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Sasi Vimon , [Chonticha Romyasamit](#) , Rawiwan Chanpakdee , Sumaree Boonplu , [Chackrit Nuengjamnong](#) , Fonthip Makkliang , [Suthinee Sangkanu](#) , [Veeranoot Nissapatorn](#) , Phirabhat Saengsawang , Tina S. Dalgaard , [Watcharapong Mitsuwana](#) *

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Article

Eco-Friendly Microencapsulation of *Lacticaseibacillus paracasei* Using *Cissampelos pareira* Leaf Extract as Natural Encapsulating Materials

Sasi Vimom^{1,2}, Chonticha Romyasamit³, Rawiwan Chanpakdee¹, Sumaree Boonplu¹, Chackrit Nuengjamnong⁴, Fonthip Makkliang⁵, Suthinee Sangkanu⁶, Veeranoot Nissapatorn^{3,7}, Phirabhat Saengsawang^{1,2}, Tina S. Dalgaard⁸ and Watcharapong Mitsuwan^{1,2,9,*}

¹ Akkharatchakumari Veterinary College, Walailak University, Nakhon Si Thammarat, 80160, Thailand; Sasi.vi@wu.ac.th (S.V.); rawiwanchan57@gmail.com (R.C.); boonplu.su@gmail.com (S.B.); phirabhat.sa@wu.ac.th (P.S.); watcharapong.mi@wu.ac.th (W.M.)

² One Health Research Center, Walailak University, Nakhon Si Thammarat, 80160, Thailand

³ School of Allied Health Sciences, Walailak University, Nakhon Si Thammarat, Thailand; chonticha.ro@wu.ac.th (C.R.); nissapat@gmail.com (V.N.)

⁴ Department of Animal Husbandry, Faculty of Veterinary Science, Chulalongkorn University, Bangkok 10330 Thailand; Chackrit.n@chula.ac.th (C.N.)

⁵ School of Languages and General Education, Walailak University, Thasala, Nakhon Si Thammarat, 80160, Thailand; fonthip.ma@wu.ac.th (F.M.)

⁶ Department of Pharmacognosy and Pharmaceutical Botany, Faculty of Pharmaceutical Sciences, Prince of Songkla University, Hat Yai, Songkhla 90112, Thailand; suthinee.9938@gmail.com (S.S.)

⁷ World Union for Herbal Drug Discovery (WUHeDD), and Research Excellence Center for Innovation and Health Products (RECIHP), Walailak University, Nakhon Si Thammarat, Thailand

⁸ Department of Animal and Veterinary Sciences, Aarhus University, Tjele, Denmark; tina.dalgaard@anivet.au.dk (T.S.D.)

⁹ Center of Excellence in Innovation of Essential Oil and Bio-active compound, Walailak University, Nakhon Si Thammarat, 80160, Thailand

* Correspondence: watcharapong.mi@wu.ac.th and/or 1234_k@hotmail.co.th

Abstract: Microencapsulation using polymer materials is a potent process to protect and prolong the survival of probiotics. *Cissampelos pareira* leaf contains natural gelling agents that possess solidifying properties. This study aimed to investigate the development of microencapsulation containing *Lacticaseibacillus paracasei* using *C. pareira* extract as a natural encapsulating material. The absorption bands near 1603 cm⁻¹ and 1725 cm⁻¹ detected by Infrared spectroscopy (FTIR) were identified as pectin in *C. pareira* structure. The *L. paracasei*-*C. pareira* microcapsules (LP-CP) showed high encapsulation efficiency by 90.5% which was confirmed by the evaluation of their survival rate. Under thermal conditions (85°C), bacterial viability detected in the microcapsules was 69% as opposed to non-encapsulated bacteria where viability was as low as 5%. Furthermore, the microcapsule presented 75% bacterial viability whereas the free cells showed 30% under acidic conditions (pH 2). During storage conditions, LP-CP viability remained at 50% when the storage time was extended to 90 days whereas, the survival rate of free cells significantly decreased by 100% after 90 days. This information suggests that *C. pareira* is a potent polymer which can be used as an eco-friendly material for the microencapsulation of probiotics relevant as dietary supplementation for humans or animals.

Keywords: *Cissampelos pareira*; *Lacticaseibacillus paracasei*; microencapsulation; natural encapsulating material; probiotic supplementation

1. Introduction

Probiotics including lactic acid bacteria are beneficial microorganisms that are present in the digestive tract, oral cavity, and reproductive systems of both humans and animals [1]. The bacteria in the genus *Lactobacillus* such as *Lactobacillus paracasei* that belong to the normal mucosal microbiota of humans and animals is a well-documented probiotic and e.g. often used in dairy product fermentation [2]. Recently, the name *L. paracasei* was changed to *Lacticaseibacillus paracasei* because the genome diverged from *Lactobacillus* [2]. It has been demonstrated that bacteria in the genus *Lacticaseibacillus* are capable of creating bacteriocins, antimicrobial peptides utilized in the food and healthcare industries [3,4]. Unfortunately, factors in the gastrointestinal tract such as low pH, low enzyme levels, and low bile salts make it difficult for free probiotic cells to survive in the body after oral administration. Hence, the food and feed industry need to implement a strategy to increase the probiotics' survival rate.

Microencapsulation entails the process of encasing probiotics in microcapsules made of polymers and other organic and inorganic materials [5]. Microcapsules may be beneficial as delivery system supporting minimal damage to the live probiotic bacteria by the various conditions in the gastrointestinal tract [5]. Additionally, microencapsulation might make it easier to regulate release and successfully deliver probiotics to the action site [6]. In addition, the encapsulation could maintain the viability of probiotics during the food manufacturing process and long-term storage [6]. Several strategies to prepare microcapsules exist and we have earlier reported encapsulation of probiotics using sodium alginate in combination with calcium chloride [6]. Sodium alginate is a natural polysaccharide which is biodegradable and commonly used in industry. Sodium alginate can be extracted from brown algae cell walls but the extraction process needs several chemical reagents such as alkaline medium, sodium carbonate, and sodium hydroxide [7]. Therefore, alternative natural materials which can be applied for encapsulation of probiotics are of interest.

This study focused on *C. pareira* leaf extract. The plant species belonging to the Menispermaceae family is common in many Southeast Asia countries including Thailand where it is locally known as Khruea-ma-noi. The plant has been extensively used in the traditional medicinal system for the treatment of numerous diseases such as ulcers and wounds [8,9]. In addition, the leaves of the plant have been used as raw materials for Thai food and importantly they contain hydrocolloids, which may act as gelling agents as they possess solidifying properties [10]. In addition, pectin, a high molecular weight polysaccharide, was detected in *C. pareira* leaves [9]. Recently, a preparation of biodegradable films containing pectin extracted from *C. pareira* leaves has been documented [11].

Therefore, this study aimed to investigate the development of microencapsulation of *L. paracasei* using *C. pareira* extract as a natural encapsulating material. Furthermore, the characterization, the properties, and the storage of the microcapsules under adverse conditions were determined. In addition, the phytochemical present in the plant extract was investigated.

2. Materials and Methods

2.1. Ethical Approval

The study did not involve any live animals or humans, so ethical approval was not necessary. All the experiments were performed under the regulation of biosafety for scientific experiments (Ref. No. WU-IBC-66-042) of Walailak University, Nakhon Si Thammarat, Thailand.

2.2. Plant Collection and Chemical Reagents

C. pareira leaves were harvested in Nakhon Phanom province, Northeast Thailand in December 2023. The plant materials were kept at 4 °C until used. All the reagents containing tween 80, oil, porcine bile extract, pancreatin, lipase from porcine pancreas, pepsin from porcine gastric mucosa, and trypsin from bovine pancreas were purchased from Sigma-Aldrich, Co. Ltd. (St. Louis, MO, USA). Calcium chloride (CaCl₂) and succinic acid (C₄H₆O₄) were purchased from Merck (Darmstadt, Germany).

2.3. Bacterial Strain and Bacterial Culture Conditions

Lactocaseibacillus paracasei WU2502, a probiotic candidate, was used in this study. The bacterium was isolated from Palmyra palm sugar as described earlier by our research team [2]. In brief, the bacterium was cultured in 10 mL of Mann Rogosa Sharpe (MRS) broth (HiMedia, India) and MRS agar (HiMedia, India), incubated at 37°C for 24 h [12]. The bacterial culture was kept in MRS broth containing 20% glycerol at -80 °C until used.

2.4. Chemical Composition and Characterization of *C. pareira*

The Official Methods of Analysis (AOAC) techniques were used to assess the chemical composition of *C. pareira* leaf extract [13]. Briefly, the leaves were dried at 60°C in order to measure the levels of dry matter (DM), crude protein (CP), ether extract (EE), crude fiber (CF), ash, and nitrogen free extract. The moisture content was evaluated by drying the sample in a hot air oven at 100 °C until it achieved a consistent weight. Ash was determined after burning in a muffle furnace at 550 °C. Fat was extracted using petroleum ether and determined using a Soxtec System 1043 (Tecator, Hoganas, Sweden) after the petroleum ether was removed. Protein was calculated using the Kjeldahl technique with a multiplication factor of 6.25, and carbohydrate was calculated by subtracting the quantities of the above components from 100. The enzymatic gravimetric technique was used to determine soluble and insoluble dietary fibers. The analysis of metabolized energy was conducted using a bomb calorimeter (Mode AC-500 model manufactured by Leco in St. Joseph, MI, USA). Benzoic acid was used as the reference standard.

The chemical structure of the extracted *C. pareira* was determined by attenuated total reflection infrared (ATR-FTIR) spectroscopy using a Bruker Alpha ATR technique. Spectra were detected in the range of 4000–400 cm⁻¹.

2.5. Preparation of *L. paracasei* Encapsulated *C. Pareira* Microcapsules (LP-CP)

Fifteen grams of fresh *C. pareira* leaves were mixed in deionized water (DI) (100 mL), followed by adding tween 80 (2 mL) and stirring for 20 min at room temperature. The solution was maintained under constant stirring for 20 min followed by adding the overnight culture of *L. paracasei* (1×10^{10} CFU/mL, 10 mL) and soybean oil (10 mL). The solution was added dropwise through a needle (diameter 1.8 mm) into CaCl₂ (0.1 mol/L) aqueous solution to obtain the crude *L. paracasei* encapsulated *C. pareira* microcapsules. The distance between the syringe and CaCl₂ solution was around 4-5 cm. The microcapsules were sieved by a mesh (diameter 0.053 mm.) and washed with DI water twice before drying in a hot air oven at 37°C overnight to obtain LP-CP microcapsules. A schematic diagram of LP-CP preparation is shown in Figure 1. Alginate microcapsules was used as the positive control, was prepared as previously described by our research team [6].

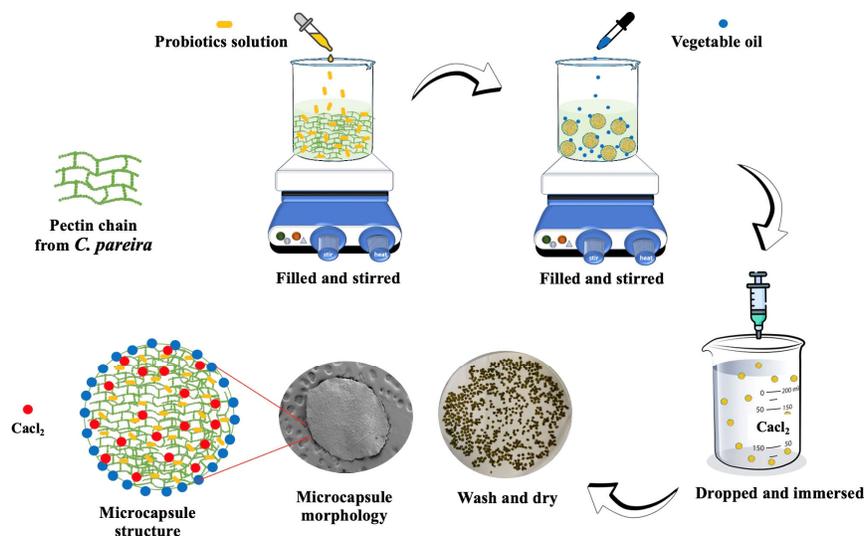


Figure 1. Preparation of *L. paracasei* encapsulated *C. pareira* microcapsules (LP-CP).

2.6. Determination of Physico-Chemical Characteristics of LC-LP Microcapsules

2.6.1. Structural Analysis

The morphologies of LC-LP microcapsules as well as sodium alginate microcapsules were identified by scanning electron microscopy (SEM-Zeiss, Munich, Germany). The microcapsules of sodium alginate and LP-CP were cut across by a razor blade to observe the cross-section. The chemical structures of LP-CP were evaluated by a Bruker Alpha Fourier transform infrared spectrometer using ATR technique in the range of 4000 cm^{-1} to 400 cm^{-1} [14]. *L. paracasei* viability was evaluated after heating LP-CP at $75\text{ }^{\circ}\text{C}$, $80\text{ }^{\circ}\text{C}$, $85\text{ }^{\circ}\text{C}$, $90\text{ }^{\circ}\text{C}$, $95\text{ }^{\circ}\text{C}$, and $100\text{ }^{\circ}\text{C}$ for 3 min.

2.6.2. Encapsulation Efficiency

The enumeration of the viable cells was determined in triplicate using a drop plate technique in an MRS agar as previously described [15]. Briefly, dilutions of the culture were prepared by adding the samples (1 mL) into a tube containing peptone water (9 mL, 0.1 g/100 mL). The mixture was vortexed for 10 min until homogenization. The sample (20 μL) was cultured on MRS agar and incubated at $37\text{ }^{\circ}\text{C}$ for 48 h. The bacterial viability was investigated and represented as CFU/mL. The viability of *L. paracasei* in LP-CP was carried out by homogenization LP-CP (1 g) in sodium citrate (10 mL) to dissolve the encapsulating material. Then, the samples were serial diluted and cultured on MRS agar as described above. The encapsulation efficiency (EY, %) was calculated as described by Vimont et al. [16].

$$\text{EY (\%)} = (N/N_0) \times 100 \quad (1)$$

where N is the number of viable entrapped cells released from the microcapsules, and N_0 is the number of free cells added to the LP-CP.

2.6.3. Swelling Properties

The swelling of LP-CP was evaluated in the simulated poultry digestive tract following the procedure described by Azad et al. [17], with a few modifications. Simulated gastric fluid (SGF) and simulated intestinal fluid (SIF) were freshly prepared according to Chitprasert & Ngamekaue [18]. SGF was made by dissolving sodium chloride (2 g) in deionized water (100 mL) and then adding 0.1 N HCl until the pH reached 1.2. The final volume was increased to 1 L. SIF was prepared by dissolving dipotassium phosphate (6.8 g) in sodium hydroxide (0.1 M, 190 mL). The pH of the solution was adjusted to 7.4 before increasing the final volume to 1 L.

For simulated gastric stage study, LP-CP (1.0 g \pm 0.05 g) was added into SGF (100 mL). The pH of the solution was adjusted from 1.2 to 2.0 and LP-CP was incubated at 39.5 °C \pm 0.5 °C in a Memmert WNB 14 thermostat water bath for 60 min, 120 min and 180 min, respectively, to represent the time of the microcapsules in the stomach *in vivo*. After incubation at each point, the LP-CP was filtered and weighed whereas the SGF solution was collected for use in the intestinal stage.

For simulated intestinal stage study, trypsin solution (2 mg/mL, 1 mL), bile solution (40 mg/mL, 14 mL), pancreatic solution (3.2 mg/mL, 7.5 mL), and SIF (7.5 mL) were mixed with the above SGF solution. The pH of SIF was adjusted from 7.4 to 5.5 whereas LP-CP was incubated at 39.5 °C \pm 0.5 °C in each pH for 220 min and 240 min, respectively, to represent the time in the intestinal tract *in vivo*. The treated LP-CP was collected and weighed at each incubation time. The results were calculated by equation (2) to indicate the percentages of swelling [18].

$$DS = [(W_s - W_0) / W_0] \times 100 \quad (2)$$

where W_0 and W_s are the weights of the dry and the swollen microcapsules after 4 h, respectively.

2.6.4. Release Performances

The release performance of LP-CP was carried out as described by Azad et al. [17]. After incubation at each point, the supernatant (1 mL) was collected, and the bacterium released was determined using the pour plate technique in MRS agar. The index of cell release was calculated by equation (3).

$$\text{Cell release (\%)} = \frac{\text{Released bacteria count at different times (log CFU/mL)}}{\text{Initial bacteria count (log CFU/mL)}} \times 100 \quad (3)$$

2.6.5. Impact of Storage Condition

The storage stability of *L. paracasei* and LP-CP was studied in three replicates according to Mitsuwan et al. [6]. Briefly, free *L. paracasei* (10 mL) and LP-CP (10.0 g \pm 0.5 g) were separately sealed in glass vials and wrapped with aluminum foil. The containers were stored at room temperature for 90 days. Samples were taken after 30 days, 60 days, and 90 days of storage to determine the viability in terms of tolerance characteristics.

2.6.6. Effect of Acids, Enzymes and Temperature on the Viability of Free Cells and LP-CP

Tolerance characteristics were measured thrice with 5 mL of free cells and 5.0 \pm 0.5 g of LP-CP according to Mitsuwan et al. [6]. Acid tolerance was investigated by soaking the sample in citrate phosphate buffer pH 2 (0.2 M, 20 mL). Bile salt tolerance was carried out by dissolving 3.0 g of porcine bile extract into DI (100 mL) with the sample. Trypsin tolerance was tested by preparing trypsin from bovine pancreas (1.0 g) in DI (100 mL) containing the sample. All tubes were incubated in a thermostat water bath vibrator at 39.5 °C \pm 0.5 °C for 30 min. For thermal treatment, the sample was tested at 85 °C for 1 min (simulation of feed pelleting condition). The treated samples were immediately removed to measure the viability as in the previous procedures.

2.7. Statistical Analysis

All samples were statistically analyzed in triplicate by one-way analysis of variance (ANOVA) in a completely randomized design. The mean was further evaluated using Duncan's new multiple-range post-hoc test. The statistical significance was considered at $P < 0.05$. Results were presented as mean \pm standard deviation.

3. Results and Discussion

3.1. Property and Phytochemical Composition of *C. pareira* Leaf Extract

C. pareira leaf was mixed and crushed with deionized water to prepare the plant extract. It was noticed that the extraction procedures did not degrade the extract's gelling characteristics. The extract preserved its original pH of 3.4 and formed a gel that could be reversed by temperature adjustments. The extract was a deep shade of impenetrable green. The extraction procedure may remove a large proportion of insoluble fiber and protein while keeping fiber-free carbohydrates, which include phenolic compounds, organic acids, starch, sugars, and calcium. The proximate composition of the extract is shown in Table 1. The presence of divalent cations may significantly affect the interactions between soluble fiber molecules. It is well-known that soluble fiber, divalent cations, and fiber-free carbohydrates are the most significant components of freshly extracted (manually prepared) *C. pareira* for gel production [19]. The presence of substances that release green color, such as phenolic compounds, significantly influences the process [20].

Table 1. Calculated composition of *C. pareira* (as-fed basis).

Items	Calculated nutrient composition (%)
Dry matter	94.55
Metabolizable energy (kcal/kg)	3,846
Crude protein	16.64
Ether extract	1.33
Crude fiber	16.22
Soluble dietary fiber	13.22
Insoluble dietary fiber	3
Total Carbohydrate	48.90
Total Ash	11.47

The chemical structure of *C. pareira* was verified using FTIR spectroscopy data (Figure 2). The absorb peak at 3320 cm^{-1} was due to O-H stretching vibrations. The peak at 2915 cm^{-1} was attributed to C-H absorption bands of CH_2 , CH_3 , and O- CH_3 stretching vibrations. Peak at 1403 cm^{-1} was due to asymmetric C-H vibrations of CH_3 . The characteristic peaks at 1603 cm^{-1} (COO^- asymmetric stretching) and 1347 cm^{-1} (COO^- symmetric stretching) could correspond to carboxylic acid vibrations, while esterified C=O vibrations were found at 1725 cm^{-1} (symmetric stretching) [20]. In general, the absorb peak at 1603 cm^{-1} and 1725 cm^{-1} could be classified as pectin. Pectin that exhibits low levels of esterification was earlier reported to have a higher peak intensity of around 1603 cm^{-1} [21]. In contrast, Pectin with greater esterification has a higher peak intensity at 1725 cm^{-1} (ester carbonyl band). The isolated *C. pareira* pectin exhibited a high peak intensity at 1635 cm^{-1} , implying low levels of esterification. The "finger print" area for pectin, located at 829-1134 cm^{-1} , includes C=O stretching and C-H bending modes [21]. This is qualitatively consistent with prior research, which revealed the characteristics of pectin with FTIR [11].

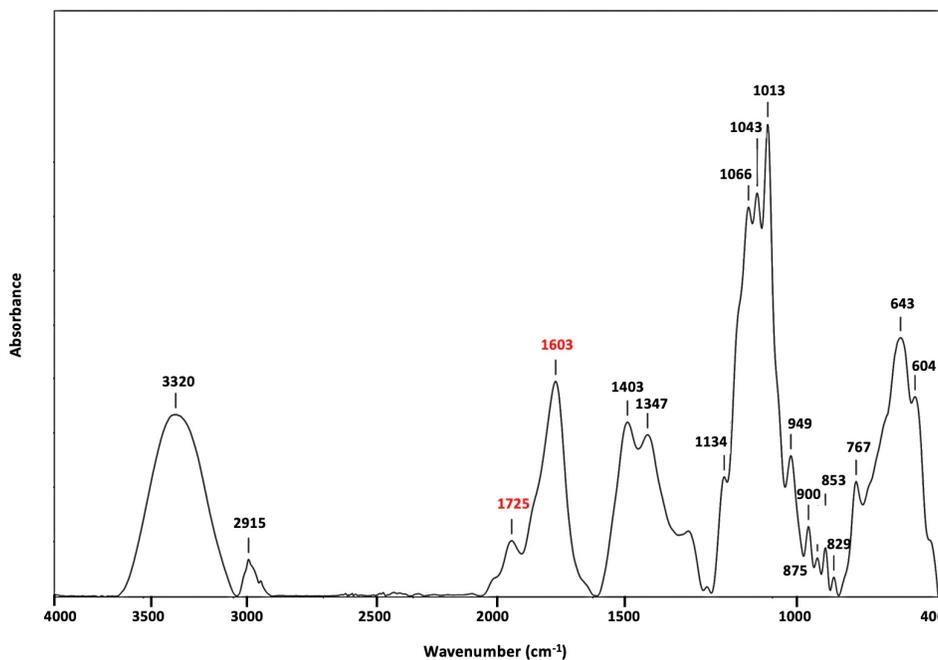


Figure 2. FTIR spectra of *C. pareira* extract.

3.2. Characterization and Performance of Microcapsules

3.2.1. Structural Analysis

The structural characterization of the microcapsule was carried out by FTIR technique to analyze the functional groups and chemical interactions (Figure 3). The spectrum of *C. pareira* revealed the typical pyranose ring at 1630 cm^{-1} apart from hydroxyl group (OH stretching at 3423 cm^{-1}) (Figure 3a) [21,22]. In case of *L. paracasei* which are proteins, the amino groups (NH_2) and acidic carboxyl groups (COOH) can be confirmed at 1150 cm^{-1} (amide I), and 1010 cm^{-1} (amines) respectively (Figure 3b). For *C. pareira* microcapsule, in the absence of Ca^{2+} , the O-C-O stretching vibrations of these carboxylates give two characteristic infrared absorption bands near 1400 cm^{-1} and 1597 cm^{-1} . When Ca^{2+} ions were present, the bands were shifted to 1430 cm^{-1} and 1630 cm^{-1} , respectively (Figure 3c). Changes in the position of the carboxylate bands typically indicate that significant interaction has occurred between the carboxylates and the metal ion [16]. In the case of LP-CP (Figure 3d), the characteristic peaks of carboxyl, hydroxyl, amino groups and calcium chloride were shifted implying the successful preparation.

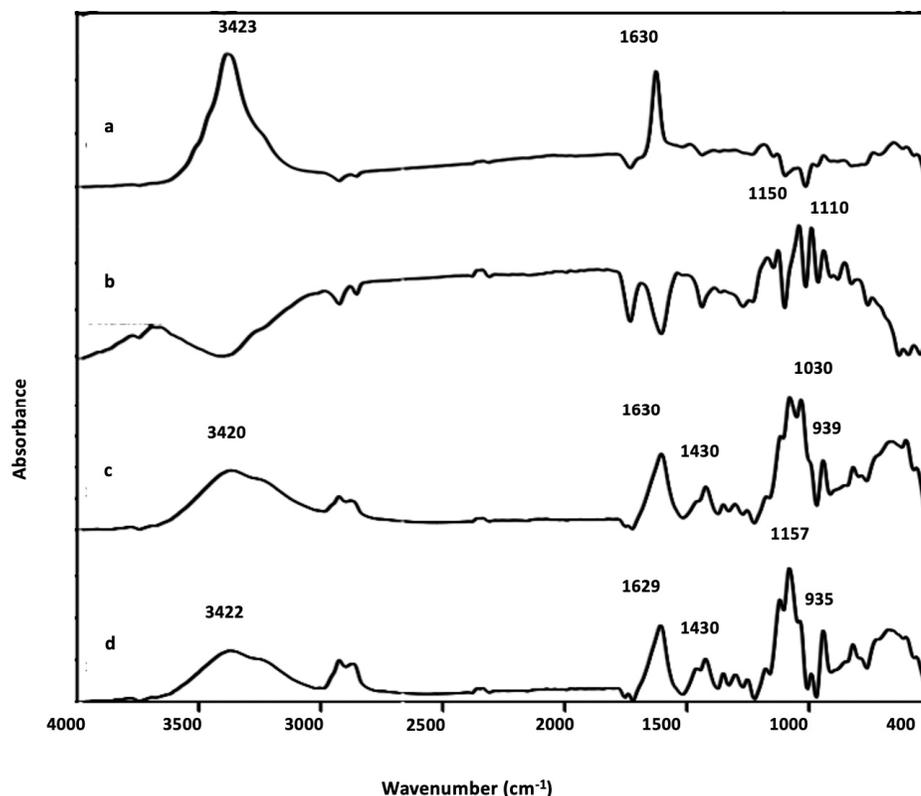


Figure 3. FTIR spectra of (a) *C. pareira*, (b) *L. paracasei*, (c) *C. pareira* microcapsule, (d) LP-CP microcapsule.

3.2.2. Morphology Analysis

The morphologies of sodium alginate and LP-CP were observed by SEM. On the sodium alginate surface, there are notable porous networks, as seen in Figures 4 (a)–(b). The diameter of the pores ranged from 45 to 65 μm (Figure 4c). CaCl_2 generates interconnected structures by ionic gelation with sodium alginate. In the case of LP-CP microcapsules, the surface of microcapsules became smooth (Figure 4d–4e). However, the drying process caused the shrinkage of the capsules and resulted in flat shapes, as shown in Figure 3d. The flat and irregular surface was due to water loss during the drying process [22]. On the surface, the porous structure of LP-CP was identifiable. (Figure 4f), demonstrating that the porous networks of *C. pareira* were preserved. The surface reveals a dense network, particularly on the surface, that might be related to the *C. pareira*- CaCl_2 chains formed through crosslinking. The average diameter of LP-CP was 300 μm which is within the range of the feed ingredients or feedstuffs (approximately, 400 μm - 2,000 μm) as reported by Abdollahi et al. [23]. The LP-CP exhibited high encapsulation efficiency at $90.5 \pm 0.1\%$, indicating successful encapsulation.

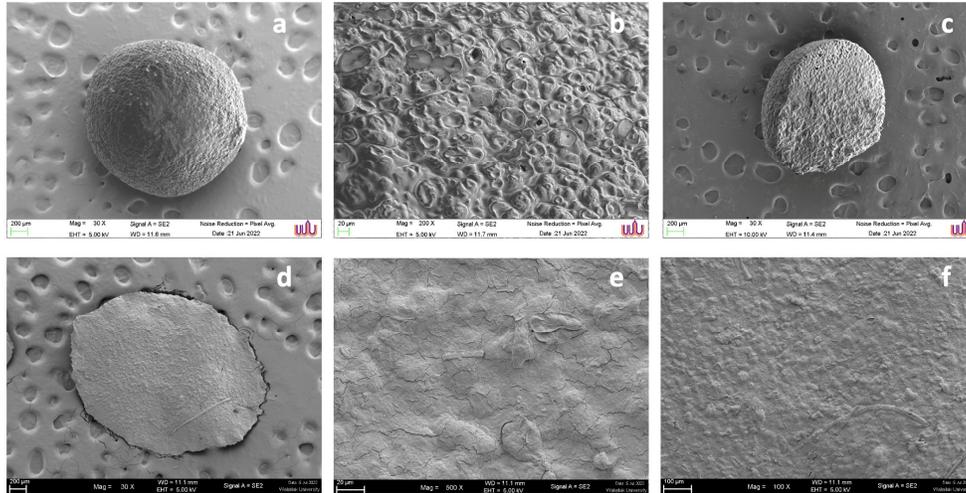


Figure 4. SEM micrographs of; (a) AL microcapsule, (b) surface of (a), (c) cross section of (a), (d) Encapsulated LP-CP microcapsules, (e) surface of (d), and (f) Detail on surface of (d).

3.2.3. Viability of LP-CP Under Pelleting Temperature

In order to produce pelleted animal feed, the pelleting process requires temperatures between 75 °C and 85 °C for 2-3 minutes [23]. The question is whether *L. paracasei* encapsulation by *C. pareira* maintains cell viability at these temperatures. According to Su et al. [24], the viability of free cells (*L. plantarum*) after heating at 85 °C was as low as 70-80 % from the initial. Similarly, Santos et al. [25] showed that after treating at 80 °C, the vitality of free *Lactobacillus* spp. was between 20% and 30%. The survivability of LP-CP microcapsules after 3 minutes of heating at various temperatures is summarized in Figure 5. In the beginning of the experiment, 91.12% of *L. paracasei* was determined to be viable. After three minutes of isothermal treatment at 75 °C and 80 °C, the viability of *L. paracasei* remained constant. ($P < 0.05$). When subjecting the microcapsules to heat treatment at 85 °C, the viability of *L. paracasei* slightly declined (75.60%). It demonstrates that the LP-CP microencapsulation technology may be compatible with feed pelleting. The results were corresponding with data from the components of microcapsules. (Figure 3c).

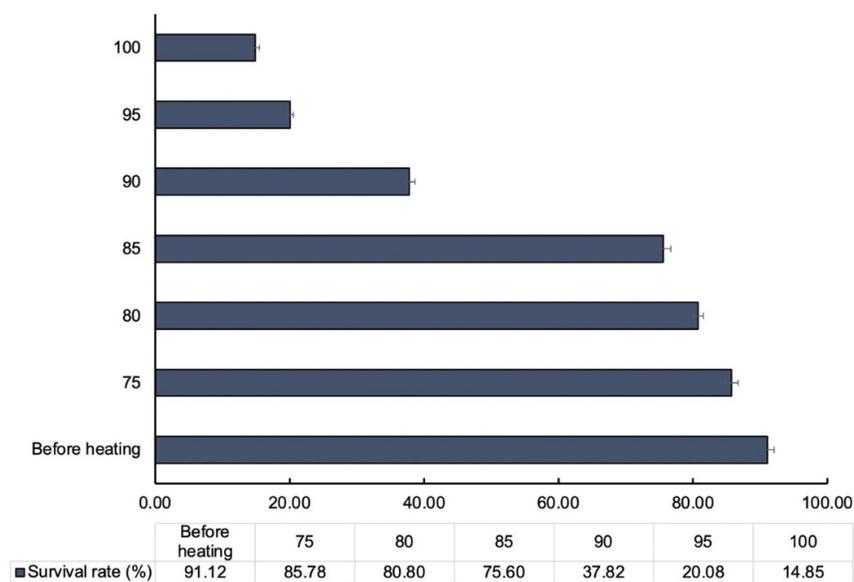


Figure 5. Survival of *L. paracasei* in *C. pareira* microcapsule after heating at different temperatures for 3 min.

3.3. In Vitro Studies of Microcapsules

3.3.1. Swelling and Cumulative Release Studies

To assess the release of *L. paracasei* in the animal digestive tract based on SIF and SGF, swelling and cumulative release tests were conducted. The swelling at different incubation periods is shown in Figure 6a. For the intervals of 60, 120, and 180 minutes in SGF, the swelling values were $17.63\% \pm 1.0\%$, $35.23 \pm 1.5\%$, and $58.04\% \pm 2.0\%$, respectively. The total weight of LP-CP microcapsules rapidly increased to double times of the original weight. After 180 minutes of SIF treatment, LP-CP microcapsules swelled by $89\% \pm 2.5\%$.

The swelling indicates how *C. pareira* responds to digestive fluid. During SGF treatment, gastric acidic pH (pH 2-3) is expected to induce CaCl_2 protonation, allowing water penetration and swelling. Continuous swelling is initiated by deprotonation of *C. pareira* due to treatment in SIF, which is moderately neutral (pH 6.5) but higher than the pKa of *C. pareira* (3.6). Anionic hydrogels have a special characteristic whereby an increase in pH causes an increase in swelling [17].

There is a definite interaction between LP-CP swelling (Figure 6a) and *L. paracasei* cumulative release (Figure 6b). At the initial stage of the incubation (60 minutes), the release rate was approximately 27%. The first burst of release may be related to *L. paracasei* present on the *C. pareira* microcapsule surface. The cumulative release of *L. paracasei* in SGF was approximately 66% at 180 minutes. The release in SGF is almost 38% if the pretreatment to exclude burst release (28%) was taken into consideration. The cumulative release in SIF reaches 90% at 240 minutes. The controlled release of core matter is known to depend on the type of encapsulation material and its characteristics, such as cross-linking amount, pH, chemical interaction, and incubation period. [26]. In this study, controlled release by *C. pareira* might function synergistically in the following approaches. The degradation of the CaCl_2 cross-linker was initially interrupted because of an ion exchange between calcium and chloride ions in SGF. The structural destruction of microcapsules facilitated SGF penetration of *C. pareira* micropores. Although variations in pH and endogenous enzymes in the intestinal tract, the carboxylate groups of *C. pareira* chains and their networks in SGF retained *L. paracasei*. In the last stage, the *C. pareira* crosslink network began to break down as *C. pareira* deprotonated in the SIF, leading to diffusion in *L. paracasei*. Corstens et al. [27] reported that carboxyl groups swelled during the ion exchange between calcium and phosphate ions in SIF. In this work, calcium ion crosslink network was disrupted, allowing *L. paracasei* to be slowly released. According to the above-mentioned about the release mechanism, structure and ionic crosslinks of *C. pareira* with CaCl_2 resulting in the successful release of *L. paracasei*. This is supposed to occur in the lower region of the gut.

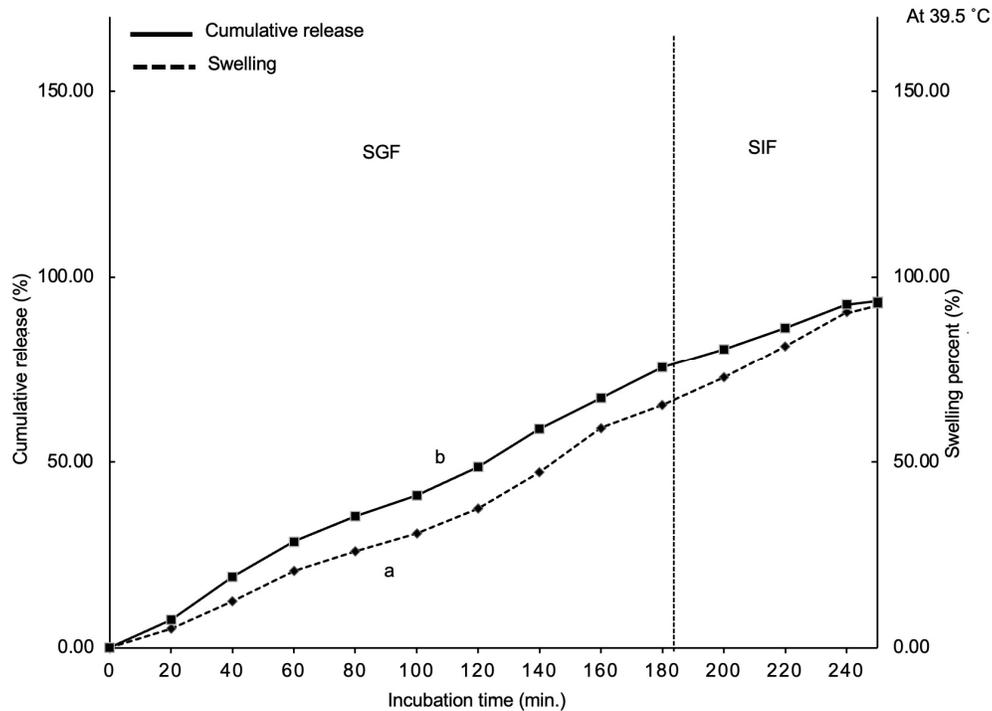


Figure 6. Performances of LP-CP in terms of (a) swelling percent (%), and (b) cumulative release (%) in SGF and SIF at 39.5 ± 0.5 °C.

3.3.2. Effect of Acids, Enzymes and Temperature on the Viability of Free Cells and LP-CP

When probiotics are encapsulated, it is important to make sure that the barrier of protection is effective in a substance that simulates gastrointestinal fluid, including bile, acid, and enzyme [28]. Based on the number of bacteria in the sample, the functions of *C. pareira* in retaining the survival of *L. paracasei* were examined. At the beginning of the experiment, viability amounts for *L. paracasei* and LP-CP were 91.48 ± 0.1 and $91.83\% \pm 0.2\%$, respectively (Figure 7). Apparently, acid and heat are the worst conditions. Microencapsulated *L. paracasei* still able to maintain the viability as high as 69%. In contrast, free cell viability decreased significantly to 18%-95% ($p < 0.05$). According to Mitsuwan et al [6] reported that under acid conditions, microcapsules of probiotics showed a significantly enhanced survival rate of probiotic bacteria by approximately 80% when compared with the non-encapsulated cells. Likewise, another previous research demonstrated that a decrease in survival of free cells was found to be nearly twice when compared with encapsulated cells (at pH 2) [28]. Our results reflect that the structure of *C. pareira*, which consists of pectin, may provide probiotics with protection from the harmful effects of severely unsuitable conditions.

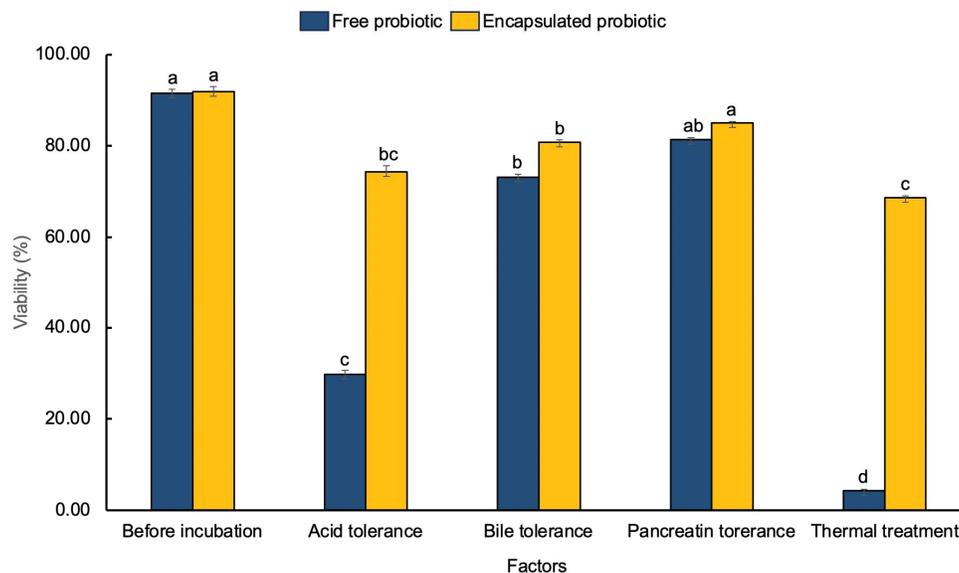


Figure 7. Survival of Free cell and LP-CP was treated under acid, bile, and trypsin tolerances including thermal treatment.

3.4. Impact of Storage Condition

Probiotics need to be kept for a certain amount of time before usage in typical applications. During that period, the probiotics may lose viability due to storage conditions such as temperature, humidity, oxygen content, and light exposure [29]. These main issues harm bacteria cell walls, reducing their effectiveness and leading to shorter shelf life. Therefore, the viability of *L. paracasei* during storage was investigated in this study. Table 2 demonstrated that *L. paracasei* survival gradually decreased over the first 30 days and significantly declined by almost 100% after 90 days ($P < 0.05$) at both 4°C and 32°C. Regarding LP-CP, almost no decrease in viability was observed in any storage durations at 4 °C or 32 °C ($P < 0.05$) at 30 days. The viability declined (40%–50%) when the storage time was extended to 60 and 90 days. This supports our previous study [16] where also sodium alginate encapsulation was reported to extend probiotic viability under storage conditions when compared with the free cells. This agrees with Silva et al. [29], revealing that encapsulation *L. acidophilus* was viable for a storage period of 45 days at 5 °C and had a count greater than 8 log CFU/g. The loss in viability of free cells and microcapsules at high temperatures is probably caused by the oxidation of lipids membrane and the denaturation of proteins, which leads to the destruction of macromolecules in bacterial cells [30]. The reason might be that probiotic microorganisms contained in protective carriers preserve cellular structures, reducing external stress by preventing molecular mobility.

It is established that microcapsules can be used as the ingredients to produce human food and animal food. Our results could suggest that the microcapsules prepared by *C. pareira* extract may be applied to animal feed or human food production. The plant extract acted as the barrier to protect from various conditions simulating the GI tract as well as the high temperature during the process of feed production. Although, the microcapsules exhibited a flat rather than a sphere shape, but they could maintain *L. paracasei* cells.

Table 2. Survival of free cells and LP-CP expressed as a function of storage time at difference temperatures.

Sample	Storage Time (Day)	Survival Amount	Survival Amount
		at 4°C (%)	at 32°C (%)
Free cell	0	91.10 ^a	90.85 ^a
	30	79.15 ^b	65.50 ^b
	60	56.24 ^c	13.05 ^c
	90	30.55 ^d	0 ^d
Microcapsule	0	90.50 ^a	90.75 ^a
	30	86.35 ^{ab}	82.95 ^{ab}
	60	71.52 ^b	62.68 ^b
	90	60.97 ^c	50.10 ^c

Superscripts a, b, c, and d, are for significant differences ($P < 0.05$) compared with before storage (day 0).

4. Conclusions

The present study revealed the microencapsulation of *L. paracasei* using *C. pareira* extract as the natural encapsulating material forming gelatin like structure with CaCl₂. The microcapsules showed a desirable result of encapsulation efficiency (90.5%) which was confirmed by the analyses of their chemical structures. Under extreme acid and thermal conditions, bacterial viability in the microcapsules was significantly increased when compared to non-encapsulated bacteria. During storage conditions of 90 days, LP-CP viability remained at 50%, whereas the survival rate of free cells significantly decreased by 100%. This information suggests that *C. pareira* is a potent polymer to be used as an eco-friendly material for the microencapsulation of probiotics.

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