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## Article

# Graphite Nanoplatelets Films as Multifunctional Protective Layer in Kevlar/Nomex Sandwich Composites

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**Featured Application:** Multifunctional protective layer for fiber reinforced composites.

**Abstract:** In the aerospace sector, structural and non-structural components are usually subjected to a wide range of environmental conditions. Although lightweight, the use of composite parts for these applications is restricted, due to their poor performance in severe environmental conditions. Specifically, moisture can seriously damage the materials' performance, reducing their mechanical, thermal, electrical, and physical properties as well as the service time. A strategy to improve the materials' resistance in humid environments is the use of protective lightweight barrier coatings to reduce the diffusion of gases and/or liquids in composites. Nanolamellar nanocomposite characterized by the high in-plane orientation of nanoplatelets effectively works as barrier structures by generating tortuous paths for molecule diffusion. In this work, the effectiveness of protection against water uptake of nanocomposites reinforced with Graphite NanoPlatelets (GNPs) at high filler content (70, 80 and 90%wt) for Kevlar sandwich panels of a nacelle engine of an ATR42 aircraft has been investigated. Moisture uptake and Ground Air Ground (GAG) tests were carried out in an environmental chamber to reproduce temperature and relative humidity profiles representative of a real-scale application for a regional aircraft composite part. The high filler content of the graphene films ensures a high level of tortuosity by delaying and reducing moisture absorption by −71% compared to the unprotected panel, showing good barrier properties.

**Keywords:** graphite nanoplatelets; moisture uptake; barrier properties

## 1. Introduction

The use of composite structures in a wide range of fields is still increasing, thanks to their high specific strength and stiffness over metals. The significant weight reduction of the structures translates into a benefit for the industry in terms of cost savings and environmental benefits, especially in terms of fuel savings [1]. However, it is well known that the mechanical properties of fibre-reinforced composites can be very sensitive to environmental parameters. Temperature, humidity, light, or environmental impact phenomena such as rain, lightning or hail affect the durability of composite materials becoming a crucial issue when they are used as structural materials [2]. These severe environmental conditions can limit the use of polymeric composites, especially for applications in which high performances are required. In an aircraft, some primary structures are always subjected to varying temperature and moisture exposures throughout their entire service life.

Continuous exposure to high temperature and moisture levels may lead to a change in the mechanical behaviour of the aircraft's structural components [3,4]. Furthermore, cyclic thermal, moisture and mechanical loadings could even further deteriorate the performance of the structures leading to premature failure. These environmental conditions, especially temperatures and moisture exposures, vary widely during a flight from the take-off to the landing.

Specifically, Kevlar-reinforced composites are prone to moisture uptake, due to the strong hygroscopic nature of both the resin and the fibre. Epoxy resins are strongly sensitive to water absorption because of large amounts of polar OH groups in the networks which facilitate attractive interactions with polar molecules of water [5–7]. It was found that the maximum moisture absorption for epoxy is 2% at 70 °C and 100% RH [8] and 6% for Kevlar 49 aramid fibres when they are exposed to a high-humidity environment (96% RH) [9]. The absorbed moisture leads to the plasticization phenomenon in the matrix resin, decreasing the glass transition temperature and also generating changes in the state of stress in favour of cracking through swelling [10–12], degrading the mechanical properties of the material [2–4]. As a consequence, the stress transfer at the fibre-matrix interphase is compromised and the overall behaviour of the composite becomes more brittle: the impact strength and fracture toughness can even drop by 50% [13].

In aerospace applications, the use of Kevlar honeycomb sandwiches is widespread, thanks to their excellent mechanical performances, i.e., high bending stiffness to weight ratio [14]. Nevertheless, the honeycomb core absorbs water by capillary phenomenon [15,16]. The moisture locally accumulated inside the honeycomb cells, thereby jeopardising the original purpose of adopting sandwich structures for weight reduction and seriously affecting the integrity of composite structures and aircraft safety [17,18]. The solution currently used to resist water adsorption is based on metallic paints, applied through flame spray [19–21], which has some drawbacks, i.e., weight, and fragility [22]. Among a variety of options, polymer-based films, such as polyvinyl fluoride (PVF) films, are extensively used in aerospace applications thanks to the low permeability for vapours and low ignition time [23,24]. These hydrophobic films are usually co-bonded to surfaces of the glass, carbon, and Kevlar composite structures.

In the last decade, it has been investigated the possibility of using graphite nanoplatelets (GNPs) to improve the barrier properties of polymers [25]. These nanoparticles are promising nanomaterials in gas or liquid-barrier applications because perfect graphene sheets do not allow the diffusion of small gases or liquids through their plane. To fulfil this function, films with special architecture can be employed as protective coatings for composites [26]. Nanoarchitectures with a high level of nanoplatelet orientation and high nanofiller content (>50 vol%) ensure a high tortuosity factor resulting in high barrier and low permeability properties [27,28].

In this work, the authors investigated the possibility of employing high-content nanocomposites made with GNPs as a protective layer for reducing moisture absorption. Graphene films with high filler content have been integrated into Kevlar/Epoxy panels in a one-step manufacturing process. Panels have been fabricated with protective layers at different filler content (70, 80, 90 wt%). The effectiveness of protection against water uptake has been investigated through moisture uptake and GAG tests in the environmental chamber and the level of damage in the panels has been investigated with non-destructive inspection (NDI). Results indicate that graphene films are effective in protecting the panel from moisture, with a reduction of the maximum absorption after 40 days of 30%. The easy integration of the films in the manufacturing process of composites in a co-curing process ensures a reduction in the processing time and improves energy saving.

## 2. Materials and Methods

### 2.1. Materials and manufacturing procedure

Kevlar/Nomex sandwich panels were selected since they are commonly employed in the construction of an engine nacelle. Square samples of 180 mm side were fabricated, with a tapered core with an angle of  $18^\circ \pm 3^\circ$ , such that the footprint varies from  $140 \times 140 \text{ mm}^2$  to  $50 \times 50 \text{ mm}^2$  on the lower and upper surfaces, respectively (Figure 1).

The GNP-rich films were fabricated following a top-down approach, based on a spray deposition process. These self-standing films are pre-impregnated with an epoxy resin (Hexcel RTM6) at different contents (10, 20, 30 wt%). Details of the fabrication process are reported in previous work [29]. Films have a thickness of  $70\pm6\text{ }\mu\text{m}$  and a weight of  $80\text{-}90\text{ g/m}^2$ .

To evaluate the effectiveness of the GNP layer in reducing the water uptake uncoated and coated Kevlar sandwich panel samples were fabricated. Seven panels were manufactured including three couple of panels coated with films with different graphene content (70%, 80%, 90%) and one reference sample without protection, and are listed in Table 1.

Table 1. List of fabricated panels.

Panel ID	Description
P-REF	No coating
P70-A; P70-B	70 wt% GNP coating
P80-A; P80-B	80 wt% GNP coating
P90-A; P90-B	90 wt% GNP coating

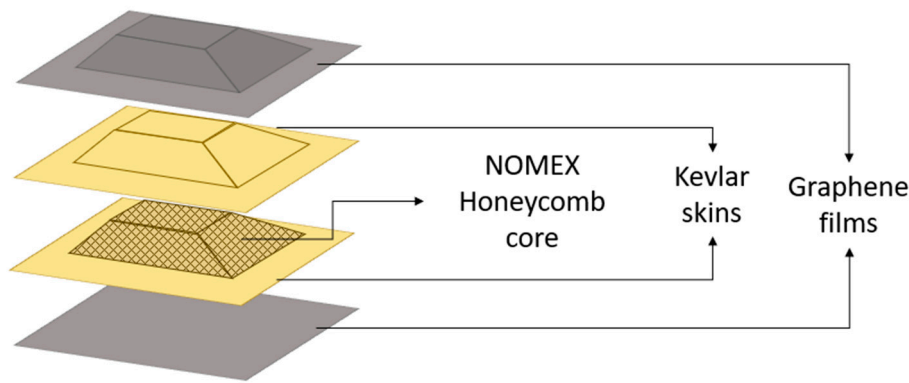
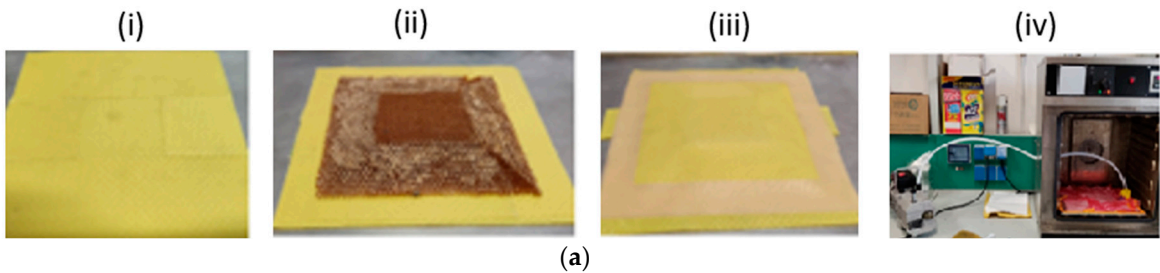


Figure 1. Geometry of the Kevlar sandwich panel.

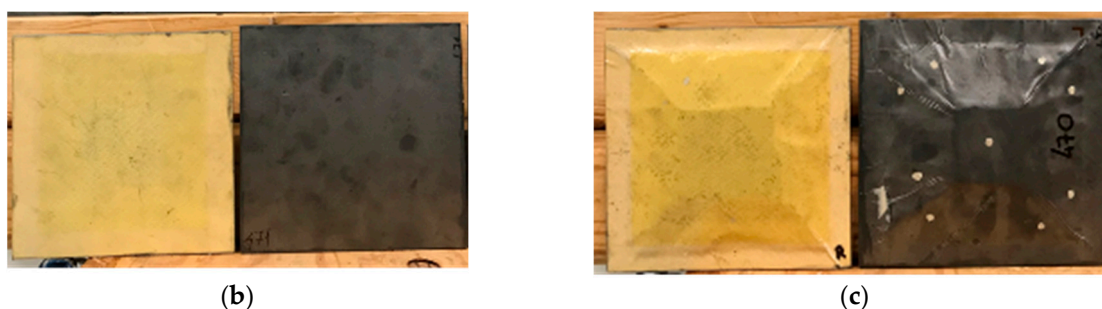
The manufacturing process was carried out by the following steps:

- i. Stacking of the lower skin according to the layout sequence (0/90)<sub>4</sub>.
- ii. Core positioning, the core was prepared according to the final geometry before lamination.
- iii. Stacking by prepregs the core contour (0/90)<sub>6</sub>.
- iv. Stacking the upper skin (0/90)<sub>4</sub>.
- v. Curing in an oven for 2h at 120 °C in a vacuum bag.

Finally, panels are trimmed and sealed on the edges. The GNP films on both the lower and upper surfaces were considered as additional layers in the stacking (i.e., tool side layer in the lower skin, last layer in the upper skin). Figure 2 shows the panels’ manufacturing process.







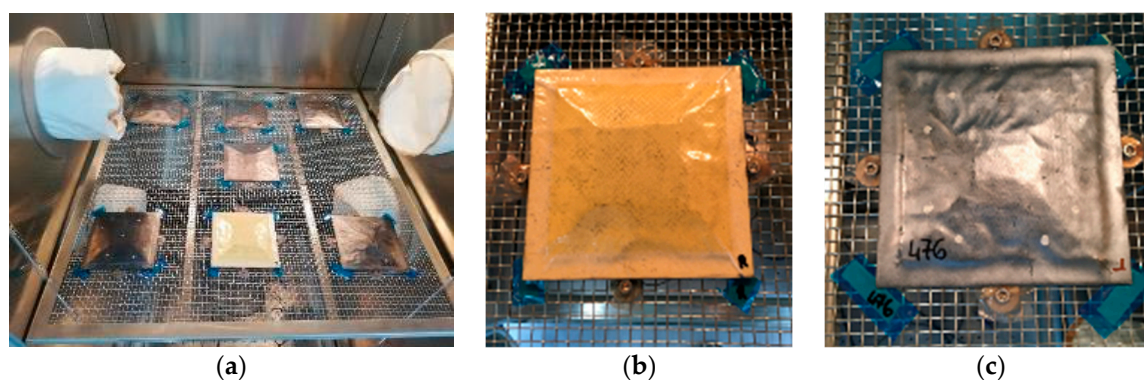
**Figure 2.** Panel manufacturing, stacking and vacuum bagging cure (a); External surfaces both uncoated and coated samples (b,c).

## 2.2. Experimental characterization

Thermogravimetric analysis (TGA) (TA Instruments Q500) was conducted to evaluate the real filler/matrix composition of GNP-rich films. Measurements were performed in an inert atmosphere, using nitrogen gas, with a temperature ramp of 10 °C/min from room temperature to 800 °C. The weight loss is evaluated at 600 °C, the temperature at which the resin residue is 10.3%.

The electrical resistivity of the graphene films was assessed using the Keithley 6221 picoammeter and the Metex M3850D digital multimeter. The picoammeter was used to generate currents with different densities, in the 1-100 mA range, while the multimeter was used to monitor the voltage. The measurement was conducted on rectangular samples of 5 cm x 1 cm size and a silver conductive paste was used to facilitate the measure. To prevent the samples from breaking during the measurements, they were reinforced with transparent adhesive tape.

To assess the effect of protection of GNP films, a moisture uptake test was performed according to ASTM D5229 [30]. All samples listed in Table 1, were placed in a climatic chamber (Angelantoni CH 2000) at a controlled temperature of 70 °C, with a relative humidity of 85% for 40 days. The increase of weight with time was measured by weighting the panels at different time steps. To avoid undesired movements of the specimens during the test, the panels were fixed on a metallic grid using four bolts accurately covered with tape, as shown in Figure 3. Before starting the test, the panels were conditioned in a vacuum oven at 80 °C for 3 weeks in order to remove any traces of moisture and fix the same starting point. The samples were weighed every 3 days in the first week and then every week as indicated in Table 2. Before each weighing, samples were dried with a paper towel to eliminate the excess water. The measurements were repeated three times, and the average value was computed and registered.



**Figure 3.** (a) Panels positioning on the metallic grid; (b) Uncoated panel; (c) GNP coated panel.

**Table 2.** Weighing schedule for moisture uptake test.

Weight	time (days)	time (hours)
$W_0$	0	0
$W_1$	3	72
$W_2$	6	144
$W_3$	13	312
$W_4$	20	480
$W_5$	27	648
$W_6$	34	816
$W_7$	41	984

Finally, non-destructive inspections (NDI) were conducted to investigate the internal structure of the panels. Analyses were conducted before and after the test, to assess the potential damage of the samples when placed in a moist environment. Lock-in thermography has been employed and images of the samples at 0.1 Hz and 0.04 Hz were acquired.

### 3. Results

#### 3.1. GNPs-protective coating functional properties

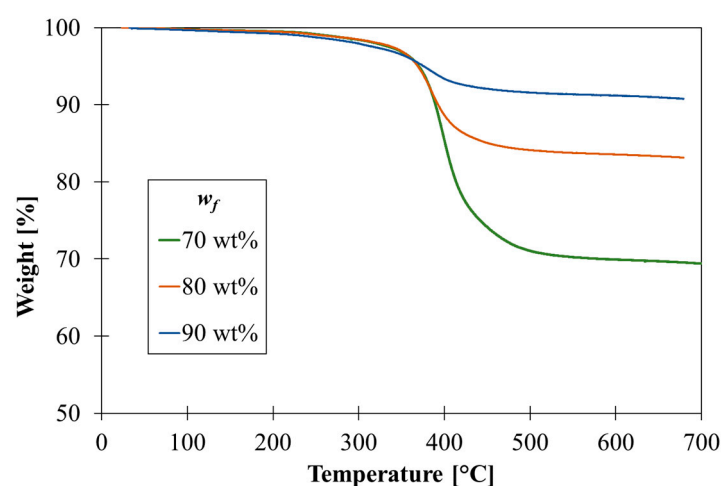
The real GNPs content has been assessed by TGA analysis. The thermograms reported in Figure 4 confirmed the correspondence between nominal and actual GNPs content in the films. The real filler content,  $w_{f,real}$ , is computed according to Equation (1):

$$w_{f,real} = \frac{R_i - R_m}{100 - R_m} \% \quad (1)$$

where  $R_i$  is the residue from TGA at 600 °C of the sample and  $R_m$  is the residue of the pure epoxy resin equal to 10%.

**Table 3.** Results of TGA and values of filler content.

$w_{f,nominal}$ [wt%]	$R_i$ [wt%]	$w_{f,real}$ [wt%]
70	72	69
80	86	84
90	91	90

**Figure 4.** TGA curves of graphene films at different GNPs content.

The electrical resistance ( $R$ ) of GNP-rich films with different filler contents was measured with the Volt amperometric test. The resistivity  $\rho$  is computed according to Equation (2), where  $R$  is the measured resistance and  $w$  and  $L$  are the width and length of the sample.

$$\rho = R \cdot \frac{w \cdot t}{L} \quad (2)$$

The films exhibit low electrical resistivity, which slightly decreases with increasing filler content as reported in Table 4. Also, these values are compatible with the electrical resistivity of graphite ( $1.3 \times 10^{-5}$ ), which represents the theoretical natural limit [31]. Graphene coating also improves the panels' electrical conductivity, providing as an additional feature, the dissipation of the electrostatic discharge (ESD) that can accumulate on the surface [32,33].

**Table 4.** Values of electrical resistivity of graphene papers.

Filler content [wt%]	Resistance, $R$ [ $\Omega$ ]	Width, $w$ [mm]	Length, $L$ [mm]	Thickness, $t$ [ $\mu\text{m}$ ]	Resistivity, $\rho$ [ $\Omega\text{m}$ ]
80	4.70	10	55	72	$8.54 \times 10^{-5}$
85	3.49	11	57	71	$6.75 \times 10^{-5}$
90	1.82	16	40	73	$4.96 \times 10^{-5}$

### 3.2. Moisture diffusion in sandwich plates

The sorption curves of the Kevlar/epoxy sandwich with and without the protection are shown in Figure 5, where moisture uptake is plotted against the square root of time. The moisture uptake,  $M$ , is evaluated according to Equation (3) [30]:

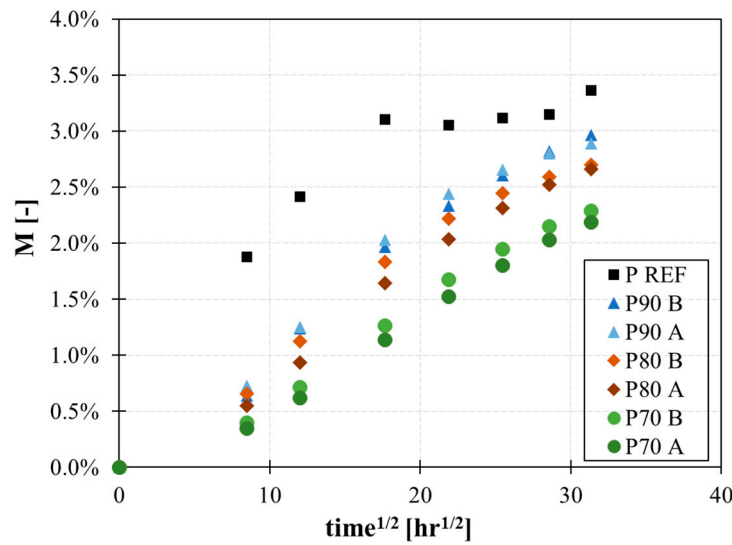
$$M(t) = \frac{W(t) - W_0}{W_0} \% \quad (3)$$

where  $W_0$  is the panel weight at time  $t=0$ , before starting the test, and  $W(t)$  is the weight of the panel at each weighing step of Table 2.

Results show that the graphene layer improves the resistance to moisture uptake, both in terms of maximum absorbance and diffusion rate. A reduction of 30% of the maximum moisture absorption has been observed in protected samples with respect to the reference Kevlar/epoxy sandwich (Table 5). The graphene layer also modifies the absorption curve. In the uncoated panel, the absorption rate is higher, and the saturation is reached earlier compared to coated panels. The presence of the graphene films lowers the absorption rate, thanks to the tortuous path created by the well-oriented nanoparticles.

**Table 5.** Maximum moisture uptake after 41 days and water diffusivity at 3 days ( $t=259200$  s) for tested panels.

Panel ID	$W_{f, \text{real}}$ [%]	M @ 41 days [%]	$\Delta M / \Delta M_{\text{max}}$ @ 3days [-]	$D$ [ $\text{m}^2/\text{s}$ ]
P REF	-	3.4	0.55	$3.8 \times 10^{-12}$
P70 A	70	2.2	0.16	$3.0 \times 10^{-13}$
P70 B	70	2.3	0.17	$3.7 \times 10^{-13}$
P80 A	77	2.7	0.21	$5.2 \times 10^{-13}$
P80 B	86	2.7	0.24	$7.2 \times 10^{-13}$
P90 A	91	2.9	0.25	$7.7 \times 10^{-13}$
P90 B	91	3.0	0.26	$8.0 \times 10^{-13}$



**Figure 5.** Moisture uptake with time of uncoated and coated panels.

Furthermore, the material's diffusivity ( $D$ ) has been calculated from the results of moisture uptake test as a function of time, according to Fick's second law. It describes the time-dependent absorption of a gas or liquid into a sheet by accounting for accumulation into the material. If a gas or liquid is absorbed in a homogeneous and infinitely long sheet, an increase in mass ( $\Delta M$ ) occurs as a function of time ( $t$ ) until saturation ( $\Delta M_{\max}$ ). At short times ( $Dt/h^2 < 0.06$ , or equivalently  $\Delta M/\Delta M_{\max} < 0.55$ ), Fick's second law can be approximated to a linear relationship between the relative mass gain and the square root of time (Equation (4)) [34].

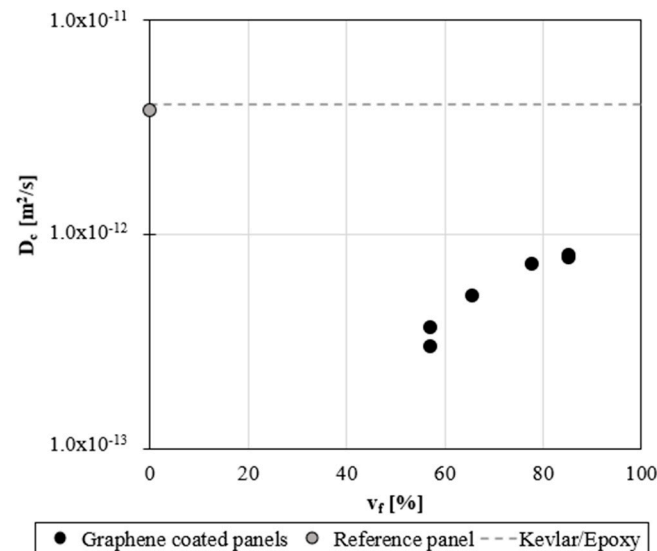
$$\frac{\Delta M}{\Delta M_{\max}} = \frac{4}{\sqrt{\pi}} \cdot \sqrt{\frac{D \cdot t}{h^2}} \quad (4)$$

$$D = \frac{\pi}{16} \cdot \frac{h^2}{t} \cdot \left( \frac{\Delta M}{\Delta M_{\max}} \right)^2$$

where  $\Delta M/\Delta M_{\max}$  is the relative change in mass, and  $t$  and  $h$  are the time and the thickness of the sample respectively.

A sensible reduction of the water diffusivity is found for protected panels, as shown in Figure 6. In the case of the unprotected panel, the diffusivity is  $3.8 \times 10^{-12} \text{ m}^2/\text{s}$ , consistent with data reported in the literature [35]. When graphene coating is applied, the diffusivity reduces of an order of magnitude, as shown in Figure 6. This reduction depends on the GNPs content: a minimum value is observed in the case of P70 with a  $\sim 90\%$  reduction compared to P REF. By increasing the GNP content, the efficiency of protection of the coating reduces; the diffusivity in P90 reduces by  $\sim 80\%$  compared to the reference.





**Figure 6.** Water diffusivity in the panels at 3 days.

## 4. Discussion

### 4.1. Influence of filler content on water uptake

Once the moisture is absorbed on the upstream surface of the material, it diffuses across the thickness of the protective films and penetrates the panel, accumulating in it. The use of protective coating with nanolamellar architecture showed a significant reduction of the moisture uptake and the absorption rate. The hydrophobic nature of GNPs combined with the very high in-plane orientation of nanoplatelets and the low matrix content, create an impervious path for vapour and gases [36].

For a material filled with impenetrable nanoplatelets, oriented perpendicular to the diffusion direction, the diffusivity depends on the matrix diffusivity and on the path that the molecules follow when they cross the two phases [37]. The longer the distance travelled by the diffusing molecule across the film thickness, the more tortuous the pathway.

A tortuosity factor ( $\tau$ ) is defined and described by Equation (4) for nanocomposites reinforced with lamellar-oriented nanoplatelets, which depends on the filler aspect ratio ( $AR$ ) and volume fraction ( $v_f$ ) [37]. It increases with the volumetric filler fraction, reducing asymptotically to zero.

$$\tau = 1 + \frac{AR}{2} \cdot v_f \quad (5)$$

According to Equation (4), the tortuosity factor in the case of graphene films loaded with 70, 80, and 90 wt% of GNPs are computed and listed in Table 6. By increasing the nanofiller content, the nanoarchitecture of the films, composed of highly oriented nanoplatelets should force the water molecules to move in an increasingly tortuous path, resulting in a reduction of diffusion through the protective film.

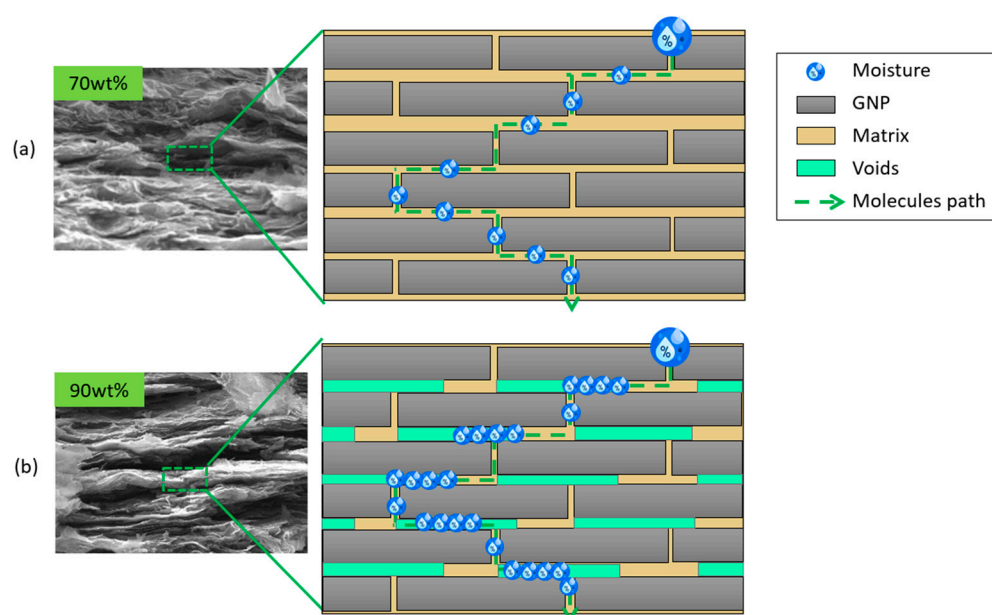
**Table 6.** Tortuosity factor in graphene films.

$w_{f,real}$ [wt%]	$v_f$ [vol%]	$\tau$ [-]
69	56	598
84	76	810
90	84	897

On the contrary, experimental data showed that in panels coated with GNP films, the absorption depends on the coating composition and increases with the increasing nanofiller content. The lowest absorption is found in the panel protected with graphene film @70 wt% of GNP.

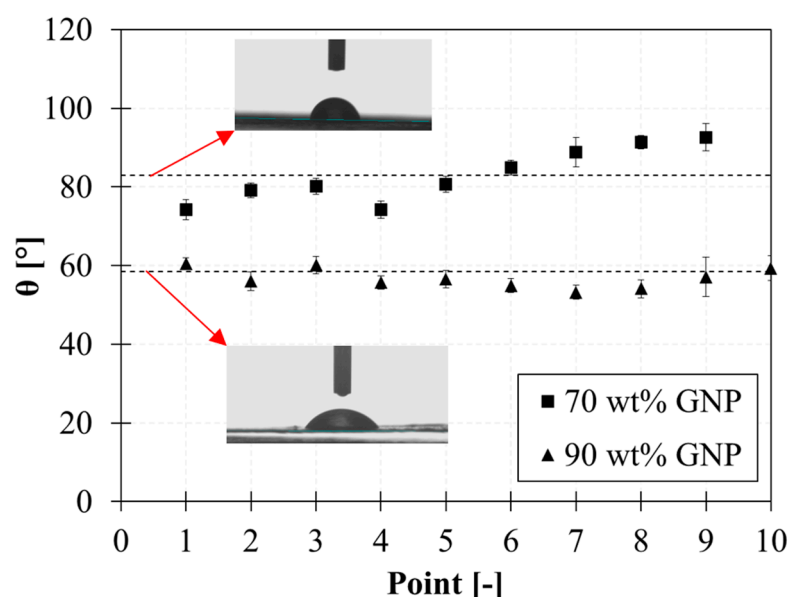
A rationale for this behaviour is found in the material's inner architecture. Morphological observations of previous works have revealed a maximum level of compaction in graphene nanocomposites with 70 wt% of GNPs [38]. By increasing the filler content up to 90 wt%, the nanostructure modifies being affected by dry spots and voids, as shown in SEM images of Figure 7. The matrix phase at high filler contents appears to be discontinuous by partially covering the nanoplatelets' surface [29,39]. These uncovered areas may be detrimental to the barrier effect. Although the water molecules travel a more impervious path, they cross areas with different diffusivities (air and matrix) resulting in a higher absorption (Figure 7).

Therefore, the moisture uptake in panels protected with graphene films with 90 wt% filler content is higher than that of panels coated with graphene film with 70 wt% filler content. However, this value is significantly lower than that of the uncoated panel (3.4%).



**Figure 7.** SEM images of graphene films and sketch their nanoarchitecture at the various filler content: (a) 70 wt% and (b) 90 wt%.

Further investigations based on contact angle measurements demonstrated that graphene films @90wt% of GNP have a more hydrophilic behaviour compared to graphene films @70 wt% as shown in Figure 8. Analyses are conducted using optical contact angle measuring (OCA 20 DataPhysics) and contour analysis systems and using distilled water drops of 1 mL. Figure 8 reports the contact angle ( $\theta$ ) geometrically defined as the angle formed by the liquid with the solid surface.



**Figure 8.** Contact angle measurements on graphene films @70 wt% and 90 wt%.

An average contact angle of  $83^\circ$  and  $59^\circ$  is found for films with 70 wt% and 90 wt% of GNPs, confirming that by increasing the nanofiller content [40,41]. These results indicate that the wetting states of membranes influence their permeability and corroborates the hypothesis that nanocomposites become more inclined to water absorption with increasing GNPs content.

#### 4.2. Ground Air Ground (GAG)

The Ground Air Ground (GAG) test was performed to assess the performance of the material during the flight. The test consists of subjecting the specimens to the combined effects of temperature, pressure, and relative humidity, thus reproducing the flight conditions. The GAG test was conducted using the same combined environmental chamber used for the moisture uptake test and samples were tested for 10 repeated cycles. The test procedure is shown in Figure 9. According to ASTM D5229 [30], samples were subjected to a rapid temperature drop to  $-54^\circ\text{C}$ , followed by a rapid temperature rise to  $70^\circ\text{C}$ , under conditions of maximum moisture saturation (95% RH). Specifically, each cycle has a duration of 240 minutes and follows the following steps:

1. Parking the aircraft on the ground in a cool environment: lower the chamber temperature to  $-54^\circ\text{C}$  at a rate of  $5^\circ\text{C}/\text{min}$  and keep these conditions for 45 minutes.
2. Lowering the chamber pressure to 120 mbar with a minimum temperature of  $-54^\circ\text{C}$ . This condition represents the take-off up to 15000 metres with a maximum ascent rate. Keep these conditions for 40 minutes.
3. Increase the pressure to 1013 mbar and raise the temperature to  $70^\circ\text{C}$  simulating landing in a warm environment and parking on the ground at  $70^\circ\text{C}$  for 25 minutes.
4. Increase relative humidity to 95% and remain in this condition ( $70^\circ\text{C}$  and RH 95%) for 60 minutes (park on the ground in a hot, humid environment).
5. Return chamber conditions to  $20^\circ\text{C}$  and 50% and remain for 10 minutes.

All panels were subjected to the GAG test. Before starting the test, all samples were dried in a vacuum oven at  $80^\circ\text{C}$  for 3 weeks and weighed. Further weighing was done after the first and tenth cycles.

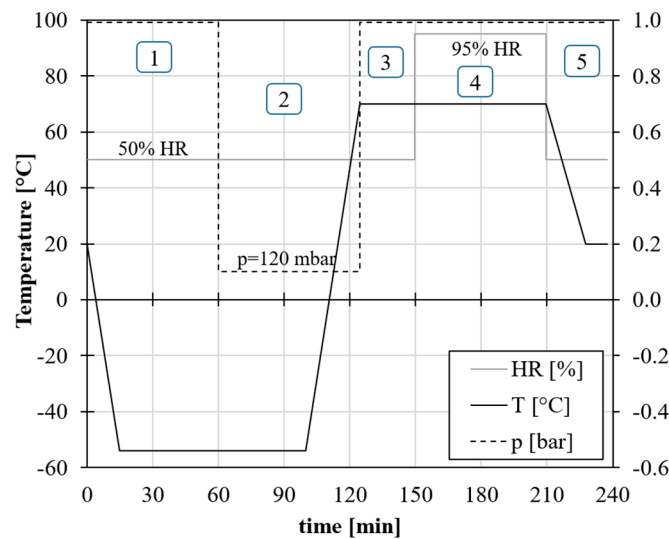


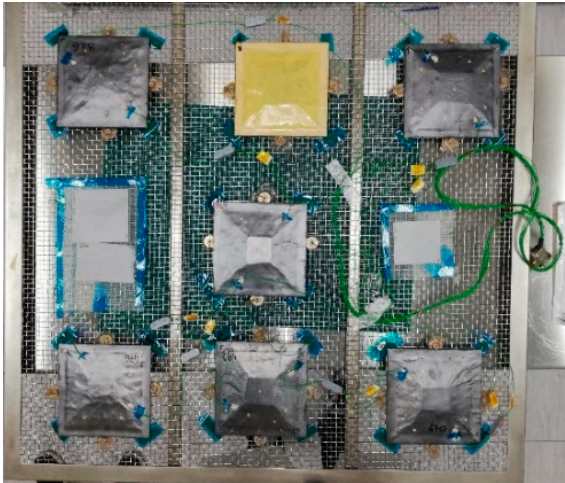
Figure 9. GAG test: temperature, humidity, and pressure profiles.

After ten cycles, the maximum absorption in panels coated with the graphene layer is sensibly reduced as well as the absorption rate. The barrier effect decreases with increasing graphene content: sample P70 shows the lowest value of absorption. Results of the GAG test conducted on panels are reported in Table 7 in terms of moisture uptake.

Table 7. Moisture uptake during the GAG test for each panel.

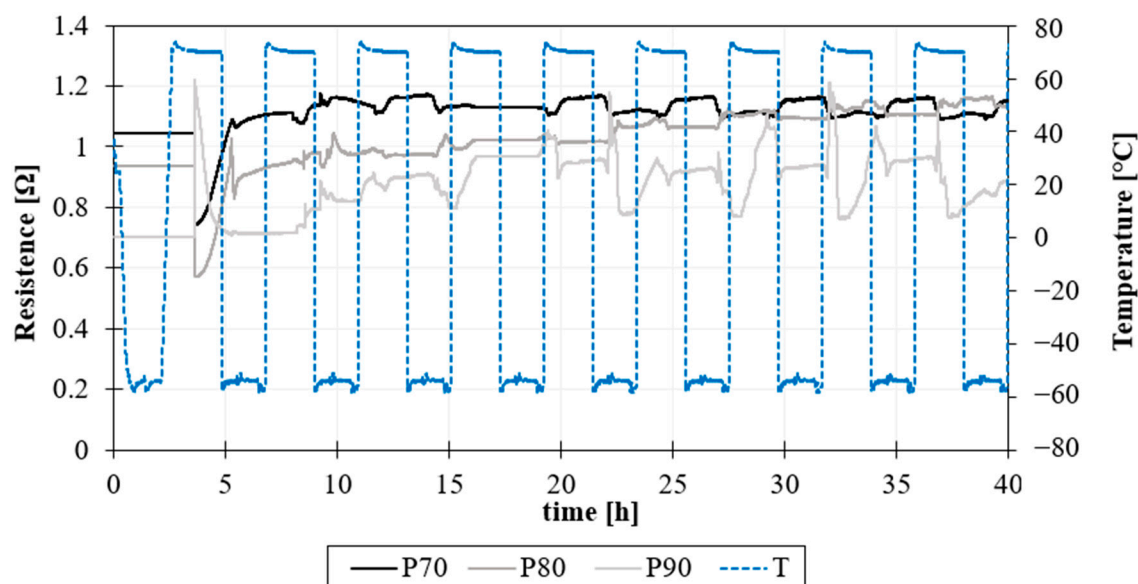
Panel ID	M @ Cycle 1	M @ Cycle 10
	[%]	[%]
P REF	0.59	1.81
P70	0.07	0.28
P80	0.18	0.64
P90	0.17	0.68

During the GAG test, the integrity of the graphene layer on the surface was monitored by measuring the electrical conductivity of the layer. Two electric wires were fastened with a conductive paste on the diagonal of each specimen and the different specimens were connected in series, as shown in Figure 10. The samples were fed with a 100 mA current, and voltage was recorded across each sample.

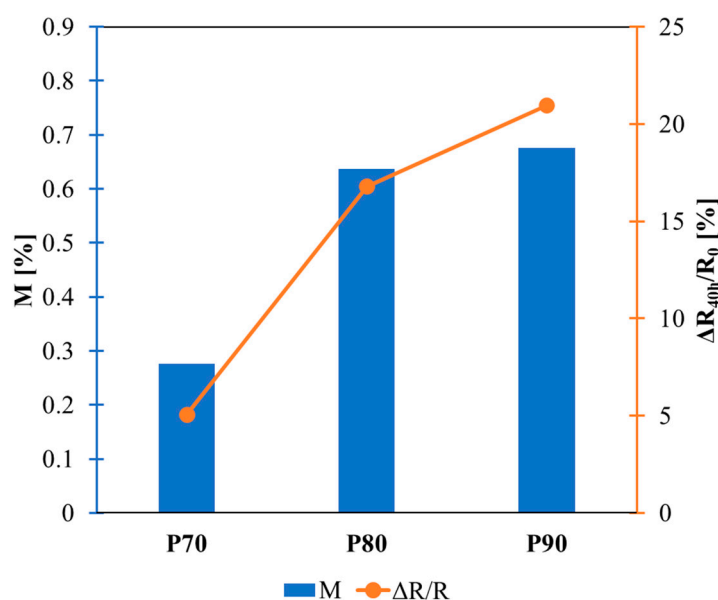


**Figure 10.** Kevlar sandwich panel samples: positioning in the chamber and set up for electrical conductivity measurements.

The results of the surface resistance measurements of specimens during the GAG test, until the 10th cycle, are reported in Figure 11. The resistance slightly increases during the test for all samples due to water absorption, as reported in Figure 12.



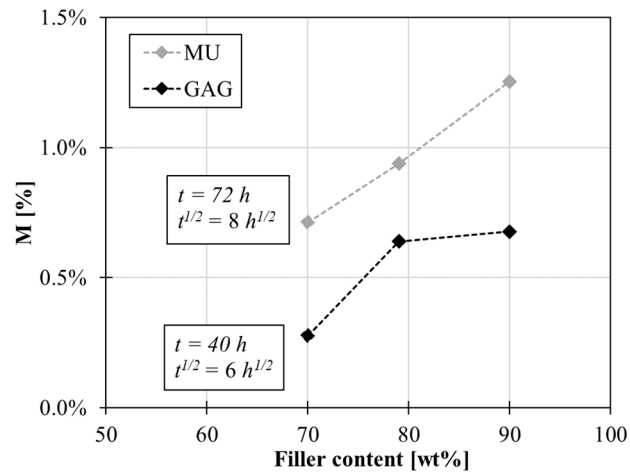
**Figure 11.** Electrical resistance vs. time for each panel during the 10 cycles (40 h) of the GAG test.



**Figure 12.** Trend of moisture uptake and variation of resistance at the end of the GAG test.

By comparing the results of the moisture uptake and GAG test, similar behaviours with graphene coating composition are found, as shown in Figure 13. In the case of the MU test after 72h the absorption sensible reduces in the panels coated with graphene layers compared to the uncoated panel by -71% for P70, -61% for P80 and -41% @90wt% for P90. Likewise in the case of the GAG test after 40h the panel absorption reduces by -85% for P70, -65% for P80 and -63% for P90 compared to P REF.

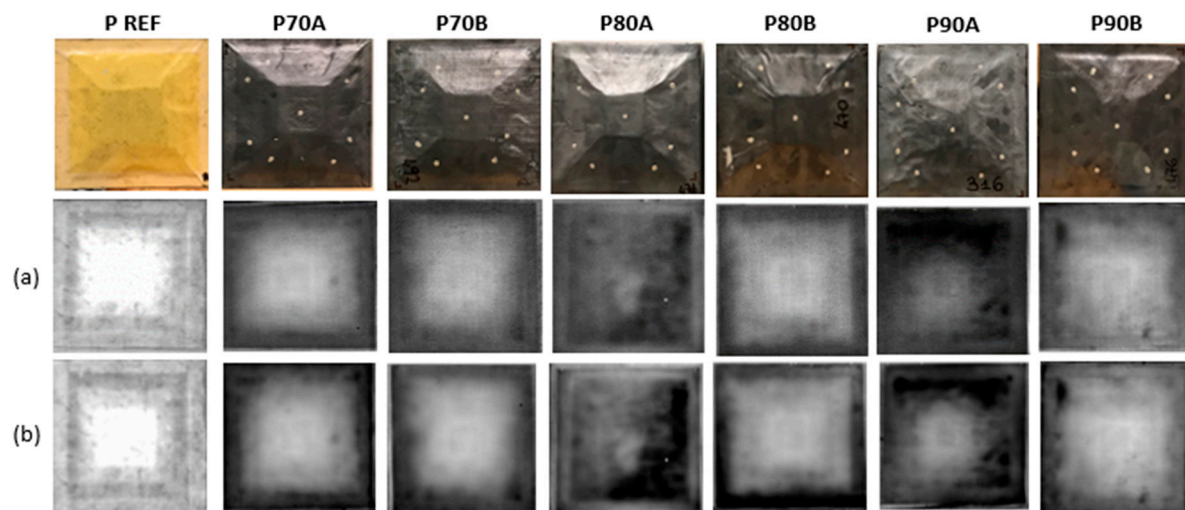




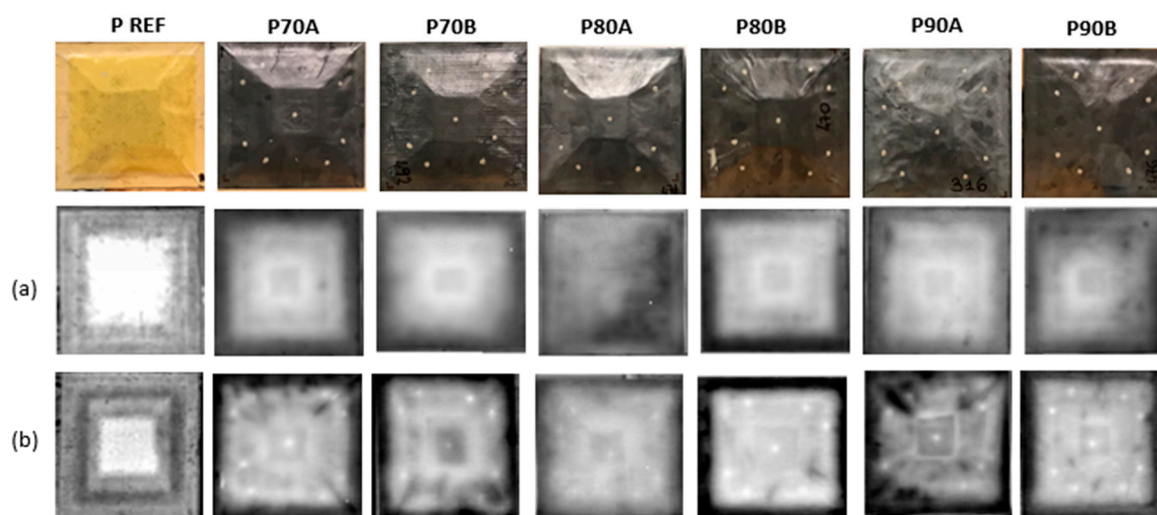
**Figure 13.** Comparison between MU and GAG tests.

#### 4.2. Non-destructive inspection of sandwich plates

Non-destructive inspections on specimens before and after the moisture uptake test revealed that the panels did not experience damage. Figures 14 and 15 show the results of the NDI analysis conducted on the lower surface of the samples at different frequencies of 0.1 Hz and 0.04 Hz. By decreasing the frequency, different depths of the samples are investigated, at 0.04 Hz it is even possible to observe the taper of the inner core. By comparing the images of samples before and after the test, for both sampling frequencies, it appears that the induced stress of the moisture uptake test does not produce any damage. Analogous results are obtained from the NDI after the GAG test. The induced stress of flight cycle simulations did not cause internal damage.



**Figure 14.** Thermal images: (a) before performing the test; (b) after the moisture uptake test at 0.1 Hz.



**Figure 15.** Thermal images: (a) before performing the test; (b) after the moisture uptake test at 0.04 Hz.

## 5. Conclusions

The effectiveness of protection against moisture uptake of graphene-rich films has been investigated through a moisture uptake test in an environmental chamber. The behaviour of Kevlar sandwich panels, representative of a regional aircraft composite part, in a humid environment, has been investigated, and the ability of graphene films to protect Kevlar/Epoxy panels from water absorption has been demonstrated.

After 40 days in a humid environment, the maximum sorption is reduced by  $\sim 30\%$  in samples protected with films @70 wt% of GNPs. Also, an evident delay in the absorption rate into the composite panels due to the presence of the protective layer has been observed. Similarly, the GAG test showed that after 10 cycles, the maximum sorption of the panel coated with films @70 wt% of GNPs is about 70% lower than that of the uncoated one. NDI analysis with lock-in thermography revealed the absence of damage in all panels after the test.

Thanks to the nanoplatelet alignment and the intrinsic hydrophobicity of GNPs, the film acts as a barrier to moisture. The effect of graphene film architecture has been investigated, showing that the nanocomposite inner architecture plays an important role in the barrier properties. Although with the increasing filler content the tortuosity factor of nanolamellar nanocomposites increases, the nanocomposite nanostructure is affected by dry areas, which promotes moisture absorption. In addition, a more hydrophilic behaviour is found for nanocomposites with 90 wt% compared to those with 70 wt%.

Experimental findings suggested that graphene films @70 wt% of GNPs represent a promising technology to be employed as lightweight barrier coating in the aeronautics field, thanks to the significant reduction of the moisture's permeation inside a composite panel and the easy integration in the manufacturing processes. In addition, thanks to the low electrical resistivity, the rich graphene films provide additional functionality, extending their application to shielding against electrostatic discharges and electromagnetic waves.

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