

Supplementary Materials

Understanding the Origin of Luminescence in Porous Organosilica Films with Various Organic Components

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Table S1: Physical properties of organosilicate glass (OSG) films: first type: both ethylene bridges and methyl terminal groups with different precursor ratios (BTMSE/MTMS = 47/53 and 25/75); second type: a mixture of 1,3,5- and 1,3-benzene groups with a ratio of 1,3,5/1,3 = 1:3, 1,3,5/1,3 = 1:7 and 1,4-benzene bridges without porogen (1,4-BB) and with 30 wt% porogen (1,4-BB-p); third type: contains only methyl terminals without bridging groups (MTMS – without porogen, MTMS-p – with porogen). All films are completely cured in the air.

OSG films	EP data				FTIR data		
	Porogen concentration (wt%)	<i>d</i> (nm)	RI	Open porosity (%)	Pore radius (nm)	Surface area (m ² /cm ³)	Si-CH ₃ /Si-O-Si ratio
BTMSE/MT MS = 47/53	0	478.0	1.397	1.2	0.81	31.72	0.151
	10	493.6	1.349	12.2	0.81	303.26	0.142
	20	485.2	1.335	25.1	0.81	621.81	0.150
	30	465.0	1.306	32.5	1.42	459.89	0.183
	50	434.2	1.223	45.5	2.34	391.19	0.186
BTMSE/MT MS = 25/53	0	617.3	1.453	8.7	0.81	216.43	0.299
	10	622.0	1.370	12.4	0.81	307.78	0.274
	20	598.3	1.357	19.4	0.97	401.66	0.283
	30	629.2	1.308	27.7	1.42	391.73	0.290
	50	585.1	1.200	50.6	4.65	219.34	0.298
1,3,5/1,3BB (1:3)	0	254.9	1.477	-	-	-	0.003
	17	356.4	1.316	28.9	0.81	716.39	0.026
	23	402.5	1.285	38.7	0.81	958.04	0.033
	29	420.1	1.270	42.4	0.81	1049.29	0.053
	33	416.3	1.268	42.9	0.92	935.01	0.059
	38	413.3	1.256	41.2	0.92	898.07	0.033
	41	398.9	1.285	37.9	1.1	691.59	0.053
1,3,5/1,3BB (1:7)	0	315.7	1.470	-	-	-	0.080
	17	377.7	1.324	26.2	0.81	649.56	0.014
	23	401.9	1.281	36.7	0.81	908.66	0.026
	29	405.4	1.272	40.4	0.81	999.99	0.058
	33	401.5	1.266	41.9	0.92	913.36	0.079
	38	419.3	1.259	41.9	0.92	913.25	0.038
	41	425.0	1.25	41.9	0.92	913.22	0.054
1,4-BB	0	177.0	1.559	10.3	0.92	229.56	0.001
1,4-BB-p	30	169.0	1.428	29.3	0.76	776.96	0.012
MTMS	0	266.0	1.363	7.5	0.71	215.02	0.048
MTMS-p	30	226.0	1.300	33.1	4.1	165.88	0.044

*OSG – Organosilicate glass, EP – Ellipsometric Porosimetry,

FTIR – Fourier Transform Infrared Spectroscopy, *d* – thickness, RI – Refractive index

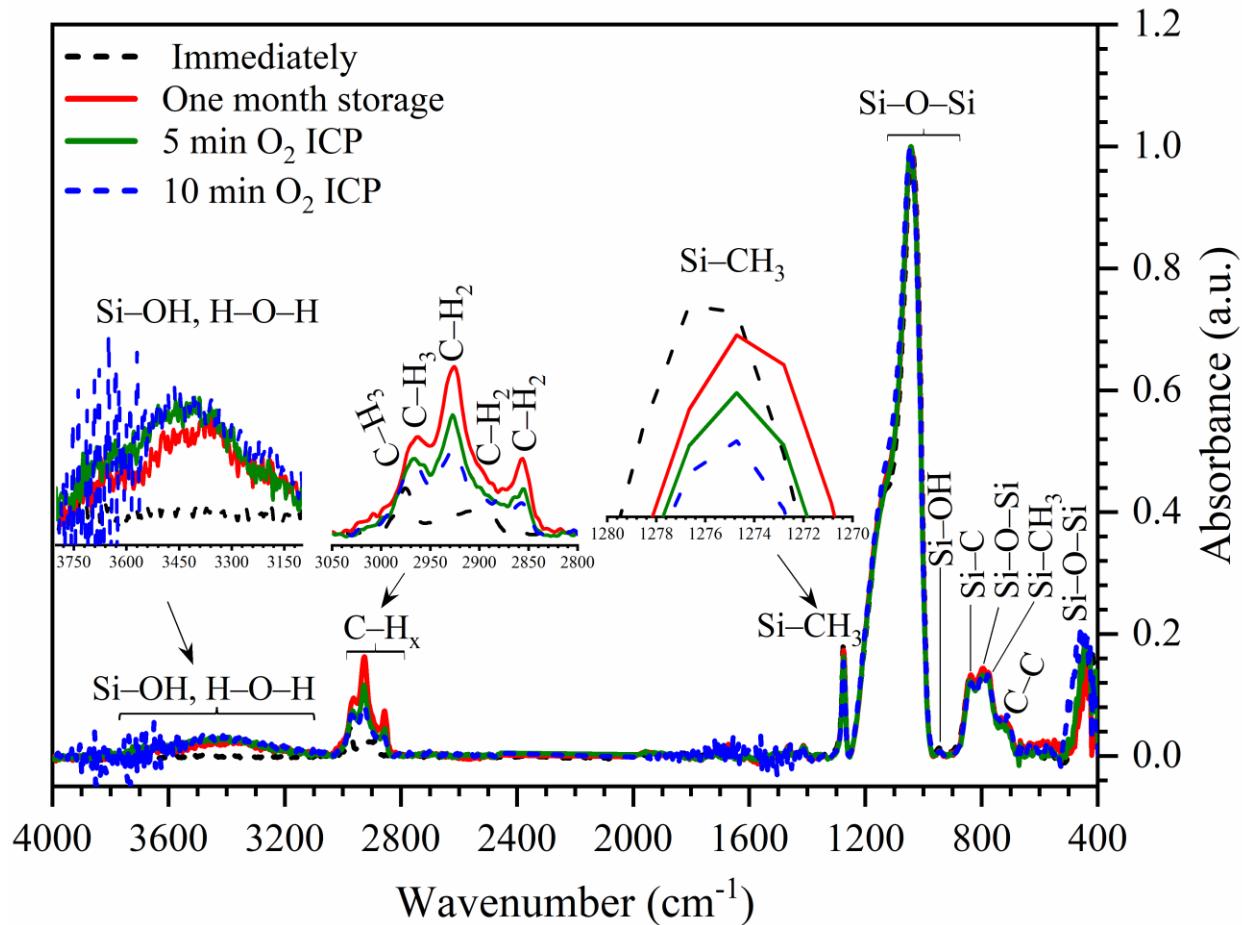


Figure S1: Effect of chemical composition during storage of film (BTMSE/MTMS = 47/53, 20 wt% Brij[®]30) in clean room environment and after soft oxygen ICP plasma. Loss of Si-CH₃ groups and hydrophilization was observed for films by using soft oxygen ICP plasma (remove the chemical residue).

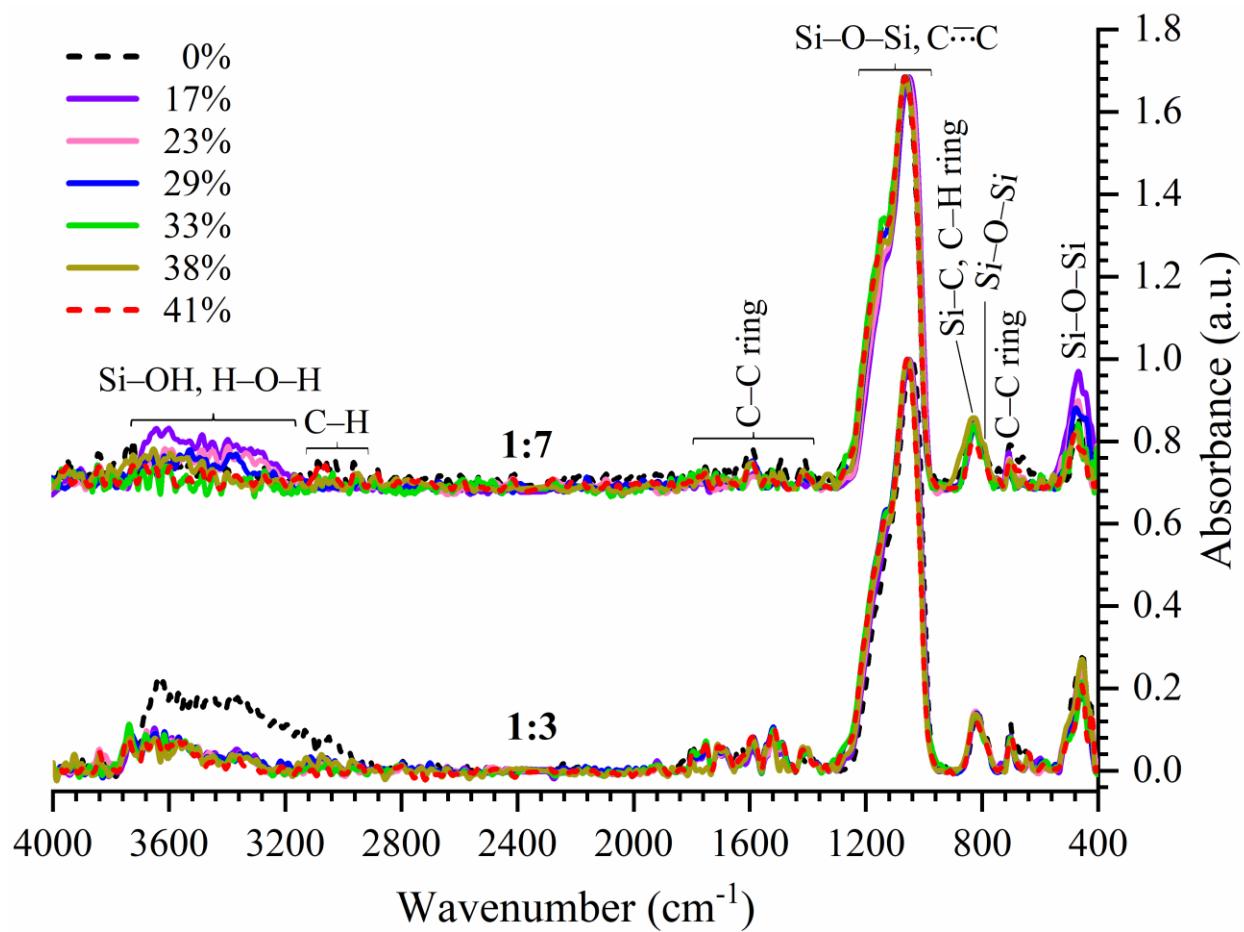


Figure S2. FTIR spectra of 1,3,5- and 1,3-benzene bridged ratio of 1:3 (bottom) and 1:7 (top) organosilica films with different porogen concentration s (0–41 wt%), soft bake at 200 °C and hard bake at 400 °C in air.

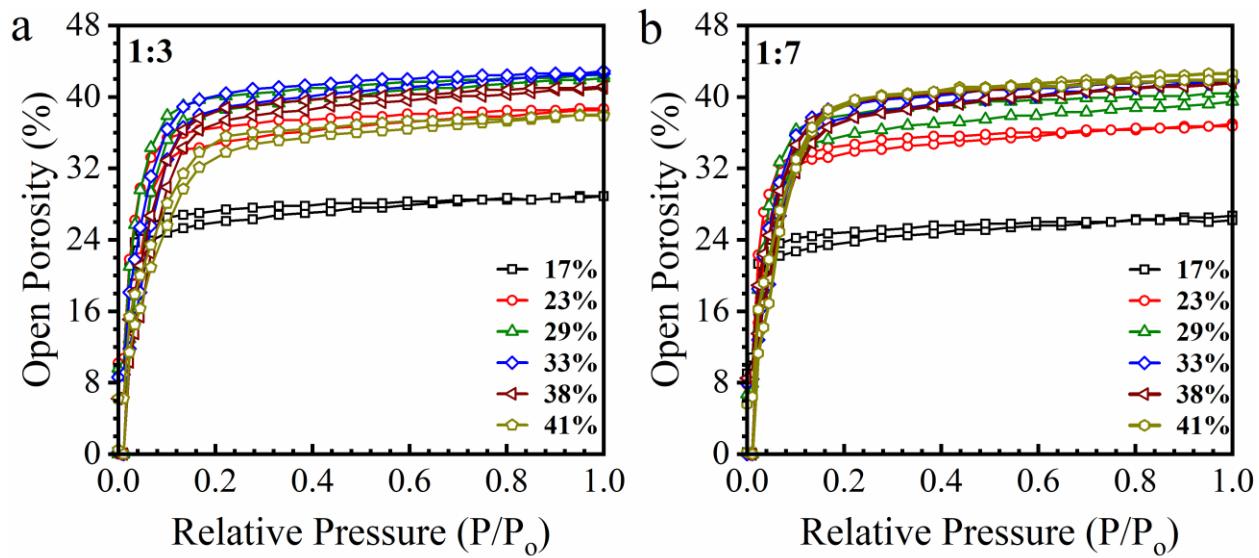


Figure S3: Adsorption-desorption isotherms for determining the value of open porosity, generated by ellipsometric porosimetry, of 1,3,5- and 1,3-benzene bridges ratio of 1:3 (a) and 1:7 (b) organosilica films with different porogen concentrations (17–41 wt%), soft bake at 150 °C and hard bake at 400 °C in air.

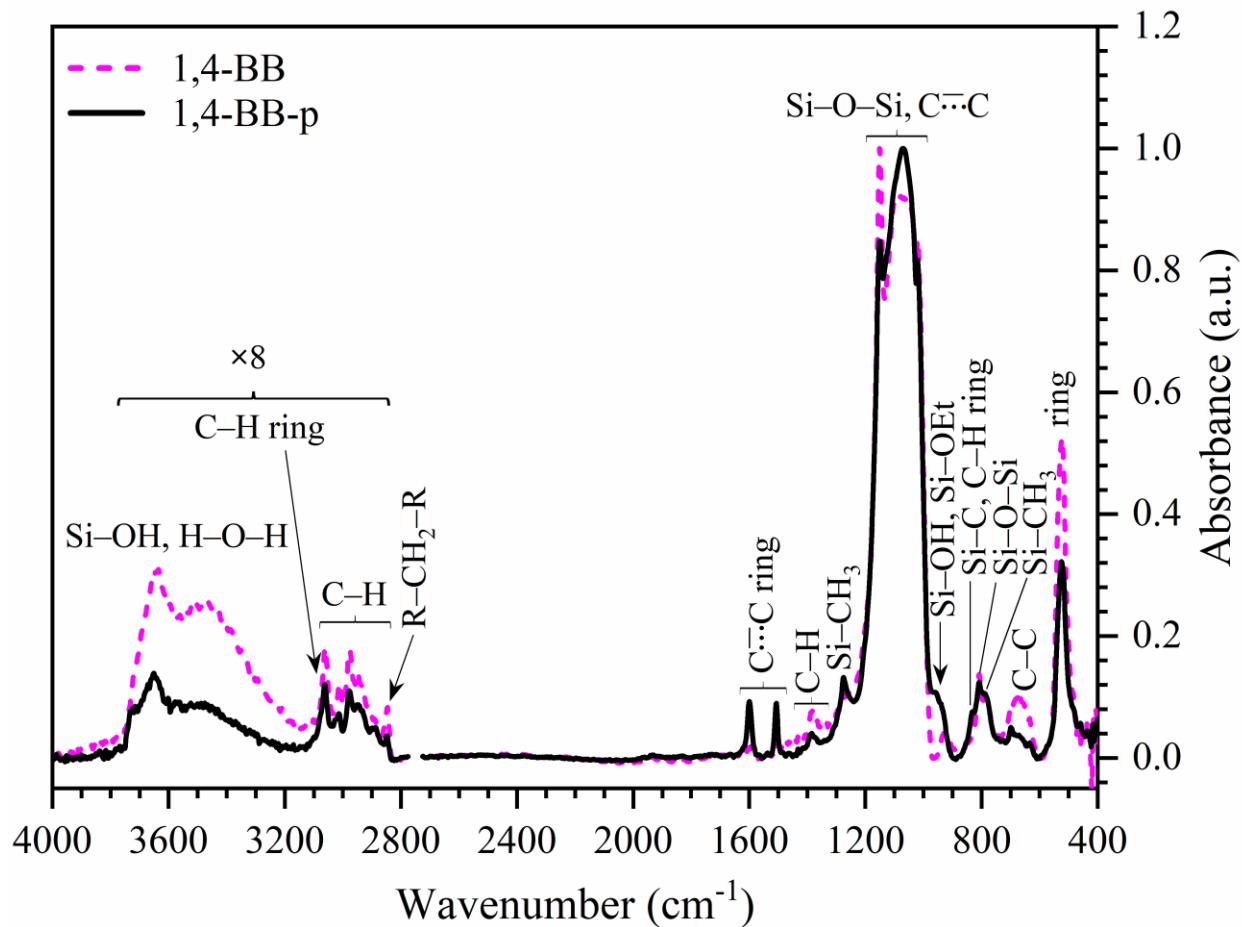


Figure S4: FTIR spectra of pure 1,4-benzene bridged organosilica films: dense (1,4-BB) and with 30 wt% porogen content 1,4-benzene bridged (1,4-BB-p) organosilica films, both the films were completely cured in air.

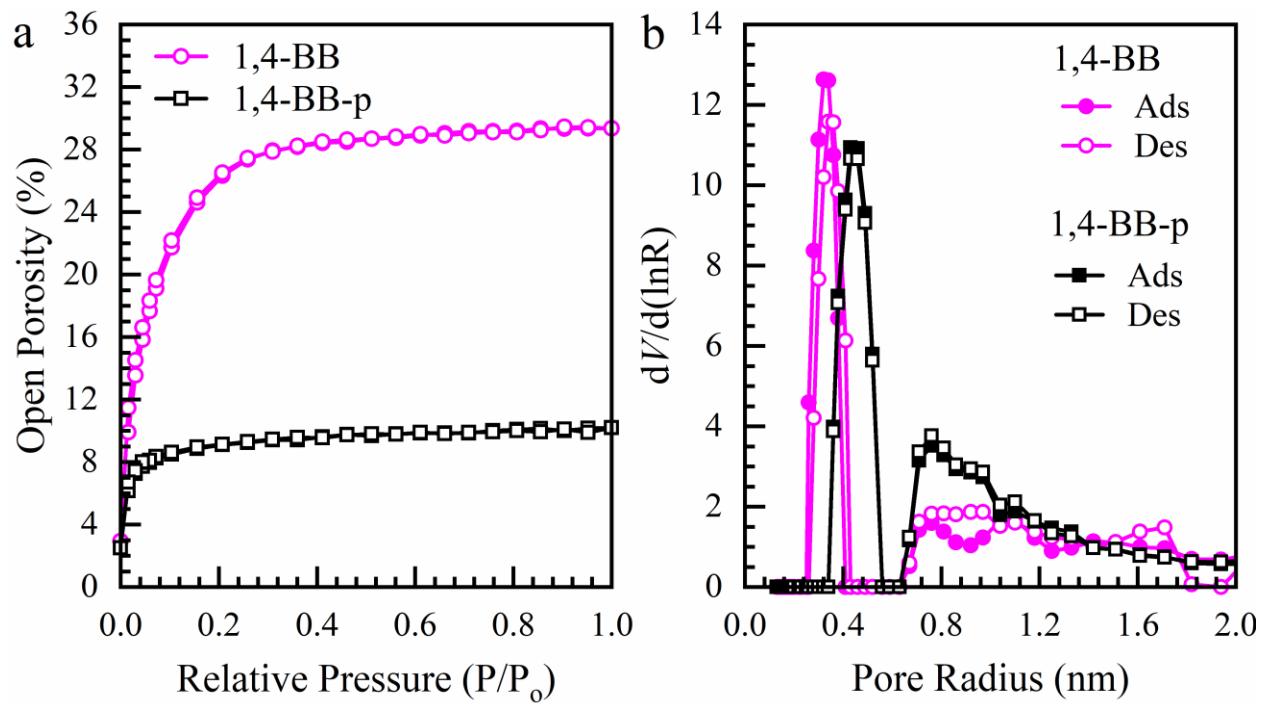


Figure S5: Adsorption-desorption isotherms for determining the value of open porosity (a) and pore size distribution (b), generated by ellipsometric porosimetry, of pure 1,4-benzene bridged organosilica films: dense (1,4-BB) and with 30 wt% porogen content 1,4-benzene bridged (1,4-BB-p) organosilica films, both the films were cured completely in air.

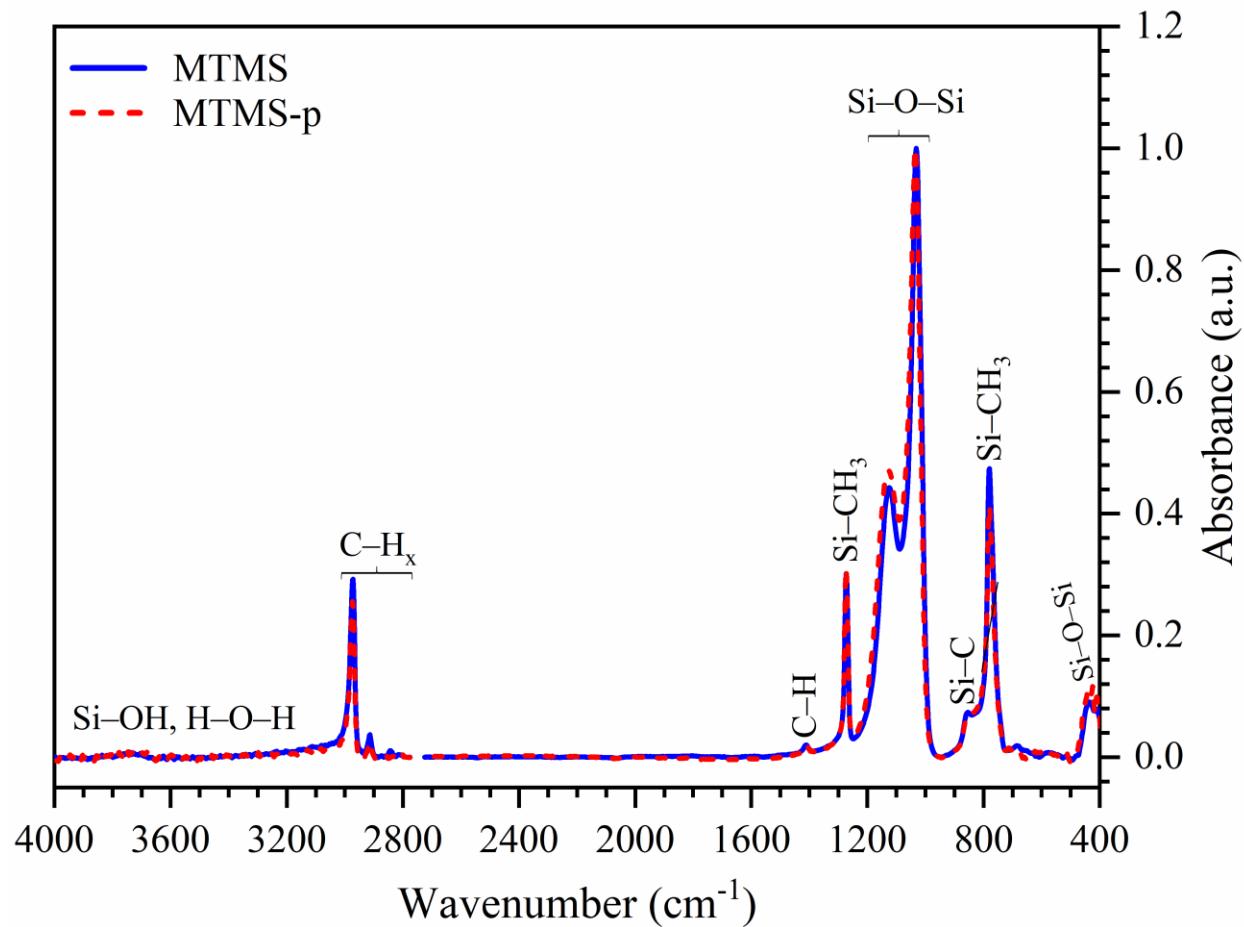


Figure S6: FTIR spectra of pure methyl-terminated organosilicate : dense (MTMS) and with 30 wt% porogen content methyl-terminated (MTMS-p), both the films were cured completely in air.

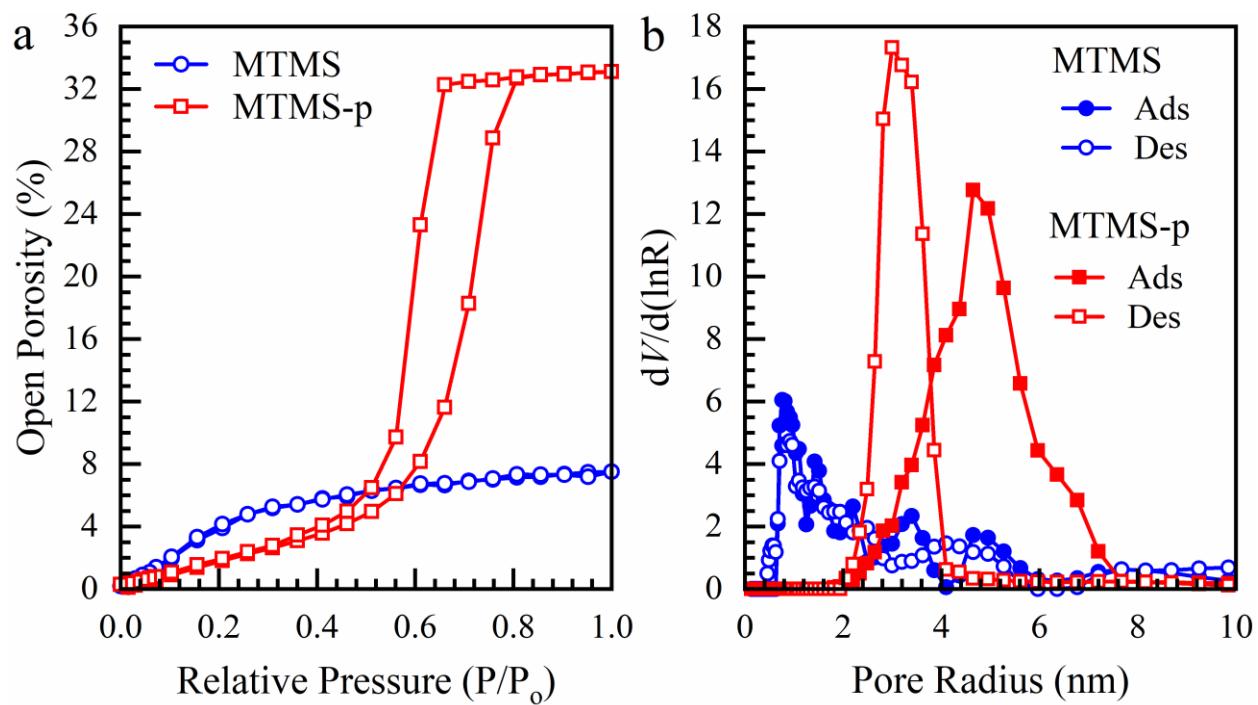


Figure S7: Adsorption-desorption isotherms for determining the value of open porosity (a) and pore size distribution (b), generated by ellipsometric porosimetry, of pure methyl-terminated organosilica films: dense (MTMS) and with 30 wt% porogen content methyl terminated (MTMS-p) organosilica films, both the films were cured completely in the air.

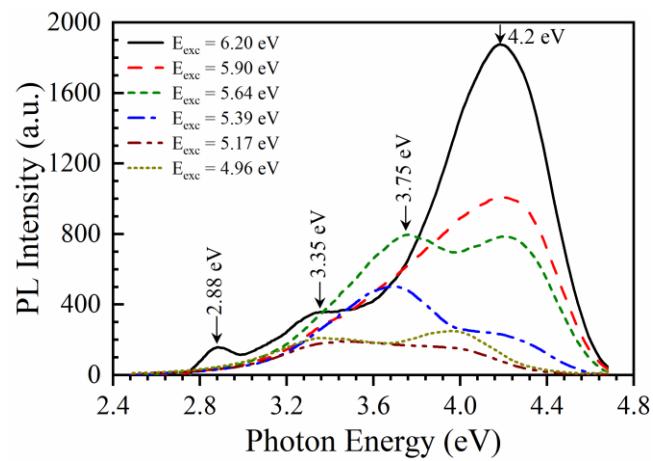


Figure S8: UV induced PL spectra measured at the different excitation energy for the ethylene bridged films (BTMSE/MTMS = 47/53).

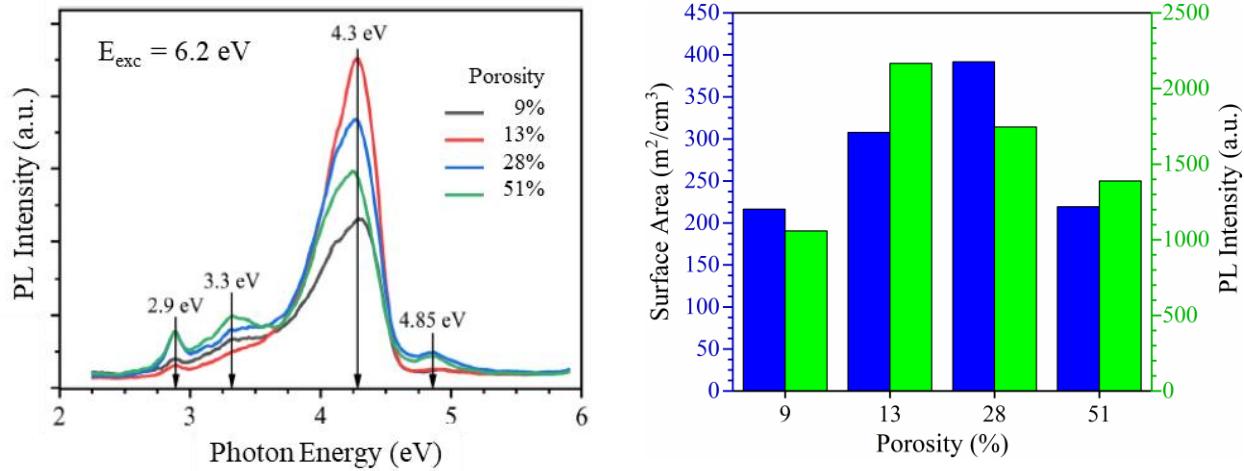


Figure S9: Correlation between 4.3 eV UV induced (upon excitation with light of 6.2 eV) room temperature PL emission with the measured surface area in hard baked OSG low-k films containing both methyl terminal and ethylene bridging groups (BTMSE/MTMS ratio of 25/75) with different porosity. However, the PL bands at 2.9, 3.3 and 4.85 eV have the highest intensity in highly porous films. It can be assumed that the reason is that the terminal methyl groups and hydrocarbon residues are mainly located on the pore wall surface. However, this assumption needs further study.