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Article

Catalase Detection via Membrane-Based Pressure Sensors

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Abstract: Membrane-based sensors (MePSs) exhibit remarkable precision and sensitivity in detecting pressure changes. MePSs are commonly used to monitor catalytic reactions in solution, generating gas products crucial for signal amplification in bioassays. They also allow catalyst quantification by indirectly measuring the pressure generated by the gaseous products. This is particularly interesting to detect enzymes in biofluids associated with disease onset. To enhance the performance of a MePS, various structural factors influence membrane flexibility and response time, ultimately dictating the device's pressure sensitivity. In this study, we fabricated MePSs using polydimethylsiloxane (PDMS) and investigated how structural modifications affect the Young's modulus (E) and residual stress (σ_0) of the membranes. These modifications have a direct impact on the sensors' sensitivity to pressure variations, observed as a function of the volume of the chamber (Σ) or of the mechanical properties of the membrane itself (S). MePSs exhibiting the highest sensitivities were then employed to detect catalyst quantities inducing the dismutation of hydrogen peroxide, producing dioxygen as gaseous product. As a result, catalase enzyme was successfully detected using these optimized MePSs, achieving a remarkable sensitivity of $(22.7 \pm 1.2) \mu\text{m/nM}$ and a limit of detection (LoD) of 396 pM.

Keywords: pressure sensors; PDMS membranes; catalase; bioassays; membrane deflection

1. Introduction

Pressure sensors are important tools for achieving a high control over progression of chemical reactions and overflow rates in “Lab-on-a-Chip (LoC)” platforms. Given the reduced space available for their integration, conventional mechanical pressure gauges are often located outside the LoCs leading to complications of the experimental set-up. Although commercial systems are convenient, they are not appropriate for LoC miniaturized systems due to their limited response time and high-pressure range [1].

To overcome these problems, innovative miniaturized technologies for detecting pressure changes in LoCs are currently under study [2]. As common approach, the pressure-induced deformation of a membrane integrated to a LoC is detected using a variety of electrical, piezoresistive and optical methods [3–10].

The thickness of the membranes is usually in the range of hundreds or tens of micrometers and the material is often polydimethylsiloxane (PDMS) because of its mechanical properties, optical transparency and easy manufacturing that explain its widespread use in LoCs and microfluidics [11–16]. The pressure range of many integrated membrane pressure sensors is a few kPa [7,8,17].

In a previous work [18], we have monitored the dioxygen (O_2) evolution resulting from catalytic hydrogen peroxide (H_2O_2) dismutation by using a Membrane-based Pressure Sensor (MePS) operating in a range between 2 and 50 Pa. A MePS chip consists of an array of microchambers of 8

mm diameter integrating a 2 μm thin PDMS membrane whose deflection as response to pressure changes is detected using a high-resolution camera. The interest in H_2O_2 dismutation catalyzed by ruthenium embedding microcapsules was due to its applications in pumping-free LoCs [19] and fast mixing microfluidic platforms [20]. However, H_2O_2 dismutation induced by catalase-like catalysts is also widely exploited in the biosensing field to monitor the glucose-driven cellular respiration [21] and to amplify the immunoassay signal in highly sensitive chips [22,23].

Among H_2O_2 -fed biosensors, Volumetric Bar-Chart Chips (V-chips) have been recently attracted attentions as cheap and power-less microsystems for detecting circulating tumor cells [24,25] and myocardial infarction biomarkers in serum [25], aflatoxin B1 [26] and Ochratoxin A [27] in beer, and lead ions in biological and environmental samples [28]. In V-chips, O_2 generation due to dismutation of H_2O_2 is directly proportional to the amount of analytes that interact with a suitable probe conjugated with the catalyst [24,25,29,30] or that react with a gel (formed by the capture probe and the catalyst) by releasing a proportional amount of catalyst in the supernatant [22,26,28,31]. The gas production leads to a volumetric expansion that is readable through an integrated bar-chart.

On the other hand, H_2O_2 -fed biosensors enable the direct quantification of enzymes producing gaseous products, like catalase, with an important role in biological defense system [32]. In humans, the abnormal levels of catalase are related to diseases such as diabetes [33], cancer [34], cardiovascular [35] and Alzheimer's diseases [36]. Among current studies of catalase detection, based on spectrophotometric [37], electrochemical [38] and colorimetric methods [39], liquid crystal-based sensing platforms have been proposed as cheap and sensitive tools for catalase sensing in human serum [40,41].

Here, we have developed a catalase sensor, as a strategic component for LoC platforms, biosensors or biological assays. The sensor is composed of a MePS chip, optimized in its operation through the modulation of structural parameters such as Young's modulus (E) and residual stress (σ_0) of the membrane it comprises. These parameters are crucial for determining the sensor's sensitivity to pressure variations, both as a function of the chamber volume and the mechanical characteristics of the membrane itself (Σ and S , respectively). The use of MePS can, therefore, be finely directed toward the detection of small quantities of an enzyme that produces O_2 during the dismutation of H_2O_2 . The achieved results are explicative to develop catalase biosensors for diagnostic applications and immunological assays based on MePS.

2. Results and Discussion

The design of the MePS has been optimized to enable highly sensitive detection of catalase activity. To this aim, several types of MePSs differing for their chamber diameter, membrane thickness and internal volume have been produced and characterized (see Table 2 for a detailed description). For each type of MePSs, parameters like Young's modulus E , residual stress σ_0 , and sensitivities both of the chamber dimension and of the membrane.

The operational performance of MePS was evaluated in relation to its sensitivity to the most significant structural parameters: Young's modulus E and residual stress σ_0 . E measures the elastic resistance of the membrane to deformation, influencing the sensitivity of the sensor and its response time. The residual stress σ_0 , which remains on the membrane after the initial stimulus has been removed, can influence the response of MePS to long reaction times. This happens especially in the analysis of slow reactions and affects the re-usability of the chip itself. More importantly, the effect of residual stresses on the structural elements of the MePS sensing device (i.e., membrane thickness or chamber volumes) could cause loss of linearity and deterioration of load capacity. (Σ and S , respectively) have been calculated (see Table 1).

Table 1. Young's modulus E , residual stress σ_0 and Sensitivity values of the chamber Σ for all the chips as extracted from the pressure-deflection data.

d (μm)	2r = 5 mm	2r = 8 mm	2r = 10 mm
	PL-MePS1	PL-MePS2	PL-MePS3
2	$E = 1,75 \pm 0,90$ MPa	$E = 3,28 \pm 0,46$ MPa	$E = 5,32 \pm 0,92$ MPa
	$\sigma_0 = 0,094 \pm 0,003$ MPa	$\sigma_0 = 0,220 \pm 0,055$ MPa	$\sigma_0 = 0,310 \pm 0,094$ MPa
	$\Sigma = 0,525 \pm 0,003$ Pa/μL	$\Sigma = 0,2090 \pm 0,0008$ Pa/μL	$\Sigma = 0,1370 \pm 0,0008$ Pa/μL
	PL-MePS4	PL-MePS5	PL-MePS6
10	$E = 1,51 \pm 0,98$ Mpa	$E = 1,73 \pm 0,67$ MPa	$E = 2,11 \pm 0,74$ MPa
	$\sigma_0 = 0,004 \pm 0,002$ MPa	$\sigma_0 = 0,051 \pm 0,007$ MPa	$\sigma_0 = 0,068 \pm 0,008$ MPa
	$S = 0,525 \pm 0,002$ Pa/μL	$S = 0,2070 \pm 0,0003$ Pa/μL	$S = 0,1360 \pm 0,0004$ Pa/μL
	PL-MePS7	PL-MePS8	PL-MePS9
50	$E = 1,23 \pm 0,78$ Mpa	$E = 1,43 \pm 0,81$ MPa	$E = 1,59 \pm 0,51$ MPa
	$\sigma_0 = 0,003 \pm 0,001$ MPa	$\sigma_0 = 0,074 \pm 0,003$ MPa	$\sigma_0 = 0,011 \pm 0,002$ MPa
	$S = 0,513 \pm 0,002$ Pa/μL	$S = 0,2060 \pm 0,0008$ Pa/μL	$S = 0,1340 \pm 0,0002$ Pa/μL

The sensitivity of the chamber dimension $\Sigma = \Delta P / \Delta V$, represents the ratio between the variation of pressure applied and the increment in the loaded liquid volume. A high value of Σ indicates that, with the same volume loaded, the device can detect a greater variation in the reaction pressure as output. The membrane sensitivity $S = \Delta w / \Delta P$ of a MePS can be described as the minimum input of the internal pressure change that creates a detectable output of membrane deflection. MePS devices with higher S values are more efficient since they allow to detect greater variations of membrane deflections with the analogous reaction pressures.

The reliability of a membrane sensor is ensured through two key factors: i) maintaining a linear regime for deformations and residual stresses, and ii) exhibiting an elastic response to varying degrees of deformation. To identify the most efficient MePS from those listed in Table 2, the study involved the analysis of the structural parameters and characteristic curves obtained from bulge tests.

2.1. Bulge Tests Theory

The bulge test is a method to evaluate mechanical properties of a circular membrane by applying a uniform strain in the radial direction (Figure 1a). The membranes, assembled to the MePS chambers, are deformed by the weight force of water droplets added with aliquots of 10 or 20 μL volumes. The applied pressure P determines a hemispherical deflection of the membrane whose maximum w is measured by data processing of an image acquired by a high-resolution camera. Typical plots of deflection w (mm) versus the added volume V (μL) are shown in Figure 1b and in Figure 2. Considering that P in this specific situation is essentially due to the weight of liquid volume, it can be calculated by applying the fundamental law of the hydrostatic pressure of a liquid by the action of gravity and the forces acting on the liquid surface (Stevino's law), equation 1:

$$P = \rho g (h + w) \quad (1)$$

with ρ representing the water density, g the gravity acceleration, and h the height of the water level in the chamber. The loading pressure P and the maximum membrane deflection w are related by the following relationz, equation 2

$$P = \frac{C_1 d}{r^2} \sigma_0 w + \frac{C_2 f(v) d}{r^4} \frac{E}{1 - \nu} w^3 = Aw + Bw^3 \quad (2)$$

where r is the membrane radius, d is its thickness, and ν is Poisson's ratio. The geometrical coefficients C_1 , C_2 , and $f(v)$ for circular membranes are 4, 2.67, and 1, respectively.

The equation 2, written in a synthetic form with coefficients A and B , has been used as fitting function of the P/w curves shown in Figures 3a, S1a, S2a, S3b-c. A and B describe the elastic response of the membrane as residual stress σ_0 and Young's modulus E .

On the other hand, by plotting the Pressure P (equation 1) versus the water loaded volume V (Figure 3b, S1b, S2b) and the deflection w versus the Pressure P (shown in Figure 4), the sensitivity of the chamber Σ (Pa/ μ L) and the sensitivity of the membrane S (μ m/Pa) can be respectively calculated.

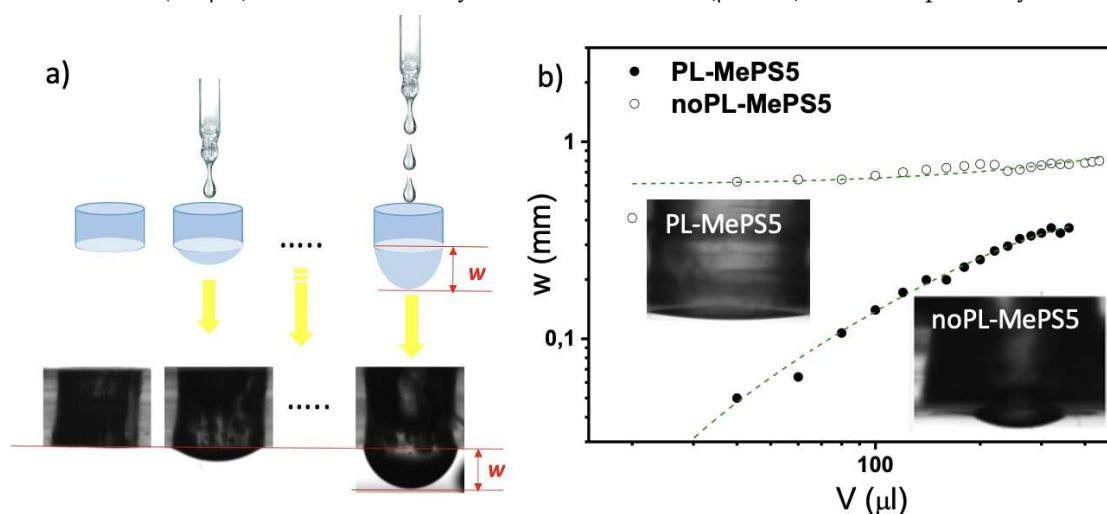


Figure 1. a) Scheme of a MePS chip and of the method used to detect membrane deflection during the calibration steps of the bulge test (upper part). In the lower part, real images of the chip acquired by a high-resolution camera are shown; b) Graph in the log-log scale showing deflection (w) versus the water loaded volume (μ L) during the bulge tests performed on two different devices with the same chamber dimensions (diameter of 8 mm) and membrane thickness (10 μ m) but assembled without (noPL-MePS5, empty dots) or with the use of plasma treatment (PL-MePS5, full dots). The reported dotted lines better show the linear trend of PL-MePS. Images of membrane deflection of PL-MePS and no-PLMePS are reported as insets.

2.2. Plasma-Induced effects on Membranes

In the integration process, two distinct methods were employed to incorporate the sensing membrane into the MePS chips: oxygen plasma treatment (PL-MePS) and a mortar layer (noPL-MePS) (details in Section 2.3 of Materials and Methods and par. "Plasma-induced effects on membranes" in SI). We find that plasma treatment increased membrane hydrophilicity, by modifying surface composition with the creation of a thin silica-like layer at the interface [42–44]. This enhanced hydrophilicity in PL-MePS chips led to more consistent curves with a broader linear range of responsivity. Therefore, subsequent studies were conducted exclusively using PL-MePS sensors. NoPL membranes, on the contrary, show reduced sensitivity due to non-linear behavior.

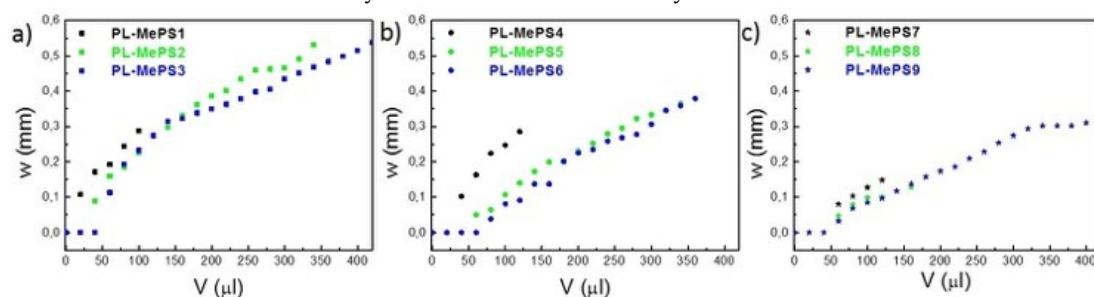


Figure 2. Graphs showing three different sets of measurements (deflection (w) versus the water volume (V) loaded in the chamber) acquired on PL-MePSs with variable chamber diameters: 5 mm (PL-MePS1; PL-MePS4; PL-MePS7); 8 mm (PL-MePS2; PL-MePS5; PL-MePS8) and 10 mm (PL-MePS3; PL-MePS6; PL-MePS9) and membrane thicknesses fixed at a) 2 μ m, b) 10 μ m, and c) 50 μ m.

The reproducibility in the PL-MePSs is reported in Figure S4 and discussed in SI (par. “Curve reproducibility”). At lower pressures, some differences due to the fabrication process may cause a higher inertia to mechanical deformation [45]. A proper calibration of each MePSs before use may eliminate this problem.

2.3. MePS Calibrations

Hence, all the plasma-treated MePSs produced have been tested. Their curves are reported in Figure 2. Thinner membranes deflect more efficiently than thicker ones at similar loading volumes. The behaviour of curves in 8 and 10 mm diameter chambers sometimes overlaps. Membranes interfaced to smaller chambers deflect more than those interfaced to large diameter chambers; indeed, for membranes with the same thickness, a fixed water volume exerts a larger pressure on a smaller surface (as in the 5 mm diameter chambers), thus determining larger deflections. Furthermore, membranes in larger chambers need higher volume of liquids to overcome the original inertia to deflection ($V > 50 \mu\text{L}$) and show different slopes of the w trend as function of loaded volume.

From the curves of Figure 2, further fittings have allowed to evaluate the parameters above mentioned: Young's modulus E , residual stress σ_0 , and sensitivity of the chamber dimension Σ . The plots from which all these values have been calculated are reported in Figure 3 for PL-MePS1-3 with membrane thickness of $2 \mu\text{m}$ and for all the other chips with thicker membranes (PL-MePS4-6 with $10 \mu\text{m}$ thick membrane and PL-MePS7-9 with $50 \mu\text{m}$ thick membrane) in Figure S1 and S2.

The results of all the fits are shown in Table 1. An average value of (1.75 ± 0.90) MPa is found for PL-MePS1, which is in agreement with previous works on plasma-treated PDMS membranes of similar thickness [46–49].

Considering E , a decrease of its absolute value is observed by increasing the thickness of the membrane. An explanation of the experimental differences in the E values can be attributed to the fabrication method of membranes. Indeed, PDMS membranes are well known to have Young's modulus values larger than bulk PDMS [46–49] in a range between 12 kPa–2.50 MPa, depending on the processing conditions. The differences have been attributed to a pre-stretching of the PDMS chains of the membranes produced by spin coating (at 6,000 RPM for a relatively long time up 150 s) [45] compared to the relaxed state of PDMS chains of membranes prepared by pouring on a flat substrate. Here, we believe that the higher rigidity of the $2 \mu\text{m}$ thin membranes of PL-MePS1 is due to the higher rotation speeds that are applied during spin coating fabrication procedure compared to $10 \mu\text{m}$ (PL-MePS2) and $50 \mu\text{m}$ thick (PL-MePS3) membranes (2500 RPM and 3000 RPM for $50 \mu\text{m}$ and $10 \mu\text{m}$ membranes, respectively, *versus* 6500 RPM for $2 \mu\text{m}$ thin membranes). The stiffness of PDMS membranes also increases with its diameter, as also observed in other works [45].

Regarding σ_0 , $2 \mu\text{m}$ thin membranes have a value one order of magnitude higher than 10 and $50 \mu\text{m}$ thick membranes that show more similar values. Chambers of 5 mm diameter have σ_0 values one order of magnitude smaller than 8 and 10 mm chambers. So thicker membranes of small diameters dissipate residual stress better than thinner one, as expected for elastic materials.

Concerning the sensitivity of the chamber Σ , according to the data reported in Table 1, definitely small diameter chambers (5 mm) are more sensitive to pressure changes as function of the loaded volume (first column data). Furthermore, compared to 8 and 10 mm chambers, they also show a lower elastic resistance to the deformation (lowest E values compared to larger chambers) and a lower residual stress σ_0 which means higher deformability and less impact on linearity and load capacity of the sensors.

However, since the deformation of chips of same radius in response an applied internal pressure represents a delicate balance between d , E and σ_0 (see equation 1), it is important to calculate the parameter sensitivity of the membrane, S , to evaluate the more convenient device between the three 5 mm chamber devices with different membrane thickness: PL-MePS 1, 4, or 7. S is maximized when at small pressure change ΔP corresponds large membrane deflection Δw . Figure 4 is a plot of the deflection dependence $w = f(P)$ for the 5 mm chambers with higher sensitivity Σ . As reported in Figure 4, the highest membrane sensitivity value S value is observed for the PL-MePS1 chip types and corresponds to $(5.9 \pm 0.41) \mu\text{m}/\text{Pa}$. Hence, although characterized by a higher stiffness and residual

stress, the chip more sensitive to small pressure changes was found to be the PL-MePS1. Indeed, considering the dependence of P from w of equation 1, it seems that the reduction of the geometric parameter d impacts more than the increases of σ_0 and E (that represent the mechanical properties of the membrane) on the deflection w that increases for thinner membranes.

Given the performances of the different devices, it was decided to use PL-MePS1 as sensor for the catalase study.

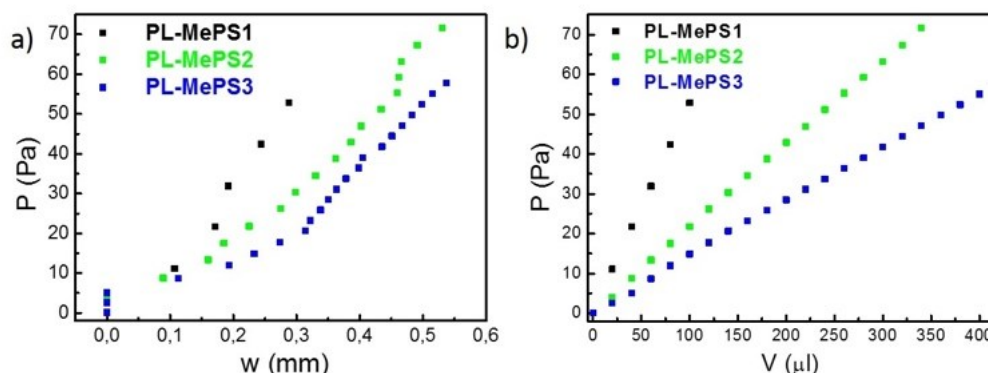


Figure 3. Plots from curves of Figure 2a to calculate a) the residual stress σ_0 and Young's modulus E , b) Sensitivity of chamber dimension for PL-MePS1-3 with membrane thickness of 2 μm and diameter of 5, 8 and 10 mm respectively.

2.4. Catalase Studies

The study on catalase was performed at the loading volumes of 40 μL and 100 μL to detect a different ranges of catalase concentrations. These volumes were chosen since at $V \geq 40 \mu\text{L}$ the deflection w of different sets of chips was found less dependent on the fluctuations due to the fabrication process, as discussed above and in SI (par. "Curve reproducibility"). 20 μL or 50 μL of catalase at concentrations ranging from 50–500 nM and 0.5–100 nM or were mixed with 20 μL or 50 μL of 30% H_2O_2 and inserted into the chamber of the MePS. The MePS was sealed as reported in the experimental section and the initial membrane deflection w_0 , due to the weight of the liquid, was calculated.

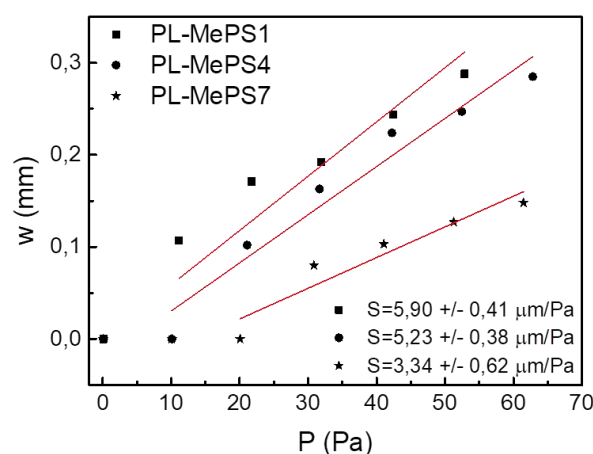


Figure 4. Plots from black curves of Figures 2 to calculate the Sensitivity of the membrane S for PL-MePS1, PL-MePS4 and PL-MePS7 with membrane thickness of 2, 10 and 50 μm , respectively. The inset in the bottom reports the sensitivity values obtained by a fitting procedure.

Figures 5a and 5b represent the evolution over the time of the value $w-w_0$ which corresponds to the membrane deflection resulting from the generation of O_2 during the reaction while Figure 5c reports the pressure exerted during the reaction. As expected, the production of O_2 steadily increases

with rising catalase concentrations until it reaches a plateau. After 300s for both loading volumes, the O_2 amount remained nearly constant across all tested concentrations. The sensitivity and the limit of detection (LoD) of MePS sensor are calculated from the experimental calibration curve of deflection versus catalase molarity (see Figure 5d) [50,51]. Sensitivity of the MePS, obtained at the equilibrium state (around 300 s from the reaction starting point) as slope of the graph of Figure 5d, is $(22,7 \pm 1,2) \mu\text{m/nM}$ for a 100 μL loading volume and $(3,3 \pm 0,7) \mu\text{m/nM}$ for a 40 μL loading volume.

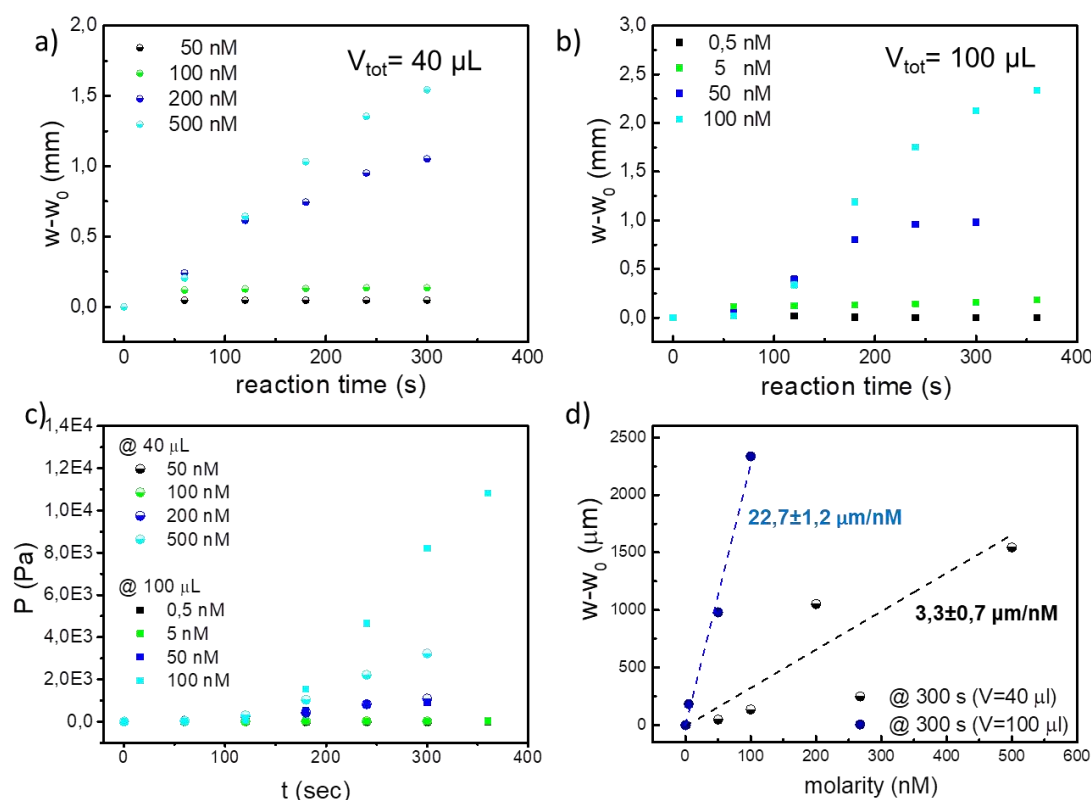


Figure 5. Deflection of the thin membrane during the catalase reaction as function of time, for PL-MePS1 filled with a) 20 μL of catalase solution (concentration of 50, 100, 200 and 500 nM) and 20 μL of 30% H_2O_2 and b) 50 μL of catalase solution (concentration of 0.5, 5, 50, 100 nM) and 50 μL of 30% H_2O_2 . c) Pressure evolution due to the catalytic reaction is reported as function of time. d) Deflection due to the reaction, detected at 300 s from the starting point of the reaction, is reported as function of the solution molarity. The plot is used as a calibration curve for the sensitivity calculation. A sensitivity of $22,7 \mu\text{m/nM}$ is determined for the completely loaded chamber.

Differences in the PL-MePS1 sensitivity of catalase at different loading volumes can be attributed to the presence of a dead volume in the chamber not occupied by the liquid and hence, to an equilibrium between the O_2 partitioned in gas phase and O_2 dissolved in liquid phase, which may impact on the overall pressure generated inside the chamber and evaluated as membrane deflection.

Then the LoD is obtained by dividing the sensor resolution with its sensitivity and a value of 396 pM for the catalase PL-MePS1 completely loaded is estimated. This LoD value gives a confidence interval of the analyte concentration for which sensor responses are reliable.

The O_2 moles produced during the catalytic reaction are calculated using the previously elaborated model for MePS sensors [18]. The O_2 amount for each well is then analyzed as a function of membrane deflection at 180 s from the start of the catalytic reaction (corresponding to a timescale of sensor linearity range). The graph in Figure 6a demonstrates that regardless of the sensor loading volume, MePS deflection can reveal the amount of oxygen produced by the reaction.

Sensors filled with a different volume of sample appear to have different behaviors when compared as a function of catalase molar concentration within the chamber (Figures 6 b,c). The

deflection produced during the reaction (Figure 6 b) and the amount of O_2 (Figure 6 c), released in the linearity range of the sensor (at 180 s from the start of the reaction), are clearly dependent on the volume loaded into the MePS chamber, as shown by the fitting values in Figures 6 b and c, and correlated to the MePS sensitivity.

However, if the O_2 concentration is plotted as a function of the reaction deflection for the two different loading volumes (as in Figure 6 d), it is found that the normalized behaviours of the two sensors are analogous and comparable. A similar quantity of oxygen produced during the reaction corresponds to an analogous deformation of the membrane. More importantly, the trend of O_2 produced follows a cubic power-law, as in the standard bulge test, with a fitting $R^2 > 99\%$.

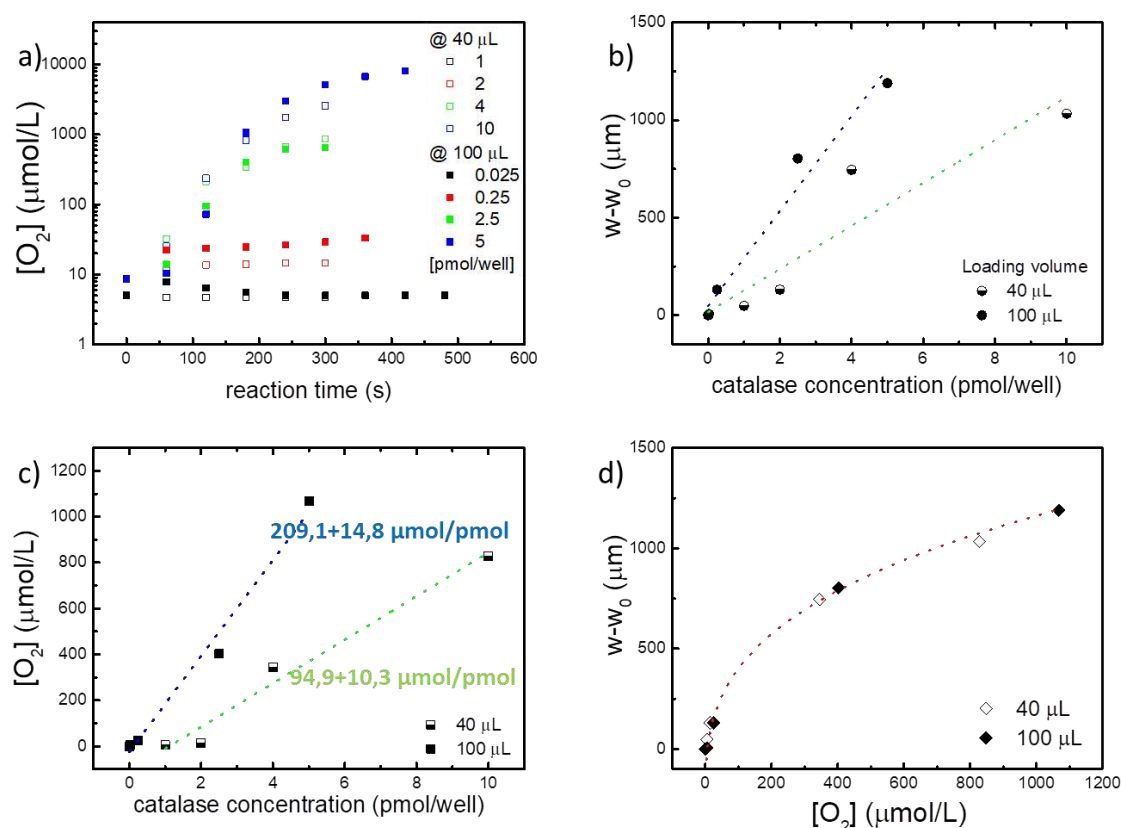


Figure 6. a) Molar concentration of O_2 ($[O_2]$, $\mu\text{mol/L}$) produced during the catalytic reaction at different catalase concentration and for MePS microchamber filled with sample volumes of 40 μL and 100 μL . The O_2 molar concentrations are obtained by measuring the deformation of the membrane during the reaction inside MePS. b) Deflections due the catalytic reaction and c) O_2 molar concentration at different catalase concentrations are analysed at the reaction point of 180 s, corresponding to a linearity regime for MePS sensor. A fitting over both datasets shows a twice sensitivity of the sensor when the chamber is full at 100 μL . d) The O_2 molar concentration as a function of membrane deflection at 180 s from the start of the catalytic reaction for the two different load volumes.

Therefore, regardless of how the sensor is filled, its linearity and correspondence with the expected behavior are confirmed. MePS is reliable and its behavior is highly predictable and always respected, even when the chamber is partially or totally full. This result suggests the extreme design flexibility of the MePS according to the specific use. In addition to the possibility of modulating the geometric and membrane characteristics (thickness, diameter, Young's modulus), it is also possible to modulate the load volume based on the concentrations to be detected. We have seen that, for example, for catalase concentrations higher than 1 pmol/well even partially filling the chamber is

sufficient. This allows not to saturate the deflection of the membrane with the gravitational effect due to the weight of the liquid.

Our MePS-based chips for catalase detection show quite relevant results, compared to the known literature, in terms of sensitivity ($(22.7 \pm 1.2) \mu\text{m/nM}$) and LoD (396 pM) [22,26,30,31,38–41]. Furthermore, considering that MePS technology can easily evolve towards low-cost portable devices (e.g., by smartphone integration), Point-of-Care tests for catalase detection in biofluids or immunoassays based on H_2O_2 -fed biosensors can be implemented.

3. Materials and Methods

Soda-lime microscopic glass slides were provided by Pearl; CLEVIOS PH 500 were purchased from Heraeus Clevios GmbH (Germany) and toluene from J. T. Baker (U.S.A.). Sylgard-184; a two parts poly(dimethylsiloxane) (PDMS) elastomer, was purchased from Dow Corning (U.S.A.); sulfuric acid (H_2SO_4 , 98%), hydrogen peroxide (H_2O_2 , 30%), catalase from bovine liver (2,000-5,000 units/mg protein) and potassium phosphate buffer (PB) were purchased from Sigma-Aldrich (Milan, Italy). Milli-Q water with a resistivity of 18.2 M Ω cm was used. PES syringe filters (0.45 μm) were purchased from Sartorius Stedim (Germany).

3.1. Fabrication of MePS Components

All the MePS chips reported in Table 2 consist of 1) reaction chambers of variable volumes and diameters and of 2) PDMS membranes of different thicknesses. To fabricate the chambers, a PDMS pre-polymer/curing agent ratio fixed to 10:1 in weight was mixed and poured in a glass Petri dish to obtain (6.5 ± 0.5) mm thick slides after curing at 140 °C for 15 min in oven. After removal from the glass dish, reaction chambers of different diameters were produced using suitable punchers. PDMS membrane preparation consists of several steps: a) cleaning of the glass slides using piranha solution (3:1 H_2SO_4 : H_2O_2), then washing with milli-Q water and drying under nitrogen flow, b) deposition by spin coating of CLEVIOS PH 500 solution (filtered through the PES syringe filter) on the clean glass acting as sacrificial layer followed by bake on a hot plate at 120 °C for 5 min, c) preparation of pre-polymer/curing agent solution (10:1 weight ratio) diluted with toluene (67% in weight), and d) spin-coating of PDMS-toluene solution on the sacrificial layer and curing in oven at 70 °C overnight.

3.2. MePS Assembly

MePS chips were assembled with two procedures: by plasma treatment (PL-MePS) and without plasma treatment (noPL-MePS). In PL-MePS chips, the punched slide and the PDMS thin membranes deposited on the glass substrate were both plasma treated and put in conformal contact. After sealing with further pre-polymer/curing agent and polymerization at 140 °C for 15 min (using additional PDMS as glue), the slide-membrane assembly was put into water while stirring, to transfer the membrane on the chambers and remove the glass substrate. Then, washing by pure water was performed to remove CLEVIOS PH 500 residues on the membrane side that was in contact with the glass substrate. In noPL-MePS chips, the plasma treatment was replaced with the “mortar layer” method [52] using a 10 μm layer of PDMS diluted in toluene deposited by spin coating on a cleaned glass substrate. The punched slide and the edges of the PDMS membranes were put in contact for 10 s with the PDMS uncured layer and bonded at 140 °C for 15 min in oven. No additional sealing with PDMS as glue was used prior to membrane transfer step.

3.3. Bulge Test

To test the sensitivity of the MePS chips reported in Table 2, water was dropped into the chamber with steps of 10 or 20 μL that locally deformed the membrane. As the volume of water increased, the membrane underlying the droplet extended under the gravity force. The membrane deflection (w) of the MePS chips was monitored by a CAM 200 (KSV Instruments Ltd., Finland) instrument, measured by using ImageJ software analysis and plotted versus water volume (see Figure 1a). The error on the w estimation was $\pm 3 \mu\text{m}$.

3.4. Catalase Experiments

For the experiments, the complete sensor was directly loaded with the test solutions. The solutions of catalase in 50 mM PB used were 0.5, 5, 50, 100, 200 and 500 nM. 50 μ L or 20 μ L of catalase were mixed in the reaction chamber of MePS chips respectively with 50 μ L or 20 μ L of 30% H₂O₂; then, the chips were immediately sealed, and the membrane deflection was imaged by the camera and followed at the timepoints of 60 s up to 5-8 min (see Figure S5).

Table 2. Types of MePS chips produced with main features like chamber diameter (2r), membrane thickness (d) and chamber internal volume (V_{in}).

MePS chips	2r (mm)	d (μ m)	V _{in} (μ L)
PL-MePS1	5	2	120
noPL-MePS1	5	2	120
PL-MePS2	8	2	340
PL-MePS3	10	2	500
noPL-MePS3	10	2	500
PL-MePS4	5	10	120
PL-MePS5	8	10	340
noPL-MePS5	8	10	340
PL-MePS6	10	10	500
PL-MePS7	5	50	120
PL-MePS8	8	50	340
PL-MePS9	10	50	500

4. Conclusions

Membrane-based sensors are highly sensitive transducers for detecting pressure changes within confined microsystems. Reactions generating gases and triggering pressure shifts can serve as catalyst detectors and as amplification mechanisms for analyte detection in bioassays. One key reaction generating O₂ involves the disproportionation of H₂O₂, a process naturally catalyzed by the enzyme catalase but also achievable through various synthetic catalase-like compounds.

In humans, catalase has been linked to numerous physiological and pathological conditions. Recent studies have unveiled unknown functions of this enzyme as well as the importance of direct catalase detection in biofluids for diagnostic purposes. Catalase-like systems have been also used in biological assays as amplification methods to detect analytes with a concentration proportional to the catalyst, by estimating the volume of the gaseous products produced. In membrane-based sensors, these gaseous products induce a pressure increase, measured by assessing membrane deflection.

In this work, we report the fabrication of an effective PDMS membrane-based sensor (MePS) obtained from optimized parameters of Young’s modulus E , residual stress σ_0 , and sensitivity of both chamber size Σ and membrane S . Several MePS have been fabricated and characterized as sensors, modifying membrane structural parameters (such as membrane thickness or chamber diameter) and monitoring their role in sensing during a catalytic reaction.

Besides the effect of an oxygen plasma treatment on the reliability and response linearity of the membrane device used as a sensor has been clarified. Oxygen plasma treatment was found to be crucial to enable a uniform deflection of the membrane and a reproducible response mainly due to the resulting increased hydrophilicity of the membrane.

Finally, our results demonstrate that the sensitivity of the MePS is dependent on the loading volume, although its linearity and correspondence with the expected behavior are always preserved. By loading 100 μ L of reactive solution, a sensitivity of $(22,7 \pm 1.2) \mu\text{m/nM}$ was obtained with LoD of 396 pM. According to these results, the MePS technology appears very sensitive to volumetric changes and is amenable for portability, improvements in sensitivity and LoD (i.e., using more performant catalase-like systems), and integration into multifunctional chips because of its easy miniaturization.

Supplementary Materials: The following supporting information can be downloaded at the website of this paper posted on Preprints.org, Figure S1: title; Table S1: title; Video S1: title.

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