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Communication

# Improved Synthesis of $\beta$ -Adrenergic Agonist and Related Structures by Microwave-Assisted Metal-Catalysed Reductive Amination

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**Abstract:** Reductive amination is a powerful tool in sustainable organic synthesis that allows chemists to access a wide range of valuable amine products using renewable feedstocks, mild reaction conditions, and with minimal waste generation. Practical applications can be found in various fields, including pharmaceuticals, contributing to greener and more sustainable chemical processes. In this work we present a heterogeneous (Rh and Pt) catalysed protocol for the fast and efficient synthesis of ractopamine hydrochloride ( $\beta$ -adrenergic drug) under MW-assisted reductive amination protocol starting from raspberry ketone and octapamine. Microwave (MW) successfully accelerated the hydrogenation reaction and reduced the reaction time from 13 hours to only 3 hours under mild conditions (50°C at 10 bar). The best catalysts were Pt/C and Rh/C, which led to high conversion and selectivity towards ractopamine:HCl. The solvent has a significant effect on the reaction outcome. The replacement of methanol with other solvents had a negative impact on the reaction yield. We also replaced the raspberry ketone with other ketone substrates, especially cyclohexanone. These preliminary experiments may be useful for further process improvements in the synthesis of  $\beta$ -adrenergic agonists and related structures.

**Keywords:** reductive amination; process intensification; microwave-assisted reactions; transition-metal catalysts; ractopamine:HCl; raspberry ketone

## 1. Introduction

Reductive amination is an important and versatile organic reaction that finds extensive applications in sustainable synthetic processes and has been intensively investigated both in both academia and industry [1,2]. Mechanistically, the reaction begins with a condensation step in which the carbonyl compound reacts with ammonia or an amine to form the corresponding imine, followed by reduction of the imine to the alkylamine product [3]. Many of these reduction steps require the presence of a catalyst to activate the reducing agent [4] and transition-metals (such as Ru, Rh, Pd, Ag, Ir, Pt, and Au) have been used extensively [5] also in the preparation of active pharmaceutical ingredients (APIs) [6,7]. Despite its cost, Rh also shows remarkable activity for such reaction as shown in recent literature [8–11] and has the potential to be employed in robust hydrogenation catalysts with low metal loading. A Rh/Al<sub>2</sub>O<sub>3</sub>-catalyzed reductive amination of furfural and other aldehydes, especially those associated with biomass, was recently reported by Chatterjee, Kawanami, and coworkers using an aqueous solution of NH<sub>3</sub> [12]. A very high selectivity to furfurylamine (~92%) was documented by the authors in only 120 min at 80 °C with the chance to reuse the catalyst several times.

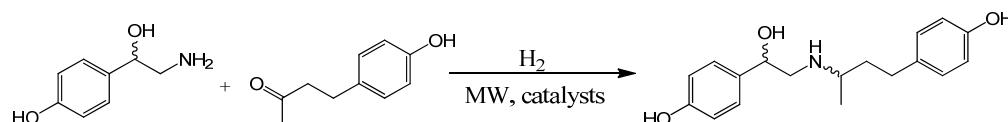
Moreover, reductive amination has attracted considerable attention in recent years in sustainable organic synthesis due to its environmental advantages and broad synthetic utility [13,14].

Apart from its versatility, as primary, secondary, and tertiary alkylamines are readily available, reductive amination usually exhibits good atom economy, which means that most of the atoms present in the starting materials end up in the desired product. This minimizes waste generation and maximizes resource utilization. In addition, reductive amination can often be carried out under mild conditions with remarkable energy saving [15]. In this framework, microwave heating (MW) is a mature technology that could be exploited to boost the sustainability of reductive amination protocols. Indeed, MW promotes the remarkable reduction of reaction time, yields enhancement, and cleaner reactions compared to conventional thermal heating. MW irradiation is, therefore, a suitable tool to overcome the limitations encountered in challenging organic reactions, such as direct reductive amination in the presence of heterogeneous catalysts and gaseous reagents. In particular, the use of metallic nanoparticles as catalysts also allows a positive synergy with microwaves (MW) heating [16,17], an alternative technology that allows faster heating rate and less side-reaction on the surface of the reactor, [18] which with convective heating are usually hotter than the bulk, at least in the laboratory scale. In this work we present a Rh catalysed optimized protocol for the fast and sustainable synthesis of ractopamine (a  $\beta$ -adrenergic compound) under MW-assisted protocol.

Ractopamine hydrochloride is the common name for the 4-[3-[(2-hydroxy-2-(4-hydroxyphenyl)ethyl] amino] butyl] phenol hydrochloride. It occurs in four stereoisomers (RR, RS, RS, SS) and it is a bioactive molecule defined as a phenyl ethanolamine  $\beta$ -adrenoceptor agonist ( $\beta$ -agonists) [19]. The  $\beta$ -agonists redirect nutrients away from fat deposition to muscle deposition involving the modulation of metabolic pathways and signals in muscle and lipid cells to enhance protein accretion. Other mechanisms also include regulation of hormone release and modification of blood flow [20]. Ractopamine :HCl is currently used as a feed additive to improve the feed efficiency and the carcass leanness in swine and cattle, as allowed in the USA, Canada, Japan and Mexico [21].

The synthetic method of relevant ractopamine hydrochloride has been introduced following several routes in existing literature. In particular, two methods are introduced in Lilly Co., Eli's patent [22]. The first method involves the synthesis of ractopamine:HCl using raspberry ketone and right-hydroxymandelic acid as raw materials. The latter involve the use of *p*-hydroxyphenylethanol amine instead of hydroxymandelic acid. These synthetic pathways do not lead to complete conversions and high selectivity, leading to the formation of numerous by-products. Additionally, the reaction conditions, separation, and purification processes are all quite challenging. Another synthetic approach was proposed by the Chengdu Organic Chemistry Institute, Chinese Academy of Sciences, in their patent [23] and developed by Jie Shaoing. Starting from raspberry ketone, this method included a ketoxime reaction followed by a condensation reaction with  $\omega$ -bromo-parahydroxyacetophenone and triethylamine under low-temperature conditions. Subsequently, a catalytic hydrogenation step was required to yield the ractopamine:HCl. However, ractopamine is currently synthesized via reductive amination between raspberry ketone and octopamine (Scheme 1) [24] in which the hydrogenation step can be easily catalysed by heterogeneous metal-based catalyst [25,26] such as Pt [27–29] or Pd [30,31], which can be easily recovered and reused.

In this piece of work, we present a study on the MW-assisted synthesis of ractopamine, in which Pt/C, Ru/C and Rh/C (all 5 wt.%) were tested as heterogeneous catalysts. Their activity was studied at different temperatures and hydrogen pressure. Moreover, the reaction was carried out in different solvents and different ketone substrates were considered for the reaction with octopamine under optimized conditions to highlight the positive influence of dielectric heating and heterogeneous catalysis.



**Tested catalysts: Pt/C, Rh/C, Ru/C, Co/C**

**Scheme 1.**

## 2. Materials and Methods

The heterogeneous Pt/C, Ru/C and Rh/C 5 wt.% catalysts were all purchased from Merk. Octopamine HCl, raspberry ketone and ractopamine HCl were supplied from Huvepharma Italia. The other ketone substrates were purchased from Merck KGaA (Darmstadt, Germany). The batch reaction with conventional heating was performed in a 300 mL PolyBlock reactor equipped with a PTFE impeller for stirring. The reactor has a minimum working volume of 41 mL and operates up to 100 bar and 250 °C. A sampling valve on the bottom allows the removal of the crude reaction mixture for analysis. The reactor was flushed three times with 2 bar of N<sub>2</sub> before the appropriate pressure of H<sub>2</sub> was introduced. The MW-assisted reductive amination reactions were performed in a MW autoclave (SynthWAVE, Milestone srl - Bergamo) in 15 ml glass vials provided with magnetic stirring (500 rpm) all inside a 1 L PTFE reaction chamber filled with 200 mL of brine solution used to guarantee a regular MW absorption. The reaction chamber was flushed three times with 2 bar of N<sub>2</sub> before the appropriate H<sub>2</sub> pressure was introduced. A heating ramp of 2 min was used, with a maximum MW power of 800 W.

After the reaction the crude was recovered and filtered over paper to recover the catalyst. The filtrate was dried under vacuum, redissolved in pyridine and derivatised with N,O-Bis(trimethylsilyl)trifluoroacetamide (BSTFA) prior to GC-MS analysis.

The GC-MS equipment was an Agilent Technologies 6850 Network GC System fitted with a 5973 Network Mass Selective Detector, a 7683B Automatic Sampler and a capillary column Mega 5MS (length 30 m; i.d. 0.25 mm; film thickness 0.25 μm, Mega Srl, Legnano, Italy). The HPLC-DAD equipment used for the diastereomeric ratio determination was an Agilent 1100 using an Hypersil ODC C18 column (5 μm particles diameter and 4.6 mm x 250mm). The mobile phases were an ammonium hydroxide orthophosphate buffer (A) and acetonitrile (B). The peaks were recorded at 226 nm.

## 3. Results

Optimization of ractopamine synthesis was primarily investigated using a commercial Pt/C (5 wt.%) catalyst as a reference material for the hydrogenation step. The reductive amination between octopamine and raspberry ketone was first carried out in batch at lab scale under conventional heating (Scheme 1). A KOH solution in MeOH was prepared and reagents were slowly added. The pH was adjusted to 6.5 with glacial acetic acid. The hydrogenation step was carried out at 50 °C with 5 bar H<sub>2</sub>. Under these conditions, 13 hours were required to achieve complete coupling and hydrogenation of the reagents with a final yield of 98.9% in ractopamine:HCl. Using these batch conditions as benchmark, we moved on to the MW experiments (Table 1). First, the effect of temperature and H<sub>2</sub> pressure on ractopamine yield was investigated. It was found that increasing the temperature from 50 to 80 °C increased the yield, but at the expense of selectivity. In contrast, increasing the H<sub>2</sub> pressure from 5 to 10 bar resulted in an increase in both conversion and selectivity at 50 °C.

**Table 1.** Influence of temperature and hydrogen pressure.

Entry	T (°C)	H <sub>2</sub> (bar)	Conversion (%)	Selectivity (%)	Yield (%)
1	50	5	10.0	63.2	6.3
2	80	5	87.2	47.0	41.0
3	50	10	91.6	94.5	86.6

Reaction conditions: 1 eq. raspberry ketone, 1 eq. octopamine, 1 eq. KOH, 5 mL MeOH/H<sub>2</sub>O 1:1, 10 mg Pt/C 5 wt.%, 3 h.

The main by-product observed is imine, which is formed from the coupling between raspberry ketone and octopamine but is not hydrogenated. HPLC analysis (Appendix A, Figure 1A, Table 1 A) to determine the diastereomeric ratio (RS-SR /SS-RR) gave a value of 47/53, which falls within the

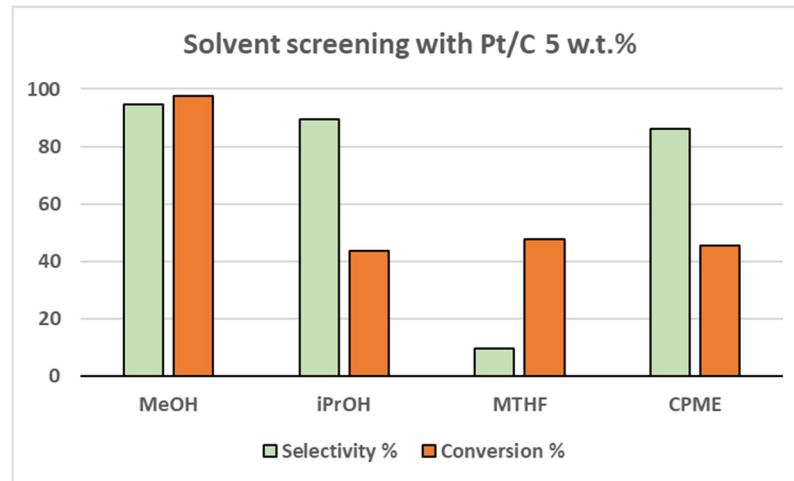
reference ratio of the commercial product (RS-SR 45-49%). After setting 50 °C and 10 bar H<sub>2</sub> as the optimal reaction conditions, we compared the activity of Pt/C 5 wt% with other heterogeneous catalyst based on Rh and Ru with the same support and metal loading (Table 2). Results show that Pt and Rh have similar activities, while Ru, despite comparable conversions, does not promote the hydrogenation step and thus leads only to an imine intermediate. Rh/C was also tested with 5 bar H<sub>2</sub>. In this case, the activity was higher than Pt/C, but both the conversion and selectivity decreased compared to the reaction carried out at 10 bar. The crude reaction product using Rh/C was also within the acceptable diastereomeric ratio and with the same value of 47/53 as the Pt/C test (Appendix A, Figure 2A, Table 2A).

**Table 2.** heterogeneous catalyst screening.

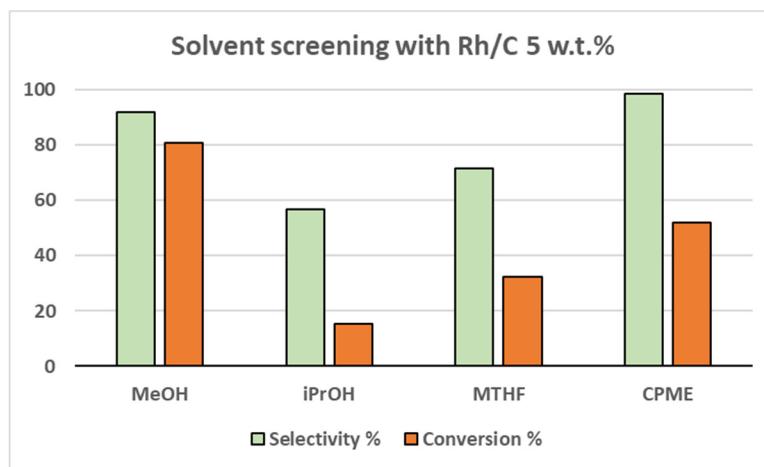
Entry	Catalyst	H <sub>2</sub> (bar)	Conversion (%)	Selectivity (%)	Yield (%)
1	Pt/C	10	91.6	94.5	86.6
2	Rh/C	10	81.0	91.9	74.4
3	Ru/C	10	73.0	n.d.	-
4	Rh/C	5	47.0	80.4	37.8

Reaction conditions: 1 eq. raspberry ketone, 1 eq. octopamine, 1 eq. KOH, 5 ml MeOH/H<sub>2</sub>O 1:1, 10 mg Pt/C or Rh/C 5 wt.%, 50°C, 10 bar H<sub>2</sub>, 3 h.

Different solvents were also tested for the reaction in order to scout for greener and safer alternatives to methanol (Figures 1 and 2). Isopropanol (*i*PrOH), 2-methyltetrahydrofuran (MTHF) and cyclopentylmethylether (CPME) were thus selected. However, the methanolic solution proved to be the optimal solvent for both catalysts. The second-best result was achieved with CPME both over Pt/C (39.1% yield) and Rh/C (51.3% yield).



**Figure 1.** Solvent screening over Pt/C 5 wt.%.



**Figure 2.** solvent screening over Rh/C 5 wt.%.

Finally, to broaden the scope of our study, other ketones were used as substrates for the reaction (Table 3). Acetophenone, cyclohexanone and 2-butanone were selected to provide an array of cyclic, linear, aromatic and aliphatic substrates. While acetophenone gave low selectivity in the tested conditions, both cyclohexanone and 2-butanone gave excellent results over both Pt/C and Rh/C. Also, using 10 bar of H<sub>2</sub> was a disadvantage since selectivity was lower than with 5 bar only.

**Table 3.** conversion of ketone substrates over Pd/C and Rh/C 5 wt.%.

Entry	Catalyst	Substrate	H <sub>2</sub> (bar)	Conversion (%)	Selectivity (%)	Yield (%)
1		Acetophenone	5	n.d.	n.d.	-
2			10	89.6	29.2	26.2
3	Pt/C	Cyclohexanone	5	>99	>99	>99
4			10	>99	>99	>99
5		2-Butanone	5	>99	80.2	80.2
6			10	>99	96.3	96.3
7		Acetophenone	5	>99	8.2	8.2
8			10	>99	8.0	8.0
9	Rh/C	Cyclohexanone	5	>99	82.7	82.7
10			10	>99	22.1	22.1
11		2-Butanone	5	>99	92.4	92.4
12			10	>99	50.2	50.6

Reaction conditions: 1 eq. ketone, 1 eq. octopamine, 1 eq. KOH, 5 ml MeOH/H<sub>2</sub>O 1:1, 10 mg Pt/C or Rh/C 5 wt.%, 50°C, 3 h.

## 5. Conclusions

The effect of temperature and hydrogen pressure was studied for the synthesis of ractopamine by reductive amination. Increasing the temperature from 50 to 80 °C had a positive effect on the conversion, but at the expense of selectivity, while increasing the H<sub>2</sub> pressure from 5 to 10 bar improved the yield from 6.3 to 86.6%. Different heterogeneous catalysts were tested and both Pt/C and Rh/C (5 wt.%) were found to be suitable for the reaction, which was completed in 3 hours instead of 13 hours as in the batch tests using conventional heating. Finally, various ketone substrates were coupled with octopamine, demonstrating the flexibility of the setup that allows yields up to 99% even at 5 bar of H<sub>2</sub>.

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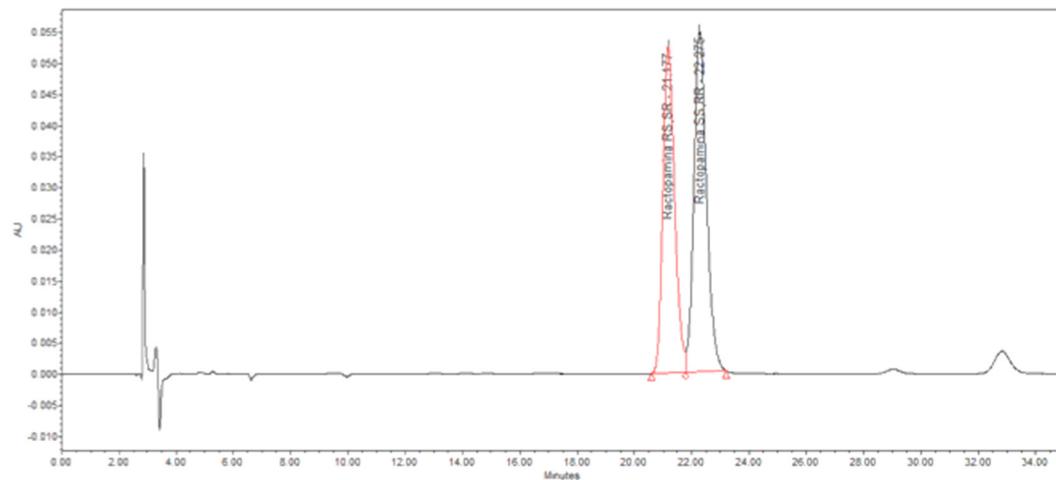
and editing, E.C.G. and G.C.; supervision, E.C.G. and G.C. All authors have read and agreed to the published version of the manuscript.

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## Appendix A

The best reaction conditions for the MW-assisted synthesis of ractopamine (50 °C under MW 180 min and H<sub>2</sub> (10 bar), using Pt/C (5% w/w) (Table 2, entry 1) were studied in detail and products characterized by HPLC-DAD. Analyses were performed with a validated HPLC method described in paragraph 2. The results (Figure 1A and Table 1A) indicate that the newly synthesized ractopamine:HCl is on specification [32].

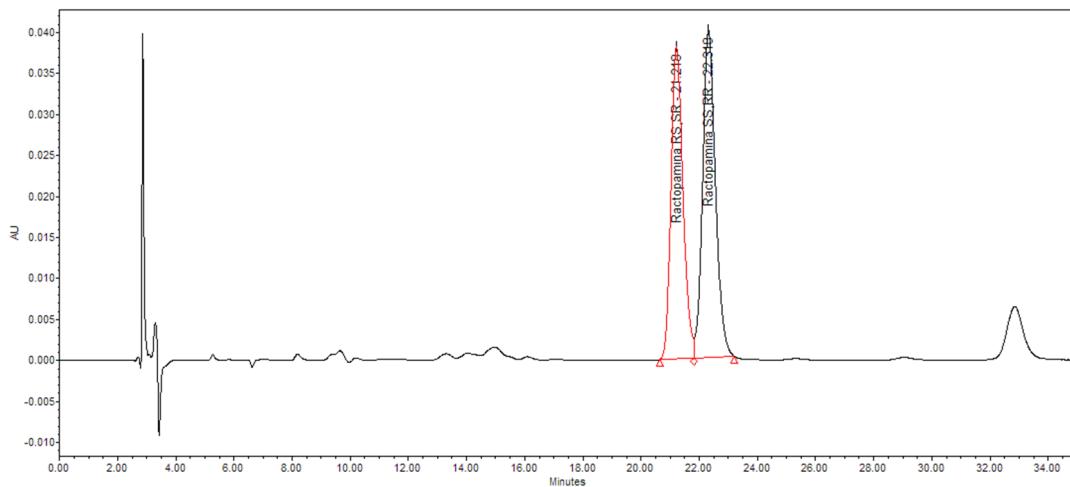


**Figure 1A.** HPLC Chromatogram - diastereomeric ratio determination at the best reaction conditions: 50 °C with MW for 180 min and H<sub>2</sub> (10 bar), using Pt/C 5% as catalyst.

**Table 1A** - Diastereomeric ratio of the sample obtained at the best reaction conditions: 50 °C with MW for 180 min and H<sub>2</sub> pressure at 10 bar, using Pt/C 5% w/w as catalyst.

Name	Retention Time	Area	% Area	Specification
Ractopamine RS,SR	21.177	1497290	47.00	RS,SR 45–49%
Ractopamine SS,RR	22.275	1688481	53.00	

The RS,SR diastereomeric ratio was also evaluated for the ractopamine:HCl sample synthetized under MW 180 min at 50 °C and H<sub>2</sub> (10 bar) with Rh/C as catalyst (Table 2, entry 2). Analyses were performed following a validated HPLC method described in paragraph 2. and the results showed that the sample is on specification (Figure 2A, Table 2A). [31]



**Figure 2A.** HPLC Chromatogram - diastereomeric ratio determination at the best reaction conditions: 50 °C with MW for 180 min and H<sub>2</sub> (10 bar), using Rh/C 5% w/w as catalyst.

**Table 2A** - Diastereomeric ratio of the sample obtained at the best reaction conditions: 50 °C with MW for 180 min and H<sub>2</sub> (10 bar), using Rh/C 5% w/w as catalyst.

Name	Retention Time	Area	% Area	Specification
Ractopamine RS,SR	21.218	1085144	47.06	RS,SR 45–49%
Ractopamine SS,RR	22.318	1220961	52.94	

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