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Article

Luminescence of ZrO₂: Ti ceramics irradiated with high-energy xenon ions

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Abstract: Samples of ZrO₂ ceramics with different concentrations of impurity titanium ions were synthesized by mixing powders of zirconium and titanium oxides in various mass ratios. The phase composition and surface morphology of the studied ceramics were determined by X-ray phase analysis and scanning electron microscopy. It was found that irradiation of samples with xenon ions (220 MeV) with fluences 10¹⁰ and 10¹² ion/cm² leads to a decrease in the intensity of the pulse cathodoluminescence band at 2.5 eV. It is shown that ion irradiation causes the appearance of a new peak of thermoluminescence at 450–650 K of a non-elementary form associated with radiation-induced traps of charge carriers. In contrast to electron irradiation, an increase in the fluence of ions leads to a decrease in the intensity of this peak. In contrast to electron irradiation, an increase in the fluence of ions leads to a decrease in the intensity of this peak. A complex nonmonotonic dependence of the cathode and thermoluminescence intensity on the dopant concentration was found, which may be due to the effects of concentration quenching and aggregation of defective centers.

Keywords: zirconium dioxide; thermoluminescence; pulse cathodoluminescence; ceramics; scanning electron microscopy

1. Introduction

Zirconium dioxide (ZrO₂) has a high refractive index, thermal and chemical stability, which allows it to be used in many industries, in particular as heat-shielding coatings, electron-optical materials and ionizing radiation detectors [1]. There are different types of detectors, with luminescent detectors and thermoluminescent dosimeters being notable examples. Thermoluminescence dosimetry (TLD) involves the use of crystals or ceramics with specific lattice defects of impurity and/or intrinsic origin, known as crystal phosphors. These substances can emit light when exposed to ionizing radiation and subsequently heated. In this case, the amount of ionizing radiation absorbed is directly proportional to the number of emitted quanta [2]. TL-dosimetry is widely used for individual dosimetry of personnel, environmental monitoring, and medical exposure control. Thermoluminescent detectors use a variety of materials, such as alkaline-halides (LiF: Mg, Ti; LiF: Mg, Cu, P), sulphates (BaSO₄: Eu, CaSO₄: Dy), borates (MgB₄O₇: Dy), oxides (Al₂O₃, ZnO, ZrO₂, MgO), and complex compounds (K₃Na(SO₄)₂: Eu, Ba_{0.97}Ca_{0.03}SO₄: Eu, YAG etc.) [3–28].

It is known that the luminescent properties of nominally pure and doped zirconium dioxide ZrO₂ depend on the method of preparation, grain size, crystal structure and morphology of the samples [29–40]. Nominally pure ZrO₂ is characterized by its own luminescence at 2.5–2.7 eV (470–

490 nm) [29,30,35,38]. According to one point of view, the indicated luminescence band in ZrO_2 is due to the relaxation of centers associated with intrinsic defects in the anionic sublattice (oxygen vacancies) [30,31,35]. An alternative point of view connects the origin of the luminescence band at 2.6 eV with the relaxation of impurity ions of titanium [32,34,36].

It is known that the introduction of titanium as a dopant can significantly change the luminescent properties of zirconium dioxide. Thus, doping with titanium (0.5 mol%) of nanostructured ZrO_2 samples led to an increase in photoluminescence (PL) at 480 nm, excited by UV radiation [36]. A similar pattern with increasing dopant amount for the above PL band was observed by other authors. In this case, the maximum intensity was achieved at a titanium concentration of 0.15 weight % [32]. At higher concentrations, a drop in PL was recorded. A similar increase in thermoluminescence (TL) intensity was also discovered when ZrO_2 was doped with titanium (0.1 mol%) [41,42].

The study of the role of titanium impurities in the formation of the luminescent properties of zirconium dioxide irradiated with high-energy ions, in particular, heavy xenon ions, is of particular interest. It is known that under such irradiation new radiation-induced defects and their complexes are formed in oxide dielectrics [43,44]. These complexes can include both oxygen vacancies and impurity ions in different charge states. It was shown in [45] that complex defects containing oxygen vacancies and titanium ions may be responsible for the formation of the luminescence band at 2.6 eV. In this work, these defects were formed by high-temperature annealing of nominally pure nanostructured zirconium dioxide compacts in vacuum under reducing conditions provided by the presence of carbon in the form of graphite. It can be assumed that such vacancy-impurity complexes can also be generated in samples pre-doped with titanium as a result of high-intensity ion irradiation. In this case, it is of interest to establish the patterns of influence of the concentration of the introduced titanium impurity on the luminescence intensity of samples subjected to such irradiation. In practical terms, such information will be useful for the development of luminescent detectors of ion radiation.

The purpose of this work is to synthesize samples of ZrO_2 ceramics with different dopant concentrations and evaluate the effect of titanium impurities on the luminescent properties of samples irradiated with xenon ions.

2. Materials and Methods

The studied samples were obtained by mixing zirconium and titanium dioxides in the following ratios ($\text{ZrO}_2\text{:TiO}_2$): 99.9:0.1; 99:1; 95:5; 90:10 and 85:15 wt.%. A 99.1% purity zirconium dioxide nanopowder with a particle size in the range of 30-70 nm was used as the starting materials obtained by plasma chemical method (Plasmotherm Company, Russia), and titanium dioxide powder (purity 99.8%, Component Reagent company, Russia). Mixing of the powders was carried out using the dry method in an agate mortar with a binder (ethyl alcohol). The mixing time was 1-2 hours for a quantity of powder weighing 2 g for better homogenization to a dry consistency. Samples in the form of disks were obtained from a mixture of powders (100 mg for each compound) by cold uniaxial pressing at a pressure of 500 MPa. The diameter of the resulting compacts was 6 mm, and the thickness was 1-1.5 mm. Compressed pellets were annealed in air in a high-temperature furnace Linn High Therm HT-1800-M at a temperature of 1200 °C for 1 hour. The selected mode made it possible to ensure the mechanical strength of the samples. Also, in this case, the formation of the ZrTiO_4 phase did not occur, in the presence of which the luminescence intensity sharply decreased. In this way, samples of $\text{ZrO}_2\text{:Ti}$ ceramics with different dopant concentrations were obtained.

The luminescent properties of ceramics were studied using pulsed cathodoluminescence (PCL) and TL methods. To excite TL, the studied samples were irradiated with xenon ions with an energy of 220 MeV (fluences of 10^{10} and 10^{12} ions/cm²) at the DS-60 heavy ion accelerator (Astana, Republic of Kazakhstan). To carry out a comparative analysis of luminescent properties, the samples were also irradiated with a pulsed electron beam (60 A/cm², 2 ns) from the RADAN EXPERT accelerator with an electron energy of 130 keV at room temperature. The electron irradiation dose was 1.5 kGy per pulse. This electronic radiation, unlike ion radiation, does not lead to the formation of new radiation-induced defects in the material under study, but only changes the charge state of existing ones. The electron beam described above was also used to excite PCL. TL was measured in linear heating mode at a rate of 2 K/s. To register TL, a FEU-130 with a maximum spectral sensitivity of 400-420 nm was used.

3. Results

The phase composition and crystal structure of the ceramic samples were determined by X-ray diffraction (XRD) on a Rigaku MiniFlex 600 diffractometer using Cu K α -radiation (wavelength 1.5406 Å, scanning speed 0.3 O/min). Phase identification was conducted using an ICDD PDF-2 database with further Rietveld refinement analysis via the FullProf software.

The resulting diffraction patterns are shown in Figure 1. It can be seen that with the selected synthesis parameters in ceramics with the amount of introduced titanium dioxide of 10% or less, only two phases are observed: baddeleyite (monoclinic ZrO₂) and tazheranite ((Zr, Ti, Ca) O_{2-x}). In this case, the baddeleyite phase predominated for all samples, and its amount was more than 90%. The figure shows the most characteristic reflexes for this phase. In samples with an amount of introduced titanium dioxide of 15%, the rutile phase (TiO₂) was also recorded in a quantitative content of less than 5%. The inset in Figure 1 shows the most intense diffraction peak belonging to rutile. The crystal lattice parameters of the above phases are given in Table 1.

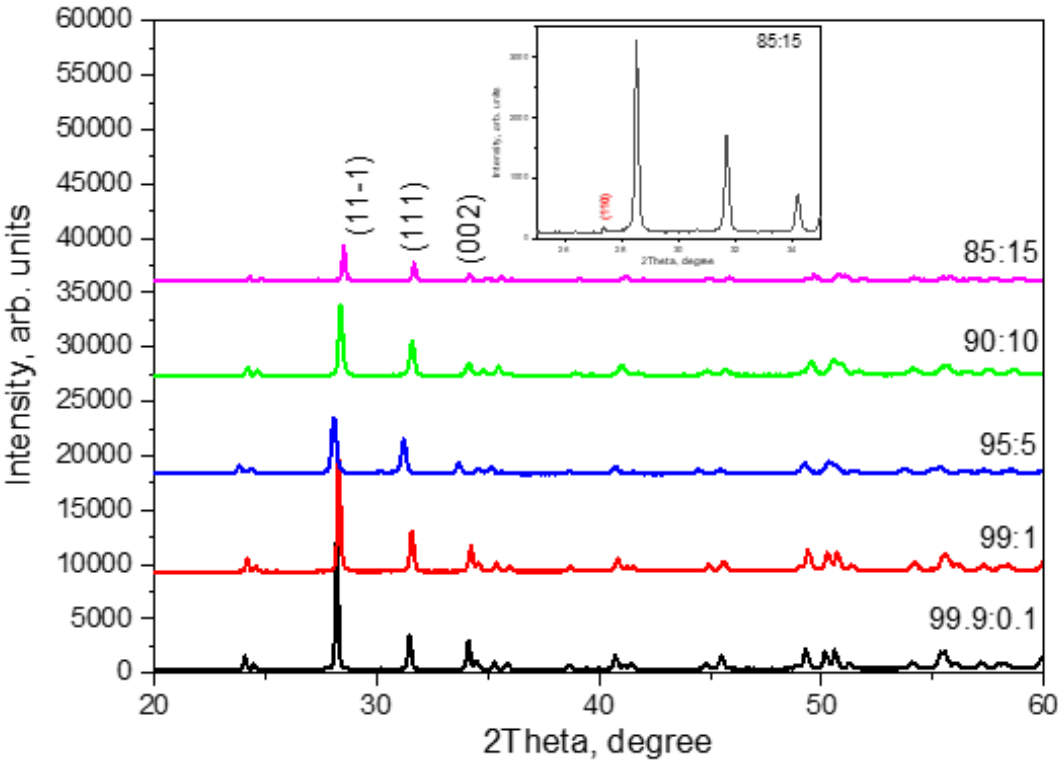


Figure 1. X-ray diffraction patterns of synthesized ceramics with different dopant concentrations.

Table 1. Parameters of crystal lattices of various phases in the samples under study.

Phase name	Singonia	a, Å	b, Å	c, Å	α , °	β , °	γ , °
Baddeleyite (ZrO ₂)	Monoclinic	5.1453	5.2002	5.3236	90.000	99.094	90.000
Tazheranit (Zr,Ti,Ca)O _{2-x}	Cubic	5.092	5.092	5.092	90.000	90.000	90.000
Rutile (TiO ₂)	Tetragonal	4.6063	4.6063	2.9773	90.000	90.000	90.000

Using a scanning electron microscope, images of the surface of the studied ceramics with different contents of introduced titanium dioxide were obtained (Figure 2). It can be seen that the resulting ceramics have a small number of pores and grains of micron and submicron size. From the analysis of SEM images, we can conclude that all samples are characterized by two groups of grain sizes (0.2-0.5 μ m and 1-2 μ m). The sample with a mass fraction of introduced titanium dioxide of 5% has the largest grains (more than 1 μ m), and the sample with the largest amount of introduced titanium dioxide (15%) has the densest packing among all the ceramics studied.

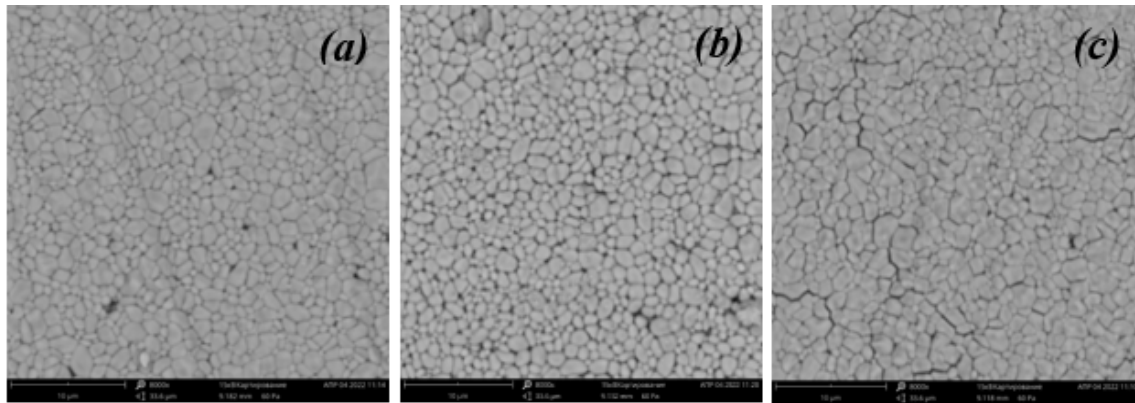
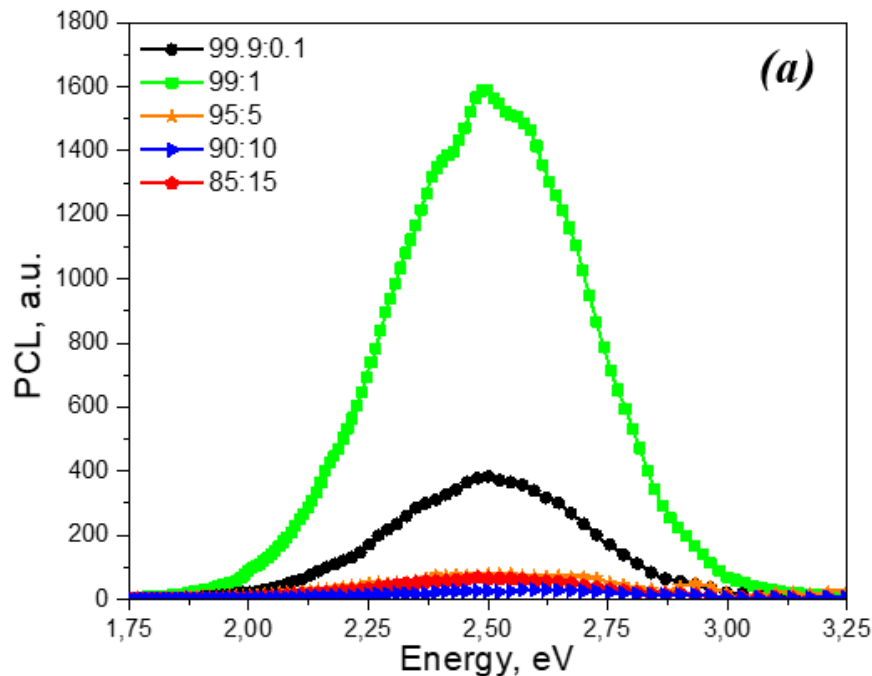


Figure 2. SEM images of synthesized ZrO_2 : Ti ceramics with different percentages of components introduced during synthesis: (a) ZrO_2 : TiO_2 (85:15), (b) ZrO_2 : TiO_2 (90:10) and (c) ZrO_2 : TiO_2 (95:5).

Figure 3 shows the PCL spectra of unirradiated ZrO_2 : Ti ceramics with different dopant concentrations, as well as samples irradiated with xenon ions with fluences of 10^{10} and 10^{12} ions/ cm^2 . It can be seen that all spectra contain one broad band with a maximum at 2.5 eV. In this case, irradiation with xenon ions does not lead to the appearance of new PCL bands. At the same time, in samples irradiated with ions, a decrease in PCL intensity is observed compared to unirradiated samples with the same dopant concentration. The dependence of the maximum luminescence intensity in the 2.5 eV band on the titanium concentration is non-monotonic. The highest PCL intensity in this band is characterized by samples with a concentration of introduced titanium dioxide of 1%. For ceramics with a higher content of introduced titanium dioxide (5 and 10%), this band is characterized by extremely low intensity. At a titanium dioxide concentration of 15%, a slight increase in PCL is observed at 2.5 eV.



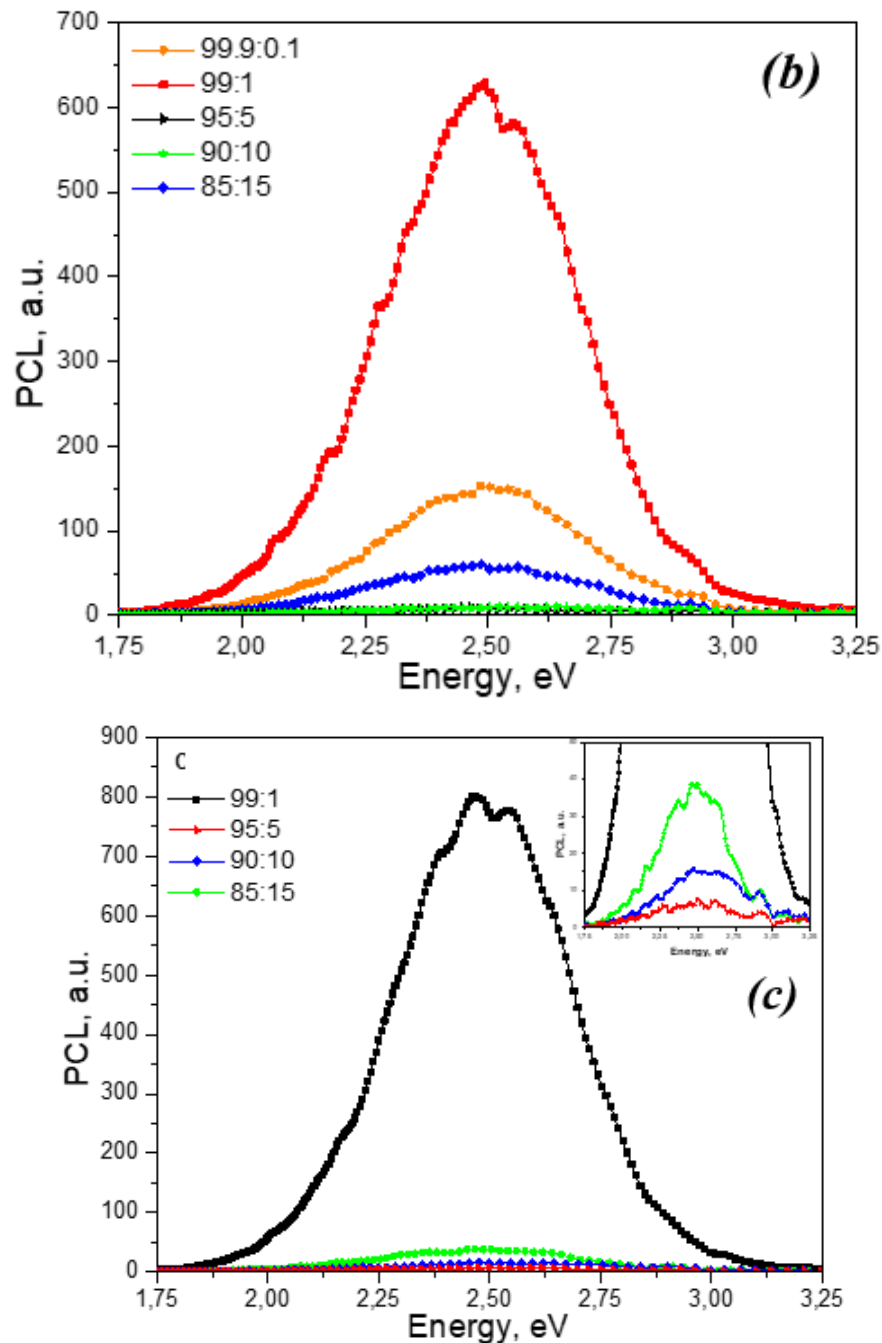


Figure 3. PCL spectra of various samples of unirradiated ZrO₂: Ti ceramics (a) and ceramics irradiated with xenon ions with a fluence of 10¹⁰ ions/cm² (b) and 10¹² ions/cm² (c).

Figure 4 shows the TL curves of ZrO₂: Ti ceramics corresponding to concentrations of introduced titanium dioxide of 5 and 10%, irradiated with various sources: xenon ions with fluences of 10¹⁰ and 10¹² ions/cm² and a pulsed electron beam with doses of 3 and 15 kGy. For samples irradiated with electrons, an intense TL signal is observed at 350-450 K. Moreover, with increasing irradiation dose (from 3 to 15 kGy), its intensity increases.

The TL curves of the same samples irradiated with xenon ions differ significantly from the curves of electron-irradiated samples. Low-temperature TL at 350-450 K is characterized by extremely low intensity, and a new high-temperature TL signal appears at 450-650 K. This signal may be associated with the appearance of new radiation-induced defects as a result of ion irradiation, which play the role of charge carrier traps. The possibility of creating such defects was proven in [46] based on the results of studying EPR spectra. From Figure 4 it is clear that the TL curve at 450-650 K has a non-elementary shape and is a superposition of several TL peaks. This is also evidenced by the

change in the temperature of the TL maximum with variations in the dopant concentration and ion fluence. In this case, with increasing fluence of ion irradiation, a drop in TL intensity is observed at 450-650 K. This pattern was observed for all samples with different concentrations of introduced TiO₂ (from 0.1 to 15%). In this case, the decrease in TL intensity with increasing ion fluence may be associated with radiation destruction of the luminescence centers responsible for TL and the formation of more complex defects. A similar effect of a decrease in the TL yield with increasing ion energy density was observed, in particular, in aluminum oxide single crystals irradiated with pulsed C⁺/H⁺ ions [47].

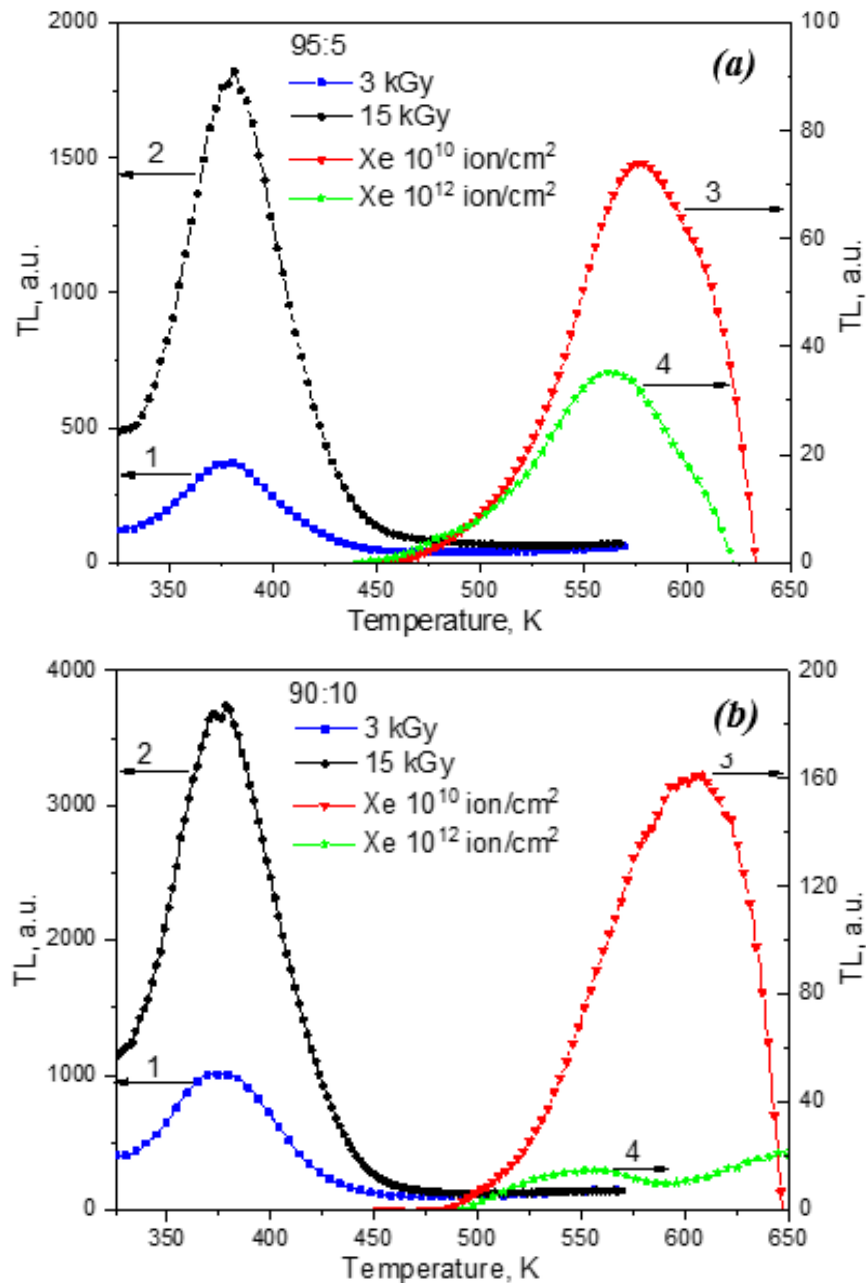


Figure 4. TL curves of samples with a concentration of introduced titanium dioxide of 5% (a) and 10% (b), irradiated with electrons at a dose of 3 kGy (1) and 15 kGy (2) and xenon ions with a fluence of 10^{10} ion/cm² (3) and 10^{12} ion/cm² (4).

Figure 5 shows the TL curves of samples irradiated with ions with different dopant concentrations. It can be seen that the dependence of the TL intensity at 450-650 K on the titanium concentration is, as in the case of PCL, non-monotonic. The maximum intensity is observed for a concentration of introduced titanium dioxide of 1%, which coincides with the results of studying the

PCL spectra. A further increase in the concentration of the dopant causes a decrease in the TL intensity. At a concentration of introduced titanium dioxide of 15%, an increase in the TL signal is observed, as in the case of PCL (Figure 3).

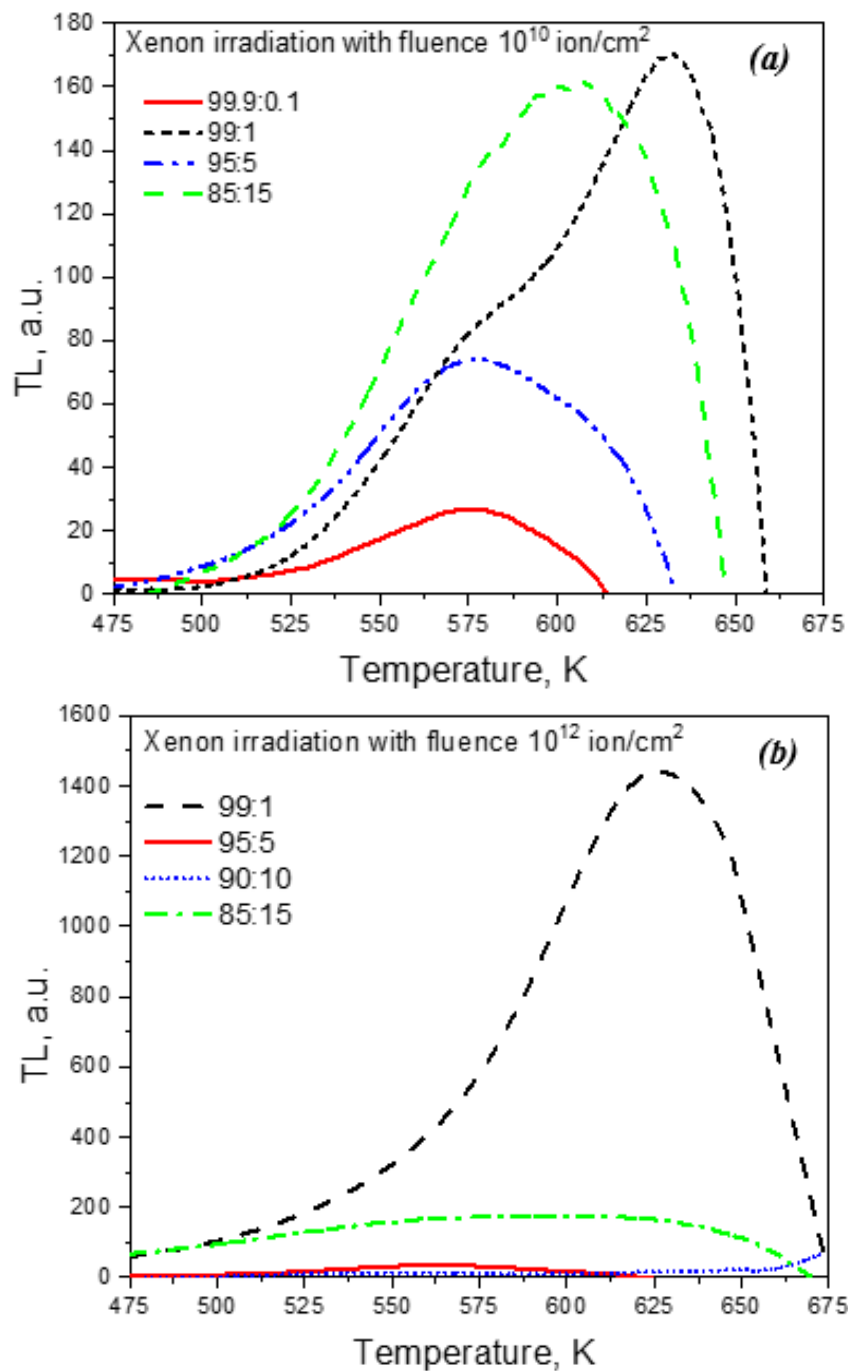


Figure 5. TL curves of samples with different dopant concentrations irradiated with xenon ions with a fluence of 10^{10} ions/cm² (a) and 10^{12} ions/cm² (b).

A nonmonotonic dependence of the PL and PCL intensity on the the concentration of titanium for monoclinic ZrO₂ samples not subjected to any irradiation was observed in [45]. Moreover, its drop at a high dopant content was associated with concentration quenching of luminescence. The increase in PCL and TL observed by us at the maximum concentrations of introduced titanium dioxide (15%) may be associated with the formation of complex cluster defects, which are responsible for the luminescence band at 2.5 eV. In [45], it was suggested that the nature of this band in monoclinic ZrO₂ is related to complex vacancy-impurity complexes containing, in particular, titanium ions. A similar relationship may also occur in the samples studied in this work.

4. Conclusions

In this work, zirconium dioxide ceramics with different concentrations of titanium impurity ions were synthesized. It was found that irradiation of samples with high-energy xenon ions leads to a decrease in the PCL intensity at 2.5 eV. In this case, no new emission bands are observed in the PCL spectra. It has been shown that ion irradiation leads to the appearance of a new TL signal at 450-650 K of a non-elementary form, presumably associated with radiation-induced defects that trap charge carriers. Unlike electron irradiation, an increase in ion fluence leads to a decrease in TL intensity. A complex non-monotonic dependence of the PCL and TL intensity on the dopant concentration was discovered, which may be due to the effects of concentration quenching and aggregation of defect centers.

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Conflicts of Interest: The authors declare no conflict of interest.

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