### Fate of β-Carotene within Loaded Delivery Systems in Food: State of Knowledge

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## Short-title: β-Carotene fate in loaded delivery systems

#### **Abstract**

Accruing evidence on the influence of β-carotene regarding the prevention of several chronic diseases - in addition to its well-acknowledged role in vision has been a strong driver for developing alternative delivery systems. Though oral delivery is accepted as the most fitting, mild and safe path for delivering bioactive agents,  $\beta$ -carotene delivery via food items challenges due to its lipophilic nature, poor water-solubility, chemical/photochemical instability and poor oral bioavailability. Nanotechnology has opened new windows for delivering bioactive agents. Their physiochemical characteristics, i.e. small size, high surface area, unique composition, biocompatibility and biodegradability make these nanomaterials an attractive tool for  $\beta$ -carotene delivery. Delivering  $\beta$ -carotene through nanoparticles does not only improve its bioavailability/bioaccumulation in target tissues, but also lessens its sensitivity against environmental factors during processing. Regardless of these benefits, nanocarriers inherit some limitations, such as variations in sensory quality, modification of the food matrix, increasing costs, as well as limited consumer acceptance and regulatory challenges. This research area has been rapidly evolved, with a plethora of innovative nano-engineered materials, including micelles, nano/microemulsion, liposomes, niosomes, solid-lipid nanoparticles and nanostructured lipid carriers. These nano-delivery systems make conventional delivery systems appear archaic and promise better solubilization, protection during processing, improved shelf-life, higher bioavailability as well as controlled and targeted release. This review provides information on the state of knowledge on β-carotene nano-delivery systems adopted for developing functional foods: depicting their classification, composition, preparation methods, challenges, release-and absorption of  $\beta$ -carotene in the GIT and possible risks and future prospects.

**Keywords:** Beta-carotene; bioavailability; delivery system; encapsulation; engineered nanomaterial; SLNs; NLCs

#### **Abbreviations**

Cremophor 40 PEG hydroxylated castor oil

EC Ethylcellulose GA Gum arabic

LCT Long-chain triglyceride

MCT Medium-chain triacylglycerides
NaCMC Sodium carboxymethyl cellulose
OSA N-octenyl succinic anhydride

PC Phosphatidylcholine PEA Phosphatidylethanolamine

PHBV Poly(hydroxybutirate-co-hydroxyvalerate)

Poly(lactic) acid **PLA** PS Phosphatidylserine Polyvinyl alcohol PVA Sodium alginate SA Sodium caseinate SC **SGF** Simulated gastric fluid SIF Simulated intestinal fluid SPI Soy protein isolate SSF Simulated saliva fluid

SSPS Soybean soluble polysaccharides
Tween Polyoxyethylenesorbitan monolaurate

WPC Whey protein concentrate
WPI Whey protein isolate

### 1. Introduction

Vitamin A deficiency is one of the most diagnosed micronutrient deficiency disorders worldwide, especially in developing countries. However, its magnitude becomes more widespread in the vegetarian population [1]. Across the globe, approximately 250 million preschool children are estimated to be affected by vitamin A deficiency [2]. Further, occurrence of disease has intimate relation with low antioxidant load in daily diet. Furthermore, lifestyle (exercise, smoking, drinking and high consumption of meat-based and processed foods), environment (emotional and social stress), and cultural constraints triggers the expression of housekeeping genes to adopting gene to retain the cellular, organ or body homeostasis [3]. Aforesaid stimuli also cause generation of reactive oxygen species (ROS) resulting oxidative homoeostasis imbalance at cellular and tissue thus generating the oxidative stress [4]. Oxidative stress can be defined as a phenomenon triggered by an imbalance between generation and accumulation of ROS. In general, ROS including organic hydro peroxides, hydrogen peroxide, nitric oxide, hydroxyl radicals and superoxide are generated as by-product of oxygen metabolism, despite this, environmental stimuli (UV, pollutants, heavy metals, and xenobiotics (including antiblastic drugs, antiallergic drugs, immunosuppressant drugs) equally contributes ROS production, thus causing oxidative stress [5]. Accruing scientific evidences are accumulating on involvement of oxidative stress in the occurrence of several health complications which are attributed to inactivation of metabolic enzymes and damage vital cellular components, oxidization the nucleic acids, resulting in eye disorders, atherosclerosis, cardiovascular diseases, joint and bone disorders, neurological diseases (amyotrophic lateral sclerosis, Parkinson's disease and Alzheimer's disease) and different organ misfunctioning including lung, kidney, liver and reproductive system [6]. ROS are primarily generated in mitochondria under both pathological as well as

physiological conditions [7]. Cells activate an antioxidant defensive system which primarily include enzymatic components such as superoxide dismutase, glutathione peroxidase, catalase to minimize the oxidative stress cell [8].

## 1.1. Oxidative stress and antioxidant

ROS generation is attributed to both nonenzymatic and enzymatic and reactions. Enzymatic processes which have intricate involvement in respiratory chain, phagoscytosis, prostaglandins biosynthesis, and cytochrome P450 system are responsible ROS generation. Superoxide radical, produced as result of enzymatic acid of NADPH oxidase, peroxidases and xanthine oxidase initiate the chain reaction for ROS including hydrogen peroxide, hydroxyl radical, peroxynitrite, hypochlorous acid and so on [9]. Hydroxyl radical (OH $\bullet$ ) is considered as the most reactive among all ROS in vivo and is produced as a result of catalysis of H<sub>2</sub>O<sub>2</sub> in presence Fe<sup>2+</sup> or Cu<sup>+</sup> (Fenton reactions).

In addition, some nonenzymatic process also contributes to ROS generation especially when oxygen is either exposed to ionizing radiations or react with organic compounds. ROS are produced due to exogenous and endogenous sources. Exogenous source of ROS includes inflammation, immune cell activation, infection, ischemia, cancer, mental stress, excessive exercise and aging [4,10]. Exogeneous ROS generation rely on exposure to radiation, heavy metals [11], environmental pollutants [12], certain drugs (bleomycin, cyclosporine, gentamycin, tacrolimus) [13], toxic chemical and solvents [13], food processing (used oil and fat and smoked meat) [14], cigarette smoking and alcohol consumption [10]. ROS are essential part of several biological processes when they remain at low or moderate concentrations. For instance, these ROS are obligatory for synthesis of some cellular structures which have vital role in the host defense system to contest with pathogens [14,15]. In fact, macrophages synthesize and store ROS to kill pathogenic microbe [16]. The critical role of ROS in the immune system is well recognized as patient unable to product ROS are more prone to pathological infections [17]. In addition, ROS are also integrated with an array of cellular signalling pathways as they play regulatory role in intracellular signalling cascade including endothelial cells, fibroblasts, cardiac myocytes, vascular smooth muscle cells and thyroid tissue. Nitric oxide (NO) is considered as a key cell-to-cell messenger which plays vital role in cell signalling and intricately involve in several processes such as blood flow modulation, thrombosis and normal neural functioning [18]. Nitric oxide also demonstrates close association in nonspecific host defense in eliminating the tumor cells as well as intracellular pathogens [19]. In addition to beneficial effects, ROS also pose several negative impacts by affecting cellular structure including plasma membrane, proteins, lipoprotein, proteins and nucleic acids (deoxyribonucleic acid, DNA; ribonucleic acid, RNA). Since oxidative stress is result of ROS imbalance between its rate of generation and rate of its clearance within the cell [20]. This excess ROS thus cause damage in plasma membrane by lipid peroxidation and form malondialdehyde and conjugated dienes which are cytotoxic and mutagenic in nature. Being a chain reaction cascade, lipid peroxidation spreads very rapidly damaging a significant number of lipids, proteins and nucleic acids hence hampering their functionalities [21]. In summary, ROS impart beneficial effect when they are maintained at low or moderate concentration while they negatively affect several cellular structures at higher concentration.

Human body adopt several strategies to combat the negative effect generated due to oxidative stress depending upon enzymatic (superoxide dismutase, glutathione peroxidase and catalase) or nonenzymatic (L-arginine, glutathione, coenzyme Q10 and lipoic acid) antioxidant molecules. In addition to aforesaid molecules several exogenous antioxidants molecules from animal or plant origin are deliberately incorporated in diet [5].

### 1.2. Mode of action of $\beta$ -carotene against oxidative stress

 $\beta$ -carotene, a key member of the carotenoid family, is recognized as one of most potent antioxidants [22] and the major provitamin A carotenoid available within the human diet. The health benefits of  $\beta$ -carotene are attributed to given biological properties [21] (a) as antioxidants that scavenge and quench ROS of oxidative metabolism, (b) as provitamin A compounds that activate retinol-mediated pathways, (c) as electrophiles that boost endogenous antioxidant systems, (d) by hampering inflammation-related processes mediated by nuclear factor  $\kappa$ -light-chain-enhancer of activated B cells (NF- $\kappa$ B) pathway, and/or (e) by directly bonding nuclear receptors (NRs) and other transcription factors in target cells .

Retinoic acid acts as ligand for the retinoid X receptors (RXRs) and canonical retinoid acid receptors (RARs) that influence the expression of a number of responsive genes having intimate relation with fatty acid, cholesterol,  $Ca^{2+}$  and phosphate homeostasis [23].  $\beta$ -Carotene also demonstrated tumor cell suppression activity and enhancing the intercellular communication at gap junctions [3]. It is believed that consumption of  $\beta$ -carotene may cause low incidence of hepatic oxidative stress and lipid oxidation. The assumption was supported by mice model study where expression of 1207 gene (approximately 4% genes) of total 30855 genes in a hepatic transcriptome was influence when mice was fed with  $\beta$ -carotene as compared to control mice [24]. Remarkably, numerous of the differentially expressed genes had intimate involvement in energy metabolism, lipid metabolism, and mitochondrial redox homeostasis

β-carotene is the main contributor of vitamin A to human being if pre-formed vitamin A intake is insufficient. It acts as a precursor of vitamin A, with the potential to yield two retinal molecules following cleavage by beta-carotene oxygenase 1 in the intestine, as compared to other carotenoids which generally yield only one retinal molecule. Despite it's indispensable role in vision, it may furthermore play a role as a bioactive compound, due to its potential anti-oxidant effects [25], and its interaction with nuclear receptors, mainly RAR/RXR, important for cell differentiation and immunity [26]. These properties have contributed to making β-carotene one of the most investigated biological molecules, both in the academia and the industry. Though its multifunctionality in humans is yet to be fully understood, several epidemiologic studies have demonstrated its relation to a decreased incidence of chronic diseases such as blindness [27], xerophthalmia [28], cancer [29], cardiovascular diseases [30], diabetes [31] and premature death [32] and found to have intricate relation with its antioxidant nature.

## 1.3. Challenges associated with $\beta$ -carotenein food fortification

β-carotene is naturally found in various foods and is also commonly used as a natural pigment in the food, pharmaceutical and cosmetic industry. This lipophilic molecule is characterized by the presence of a polyene structure having 11 conjugated double bonds with 2 β-ionone rings (Figure 1). Under environmental stress (temperature, humidity, pH, ionic strength and radiation), β-carotene may undergo transformation resulting in the formation of different isomers such as 15-cis-β-carotene, 13-cis-β-carotene and 9-cis-β-carotene and several trans-β-carotenes [33,34]. Cis-isomers have bent structures and are likely to be more readily solubilized and adsorbed compared to trans- β-carotene which possesses a linear and rigid structure and has high tendency to crystallize and aggregate as compare to the cis-isomers [35,36]. The unsaturated structure makes β-carotene prone to oxidation, resulting in the loss of its vitamin A functionality. Furthermore, β-carotene is also susceptible to isomerization when confronted with acidic conditions, high-salt, temperature, metal ions, peroxides and radiation during food processing and storage before consumption [36]. In addition, naturally occurring β-carotene is often complexed with protein molecules which

limit its solubility and distribution in the food matrix, as well as its adsorption in human body [37].

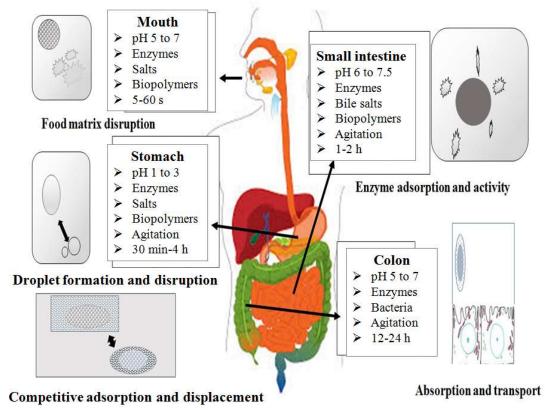


Figure 1: Schematic diagram of the human digestive system and the various physiochemical and physiological processes involved in the digestion and absorption of  $\beta$ -carotene.

Currently,  $\beta$ -carotene is one of the most exploited carotenoids to develop functional foods [38], to formulate pharmaceutical supplements and to prepare cosmetic products. However, food fortification i.e. incorporating  $\beta$ -carotene within functional foods; is recognized as the most natural, appropriate and safe way as compared to other drug administration routes including intravenous, intramuscular and subcutaneous [39]. However, within these functional food products,  $\beta$ -carotene is prone to undergo physico-chemical degradation during the production, processing and storage before food-consumption. These limiting factors makes  $\beta$ -carotene difficult to incorporated into the food matrix, in addition to its low bioavailability within the human gastrointestinal tract and hence significantly strangles its efficacy as a health beneficial plant compound.

Nanotechnology seems a logical solution to address these limiting factors, as it has demonstrated it's potential to encapsulate, protect and delivery bioactive compounds using several delivery systems to improve their physico-chemical stability, solubility, dispersibility and bioavailability upon ingestion [40-44]. Researchers have nano-engineered various kinds of delivery systems, such as microemulsion, liposomes, solid lipid carriers, nanostructured lipid carriers, nanocapsules and nanospheres to encapsulate and deliver bioactive compounds. These delivery systems are capable of improving stability, dispersity and bioavailability of bioactive compounds within the target food matrix. Although several excellent reports have already been published emphasizing the factors affecting the chemical stability of carotenoids [45], encapsulation techniques to protect them against environmental stress [46], production methods to prepare nano-engineered delivery systems [47] and delivery systems to improve

their solubility or bioavailability [48], there is lack of reviews regarding  $\beta$ -carotene delivery systems, in particular with food applications.

The present article aims to contribute to the state of knowledge on the delivery systems used for  $\beta$ -carotene to improve its stability, solubility, dispersibility, bioavailability, as well as the development of functional foods. Before opting designing an oral delivery system for  $\beta$ -carotene, it is paramount to understand its metabolism (digestion and absorption) as well as the factors affecting the physicochemical attributes of delivery system and their health risk and safety issues. Additionally, this review article will aid to better understand the evolution of delivery systems for the encapsulation of  $\beta$ -carotene in food science.

## 2. Methodology

To search the literature, three most popular search engines of food and medical sciences, Google Scholar, Science Direct and PubMed as well as Scopus database were employed with the keywords " $\beta$ -carotene", " $\beta$ -carotene encapsulation", " $\beta$ -carotene delivery system", "engineered nanomaterial and  $\beta$ -carotene", " $\beta$ -carotene bioavailability", "oxidative stress and  $\beta$ -carotene". The time-line search (year) was as follow (a) 1900 - 1990, (b) 1991 - 2000, (c) 2001 - 2010 and (d) 2011 – 2020 in these search engines. After searching, with each keyword in the mentioned time line the first 100 most relevant entries were screened with direct observation. Adopting this method of literature search, nearly 2400 articles were screened and out of which, based on the relevance of the topic nearly 400 articles were summarized in the present review. The articles having food applications were prioritized to conduct the review.

# 3. β-carotene metabolism

The fate of  $\beta$ -carotene in the human gastro-intestinal tract (GIT) is determined by various factors, including the complexity of the ingested food matrix, its release form the food matrix, the transfer of the released molecule to the oil phase, its incorporation into mixed micelles, the entrance route into enterocytes and its incorporation into chylomicrons [49]. In the following, these processes are briefly explained.

#### 3.1. Release of β-carotene from the food matrix

Release of  $\beta$ -carotene from the food matrix is a multistage process, which begins by mastication in the mouth, followed by enzymatic and physiochemical process in the stomach and the small intestine [49]. The release of  $\beta$ -carotene begins the physical disruption of ingested food particles in the buccal cavity of GIT to make  $\beta$ -carotene bioaccessible for absorption.

Bioaccessibility is defined as the quantity or extent of  $\beta$ -carotene which is released from food matrices in gastrointestinal tract and remains bioavailable for absorption in intestine.

Bioaccessibility = 
$$B_r \times 100/B_t - B_e$$

Where, Br represents quantity of  $\beta$ -carotene released in GIT fluid in consequence of food matrices digestion, Bt: Total quantity of  $\beta$ -carotene existing in the food matrices, Be:  $\beta$ -carotene secreted in the duodenal compartment along with bile salt.

The complexity of the food matrix has a great impact on the bioavailability as well as bioavailability of  $\beta$ -carotene, as its release from food the matrix is the major limiting factor for its bioavailability [37,49-52].

The bioavailability of lipophilic compounds such as  $\beta$ -carotene can be defined as the part of the ingested  $\beta$ -carotene that eventually is recovered in the systemic (blood) circulation

as an active form. Only then, the  $\beta$ -carotene will be available to allocate to the target tissues and organs where it can exert their beneficial health effects. For ingested  $\beta$ -carotene, there are several limitations, which limits the proportion that arrives in the systemic circulation in its native form, e.g. chemical instability during the digestion process, poor solubility in the gastrointestinal tract (GIT), slow uptake from the GIT, cleavage by BCO1 in the enterocyte (producing 2 molecules of retinal) [53], and first-pass metabolism (Figure 4). The oral bioavailability (F) of encapsulated  $\beta$ -carotene in delivery systems can be determined by the following equation

$$F = F_B \times F_A \times F_M$$

 $F=F_B\times F_A\times F_M$  Here,  $F_B$  is the fraction of consumed  $\beta\text{-carotene}$  that survived through the upper GIT and released from the food matrix/delivery system into the GIT, thus becoming bioaccessible for uptake by brush-bordered enterocytes. F<sub>A</sub> is the fraction of the bioaccessible β-carotene, which is eventually absorbed by the enterocytes and then reaches the portal blood or, rather the lymph (and thus the systemic circulation). F<sub>M</sub> is the proportion of absorbed β-carotene which is preserved in its active form after first-pass metabolism in the GIT and the liver (and any other forms of metabolism or breakdown).

Naturally, β-carotene is present in different physical forms within the chloroplast and chromoplast. In the chromoplasts, β-carotene is available either in crystalline form (e.g. in carrots and tomatos) or in oil droplets (mango and papaya). It was noticed that bioaccessibility of β-carotene dissolved in oil droplets (10.1% for mango and 5.3% for papaya) is higher as compared to crystalline form (3.1% for tomato and 0.5% carrot) [54].

The release equally depends on the degree of structural disruption of the food matrix, which can be enhanced by subjecting various food processing techniques (mechanical and thermal) before ingestion. It is believed that mechanical processing (homogenization, cutting, crushing and pureeing) may significantly improve bioavailability as it reduces food particle size, hence offering greater surface to volume ratio for digestive fluids and enzymes to act upon, resulting in a higher release of β-carotene [49]. An 18% higher bioavailability (in vitro) in homogenized carrot as compared to chopped raw carrot supports this assumption [55]. Similarly, a two-fold higher bioaccessibility (in vitro) was witnessed for a 125 nm particle size as compared to a particle size of 126–160 µm [55].

Thermal treatments are also considered a good option for improving the bioavailability, as it facilitates softening and disintegration of plant tissues and denaturation of β-carotene-protein complexes. Rock and team observed 3-fold increases in β-carotene serum level when spinach was incubated for 40 min at 120°C after canning and sterilization [56]. Similarly, commercially available carrot puree (subjected retort processing after cooking) has shown a higher bioavailability (in vivo) as compared to the carrot puree meshed in a grinder after 40 min of boiling [57]. Additionally, carrot finely peeled and chopped after boiling at 100°C for 15 min was found to be more effective in raising the β-carotene serum level as compared to raw carrot [58]. Differences between the bioaccessibility observations from in vitro and in vivo bioavailability studies, such as higher bioavailability found in vivo, may be attributed to differences in food preparation methods and gastro-intestinal simulation methods chosen, plus the inherent limitations of all *in vitro* methods [36].

Comparing various treatments, the thermal treatments were found to be often more effective in improving the bioavailability of β-carotene versus mechanical processing [59]. It is also assumed the simultaneous application of thermal and mechanical processing may offer better release of β-carotene from food matrix. This assumption was supported when researchers observed a higher increase β-carotene serum level when fed with food subjected to homogenization and thermal treatment as compared to thermal processed or mechanical process food alone [60]. From the above observation, it can be postulated that the bioavailability of β-carotene is a function of particle size as well as of thermal processing.

The improved bioavailability of  $\beta$ -carotene after simultaneous application of thermal and mechanical processes could be attributed to a reduction of particle size due to homogenization and degradation of  $\beta$ -carotene-protein complexes by thermal processing [37,60].

# 3.2. Mass transfer to oil phase

Once  $\beta$ -carotene is released from the food matrix, it is either solubilized into oil phase or form emulsion before the absorption. Several factors drive the transfer of released βcarotene into oil phase [50,61]. The availability of the oil phase in the digesta is the primary limiting factor for the mass transfer of β-carotene into the oil phase, which may not accessible due to incomplete digestion of ingested food in stomach resulting incomplete release of oil phase [62]. Reduced particle size also improves its transfer, as it offers greater surface to volume ratio, hence facilitating the partition of released β-carotene into the oil phase of the digesta [50,63]. In contrast, soluble proteins may limit the β-carotene bioavailability as they hinder the incorporation of β-carotene into emulsions resulting after gastric digestion. Addition of 30% and 60% raw supernatant, containing soluble proteins, to blanched carrot juice resulted in 10% and 20% reduction in β-carotene transfer to the oil phase [63]. Further, it was also observed that the decrease in surface charge on emulsions (by reducing pH) improves the solubilization β-carotene into oil phase. Moreover, it is believed that a low pH reduces the solubility of soluble proteins, resulting in acceleration in the rate of transfer to the oil phase. Rich and team recorded a one-hour increase transfer to oil in case of in vitro digested digesta at pH 2.1 as compared to in vitro digested digesta at pH 6.2 [64]. However, it has also been reported that under some conditions, proteins can aid in the emulsification of carotenoids including β-carotene in the digesta, improving its transfer into lipid droplet and thus later intestinal bioaccessibility [53]. This seemed to be the case especially under marginal digestion conditions, i.e. under low enzymatic digestive activity. It appears that both positive (emulsifying) and negative effects (by hampering e.g. enzymatic access to proteincoated lipid droplets) are present, and it depends on individual digestive conditions, testmeals, and carotenoids which effects overwhelm [65].

In addition, the solubility of  $\beta$ -carotene in the oil, the amount of  $\beta$ -carotene in the digesta and quantity of oil ingested and food-matrix aspects equally determine the amount and rate of transfer into the oil phase [64]. For example, dietary fiber is alleged to be a vital factor limiting the transfer of released  $\beta$ -carotene as it causes interference i) hindering micelle formation; ii) affecting triacylglycerol lipolysis and emulsification of fat-soluble food compounds which facilitate the transfer of released  $\beta$ -carotene; iii) limiting the release of lipophilic nutrients from the fat droplets (oil phase); iv) raising the viscosity of chyme, restraining the diffusion of lipophilic  $\beta$ -carotene from micelles into enterocytes [62,66].

#### 3.3. Micelle generation

The passage of the digesta into the small intestine stimulates the secretion of bile salts [50,67]. These bile salts (cholic, chenodeoxycholic, deoxycholic and lithocholic acids) have a high surface activity, which aid in converting small lipid droplets into mixed micelles. The surface-active nature of these bile salts further improves the incorporation of  $\beta$ -carotene into mixed micelles by reducing their size to about 80 Å [68]. The incorporation of  $\beta$ -carotene into mixed micelles is regarded as obligatory for its uptake by the intestinal epithelium, as it assures aqueous solubility and the diffusion to the unstirred water layer. Hence, factors affecting mixed micelle formation can significantly impact the bioavailability of  $\beta$ -carotene. An array of factors affecting the formation of micelles has been meanwhile reported on, including the amount of lipids in the [56,69,70] digesta, type of fatty acids [71], degree of

unsaturation and length of fatty acid [71], presence/absence of dietary fibers [49], and the presence of high amounts of minerals [72,73].

Dietary fat is one of the most important factors, as it not only facilitates the incorporation of β-carotene into mixed micelles, but also stimulates the secretion of bile salts. Prince and Frisoli [74] reported a 2.5-fold increase in β-carotene serum levels 40 h postprandial when β-carotene was ingested along dietary fat as compared to β-carotene ingested without dietary fat [74]. Furthermore, a rise in β-carotene serum levels (and other carotenoids) was also recorded when salad was ingested along with avocado oil (24 g) or avocado (150 g avocado) compared to salad alone [75]. A rise in β-carotene serum level of human subjects was also noticed when they were fed with β-carotene (8 mg) along with increasing quantity of hot bread spread (from 3 g to 36 g) [69]. In their totality, these results clearly indicate that there must be a minimum threshold for the amount of dietary fat present in test meals to enable optimal β-carotene absorption, an amount which is likely of at least 3 g of dietary fat for the uptake of β-carotene for a typical meal containing approx. 8 mg of the carotene. Nevertheless, the proposed threshold (3 g fat for 8 mg β-carotene) still remains a matter of debate and is likely to depend on matrix factors and perhaps host factors. Moreover, Castenmiller and his team proposed 5 g of fat per meal for optimal absorption of β-carotene [70]. This proposal was also supported by Hedren et al. [55] when adding 20% of cooked oil into homogenized carrot pulp improved β-carotene in vitrobioaccessibility by 27% [55]. In addition to the amount of dietary fat, chain length of fatty acids equally influences micelle formation, as well as β-carotene incorporation within the mixed micelles. Hugo and team registered a significant increase (4.9 to 8.6 to 14.9%) in micelle efficiency with increased fatty acid chain length from butanoic acid (4) to octanoic acid (8) to oleic acid (18), respectively. This may not be surprising, given that short and even medium-chain fatty acids can be absorbed via the portal vein [76], and not necessarily contribute to mixed micelle formation. Moreover, the degree of unsaturation in fatty acids also have shown significant impact on bioavailability i.e. higher bioavailability of β-carotene; was observed when it was ingested along with unsaturated vegetable oil when compared to saturated vegetable oil [77]. In contrast, the micelle efficiency was not significantly influenced with increase in degree of unsaturation from 1 (oleic acid, c18:1), to 3 (linoleic acid, c18:3) [77].

As for matrix release and oil droplet incorporation, dietary fiber is thought to limit  $\beta$ -carotene bioavailability. The inhibitory effect of dietary fibers on  $\beta$ -carotene bioavailability has been demonstrated by several *in vivo* and *in vitro* studies [67,78-80]. The inhibitory effect of dietary on  $\beta$ -carotene bioavailability could be attributed to a number of factors, including hindrance in micelle formation; alteration on triacylglycerol lipolysis and emulsification of lipophilic compounds; and finally, restraining the diffusion of  $\beta$ -carotene from mixed micelle to enterocytes.

# 3.4. Absorption

Following diffusion through the mucus-layer in the small intestine, micelles incorporating  $\beta$ -carotene come into contact with enterocytes, resulting eventually in the uptake of  $\beta$ -carotene into the cytosol of the enterocyte. Absorption of  $\beta$ -carotene is thought to be concentration dependent process i.e. at lower concentration it absorbed via protein transporters including CD 36 (cluster determinant 36) and SR-BI (scavenge receptor class B type 1) while at higher concentration is follows passive diffusion [81].

Passive diffusion is thought to be the primary mechanism for  $\beta$ -carotene absorption and is mediated by the difference between micelle and plasma membrane of enterocyte [49,50,81]. Viscosity is thought to be a limiting factor also for this diffusion process, as it interferes with the mobility of the mixed micelles [82]. Several other factors, such as physiochemical state of  $\beta$ -carotene (molecular forms, potency and their physiological

linkages), presence of lipophilic compounds, phytosterols, soluble proteins, surface active compounds (phosopholipids/surfactant), inhibitor/enhancer β-carotene and host related factors (age, disease, surgery, obesity, genetic variation) are equally responsible for influencing the bioavailability of β-carotene, by a variety of factors such as competitive mechanisms, SNP-expression, available surface for absorption etc., which have been comprehensively reviewed in our previous article [49,83]. After absorption, β-carotene needs to be incorporated into chylomicrons before entering the lymphatic system and systemic circulation [37,61]. The transport through the cells has been the topic of some discussion, but has not been fully elucidated. It may include unidentified transport proteins, BCO1, retinol binding proteins, and others [84-86].

# 4. Bioavailability assessment

Determining the bioavailability in human subjects is considered to be ideal, but *in vivo* bioavailability studies with many conditions and thus participants seem impractical due to large variation among the population, cost issues, noncompliance of ethical restriction and time-consuming nature of experimentation. *In vitro* digestion models are gaining popularity as they are reproducible, rapid and allow for handling a large number of samples in parallel. Even though *in vitro* digestion protocols to evaluate the bioavailability of bioactive agents (including  $\beta$ -carotene) have been developed and advanced in the last decade, there is still some remaining controversy around standard digestive models which can be used for assessing  $\beta$ -carotene bioavailability.

Selection of a suitable in vitro digestion model is the first stage for evaluating the bioaccessibility of a nutrient. Currently primarily two types of in vitro digestion models, static and dynamic models are employed for determining the bioavailability of bioactive compounds [49,51]. The static digestion models rely on set physico-chemical conditions (pH, bile salt concentration, enzyme) occurring during the digestion process without imitating peristalsis, fluid flow and mixing occurring during digestion. Dynamic models rely on mechanical forces that occur during digestion along with imitating the enzymatic and chemical changes (changes of enzyme, mineral and bile concentrations and pH) over time and between the different compartments. Dynamic models offer better control over pH, enzyme concentration and mechanical forces, but are more difficult to set up. Selection of suitable digestion models solely relies on the scope of measurements as well as the nature of samples to digest. Discrepancies in the measurement of β-carotene bioaccessibility between such methods has been reported, e.g. from almond butter by dynamic in vitro digestion (87.1%) vs. a static model (51.0%) [87]. These observations suggested that static in vitro models suit simpler samples with perhaps higher throughput, while dynamic in vitro digestion models are more suitable for solid or semisolid food matrix. Several in vitro models (gastric as well intestinal) have been applied to determine  $\beta$ -carotene bioavailability which is primarily derived from the model proposed by Garret [88]. Each model has its advantages and limitation which have been comprehensively reviewed in our previous article [49]. However, a huge step forward was made with the proposed INFOGEST consensus model, published in 2014 [89], with a follow-up update a few years later [90], which was based on both physiological meaningful conditions as well as practicability aspects.

Several factors, including food composition, complexity of the matrix, degree for processing and genetic variations play vital roles in the bioavailability of  $\beta$ -carotene [36,83]. Generally, when  $\beta$ -carotene is released from food matrix, it has to be incorporated into oil droplets, either formed during lipid digestion or present in the original food (e.g., emulsions). The attachment of lipases from digestive juices at the oil droplet surface initiates lipid digestion. The digested lipid products, particularly some free fatty acids and monoacylglycerols take part in the formation of mixed micelles (also contain bile salts and

phospholipids), which behave as carriers to solubilize  $\beta$ -carotene and transport it to the epithelium cells before adsorption [61]. Therefore, the ingestion and hydrolysis of lipids have been regarded as essential steps in the bioavailability of  $\beta$ -carotene [91,92]. Technically, any factors that influence lipid digestion would affect the bioavailability of  $\beta$ -carotene.

# 4.1. Improving bioavailability of $\beta$ -carotene by encapsulation

A variety of foods are being fortified with  $\beta$ -carotene, but direct addition of  $\beta$ carotene in food may result in inescapable interactions, which lead to compromises regarding food quality, taste, appearance and the bioavailability of β-carotene that can significantly diminish its efficacy as disease-combating agents [93,94]. In addition, the obligatory role in human health and the mentioned physico-chemical challenges of β-carotene drives the development toward more efficient, biocompatible, patient compliance, and safer delivery systems, such as using nanotechnology for better incorporation in target foods [95]. These challenges open new windows of opportunity to food technologists to utilize nanotechnology and to develop β-carotene delivery systems that do not compromise food quality. Encapsulation is regarded as an indispensable tool to fabricate delivery systems with improved bioavailability, by stabilizing of β-carotene in the target foods and also during gastrointestinal (GIT) passage, improving its solubility in digestive fluids hence enhancing its absorption from the GIT, and possibly even evading first-pass metabolism loss in various tissues. The bioavailability of encapsulated lipophilic compounds including β-carotene is compromised by a range of factors and has been reviewed by various researchers in excellent reviews [46,48,51,55,96-99].

In order to attain the desired solubility, dispersity, stability and bioavailability for  $\beta$ -carotene, a range of delivery systems, differing in design, structure, composition and production processes, have been tested to validate their potential to encapsulate  $\beta$ -carotene and to be an efficient carrier for  $\beta$ -carotene delivery in food systems [51]. From the origins of nanostructure as delivery system for  $\beta$ -carotene to the present date, the number of publications based on delivery systems has significantly increased. There are three major reasons that can explain their success (i) the improvements in delivery system development (ii) advancements regarding innovative technologies for delivery system synthesis avoiding organic solvents; (iii) applications of newly developed drug delivery systems for food applications.

The success of the inclusion of delivery systems encapsulating lipophilic compound like  $\beta$ -carotene in food items solely relies on the achievements of the following targets [100,101] (i) reduce solubility complications between  $\beta$ -carotene and the food matrix; (ii) protecting  $\beta$ -carotene against pH, temperature, moisture, oxidation and other detrimental external environment conditions; (iii) demonstrating improved bioavailability, also considering the potential for controlled and site-specific release of encapsulated  $\beta$ -carotene; (iv) avoidance of interferences with desired physiochemical properties of the food system.

# **5.** Delivery systems for β-carotene

β-carotene is often utilized as a natural colorant and additive in food in spite of having poor water solubility, a high melting point, susceptibility against environmental conditions, chemical instability, heterogenous distribution in food matrices, and low bioavailability; all factors which limit its potential for the food industry. In this regard, encapsulation techniques have allowed researchers to develop a range of delivery systems with desired functionalities, such as enhanced stability, high dispersibility, improved solubility and targeted/controlled release and improved bioavailability [102,103].

Delivery system corresponds to a technology where a bioactive ingredient is enclosed in nano-/microstructure not only to protect bioactive compounds against environmental

degradation due to e.g. such as oxidation, pH and enzyme, but also release them at a particular target site and at a defined rate [51]. Food application delivery systems needs to be considered as being generally recognized as safe (GRAS) ingredients. At present, the most investigated delivery systems adopted for β-carotene canprimarily be categorized into two groups: polymer-based delivery systems (PBDS) and lipid-based delivery systems (LBDS).

### 5.1. Polymer-based delivery systems

Polymer-based delivery systems (PBDSs) use the intrinsic diversity of polymers to develop encapsulating bioactive compounds in nano-delivery forms with improved functionalities. PBDSs either fabricated with a synthetic polymer whose long-term health risk is regarded as minimal or made up of natural polymers such as proteins and carbohydrates. However, the latter are either hard to scale-up as they require several heat and often complex treatments which are hard to control or result in porous micro-/nanoparticles, thereby failing the objective of the encapsulation. A range of PBDSs have been reported in the literature. In present review, we have included only those PBDSs which are derived from either natural food grade materials or are generally recognized as safe polymers. Typical PBDSs include nano-/microspheres, nano-/microcapsules, hydrogel micelles, colloidal nano-/microemulsions and nanofibers, all of which mainly consist of synthetic or natural polymers (Figure 2A and 2B).

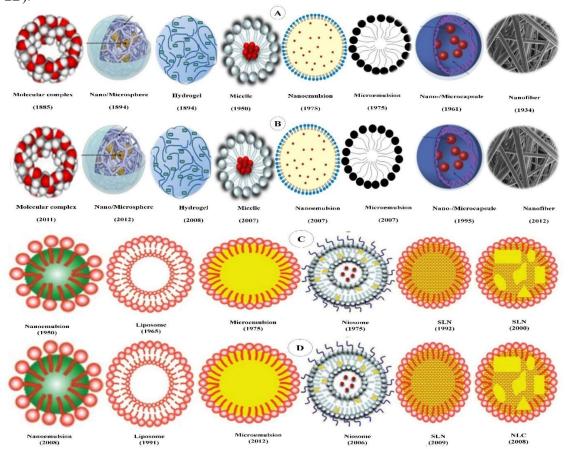
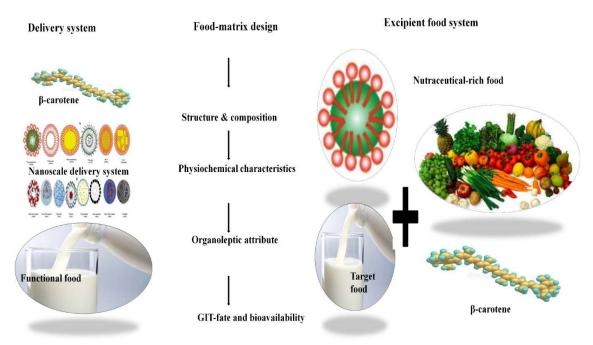


Figure 2: (A) Historical event in the evolution of polymer based delivery systems, (B) Historical event in the application of polymer-based delivery system for encapsulating  $\beta$ -carotene, (C) Historical event in the evolution of lipid-based delivery systems, (D) Historical event for applying lipid-based delivery system for encapsulating  $\beta$ -carotene.



**Figure 3:** Strategy to improve the bioavailability of lipophilic constituents in foods.

# 5.1.1. Inclusion complexes

Inclusion complexes are one of the most adopted delivery systems for encapsulating bioactive compounds. These inclusions occur at molecular level, involving the entrapment of the targeted bioactive compound by a host molecule, through weak physico-chemical forces including van der Waal forces, hydrogen bonding or hydrophobic interactions. The complex formation between the bioactive compound and the host molecule occurs only in the presence of water. Cyclodextrin molecules are the most widely used host molecule for the preparation of molecular complexes. Cyclodextrins are macrocyclic oligosaccharides, comprised of  $\alpha(1,4)$ -linked glucopyranose subunits that contain a distinctive hydrophilic outer surface and a lipophilic central cavity [104]. This molecule offers a cage-like supramolecular structure, appropriate to interact with the structures of various lipophilic bioactive agents. Utilizing their ability to form inclusion complexes with a range of "guest" molecules, cyclodextrins are recognized as being among the most important supramolecular host molecules [105].

The literature describes various methodologies such as solvent evaporation, chemical modification and isoelectric precipitation-fabricated inclusion complexes [106-111]. This paper focusses on those methodologies which allowed the formation of β-carotene inclusion complexes (Table 1). Both human and animal studies suggests that cyclodextrins can be used to enhance lipophilic bioactive compound like β-carotene delivery in food matrix [104,112-114]. There is only a single report on the encapsulation of  $\beta$ -carotene in cyclodextrins which was published in 2011, in which the authors assessed the solubility of cyclodextrincomplexes encapsulating β-carotene [113]. Furthermore, researchers have also utilized maltodextrin's ability to encapsulate β-carotene [115]. Moreover, a research team also validated the suitability of the amylose molecules to encapsulate lipophilic β-carotene [116]. For that purpose, they encapsulated β-carotene in spherical microparticles (mean diameter 8 mm) using emulsion method and carried out it's stability studies against oxidative stress (FeCl<sub>3</sub>), photodegradation and release kinetics in simulated digestive fluid (gastric as well as intestinal fluid) [116]. These amylose microparticle were not able to retain  $\beta$ -carotene activity upto 70% as compared to nonencapsulated β-carotene after 7 h of UV exposure but also had higher stability (75% retention) as compared to nonencapsulated β-carotene (18%) after 7 h of FeCl<sub>3</sub> exposure [116]. Further, simulated digestion studies also suggested that amylose microparticle were not resistant of acid conditions (resistant to gastric digestion) but demonstrated high release (25% of encapsulated  $\beta$ -carotene) in simulated intestinal fluid during 3 h treatment [116].

Despite the high stability of entrapped bioactive compounds, molecular inclusion inherits several limitations, including poor release of the encapsulated bioactive compound, low loading capacity, as well as high cost and failure of legislative compliance, as cyclodextrins are not legally permitted in food systems in some countries. These limitations do not allow researchers to use the full potential of cyclodextrins to encapsulate  $\beta$ -carotene in a suitable delivery system for food applications. To deal with regulatory compliance, researchers have come up with specific carbohydrate molecules (amylose and maltodextrin) which display unique binding properties to lodge lipophilic ligands in their hydrophobic patches. These molecules (amylose and maltodextrin) offer high encapsulation, protection against oxidative, chemical and photodegradation for  $\beta$ -carotene could be attributed to three-way interaction (i) the helical cavity/ hydrophobic patches of these carbohydrate molecules demonstrate greater affinity for lipophilic  $\beta$ -carotene possibly due to their "slim" and hydrophobic alkyl chain (ii) altered microparticles matrices' viscosity profiles resulting in the formation of a soluble high molecular weight nano-complex (iii) offer better linkage for carbohydrate-surfactant- encapsulant compound ( $\beta$ -carotene) in ternary structure [117].

## 5.1.2. Micro-/Nanospheres

Micro-/nanospheres are uniform spherical structures derived from non-biodegradable or biodegradable natural or synthetic polymers having particles size between 1-1000  $\mu m$  (microspheres) and or 1-1000 nm (nanospheres). The water-soluble polymer or mixture of polymers is dispersed in an organic phase to form spherical structures in the presence of cross-linking agents. Bioactive compound can be encompassed into the inner hollow core of nanospheres or entrapped into the polymeric matrix of a solid micro-/nanosphere.

The literature describes several methodologies for the preparation of nano-/microspheres, such as single emulsion, double emulsion, coacervation phase separation, and polymerization methods adopted for encapsulating various bioactive compound [118-121]. These delivery systems are renowned for their rather ease of optimization to obtain the desired functionalities, such as dictated by pharmaceutical needs, including targeted and temporal control of release of encapsulated drug, efficacy and *in vivo* stability as well as biocompatibility. Micro- and nanospheres are well acknowledged delivery systems in biomedical and pharmaceutical industries. These desirable functionalities have attracted researchers to apply these microspheres and nanospheres to food products to deliver bioactive compound through encapsulations.

In spite of the great potential in the pharmaceutical field for drug delivery, nano-/microspheres remain under-utilized for  $\beta$ -carotene encapsulation. In order to obtain better knowledge about the role of nano-/microsphere for  $\beta$ -carotene delivery, we have discoursed about those methodologies that are involved in  $\beta$ -carotene encapsulation. The encapsulation of  $\beta$ -carotene in micro-/nanosphere was first carried out with a carrageenan/carboxymethyl cellulose based microsphere to determine the release kinetics of encapsulated  $\beta$ -carotene from genipin-cross-linked kappa-carrageenan/carboxymethyl cellulose [122]. During course of time, several studies were carried to evaluate the potential of polymeric micro-/nanospheres as an alternative delivery system for  $\beta$ -carotene encapsulation [119,123-130]. Nevertheless, there is a scarcity of data on the use of nano-/microsphere for the purpose of  $\beta$ -carotene fortification in food systems. Though these micro-/nanospheres are relatively easy to scale-up as they do not require sophisticated instrumentation, there are also several challenges such as poor loading capacity [131], premature release and degradation by

enzymes [132] that could be the cause for micro-/nanospheres not being among the more accepted species for the encapsulation of  $\beta$ -carotene.

## 5.1.3. Nanohydrogels

Nanohydrogels, three-dimensional soft gels, are generally made by cross-linking the water-soluble material, which is comprised of a wide range of chemical compounds and bulk physical properties. The use of hydrogels as a delivery system results in a number of advantages, including reduced systemic side effects [133], sustained and site-specific drug delivery under desired external stimuli such as thermal, pH or mechanical changes [134] and reduced systemic side effects attributed to loss in encapsulated bioactive compounds (i.e.β-carotene) during digestion and inevitable interaction with other components of food matrices hence offering improved bioavailability [135]. The literature has been updated with excellent reviews on preparation methods for nanohydrogels including sonication methods, cross-linking and inverse-suspension polymerization [136-138].

To gain crisp knowledge on nanohydrogels to design delivery systems encapsulating β-carotene for the purpose of food fortification, the focal point was kept on those reports which involved natural or food grade ingredients. Chu et al. [139] compared the suitability of Sodium caseinate (SC) (mean diameter 17 nm) and whey protein (mean diameter 45–127 nm) based hydrogels to protect encapsulated β-carotene against physico-chemical stress including heat, salt and pH [139]. It was observed that β-carotene encapsulated within Sodium caseinate based hydrogels had higher stability (minimal change in particle size and zeta potential) as compared to whey protein-based hydrogels against various stress conditions [139]. Similarly, β-carotene loaded κ-carrageenan hydrogel was also synthesized and tested for photodegradation, thermal stability and simulated digestive release kinetics. It was observed that approximately 75% of encapsulated β-carotene was retained in κ-carrageenan hydrogel after 24 h of UV exposure while approximately 89% of encapsulated β-carotene was found to be reattained when they were incubated at 4°C as compared to hydrogel incubated at 25°C (> 35%) [140]. Further, alginate nanohydrogel was found to be more effective in providing stability to β-carotene under accelerated storage conditions (55°C), bioaccessibility and bioavailability as compared to β-carotene encapsulated in nanoemulsion [141]. The high structural and chemical stability of the developed hydrogel system against pH, heat and salt, encouraged further progress in designing hydrogels as an efficient delivery system for β-carotene (Table 1) [115,122,141-147]. Nevertheless, the great potential hydrogel also carries several limitations including poor loading capacity [135], premature release and oxidation of  $\beta$ -carotene [122,141]. These could be among the reasons that hydrogels have not well being adopted as species for the encapsulation of  $\beta$ -carotene for food applications.

#### 5.1.4. Micro-/Nanocapsules

Recently, microspheres have been well adopted for site-specific drug delivery, such as anticancer agents; intestinal targeted drug and mucoadhesive drug delivery systems, due to their proficient properties including high loading capacity, high encapsulation efficiency, enhanced stability and great biocompatibility. In general, micro-/nanocapsules belong to the vesicular system family in which the bioactive compound is situated within a cavity comprised of an inner liquid core fenced by a polymeric membrane, with a range of sizes, microspheres (1-1000 µm) and nanospheres (1-1000 nm). Solvent displacement and spraydrying are some of the well adopted techniques for fabricating nano-/microcapsules. These delivery systems are recognized as substitute to liposomes due to its cost-effective and triggered release under specific stimuli.

The first report on the use of microcapsules to encapsulate carrot-derived  $\beta$ -carotene was published on  $\beta$ -carotene loaded microcapsules which were prepared by using spray-

drying to evaluate the effectiveness of microcapsules to retain encapsulated  $\beta$ -carotene [148]. In the following, a research team developed  $\beta$ -carotene loaded nanocapsules (differeing in gum arabic concentration 15 to 30%) to study the impact of the effect of increase gum arabic concentration (15 to 30%) on the stability of  $\beta$ -carotene and it was found that microcapsules fabricated with 25% of gum arabic had highest retention capacity for  $\beta$ -carotene [149]. Thereafter, various reports have been published on the production of micro-/nanocapsules [150-156] (Table 1).

Despite these available gained insights, only few food technologists have prepared  $\beta$ -carotene loaded nanocapsules that are suitable for the purpose of food applications [150-154]. However, there is still a scarcity of data on micro-/nanocapsules encapsulating  $\beta$ -carotene that have been used for the purpose of food fortification. This could be because of their operative limitations such as complexity in their fabrication process [157], the use of synthetic polymers [158] and the susceptibility for leakage of  $\beta$ -carotene which is adsorbed on their surface or can be imbibed within the polymeric membrane [159]. These limitations are also further aggravated by the failure of technology to resolve stability issues such as aggregation, fusion, leakage and sedimentation. Once these aforementioned limitations are addressed and solved, there is great potential for micro-/nanocapsules to act as efficient delivery systems for  $\beta$ -carotene in food applications.

#### 5.1.5. Nanofibers

The exclusive properties of nanofibers such as their nanoscale dimensions, quick wetting properties, rapid release and temperature independence nature makes nanofiber-based delivery systems a good technique for the delivery of heat sensitive bioactive agents such as β-carotene [160]. Electrospinning, freeze-drying and centrifugal spinning are extensively adopted encapsulation techniques for heat susceptible bioactive compounds [161,162]. A range wall material is used to fabricate nanofibers to attain aforesaid functionality which include which are broad categorized in two classes (i) natural and (ii) synthetic based to the nature. Natural wall materials involve cellulose, chitosan, pullulan, cyclodextrins, starch, gelatin, zein protein, egg albumin, soy protein, whey protein while synthetic wall materials include polyvinyl alcohol, cellulose acetate, hydroxypropyl methyl cellulose, ethyl cellulose, methyl cellulose [163,164].

In spite of these promising properties, nanofibers have remained untapped for encapsulating  $\beta$ -carotene. It is evident that there is a scarcity of reports addressing  $\beta$ -carotene encapsulation in nanofibers [165-169]. One major reason is that the porous nature of nanofibers makes them liable to oxidative degradation of  $\beta$ -carotene, which makes it unfit as a delivery system for  $\beta$ -carotene encapsulation [163].

## 5.2. Lipid based delivery systems

Lipid-based delivery systems (LBDSs) involve those delivery systems which are principally composed of physiological lipid analogs such as surfactants as stabilizers (Figure 1A and B). LBDSs have been recognized for their promising biocompatibility, competency in GIT penetration, easy to scale-up and broad application [102,170]. LBDS have been admired for their potential for drug delivery through various administration routes, particularly for the oral delivery of lipophilic drugs, because of their competence to mimic the food lipids during the digestive process [171,172]. With their properties, lipid-based delivery systems offer an array of advantages over polymer-based systems as shown in Table 2. Some of these advantages of lipid based nanodelivery systems entail (i) biocompatibility and use of nontoxic excipients [170,173]; (ii) high drug payload [174]; (iii) viability of incorporating both lipophilic and hydrophilic bioactives [170]; (iv) prospect of controlled release and drug targeting; (v) improved drug stability [175]; (vi) averting of organic solvents [176]; (vii) cost

effectiveness [177]; as well as (viii) ease of scale-up during production and sterilization [95]. During the course of time, a range of lipid-based delivery systems have been developed for encapsulating bioactive compounds such as micelles, micro and nanoemulsions, liposomes, niosomes, solid lipid carriers, nanostructured lipid carriers, bilosomes, cubosomes etc. [178]. However, in the present review the emphasis has given those LBDSs which have been adopted for encapsulation  $\beta$ -carotene in particular particularly micelles, micro and nanoemulsions, liposomes, niosomes, solid lipid carriers, nanostructured lipid carriers which are discussed in the following sections.

**Table 1:** Engineered nanoparticle-based delivery systems for enhancing the bioavailability of  $\beta$ -carotene.

Class of delive ry syste ms	Subclass of delivery system	Delivery system	Ingredients	Technique/Prepa ration method	Physio- chemical studies	Encapsula tion efficiency	Release studies	Particle Size	Cellular/ani mal studies	Applicatio ns	Referen
Lipid derive d delive ry syste	Self- assemble d delivery system	Liposom	<ul><li>Hydrogenated soy PC</li><li>Lipoid gmbh</li><li>Xanthan gum</li></ul>	Spray-drying	DSC, small- angle X-ray scattering (SAXS), TEM, DLS, ELS	NA	NA	700 to 3000 nm	NA	NA	[228]
ms			<ul><li>Hydrogenated soybean PC</li><li>Lipoid</li></ul>	Ethanol injection method	FTIR, SEM, Raman microspectros copy, UV-vis irradiation	NA	NA	NA	NA	NA	[166]
			Gama-oryzanol	Modified thermal method	FTIR	NA	NA	64 nm- 500 nm	NA	NA	[230]
			<ul><li>Egg yolk phospholipid</li><li>Tween 80</li></ul>	Thin-film evaporation method	DLS, AFM	NA	SGF, SIF	600 nm	NA	NA	[233]
			<ul> <li>Phospholipids         (Lipoid S-100-H and Lipoid S-40, Lipoid gmbh)     </li> <li>Sucrose</li> </ul>	Spray drying	DLS, ELS, XRD, SEM	NA	NA	285 - 1695 nm	NA	NA	[232]
			<ul><li>PC</li><li>PS</li><li>PEA</li></ul>	Dehydration/rehyd ration method	NA	NA	NA	NA	Hamster	pharmaceuti cal	[231]
			• PC	Dehydration/rehyd ration method	NA	NA	NA	NA	Microsomes /Rat	pharmaceuti cal	[225]
		Niosome	• Spans 40, 60, 80 • Tween 20, 40, 60	Dehydration/rehy dration method	DLS, EE, TEM	16.0 -51%	NA	273.2- 367.9 nm	RAT-1 immortalize d fibroblasts	pharmaceuti cal	[243]

		• Cholesterol								
Particulat e delivery	Solid lipid nanoparticl	<ul><li>Brij 30</li><li>Octadecane</li></ul>	Phase inversion temperature	DLS, DSC	NA	NA	109 and 128 nm	NA	NA	[297]
systems	es	<ul><li>SPI</li><li>Xanthan gum</li><li>Palm stearin</li></ul>	Hot homogenization	DLS	NA	NA	1.20- 1.70 μm	NA	Food application( ice creams)	[255]
		<ul> <li>Tristearin</li> <li>Sunflower oil</li> <li>Hydrogenated soy lecithin</li> <li>Tween 80</li> </ul>	Hot pressure homogenisation	DLS, ELS, DSC	NA	NA	NA	NA	NA	[298]
		<ul><li>SC</li><li>WPI</li><li>SPI</li></ul>	Microfluidization	DLS, TEM	99.1%, 98.8%,	NA	77.8- 190.9 nm	Caco-2 cells	NA	[247]
		<ul><li>Hydrogenated palm oil</li><li>Cocoa butter</li><li>Tween 20</li></ul>	Hot high-pressure homogenization method	DLS, DCS, NMR	NA	NA	168 - 227 nm	NA	NA	[299]
		<ul><li>Polyoxyethylene</li><li>Tween 80</li></ul>	Phase inversion temperature	AFM, DLS, DSC, XRD	NA	NA	<400 nm	NA	NA	[300]
		<ul><li>Stearic acid</li><li>Sunflower oil</li><li>Tween 80</li></ul>	Hot agitation	DLS, DSC, XRD	NA	NA	<5 μm	NA	NA	[301]
		<ul><li>WPI</li><li>Corn oil</li></ul>	Homogenization	DLS, ELS, TEM, SEM	NA	NA	<200 μm	NA	NA	[254]
		<ul> <li>Hydrogenated canola stearin</li> <li>Polyoxyethylene</li> <li>Sorbitan monolaurate</li> </ul>	Hot homogenization	DLS, DSC, ELS, Cryo TEM, NMR, XRD	NA	NA	111.7- 170.8 nm	NA	NA	[210]
		<ul> <li>Tripalmitin</li> <li>Phospholipid</li> <li>Polyethylene glycol sorbitan monooleate</li> </ul>	Hot high-pressure homogenization	DLS, ELS, DSC	NA	NA	0.20- 6.09 □ □ m	NA	NA	[253]
	Nanostruct	<ul> <li>Glyceryl</li> </ul>	Solvent	DLS, DSC	NA	NA	500 nm	NA	NA	[260]

ured lipid carriers	tristearate High oleic sunflower Tween 80	displacement technique							
	<ul> <li>Cremophor RH40</li> <li>Span 80</li> <li>Cupuacu butter</li> </ul>	Phase inversion temperature	DLS, DSC, TEM	NA	Gastric fluid, Duoden al fluid, Jejunal fluid, Ileal fluid	31.6 - 34.08 nm	NA	NA	[302]
	<ul> <li>Tween 80</li> <li>Tween 60</li> <li>Tween 80</li> <li>Phosphatidylcho line</li> <li>Grape seed oil</li> </ul>	Hot homogenization	DLS, ELS, DSC, TEM	65.26- 74.35%	NA	85.2- 129.2 nm	NA	NA	[271]
	<ul> <li>Propylene glycol monostearate</li> <li>Propylene glycol monoand distearates</li> <li>Propylene glycol monoand dipalmitates</li> <li>Sunflower oil</li> </ul>	Hot homogenization	DLS, DSC	NA	NA	82 - 217 nm.	NA	NA	[270]
Micro emulsion	<ul> <li>Span 80</li> <li>Span 40</li> <li>Tween 80</li> <li>virgin coconut oil</li> <li>Palm oil</li> </ul>	Spontaneous emulsification method	DLS, ELS	NA	NA	20 22.60 nm	NA	NA	[207]
	<ul><li>Tween 20</li><li>Corn oil</li></ul>	Microfluidization	DLS, CFFM	NA	SSF, SGF	0.21 - 23 μm	NA	NA	[211]

•	Hydrogenated canola stearin Tween 20	Hot homogenization	DLS, DSC, ELS, Cryo TEM, NMR, XRD	NA	na	115 nm	NA	NA	[210]
•	Sucrose monolaurate Lactoglobulin Whey proteins	Microchannel Device	DLS	NA	na	27.9 μm	NA	NA	[209]
•	Lactoferrin B- Lactoglobulin	Microfluidazion	DLS, ELS	NA	NA	< 250 nm	NA	NA	[208]
Vano emulsion	Corn oil Tributyrin	Homogenization	DLS	NA	SSF, SGF, SIF	1.25 - 1.34 μm	NA	NA	[205]
•	Tween 20 Corn oil	Microfluidization	DLS, DSC	NA	SSF, SGF, SIF	0.2 -23 μm	NA	NA	[211]
•	Miglyol 812 (MCT) Corn oil (LCT)	Microfluidization	DLS, ELS, DSC	NA	SSF, SGF, SIF	146 to 415 nm,	NA	NA	[303]
•	Corn oil Lemon oil Sucrose Ponopalmitate Lysolecithin	Microfluidization	DLS	NA	SSF, SGF, SIF	<150 nm)	NA	NA	[91]
•	Long chain triglyceride medium chain triglyceride Tween 20	Microfluidization	DLS	NA	SSF, SGF, SIF	140– 170 nm	NA	NA	[92]
•	Sunflower lecithin Tween 20 Peppermint oil	Heating and stirring	DLS	NA	NA	<10 nm	NA	NA	[304]
•	Orange oil B-lactoglobulin	Microfluidization	DLS	NA	NA	<100 nm).	NA	NA	[305]

			• Tween 20								
			• Tween 80	Supercritical fluid	DLS	NA	NA	50-150 nm	NA	NA	[200]
			Corn oil	Hot homogenization	DLS	NA	SSF, SGF, SIF	< 200 nm	NA	NA	[174]
			<ul><li>Tween 80</li><li>Stearic acid</li></ul>	High-speed homogenization	DLS	NA	NA	418.8- 1689.0 nm	NA	NA	[206]
			<ul> <li>Miglyol-812 (caprylic/capric triglycerides</li> </ul>	Spontaneous emulsionfication method	DLS, SEM	NA	NA	100 - 300 nm	NA	NA	[306]
			<ul><li>Compritol</li><li>Poloxamer 407</li></ul>	Hot-High shear Homogenization	DLS, ELS	NA	NA	79 -115 nm	NA	NA	[272]
			MCT oil	Microfluidization	DLS, ELS	NA	NA	97.2- 416.0 nm	NA	NA	[197]
Polym er derive	Self- assemble d	Starch based emulsion	<ul><li>Lactoferrin</li><li>B- Lactoglobulin</li></ul>	Microfluidazion	DLS, ELS	NA	NA	208-385 nm	NA	NA	[208]
d delive ry syste ms	polymer derived delivery systems		<ul> <li>Medium-chain triacylglycerol</li> <li>MCT oil</li> <li>OSA-modified starches</li> </ul>	Spray drying	DLS, ELS, SEM	NA	NA	114 - 118 nm,	NA	NA	[150]
			<ul><li>NaCMC</li><li>Kappa- carrageenan</li></ul>	Cross linking	SEM	NA	NA	700nm	NA	NA	[122]
			Modified starches	High-pressure homogenization	DLS	NA	SGF, SIF	17 nm	NA	NA	[307]
			OSA-starch	Ultrasound emulsification	SEM	NA	NA	300-600 nm	NA	NA	[308]
			• SSPS	Layer-by-layer	DLS,	NA	NA	250.0e3	NA	NA	[309]

	Beet <u>pectin</u>	electrostatic deposition method	ELS			06.3			
	<ul><li>OSA-modified starch</li><li>MCT</li></ul>	Microfluidization	DLS	NA	SGF, SIF	80.0 ± 1.3 nm	NA	NA	[310]
	<ul><li>SSPS</li><li>Beet <u>pectin</u></li></ul>	Layer-by-layer electrostatic deposition method	DLS, ELS	NA	NA	250.0- 306.3 nm 304.5- 466.6 nm	NA	NA	[309]
Protein based emulsion	A-lactalbumin     catechin	Microfluidization	CD, DLS, ELS	NA	NA	158.8 and 162.7 nm	NA	NA	[311]
	<ul><li> Protein powders</li><li> Sucrose syrup</li></ul>	Homogenization	DLS, DSC	NA	NA	0.48- 0.66μm	NA	NA	[312]
	<ul><li>Sunflower oil</li><li>Hydrogenated palm kernel oil</li><li>WPI</li><li>SC</li></ul>	High speed homogenization	DLS, XRD	NA	NA	0.46_0. 50 μm	NA	NA	[313]
	A-lactalbumin     catechin	Microfluidization	CD, DLS, ELS	NA	NA	158.8 and 162.7 nm	NA	NA	[311]
	• WPI	Ph-cycling method	DLS, ELS, FTIR, SEM	NA	SGF, SIF	409.7 nm	NA	NA	[314]
	<ul><li>Beta- lactoglobulin</li><li>Catechin</li></ul>	Microfluidization	DLS, ELS	NA	NA	160 – 170 nm	NA	NA	[315]
	<ul><li>WPI</li><li>sunflower oil</li><li>Gum arabic</li></ul>	Layer-by-layer electrodeposition technique	DSC, Dynamic Mechanical Analyses (DMA)	NA	NA	NA	NA	NA	[316]

	<ul><li>SC</li><li>Corn oil</li></ul>	Microfluidization	DLS	NA	NA	124-368 nm	NA	NA	[317]
	• SC • Tween 20	Solvent displacement technique	DLS, ELS	NA	NA	30 nm- 206 nm,	NA	NA	[318]
	<ul><li>Lactoferrin</li><li>MCT</li></ul>	Homogenization	CD, DLS, ELS, FSS (Fluorescence spectroscopy)	NA	NA	302 - 583 nm	NA	NA	[319]
	SC WPC	Solvent- displacement method	DLS, ELS	NA	NA	45–127 nm	NA	NA	[139]
	<ul><li>SC</li><li>Alginic acid</li></ul>	Microfluidization	DLS, ELS, FSS	NA	SGF, SIF	0.48 ± 01.87 μm	NA	NA	[320]
	<ul><li>Corn oil</li><li>Canola oil</li><li>Olive oil</li><li>SC</li></ul>	Microfluidization	DLS	70.9%	SGF, SIF	167.4- 178.8 nm	Caco-2, Cell toxicity	Pharmaceut ical	[321]
	• SC	Spontaneous emulsification	DLS, SEM	100 ± 1%	NA	50 to 500 nm	NA	NA	[190]
Carbohydr ate based emulsion	SA Tween 80	Sonication and hot homogenization	DLS, ELS, CFSM	NA	SSF, SGF, SIF	0.2-23 μm	NA	NA	[322]
	<ul><li>Mannitol</li><li>Gelatin</li></ul>	Freeze-dryer	DSC	NA	NA	NA	NA	NA	[323]
Micelle	• SC	Spontaneous emulsification	DLS, SEM	100 ± 1%	NA	50 to 500 nm	NA	NA	[190]
	<ul><li>SC</li><li>Whey protein hydrolysate</li></ul>	Solvent displacement	DLS, ELS, FSEM	NA	NA	13-171 nm	NA	NA	[188]
	<ul><li>Hydroxyethyl cellulose</li><li>Lionic acid</li></ul>	Sonication	DLS, FTIR, NMR, SEM, TEM	84.67%	SSF, SGF, SIF	20–50 nm	NA	NA	[191]
	• Casein	Microfiltration	DLS, FTIR,	NA	NA	0.04 to	NA	NA	[192]

				TEM			0.4 lm			
		<ul><li>Chitosan</li><li>PLA</li></ul>	Polymerization	DLS, FTIR, NMR, XRD, TEM	NA	NA	14 nm	NA	NA	[193]
		<ul> <li>Soybean oil</li> <li>Tween 20</li> <li>Tween 40</li> <li>Tween 80</li> <li>Glycerol monocaprylocap rate</li> <li>Propylene glycol dicaprylate/dica prate</li> <li>Caprylic/capric triglyceride</li> </ul>	Homogenization	DLS, ELS, TEM	NA	NA	12–100 nm.	Caco-2, Cell toxicity study	Food application	[194]
		<ul><li>PLA</li><li>Tween 80</li></ul>	Solvent displacement method	DLS, ELS	NA	NA	0.087 - 1.158 μm	NA	NA	[324]
Particulat e nanoparti	Molecular complex	• Γ-cyclodextrin	Co-precipitation and physical mixture techniques	FTIR, FESEM	NA	NA	NA	NA	NA	[113]
cles		• Amylose	Sonication	DLS, TEM, SEM, XRD	65%	NA	12 ± 3 nm	NA	NA	[116]
		<ul><li>Sunflower seed oil</li><li>acacia gum</li><li>Maltodextrin</li></ul>	Spray drying	SEM	NA	PBS	NA	NA	NA	[115]
	Nanospher e	• Rice protein isolate	Homogenization	CD, DLS, FTIR, CLSM	NA	SGF, SIF	300–400 nm	NA	NA	[290]
		• Zein	Microfluidization	DLS, ELS, TEM	NA	SGF, SIF	32.44 ± 0.87 -	NA	Food application	[325]

						168.17 ± 22.36 nm		(milk)	
	<ul><li>Sunflower oil</li><li>WPI</li><li>Trehalose</li><li>Gum Arabic</li></ul>	Microfluidization	DLS, ELS, Raman-FIB- SEM	NA	NA	46.77 ± 0.17 48.23 ± 0.13 μm	NA	NA	[326]
	Corn starch	Nanoprecipitation method	DLS, DSC, XRD	NA	SIF	0.77 to 0.89 m	NA	NA	[327]
	<ul> <li>Poly[poly(oxyet hylene-1500)-Oxy-5-dodecanyloxyiso phthaloyl</li> <li>Poly [poly-(oxyethylene-1500)-oxy-5-hydroxyisophthaloyl]</li> </ul>	Homogenization	DLS, SEM, TEM, NMR	22.60- 28.08%	Water, Buffer	<100 nm	NA	NA	[328]
	<ul> <li>OSA -modified starches</li> <li>OSA-dextrin</li> </ul>	High-temperature, high-pressure emulsification and antisolvent precipitation	DLS	70-80%	NA	137- 135,900 nm	NA	NA	[329]
	<ul><li>SC</li><li>WPI</li><li>SPI</li></ul>	Homogenization- evaporation method	DLS, DSC, ELS, FTIR, XRD	NA	SGF, SIF	NA	Caco-2 cells	NA	[330]
	<ul><li>PLA</li><li>Stearyl amine</li><li>Stearoyl polyoxyl-32 glycerides</li></ul>	Nanoprecipitation method	DLS, ELS	NA	PBS	117.1 ± 4.6 nm)	MCF-7 breast cancer cells, Cell toxicity studies	Pharmaceut ical	[118]
Microspher e	<ul><li>Almond gum</li><li>Gum Arabic</li></ul>	Spray drying and freeze drying	DLS	66- 70%	Sunflow er oil	1.20- 2.30μm	NA	NA	[331,33 2]

	OSA-modified starches OSA-dextrin	Precipitation	DLS, SEM	65-90%	NA	300-600 nm	NA	NA	[119]
•	Caseins	Spray drying	Photodegradat ion study	NA	NA	na	NA	NA	[333]
	K-carrageenan Oil	Ionic gelation	DLS	NA	SGF, SIF	80–94 nm, 91– 106 nm, 128– 134 nm	NA	NA	[120]
	Canola oil Ethylcellulose	Ionic gelation	Lipid lipolysis	NA	NA	na	NA	NA	[128]
	OSA- <u>modified</u> <u>starches</u> <u>Flax seed</u> oil	Microfluidization	DLS, ELS, FESEM	90%	NA	165.0- 129.1 nm	NA	NA	[334]
	Casein Dextran	Dry heating method	DLS, DSC, FSM	73.64- 74.53	SGF, SIF	111.1- 127.3 nm	NA	NA	[335]
•	WPI Dextran	Glysocylation conjugation	CD, DLS, ELS	NA	SGF, SIF	165.6- 176.0 nm	NA	NA	[121]
	WPI Corn oil	Microfluidization	DLS, ELS	NA	NA	0.14– 0.16 μm)	NA	NA	[336]
	MCT coconut oil Corn oil span 20 Monostearin	High pressure homogenization	DLS	NA	NA	176.3 228 nm	CACO-2 CELLS, RATS	PHARMAC EUTICAL AND FOOD	[337]
•	SPI	Freeze drying	AFM DLS, ELS	NA	SGF, SIF	55 nm	NA	NA	[129]
	PLA	Electrospinning	SEM	NA	NA	NA	NA	NA	[168]
	<u>Casein</u> <u>Maltodextrin</u>	Microfluidization and Spray drying	DLS, ELS	NA	NA	230- 277nm	NA	NA	[127]
	WPI SC	High-pressure homogenization	DLS, ELS	NA	SGF, SIF	142 ± 6- 160	CACO-2 CELLS	NA	[338]

					± 10			
SC     Maltodextrin	High pressure homogenization	DLS, LD, TEM	NA	NA	262.8 ± 4.10-307.1 ± 5.40 nm	NA	NA	[339]
<ul> <li>OSA-modified starch</li> <li>Tween-80</li> <li>Flax seed oil</li> <li>MCT</li> </ul>	Microfluidization	DLS, ELS	NA	NA	123.9- 207.2 nm	NA	NA	[340]
• Shellac	Syringe microfluidzation	SEM	NA	NA	19-84 μm	NA	NA	[126]
<ul><li> Xanthan</li><li> Gum</li><li> Palm stearin</li><li> Hydrolyzed SPI</li></ul>	Homogenization	DLS, DSC, ELS, FFS	NA	NA	1- 1.5 μm	NA	NA	[341]
Chitosan	Cross linking and sonication	DLS, SEM	NA	NA	1,570.0 nm.	NA	Food application (hamburger patties)	[342]
<ul><li>Soybean oil</li><li>Ulva fasciata polysaccharide</li></ul>	Microfluidization	DLS	NA	SSF, SGF, SIF	0.82 μm	NA	NA	[343]
<ul><li>Zein</li><li>Carboxymethyl chitosan</li></ul>	Rotating evaporation	DLS, DSC, ELS, FTIR, SEM	56.5- 92.7%	SGF, SIF	70.41 ± 0.67- 420.9 ± 2.34 nm	NA	NA	[344]
<ul> <li>Flax seed oil</li> <li>MCT</li> <li>OSA modified starch</li> <li>Tween-80</li> </ul>	Microfluidization	DLS, ELS	NA	NA	123.9- 207.2 nm	NA	NA	[345]
• Casein	Microfiltration	DLS, FTIR, TEM	NA	NA	0.04 to 0.4 lm	NA	NA	[192]

n • I		High pressure homogenization	DLS	NA	NA	1.38e1. 96 mm.	NA	NA	[124]
• <u>C</u>		Polymerization	DLS	NA	SGF, SIF	127- 149 nm	NA	NA	[346]
• 0	Casein I Guar gum a	Homogenization and coacervation pr	DLS, ELS, FTIR, SEM	65.95 ± 5.33%	SGF, SIF	176.47± 4.65 μm	NA	NA	[347]
• E		High-pressure homogenization	DLS	NA	NA	10.1 ± 0.7- 14.5 ± 0.6	NA	NA	[348]
• 0		High-pressure homogenization	DLS	NA	SGF, SIF	170 nm	NA	NA	[349]
		High pressure homogenization	Effect of digestion on particle size	NA	SSF, SGF, SIF	NA	NA	NA	[350]
1 1		Electrostatic complexation	DSC, FTIR	NA	NA	NA	NA	NA	[125]
	Almond gum Gum arabic	Freeze drying	DLS	66-70%	Sunflow er oil	2.10- 3.2 μm	NA	NA	[332]
	, ,	Spontaneous emulsification	DLS	14.18- 64.39%	NA	655- 3418nm	NA	NA	[130]
• 0	Maltodextrin Gum arabic Gelatin	Spray drying	Stability of carotene in powder	NA	NA	NA	NA	Food application	[351]
la la	•	Solvent evaporation	DLS	14%	NA	260 nm	NA	Pharmaceut ical	[352]
c		Homogenization and sonication	DLS, SEM	79.63 ±1.41- 84.32 ± 1.08%	SGF,	210.5 1.23 nm	NA	NA	[353]

			<ul><li>Soybean soluble polysaccharides</li><li>Chitosan</li></ul>	Homogenization	DLS, ELS	NA	NA	0.52 μm.	NA	NA	[354]					
	Capsular nanoparti cles	Microcaps ule	<ul><li>Casein</li><li>Gum</li><li>Tragacanth</li></ul>	Complex coacervation	CLSM, DLS, FTIR, SEM, TGA, XRD	79.36±0.54 1%	SGF	159.71± 2.16 μm	NA	NA	[153]					
			Poly-(3- hydroxybutyrate -co-3- hydroxyvalerate )	Supercritical carbon dioxide micronization technique	NA	NA	organic solvent	NA	NA	NA	[355]					
			Arabic gum	Spray-drying	NA	NA	NA	NA	NA	NA	[149]					
			Hydrolyzed starch	Homogenization	Stabilitystudy	NA	NA	NA	NA	NA	[148]					
			<ul> <li>Poly(hydroxybut irate-co-hydroxyvalerate)</li> </ul>	Supercritical fluid	SEM	NA	NA	NA	NA	NA	[356]					
			<ul> <li>Poly(hydroxybut irate-co-hydroxyvalerate)</li> </ul>	Supercritical fluid	SEM	0.95- 55.54%	NA	1.3 51.9 μm	NA	NA	[357]					
			<ul><li>Chitosan</li><li>Oleic acid</li><li>Fe<sub>3</sub>O<sub>4</sub></li></ul>	Solvent displacement technique	SEM, XRD	78.74- 81.2%	PBS	NA	NA	NA	[358]					
								• Dextrin	Precipitation	DLS, DSC, TEM, XRD	NA	SGF	16-30 nm	NA	NA	[359]
									<ul><li>Chitosan</li><li>SA</li></ul>	Spray-drying	DLS	34-55%	SIF	852 - 958 μm	NA	NA
			Gum arabic	Spray-drying	DLS, SEM	NA	NA	19.69- 20.98 μm	mouse bone marrow and peripheral blood cells/Wistar albino rats,	Pharmaceut icals	[151]					

	Gum arabic Gelatin Maltodextrin	Freeze-dryer	DSC	NA	NA	NA	NA	NA	[360]
	Oil Tween 20	Homogenization and evaporation	CFLS, DLS	NA	NA	161.98 ± 17.19- 189.45 ± 22.69 nm	NA	NA	[361]
	Soybean oil SPI	Homogenization	CLSM	NA	NA	0.23 ± 0.02 - 6.68 ± 0.65 lm	NA	NA	[289]
	WPC Tween 20	Membrane emulsification	DLS	NA	NA	$\begin{array}{ccc} 1.28 & \pm \\ 0.02 & - \\ 1.69 & \pm \\ 0.49 \mu m \end{array}$	NA	NA	[362]
	Maltodextrin Tween 80	Freeze drying	CFLM, DLS, ELS	NA	SGF, SIF	$\begin{array}{ccc} 0.23 & \pm \\ 0.02 & - \\ 0.24 & \pm \\ 0.01 \ \mu m \end{array}$	NA	NA	[115]
	Pea protein concentrate Maltodextrin	Spray drying	DLS, SEM	NA	Water	4.9 + 2.4- 6.0 + 3.0 μm	NA	NA	[363]
	Lactose Trehalose	Spray-drying	DLS, DSC	NA	NA	0.2-0.8 μm	NA	NA	[364]
	Poly-ε- caprolactone	Emulsification- diffusion method	DLS, ELS, SEM	NA	NA	250-650 nm.	NA	NA	[152]
	Medium-chain triacylglycerol MCT	High speed homogenization, spray-drying	DLS, SEM	NA	NA	114-159 nm	NA	NA	[150]
	WPI Alginate Chitosan	Emulsion-ionic gelation technique							[87]
Nanocapsu le	Poly-ε- caprolactone	polymer method	DSC, ELS, TEM	99.65 - 99.75%	NA	142.33- 190.33n	NA	NA	[154]

		polymer  Tween 80 Triglycerides of the capric and caprylic acids Poly-ε-caprolactone Lecithin	(Nanoinjection and stirring)  Emulsification-diffusion method Homogenization	DLS, ELS, SEM DLS, DSC,	NA 2.23±1.42	NA PBS	250-650 nm. 255.9±1	NA NA	NA NA	[152]
		• Tween20	and ultrasonication	SEM, XRD	%		.63 nm			
Fibrous nanoparti	Nanofiber	<ul> <li>Polyethylene</li> </ul>	Electrospinning	DSC, FTIR, SEM	NA	NA	na	NA	MA	[165]
cles		• PLA	Electrospinning	SEM	NA	NA	na	NA	MA	[168]
		<ul><li>Maltodextrin</li><li>Alginate</li><li>Chitosan</li></ul>	Spray drying	DLS, SEM	NA	SSF, SGF, SIF	10.5- 942.8 μm	NA	Food application	[167]
	Nanotube	<ul><li>PVA</li><li>Polyethylene oxide</li></ul>	Electrospinning	FTIR, SEM, RSM	NA	NA	250 nm)	NA	NA	[166]
Gelatinou s nanoparti	Hydrogel	<ul><li>Rice starch</li><li>Xanthan gum</li><li>WPI</li></ul>	Microfluidization	CFSL	NA	SSF, SGF, SIF	450 nm	NA	NA	[142]
cles		<ul><li>WPI</li><li>Alginic acid</li></ul>	Microfluidization	CFLS, DLS, ELS	NA	SSF, SGF, SIF	285 - 660 mm).	NA	NA	[141]
		<ul><li>Pea protein isolate</li><li>Sunflower oil</li></ul>	Microfluidization	DLS	NA	SSF, SGF, SIF	3.16- 22.1µM	NA	NA	[144]
		<ul> <li>Soy glycinin</li> </ul>	Microfluidization	CFLS, DLS	NA	NA	1.5- 9.7 μm	NA	NA	[366]
		<ul> <li>Codium alginate</li> <li>Δ-glucono- lactone</li> <li>Tween 80</li> </ul>	Spontaneous emulsification	Bioaccessibilit y, DLS	NA	SSF, SGF, SIF	79-138 nm	NA	NA	[145]
		<ul><li>Ethylcellulose</li><li>Canola oil</li></ul>	Heating and stirring	Bioaccessibilit y	NA	SSF, SGF, SIF	NA	NA	NA	[143]

Corn oil, WP.     Rice starch	Hot homogenization	CFSL, Bioaccessibilit	NA	SSF, SGF, SIF	NA	NA	NA	[367]
• WPI	Ultrasonic emulsification	CFSL, DLS, ELS	NA	SGF	78- 252nm	NA	NA	[146]
<ul> <li>Sodium carboxymethy cellulose</li> <li>Kappa-carrageenan</li> </ul>	Cross linking	SEM	NA	SGF	NA	NA	NA	[122]
• SA • Calcium alginate	Freeze drying	SEM	NA	PBS	NA	NA	NA	[115]
• SC	Solvent- displacement method	DLS, ELS	NA	NA	45–127 nm	NA	NA	[139]

Note: NA: not applicable, AFM: atomic force microscopy, CFM: confocal fluorescent microscope, CLSM: confocal laser scanning microscopy, DLS: dynamic light scattering (used for size determination), DSC: differential scanning calorimetry, EE: encapsulation efficiency, ELS: electrophoretic light scattering (used for zeta potential determination), FRF: fractional residual fluorescence, FSM: fluorescence spectrophotometer, FTIR: Fourier transform infrared spectroscopy, NMR: nuclear magnetic resonance, PBS: phosphate buffered saline, SEM: scanning electron microscope, SGF: simulated gastrointestinal fluid, TEM: transmission electron microscope, XRD: X-ray diffraction, FSP: Florescence spectrophotometry, CM: confocal microscopy, FRF: fractional residual fluorescence, SRB: cellular proliferation assay (colorimetric) and MTT: cellular viability assay (colorimetric)

Table 2. Various factors that need to be considered prior to selecting a delivery system for encapsulating any bioactive agent

								<del>-</del>	
ENMS	Class of delivery	Subclass of delivery system	Ability to deliver lipophilic and	Physical stability	Biological stability	Biocompatibility	Drug targeting	Drug loading	Feasibility to be delivery system for
			lipophobic BA						beta carotene
Lipid	Self-	Liposome	Yes	poor	Poor	Good	Moderate	Low to	Poor
derived	assembled							moderate	
delivery	delivery								
system	system								
		Niosome	Yes	moderate	Poor	Moderate	Moderate	Moderate	Poor
	Particulate	Solid lipid nanoparticles	Only lipophilic	Good	Moderate	Good	Moderate	Moderate	Moderate

		Nanostructured lipid carriers	Only lipohilic	Good	High	Good	Moderate	High	Good
	Emulsion	Micro emulsion	Yes	Moderate	Moderate	Good	Poor	High	Good
		Nanoemulsion	Yes	poor	Moderate	Good	Poor	High	Poor
Polymer derived delivery system	Self- assembled delivery system	Starch based Micelle	Yes	Good	Good	Moderate	Poor	Poor	Good
		Protein based micelles	Yes	Poor	Good	Moderate	Moderate	Poor	Good
		Carbohydrate							Poor
		Hydrogel	Yes	Good	Good	Poor	Poor	Poor	Good
		Colloidal nanoemulsion	Yes	Moderate	Moderate	Good	Poor	High	moderate
		Nano emulsion	Yes	poor	Moderate	Good	Poor	High	Poor
		Molecular complexes	Only lipohophilic	Good	Moderate	Poor	Poor	Low	Poor
	Particulate	Protein inclusion complexes	Yes	Good	Moderate	Moderate	Moderate	Low	Poor
		Nanosphere	Yes	Good	Moderate	Moderate	Moderate	Moderate	Poor
		Microsphere	Yes	Good	Moderate	Moderate	Moderate	Low	Moderate
	Fibrous	Nanofiber	Yes	Good	Moderate	Moderate	Moderate	Low	Poor
	Capsular	Microcapsule	Yes	Good	Moderate	Moderate	Moderate	Low	Poor
		Nanosphere	Yes	Good	Moderate	Moderate	Moderate	Moderate	Poor

#### 5.2.1. Micelles

Micelles are distinguished as colloidal dispersions (with particle sizes ranging between 5 to 100 nm), related to a large family of dispersed systems containing particulate matter (called the dispersed phase), distributed within a continuous phase [179]. They belong to a class of amphiphilic colloids, which are generated spontaneously under a certain concentration and temperature from amphiphilic substances or surfactants, whose distinct regions confer opposite affinities towards a solvent phase. The hydrophobic regions of amphiphilic molecules form the core of the micelle while hydrophilic regions form the micelle's shell. When micelles are used as delivery systems for lipophilic beta-carotene in aqueous phases (food items and beverages), fat soluble molecules are imbibed on the micelle surface [180].

Several researchers have reproduced excellent reviews highlighting the chronological developments in the design, preparation, characterization and evaluation of polymeric micelles to attain efficient delivery of lipophilic drugs [181-184]. Micelles promise an array of advantages over polymeric nanoparticles, such as higher water solubility to lipophilic bioactive compounds [185], better penetration across physiological barriers [186], reduced toxicity and other adverse effects and effective bioactive drug distribution among tissues as well as organs [47,187]. These attractive attributes fascinated food technologists to exploit  $\beta$ carotene encapsulation. Chu et al. [188] encapsulated β-carotene in sodium caseinate-based micelles to correlate the changes in the particle size and  $\zeta$ -potential of the nano dispersions with their composition [188]. These β-carotene loaded micelle displayed a better stability than that of empty micelles [188]. β-carotene loaded α-lactalbumin micelles was not only found to be effective in protection β-carotene (40% to total encapsulated β-carotene) against thermal degradation (after 24 h of incubation at 60°C) but also demonstrated high cellular uptake of micelles encapsulating fluorescent dye by Caco-2 cell which also signifies higher absorption of encapsulated β-carotene [189]. These observations attracted food technologists to encapsulate β-carotene in micelles, using different food grade ingredients including casein, α-lactalbumin, β-lactoglobulin [190-194]. Low loading capacity, premature release of drugs and poor stability has nevertheless limited the use of micelles in food applications [47].

## 5.2.2. Micro/Nano emulsions

Oil-in-water nanoemulsions and microemulsions are two basic colloidal dispersion systems suitable for the delivery of lipophilic  $\beta$ -carotene for food applications. Literature also reports several techniques for the preparation of micro/nano emulsions, such as emulsion phase inversion [195], high pressure homogenization [196], microfluidization [197,198], supercritical fluid methods [199,200], spontaneous emulsification [201] and phase-inversion temperature [202].

Micro/nanoemulsions are recognized as colloidal dispersion systems of small liquid droplets, depending on the size ( $\leq 100$  nm for micro emulsion and  $\leq 50$  nm for nanoemulsion) [203]. Both microemulsion and nanoemulsion dispersion systems are formulated from the same ingredients (oil, water and surfactant), differing in their proportion. Due to various similarities in composition, dimension, structure and fabrication methods, it is necessary to clearly define differences. The difference between microemulsions and nanoemulsions was explained "an oil-in-water microemulsion is a thermodynamically stable colloidal dispersion consisting of small spheroid particles (comprised of oil, surfactant, and possibly cosurfactant) dispersed within an aqueous medium, while an oil-in-water nanoemulsion is defined as a thermodynamically unstable colloidal dispersion consisting of two immiscible liquids, with one of the liquids being dispersed as small spherical droplets (r < 100 nm) in the other liquid" [203]. The main difference between these two kinds of colloidal systems is thus their thermodynamic stability, i.e. microemulsion being thermodynamically stable while

nanoemulsion being thermodynamically unstable [203]. It is assumed that type of carrier oil and degree of saturation have significant impact on the  $\beta$ -carotene bioaccessibility. For that purpose β-carotene was encapsulated in three different nanoemulsion differing in their carrier oil (long chain fatty acid, medium chain fatty acid and orange oil) and it was found that nanoemulsion derived from long chain fatty acid had higher bioaccessibility (≈66%) as compared to medium chain fatty acid ( $\approx 2\%$ ) and orange oil (negligible) [92]. Tea polyphenols (TP) nanoemulsion was also fabricated to encapsulate  $\beta$ -carotene with the hypothesis that being an antioxidant itself the tea polyphenols could protect the encapsulated  $\beta$ -carotene. It was observed that addition of TP prevented the degradation of  $\beta$ -carotene during storage and improved the bioaccessibility of β-carotene after simulated oral and stomach digestion [204]. These observation have encouraged food food technologiest to develop noval nanoemulsion incorporating β-carotene [91,174,197,200,205-211]. Nethertheless, β-carotene incorporation into nanoemulsions and microemulsions for food applications has shown to be limited due to technical and practical hurdles, such as scarcity of food grade surfactants [212], complexity in fabrication method (most of them involving organic solvents), poor loading capacity and instability during storage [213].

### 5.2.3. Liposomes

In general, liposomes are spherical liquid structures with an aqueous core enveloped by as single (unilamellar) or multiple lipid bilayers (multilamellar liposomes) and promise high biocompatibility with animal tissues as they have demonstrated similarity to natural plasma membranes. According to the size, they are also defined as nanoliposomes ( $\leq$ 200 nm). The ability to incorporate both hydrophilic and hydrophobic compounds individually or simultaneously make liposomes most adopted delivery systems. Their broad application is also endorsed by their structure flexibility, size and composition. Various fabrication methods for preparation of liposomes have been developed, including lipid film hydration, micro emulsification, sonication, membrane extrusion, dried reconstituted vesicles, solvent dispersion method, detergent removal technique and supercritical fluid method [181,214-221].

Liposomes is one of most widely used delivery system to encapsulate and deliver lipophilic as well as hydrophilic bioactive compounds for cosmetics, pharmaceuticals and food industry [222,223]. It is assumed that the stability of encapsulated  $\beta$ -carotene can be further improved by the addition of antioxidants, though this may compromise the loading capacity. This assumption was varied for a study where  $\beta$ -carotene was found to be more stable (approximately 88%) when encapsulated along in liposome with vitamin C as compared to liposome without vitamin C (approximately 36%) during 30 days of storage at 4°C [224].

Liposomes are comprised of a hydrophilic core and a lipophilic crust, thus being able to incorporate bioactive compounds differing in their hydrophilicity. Hence, the solubility of any bioactive compounds governs its loading capacity as well as its location within the liposome [225]. For instance, the loading capacity of  $\beta$ -carotene was compromised when  $\beta$ -carotene was encapsulated in liposomes along with additional antioxidants such as lutein and lycopene [226,227]. Xanthan gum coated liposome has shown high retention ability for encapsulated  $\beta$ -carotene (2 molar  $\beta$ -carotene) during 90 days of storage at under refrigerated conditions [228]. L- $\alpha$ -Dipalmitoylphosphatidylcholine based liposomes was evaluated for release of  $\beta$ -carotene in simulated digestive system and it was observed that only 5–10% of total encapsulated  $\beta$ -carotene was released under gastric digestion conditions while 30–40% of total encapsulated  $\beta$ -carotene was released under intestinal digestive fluid [229]. Liposome have also been reported improved stability for encapsulated  $\beta$ -carotene [166,225,228,230-233].

Though liposome is the most widely adopted delivery system for food bioactive it also inherits several limitations such as hard to scale-up due to their vulnerability to shear, sedimentation, aggregation, fusion and environmental stress (osmotic pressure, pH, temperature, oxidation etc.), which may result in premature release and degradation of encapsulated β-carotene. To overcome this hurdle, food technologist came up with a proliposome strategy, nanometric version of liposomes, which offer more surface volume ratio, offer improved solubility to lipophilic compounds, enhance bioavailability, improve controlled release and enable site directed release of encapsulant, high stability during processing and storage [234]. Regardless of their great stability, proliposomes also carry technical limitations, such as the need of a vacuum or nitrogen atmosphere during their fabrication and storage [235]. It is also evident that for these reasons the food industry has not adopted this technique. Additional challenges with liposomes/proliposomes include low water solubility, short half-life, sedimentation, aggregation, fusion and phospholipid hydrolysis and/or oxidation, and high production costs remain high [236].

#### 5.2.4. Niosomes

To stretch the boundaries coupled with liposomes and proliposomes, researchers did come up with better alternatives, including noisomes. Niosomes are vesicles formed as a result of unfavorable interactions between s nonionic surfactants and water molecules resulting in closed bilayer structures and can also encapsulate lipophilic, hydrophilic and amphiphilic compounds [237]. Niosomes are preferred over liposomes as they offer better mucosal permeability, sustained and site-specific release and higher stability and are cost effective [238]. Niosomes promise higher chemical stability, simultaneous encapsulation of hydrophilic and hydrophobic bioactive compounds and reduced toxicity due to their nonionic nature [237]. It also resolves the issue coupled with liposomes such as challenges during sterilization, phospholipid purity and high cost [239]. In addition, the scale up of nisomes are also simple, as it does not require any specific conditions, organic solvents and other precautions such as vacuum [181,221,240-242]. In spite of the great potential, niosomes are not well adopted for food fortification. There is a scarcity of data on niosome encapsulation of  $\beta$ -carotene and only a single study is available for  $\beta$ -carotene in niosome where high stability for β-carotene was observed when it is encapsulated in noisome (20μM) than that of dissolved in tetrahydrofuran (10µM) after 96 h incubation at 50°C [243]. Furthermore, it is evident that not a single report was generated on applications of niosomes regarding encapsulating β-carotene for the food industry. This could be due to failure in resolving major stability issues, such as aggregation, fusion, leakage and sedimentation that were also observed in liposomes.

# 5.2.5. Solid lipid nanoparticles

To overcome the restriction inherited with available conventional delivery systems (emulsions, liposomes, niosomes and other polymeric delivery systems), nanotechnologists have evolved further to the next generation delivery system termed solid lipid nanoparticles (SLN), where the liquid lipid (oil) has been substituted by a solid lipid [244]. SLN promise exclusive properties such as a better interaction of phases at the interface, greater stability of encapsulated bioactive compounds, controlled and or/targeted drug release, large surface area and small size and ease in scaling up, which make it an promising delivery system for hydrophilic bioactives [245].

Studies are available on SLNs fabrication methods, such as evaporation or diffusion [246,247], high pressure homogenization at high or low temperatures (including cold homogenization and hot homogenization) [248], phase inversion methods [248], solvent emulsification [249], supercritical fluid (supercritical fluid extraction of emulsions (SFEE)

[250], homogenization or high-speed assisted with ultrasonication [251] and spray drying [252].

The potential of SLN to encapsulate and protect  $\beta$ -carotene was recognized during its initial development phase where  $\beta$ -carotene was incorporated into SLN to evaluate the effect of surfactants on the oxidative stability of encapsulated  $\beta$ -carotene. It was also observed that high melting surfactants better protected encapsulated  $\beta$ -carotene against chemical degradation [253]. It was assumed that the incorporation of protein molecules into SLN also improves the stability of encapsulated  $\beta$ -carotene. In order to verify these aspects, a study was carried out to assess the impact of whey protein on the stability of encapsulated  $\beta$ -carotene in SLN [254]. Though this study demonstrated better stability of  $\beta$ -carotene in SLN, there is sacristy of data on SLN for food applications. Only one paper was published addressing  $\beta$ -carotene loaded SLN for food fortification [255]. The scarcity of data on the incorporation of  $\beta$ -carotene loaded SLN in food matrices could be due to various limitations coupled with SLN, which involves low drug loading capacity, drug expulsion after polymeric transition during storage, particle size growth, random gelation tendency, unforeseen dynamics of polymeric transitions and sometimes burst releases [244,256].

## 5.2.6. Nanostructured lipid carriers (NLC)

The NLCs contain an unstructured solid-lipid core matrix, which consist of a mixture of liquid and solid lipids and an aqueous phase consisting of a surfactant or mixture of surfactants. Usually, liquid and solid lipids are a blend in a defined ratio that could vary from 70:30 to 99.9:0.1, while the surfactant content is kept between 1.5-5% (w/v) [257].

Current literature reports use of various fabrication methods for NLCs including, high pressure homogenization at high or low temperatures (including cold homogenization and hot homogenization) [258], solvent emulsification-diffusion techniques [251], use of supercritical fluid (supercritical fluid extraction of emulsion) [259], solvent emulsification evaporation [258], solvent displacement [260], solvent diffusion [261], phase inversion [262,263], melt emulsification [264], sonication [257], spray drying [264,265], and solvent evaporation [258].

Among the aforementioned preparations methods, hot homogenization process is preferred for NLCs fabrications when it comes in reference to food applications, as it does not involve organic solvents [266]. NLCs are partly crystallized lipid nano-delivery particles with an average diameter of ≤100 nm, dispersed in an aqueous phase containing an emulsifier. Typically, liquid and solid lipids are blended in a defined ratio. The unstructured/partially solid matrix produces interesting nanostructures, which improves the stability of entrapped bioactive compounds, offers high loading capacity and controlled/target release [267]. Moreover, it is believed that the addition of antioxidant aqueous and or lipid phase may increase the physiochemical stability of NLC as well as the entrapped bioactive compound.

To verify this hypothesis, various antioxidants (ascorbic acid, coenzyme Q, EDTA, tocopherol) were integrated into NLCs (formulated with lecithin and Tween 80), encapsulating astaxanthin also a carotenoid [268]. Ethylenediaminetetraacetic acid and tocopherol were shown to offer better oxidative stability to carotenoids (astaxanthin), while ensuring higher physical stability of NLC particles [268]. It was also hypothesized that the surfactant and emulsifiers utilized in NLCs preparations might negotiate the physiochemical stability of NLC particle as well as the encapsulated bioactive compounds. In order to verify this assumption, a study was devoted to formulate NLCs with two types of lipids differing in their melting point, i.e. low melting (LM) and high melting (HM) lecithins encapsulating tristerin and omega-3 fish oil [269]. The observation clearly suggested that NLCs formulated with HM-lecithin demonstrated greater inhibition ability against oxidation of omega-3 fatty acids than that of LM-lecithin [269].

Despite being a promising technique for drug delivery, NLCs remained underutilized for β-carotene encapsulation. Till date, only a few dedicated reports were produced dealing with β-carotene encapsulation in NLCs, showing the potential of NLCs to be used for food fortification purpose. In the first report, β-carotene loaded NLCs were fabricated by the hot homogenization method and the physico-chemical properties were evaluated [270]. Lacatusu et al. (2012) used high pressure homogenization method to encapsulate β-carotene in NLCs containing two natural oils (squalene and grape seed oil) [271]. The impact of the surfactant on physiochemical properties of NLCs encapsulating β-carotene was also studied [272]. Optimization of β-carotene encapsulation for NLCs using solvent diffusion methods was also carried out [261]. Similarly, β-carotene-loaded NLCs differing in the oil phase were also synthesized to evaluate the impact of the change in oil phase type on the physico-chemical properties of the NLCs [202]. A high-pressure homogenization method was adopted to encapsulate 9Z-β-carotene and total β-carotene in NLCs for its physico-chemical characterization and evaluated their stability during storage stability. It was observed that βcarotene-loaded NLCs stabilized both 9Z-β-carotene and total β-carotene not only from leakage but also from degradation against pH variations (pH 3.5, 4.5, 5.5, 6.5 and 7.5) also found high stable at 37°C during 21 days of storage [273]. Despite being the most advanced delivery method for processing of the sensitive bioactive compounds, applications of NLCs for  $\beta$ -carotene have been limited and its food applications are rather rarer.

# 6. Safety compliance and risks of β-carotene-nanoparticles

The customized properties of the discussed delivery systems, including the potential for bioavailability, better absorption and controlled release kinetics of the encapsulated bioactive compounds may also impart unseen risks to biological systems [177,274]. It is assumed that utilization of biodegradable or natural materials may curtail the health hazards as compared to polymeric nanoparticles which are either derived from synthetic polymers or involve toxic organic solvents during their fabrication process [274]. Due to the ambiguity on long- or short-term effects of direct or indirect employed nanoparticles in food systems, it is paramount to evaluate the impact of nanoparticles on human health [275]. With regard to food safety, the FDA has listed certain strategies in conjunction with nanoparticle-based food and food components for mass production [276]. Regardless of the potential health concern, at present no standardized legislation for incorporation of nanoparticles in food systems particularly for nanoparticles encapsulating β-carotene are not available. Nevertheless, several agencies and governmental bodies insist to embrace the safety concerns of nanoparticle-based food products in legislative guidelines [277]. The European Food Safety Authority (EFSA) has published an excellent report on the topic (https://www.efsa.europa.eu/en/efsajournal/pub/5327). This guideline provides an overview on the required information about physico-chemical characterization and the other data requirements. It also states about the performance of risk assessment of nanomaterials in the food and feed area including novel food, FCMs, food/feed additives and pesticides. This lack of universal legislations compelled duty-bounded policy-makers to outline a guideline specifically dealing with the nanoscale materials in the food system [278].

The potentially tailored bioavailability of encapsulated bioactive compounds in delivery system is key safety concern, specifically for bioactive compounds, or the nanodelivery systems which may become toxic beyond a certain dose. To scrutinize the safety aspects, the bioavailability of bioactive compound needs to be revaluated when it is encapsulated within nano-delivery vehicles, and reflections on alterations of the Recommended Daily Allowance (RDA) as well as the Tolerable Upper Intake Level (UL) of encapsulated bioactives are needed [279].

In addition, food scientists may also need to conduct studies addressing the safety concerns associated with nanoparticles, with special attention regarding; (i) the physiochemical characterization constraints of nanoparticles utilized in food items such as food additives, enzymes, flavourings, food contact materials (FCMs), novel foods, feed additives and pesticides [280]; (ii) development of the testing strategies to determine and characterize hazards transmitted via the ENMs i.e. assays for *in vitro* genotoxicity, absorption, distribution, metabolism and excretion and repeated-dose trials to study toxicity in test animals such as rodents[281].

In addition, the interaction between food items and nanodelivery systems should also be debated, which may result in producing radical oxygen species, photoreactions, among them. In December 2014, EU legislative bodies have insisted the food industries to mention on the label if nano-food products are sold [278]. According to this guideline, particles having one or more dimensions of either 100 nm or less and agglomerates above 100 nm exhibiting EMNs characteristics, and should be considered as ENMs. In conjunction with this, FDA has drafted guidelines which clearly defines ENMs derived foods as (i) agents or products having particlesizes within the range of 1 to 100 nm with at least in one dimension being within the nanoscale; (ii) agents or products exerting biological, chemical and physical characteristic associated with nano-scale material and also being under the nano-scale even though they are not nano-sized.

In addition to legislative guidelines, there are several moral responsibilities of the food processing manufactures including (i) evaluation of the changes imparted on the food materials, i.e. impurities and physiochemical properties; (ii) evaluation of the safety of food materials after modifications; (iii) submission of the regulatory assessment report to the legislative bodies such as FDA, FSSAI, EU, FASSAI etc.; (iv) identification and a statement about the regulatory concern due to the ingestion of the nanoparticle derived food items.

Apart from the US-FDA, several other regulatory authorities from various countries including Australia, New Zealand (FSANS) and Korea (MFDS) have issued their own guidelines [282]. These agencies counseled to conduct safety experiments (*in vitro* as well as *in vivo*) to evaluate the effect of nanoparticle-containing foods and publish the data, as well as to establish guidelines before releasing these nanoparticle containing foods to the food supply chain. Nevertheless, there is lack of specific guidelines regarding nanoparticles containing foods, thus it is high time as well as the need of hour that the legislative bodies should come together to frame a more universal guideline for nanomaterial-derived food products which can then be applied or further tailored to the different country.

## 7. Fate of β-carotene loaded nano delivery systems

 $\beta$ -carotene needs to be released in the GIT fluids to be taken up by the enterocyte for adsorption in GIT. The lipophilic nature of  $\beta$ -carotene limits its bioaccessibility to the cells due to the poor solubility. Lipid-based delivery systems, such as micelles, nano-/microemulsions, liposomes, niosomes, SLNs and NLCs, have recently been recognized for enhancing the bioaccessibility of many lipophilic vitamins and fat-soluble compounds including vitamin A, D and E [51,283-286]. The nature of the carrier oil utilized to fabricate LBDS also affects their encapsulation efficiency and bioaccessibility [92,287].

Literature has also witnessed improved bioavailability of lipophilic bioactive compounds for various polymer-based delivery systems, such as micelles, nano-/microemulsions, hydrogel, nano-/microsphere, nano-/microcapsules and nanofibers [283,284]. However, a lack of dedicated comparative studies for various compounds about the use of such polymer-based delivery systems creates a research gap. Comprehensive, comparative and rigorous research is needed with the use of various delivery systems for each

category of the compound to fill the research gap. Moreover, PBDS are well celebrated in pharmaceuticals to manipulate bioaccessibility by altering the solubility of β-carotene.

Figure 4 illustrates the primary routes for  $\beta$ -carotene absorption in the small intestine. LBDSs have been primarily adopted to encapsulate lipophilic  $\beta$ -carotene and tend to enhance their absorption [87,128,167,288-290]. Mixed micelles produced as a result of digestion of LBDSs and penetrating through the aqueous mucous layer were created to make  $\beta$ -carotene bioaccessible to brush bordered enterocytes for uptake, while PBDSs undergo various digestive enzymatic alternations to release  $\beta$ -carotene from polymeric nanoparticle and ensuing absorption by enterocytes. Further release of  $\beta$ -carotene from PBDSs and LBDSs and their packaging into chylomicrons inside the enterocytes is increased as a result of the high hydrophobicity [291,292]. These chylomicrons are lipid particles which are endogenously generated within the enterocytes using lipid components (free fatty acids, monoacyglycerols, and cholesterol) originating in part from mixed micelles produced as result of fat digestion [293]. Furthermore, these chylomicrons comprising  $\beta$ -carotene are then transported to the lymphatic circulation system to the liver for further processing.

Taken together, it is also believed that a fraction of encapsulated  $\beta$ -carotene from the delivery systems could be resistant to digestion and (i) may excrete from the GIT in undigested or semi digested condition or (ii) can penetrate the biological barriers of the intestine and enter the circulatory system [294]. The excretion of nanoparticles encapsulating  $\beta$ -carotene does not seemsa viable approach; thus such nanoparticles could not be commercially a realistic strategy for  $\beta$ -carotene encapsulation inthe food sector as they could pose some unidentified health risk [295]. On the other hand, since nanoparticles could penetrate the biological barriers, the immunological and toxicokinetic aspects of them are needed to be fully understood. Thus, it is advisable to carry out various investigations addressing the distribution of nanoparticles in cells and tissues, toxicological constraints and indecorous variation ofnanoparticle properties. In summary, a suitable design for deliverysystem can overcome the safety hurdles to a great deal and the safety can be gauged in direct approaches. Furthermore, being cognizant about the full features of safety concerns is the key to a suitable design and to make nanodelivery systems commercially viable.

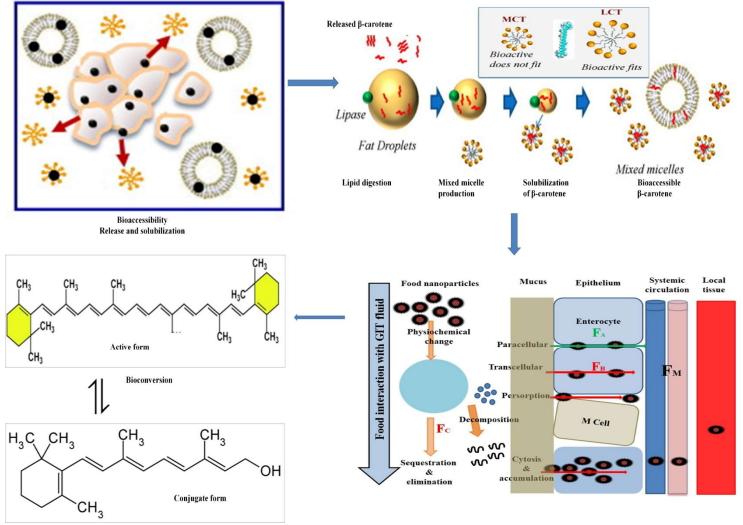


Figure 4: Factor influencing the bioavailability of  $\beta$ -carotene during absorption in the GIT. Where **A.** paracelluar absorption, **B.** M-cell uptake via Peyer's patches, **C.** Chylomicron- assisted enterocyte absorption.

#### 8. Conclusions

The selection of appropriated encapsulation techniques is the key to designing  $\beta$ -carotene delivery of food systems. Optimized doses of vital food components along with  $\beta$ -carotene can be achieved using suitable delivery system and food items which can be used as a platform to therapeutic as well as nutrient delivery.

Nanodelivery systems such as microemulsions, liposomes, SLNs, NLC, nanocapsules and nanospheres are impending carriers for β-carotene. Moreover, solvent evaporation, solvent displacement, microfluidization, thin film hydration and hot homogenization methods remain widely adopted encapsulation technologies for fabricating various β-carotenenanodelivery systems. Each delivery system has its own technical complications that affect the final properties of the resulting nanodelivery system. However, the majority of ENMs have reported sustained release, high loading capacity, lower possibility of encapsulant expulsion, low toxicity and high encapsulant protection; these data are extrapolated from pharmaceutical studies though. Further studies in this domain are surely warranted for food enrichment purpose. More focused studies are required to obtain better knowledge in designing delivery systems and to resolve the associated limitations such as the need for novel food grade polymers. Additionally, the safety of β-carotene-delivery systems in food needs to be routinely investigated. This includes in vivo and in vitro studies based data involving all available classes of available delivery systems. In summary, the stability of βcarotene-delivery systems in food matrices as well as its the delivery in the GIT need to be cautiously watched.

# 9. Future prospects and research gaps

The great potential of delivery systems in food items is the new normal which is coming into routine. In light of the global health issues, these applications in food seem imperious and indispensable to aid in combating diseases and promote healthy life. Several delivery systems have already been widely applied including micro/nano emulsion, NLCs, PBDS for food fortification in the items such as ice creams and beverages. However, information regarding safety concerns associated with the incorporation of new ingredients and technologies must be generated by accelerated *in vivo* and clinical trials to support both legislative makers and producers to provide the consumer evidence-based information. Public acceptance of delivery systems based food is gradually pending; ensuring its huge potential in many ways, such as personalized nutrition with novel functionalities for evolving human physical and mental capabilities and improving mood and satisfaction from nano-based foods.

After a comprehensive review of the literature, the gaps in the existing literature were pointed and these research gaps should be addressed by future focused studies. The future research prospects recognized from existing literature on delivery system encapsulating  $\beta$ -carotene are as follow:

- i. The field of designing nanodelivery systems for food applications is mainly trial & errors based. More interdisciplinary research needs be conducted to develop a set of universal methodology in development of delivery systems which could display high compatibility towards β-carotene, target food and their interaction with GIT fluids, cells, tissues and organs.
- ii. There is not a single report produced on the comparative assessment of the bioavailability of  $\beta$ -carotene-EMS to the above mentioned other nanodelivery systems. The data produced by devoted studies on bioavailability and health risk comparing various  $\beta$ -carotene loaded delivery systems (LBDS and PBDS), particularly PBDS, will aid in better understanding and designing suitable delivery systems for  $\beta$ -carotene.

- iii. The nature of the carrier oil (fatty acid chain length and degree of saturation) can also affect the biological fate of the lipid derived delivery systems [296]. Nevertheless, data is scarce with respect to LBDSs to draw a firm conclusion.
- iv. Although β-carotene loaded delivery systems display a high bioavailability, other lipophilic compounds and related carotenoids may manipulate the bioavailability of β-carotene. More studies, demonstrating the influences of lipophilic compounds present in the food matrix on the bioavailability of β-carotene loaded delivery systems. will be aid in a better understanding in designing better delivery system for β-carotene.
- v. Many researchers have argued that nanoparticles may enhance the bioavailability of  $\beta$ -carotene due to the transfer of intact nanoparticles across enterocytes. Nevertheless, no single study witnessed the penetration of food grade nanoparticles containing  $\beta$ -carotene across intestinal walls in the available literature.
- vi. Most of the delivery systems are fabricated based on extrapolated *in vitro and in vivo* pharmaceutical data. This cannot be applied to food grade nanoparticles; in particular polymer-based delivery systems.
- vii. Certain ingredients (EDTA, chitosan, fatty acid etc.) can manipulate the structure and integrity of the cell membrane. This is perhaps the least explored field and data generated on the effect of these ingredients on the cell membrane is necessary for better understanding and designing of efficient β-carotene delivery systems.
- viii. Various research studies displayed the improved permeability of cell membranes for certain kind of nanoparticles. Most of these are coupled with pharmaceutical formulations, containing certain nonfood grade materials to some extent. The same conclusion cannot be hypothesized for food grade nanoparticles. Thus, more devoted and rigorous investigations are needed to evaluate the impact of food grade nanoparticles on the penetration of cell membranes.
- ix. There is ambiguity on interactions between GIT fluids and nanoparticles encapsulating  $\beta$ -carotene. It is sensible to debate how the bioavailability of  $\beta$ -carotene is influenced when it is encapsulated in available delivery systems.
- x. The real time or perceived risks endorsed within the transfer of intact particles across the intestinal walls into the systemic circulation and buildup of particles or  $\beta$ -carotene in organs and the incidence of very high peak concentration of  $\beta$ -carotene in the blood. Since reliable data signifying toxicity or risks in real time are not present in the current literature, this is a debate with various unknown.
- xi. The role of digestive enzymes in the release of  $\beta$ -carotene from delivery system as well as on its bioavailability is not fully renowned. The assessment addressing the effect of enzymes individually or in array and their concentration on bioavailability  $\beta$ -carotene from delivery systems will aid in better knowledge for designing suitable delivery system.
- xii. There is ambiguity regarding the kinetics of nanoparticle transfer from food matrices GIT fluid as well as from GIT fluid to enterocyte. More focused data need to be generated to understand the transfer kinetic of nanoparticles, which will result in a better understanding for designing better delivery system for β-carotene for food applications.

#### **Author Contributions**

The first author Vaibhav Kumar Maurya and second author Amita Shakya contributed equally in the conceptualization, designing of the work, original draft preparation and method of the data collection. Prof. Manjeet Aggarwal and Prof. K.M. Gothandam provided with the critical comments on the concept and structure of the work, approved and supervised the present project. They also edited the drafts of the manuscripts and gave critical comments for improvisation. Prof. Sunil Pareek has reviewed and approved the submitted version and edited the present version of the study. Prof. Torsten Bohn reviewed the draft of the work and gave his critical comments to make the review enriched.

#### **Conflicts of Interest**

The authors declare no conflict of interest in the choice of this project; study design, data collection, analyses or interpretation, writing of the manuscript and in the decision to publish the results.

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