

Article

The Ulug-Sair Gold-Bismuth Ore Occurrence (Western Tuva, Russia): PT Parameters, The Composition of Fluids, and Isotopy of S and O

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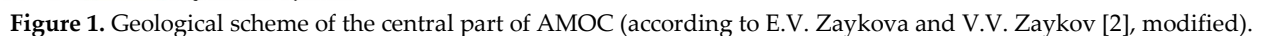
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Abstract: We examined PT parameters, geochemical peculiarities, and fluid sources of the Ulug-Sair ore occurrence attributable to class of intrusion-related gold deposits and according to ore mineral assemblages corresponding to Au-Bi type with wide Bi minerals (AgBiTe, Bi₂Te₂Se, Cu_{3,07}BiS₃, Bi), tellurides (Au and Ag), Se-tellurides (Ag and Bi), and selenides (Au, Ag, and Hg). We identified that ‘pre-gold’ quartz-tourmaline veins were deposited using an aqueous Mg-Na-K-chloride fluid with a salinity of 8–10 wt % NaCl eq. At 325–370 °C; host breasts were formed due to a CO₂-water fluid containing CH₄ and N₂, with a salinity of 0,18–6,1 wt % NaCl eq. at least 200–400 °C. Gold-bearing mineral assemblages were formed at P ~ 0,75–1,0 kbar (~ 2,3–3 km) due to CO₂-water chloride (Na-K±Fe, Mg) fluid with CH₄, Na₂SO₄, and Na₂B₂O₅, and salinities 1,7–12,5 wt % NaCl eq. during the decreasing temperatures from 360 up to 115 °C (gold-sulfide-quartz veins – 360–130 °C, and gold-telluride-sulfide-quartz veins – 330–115 °C) and variations f_{O_2} , f_{S_2} , f_{Se_2} , and f_{Te_2} . The isotopic composition of $\delta^{34}S_{H_2S}$ fluid (-0,7...+2,5 ‰) indicates the juvenile or magmatic origin of fluid and ore elements. The $\delta^{34}O_{H_2O}$ fluid indicates that, at an early substage, the formation of ore occurrence involved a fluid of magmatic or metamorphic origin (+7,3...+11,4 ‰), and, in the later substage, it mixed with meteoric waters (-2,3...+9,1 ‰).

Keywords: gold; fluid inclusions; quartz; stable isotopes; gold deposits, Western Tuva

1. Introduction

During the geological survey, searching, and prospecting work of 1952–1977 in Western Tuva, gold-quartz ore occurrences were discovered (Ulug-Sair, Aryska, Khaak-Sair, Dushkunnug, Ak-Dag, etc.) and numerous gold mineralization points associated with medium-temperature beresite and listvenite metasomatites. The largest of them are the occurrences of the Khaak-Sair in listvenites [1] and the Ulug-Sair – in conglomerates, siltstones, and beresites. They are located in the Aldan-Maadyr ore cluster (AMOC), which is considered one of the promising ore clusters for native Au (Figure 1) in the region.



Gold mineralization is controlled by narrow linear anticlines and horst anticlines of a sub-latitudinal strike and the supporting fractures of the Sayany-Tuva deep fault of the same orientation cutting them. A similar orientation of the main discontinuous and

folded structures led to a linear distribution of igneous rocks and narrow linear zones of metasomatites of the beresites & listvenite series with gold-quartz veins.

Gold mineralization is associated with intrusions of the Bayankol complex (D3bn) which are represented as dikes of as follows granodiorite- and tonalite porphyry, rhyolite and granodiorite porphyry dykes [5].

Gold mineralization is paragenetically related to the intrusions of the Bayankol complex (D3bn), which are represented as dikes of granodiorite and tonalite porphyries, rhyolites, and granodiorite porphyries [5]. The results of $^{40}\text{Ar}/^{39}\text{Ar}$ dating of synore listvenites (379.4 ± 4.4 Ma) of the Khaak-Sair ore occurrence are agreed very well with the $^{40}\text{Ar}/^{39}\text{Ar}$ age (376.5 ± 3.4 Ma) of phase III gabbro dikes of the Bayankol intrusive complex (D3bn), confirmed the genetic relationship of mineralization with the Late Devonian Bayankol intrusive complex [6,7].

AMOC is one of the perspective ore clusters for native Au in the region. Ore occurrences of AMOC are characterized by the low sulfide content ($< 3\%$) and by the presence of the Au-Bi-Te-Se mineral assemblages namely petzite, hessite, altaite, coloradoite, Te-bismuthite, tsumoite, Se-volynskite, kawazulite, wittichenite, native Bi, fischesserite, clausthalite, naumannite, and tiemannite [7,8].

The relevance of the study is to determine the mineralogical-geochemical features and fluid regime of the Ulug-Sair gold-quartz ore occurrence with a peculiar mineral composition of the ores characterized by the presence of tellurides (Ag_3AuTe_2 , Ag_2Te), selenides (Ag_3AuSe_2 , Ag_2Se , HgSe) and diverse Bi minerals (AgBiTe , $\text{Bi}_2\text{Te}_2\text{Se}$, $\text{Cu}_3\text{Bi}_2\text{S}_3$, Bi).

2. Geological setting

The Ulug-Sair ore occurrence is located in the central part of AMOC and was discovered in 1964 by Victor V. Zaykov during the geological survey in 1:50000 scales [2]. It is confined to the axial part of the host-anticline of sub-latitudinal strike and complicated by feathering faults: in the north – Arzhansky Fault, in the south – “Ore” Fault accompanied with numerous cracks with incidence angles of $75\text{--}90^\circ$. The core of the horst anticline is composed of Ordovician conglomerates, siltstones, sandstones, and Vendian-Lower Cambrian ophiolites (exposed west of the ore occurrence), the wings are composed of Ordovician siltstones, Silurian schists, and siltstones (Figure 1).

The Ulug-Sair horst anticline is 6 km long and the wingspan is 2–3 km. Lilac-gray small-pebble schist conglomerates of the Ordovician (O_{3ad1}) Lower Adyrtash subsuites, which lies in the core of the Ulug-Sair horst anticline, are wrinkled into small folds and contain a large amount of gold-quartz veins. Pebbles in conglomerates with silt cement are represented by semi-rounded, sometimes angular vein quartz and quartzite from 1 to 5 cm. It makes up 50–60 % of the rock bulk. Conglomerates contain interlayers of greenish-gray schist sandstones of 10 to 25 m thick and gravelites of 0.2–1.5 m thick. The conglomerates, by and with a gradual transition, are overlapped with lilac and gray-green siltstones of the Upper Adyrtash sub formation (O_{3ad2}). Siltstones contain greenish-gray, lilac-gray, and red-brown sandstones in the form of interlayers of 0.01 to 5 m thick, as well as packs with a thickness of several tens of meters. The total thickness of the rocks of the Upper Adyrtash Formation is 600–700 m.

The rocks of the Silurian Nizhny Chergak sub formation (S_{1cr1}) are concordantly overlapped with the rocks of the Upper Adyrtash sub formation (O_{3ad2}) and represented by gray, greenish-gray sericite-clay shales with gray-green aleurolites and gray sandstones of 68–140 m and 1.5–20 m thick, respectively.

Igneous rocks are represented by dykes, rhyolite-, granodiorite-porphyries (II phase), and diorites (III phase) of the Bayankol complex (D3 bn) in the Ordovician and Silurian rocks. The fall of dykes is vertical; contacts with host rocks are transversal. Dykes of light gray rhyolite-porphyry are confined to the western flank of the ore occurrence, and gray-green granodiorite-porphyry – to its northern flank. The dykes of greenish-gray diorites are the most widespread.

At the Ulug-Sair, quartz-tourmaline metasomatites, beresites after intrusive and sedimentary rocks, as well as quartz veins and vein zones with gold mineralization are formed (Figure 2). More than 75 gold-quartz veins and several vein zones of the EW strike with a dip close to vertical were identified. The veins are from 15 cm to 2 m thick, and 20–100 m long (less often 200 m); vein zones are 3 to 40 m thick and 20 to 120 m long [2]. The average content of ore minerals in ores does not exceed 3 %.

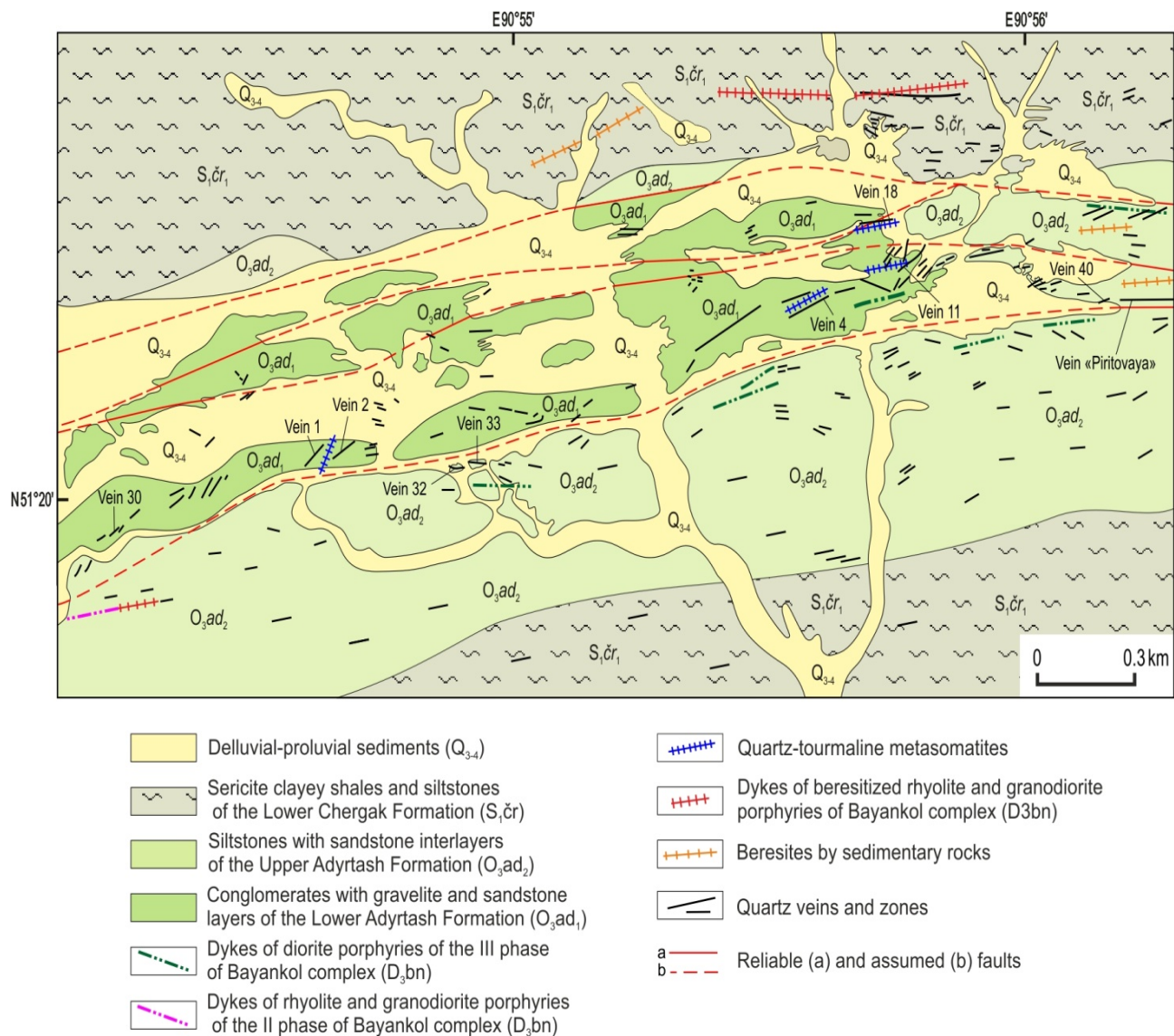


Figure 2. Geological scheme of the Ulug-Sair ore occurrence (according to E.V. Zaykova, V.V. Zaykov [2], modified).

The Au content in quartz veins ranges from 0.2 to 286 ppm; and Ag – from traces up to 300 ppm [2]. The Au resources of the P₂ category, according to the prospecting work of 'Krasnoyarskgeolsjemka' OJSC, are 20 t with average Au content of 2 ppm [9]. The Cu sulfides in quartz veins marked their gold enrichment. In ores, a positive correlation between Au and Cu, B, Ag, Sb, As, Te, Bi, Mn, Ba, Sr, Pb, Mg, Mo, Cd, Zn, and W was identified [10].

2. Materials and Methods

The optical studies were carried out using Olympus BX41, POLAM P-213M, and P-212M microscopes (TuvIENR SB RAS, Kyzyl). The chemical composition of minerals was determined using a MIRA 3 LMU SEM (Tescan Orsay Holding) with INCA Energy

450 + XMax 80 and INCA Wave 500 microanalysis systems (Oxford Instruments Nano-analysis Ltd) (Institute of Geology and Mineralogy SB RAS, Novosibirsk). We used the ontogenic peculiarities (composition, structure, zoning, and induction surfaces) to examine the sequence of mineral formation in ores, including the intersections of early mineral aggregates with the latest ones, as well as the presence of previously formed fragments in late mineral aggregates.

The Au and Au-Ag minerals (cubic solid solutions) are described using the terminology from [11,12]: native gold (1000–700‰: ultrahigh-fineness (1000–950‰), high-fineness (950–900‰), medium-fineness (900–800‰), and low-fineness (800–700‰)), electrum (700–300‰), kustelite (300–100‰), and Au-bearing silver with a fineness of <100‰.

The fluid inclusion study was carried out in the thermobarogeochemistry laboratory of the South Urals State University (Miass) and the CCE "Multielement and Isotope Studies" SB RAS (Novosibirsk) using the Linkam TMS-600 stage equipped with LinkSystem 32 DV-NC and optical Olympus BX51 microscope. The eutectic temperatures of fluid inclusions are interpreted using [13,14]. The salinity of inclusions was determined by the final melting temperatures according to [15]. The homogenization temperatures are the first temperature of the process of mineral formation [16]. The results were processed in Statistica 12.

PT for mineral assemblages is also determined using geothermometers, geofugometers, and mineral parageneses. The stability areas of the main ore minerals in the fS_2 – fTe_2 и fS_2 – fSe_2 coordinates are examined from [17–20].

The gas composition of fluid inclusions was detected by Raman spectroscopy on a Ramanor U-1000 spectrometer with a Horiba DU420E-OE-323 detector (Jobin Yvon) and a Millennia Pro laser from Spectra-Physics (analyst A.A. Redina, Institute of Geology and Mineralogy SB RAS, Novosibirsk). The pressure of fluid inclusions was calculated using the FLINCOR program using CO_2 homogenization temperatures [21].

The bulk analysis of the element composition of fluid was carried out according to [22]. The gases from inclusions (H_2O , CO_2 , and CH_4) were analyzed by an Agilent 6890 gas chromatograph. The content of anions in the aqueous extract was determined by ion chromatography (Color-3000); the cations and trace elements – by the ICP MS analysis (Elan-6100). The NSO_3 content was calculated on the balance sheet. To exclude the influence of the matrix, a repeated ("single") extraction was carried out, the analysis of which was subtracted from the first one.

The oxygen isotopic composition was analyzed on Isoprime mass spectrometer using the internal AQS standard (Akita Quartz Standard) at the University of Akita, Japan (analysts H. Kawaraya and O. Matsubaya). The values of $\delta^{18}O$ are in ppm (‰) relative to the SMOW standard.

The isotopic composition of sulfur in sulfides was analyzed at the Multi-element and Isotope Research Center SB RAS using a Finnigan MAT Delta gas mass spectrometer in dual-inlet mode (analysts V.N. Reutsky and M.N. Kolbasova, Novosibirsk). The measurement control was provided by the samples with standard isotopic composition in the range $\delta^{34}S$ from -15.1 to +21.8 ‰ relative to troilite from Canyon Diablo (CDT), including international ones: NBS-123 ($\delta^{34}S$ = +17.44) and NBS-127 ($\delta^{34}S$ = +21.8). The reproducibility of the values of $\delta^{34}S$, including sample preparation, is not more than 0.1 ‰ (2 σ). The values of $\delta^{34}S$ (‰) are given relative to the CDT standard.

3. Results

3.1. Mineral composition of ores

Pre-ore quartz-tourmaline metasomatites are replaced by the gold-sulfide-quartz mineralization in beresites within the Ulug-Sair ore occurrence. We identified based on the previous works [2] and our data that the earliest high-temperature quartz-tourmaline

stage includes two substages: tourmaline and tourmaline-quartz. Gold-sulfide-quartz mineralization includes 7 substages: pre-ore beresite and pyrite-quartz; ore gold-sulfide-quartz (I) and gold-telluride-sulfide-quartz (II); and post-ore chlorite-quartz, carbonate-quartz, and chlorite-hematite-quartz substages. Limonite, malachite, azurite, goethite, scorodite, cuprite, cerussite, bismutite, iodargyrite, and chlorargyrite are secondary and formed in the weathering crust (Figure 3).

Minerals	Stage									Supergene
	Quartz-tourmaline		Hydrothermal gold-sulfide-quartz							
	Substage									
	1	2	1	2	3	4	5	6	7	
Quartz										
Tourmaline										
Scheelite										
Rutile										
Chlorite										
Magnetite										
Pyrite										
Calcite										
Dolomite										
Siderite										
Ankerite										
Sericite										
Albite										
Chalcopyrite										
Galena										
Native gold										
Electrum										
Bornite										
Petzite										
Hessite										
Kawazulite										
Volynskite										
Wittichenite										
Fischesserite										
Tennantite										
Tetrahedrite										
Sphalerite										
Naumannite										
Tiemannite										
Native Bi										
Baryte										
Native Cu										
Hematite										
Covellite										
Chalcocite										
Malachite										
Azurite										
Other hypergene										

Figure 3. The sequence of mineral formation at the Ulug-Sair ore occurrence.

Quartz-tourmaline metasomatites and veins are widespread throughout the entire ore occurrence (Figure 4a–b). They contain rutile (including W-containing), F-apatite, scheelite, and pyrite. They are metasomatic rocks formed after Ordovician aleurolites and

conglomerates up to 5–7 m thick and composed of light green needle-prismatic tourmaline.

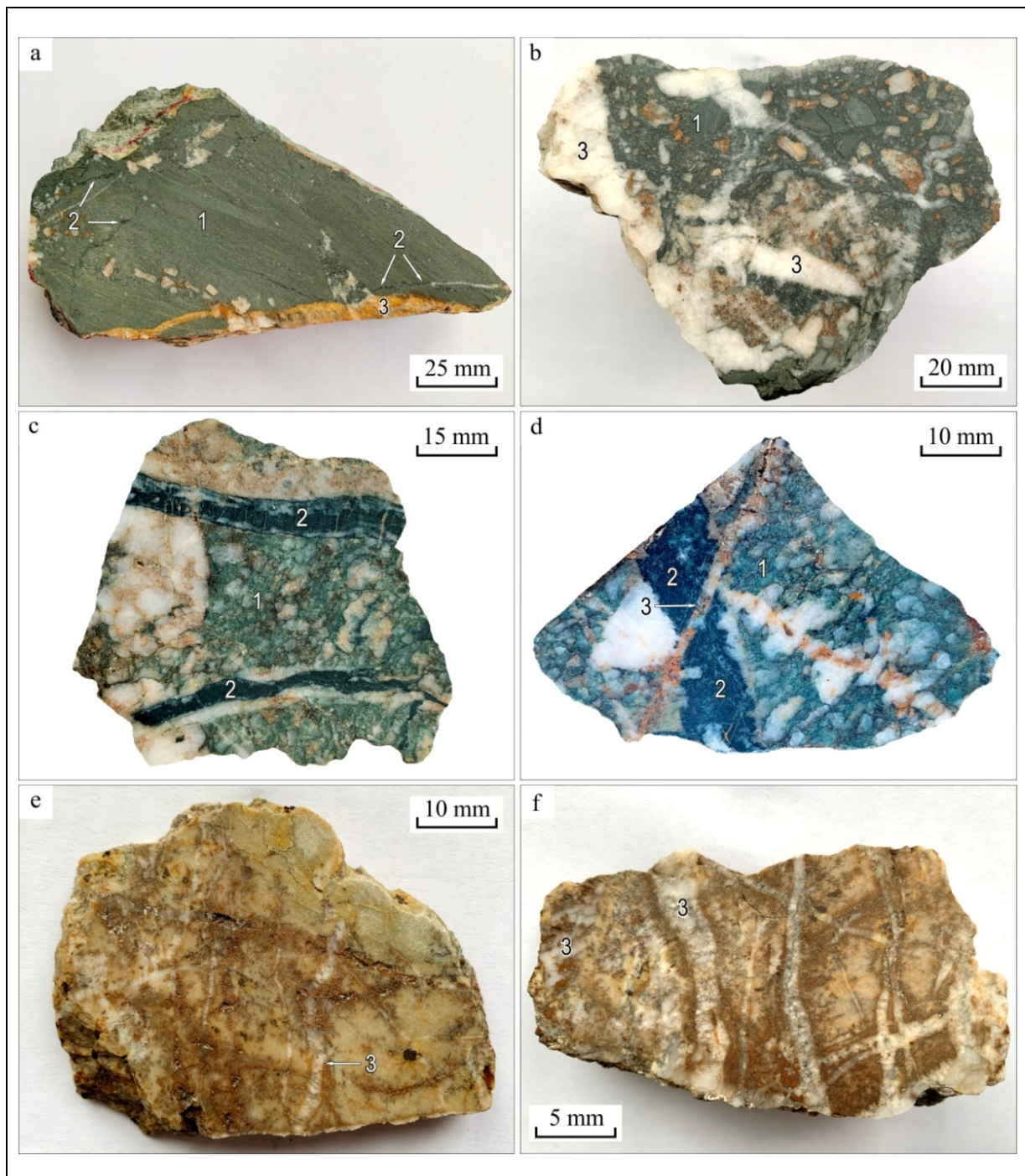


Figure 4. Metasomatites of the Ulug-Sair ore occurrence: (a) quartz-tourmaline metasomatites (1) with tourmaline-quartz (2) and gold-sulfide-quartz (3) veins of the latest substages; (b) brecciated fragments of quartz-tourmaline metasomatites (1) in gold-sulfide-quartz mineral aggregates (3); (c–d) ore-bearing conglomerates (1) with tourmaline-quartz (2) and sulfide-quartz (3) veins; (e–f) beresite with gold-sulfide-quartz veins after c) rhyolite porphyry (3); d) quartz sandstone (3).

The chemical composition of tourmaline from quartz-tourmaline metasomatites and veins shows that they are related to the intermediate members of the dravite-magnesio-foitite series, where the magnesio-foitite component predominates (X_{Fe} 0.80–1.38 cf.). The content of Fe^{3+} 0.00–0.15 cf.; $\text{Fe}^{3+}/\text{Fe}_{\text{total}}$ = 0.00–0.11; Fe^{2+} and Fe^{3+} are

calculated from the charge balance. The X_{Fe} increases from early to latest one (from 0.80–0.91 up to 0.92–1.28 cf). Also, an isomorphism of $Mg,(Al) \rightarrow Fe$ is in tourmaline. The ratio Fe^{3+}/Fe_{total} is low and ranged between 0.00 and 0.08 [10].

Two types of beresites formed after intrusive and sedimentary rocks, respectively, were formed (Figure 4c–d). In both cases, hydrothermal-metasomatic alteration is presented by silicification, sericitization, pyritization, and carbonatization of rocks. The relics of primary rocks are often in beresites: on the first way, rhyolite and granodiorite porphyries, in the second, sandstones, siltstones, and conglomerates. The alterations from beresites to them are gradual. The mineral composition of the beresites of both types is similar – yellow hydrothermally altered, fine and cryptocrystalline rocks with thin pyrite-quartz and carbonate veins, composed of quartz (30–50 %), albite (40–60 %), sericite (up to 5–10 %), calcite, ankerite (up to 10–30 %) and pyrite in the form of well-faceted cubic crystals (1–15 %) with an average size of 1–5 mm, reaching a maximum of 1–3 cm.

The dykes of rhyolite and granodiorite porphyries are intensely beresitized; in diorite dykes; beresitization is insignificant in the form of thin zones (up to 5 cm) in the selvages of thin gold-sulfide-quartz veins. The strike of dykes is sublatitudinal and consistent with the directions of the faults. They have clear contacts, a vertical or close drop to them. The shape of the dykes is plate-like, slightly curved, and sometimes vein-like; 1–3 m thick and 0.3–0.5 km long, less often up to 1 km.

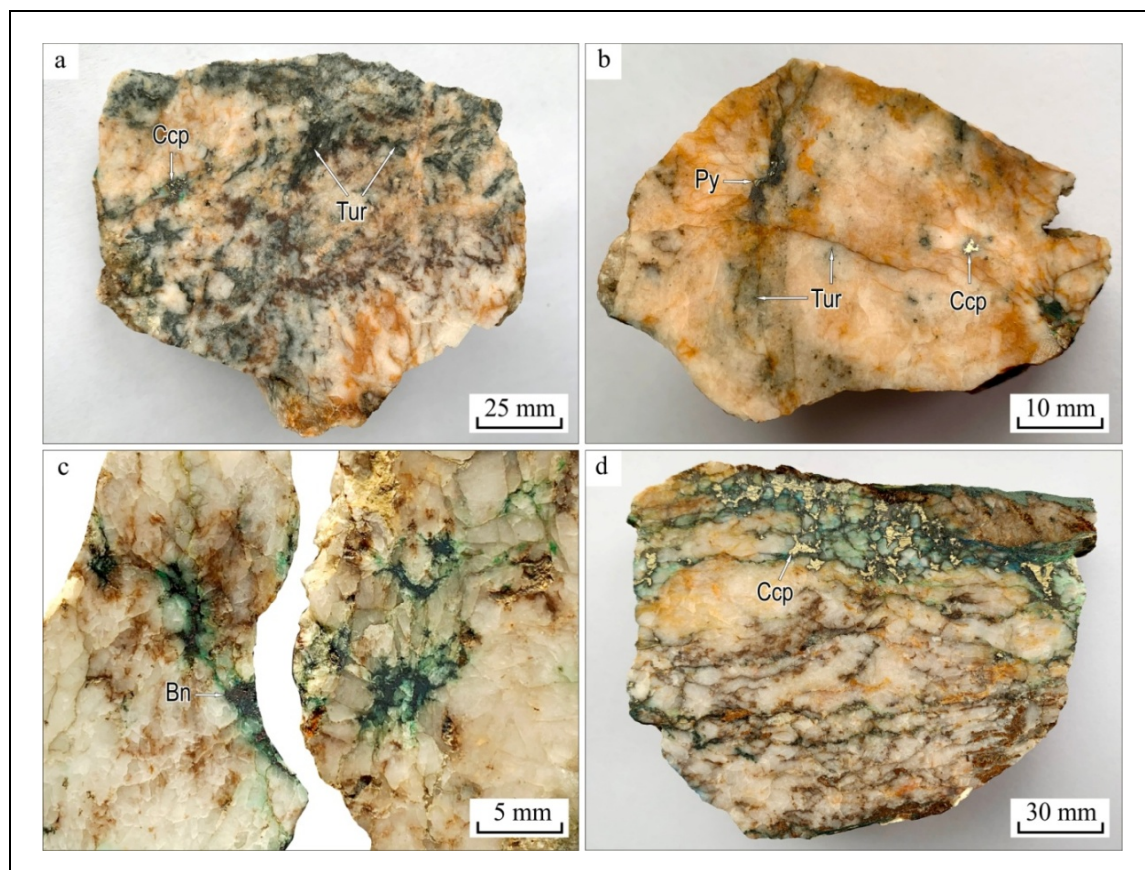


Figure 5. Ore textures of the Ulug-Sair ore occurrence: (a–b) vein-disseminated gold-sulfide-quartz ores with chalcopyrite (Csp), pyrite (Py) and early brecciated tourmaline (Tur); (c–d) vein-disseminated gold-telluride-sulfide-quartz ores with bornite (Bn) and chalcopyrite (Ccp).

Beresites after sandstones, aleurolites, and conglomerates have a "splinous", like a "ponytail" body shape and steep (with angles of incidence of 75–85°), sometimes vertical bedding by the host rocks. They vary in thickness from 0.5 to 2 m and a length of 100 to

150 m. The rhyolite and granodiorite porphyry dykes are hypsometrically lower than the beresitized sedimentary rocks in the beresitized zones along the sedimentary rocks.

Later, gold-sulfide-quartz and gold-telluride-sulfide-quartz veins are formed overlaid quartz-tourmaline metasomatites and beresites and beresitized rocks (Figure 5).

Gold-sulfide-quartz veins (I) are widespread and formed vein zones composed of quartz, chalcopyrite, pyrite, galena, gold, and electrum. The main ore mineral – chalcopyrite – formed phenocrysts and ‘nests’ in quartz. Pyrite is formed cubic and pentagon-dodecahedral crystals. Two mineral assemblages are formed in the gold-sulfide-quartz veins: early gold-galena-pyrite-chalcopyrite-quartz and the latest gold-electrum-pyrite-chalcopyrite-quartz. Their formation is accompanied by deformations of tourmaline metasomatites and followed by brecciation of tourmaline in quartz veins in the form of tourmaline xenoliths or radially radiant aggregates (up to 5 % of the vein) (Figure 5).

Gold in gold-sulfide-quartz veins (I) of lumpy, interstitial, elongated, crystallo-morphic (octahedra, cuboctahedra, crystal intergrowths, and weakly faceted aggregates) and dendritic shapes is disseminated in quartz and chalcopyrite, in intergrowth with pyrite and brecciated tourmaline (Figure 6). Electrum of lumpy-branched and interstitial shapes is detected in intergrowths with chalcopyrite in quartz; and rare.

The chemical composition of gold is (wt %) Au 72.12–96.44, Ag 3.36–27.69, Cu 0.00–0.69, Te 0.00–0.04; electrum – Ag 29.80–38.45, Au 61.55–69.71, Cu 0.00–0.46. Gold and electrum grains are weakly zoned with an increase of Ag amount to the grain periphery by 3–7 wt %.

Gold-telluride-sulfide-quartz veins (II) with disseminated, nest- and vein-disseminated mineralization are up to 2 m thick and up to 28 m long. Gold (up to 0.3 mm) of interstitial, lumpy-branched, and cementation shapes are accompanied by quartz, chalcocite, hessite, petzite, malachite, and Fe hydroxides. Lumpy and lumpy-branched electrum is found in quartz and Fe hydroxides. Gold and electrum are associated with chalcopyrite, bornite, galena (Se up to 0.64 wt %), tennantite, tetrahedrite, sphalerite, kawazulite, fischesserite, naumannite, tiemannite, Se-volynskite, wittichenite, native Bi, and Cu.

The chemical composition of this gold is (wt %) Au 72.56–90.10, Ag 9.47–27.44, Cu 0.00–0.50, and Te 0.00–0.02; electrum – Au 60.37–69.45 and Ag 30.53–40.12. In gold, the Au amount from the center to the edge of the grains is decreased by 2–5 wt % with increasing of Ag.

Small inclusions (1–50 μm) of hessite and petzite are detected in chalcocite; less often bornite and quartz (Figure 7).

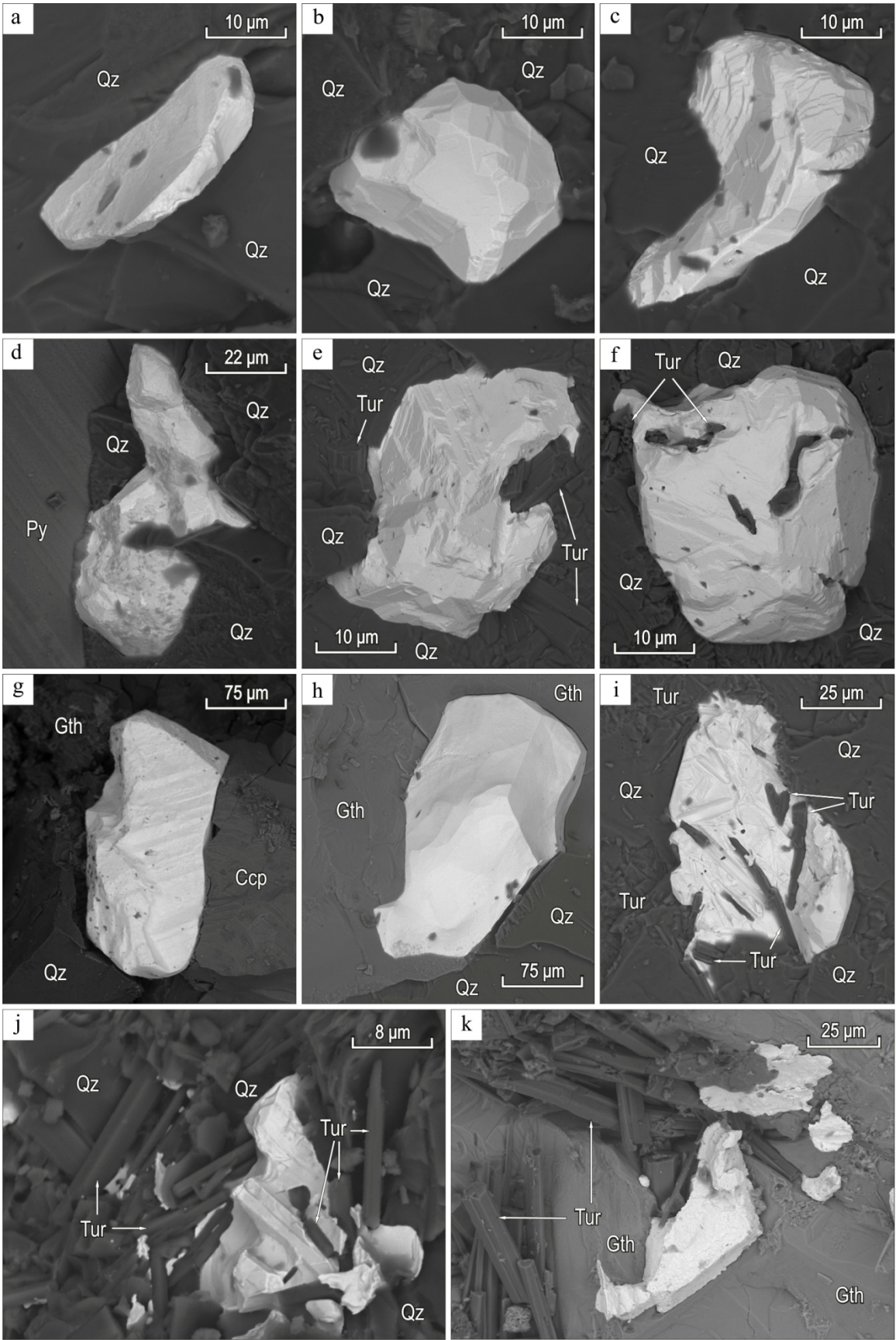


Figure 6. Gold (light) in gold-sulfide-quartz veins with pyrite, chalcopyrite, and goethite (Gth), and early tourmaline (Tur) from quartz-tourmaline metasomatites. BSE photos.

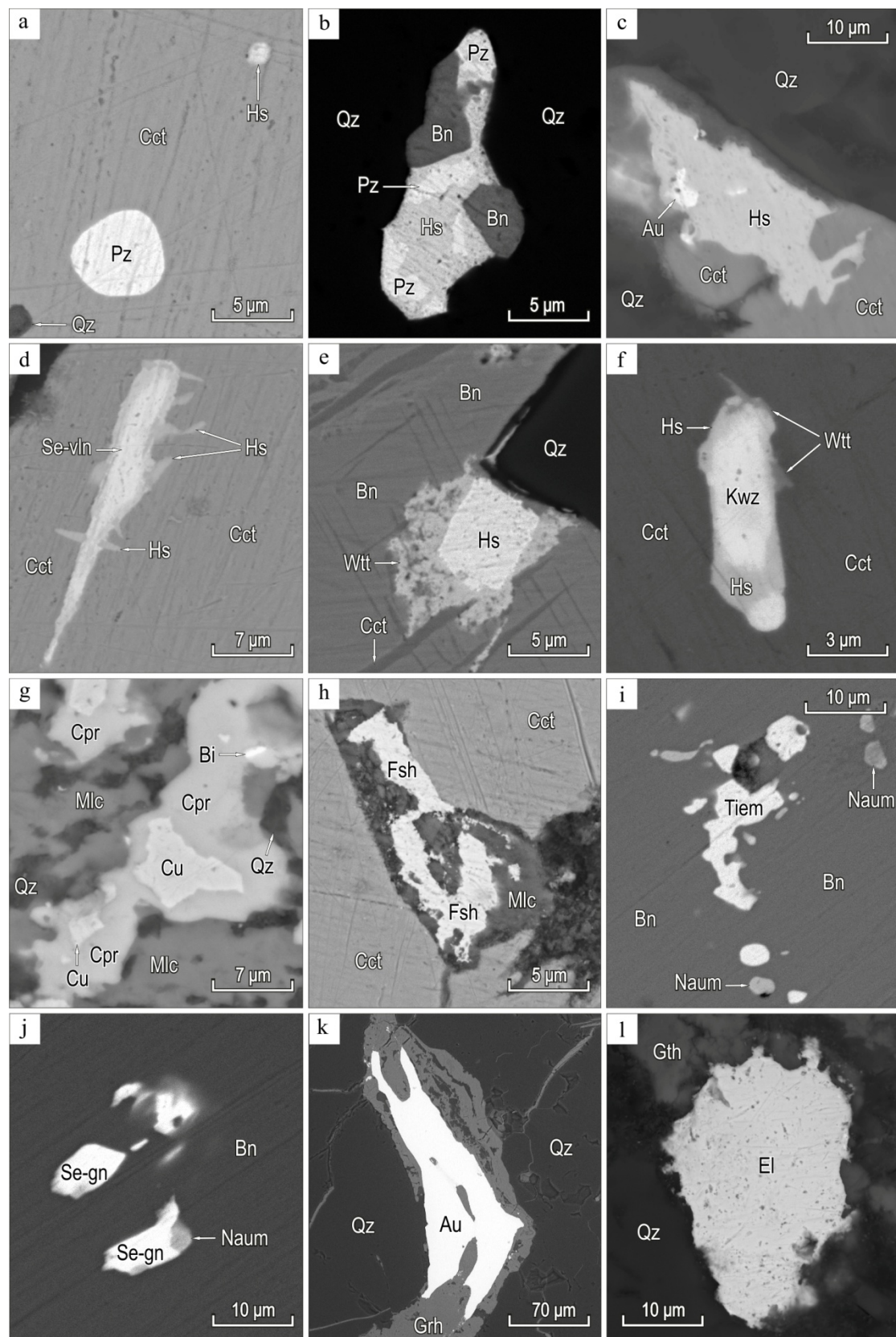


Figure 7. Hessite (Hs), petzite (Pz), Se-volynskite (Vln), wittichenite (Witt), kawazulite (Kwz), native Bi, fischerite (Fsch), naumannite (Naum), tiemannite (Tiem), Se-galena (Se-gn), gold (Au) and electrum (El) in quartz (Qz), bornite (Bn), chalcocite (Cct) with malachite (Mlc), cuprite (Cpr), native copper (Cu) and goethite (Gth) gold-telluride-sulfide-quartz veins. BSE photos.

Hessite formed the intergrowths with petzite, wittichenite, kawazulite, and Se-volynskite, sometimes up to 4 μm thick in the rims. Petzite in intergrowths with hessite often has round and oval shapes. The compositions of Au and Ag tellurides are stoichiometric (Table 1).

Table 1. Chemical composition of petzite, hessite, fischesserite, naumannite, tiemannite, Se-volynskite, kawazulite, and wittichenite, wt %.

Nº	Au	Ag	Bi	Cu	Hg	Pb	Te	Se	S	Total	Crystallochemical formula
Petzite											
1	25.21	41.56	–	–	–	–	32.69	–	–	99.46	$\text{Ag}_{3.00}\text{Au}_{1.00}\text{Te}_{2.00}$
2	25.10	41.55	–	–	–	–	33.15	–	–	99.80	$\text{Ag}_{2.99}\text{Au}_{0.99}\text{Te}_{2.02}$
Hessite											
3	–	63.12	–	–	–	–	36.47	–	–	99.59	$\text{Ag}_{2.02}\text{Te}_{0.98}$
4	–	62.22	–	–	–	–	37.29	–	–	99.51	$\text{Ag}_{1.99}\text{Te}_{1.01}$
5	–	62.29	–	–	–	–	37.47	–	–	99.76	$\text{Ag}_{1.99}\text{Te}_{1.01}$
6	–	62.08	–	–	–	–	37.34	–	–	99.42	$\text{Ag}_{1.99}\text{Te}_{1.01}$
7	–	62.38	–	–	–	–	36.86	–	–	99.24	$\text{Ag}_{2.00}\text{Te}_{1.00}$
8	–	62.25	–	–	–	–	37.55	–	–	99.80	$\text{Ag}_{1.99}\text{Te}_{1.01}$
9	–	61.85	–	–	–	–	37.37	–	–	99.22	$\text{Ag}_{1.99}\text{Te}_{1.01}$
10	–	62.64	–	–	–	–	36.76	–	–	99.40	$\text{Ag}_{2.01}\text{Te}_{0.99}$
Fischesserite											
11	27.95	48.70	–	–	–	–	–	23.07	–	99.72	$\text{Au}_{0.96}\text{Ag}_{3.06}\text{Se}_{1.98}$
12	27.39	48.86	–	–	–	–	–	23.27	–	99.52	$\text{Au}_{0.94}\text{Ag}_{3.06}\text{Se}_{2.00}$
13	26.12	49.16	–	–	–	0.70	0.32	23.36	–	99.66	$(\text{Au}_{0.89}\text{Pb}_{0.03})_{0.91}\text{Ag}_{3.07}(\text{Se}_{1.99}\text{Te}_{0.02})_{2.01}$
14	26.74	48.60	–	–	–	0.98	0.62	22.57	–	99.51	$(\text{Au}_{0.92}\text{Pb}_{0.03})_{0.95}\text{Ag}_{3.07}(\text{Se}_{1.95}\text{Te}_{0.03})_{1.97}$
Naumannite											
15	–	72.69	–	–	–	–	–	26.52	–	99.21	$\text{Ag}_{2.00}\text{Se}_{1.00}$
16	–	74.75	–	–	–	–	–	22.91	1.80	99.46	$\text{Ag}_{2.00}(\text{Se}_{0.84}\text{S}_{0.16})_{1.00}$
Tiemannite											
17	–	–	–	–	72.58	–	–	26.49	0.76	99.83	$\text{Hg}_{1.00}(\text{Se}_{0.93}\text{S}_{0.07})_{1.00}$
18	–	–	–	–	72.81	–	–	25.60	1.08	99.49	$\text{Hg}_{1.01}(\text{Se}_{0.90}\text{S}_{0.09})_{0.99}$
19	–	–	–	–	73.65	–	–	23.92	1.97	99.54	$\text{Hg}_{1.00}(\text{Se}_{0.83}\text{S}_{0.17})_{1.00}$
20	–	–	–	–	74.05	–	–	26.98	2.65	99.76	$\text{Hg}_{0.99}(\text{Se}_{0.79}\text{S}_{0.22})_{1.01}$
21	–	–	–	–	74.76	–	–	21.69	2.93	99.38	$\text{Hg}_{1.01}(\text{Se}_{0.74}\text{S}_{0.25})_{0.99}$
22	–	–	–	–	76.69	–	–	17.81	4.84	99.34	$\text{Hg}_{1.01}(\text{Se}_{0.59}\text{S}_{0.40})_{0.99}$
Se-volynskite											
23	–	19.76	39.23	–	–	–	30.58	9.59	–	99.16	$\text{Ag}_{1.00}\text{Bi}_{1.03}\text{Te}_{1.31}\text{Se}_{0.66}$
24	–	20.37	39.17	–	–	–	29.48	10.45	–	99.47	$\text{Ag}_{1.02}\text{Bi}_{1.01}\text{Te}_{1.25}\text{Se}_{0.72}$
Kawazulite											
25	–	–	54.22	–	–	–	36.01	8.98	–	99.21	$\text{Bi}_{1.98}\text{Te}_{2.15}\text{Se}_{0.87}$
26	–	–	55.37	–	–	–	36.50	7.79	–	99.66	$\text{Bi}_{2.04}\text{Te}_{2.20}\text{Se}_{0.76}$
27	–	–	55.17	–	–	–	35.56	8.74	–	99.51	$\text{Bi}_{2.02}\text{Te}_{2.13}\text{Se}_{0.85}$
28	–	–	55.26	–	–	–	35.55	8.28	–	99.09	$\text{Bi}_{2.04}\text{Te}_{2.15}\text{Se}_{0.81}$
29	–	–	55.48	–	–	–	35.61	8.78	–	99.87	$\text{Bi}_{2.02}\text{Te}_{2.13}\text{Se}_{0.85}$
Wittichenite											
30	–	–	39.93	40.09	–	–	–	–	19.70	99.72	$\text{Cu}_{3.07}\text{Bi}_{0.93}\text{S}_{3.00}$
31	–	–	40.89	39.27	–	–	–	–	19.38	99.54	$\text{Cu}_{3.05}\text{Bi}_{0.97}\text{S}_{2.98}$

Note. Mineral compositions were determined on the MIRA LM electron microscope (analyst N.S. Karmanov, IGM SB RAS). A dash is below detection limits. The formulas of petzite and fischesserite are calculated for 6 at., hessite and naumannite – 3 at., tiemannite – 2 at., Se-volynskite are calculated for 4 at., kawazulite – 5 at., wittichenite – 7 at.

We observed the inclusions of fischesserite, naumannite, tiemannite, kawazulite, wittichenite, and Se-volynskite up to 40 μm and fine intergrowths in quartz, bornite, and chalcocite (see Figure 7).

Fischesserite grains up to 20 μm are in chalcocite. Fischesserite contains small admixtures of Pb (up to 0.98 wt %) and Te (up to 0.62 wt %) (Table 1).

Naumannite and tiemannite up to 25 μm are in bornite, sometimes in the view of their fine intergrowths.

Kawazulite and Se-volynskite (Se up to 10.45 wt %) formed elongated 10–40 μm grains of 3–7 μm wide. Kawazulite formed intergrowths with hessite, less often with Se-wittichenite, which as wittichenite, formed rims (up to 2 μm) around kawazulite.

Au with high and medium fineness prevails; electrum is rare (Fig. 8) at the Ulug-Sair ore occurrence. The average gold fineness is 891 ‰ varied from 602 up to 967 ‰, and the average Au fineness of gold-sulfide-quartz veins (I) is 885 ‰ (615–967 ‰), and gold-telluride-sulfide-quartz veins (II) – 797 ‰ (601–904 ‰).

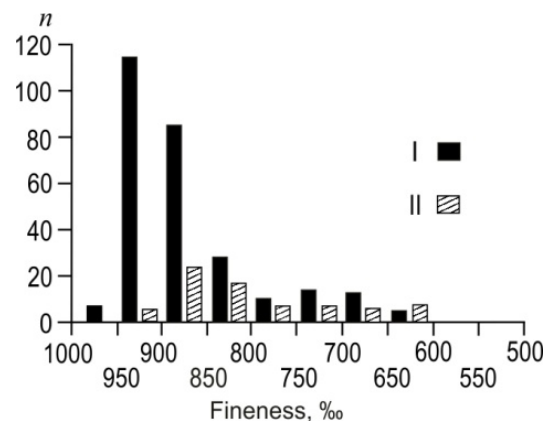


Figure 8. Frequency of native Au fineness of I and II ore subtypes of the Ulug-Sair ore occurrence.

3.2. Fluid inclusion data of the mineral assemblages

The syngenetic fluid inclusions in quartz of mineral assemblages were analyzed to study the PT parameters of gold-quartz veins from the Ulug-Sair ore occurrence (see Figure 2).

Pseudo-secondary biphasic VL fluid inclusions of 10–20 μm in size, oval or isometric shapes with smooth boundaries were studied in quartz of "pre-ore" quartz-tourmaline veins. The eutectic temperatures of the fluid ranged from -33 to -33.9°C and -23 ... -25°C , therefore the fluid contains aqueous NaCl-MgCl₂ and NaCl-KCl. Homogenization temperatures (into liquid) range from 325 to 367 $^{\circ}\text{C}$. The salinity of the fluid is 8–10 wt % NaCl eq.

Primary and pseudo secondary VL fluid inclusions are analyzed in quartz from beresites. Primary VL inclusions are in the form of negative crystals and 8–12 μm in size. The share of the gas phase is up to 30 vol. %. They consist of CO₂, CH₄, and N₂ according to Raman spectroscopy data. The fluid contains Na chloride and K hydrogen carbonate according to the eutectic temperatures are -10 ... -8°C . The single inclusions contain dark-colored mineral phases with elongated shapes are most likely presented by ore minerals. The salinity is low – 0.18–0.71 wt % NaCl eq. (temperatures of the final melting are -0.4 ... -0.1°C). Homogenization temperatures (into the liquid phase) vary from 350 to 400 $^{\circ}\text{C}$. Pseudo secondary fluid inclusions are up to 8 μm and have rounded shapes. The vapor bubble does not exceed 15 vol. %. Homogenization (into the liquid phase) occurs at 200–240 $^{\circ}\text{C}$ (Table 2).

Table 2. Fluid inclusion data of the Ulug-Sair ore occurrence.

Sample	FIA	Phases	T_{hom} , °C	$T_{hom}^{CO_2}$, °C	T_{eut} , °C	$T_{ice\ melt}$, °C	C, wt. %, NaCl eq	D , g/cm ³	P bar	Fluid salt composition
«Pre-gold» quartz-tourmaline veins										
YC-202	ΠB	VL	343–365	–	-33...-33,9	-5,2...-7,1	8,2–10,6	–	–	MgCl ₂ –H ₂ O + NaCl–KCl–H ₂ O
YC-203	ΠB	VL	325–350	–	-23...-25			–	–	MgCl ₂ –H ₂ O + NaCl–KCl–H ₂ O
Berezites										
YCB-8	Π	VL	350–400	–	-10...-8	-0.4...-0.1	0.18–0.71	–	–	chloride + CO ₂ (gas) + CH ₄ (gas) + N (gas)
	ΠB	VL, VLS	200–240	–	–	-3.8...-3	4.9–6.1	–	–	chloride
Gold-sulfide-quartz veins (I)										
YC-18	Π	VL	240–360	–	-21.3...-38.2	-6.3...-3.7	6–9.6	–	–	MgCl ₂ –H ₂ O + NaCl–KCl–H ₂ O
YC-4	Π	VL	290–330	–	–	-8.3...-4.2	6.74–12.51	–	–	chloride + CO ₂ (gas)
	Π	VLC	–	170–180	–	–	1.74–6.45	0.51–0.56	750–900	NaCl–KCl–H ₂ O + MgCl ₂ –H ₂ O + CO ₂ (gas)
YC-4 AΛ-5	ΠB	VL	200–240	–	-21.2...-38.2	-4.4...-2.7	4.5–6.8	–	–	NaCl–KCl–H ₂ O + MgCl ₂ –H ₂ O + CO ₂ (gas)
Gold-telluride-sulfide-quartz veins (II)										
YC-30	Π	VL	270–330	–	-20.9...-32.1	-4.8...-2.5	4.6–7.4	–	–	NaCl–KCl–H ₂ O + MgCl ₂ –H ₂ O
AΛP-Py	ΠB	VL	130–250	–	-21.3...-24.9	-3.5...-6.3	5.0–9.5	–	–	NaCl–Na ₂ SO ₄ –H ₂ O, NaCl–KCl–H ₂ O и NaCl–Na ₂ B ₂ O ₅ –H ₂ O
YC-40	Π	VLC	230–250	+4...+16.8	–	–	–	–	–	chloride
YC-41	Π, ΠB	VL	140–235	–	–	–	–	–	–	chloride + CO ₂ (gas)
YC-33	Π	VL	115–170	–	-22...-38	-2.1...-6.1	3.5–9.3	–	–	NaCl–KCl–H ₂ O + MgCl ₂ –H ₂ O

Note. Fluid inclusion (FIA) associations: Π — primary, ΠB — pseudo secondary, B — secondary. Phases: VLS — three-phase (vapour+liquid+solid); VLC — three-phase (vapour+liquid+CO₂); VL — biphasic (vapour+liquid). T_{hom} — homogenization temperatures, $T_{hom}^{CO_2}$ — CO₂ homogenization temperatures, T_{eut} — eutectic (first melting) temperatures; $T_{ice\ melt}$ — final melting temperatures, d — CO₂ density. A dash — not determined. * — temperatures of decrepitation.

We analyzed primary biphasic VL fluid inclusions. They are 8–15 μm with crystallographic outlines and isometric shapes in quartz from gold-sulfide-quartz vein No. 18 (I) (see Figure 2) with high-grade gold. The eutectic temperatures ranged between -31.2 and -33.7 °C; therefore, the fluid contains NaCl–MgCl₂–H₂O. Homogenization temperatures (into the liquid phase) ranged from 240 to 360 °C. The salinity is 5.5–10 wt % NaCl eq.

Also, we analyzed the primary and pseudo secondary fluid inclusions in quartz from gold-sulfide-quartz vein No. 4 (I) (see Figure 2) with high-grade gold and electrum (Figure 9). Primary VL and VLC inclusions are 12 μm and have elongated shapes. Pseudo-secondary VL inclusions are 3–10 μm and are characterized by an isometric shape.

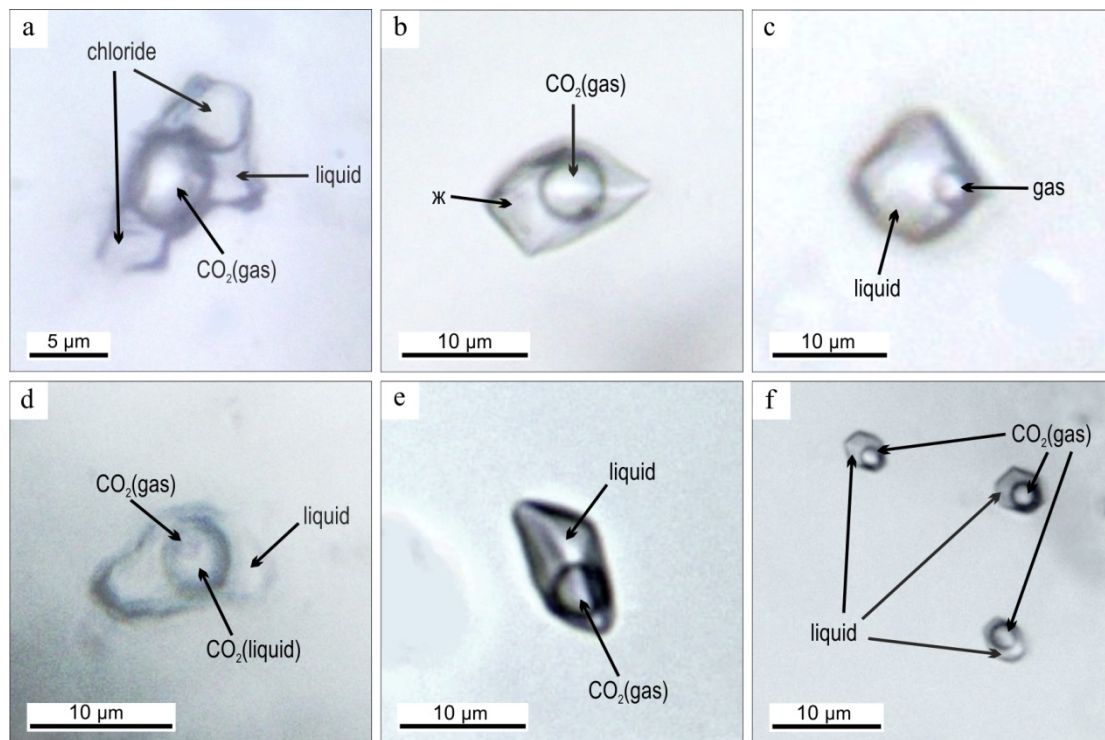


Figure 9. Fluid inclusions in quartz of the Ulug-Sair ore occurrence: (a) VLS primary (sample YC-4); (b) VL pseudo secondary (sample YC-4); (c) VL secondary (sample YC-4); (d) VLC primary (sample YC-40a); (e) VL primary (sample YC-40a); (f) VL primary/pseudo secondary (sample YC-40a). FIA see in Table 2.

We detected CO₂ in VL primary inclusions used of Raman spectroscopy. The final melting temperatures (-8.3 ... -4.2 °C) were provided to estimate the salinity of 6.7–12.5 wt % NaCl eq. The homogenization temperatures (into the liquid phase) are 290–330 °C. The CO₂ primary VLC inclusions are homogenized (into a liquid) at 30.5–31 °C. The total homogenization of these inclusions was 170–180 °C. These values correspond to CO₂ densities ≈ 0.51 – 0.56 g/cm³ and a pressure of 750–900 bars. Pseudo secondary VL inclusions also contain carbon dioxide. Their eutectic temperatures are -21.2 ... -21.7 and -23. ... -23.9 indicated Na-K-chlorides in the fluid. The final melting temperatures are -4 ... -1 °C corresponding to the salinity of 1.7–6.8 wt % NaCl eq. The homogenization temperatures are 200–240 °C.

We examined the primary VL inclusions in quartz from Au-telluride-sulfide-quartz vein No. 30 (II) (see Figure 2) with Au, petzite, hessite, Se-volynskite, kawazulite, and wittichenite. They are 8–15 μ m and isometric angular or tabular shapes with crystallographic outlines. The fractions of gas bubbles are 15–29 vol. %. The eutectic temperatures ranged between -20.9 and -32.1 °C that determined the presence of Na, K, and Mg chlorides in the fluid. Homogenization temperatures (into the liquid) are from 240 to 360 °C. The salinity is 5.5–10 wt % NaCl eq.

We examined primary and pseudo secondary VL inclusions in quartz from Pyritovaya gold-sulfide-quartz vein (II) (see Fig. 2). They are up to 15 μ m, oval and round shapes with large gas bubbles (up to 20–25 vol. %). The eutectic temperatures ranged between -21.3 and -24.9 °C indicated the presence of Na and K chlorides in the fluid. The salinity varies from 5.0 to 9.5 wt % NaCl eq. The homogenization temperatures (into the liquid) are 130–250 °C.

We also analyzed VLC and VL primary and pseudo secondary fluid inclusions of 8–12 μ m in quartz from veins No. 40 and 41 (II) with Au, electrum, acanthite, and hessite. Gas CO₂ bubbles appear only during the cooling. The CO₂ homogenization temperatures (into a liquid) were estimated upon further heating and reached +4 ... +16.8 °C. The temperatures of total homogenization have not been detected since at 230–250 °C as inclu-

sions were decrypted. Primary VL inclusions are 4–10 μm and rounded and elongated. The gas phases of inclusions contain CO_2 according to Raman spectroscopy data. The homogenization of inclusions (into a liquid) occurred at 220–235 $^{\circ}\text{C}$. At a temperature of about 240 $^{\circ}\text{C}$, we observed the decrepitation of inclusions. The primary VL inclusions are 8–12 μm and in the form of negative crystals. The gas phase contains CO_2 . The homogenization temperatures are 140–160 $^{\circ}\text{C}$.

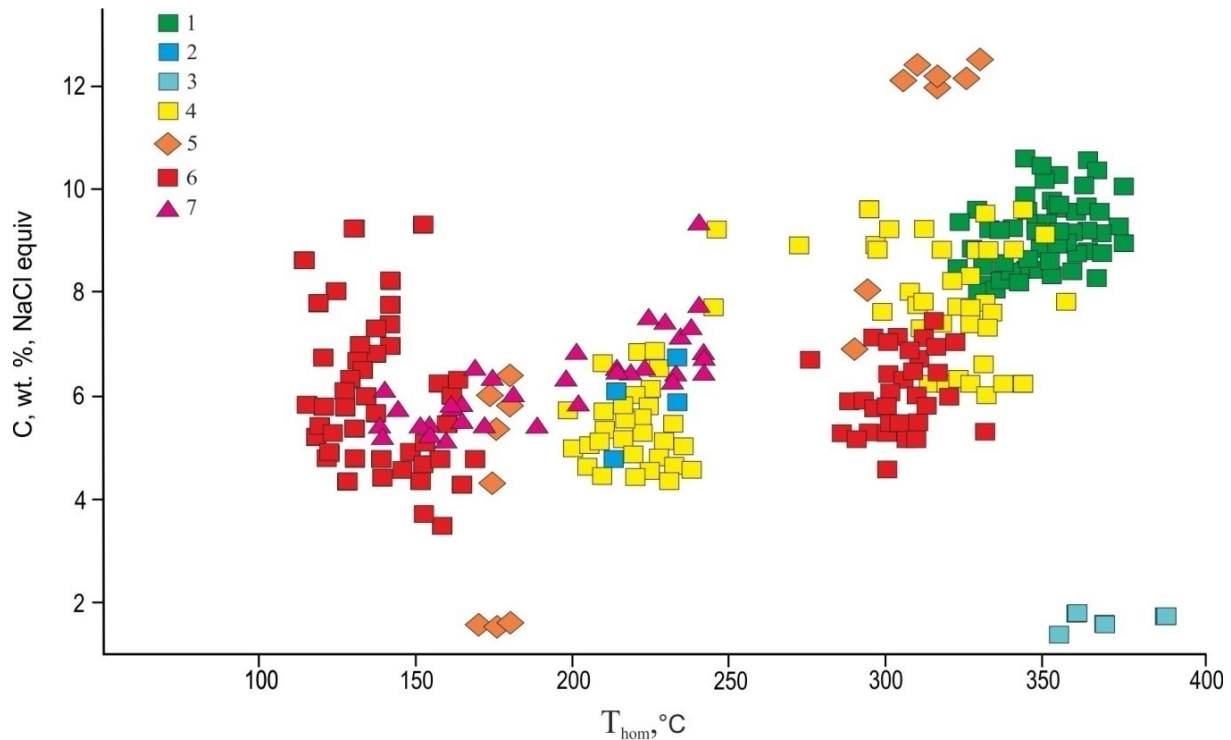


Figure 10. Homogenization temperatures vs salinity plot of fluid inclusions in quartz: 1 – "pre-gold" quartz-tourmaline metasomatites; 2, 3 – berezites (2 – PS, 3 – P); 4, 5 – gold-sulfide-quartz veins (I) (4 – P, 5 – PS); 6, 7 – gold-telluride-sulfide-quartz veins (II) (6 – P, 7 – PS). P – primary, PS – pseudosecondary fluid inclusions.

The primary VL inclusions in quartz from No. 33 veins (II) are up to 15 μm , acute-angled with crystallographic outlines, and large (up to 20 vol. %) gas bubbles. Their eutectic temperatures ranged between -22.7 and -23.9 $^{\circ}\text{C}$ corresponding to a fluid with Na and K chlorides. Large inclusions (more than 20 μm) have lower eutectic temperatures (-29.7 ... -37.2 $^{\circ}\text{C}$) corresponding to $\text{MgCl}_2\text{-H}_2\text{O} \pm \text{FeCl}_2\text{-H}_2\text{O}$ fluid. The salinity is 4–9.5 wt % NaCl eq., and homogenization temperatures (into a liquid) are 110–170 $^{\circ}\text{C}$.

3.3. Geochemical composition of fluids

The bulk analysis of water and gas extracts from fluid inclusions in quartz powder provided the data about the composition of fluids of the Ulug-Sair ore occurrence. In the fluid from inclusions in quartz of Au-sulfide-quartz veins (I) (Table 3), Na (5.73–6.48) prevails among cations (g/kg H_2O); but K (0.14–0.32), Ca (0.01–0.04), and Mg (0.101–0.005) are subordinate. The significant amounts (g/kg H_2O) are detected for CO_2 (41.31–56.24), HCO_3^- (10.04–11.61), Cl $^-$ (2.65–4.31), and CH_4 (0.027–0.048). The significant trace elements are (mg/kg of the fluid): B (74.40–295.5), Ba (53.31–57.85), Sr (17.19–27.67), As (45.42–66.74), Cu (0.03–15.53), Sb (6.27–8.07), Mn (0.40–9.23), Ni (1.26–3.0), and Fe (1.62–2.03).

Table 3. Element composition of fluid inclusions of the Ulug-Sair ore occurrence.

Elements	I		II		Average
	Samples				
	A/1-18-5	AM-13	A/1P-Py	AM-14-2	
Macrocomponents, g/kg H ₂ O					
CO ₂	56.24	41.31	85.58	69.77	63.23
CH ₄	0.05	0.03	0.19	0.04	0.08
Cl ⁻	2.65	4.31	4.51	2.81	3.57
HCO ₃ ⁻	11.61	10.04	35.74	13.13	17.63
Na	5.73	6.48	12.91	6.70	7.96
K	0.32	0.14	1.32	0.16	0.49
Ca	0.01	0.04	1.96	0.00	0.50
Mg	0.1	0.005	0.26	0.00	0.09
Microcomponents, 10 ⁻³ g/kg H ₂ O					
B	295.55	74.40	696.31	129.53	298.95
Ba	57.85	53.31	153.19	637.97	225.58
Cu	15.53	0.03	780.13	0.00	198.92
As	45.42	66.74	208.87	101.43	105.62
Sr	27.67	17.19	82.27	19.24	36.59
Zn	0.00	0.00	58.48	0.00	14.62
Mn	9.23	0.40	42.68	0.02	13.08
Sb	6.27	8.07	27.87	7.37	12.40
Ni	3.00	1.26	20.38	0.47	6.28
Fe	1.62	2.03	9.17	0.00	3.20
Mo	0.73	0.07	10.89	0.47	3.04
W	0.00	0.00	9.07	0.00	2.27
Li	1.59	2.34	3.72	1.24	2.23
Rb	0.40	0.16	2.17	0.33	0.76
Pb	0.00	0.17	1.18	0.00	0.34
Cs	0.10	0.19	0.61	0.25	0.29
Cd	0.08	0.06	0.66	0.01	0.20
Hg	0.00	0.00	0.58	0.00	0.14
Ag	0.00	0.18	0.21	0.00	0.10
Co	0.04	0.00	0.35	0.00	0.10
Au	0.03	0.00	0.17	0.02	0.05
Ge	0.03	0.04	0.07	0.05	0.05
Sn	0.00	0.08	0.00	0.00	0.02
REE	0.07	0.08	0.08	0.38	0.15
Tl	0.01	0.01	0.00	0.00	0.005
Bi	0.00	0.00	0.01	0.00	0.0025
U	0.00	0.00	0.01	0.00	0.0025
Na/K	17.90	46.29	9.78	41.87	28.96
CO ₂ /CH ₄	1171.67	1530	436.63	1701.7	1210
K/Rb	800	875	608	484	645

Note. Analyses were performed in Central Research Institute of Geological Prospecting for Base and Precious Metals (TsNIGRI) (analyst S.G. Kryazhev).

In extracts from fluid inclusions in Au-telluride-sulfide-quartz veins (II) (g/kg H₂O) Na (6.7–12.9) prevails, while K (0.16–1.32), Ca (0.00–1.96), and Mg (0.00–0.25) are subordinate too. The amounts of the volatiles are identified in (g/kg H₂O): CO₂ (69.77–85.58), HCO₃⁻ (13.13–35.74), Cl⁻ (2.81–4.51), and CH₄ (0.041–0.196). The significant trace elements are (mg/kg of fluid): B (129.53–696.3), Ba (153–638), Cu (0.00–780), Sr (19.24–82.27), As (101.4–208.8), Sb (7.37–27.87), Ni (0.47–20.38) и Fe (0.00–9.17), Zn (0.00–58.48), Pb (0.00–1.18), W (0.00–9.07), and Mo (0.47–10.89).

The average chemical composition of the fluid is shown in Figure 10.

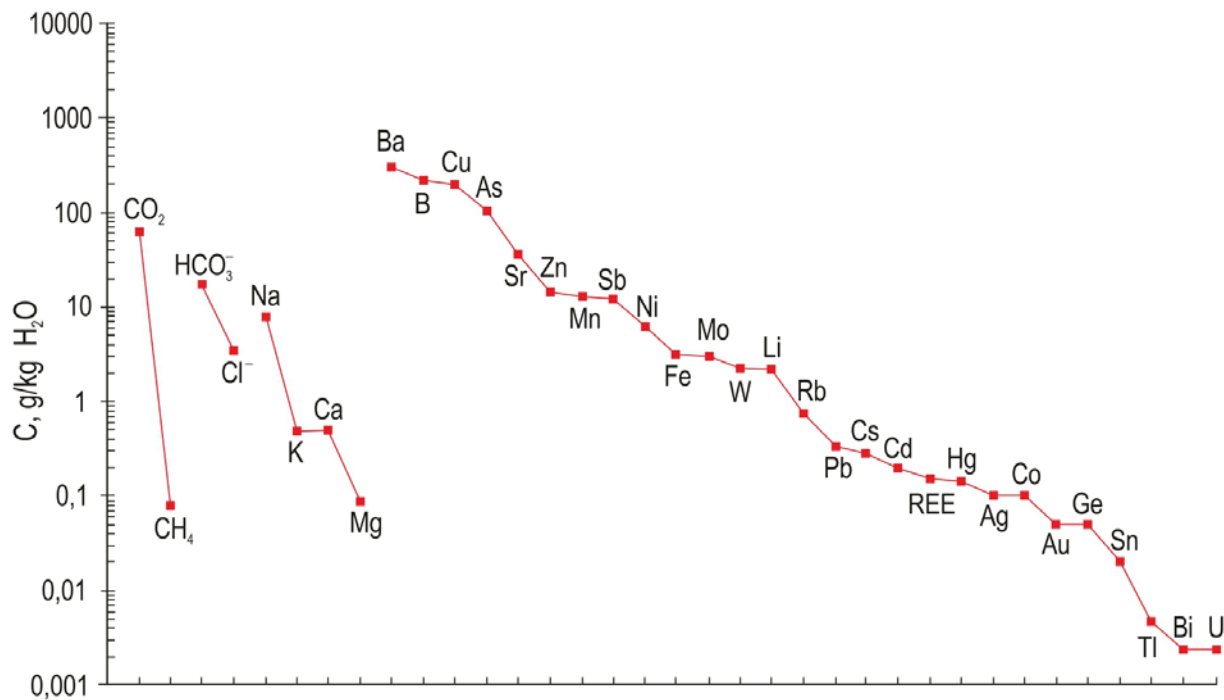


Figure 11. The average chemical composition of the Ulug-Sair ore occurrence fluid.

3.4. Fluid isotopic composition of S and O

The isotopic composition of S in pyrite from I ore substage I is 3.5 ‰, and II – 1.6 ‰, i.e. the values of the sulfur isotopic composition in pyrite are characterized by stable near-zero values ranging between +1.6 and +3.5 ‰.

The isotopic composition of S in fluid ($\delta^{34}\text{S}_{\text{H}_2\text{S}}$) was calculated using the homogenization temperatures of fluid inclusions in quartz by the fractionation equation [22,23] based on the assumption that H_2S prevailed in fluids:

$$\Delta_{\text{pyrite} - \text{H}_2\text{S}} = \delta^{34}\text{S}_{\text{pyrite}} - \delta^{34}\text{S}_{\text{H}_2\text{S}} = 0.4 (10^6/T^2); \quad (1)$$

T is a temperature according to Kelvin.

The values of the isotopic composition $\delta^{34}\text{S}_{\text{fluid}}$ of Au-sulfide-quartz veins (I) vary from +1.8 to +2.5 ‰ ($T = 360\text{--}210^\circ\text{C}$), Au-telluride-sulfide-quartz veins (II) – from -0.7 to +0.3 ‰ ($T = 280\text{--}160^\circ\text{C}$) indicated the magmatic (0 ± 5 ‰) or mantle (0 ± 3 ‰) genesis of S [23,25,26] and suggested that the fluid and ore elements are magmatic, too.

The $\delta^{18}\text{O}$ values for quartz of Au-bearing veins vary from 17.3 to 18.5 ‰, in particular, for Au-sulfide-quartz veins (I) – from 17.2 to 17.5 ‰, and Au-telluride-sulfide-quartz veins (II) from 17.3 to 18.5 ‰ [27].

The isotopic composition of fluid oxygen was calculated using the temperatures of homogenization of fluid inclusions in quartz by the fractionation equation [28,29]:

$$\Delta_{\text{quartz} - \text{H}_2\text{S}} = \delta^{18}\text{O}_{\text{quartz}} - \delta^{18}\text{O}_{\text{H}_2\text{S}} = 3.306 (10^6/T^2) - 2.71; \quad (2)$$

T is a temperature according to Kelvin.

The values of $\delta^{18}\text{O}_{\text{H}_2\text{O}}$ fluid in equilibrium with quartz (I) vary from +7.3 to +11.1 ‰ ($T = 340\text{--}240^\circ\text{C}$), and (II) – from -2.3 to +9.1 ‰ ($T = 250\text{--}130^\circ\text{C}$). The points of the oxygen isotopic composition of the fluid of the early mineral association (I) fall into the range of fluids with both magmatic and metamorphic origins, and the latest mineral association (I) indicates a mixture of ore-bearing magmatic fluids and meteoric water (Figure 12).

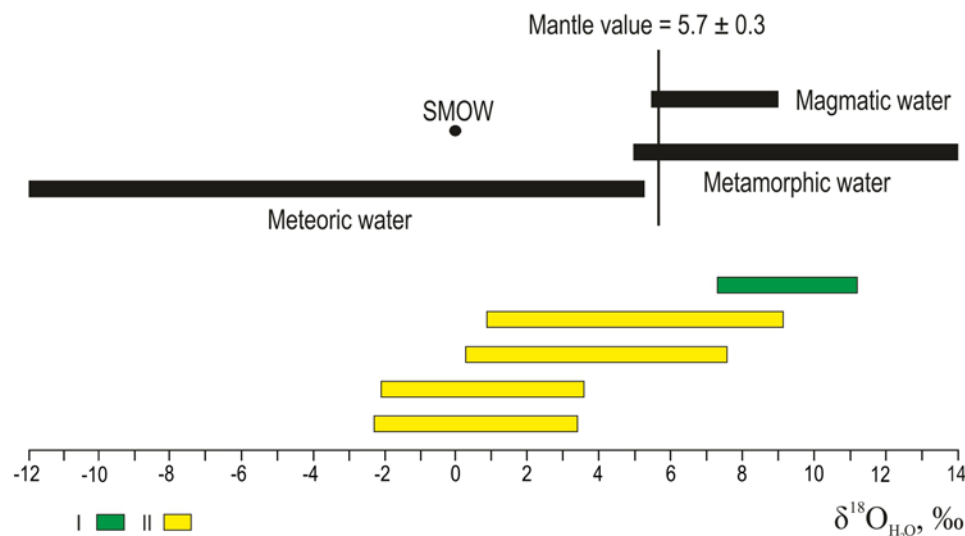


Figure 12. Oxygen isotopic composition of the fluid of I and II ore substages of the Ulug-Sair ore occurrence.

The isotopic values of the fluid oxygen (from -2.3 to +9.1 ‰) of the II ore substage indicated the mixing of the mineral-forming magmatic fluid with meteoric water (see Fig. 12). Note that the values of $\delta^{34}\text{S}_{\text{H}_2\text{S}}$ fluid from -0.7 to +2.5 ‰ suggested the magmatic source of sulfur (0 ± 3 ‰).

4. Discussion

Gold mineralization at the Ulug-Sair ore occurrence is confined to dykes of the rhyolite- and granodiorite-porphyry of the Bayankol complex (D₃) and accompanied by beresitization of intrusive and sedimentary rocks. It should be noted that the formation of beresites with quartz, pyrite, sericite, ankerite, and siderite for both intrusive (acidic, middle) and sedimentary and volcanic-sedimentary rocks is described for the Mechnikovskoye, Ganeevskoye (South Urals), and Kedrovoye (Transbaikalia) deposits [30–34].

Early high-temperature quartz-tourmaline metasomatites are widespread at the Ulug-Sair ore occurrence. For the nearby Khaak-Sair ore occurrence, they are confined to intrusions of tonalite-porphyry of the Bayankol complex (D₃ bn) [5]. The widespread tourmaline mineralization suggested the presence of the granitoid intrusion on a depth of mesoabyssal facies. The presence of boron is indirectly indicated by data on the salt composition of the fluid according to the thermometry results and ICP-MS. We note that quartz-tourmaline metasomatites of the Berezovsky gold deposit are confined to the Shartash intrusion [35].

The Au-bearing veins on the Ulug-Sair are hosted by not only sedimentary rocks but also overlaid on quartz-tourmaline metasomatites, beresites, and beresitized rocks.

Native gold was formed during two substages of mineral formation. The grains of gold in both ore substages are similar in composition, but the II ore substage is characterized by exceptional mineral diversity and the presence of tellurides (petzite Ag_3AuTe_2 , Ag_2Te), selenides (fischesserite Ag_3AuSe_2 , naumannite Ag_2Se , and tiemannite HgSe), and diverse Bi-minerals (kawazulite $\text{Bi}_2\text{Te}_2\text{Se}$, Se-volynskite AgBiTe_2 , native Bi, and wittichenite Cu_3BiS_3).

Raman spectroscopy and fluid inclusions data demonstrated that ‘pre-gold’ metasomatites and Au-bearing veins and metasomatites of the Ulug-Sair ore occurrence were formed by carbon dioxide-water-chloride ($\text{Na}+\text{K}+\text{Mg}$) fluid. The fluid oxidation ratio ($\text{CO}_2/\text{CO}_2+\text{CH}_4$) during a mineral formation is stable and ranged between 1.04 and 1.12. The presence of selenides and Se-bearing minerals indicates the oxidizing fluid, because

according to [36] the presence of selenides in ores indicates a high oxidation potential in the ore formation environment.

Fluid inclusions data resulted that the beresites were formed due to methane-carbon dioxide-water-chloride low-medium-salt fluid with a concentration of 0.18–6.1 wt % NaCl eq. at temperatures of at least 200–400 °C that correspond to the temperatures of the formation of mesothermal gold deposits, including large deposits of the Urals associated with beresites and listvenites [37–39].

Fluid salinity when forming the early gold-sulfide-quartz veins ranged from 12.5 to 1.7 wt % NaCl eq and temperatures of 200–360 °C. These data are consistent with previous studies of fluid inclusions in quartz of such veins [40], who found that they were deposited at temperatures of 250–370 °C from fluid with salinities of 4–10 wt % NaCl eq. and pressures of 0.9–1 kbar. The pressure during the formation of early gold-sulfide-quartz veins was ~ 0.75–1.0 kbar (~2.3–3 km) according to our results on three-phase inclusions.

Mineral paragenesis of Au-sulfide-quartz veins marked that ore formation occurred with high fugacity (f) sulfide sulfur $\lg f(S_2) = 10^{-14.3} - 10^{-8.6}$ ($T = 250$ °C) [17,18].

The latest gold-telluride-sulfide-quartz veins (II) were formed by fluid with a salinity of 9.5–3.5 wt % NaCl eq. with wide variations of temperatures – 110–330 °C.

We note that the parameters of the formation of Ag_3AuTe_2 – Ag_2Te –Au parageneses ranged between 145 and 280 °C, at $f(Te_2) = 10^{-18} - 10^{-10}$ [41]. The paragenesis of tellurides, sulfides, and selenides in gold-telluride-sulfide-quartz veins was formed at fTe_2 from 10^{-21} to 10^{-9} , $fS_2 - 10^{-25} - 10^{-9}$, $fSe_2 - 10^{-21} - 10^{-12}$ at 200 °C [17–20].

ICP-MS data showed that hydrocarbonate with a concentration significantly higher than chlorine prevailed among the anions in the fluid. The fluid within the cations is most enriched in Na with impurities of Ca, K, and Mg; and it can be attributed to the hydrocarbonate-chloride-sodium type that does not contradict the fluid inclusion data. The elevated amounts of Ca and Mg (Mn) and the bicarbonate ion are expressed in the formation of carbonates in the host beresites. The fluid composition with ore elements (B, As, Sb, Cu, Fe Ag, Ba) reflects the composition of the gold-bearing mineral assemblages.

The isotopic composition of $\delta^{34}S_{fluid}$ (from -0.7 to +2.5 ‰) indicates the juvenile or magmatic origin of ore elements. The isotopic composition of $\delta^{34}O_{fluid}$ indicates that, during the early substages, the formation of ore occurrence involved a fluid of magmatic or metamorphic origin (from +7.3 to +11.1 ‰), and during the latest substages, it mixed with meteoric waters (from -1.8 to +9.1 ‰). According to [36], Se-containing minerals in ores indicate a high oxidation rate of the ore formation environment that could be caused by a mixing ore-bearing fluid with highly aerated meteoric waters.

The fluid salinity up to 12.51 wt % NaCl eq. and complicated composition with Na, K, Mg chlorides, Na hydrocarbons, and borates indicate the magmatic genesis of fluid. The elevated amounts of boron in fluid and the presence of “magnaphile” elements (W, Sb, Mo, Sb) also confirm the magmatic genesis of the fluid [42–44].

The salinity reduction during the mineral formation from 12.5 to 1.74 wt % NaCl eq. may be caused by a dilution of igneous fluid with increased salinity and heated meteoric waters [45].

The involvement of magmatic, metamorphic, or meteoric fluids in ore formation is typical for gold deposits in Russia (Beresovskoye, Kochkar) with the dominant role of magmatic fluid [38].

In terms of mineralogical and geochemical peculiarities and formation conditions the Ulug-Sair ore occurrence is close to gold-bismuth (gold-bismuth-telluride) mineralization type deposits [46,47], which are confined to granitoid intrusions and belong to “granitoid related deposits” [48]. Deposits of this type in Alaska and Yukon (Smith et al., 1999; Hart et al., 2002) are large with reserves of 100 t and more. The gold-bismuth mineral type deposits in Russia are the Pogranichnoye (Eastern Sayany), Ergelyakh, Kurumskoye, Tuguchak, Basaguninskoye, Chuguluk, Nennely, and Galechnoye (North-Eastern Russia) vein deposits, and Levodybinskoye, Teutedzhak (North-Eastern

Russia) stockwork deposits confined to the apical near-contact zones of granitoid plutons or their marginal near-contact fracture zones [46,49–52]. It will be noted that deposits of this type are formed mainly from magmatic fluids with the involvement of metamorphic and meteoritic sources [52–56]. They are formed over a wide range of temperatures (437–155 °C, mainly at 400–250 °C) and pressures (1700–90 bar) due to aqueous fluids with Na and K chlorides with wide salinity (46.0–1.1 wt %), at fO_2 - fS_2 variations [51,52].

5. Conclusions

The isotopic data suggest that the Ulug-Sair ore-forming fluid was predominantly magmatic sourced from hidden granite pluton, and, it was mixed with meteoric waters at the final substages. The results supported the hypothesis that mineralization is associated with magmatic activity whereby sub-latitudinal tectonic disturbances become a favorable environment for the circulation of hydrothermal fluids generated by dykes and small granitoid intrusions of the Late Devonian Bayankol complex, which are assumed to be deep. This resulted the formation of linear zones with gold-quartz veins in quartz-tourmaline metasomatites, beresites, conglomerates, and aleurolites. The Ulug-Sair ore occurrence is close to deposits of gold-bismuth type belonging to attributable to class of intrusion-related gold deposits.

Author Contributions: Conceptualization, R.V.K.; methodology R.V.K., N.N.A.; expeditionary work R.V.K., A.A.M. and Yu.V.B.; investigation R.V.K., N.N.A., and Ch.O. K.; visualization R.V.K. and S.N.V.; writing-original draft preparation R.V.K. and N.N.A.; writing—review and editing R.V.K., N.N.A., S.N.V., and A.A.M. All authors have read and agreed to the published version of the manuscript.

Funding: This research was funded by the Ministry of Science and Higher Education of Russian Federation (State Contracts no. 0307-2021-0002 and AAAA-A19-119061790049-3).

Data Availability Statement: Data is contained within the article or supplementary material.

Acknowledgments: The authors are grateful to Anna A. Redina, Ilya R. Prokopyev for their analytical works. The authors are deeply grateful to the reviewers for allowing us to make this paper better.

Conflicts of Interest: The authors declare no conflict of interest.

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