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# Indigo Carmine: Between Necessity and Concern

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Abstract: Dyes, as indigo carmine have become indispensable to modern life being widely used in food, textile, pharmaceutical, medicine and cosmetic industries. Although indigo carmine is considered toxic and presents many adverse effects, it is heavily used in the food industry because the blue pigment is difficult to obtain from natural sources and is one of the most used dyes in the textile industry, especially for dyeing denim. Also, indigo carmine is one of the dyes used in medicine as diagnostic agent because it has impressive applicability in terms of diagnostic methods and surgical procedures. In the literature it is reported that indigo carmine is toxic for humans and can cause various pathologies, such as hypertension, hypotension, skin irritations, corneal and conjunctival disorders or gastrointestinal disorders. In this review, we discuss the structure and properties of indigo carmine, its use in various industries and medicine, the adverse effects of its ingestion, injection or skin contact, the effects on environmental pollution and its toxicity testing.

Keywords: indigo carmine; food dye; textile dye; diagnostic agent; toxicity

#### 1. Introduction

Dyes have been intensively used in various fields, such as food, textile, pharmaceutical, medicine, cosmetic, plastic, rubber and paper industries. These industries produce impressive amounts of wastewater containing carcinogenic and toxic dyes that end up in the environment causing water and soil pollution [1–6]. Thus, dyes represent a threat to the environment and the health of the entire population [2]. Globally, approximately  $7 \times 108-1 \times 109$  kg/year of dyes are produced [7] and among these industries, the largest quantities of dyes are used in the textile industry [8]

Although indigo carmine is considered toxic and presents many adverse effects, it is part of the group of synthetic dyes intensively used in many fields, such as the food, textile, pharmaceutical, medicine and cosmetic industries [9–13].

In recent years, there are many articles about indigo carmine in the literature. Most of them are about removing indigo carmine from wastewater, testing its toxicity, and its uses in medicine. Even if indigo carmine is a widely used dye, there are no reviews in the literature summarizing all its applications, the adverse effects on humans and the environment and its toxicity.

#### 2. Materials and Methods

To the best of our knowledge, no reviews on the uses and potential impact of exposure to indigo carmine have been published. An in-depth search of articles related to indigo carmine was performed using PubMed, Scopus and Google Scholar databases. We used the following keywords: "indigo carmine" AND "Food," "indigo carmine," AND "textile," "indigo carmine" AND "medicine", "indigo carmine" AND "pollution," "indigo carmine" AND "toxicity."

A total of 137 studies were considered relevant by searching abstracts and titles. Of these, 104 studies were original articles, 21 were review articles and 6 were book chapters. We included in our review 17 articles that mentioned indigo carmine as food additive, 5 articles about indigo carmine in textile industry, 48 articles about environmental pollution with indigo carmine (including methods developed for the removal of indigo carmine from wastewater), 42 articles about uses of indigo carmine in medicine and 13 toxicology studies (9 in vivo toxicology studies and 4 in vitro toxicology studies).

Additional data sources (EFSA/European Food Safety Authority, EINECS/European Inventory of Existing Commercial Chemical Substances; FAO/Food and Agriculture Organization of the United Nations, General Standard Food Additives, IUPAC/International Union of Pure and Applied Chemistry and JECFA/Joint FAO/WHO Expert Committee on Food Additives) were also searched.

## 3. The structure and properties of indigo carmine

Indigo carmine is a dye also known as indigotine, E132, CI natural blue II, FD&C Blue No.2, Acid blue 74, Brilliant blue, CI (1975) No. 73015 [14–20]. According to the IUPAC (International Union of Pure and Applied Chemistry), indigo carmine is also called 3,3'-dioxo-2,2'-bis-indolyden-5,5-disulfonic acid disodium salt [14,17,21].

Indigo carmine is a dark blue powder with molecular formula C<sub>16</sub>H<sub>8</sub>N<sub>2</sub>Na<sub>2</sub>O<sub>8</sub>S<sub>2</sub> and a molecular weight of 466.367 [14–18,20,22]. Its chemical structure presents two essential groups, NaSO<sub>3</sub> and a chromophore group. NaSO<sub>3</sub> provides the property of the dye to dissolve in water, and the chromophore group gives the compound its characteristic color. The chromophore group is a conjugate system of a C=C bond replaced by two C=O groups and two NH groups [18,23].

According to European Commission Regulation 231/2012, Joint FAO/WHO Expert Committee on Food Additives and U.S. Food and Drug Administration, indigo carmine consists of a mixture of disodium 3,3'-dioxo-2,2'bi-indo-lydidene-5,5'-disulphonate and disodium 3,3'-dioxo-2,2'-bi-indolylidene-5,7'-disulphonate, as well as auxiliary coloring substances and colorless components represented by sodium chloride and/or sodium sulfate. Thus, indigo carmine is often described as a sodium salt, but also as calcium and potassium salt [16,19]

Indigo carmine must contain at least 85% total dyes calculated as the sodium salt, and disodium 3,3'-dioxo-2,2'-bi-indolylidene-5,7'-disulphonate not more that 18%. It also contains 0.2% water-insoluble substances, 1% other auxiliary coloring substances, 0.5% organic compounds, 0.01% unsulfonated primary aromatic amines and 0.2% neutral ether-extractable substances [16,19].

Indigo carmine has low pH stability, so after a week it loses its intensity if kept at pH 3-5, it fades considerably at pH 7 and completely at pH 8. Also, it has very poor stability to light and oxidation, moderate stability to heat, and is the least soluble of the dyes, with a solubility of 1.6 g in 100 ml of water at 25°C and slightly soluble in ethanol. Indigo carmine appears blue in solution at neutral pH, blue-violet at acidic pH, and green to yellow-green at basic pH. If it is dissolved in concentrated sulfuric acid it shows a blue-violet color which turns blue when diluted with water [16,24,25]. Table 1 shows the properties of indigo carmine.

**Table 1.** Properties of indigo carmine dye.

	Properties of indigo carmine dye	Authors	
Molecular formula	$C_{16}H_8N_2Na_2O_8S_2$	17.10.27	
IUPAC name	3,3'-dioxo-2,2'-bis indolyden-5,5-disulfonic acid disodium salt	17,18,26	
EINECS	212-728-8	16	
Consistency	powder	17 10 27 27	
Color	dark blue	17,18,26,27	
Color index number	73015	16	
Molecular weight	466,367 g/mol	14,17,28	
Melting point	>300°C		
Maximum absorption	608-612 nm	15	
Composition	Dye content~85%; <18% disodium 3,3'-dioxo-2,2'-bi-indolylidene-5,7'-disulphonate	17	
Reactive group	Amines, phosphines, pyridines, salts basic, ketones, hydrocarbons, aliphatic unsaturated		
Purity	Water insoluble matter <2%; Auxiliary coloring substances <1%; Organic compounds (isatin-5-sulfonic acid, 5-		
	sulfoanthranilic acid, anthranilic acid) < 1%; Unsulfonated primary aromatic amines < 0,01%; Ether extractable matter <0,2%;	16,29	
	Arsenic <3 mg/kg; Lead <2 mg/kg; Cadmium <1 mg/kg; Mercury <1 mg/kg		
Solubility	10 g/L	18	
Density	0.4 - 0.6	10	

IUPAC=International Union of Pure and Applied Chemistry; EINECS=European Inventory of Existing Commercial chemical Substances.

# 4. The manufacture of indigo carmine

Indigo is among the first known dyes and was originally obtained from the leaves of *Indigofera tinctoria*, *Indigofera suifruticosa* or *Isatis tinctoria*, where it occurs as indican that is an indoxyl glycoside. Enzyme indimulsin hydrolyzes the indican into indoxyl and glucose. By oxidizing indoxyl it is transformed into indigo. Since indigo carmine is increasingly used since the 20th century, its extraction from plants has been replaced by its production through chemical synthesis. For instance, reactants such as N-phenylglycine or N-(-2carboxyphenyl)glycine are heated to produce indoxyl which undergoes oxidation processes to form indigo [16,30].

Indigo is produced synthetically by fusing N-phenylglycine in a molten mixture of sodamide and sodium potassium hydroxide. Indigo is isolated and subjected to purification procedures. Indigo carmine is obstained by sulfonation of indigo (2,2'-bindoline-3,3'-dione), which involves heating indigo in the presence of sulfuric acid, and is subsequently subjected to purification processes [16,26].

# 5. Indigo carmine in food industry

Food industry represents one of the main fields in which indigo carmine is intensively used, being considered a food additive [18,31]. Food additives are widely used in developed countries to improve the taste, color, aroma, texture and extend the shelf life of food or beverages [31–33]. In the food industry, synthetic dyes are preferred, because present numerous advantages, such as stability in contact with light, oxygen or another pH, provide uniform color to food, excellent solubility, low risk of contamination with microorganisms and low production costs [25,27,32,34,35]. Before the approval of a dye in the food industry, numerous toxicity tests are required, such as acute and chronic effects, carcinogenicity, mutagenicity, teratogenicity, toxicity on the reproductive system, the degree of accumulation in the body, but also the identification of the lethal dose 50 [36,37].

Indigo carmine is heavily used in the food industry because the blue pigment is difficult to obtain from natural sources, and the dyes that are obtained do not have same stability, coloring power and are not as affordable as synthetic dyes [25,35]. Thus, indigo carmine is part of group III of synthetic food dyes used for a wide range of food products [26,33].

The Codex Alimentarius Commission allows the use of indigo carmine in food and beverages with maximum permitted levels between 50 – 450 mg/kg, as mentioned in the General Standard Food Additives [38]. Also, the European Food Safety Authority has approved maximum levels for the use of indigo carmine in food and beverages ranging between 50 and 500 mg/kg [16]. U.S. Food and Drug Administration (USFDA) has not decided maximum levels for this food additive, but allows use under good manufacturing practice in a number of foods as established in the Code of Federal Regulations [39].

#### 5.1. Adverse effects of indigo carmine consumption

According to EFSA [16] and GSFA [38], indigo carmine is found in many food products (Table 2), although it presents a potential risk to human health [18,27]. The ingestion of indigo carmine can cause nausea, vomiting, diarrhea and skin irritation [18,32].

Table 2. Maximum permitted levels (MPL) of indigo carmine (E 132) in food according to the Annex II to Regulation (EC) No 1333/2008 and GSFA [16,38].

Food category no.	Food category	EFSA MPL mg/kg	GSFA MPL mg/kg
01.1.2	Dairy-based drinks, flavoured and/or fermented		300
01.4	Flavoured fermented milk products including heat-treated products	150	_
01.6.1	Unripened cheese	150	200
01.6.2.2	Rind of ripened cheese	-	100
01.6.3	Other creams	150	-
01.6.4.2	Flavoured processed cheese, including containing fruit, vegetables, meat, etc	-	100
01.6.5	Cheese analogues	-	200
01.7	Dairy-based desserts - pudding, fruit or flavoured yoghurt	-	150
01.7.3	Edible cheese rind	quantum satis	-
01.7.6	Cheese products	100	-
02.1.3	Lard, tallow, fish oil and other animal fats	-	300
02.3	Fat emulsions mainly of type oil-in-water, including mixed and/or flavoured products based on fat emulsions	-	300
02.4	Fat-based desserts excluding dairy-based dessert	-	150
03.0	Edible ices, including sherbet and sorbet	150	150
04.1.2.11	Fruit fillings for pastries	-	150
04.1.2.5	Jams, jellies, marmalades	-	300
04.1.2.6	Fruit-based spreads	-	300
04.1.2.7	Candied fruit	-	200
04.1.2.8	Fruit preparations, including pulp, purees, fruit toppings and coconut milk	-	150
04.1.2.9	Fruit-based desserts, including fruit-flavoured water-based desserts	-	150
04.2.2.3	Vegetables (mushrooms and fungi, roots and tubers, pulses and legumes, and aloe vera), and seaweeds in vinegar, oil, brine or soybean sauce	-	150
04.2.2.7	Fermented vegetable (mushrooms and fungi, roots and tubers, pulses and legumes, and aloe vera) and seaweed products	-	300
04.2.4.1	Fruit and vegetable preparations excluding compote	200	-
05.2	Confectionery including hard and soft candy, nougats	200-300	300
05.3	Chewing gum	300	300
05.4	Decorations for fine bakery wares, toppings (non-fruit) and sweet sauces	300-500	300
05.1.4	Cocoa and chocolate products	-	450
05.1.5	Imitation chocolate, chocolate substitute products	-	300

06.5	Cereal and starch based desserts (rice pudding, tapioca pudding)	-	150
06.6	Batters	500	-
07.2	Fine bakery wares (sweet, salty, savoury) and mixes	200	200
08.2.3	Casings and coatings and decorations for meat	500	-
09.1.1	Fresh fish	-	300
09.2	Processed fish and fishery products including molluscs and crustaceans	500	-
09.2.1	Frozen fish, fish fillets, and fish products, including molluscs, crustaceans and echinoderms	-	300
09.2.4.1	Cooked fish and fish products	-	300
09.2.4.2	Cooked molluscs, crustaceans and echinoderms	-	250
09.2.5	Smoked, dried, fermented, and/or salted fish and fish products, including molluscs, crustaceans and echinoderms	-	300
09.3	Fish roe	300	-
09.3.3	Salmon substitutes, caviar and other fish roe products	-	300
09.3.4	Semi-preserved fish and fish products, including molluscs, crustaceans and echinoderms, excluding products of food	_	300
09.3.4	categories 09.3.1 – 09.3.3	_	300
09.4	Fully preserved, including canned or fermented fish and fish products, including molluscs, crustaceans and echinoderms	-	300
10.1	Fresh eggs	-	300
10.4	Egg-based desserts (e.g., custard)	-	300
11.4	Other sugars and syrups (e.g., xylose, maple syrup, sugar toppings)	-	300
12.2.2	Seasonings and condiments	500	300
12.4	Mustards	300	300
12.5	Soups and broths	300	50
12.6	Sauces and like products	500	300
12.9	Protein products	100	-
13.2	Dietary foods for special medical purposes defined in Directive 1999/21/EC	50	-
13.3	Dietetic foods intended for special medical purposes (excluding products of food category 13.2)	50	50
13.4	Dietetic formulas for slimming purposes and weight reduction	-	50
13.5	Dietetic foods (e.g., supplementary foods for dietary use) excluding products of food categories 13.1-13.4 and 13.6	-	300
13.6	Food supplements	-	300
14.1.4	Water-based flavoured drinks, including sport, energy or electrolyte drinks and particulated drinks	100	100
14.2.2	Cider and perry	200	200
14.2.4	Wines (other than grape)	200	200
14.2.6	Distilled spirituous beverages containing more than 15% alcohol	200	300
14.2.7	Aromatized alcoholic beverages (e.g., beer, wine and spirituous cooler-type beverages, low alcoholic refreshers)	200	200
14.2.7.1	Aromatised wines	200	-

14.2.7.2	Aromatised wine-based drinks	200	-
14.2.7.3	Aromatised wine-product cocktails	200	-
14.2.8	Other alcoholic drinks including	200	-
15.1	Snacks – potato, cereal, flour or starch based (from roots and tubers, pulses and legumes)	100	200
15.2	Processed nuts, including coated nuts and nut mixtures (with e.g., dried fruit)	100	100
16	Desserts excluding products covered in categories 01, 03 and 04	150	-
17.1	Food supplements supplied in a solid form (capsules and tablets and similar forms, excluding chewable forms)	300	-
17.2	Food supplements supplied in a liquid form	100	-
17.3	Food supplements supplied in a syrup-type or chewable form	100-300	-

GFSA: General Standard on Food Additives; EFSA: European Food Safety; Authority; MPL: Maximum permitted level.

The toxicity of indigo carmine was first evaluated by JECFA (Joint FAO/WHO Expert Committee on Food Additives) who established a daily intake of 0-2.5 mg/kg body weight in 1969, increasing to 5 mg/kg body weight in 1975. EFSA stated, based on available research, that indigo carmine does not show developmental toxicity, genotoxicity or changes in hematological parameters in chronic toxicity tests [34,35]. Although the dye concentrations used are evaluated for consumer safety, their purity specifications allow certain concentrations of unsulfonated aromatic amines, which can reach 100 mg/kg of dye. If the average dye concentration is 500 mg/kg of food, the amine concentration can reach 50  $\mu$ g/kg of food. If an amount of 15 mg/kg body weight of dye is ingested daily, the level of aromatic amines may be 1.5  $\mu$ g/kg body weight. Thus, indigo carmine, but also other synthetic blue dyes, can also affect human health through the indirect consumption of amines that have genotoxic and carcinogenic effects [35].

The interaction of food dyes with metal ions can change their stability, toxicity and physicochemical properties of both the dyes and the metal ions. Indigo carmine forms stable complexes with various ions. The interaction between copper ions and indigo carmine in an alkaline medium led to the formation of the Cu2(IC) complex. Following spectrophotometric studies it was observed that the stoichiometric ratio between indigo carmine and copper is 2:1. This aspect is important because the interactions between food colorings and metal ions can change the stability, toxicity, but also other physico-chemical properties of both toxic compounds [40].

Indigo carmine is generally unstable in the presence of oxidizing or reducing agents, sugars, and salts because they alter the unsaturated conjugate group that gives it its characteristic color. Therefore, the recommended shelf life of indigo carmine in the food industry is 4 to 6 years [16].

There are numerous analytical methods used for the identification and determination of indigo carmine in food. The most commonly used are thin layer chromatography, high performance liquid chromatography (HPLC) [41–44], ion chromatography, spectroscopy [31], voltammetry, differential pulse polarography and capillary electrophoresis [43,45–48].

#### 6. Indigo carmine in textile industry

Dyes have been used in textile industry for more than 4000 years, but until the 19th century they were obtained from natural sources, mostly plants, insects and minerals [49]. Originally, indigo was obtained from the plants *Indigofera tinctoria* and *Isatis tintoria* which were used to dye textiles in India, China and Egypt [30,50]. Thus, the dyes were less toxic, less polluting and produced fewer allergies compared to synthetic ones [49].

Indigo carmine has been identified in textile objects dating back to the 18th century. It was mainly used for dyeing silk, wool and leather products [51], but also for dyeing polyester and denim fibers [2,52].

At the end of 19th century, with the development of chemical synthesis techniques, indigo became one of the most important and profitable products in the chemical industry [30]. In the 20th century, synthetic indigo still represent one of the most important products of the textile industry, almost completely replacing plant-derived indigo [30,53].

Among all dyes, indigo carmine remains one of the most used dyes in the textile industry [9,50,54,55] especially for dyeing denim. For instance, between 3 – 12 g of dye are needed to color a pair of jeans [30,50,52]. Thus, in the last decades its production has increased, reaching up to 50,000 tons/years, being necessary to obtain it through chemical synthesis [55].

It has been estimated that approximately 100,000 types of synthetic and natural dyes are produced globally, which represents between  $7 \times 108-1 \times 109$  kg/year, but a significant amount is lost during the manufacturing processes [7].

### 6.1. Indigo carmine and water pollution

The textile industry is considered a major source of pollution due to the discharge of non-biodegradable, acidic pH and strongly oxidizable contaminants into natural water sources [17,50,52,56–60]. Dyes, including indigo carmine, cause extensive environmental pollution and one

of the consequences of water contamination with dyes is the impairment of photosynthesis [50,56,57]. In addition, indigo carmine often ends up in waters it contaminates, posing a threat to aquatic organisms and public health [25,50,57,60–63]. Also, it presence in water changes the color and smell of water, even at very low concentrations [64]. Therefore, several methods have been tried to remove indigo carmine from the aquatic environment, such as adsorption [5,11,52,59,63,65,66], ultrafiltration [67,68], nanofiltration [69,70], reverse osmosis [69], electrocoagulation [60], electrodialysis [17], oxidation [64], micellar catalysis [71], photocatalytic degradation [72], photolysis [73], bacterial treatment [74], etc.

Adsorption is considered the most advantageous due to ease of operation, high efficiency in a wide pH range, high performance and the possibility of reusing the adsorbent by regeneration [63,75]. Adsorption is a physical or chemical phenomenon, in which the pollutant is attached to the surface of the adsorbent material which can be activated carbon [76–78], various organic or inorganic matrices [71], magnetic composites [59], geopolymers [79], chitosan [80],  $\beta$ -cyclodextrin, chitinchitosan [63], synthesized cationic hydrogels [52], ionic/nonionic polystyrene adsorbents [81], magnesium oxide, magnesium hydroxide, calcium oxide, calcium hydroxide [82], etc.

The adsorption capacity of indigo carmine also depends on the concentration of the dye and the pH [52], and the adsorbent materials must show high adsorption capacity, mechanical and thermal stability [63].

Relatively recently, certain wastes have begun to be used as adsorbent material to remove various dye from waters and have been found to be very effective in removing indigo carmine from the aquatic environment. The most used wastes for water decontamination are sunflower stalks, corn cobs, sugarcane stalks, rice husks, wheat husks, clay and paper [59,83–86].

Electrocoagulation is an electrochemical treatment process that uses electrodissolvable metal electrodes to clean polluted water with metal hydroxides and can remove even the smallest dye particles. Through this process, 82.55% of the dye was removed from a solution containing 20.01 mg/l indigo carmine, thus being an effective but also affordable method [60].

Also, to remove indigo carmine from waters, photocatalytic methods can be used. They break down the dye using light sources such as 172 nm excimer lamps and 254 nm Hg lamps [73]. MnO2 is considered to be an effective sonocatalyst and sonophotocatalyst for complete removal of indigo carmine from water, including small amounts. The rate of degradation depends on substrate concentration, reaction volume, ultrasound frequency and is facilitated by acidic pH. The use of sonocatalysis together with sonophotolysis increases the efficiency of degradation and mineralization of indigo carmine, so it can be removed from polluted waters [87].

Another method of removing indigo carmine from water is ozonation which can also be associated with sonolysis and photocatalytic processes. The best results were obtained when ozone was used together with ultrasound because they favor the diffusion of gases in the reaction medium and the formation of oxidizing species, such as the hydroxyl radical and peroxydryl. Although these methods remove the dye from the waters, toxic compounds are subsequently produced that further pollute the aquatic environment [64].

#### 7. Indigo carmine in medicine

Indigo carmine is one of the dyes used in medicine because it has impressive applicability in terms of diagnostic methods and surgical procedures due to its intense color that allows visualization of structures to be analyzed [21,88]. Indigo carmine was used for the first time in medicine in 1904 [89] and is generally considered to be biologically inert [90–92] and relatively safe for human health due to its predominantly renal excretion [21]. However, adverse effects of using indigo carmine have been reported in the literature [90,92].

Indigo carmine is mainly excreted by the kidneys [93,94], it quickly reaches the urine and due to the small size of the particles it allows them to be easily filtered by the kidneys without tubular resorption [95,96]. It has been observed that indigo carmine does not affect renal function and is cleared from the circulation in the bladder within 5-7 minutes [95].

Indigo carmine is used in orthopedics and trauma surgery for staining cystic lesions at the knee and hip level (a 0.8% solution) or in herniated disc surgery (a 10 or 20% solution) [91,97]. It is also used for intradiscal visualization through the preferential staining of the degenerated nucleus pulposus [98]. Thus, lesions and degenerated and abnormal areas are colored and can be visualized during the intervention [91,98,99]. To visualize the ganglions with the arthroscope, a 1:10,000 indigo carmine solution is injected intralesionally into the wrist ganglion. Thus, the stalk of the ganglion and the cyst appear colored and are easy to analyze [100]. Indigo carmine is a very useful method for visualizing a fistula and for differentiating superficial from intra-articular infections [97].

Indigo carmine is one of the most common types of vital dyes used in microsurgery, being often used in vasography, vasovasostomy, vasoepididymostomy and gynecological laparoscopy at a concentration of 25% [101].

Indigo carmine is used in gynecological and urological surgeries in combination with cystoscopy to evaluate urethral conditions and urine production during genitourinary surgeries [90,92,96,102–104]. It is administered intravenously and the recommended dye concentration is 0.4% [96,104]. Indigo carmine is often used in gynecology because, unlike methylene blue, it has not been reported to induce teratogenic changes [105].

Imai et al. [106], described a new technique for visualization the dissectable layer in vaginal hysterectomy by staining it with vasopressin and indigo carmine. Indigo carmine delineated and widened the dissectable layer and the organ injury is avoided [106].

Indigo carmine is also used in performing angiography to differentiate between tumor tissue and normal tissue in the case of ischemia of the lower extremities. Therefore, the use of indigo carmine in angiography provides essential visual information and is considered an important indicator in this pathology. The amount of indigo carmine used is 40 mg/5 ml without being diluted. The dye is injected through a catheter into the popliteal artery [107].

Indigo carmine is often used in chromoendoscopy for in vivo detection of pathologies of the colon mucosa, gastric mucosa and urinary bladder [108,109]. It has been observed that the use of indigo carmine in chromoendoscopy can allow the detection of gastric cancer in an early phase, but there are few clinical studies in the literature regarding its effectiveness in diagnosing the pathology. Generally, indigo carmine is considered a rapid diagnostic method and is used to distinguish between benign and malignant lesions, facilitates lesions analysis and delimits the borders of malignant lesions in the early stage and estimates cancer invasion depth [12,28,110].

In addition, indigo carmine is useful in recognition of intestinal lesions and in the diagnosis of patients infected with *Helicobacter pylori* [111]. Resindra et al. [112] highlighted that the use of indigo carmine in chromoendoscopy during stomach reconstruction interventions leads to a significant improvement to the completeness of the reconstruction result. It can also be used in chromoendoscopy to diagnose colorectal polyps [20], in screening of neoplasia in patients with inflammatory bowel disease [20,113] and for detecting dysplasia in colitis [114].

Indigo carmine is used as contrast agent during colonoscopy to differentiate between neoplastic and non-neoplastic colorectal lesions, being considered the most reliable method for diagnosis without performing a biopsy [115]. Frequently, indigo carmine is used to identify flat colorectal lesions. Normally a concentration of 0.5% is injected through the biopsy channel of endoscope, but the dye concentration can vary between 0.2-2% [116–119]. Also, another method for differentiating between neoplastic and non-neoplastic tissue is high pressure spraying with a solution of 0.035% indigo carmine [116,188]

Indigo carmine can also be used in microendoscopy of the lacrimal glands, allowing the identification of pathologies of the lacrimal mucosa.

#### 7.1. Adverse effects of using indigo carmine in medicine

Although indigo carmine is generally considered a safe, biologically inactive dye, numerous adverse effects associated with its intravenous administration have been reported [90,109,120].

Several studies have reported that indigo carmine has vasopressor properties [21,89,92-95,102,103,121–123] because its intravenous administration can cause transient stimulation of alfaadrenergic receptors, increases in diastolic and systolic blood pressure, and can reduce cardiac output, volume and rate [21,102,120,122]. It can also cause reflex bradycardia with decreased cerebral vascular volume, and arrhythmias but heart attack have been rarely reported [92,95]. Craik et al. [122] observed that indigo carmine causes a statistically significant increase in both systolic and diastolic blood pressure with a maximum value recorded 5 minutes after administration. At a difference of 10 minutes, a significant decrease in heart rate was observed [89,92,122]. The increase in blood pressure is thought to be due to the molecular similarity of indigo carmine to serotonin because the molecular structure of indigo carmine resembles 2 molecules of 5-hydroxytryptamine (5-HT) arranged in a mirror image [89,92,103,121–123]. However, cases of hypotension are very rare [34,92,103,123]. Indigo carmine has also been reported to affect blood oxygen saturation, but the decrease in oxygen levels is not very large [96].

Adverse effects of indigo carmine after skin contact have also been reported. After the use of indigo carmine in various surgeries, the dye extravasates into the skin and can cause severe irritation [18,95,124,125]. Fortunately, patients have no symptoms other than discoloration and edema that resolve within 24-48 hours without sequelae [95,124,125].

Rodriguez-Ferreras and Ruiz-Salazar [20] present the case of a 10 year-old patient who presented tooth discoloration following the administration of methylphenidate that also contains indigo carmine in concentrations of 30 - 40 mg per capsule. Isolated cases have also been reported in which the teeth of patients who ingested drugs containing indigo carmine became yellow due to the color change of the dye depending on the pH. In addition, other dental conditions have been reported following the administration of drugs containing indigo carmine, such as increased caries, gingival hypertrophy and bleeding, toothache, gingivitis, periodontitis, gingival erythema, etc. [20].

Indigo carmine can cause tumors in the areas where it is applied [18,89,93]. Also, eye contact may cause permanent damage to the cornea and conjunctiva [18,54].

After injecting a 0.4% dye solution into the lacrimal sac, indigo carmine-stained tissue was observed to exhibit superficial atrophy, with diminished goblet cells and subepithelial fibrosis leading to fibrous change due to inflammation [109].

# 8. Other applications of indigo carmine

Indigo carmine is used as a photometric detector, as a microscopic dye and as a redox indicator in analytical chemistry [9,54,126]. Because indigo carmine exhibits antioxidant properties related to the ability to scavenge anionic superoxide or dioxygen, it is used for ozone detection. It is also a chelator of minerals such as cooper, zinc and cobalt to the extent that they participate in the Fenton reaction which can help avoid oxidative stress [21,40,127,128].

Also, in the 60s indigo carmine was used to measure dissolved oxygen, so it was a quality indicator of beer [129]. In addition, indigo carmine is also considered an antibacterial agent and a bird feed additive [18,59].

#### 9. Toxicity testing of indigo carmine

The effect of indigo carmine on mortality has been tested on several species.

Gaunt et al. [130] administered concentrations of 150, 450 and 1350 mg/kg of indigo carmine to Sus domesticus for 90 days, and no mortality percentage was recorded after the experiment. A decrease in hemoglobin level and red blood cell count was observed after 45 and 90 days in males given 1350 mg/kg/day.

Hooson et al. [131] reported that they did not observe an increase in mortality rate of mice after administration of indigo carmine in concentrations of 0.4, 0.8 and 1.6% for 80 weeks.

Also, other studies were carried out to see other effects of the dye. Growth inhibition was observed in rats after administration of 2 and 5% indigo carmine for 2 years, but no effects on viability or pathological changes were reported [132]. Singh et al. [133] have highlighted that after oral administration of the dye to mice and rats, it has no carcinogenic and mutagenic effects. On the other hand, Ozaki et al. [134] reported that indigo carmine exhibits mutagenic activity and inhibits the growth of Bacillus subtilis species.

Dixit and Goyal [135] evaluated the toxic potential of indigo carmine on the male reproductive organ of mice. The mice were divided into 2 experimental groups. One group ingested 17 mg/kg/day of indigo carmine, and the other group ingested 39 mg/kg/day of dye. The experiment lasted 42 days. A significant increase in body weight and a significant reduction in tubular diameter, as well a decrease in sperm motility, were observed in both experimental groups. Also, a significant reduction in sperm density was observed in the group that received a higher dose of dye.

In the case of *Eisenia andrei*, the lethal dose 50 could not be established because the mortality rate was low even in the case of the highest concentration (166.67 mg/cm2) after 48 hours of exposure. The researchers estimated that after 72 hours of intoxication the concentration that can induce the lethal effect of 50% of individuals is 75.70 mg/cm2, considering that the toxic effect of indigo carmine is dependent on the exposure time [18].

Although the dye has antioxidant properties, it was observed that following the evaluation of the effects of indigo carmine treatment on the liver of rats subjected to ischemic and reperfusion injuries, indigo carmine is not sufficient to mitigate hepatocyte damage. In addition, the dye acts synergistically with ischemic lesions, thus favoring the destruction of hepatocyte membranes [21].

Mahadevan et al. [105], compared the effects of indigo carmine with those of metilene blue on granulosa luteal cell function *in vitro*. Methylene blue was found to significantly reduce the cells production of progesterone, while indigo carmine had no effect.

Indigo carmine was tested in vitro on cell cultures prepared from bovine coccygeal intervertebral disc tissue and it has been demonstrated that the staining efficacy and cytotoxicity were proportional to dye concentration but significant toxicity was observed at concentrations higher than 2.5 mg/ml [136]. Furthermore, the effect of indigo carmine was tested in vitro on human chondrocytes, where a significant decrease in the number of viable cells was observed at concentrations of 10% and 100% [97] and on human fibroblasts where it was observed a reduction in cell proliferation at concentrations of 100, 250, 500, 1000 and 2000 µg/ml [12].

Yoo et al. tested a dye mixture of indigo carmine and Lipiodol on porcine models. They concluded that the dye mixture had reliable stability and visibility for localizing lung lesions on porcine models, but for safety is recommended to test the dye mixture on the lung surface within 8 hours [137].

# 10. Conclusions

The use of dyes in various industries, especially in the food and textile industries, is a consequence of modern life. All dyes present many advantages, but at the same time they can also be considered a cause of concern for the health of the population and the environment. Since indigo carmine has many uses, but also numerous adverse effects, its use is a necessity, but also a concern. Although there are studies highlighting the adverse effects of indigo carmine consumption and the disastrous environmental effects resulting from its use in the textile industry, this dye is still intensively produced and used. For this reason there is a growing research interest in the removal of indigo carmine from wastewater.

Although indigo carmine has impressive applicability in terms of diagnostic methods and surgical procedures, there are many studies in the literature reporting adverse effects arising from its use. Thus, in the future, more studies are needed that aim the toxicity and effects of indigo carmine on global health and the environment and to find the most effective and affordable methods of removing the dye from wastewater. Also, more studies are needed to determine the concentrations of indigo carmine in wastewater, and in the future, there is a need to regulate the use of indigo carmine in the textile industry to minimize pollution.

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