularized Neural Network (Figure 10d). The highest emissions predicted by this model were observed for material enriched with doses of biochar 3, 6 and 9% BC550, incubated at 60°C . A significant reduction in predicted NH_3 emission was present for the SVM with RBF Kernel model (Figure 10e). For that model the maximum approximated emission was less than $200 \mu\text{g}\cdot\text{g}^{-1}$ d.m. (Supplementary Materials Table S5). In addition, all models failed to cope with data extrapolation beyond the designated time interval. The high R^2 and RMSE values for each of the examined models presented graphically demonstrate the high applicability potential of using artificial intelligence to predict NH_3 emissions during the composting process.

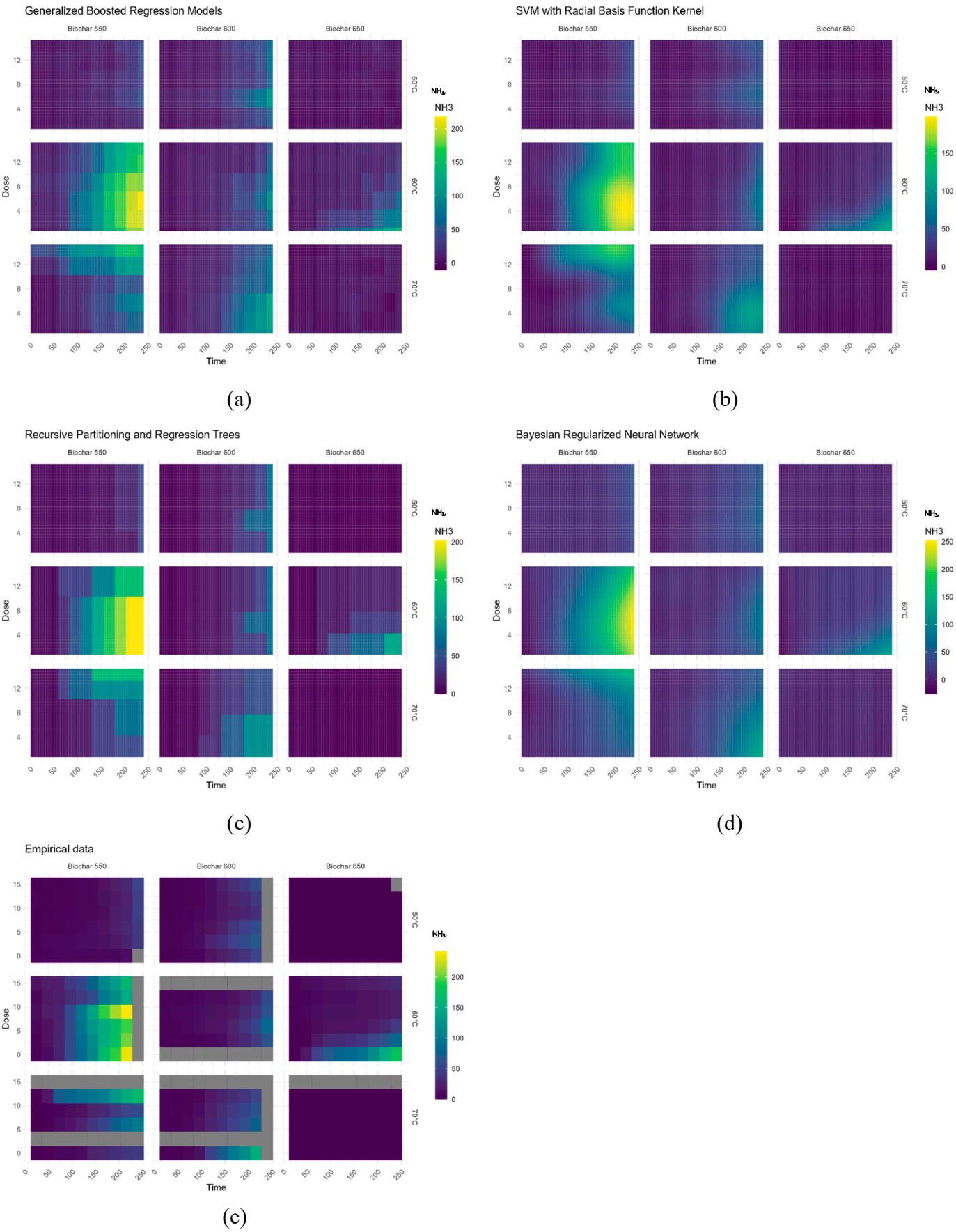


Figure 10. Predicted NH_3 production ($\mu\text{g}\cdot\text{g}^{-1}\cdot\text{d.m.}$) based on biochar temperature production, incubation temperature and dose of biochar, using a) Generalized Boosted Regression Models, b) SVM with Radial Basis Function Kernel, c) Recursive Partitioning and Regression Trees, d) Bayesian Regularized Neural Network, e) Empirical data.

4. Discussion

Improving the efficiency and quality of composting is the primary issue for sustainable composting. Although composting has many advantages in the treatment of organic waste, there are still many problems and challenges associated with emissions. Various emissions like NH_3 , VOCs, and H_2S , as well as greenhouse gases such as CO_2 , CH_4 , and N_2O are generated during the process of decomposition of organic compounds [39]. It is understood that emissions released during the composting process are influenced by both the characteristics of the feedstock and the conditions of the process itself. Effective management emissions techniques such as adsorption/optimizing C/N ratios [40] (for CO_2 reduction), minimizing N losses (for NH_3 reduction) [41], and improving pile oxygenation [42] (for H_2S and CO reduction) can help to control these emissions. One promising approach to enhancing composting conditions to reduce listed above emissions involves the use of compost's biochar in small quantities [20]. These observations may explain can observed correlations between the emissions (Supplementary Material Table S1) and support the accuracy of emissions modeling based on other emissions used in this study.

This constatation supports the network analysis (Figure 11) the use of ML in composting is mainly connected with ANN and the most analyzed parameters are temperature, nitrogen and heavy metals. The compost and biochar have a large number of connections, with mainly emissions like NH_3 , N_2O , CH_4 and CO . There is a low connection between biochar compost and machine learning, what's proof the novelty of this study.

The novel analytical method based on a mathematical model (MM) and, machine learning (ML) model can explore the relationship between different parameters and draw universal conclusions, which was used to predict emissions during green waste composting. Using modeling techniques can significantly decrease costs and expedite the process of implementing new composting practices, especially when compared to laboratory and pilot-scale investigations. This makes it an attractive option for exploring innovative composting methods [43].

Currently, the research of MM and ML on aerobic composting is still in the early stages. Mathematical models could enhance the initial mixture of biowaste streams and optimal amounts for composting and thereby help to accelerate the process [44]. In this study, the first-order kinetics equations were used to estimate the emissions potential during the first 10 days of composting (Figures 3-6, Supplementary Materials Table S1). The first-order kinetics believed the degradation of organic matter during composting is thought to be enzyme-mediated. The rate of the reaction is determined by the substrate concentration. Mathematical models can be a valuable tool for optimizing process performance in terms of costs, efficiency, and environmental impact by simulating and predicting the process outcome [45]. Interpretive and optimization methods of MM and ML can be employed to analyze conversion patterns in composting. Previous studies have demonstrated that MM can be utilized to describe the intermediate conversion patterns of biomass, primarily employing empirical equations, Monod-type equations, and first-order kinetics. The first-order kinetic equation is commonly used for composting simulation, but it is less suitable for modeling organics conversion at constant temperature parameters. On the other hand, the first-order has been effectively utilized in estimating the emissions potential from composting. In this study, first-order equations were employed to compare their usefulness with matching learning in the estimation of emissions in composting. In previous research, R^2 was primarily 0.8-0.9 for CO_2 and CO [22]. However, in this study, it was much lower at 0.5-0.9 (Supplementary Materials Table S1). For other emissions (H_2S and NH_3), the first kinetic equations effect in wide range fit value – R^2 0.1-0.9 (Supplementary Materials Table S1). As mentioned above the addition of biochar to composting effecting in change in its properties. This implies that the addition of biochar to compost, alters the

emissions production patterns, and mechanism-derived mathematical models may no longer be sufficient.

These observations made the authors focus on predicting the composting process using ML. As shown so far, an ML in composting focuses mostly on predicting the compost maturity and compost properties i.e., pH, EC, GI, TN, TOC, etc, with only a few papers concerned with emissions [46]. The accuracy of ML models used in composting process prediction changed in the range of 0.56-0.99 for R^2 , but in most cases showed good fit >0.7 . Common ML models used in composting are as follows: Random Forest (RF), Artificial Neural Network (ANN), Support Vector Regression (SVR), Decision Tree, and Decision Support (DS). RF and ANN are observed to have the best prediction performance, and the accuracy of R^2 was usually >0.9 . In comparison to this study, the best ML models were also ANN (Bayesian Regularized Neural Network), and DT (RPART).

There is a limited number of authors who concentrate on precise forecasts of CO_2 or NH_3 emissions from feedstock composting. Furthermore, no research centers on the anticipation of CO or H_2S during composting using machine learning techniques. Li. et. al. used various ML models to predict CO_2 emissions based on input variables such as TOC, TN, C/N ratio, cellulose, hemicellulose, and lignin. The different models had varying levels of RMSE, with AdaBoost at 49.8, Bagging at 80.6, Gradient Boost at 99.9, Random Forest at 83.0, KNN at 55.0, and Decision Tree at 101.8. These results are similar to ours, as shown in Table 1. Li et al. found the highest R^2 score of 0.88 accuracy for Random Forest. Bayesian Regularized Neural Network had the best accuracy of 0.81 in the study, while RF achieved an R^2 score of 0.74 for CO_2 emissions production. This indicates that further research should explore the potential of this type of ML model. In other study for predicting NH_3 emissions during composting sewage sludge with straw, Artificial Neural Network (ANN) was utilized. The ANN achieved an R^2 score of over 0.97 by using temperature, pH, EC, C/N, and N-NH_4 as input parameters [47].

The findings of this study suggest that controlling gaseous emissions from green waste composting with compost's biochar can be achieved by monitoring the emissions of other gases e.g., CO_2 output from composting is controllable by CO, H_2S , and NH_3 emissions. It is important to note that the experimental data used in this study are based on the observations from previous publications and may not fully reflect the control of CO, CO_2 , H_2S and NH_3 emissions from composting. Nevertheless, this solution can provide valuable insights for future studies and practices with a larger dataset (especially collected in field study) and more sophisticated ML techniques.

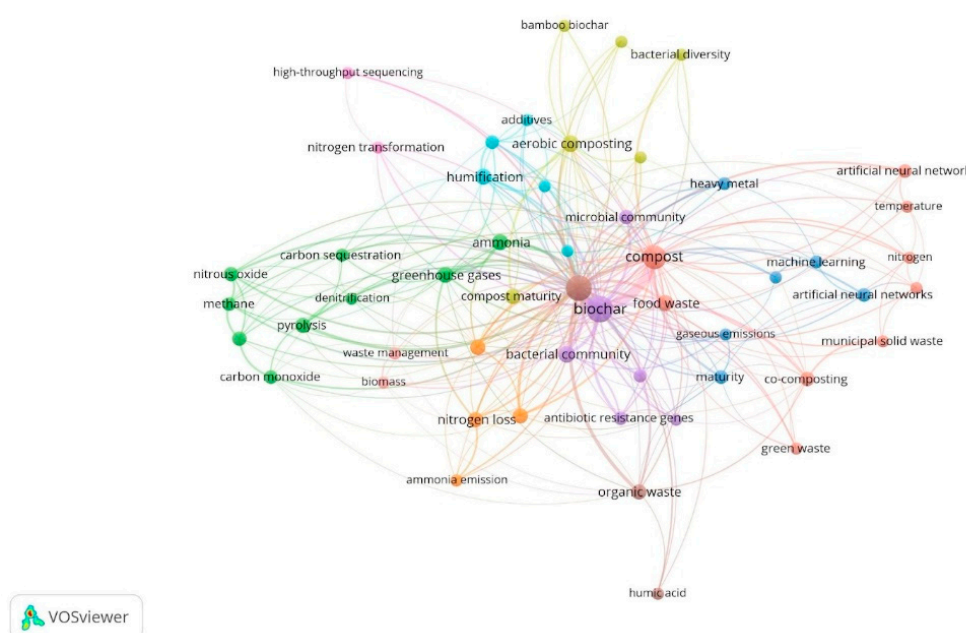


Figure 11. Network analysis of co-occurrence of important keywords.

4. Conclusions

This study utilized mathematical models (MM) and machine learning (ML) models to predict the emissions (CO, CO₂, H₂S, NH₃) during first 10 days of composting with compost's biochar addition. **For the first time the ML models to predict CO and H₂S during composting were demonstrated.** MM has not been very effective in predicting emissions, (R² 0.1 - 0.9), while ML models such as acritical neural network (ANN) and decision tree (DT) have demonstrated satisfactory results. **A quality assessment of the developed ML models has shown that the best predictive capacity was reached for ANN** (Bayesian Regularized Neural Network; R² accuracy CO:0,71, CO₂:0,81, NH₃:0,95, H₂S:0,72) **and DT** (RPART; R² accuracy CO:0,693, CO₂:0,80, NH₃:0,93, H₂S:0,65). Further research in a semi-scale and field study composting with biochar is also needed to improve the accuracy of development models. In conclusion, this study provided new insights into the enhancement of the composting emissions process.

Supplementary Materials: The following supporting information can be downloaded at the website of this paper posted on Preprints.org, Figure S1. Principal component analysis of collected data; Table S1. Estimated maximum gas production (PCO₂, mg·g⁻¹·d.m; PCO, PH₂S or PNH₃ µg·g⁻¹·d.m), production constant rate (k, h⁻¹), and average production rate (r, CO₂ mg·g⁻¹·d.m; CO, H₂S or NH₃ µg·g⁻¹·d.m), during the composting in different temperature incubations, biochar type and biochar dose; Table S2. Experimental and predicted CO emission, µg CO·g⁻¹·d.m, in different machine learning models; Table S3. Experimental and predicted CO₂ emission, mg CO₂·g⁻¹·d.m, in different machine learning models; Table S4. Experimental and predicted H₂S emission, µg H₂S·g⁻¹·d.m, in different machine learning models; Table S5. Experimental and predicted NH₃ emission, µg NH₃·g⁻¹·d.m in different machine learning models.

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