Title: Role of Nanoscience in Agriculture

Praveen Kumar

Department of Biochemistry, COBS&H, CCS Haryana Agricultural University, Hisar - 125004

Abstract

Nanoparticles are widely used in agricultural sector because of their distinctive properties. Studies have

shown the influence of nanoparticles on plant growth and production. Nanoparticles acts as effective

carriers in the delivery of agrochemicals to plants. They provide site targeted delivery of nutrients and thus,

prevents wastage of nutrients applied for plant growth and productivity. Bioremediation of pollutants is an

emerging technology which provides bio-nanomaterials for protection of agriculture from pollution. The

aim of this review is to present and focus on the latest techniques used for reduction of environmental

pollution and improved agricultural production. This review speculates about biosynthesis of nanomaterials

from different sources like plants, fungi and bacteria along with chemical and organic synthesis from

carbon, silver and gold. The role of nanoscience in detecting plant diseases and removal of heavy metals.

Application of Nanoscience in storing, production, processing and transport of agricultural materials. It is

also emphasized that Nanoscience may transform agriculture through innovation of new techniques like

Precision farming, improvement of plants to engross nutrients, targeted use of inputs, detection and control

of diseases and withstand environmental pressures. Further, efforts have been made in describing that

nanoparticles may act as a better substitute for agricultural plant's growth and nutrition improvement by

lowering the content of pollutants and pre-detection of diseases in plants. The biosynthetic route of

nanomaterial synthesis could emerge as a better and safer option for environmental pollution reduction.

Thus, nanoscience may increase agricultural production to feed huge population in near future.

Keywords: Bioremediation; Nanomaterials; Biosynthetic; Sustainable; Ecofriendly

1. Introduction

1

The challenging task of 21st-50th century is to clean the environment in an ecofriendly and sustainable way. So, that the agricultural production can be increased. Although it is a new area but still it is rapidly growing. Microbes are the emerging Nano-factories and has potentials for environmental cleanup methods. Most of the biogenic nanomaterials yielded satisfactory results. They can increase the production in agricultural system. India is an agricultural dependent country. Its economy is also dependent on agricultural production. All animals, directly or indirectly, depends on agricultural production, for their food supply. Thus any change in agricultural production may affect the development and economy of any country like India. Diseases, lack of nutrients and pollution are three major reasons of reduction in agricultural production. Among diseases, microbial diseases are becoming a vast problem in agriculture and the chemicals used in the remediation of these diseases are producing problems in animals that are consuming them. These toxic chemicals (fungicide, herbicide and bactericide) causes major loss of animal species like honey bee and birds which are beneficial for agricultural production [1]. These toxic chemicals also causes loss of beneficial bacteria's from soil which are involved in decomposition of various waste materials and thus increasing soil fertility. Now a days farmers adding nutrients more than their requirement by the plants [2]. This extra quantity of nutrients are becoming wastage and directly affecting the farmer's income from agricultural production. Among pollutants, the heavy metals are becoming a major problem in agricultural production. Heavy metals causes harmful effects to the agricultural plants and reduces their production. Heavy metals enters to animals through food sources from plants and causes toxicity in them also. In short, if we are gaining in one hand but we are losing more on the other hand. So, as an outcome we are losing a lot. This loss of agricultural production can be prevented by employing nanoparticle science technology for providing nutrients, diseases protection and protection from heavy metals pollution. Nanoscience or nanotechnology is a branch of science dealing with studies involving nano size (10<sup>-9</sup> size of meter) particles. Nano can be applied to any unit of measure for example you can consider nanograms and nanoliters. The concept of nanoscience was first time driven by Richard Feynman in 1959 and from then it was growing very fast in all areas of research and development worldwide [3]. Nowadays, nanoscience is the subject of extensive research in all areas [4]. It may bring the next industrial as well as

the green revolution in the world [5]. Nanoscience provides many environmental benefits to agriculture. These may be divided into nano-fertilizer, disease remediation and reduction of pollutants like heavy metals. The specific nanoscience application for onsite remediation of soil, air, ground water and wastewater treatment are also in development. The size of nanoparticles has significance on its physiological and biochemical properties. It behaves completely different from its main part. The reduction in the size changes redox properties, thermodynamic properties and internal cohesive forces of the particle or material. It causes easy delivery and interaction of the particle with the action site [4]. Nanoparticle synthesis using microorganism comes under green nanoscience technology. It may provide a better alternative for diseases control in plants [6]. Nanoscience is a persistent solution for our environment. It is needed to utilize new methods like nanoparticle science for better development in agriculture. Nanoscience can help us to design new materials with peculiar properties and reproducibility for agricultural development [7]. In this concern, scientific community should continuously focus on a very challenging and relevant research's direction, which is the development of nanoparticle capable for heavy metal remediation from soil and water. The concept of nanoscience may help us to understand the mechanism of transformation of heavy metals in to plants and animals, more accurately at molecular level. The synthesis of materials at nanosize dimension will facilitate us to design new useful systems for heavy metal removal from plants. This review will provide information of different systems and methods which may be utilize to overcome heavy metal toxicity among agricultural lands across the globe.

### 2. Nanoparticles

The particle of very small size (<100 nm), of any material are called nanoparticle of that particular element. They are atomic or molecular aggregates of 1-100 nm dimension and their physicochemical properties modifies drastically as compared to bulk material. The properties of nanoparticles depends and may varies with their size i.e. the size of nanoparticles define their properties. Nanoparticles are more mobile and more reactive in nature. They are roughly divided in to organic and inorganic. The first category organic nanoparticle comprise carbon nanoparticle. The second category inorganic nanomaterials comprise

magnetic nanomaterials, noble metal nanoparticles like silver, gold nanoparticle and semiconductor nanoparticles like titanium dioxide, zinc oxide. Nanomaterials are also produced naturally by volcanic and lunar eruptions [8]. These are known as natural nanoparticles but there are other class of nanoparticles which are produced by anthropogenic activities like exhaust of diesel, coal combustion and fumes of welding works; are known as incidental nanoparticles. The nanoparticles synthesized by man are called as engineered nanoparticles. The man made or engineered nanoparticles includes nanozinc, nanoaluminium, TiO2, ZnO, quantum dots etc. [9]. Smaller size of nanoparticle make them more useful for used as a sensors which can be easily deployed in remote locations. Currently, nanomaterials are being proved successful both in efficiency as well as cost effective and also environmental friendly way in utilizing them as an alternatives for current treatment materials for agricultural remediation [10], [11]. The use of biologically synthesized nanomaterial has been increasing in the field of agriculture and medicine because of their stable, ecofriendly and cost effective nature [12]. They should be utilized in wider applications. The biological synthesis of nanomaterials is more advantageous over other methods due to ease of fast synthesis, ease of toxicity control, ease of size control, cost effective and environmental friendly methodology [13]. Nanoscience is being utilized extensively for protection of agriculture from bacterial disease and pollutants including chemicals like heavy metals [14].

### 3. Unique properties of nanoparticles and their contribution in plant production and growth

The properties of nanoparticle changes with its size. Nanoparticles shows different and unique properties when compared with their bulked product. Some examples are sown here below – Table 1. These properties showed a way to encapsulate and slow release of agrochemicals used in pant growth and protection. The major roles of nanoparticles in increasing agricultural production are summarized in Table 2.

**Table 1.** Properties of nanoparticles along with their bulked forms.

Materials	Nanoparticle	Bulked

Copper	Hard	Soft
Gold	Chemically active	Chemically inactive
Silicon	Conductor	Insulator
Titanium dioxide	Colorless	White

# 4. Synthesis of Nanoparticles

Traditionally, there are several methods for generating nanoparticles, including gas condensation, abrasion, chemical precipitation, ion embedding, pyrolysis, and hydrothermal production. In abrasion, macro- or micro-scale elements are crushed in a ball mill, an earthly ball mill, or other size-reducing apparatus. The resulting particles are air categorized to recover nanoparticles. In pyrolysis, a vaporous precursor (liquid or gas) is forced through an orifice at high pressure and burned. The resulting solid (a version of soot) is air categorized to recover oxide particles from by-product gases. Old-style pyrolysis often results in aggregates and agglomerates rather than single primary particles. Ultrasonic nozzle spray pyrolysis (USP) on the other hand aids in preventing agglomerates from forming. These methods are costly for nanoparticle synthesis, there are other methods to produce nanoparticles cheaply as:

# i. Biogenic production of various nanoparticles

The microorganisms and plant extracts are used generally for a cheaper synthesis of nanoparticles (Figure 1). The main reaction involved in the biosynthesis of the nanoparticle is oxidation/reduction reaction. It is a bottom-up approach for the synthesis of the nanoparticle. The biochemical antioxidants in plant sample and biochemical enzyme in the microbial sample are responsible for the reduction of metal into their nanoparticles. Currently, there are million tons of production of nanoparticles worldwide and is probable to increase intensely in near upcoming [15]. The nanomaterials are materials having a particular dimension between 1 - 100 nm, so along with vascular plants; the microorganisms like bacteria, yeasts, algae, fungi, and actinomycetes may be used for nanoparticles biosynthesis [16].

# ii. Nanoparticles produced by plants

Does not meet shape, size and size distribution criteria

Nanoparticle synthesis from plant extract is called a green synthesis of the nanoparticle [17, 18]. Currently, it is gaining prominence because of the single-step involved in biosynthesis. So, it is a time-saving process along with no toxicants and presence of natural capping agents [16]. Plants material are easily available, they are safe to handle and possess a large variability of metabolites that aid in reduction. All these reasons make plants more advantageous over others for nanoparticle synthesis.

Generalized flow chart for Nanobiosynthesis

# Bio-reductant from bacteria, fungi, or plant parts + Metal ions (Maybe enzyme/ phytochemical) Reactant conc., pH, Kinetics, Mixing ratio, Metal nanoparticles in solution UV visible analysis (SPR) Purification and recovery Nanoparticle powder SEM, TEM, DLS, XRD Physicochemical characterization Meet shape, size, and size

distribution criteria

Biofunctionalization

End use

**Figure 1:** General outline for biosynthesis of nanoparticles

Modify process variables

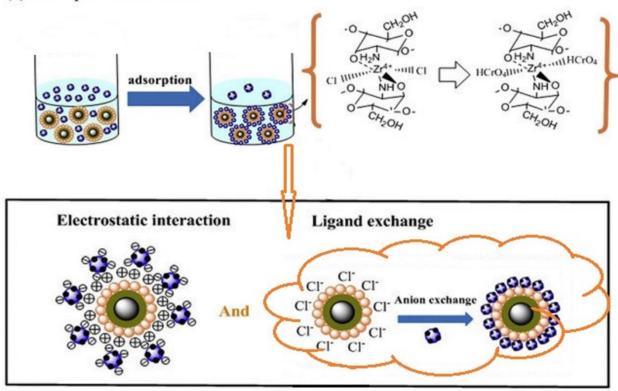
Currently, there are many plants are being investigated for their importance in nanoparticles synthesis (Table 3). Phytochemicals consume less time for the reduction of metal ions, compared to fungi and bacteria. It shows plants materials as a better candidate for the biosynthesis of nanomaterials, compared to bacteria and fungi. By utilizing plant tissue culture and downstream processing techniques, one can

synthesis nanomaterials of metallic and oxide at industrial scale if the metabolic status of the plants are addressed appropriately. It is found that the nature of nanoparticles produced from this method varies with the plant to plant, mode of application, size, and quantity [19]. It is revealed by the scientist that the investigation on nanoscience basically for plants is in its early stages; more depth study is required to reveal about biochemical, physiological and molecular mechanisms of plants in concern to nanomaterials and more work is required to explore the mechanism of nanomaterials actions, their behavior towards biomolecules and their effect on gene regulation and expressions in plants. Generally, plant based nanoparticles involve three mechanisms viz. adsorption, reduction and desorption mechanisms in heavy metal removal process from plants, soil and water (Figure 2.).

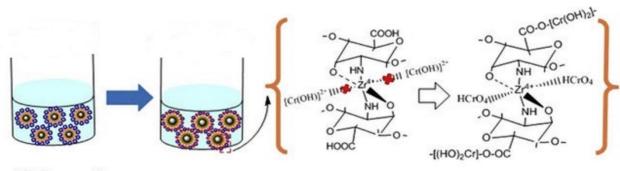
**Table 2.** Contribution of nanoscience in plant production and growth.

Sl.	Contribution of Nanoscience
No.	
1.	Nanomaterials (NM) can be effectively used in plant germination and growth.
2.	The carbon nanotubes can be used as regulators of seed germination and plant growth.
3.	Multiwall carbon nanotubes (MWCNTs) have the ability to enhance the growth of tobacco cell
	culture by 55 - 64% when compared to control at a wide range of concentrations from 5 - 500 µg
	$ml^{-1}$
4.	At low concentrations, activated carbon enhanced cell growth
5.	Nanotechnology provides controlled and efficient use of fertilizers, pesticide and other
	agrochemicals for better growth and production of plants [20].
6.	Chitosan nanoparticles are proved more effective antifungal agents against pathogen (Fusarium
	solani).
7.	Silver nanoparticles acts as antimicrobial agents and protects plants from microbial diseases.
8.	Copper nanoparticles may acts as bio-pesticides and protects plants from pests.
9.	Zinc based nanoparticles inhibits fungal growth by damaging conidiophores.

# (1) Adsorption mechanism



# (2) Reduction mechanism and readsorption



# (3) Desorption



**Figure 2.** General mechanisms involved in heavy metal removal by plant based nanoparticles from plants, soil and water [21].

**Table 3.** Plants used for the synthesis of nanoparticles.

Nanoparticles	Plants species	References
Gold and silver nanoparticles	Citrus sinensis	[22]
	Diopyros kaki (Persimmon)	[23]
	Pelargonium graveolens	[24]
	Hibiscus rosa sinensis	[25]
	Coriandrum sativum	[26]
	Emblica officinalis	[27], [28]
	Phyllanthium	[29]
	Mushroom extract	[30]
Gold nanoparticles	Terminalia catappa	[31]
	Banana peel	[32]
	Mucuna pruriens	[33]
	Cinnamomum zeylanicum	[34]
	Medicago sativa	[35]
	Magnolia kobus and Dyopiros	[36], [28]
	kaki	[37]
	Allium cepa L.	[38]
	Azadirachta indica A. Juss.	[39]
	Camellia sinensis L.	[40]
	Chenopodium album L.	[41]
	Justicia gendarussa L.	[42]
		[43]

	Macrotyloma uniflorum (Lam)	[1]	
	Verde	[44]	
	Mentha piperita L.	[31]	
	Mirabilis jalapa L.	[45]	
	Syzygium aromaticum (L)		
	Terminalia catappa L.		
	Amaranthus spinosus		
Silicon-Germanium (Si-Ge)	Freshwater diatom	[46], [47]	
nanoparticles	Stauroneis sp.		
Silver nanoparticles	Elettaria cardamomom	[42]	
	Parthenium hysterophorus	[48], [49]	
	Ocimum sp.	[2]	
	Euphorbia hirta, Nerium	[50]	
	indicum	[24], [51], [38]	
	Azadirachta indica	[27]	
	Brassica juncea	[52]	
	Pongamia pinnata	[53]	
	Clerodendrum inerme	[54]	
	Gliricidia sepium	[55]	
	Desmodium triflorum	[56]	
	Opuntia ficus indica	[15]	
	Coriandrum sativum	[57]	
	Carica papaya (fruit)	[58]	
	Pelargoneum graveolens	[59]	
	Aloe vera extract	[60]	

	Capsicum annum	[61]
	Avicennia marina	[3]
	Rhizophora mucronata	[62]
	Ceriops tagal	[11]
	Rumex hymenosepalus	[63]
	Pterocarpus santalinus	[64]
	Sonchus asper	
Palladium nanoparticles	Cinnamomum zeylanicum	[20]
	Blume.	[38]
	Cinnamomum camphora L.	[65]
	Gardenia jasminoides Ellis.	[66]
	Soybean (Glycine Max) L.	
Magnetic Nanoparticles	Aloe vera	[67]
Lead Nanoparticles	Vitus vinifera L.	[14]
	Jatropha curcas L.	[68]
Nanoparticles of silver,	Brassica juncea, Medicago	[69]
nickel, cobalt, zinc and copper	sativa and Helianthus annuus	
Platinum nanoparticles	Diopyros kaki	[35]
	Ocimum sanctum L.	[21]

# iii. Nanoparticles produced by bacteria

Bacteria possess the peculiar property of mobilization and immobilization of elements and in some cases, they may precipitate metals even in nanometer size. Therefore, Bacteria are called bio-factory for production of nanomaterials like silver, gold, palladium, titanium, magnetite, cadmium and platinum. It is a new method of nanoparticle synthesis. In this method, bacterial enzymes are involved in catalyzing a

specific breakdown reaction and produce nanoparticles [70]. The polysaccharides, vitamins, enzymes, biodegradable polymers and biological systems may be utilized for the synthesis of nanoparticle in the same way. The enzymes which are produced in extracellular secretion possess more advantage of the production of a large number of nanomaterials ranging size between 100 – 200 nm in a comparably pure form, and free from other materials like cellular proteins. These nanomaterials may be further purified by filtration easily. The metal binding and S-layer property of bacteria make them more useful in the technical application of bioremediation and nanoscience. The different types of bacteria used for nanomaterials are shown (Table 4). The mechanism involved in silver nanoparticle synthesis by *Pseudomonas aeruginosa* is explained in Figure 3. The quality of nanoparticles produced by these techniques may be controlled by optimization of bacterial growth controlling parameters like enzymatic activities and cellular activities. Thus, there is a lot more to study to make proper use of this technique for nanomaterial synthesis. It is a more appealing field because of an ecofriendly, more stable, safe and less expensive process.

**Table 4.** Bacteria used for the synthesis of nanoparticles.

Nanoparticles	<b>Bacterium species</b>	References
ZnS nanoparticles	Sulphate reducing bacteria of the	[71], [72], [73], [74], [75], [76]
	family Desulfobacteriaceae	
CdS nanoparticles	Clostridicum thermoaceticum	[77]
	Klebsiella aerogens	[78]
	Escherichia coli	[79]
As-S nanotubes	Shewanella sp.	[80], [81]
Palladium nanoparticles	Desulfovibrio desulfuricans	[82]
	NCIMB 8307	
Gold nanoparticles	Alkalothermophilic actinomycete	[5]
	Thermomonospora sp.	[5]
	Pseudomonas aeruginosa	[83]

	Lactobacillus strain	[71]
Magnetic nanoparticles	Magnetosirillium magneticum	[19]
	Sulphate reducing bacteria	[84]
Gold nanowires	Rhodopseudomonas capsulate	[85]
Silver nanoparticles	Bacillus cereus	[25]
	Oscillatory willei NTDMO1	[86]
	Escherichia coli	[87]
	Pseudomonis stuzeri	[88], [89], [90]
	Bacillus subtilis	[91]
	Bacillus sp.	[92]
	Bacillus cereus	[93]
	Bacillus thuringiensis	[4]
	Lactobacillus strains	[71]
	Pseudomonas stutzeri	[88]
	Corynebacterium	[55]
	Staphylococcus aureus	[94]
	Ureibacillus thermosphaericus	[95]

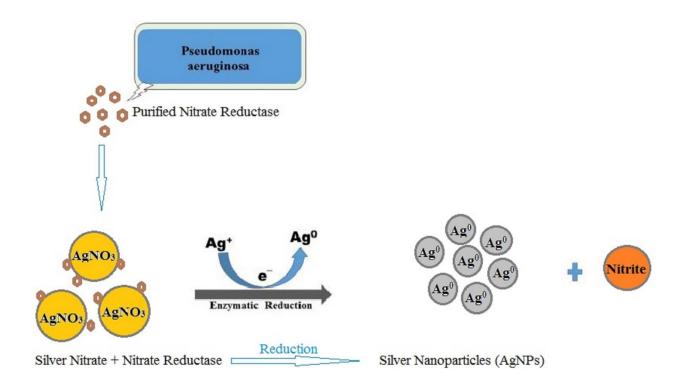
## iv. Nanoparticles produced by yeast and fungi

Fungi produce a larger amount of extracellular enzymes than bacteria, which shows more influence on nanomaterial synthesis. The documented mechanistic aspect governing nanoparticle synthesis by using fungi are explained here. The nanomaterials with well-defined dimensions can be synthesized by using fungi. Compared to bacteria, fungi can be used for larger production of nanomaterials because fungi secrete a larger amount of proteins which directly increases the formation of nanomaterials [19]. Moreover, the protein isolates from fungi can also be used for nanoparticle formation. The enzymes produced by fungi

during metal nanoparticle synthesis reduces salts to its metallic solid nanomaterials by the catalytic effect [24]. It is major weaknesses of this synthesis method by microorganism and if need be corrected if this method is to participate with other methods. The other disadvantage of this method is that it generates nanomaterials at a slower rate than detected when plant extract is used. The other things to be considered for the fungal synthesis of nanomaterials is that fungi should possess certain properties like high production of required metabolites/enzymes, high growth rate, easy handling in large scale production and require low cost for production techniques [47]. Generally, fungi are observed as the best source for nanomaterial synthesis over other biological systems because of its easy handling, cost-effectiveness, and wide diversity. Genetic engineering may be employed for an ecofriendly and improved method of nanoparticle synthesis [96]. In terms of quality of nanomaterials and utilization of nutrients for microbial growth, yeast strains have assured benefits over bacteria [12].

### 5. Role of Nanoscience in detecting plant diseases

There is a strong need to detect plant diseases in the early stage so that it can be protected from microbial agents and a great loss of food and agriculture can be prevented. Nanoscience can be utilized in real-time monitoring systems, distributed throughout the fields, to monitor soil conditions and crop conditions, by providing or connecting an autonomous nanosensor to a GPS system. This technique reduces the time required for observation of the crop in fields. It provides results within a few hours that are more accurate, simple and portable. This technique does not involve any complicated procedures and is so simple to be utilized by the farmers. The union of nanoscience with sensors technology may create new methods which allow early detection of diseases development and environmental changes. Nanoscience provides a new way of changing crop management methods and improving plant nutrition. It reduces the problem of leaching, photolytic degradation, hydrolysis, and microbial degradation. It has reduced the wastage of nutrients by providing the way in which a very low amount of required minimal effective concentration, reach the target site of crops.

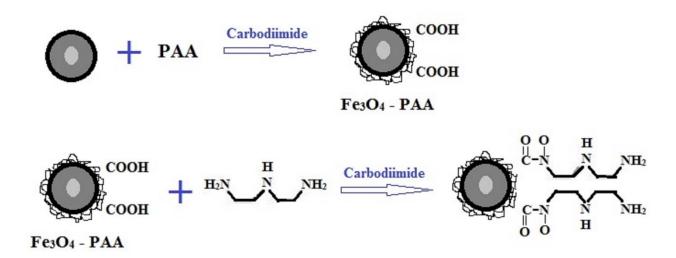


**Figure 3.** Mechanistic illustration of silver nanoparticle synthesis by bacterial involvement.

### 6. Nanomaterials for heavy metals removal

Heavy metals are distributed among soil, air, and water by natural or human activities like weathering, biodegradation, volcanic eruptions and mining, agrochemical industries; respectively. Heavy metals are increasing with human civilization. Any element having an atomic weight greater than 63.5 and has a specific gravity greater than 5.0 can be called heavy metal [97]. Heavy metals are divided in to toxic metals (Hg, Cr, Pb, Zn, Cu, Ni, Cd, Co, Sn); precious metals (Pd, Pt, Ag, Au, Ru) and radionuclides (U, Th, Ra, Am) [98]. These metals are non-biodegradable and very difficult to eliminate naturally from soil, air, and water. Generally, all metals are toxic when their values increase from its permissible limits. Heavy metals cause toxic effects in human's body when it exceeds the permissible limits [99]. Various technologies have been used for heavy metal remediation like precipitation, reverse osmosis, advanced oxidation, coagulation, photoelectrochemistry, and flocculation. Each differs in the degree of remediation efficiency [9], [100]. Currently, nanoscience has provided a great approach for heavy metal toxicity remediation by providing different nanomaterials. This technique is very beneficial over other techniques,

some of the benefits are like increased efficiency, reduced consumption and substation of more abundant and less toxic materials than ones used before. Generally adsorbing nanomaterials have shown great and effective work against heavy metal toxicity removal [12]. Nanoparticles for heavy metal removal are divided into carbon-based nanoparticles (carbon nanotubes and graphene) and inorganic nanoparticles (based on metal oxides and metals). Different combination of nanoparticles is also developed for heavy metal removal. The disposal of nanoparticles may lead to release in the aquatic ecosystem which may prove harmful for them, this should be solved. The plants utilized in heavy metal removal are summarized in Table 5. The interaction of nanoparticle with plant's biomolecules and removal of heavy metal cations and anions is explained in Figure 4.



**Figure 4.** Iron oxide nanoparticle generation and activation of poly-acrylic acid (PAA) by aminofunctionalization on iron oxide nanoparticle as a new approach in making magnetic nano-adsorbent for metal cation and anions removal.

### 7. Carbon-based nanomaterials

### i. Activated Carbon

Activated carbon is produced from freely available carbonaceous precursors such as wood, coal, coconut shells and agricultural wastes. They possess high porosity and large surface area. They are broadly

used in wastewater treatments. These nanomaterials possess high potential for removal of organic and inorganic pollutants from contaminated water. Activated carbon has a significant weak acidic ion-exchange character which is involved in the removal of trace metal contaminants from wastewater by adsorption. H. X., Zhong. [101] in his work observed the adsorption of pentavalent arsenic on granular activated carbon. The activated carbon synthesized from coconut tree sawdust was observed as an adsorbent for Cr (VI) and removes it from wastewater [102]. Activated carbon is investigated for adsorption and stability of mercury on it [103]. The adsorption ability of powdered activated carbon was tested against Cu (II) and Pb (II) and was found 0.85, 0.89 µmol g<sup>-1</sup>; respectively [19]. A new sodium polyacrylate grafted activated carbon was synthesized by gamma radiation to enhance functional groups on its surface. This procedure may be applied to other adsorbents to enhance the efficiency of metal ions adsorption by activated carbon [104]. Because of low cost and high adsorption ability of activated carbons, they are appropriate materials for heavy metal removal.

### ii. Carbon nanotubes

Carbon nanotubes are like pillars for nanoscience. They are generally used in nanomaterial synthesis. Carbon nanotubes are divided into two groups as single-walled and multiwall nanotubes [105]. Their potential as a superior adsorbent for heavy metals has been studied widely [106], [107], [108], [109]. Currently, the scientists are focusing on nanomaterial synthesis which shows specific adsorption towards a single solute in the target material and ignoring the interaction potentials between mixtures that may affect adsorption process [10], [110], [111], [112], [113]. Carbon nanotubes have been used as a better adsorbent for removal of heavy metals like Pb<sup>2+</sup>, Cd<sup>2+</sup>, Ni<sup>2+</sup>, and Cu<sup>2+</sup> [114]. The mechanism involved in heavy metal removal by carbon nanotubes comprises of electrostatic attraction, adsorption, precipitation and chemical interaction between the metal ion and functional group present on the surface of carbon nanotubes. The drops in pH of the solution, when metal binds to carbon nanotubes indicate the transfer or release of the proton from carbon nanotubes where metal binds to it. Sometimes phenolic groups are also involved in heavy metal binding to carbon nanotubes. Moreover, an increase in metal ion concentration also increases

the pH values of the solution and indicate the adsorption of heavy metal [10]. Various carbon nanotubes have been synthesized to improve adsorption capability and minimize weaknesses under different environmental circumstances, for the removal of heavy metals. The combination of magnetic properties of iron oxide with adsorption properties of carbon nanotubes may prove rapid, effective and promising technology for hazardous heavy metals removal [64]. The maximum adsorption capacity of different heavy metals on carbon nanotubes differs among different heavy metals.

**Table 5.** Plants utilized for nanoparticle generation and their role in bioremediation of heavy metals.

Plant Species used for Nanoparticle generation	Best Heavy metal Bioremediation	Reference
Noaea mucronata	Pb (98%), Zn (79.03%), Cu	[15]
	(73.38%),	
	Cd (72.04%) and Ni (33.61%)	
Euphorbia macroclada	Pb (92%), Zn (76.05%), Cu (74.66),	[115]
	Cd (69.08%) and Ni (31.50%)	

### iii. Graphenes

Graphene-based adsorbents have great adsorption capacity for removing heavy metal pollutants from water and soil [116], [117]. Many researchers are focusing on and putting their efforts in investigating the potential application of graphene oxide nanomaterials in eliminating heavy metals from water and soil. The adsorption properties of graphene nanomaterials may be improved if graphene nanomaterial synthesized by employing the Hummers method which incorporates many oxygen-containing groups like carboxyl and hydroxyl groups on the surface of nanomaterials [97]. These functional groups increase the surface area of graphene oxide nanomaterial which further increases the adsorption capacity of nanomaterials for heavy metals. The mechanism of heavy metal removal by graphene nanomaterials is the same as was discussed for carbon nanotubes. The high dispersion nature of graphene oxide nanomaterials makes it more advantageous over carbon nanotubes. The other nanomaterials that have been studied include

carbon fiber, carbon, nanoporous carbon and their complexes [118], [119]. They show exceptional adsorbent capacity for heavy metal removal and the rates of removal are persistent with pollution control requirements. Their efficiency depends on well-developed internal pore structures, large specific surface areas, and the presence of functional groups on surface nanomaterials.

### iv. Other nanomaterials

Along with carbon-based nanomaterials, other nanomaterials such as iron-based nanomaterials, photocatalytic nanomaterials, silica, alumina, and manganese dioxide nanomaterials are also explored for heavy metal pollutant removal. These nanoparticles have been engrossed for their outstanding adsorption properties. Some other nanomaterials which are utilized in the removal of heavy metal from soil, air and water are TiO<sub>2</sub> -Based Nanomaterials, ZnO –based nanomaterials, Iron oxide nanomaterials, bimetallic nanoparticle, and Metal based-nanomaterials [120]. All possess excellent efficiency for the removal of heavy metal pollution from agricultural lands and irrigation water. But still, a lot of further research is required on nanoscience to remove heavy metals more efficiently and significantly from agricultural lands and water.

Nanotechnology has an enormous role in the field of nano-fertilizer. Nano-based fertilizer can do balanced nutrient supply, stimulated crop growth, improved crop quality, regulate nutrient migration to the environment, reduction of cost of production due to its less requirement per hectare [121], helps in precision farming, increased nutrient use efficiency of plants, increased drought and salt stress tolerance to plants [122]. The two most important applications of nanotechnology in nano-fertilizers include the controlled release of fertilizers and the bio-availability of nutrients [123]. Apart from these two roles nanotechnology can also be used for the target modulation and regulation of Plant Growth-Promoting Rhizobacteria (PGPR) at the genetic or proteomic level [124]. In India, nano-fertilizers are used mostly to boost zinc uptake in wheat and other cereals, but there is a gap in research and its application by the farmers. Dr. G. Pandey in his review has shown the application and challenges of nanotechnology for sustainable agriculture in India

[125]. The impact of different nano-fertilizers on yield and growth characteristics of particular crops is illustrated in Table-6 & 7 below.

Table 6. Impact of nano-fertilizers on yields of different crops under varying climatic conditions [126].

Nanofertilizers	Crops	Yield increment (%
Nanofertilizer + urea	Rice	10.2
Nanofertilizer + urea	Rice	8.5
Nanofertilizer + urea	Wheat	6.5
Nanofertilizer + urea	Wheat	7.3
Nano-encapsulated phosphorous	Maize	10.9
Nano-encapsulated phosphorous	Soybean	16.7
Nano-encapsulated phosphorous	Wheat	28.8
Nano-encapsulated phosphorous	Vegetables	12.0–19.7
Nano chitosan-NPK fertilizers	Wheat	14.6
Nano chitosan	Tomato	20.0
Nano chitosan	Cucumber	9.3
Nano chitosan	Capsicum	11.5
Nano chitosan	Beet-root	8.4
Nano chitosan	Pea	20
Nanopowder of cotton seed and ammonium fertilizer	Sweet potato	16
Aqueous solution on nanoiron	Cereals	8–17
Nanoparticles of ZnO	Cucumber	6.3
Nanoparticles of ZnO	Peanut	4.8
Nanoparticles of ZnO	Cabbage	9.1
Nanoparticles of ZnO	Cauliflower	8.3
Nanoparticles of ZnO	Chickpea	14.9
Rare earth oxides nanoparticles	Vegetables	7–45
Nanosilver + allicin	Cereals	4–8.5
Iron oxide nanoparticles + calcium carbonate nanoparticles + peat	Cereals	14.8–23.1
Sulfur nanoparticles + silicon dioxide nanoparticles + synthetic fertilizer	Cereals	3.4–45%

**Table 7.** Impact of different nano-fertilizers on different growth and survival characteristics of particular crops [127].

Nanofertilizers	Crops	Imparted characteristics
Nanoparticles of ZnO	Chickpea	Increased germination, better root development, higher indoleacetic acid synthesis.
Nano silicon dioxide	Maize	Drought resistance, increment in lateral root roots number along with and shoot length.
Nano silicon dioxide	Maize	Increased leaf chlorophyll.
Nano silicon dioxide	Tomato	Taller plants and increased tuber diameter.
Colloidal silica + NPK fertilizers	Tomato	Increased resistance to pathogens.
Nano-TiO <sub>2</sub>	Spinach	Improved vigor indices and 28% increased chlorophyll.
Polyethylene + indium oxide	Vegetables	Increased sunlight absorption
Polypropylene + indium–tin oxide	Vegetables	Increased sunlight utilization
Gold nanoparticles + sulfur	Grapes	Antioxidants and other human health benefits.
Kaolin + SiO <sub>2</sub>	Vegetables	Improved water retention.
Bentonite + N-fixing bacteria inoculation	Legumes	Improved soil fertility and resistance to insect-pest.
Nanocarbon + rare earth metals + N fertilizers	Cereals	Improved nitrogen use efficiency
Stevia extract + nanoparticles of Se + organo-Ca + rare-earth elements + chitosan	Vegetables	Enhanced root networking and root diameter.
Nano-iron slag powder	Maize	Reduced incidence of insect-pest
Nano-iron + organic manures	Cotton	Controlled release of nutrients acts as an effective insecticide and improves soil fertility status.

Nanotechnology has a promising approach in controlling the diseases of agricultural crops. It is beneficial over conventional approaches, such as reduced input requirements, better efficacy and lower ecotoxicity [128]. Nanomaterials may act as antimicrobial agents, biostimulants that induce plant innate immunity, and as carriers for active ingredients such as elicitors, micronutrients, and pesticides in controlling diseases [129]. Thus nanotechnology may provide potential benefits in the field of disease management and agricultural adaptations in the near future.

The use of nanomaterials as agrochemicals either as fertilizers or pesticides is scientifically being explored before it could be used in agriculture for a general farm practice. The properties of nanomaterials such as type of material, shape, size, crystal phase, solubility, exposure and dosage concentrations are considered to be of the potential risk to human health [130]. However, some experts point out that food products containing nanomaterials available in the market are probably safe to eat, then also this area needs to be investigated more actively because detailed studies are required to address the impact and safety concern of nanomaterials within the human body when exposed to nanomaterials via food [131]. Researchers have to develop proper assessment approaches to assess the impact of nanomaterials or nanofertilizers on biotic and abiotic components of the ecosystem [132]. Amongst the different issues concerning the use of nanomaterials in agriculture, the nature of accumulation of nanomaterials in the environment and the focus on the edible part of plants might be the important issues before their use.

Nevertheless, Dr. S.K. Singh in his review has discussed the impending use of nanotechnology in the fields of food and agriculture such as nano-fertilizer, plant nutrition, and plant protection and suggested that nanoparticles may be toxic to the environment and biological system due to their changed physicochemical properties [133]. Thus while considering the benefits of nanomaterials researchers should also focus on the probable risk that may arise due to that nanomaterials.

### v. Future scenarios

Nanotechnology is a capable domain of recent times. Metallic nanoparticles are mainly produced during enzymatic reactions. Mechanistic understandings may provide control over morphology, stability, and rate of synthesis of nanomaterials. Environmental contaminants are remediated by biologically synthesized nanomaterials [134]. Heavy metal pollution has been seen as a significant concern because of its hazardous impacts and non-biodegradable nature [135]. Bioremediation of heavy metal pollution has been preferred over other methods such as chemical precipitation and electrochemical methods [136]. Microbes have been evolved and can offer inventive bio-based clean-up methods [137]. Metal reducing microbes promotes precipitations, minerals dissolutions and transformations such as

geochemical agents [138]. Microbial reduction methods have delivered a new way to explore nanotechnology. Bio-mineralization is among, one of them which controls the environmental impact of heavy metals [139]. The enzyme metal interaction here provides detoxification of heavy metals by a biotransformation process. Microbes which produces nitrate reductase can enhance the efficacy of bioremediation process. Other enzymes along with nitrate reductase have been studied for further possibilities of heavy metal remediation. Microbe metal interaction and the importance of their enzymes still required further investigation [140]. The studies of these microbial interactions at the genetic level may provide new genetic tools for bioremediation. These findings may assist in clarifying new phenomenon such as antibiotic resistance [141]. The upsurge in antibiotic resistance is becoming a serious concern in all over the world for modern medicine. The combination of metallic nanomaterials with antibiotics may overcome these problems such as bacterial resistance [142]. However, the co-occurrence of antibiotic resistance and metallic resistance genes have been reported recently. Thus microbes may play a critical role in controlling heavy metal pollution. The use of microbial nanomaterials for renewable energy needs further investigation.

### 8. Conclusions

Development in engineering and nanoscience technology is proving more helpful for generating new chances to develop more cost-effective and environmentally suitable pollutant treatment technology. The nanoparticle has peculiar properties which make them more suitable for pollution treatment. Current findings have indicated that nanomaterials are beneficial particles for heavy metal removal because of their unique properties. These particles are very efficient in removing heavy metals from the agricultural system. For future development, the scientists should focus on the synthesis of new nanomaterial or compounds which can remove heavy metal ions and organic dyes more efficiently and should have more adsorption capacity and specificity. The microbial threats to agricultural quality and production are also a serious public concern. Scientists should work to develop zinc, chitosan and copper based nanomaterials which possesses antimicrobial and antifungal properties which in turn can solve these microbial problems in

agriculture and may increase agricultural production. Thus further research must be made in this direction to develop nanoparticles with greater stability, specificity and capability for simultaneous removal of multiple contaminants, and microbial pathogens from agriculture. It (nanoparticle) should be economically cost-effective, recyclable adsorbents for their extensive application in agriculture. In addition, a wide range of nanoscience technology should be developed for purification of the environment for the betterment of agricultural production. Studies have demonstrated the activities of nanoparticles in plant growth and development. However, it is not clear that why and how different species of plants shows dissimilar resistance property to nanoparticles. It is recommended to discover the effects of environmental factors on the uptake and accumulation of nanoparticles by different plant species. The route of nanoparticles movement in plant's cells and their effect on genetic system of plants, should be investigated.

### **Conflict of Interests**

We corroborate that this work is original and has not been published elsewhere nor is it currently under consideration for publication elsewhere. We have no conflicts of interest concerning the publication of this paper.

### Acknowledgements

The authors are thankful to the University (CCSHAU) and Prof. K. P. Singh (V. C. Chaudhary Charan Singh Haryana Agricultural University) for providing essential information.

### 9. References

- Wiyaratn, W., Appamana, W., Charojrochkul, S., Kaewkuekool, S. and Assabumrungrat, S., 2012.
   Au/La1- xSrxMnO<sub>3</sub> nanocomposite for chemical-energy cogeneration in solid oxide fuel cell reactor. *Journal of Industrial and Engineering Chemistry*, 18(5), pp.1819-1823.
- 2. Mandal, D., Bolander, M.E., Mukhopadhyay, D., Sarkar, G. and Mukherjee, P., 2006. The use of microorganisms for the formation of metal nanoparticles and their application. *Applied microbiology and biotechnology*, 69(5), pp.485-492.

- 3. Jain, D., Kachhwaha, S., Jain, R., Srivastava, G. and Kothari, S.L., 2010. Novel microbial route to synthesize silver nanoparticles using spore crystal mixture of *Bacillus thuringiensis*.
- Li, L., Tang, Q., Li, H., Yang, X., Hu, W., Song, Y., Shuai, Z., Xu, W., Liu, Y. and Zhu, D.A.O.B.E.N., 2007. An ultra-closely π-stacked organic semiconductor for high performance fieldeffect transistors. *Advanced materials*, 19(18), pp.2613-2617.
- 5. Sathishkumar, M., Sneha, K., Kwak, I.S., Mao, J., Tripathy, S.J. and Yun, Y.S., 2009. Phytocrystallization of palladium through reduction process using *Cinnamom zeylanicum* bark extract. *Journal of Hazardous materials*, 171(1-3), pp.400-404.
- 6. Hsu, Y.S., Lou, J.C., Chou, M.S., Hsu, K.L. and Han, J.Y., 2015. Characterization of single-walled carbon nanotubes and adsorption of perchlorate in water. *Water, Air, & Soil Pollution*, 226(3), p.58.
- 7. Ndlovu, N., Mayaya, T., Muitire, C. and Munyengwa, N., 2020. Nanotechnology Applications in Crop Production and Food Systems. International Journal of Plant Breeding, 7(1), pp.624-634.
- 8. Li, S., Anderson, T.A., Maul, J.D., Shrestha, B., Green, M.J. and Cañas-Carrell, J.E., 2013. Comparative studies of multi-walled carbon nanotubes (MWNTs) and octadecyl (C18) as sorbents in passive sampling devices for biomimetic uptake of polycyclic aromatic hydrocarbons (PAHs) from soils. *Science of the Total Environment*, 461, pp.560-567.
- 9. Hočevar, M., Krašovec, U.O., Bokalič, M., Topič, M., Veurman, W., Brandt, H. and Hinsch, A., 2013. Sol-gel based TiO<sub>2</sub> paste applied in screen-printed dye-sensitized solar cells and modules. *Journal of Industrial and Engineering chemistry*, 19(5), pp.1464-1469.
- 10. Rao, G.P., Lu, C. and Su, F., 2007. Sorption of divalent metal ions from aqueous solution by carbon nanotubes: a review. *Separation and Purification Technology*, *58*(1), pp.224-231.
- 11. Rodríguez-León, E., Iñiguez-Palomares, R., Navarro, R.E., Herrera-Urbina, R., Tánori, J., Iñiguez-Palomares, C. and Maldonado, A., 2013. Synthesis of silver nanoparticles using reducing agents obtained from natural sources (*Rumex hymenosepalus* extracts). *Nanoscale Research Letters*, 8(1), p.318.

- 12. Popescu, M., Velea, A. and Lőrinczi, A., 2010. Biogenic Production of Nanoparticles. *Digest Journal of Nanomaterials & Biostructures (DJNB)*, 5(4).
- 13. Umashankari, J., Inbakandan, D., Ajithkumar, T.T. and Balasubramanian, T., 2012. Mangrove plant, *Rhizophora mucronata* (Lamk, 1804) mediated one pot green synthesis of silver nanoparticles and its antibacterial activity against aquatic pathogens. *Aquatic biosystems*, 8(1), p.11.
- 14. Petla, R.K., Vivekanandhan, S., Misra, M., Mohanty, A.K. and Satyanarayana, N., 2012. Soybean (*Glycine max*) leaf extract based green synthesis of palladium nanoparticles. *J Biomater Nanobiotechnol*, 3(1), pp.14-19.
- 15. Sekhon, B.S., 2010. Food nanotechnology–an overview. *Nanotechnology, science and applications*, 3, p.1.
- 16. Saxena, J., Sharma, M.M., Gupta, S. and Singh, A., 2014. Emerging role of fungi in nanoparticle synthesis and their applications. *World J Pharm Sci*, *3*, pp.1586-1613.
- 17. Shebl, A., Hassan, A.A., Salama, D.M., El-Aziz, A. and Abd Elwahed, M.S., 2019. Green synthesis of nanofertilizers and their application as a foliar for *Cucurbita pepo* L. Journal of Nanomaterials, 2019.
- 18. Shebl, A., Hassan, A.A., Salama, D.M., Abd El-Aziz, M.E. and Abd Elwahed, M.S., 2020. Template-free microwave-assisted hydrothermal synthesis of manganese zinc ferrite as a nanofertilizer for squash plant (*Cucurbita pepo* L). Heliyon, 6(3), p.e03596.
- 19. Xiao, H., Ai, Z. and Zhang, L., 2009. Nonaqueous sol— gel synthesized hierarchical CeO<sub>2</sub> nanocrystal microspheres as novel adsorbents for wastewater treatment. *The Journal of Physical Chemistry C*, 113(38), pp.16625-16630.
- 20. Gudkov, S.V., Shafeev, G.A., Glinushkin, A.P., Shkirin, A.V., Barmina, E.V., Rakov, I.I., Simakin, A.V., Kislov, A.V., Astashev, M.E., Vodeneev, V.A. and Kalinitchenko, V.P., 2020. Production and Use of Selenium Nanoparticles as Fertilizers. ACS omega.

- 21. Chen, X., Zhang, W., Luo, X., Zhao, F., Li, Y., Li, R. and Li, Z., 2017. Efficient removal and environmentally benign detoxification of Cr (VI) in aqueous solutions by Zr (IV) cross-linking chitosan magnetic microspheres. Chemosphere, 185, pp.991-1000.
- 22. Siddiqui, M.H., Al-Whaibi, M.H. and Mohammad, F. eds., 2015. *Nanotechnology and plant sciences: nanoparticles and their impact on plants*. Springer.
- 23. Sathyavathi, R., Krishna, M.B., Rao, S.V., Saritha, R. and Rao, D.N., 2010. Biosynthesis of silver nanoparticles using *Coriandrum sativum* leaf extract and their application in nonlinear optics. *Advanced science letters*, 3(2), pp.138-143.
- Sweeney, R.Y., Mao, C., Gao, X., Burt, J.L., Belcher, A.M., Georgiou, G. and Iverson, B.L., 2004.
   Bacterial biosynthesis of cadmium sulfide nanocrystals. *Chemistry & biology*, 11(11), pp.1553-1559.
- 25. Pope, J.H., Aufderheide, T.P., Ruthazer, R., Woolard, R.H., Feldman, J.A., Beshansky, J.R., Griffith, J.L. and Selker, H.P., 2000. Missed diagnoses of acute cardiac ischemia in the emergency department. *New England Journal of Medicine*, 342(16), pp.1163-1170.
- 26. Parida, U.K., Bindhani, B.K. and Nayak, P., 2011. Green synthesis and characterization of gold nanoparticles using onion (*Allium cepa*) extract. *World Journal of Nano Science and Engineering*, 1(04), p.93.
- 27. Arfin, T. and Yadav, N., 2013. Impedance characteristics and electrical double-layer capacitance of composite polystyrene–cobalt–arsenate membrane. *Journal of Industrial and Engineering Chemistry*, 19(1), pp.256-262.
- 28. Reddy, M.C., Murthy, K.R., Srilakshmi, A., Rao, K.S. and Pullaiah, T., 2015. Phytosynthesis of eco-friendly silver nanoparticles and biological applications—a novel concept in nanobiotechnology. *African Journal of Biotechnology*, 14(3), pp.222-247.
- 29. Ankamwar, B., Chaudhary, M. and Sastry, M., 2005. Gold nanotriangles biologically synthesized using tamarind leaf extract and potential application in vapor sensing. *Synthesis and Reactivity in Inorganic, Metal-Organic and Nano-Metal Chemistry*, 35(1), pp.19-26.

- 30. Kavitha, K.S., Baker, S., Rakshith, D., Kavitha, H.U., Yashwantha Rao, H.C., Harini, B.P. and Satish, S., 2013. Plants as green source towards synthesis of nanoparticles. *Int Res J Biol Sci*, 2(6), pp.66-76.
- 31. Philip, D., 2009. Biosynthesis of Au, Ag and Au–Ag nanoparticles using edible mushroom extract. *Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy*, 73(2), pp.374-381.
- 32. Ankamwar, B., 2010. Biosynthesis of gold nanoparticles (green-gold) using leaf extract of *Terminalia catappa. Journal of Chemistry*, 7(4), pp.1334-1339.
- 33. Bankar, A., Joshi, B., Kumar, A.R. and Zinjarde, S., 2010. Banana peel extract mediated synthesis of gold nanoparticles. *Colloids and Surfaces B: Biointerfaces*, 80(1), pp.45-50.
- 34. Arulkumar, S. and Sabesan, M., 2010. Biosynthesis and characterization of gold nanoparticle using antiparkinsonian drug *Mucuna pruriens* plant extract. *Int Res Pharm Sci*, 1, pp.417-20.
- 35. Jung, J.Y., Lee, S.H., Kim, J.M., Park, M.S., Bae, J.W., Hahn, Y., Madsen, E.L. and Jeon, C.O., 2011. Metagenomic analysis of kimchi, a traditional Korean fermented food. *Appl. Environ. Microbiol.*, 77(7), pp.2264-2274.
- 36. Blaney, L.M., Cinar, S. and SenGupta, A.K., 2007. Hybrid anion exchanger for trace phosphate removal from water and wastewater. *Water research*, *41*(7), pp.1603-1613.
- 37. Kathad, U. and Gajera, H.P., 2014. Synthesis of copper nanoparticles by two different methods and size comparison. *Int J Pharm Bio Sci*, *5*(3), pp.533-540.
- 38. Pavani, K.V., Swati, T., Snehika, V., Sravya, K. and Sirisha, M., 2012. Phytofabrication of lead nanoparticles using Grape skin extract. *International Journal of Engineering Science and Technology*, 4(7).
- 39. Parashar, V., Parashar, R., Sharma, B. and Pandey, A.C., 2009. Parthenium leaf extract mediated synthesis of silver nanoparticles: a novel approach towards weed utilization. *Digest Journal of Nanomaterials & Biostructures (DJNB)*, 4(1).

- 40. Boruah, S.K., Boruah, P.K., Sarma, P., Medhi, C. and Medhi, O.K., 2012. Green synthesis of gold nanoparticles using *Camellia sinensis* and kinetics of the reaction. *Adv Mater Lett*, *3*, pp.481-486.
- 41. Dwivedi, A.D. and Gopal, K., 2011. Plant-mediated biosynthesis of silver and gold nanoparticles. *Journal of biomedical nanotechnology*, 7(1), pp.163-164.
- 42. Fazaludeena, M.F., Manickamb, C., Ashankyty, I.M., Ahmed, M.Q. and Bege, Q.Z., 2017. Synthesis and characterizations of gold nanoparticles by *Justicia gendarussa* Burm F leaf extract. *Journal of Microbiology and Biotechnology Research*, 2(1), pp.23-34.
- 43. Aromal, S.A., Vidhu, V.K. and Philip, D., 2012. Green synthesis of well-dispersed gold nanoparticles using *Macrotyloma uniflorum*. Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy, 85(1), pp.99-104.
- 44. MubarakAli, D., Thajuddin, N., Jeganathan, K. and Gunasekaran, M., 2011. Plant extract mediated synthesis of silver and gold nanoparticles and its antibacterial activity against clinically isolated pathogens. *Colloids and Surfaces B: Biointerfaces*, 85(2), pp.360-365.
- 45. Raghunandan, D., Bedre, M.D., Basavaraja, S., Sawle, B., Manjunath, S.Y. and Venkataraman, A., 2010. Rapid biosynthesis of irregular shaped gold nanoparticles from macerated aqueous extracellular dried clove buds (*Syzygium aromaticum*) solution. *Colloids and Surfaces B: Biointerfaces*, 79(1), pp.235-240.
- 46. Yadav, K.K., Singh, J.K., Gupta, N. and Kumar, V., 2017. A review of nanobioremediation technologies for environmental cleanup: A novel biological approach. *J. Mater. Environ. Sci*, 8, pp.740-757.
- 47. Roychoudhury, A., 2020. Silicon-Nanoparticles in Crop Improvement and Agriculture. International Journal on Recent Advancement in Biotechnology & Nanotechnology [ISSN: 2582-1571 (online)], 3(1).
- 48. Gurunathan, S., Kalishwaralal, K., Vaidyanathan, R., Venkataraman, D., Pandian, S.R.K., Muniyandi, J., Hariharan, N. and Eom, S.H., 2009. Biosynthesis, purification and characterization

- of silver nanoparticles using *Escherichia coli*. *Colloids and Surfaces B: Biointerfaces*, 74(1), pp.328-335.
- 49. Vankar, P.S. and Bajpai, D., 2010. Preparation of gold nanoparticles from *Mirabilis jalapa* flowers.
- 50. Gericke, M. and Pinches, A., 2006. Biological synthesis of metal nanoparticles. *Hydrometallurgy*, 83(1-4), pp.132-140.
- Pugazhenthiran, N., Anandan, S., Kathiravan, G., Prakash, N.K.U., Crawford, S. and Ashokkumar,
   M., 2009. Microbial synthesis of silver nanoparticles by *Bacillus sp. Journal of Nanoparticle Research*, 11(7), p.1811.
- 52. Vahabi, K., Mansoori, G.A. and Karimi, S., 2011. Biosynthesis of silver nanoparticles by fungus *Trichoderma reesei* (a route for large-scale production of AgNPs). *Insciences J.*, *1*(1), pp.65-79.
- 53. Raut Rajesh, W., Lakkakula Jaya, R., Kolekar Niranjan, S., Mendhulkar Vijay, D. and Kashid Sahebrao, B., 2009. Phytosynthesis of silver nanoparticle using *Gliricidia sepium* (Jacq.). *Current Nanoscience*, 5(1), pp.117-122.
- 54. Farooqui, M.A., Chauhan, P.S., Krishnamoorthy, P. and Shaik, J., 2010. Extraction of silver nanoparticles from the leaf extracts of *Clerodendrum inerme*. *Digest Journal of Nanomaterials and Biostructures*, *5*(1), pp.43-49.
- 55. Monica, R.C. and Cremonini, R., 2009. Nanoparticles and higher plants. *Caryologia*, 62(2), pp.161-165.
- 56. Ahmad, N., Sharma, S., Singh, V.N., Shamsi, S.F., Fatma, A. and Mehta, B.R., 2011. Biosynthesis of silver nanoparticles from *Desmodium triflorum*: a novel approach towards weed utilization. *Biotechnology Research International*, 2011.
- 57. Babu, M.G. and Gunasekaran, P., 2009. Production and structural characterization of crystalline silver nanoparticles from *Bacillus cereus* isolate. *Colloids and surfaces B: Biointerfaces*, 74(1), pp.191-195.

- 58. Jain, D., Daima, H.K., Kachhwaha, S. and Kothari, S.L., 2009. Synthesis of plant-mediated silver nanoparticles using papaya fruit extract and evaluation of their anti-microbial activities. *Digest journal of nanomaterials and biostructures*, 4(3), pp.557-563.
- 59. Jain, P.K., Lee, K.S., El-Sayed, I.H. and El-Sayed, M.A., 2006. Calculated absorption and scattering properties of gold nanoparticles of different size, shape, and composition: applications in biological imaging and biomedicine. *The journal of physical chemistry B*, 110(14), pp.7238-7248.
- Chandran, S.P., Chaudhary, M., Pasricha, R., Ahmad, A. and Sastry, M., 2006. Synthesis of gold nanotriangles and silver nanoparticles using Aloevera plant extract. *Biotechnology progress*, 22(2), pp.577-583.
- 61. Lin, D. and Xing, B., 2007. Phytotoxicity of nanoparticles: inhibition of seed germination and root growth. *Environmental Pollution*, *150*(2), pp.243-250.
- 62. Gnanajobitha, G., Annadurai, G. and Kannan, C., 2012. Green synthesis of silver nanoparticle using *Elettaria cardamomom* and assessment of its antimicrobial activity. *Int. J. Pharma Sci. Res.(IJPSR)*, 3, pp.323-330.
- 63. Dhas, S.P., Mukerjhee, A.M.I.T.A.V.A. and Chandrasekaran, N., 2013. Phytosynthesis of silver nanoparticles using *Ceriops tagal* and its antimicrobial potential against human pathogens. *Int J Pharm Pharm Sci*, 5(3), pp.349-352.
- 64. Kasthuri, J., Veerapandian, S. and Rajendiran, N., 2009. Biological synthesis of silver and gold nanoparticles using apiin as reducing agent. *Colloids and Surfaces B: Biointerfaces*, 68(1), pp.55-60.
- 65. Tang, W.W., Zeng, G.M., Gong, J.L., Liu, Y., Wang, X.Y., Liu, Y.Y., Liu, Z.F., Chen, L., Zhang, X.R. and Tu, D.Z., 2012. Simultaneous adsorption of atrazine and Cu (II) from wastewater by magnetic multi-walled carbon nanotube. *Chemical Engineering Journal*, 211, pp.470-478.

- 66. Jiang, S., Lee, J.H., Kim, M.G., Myung, N.V., Fredrickson, J.K., Sadowsky, M.J. and Hur, H.G., 2009. Biogenic formation of As-S nanotubes by diverse Shewanella strains. *Applied and environmental microbiology*, 75(21), pp.6896-6899.
- 67. Philip, D., 2010. Green synthesis of gold and silver nanoparticles using *Hibiscus rosa* sinensis. *Physica E: Low-dimensional Systems and Nanostructures*, 42(5), pp.1417-1424.
- 68. Priya, M.M., Selvi, B.K. and Paul, J.A., 2011. Green synthesis of silver nanoparticles from the leaf extracts of *Euphorbia hirta* and *Nerium indicum*. *Digest Journal of Nanomaterials & Biostructures* (DJNB), 6(2).
- 69. Juibari, M.M., Abbasalizadeh, S., Jouzani, G.S. and Noruzi, M., 2011. Intensified biosynthesis of silver nanoparticles using a native extremophilic *Ureibacillus thermosphaericus* strain. *Materials Letters*, 65(6), pp.1014-1017.
- 70. Bali, R., Razak, N., Lumb, A. and Harris, A.T., 2006, July. The synthesis of metallic nanoparticles inside live plants. In *Nanoscience and Nanotechnology*, 2006. *ICONN'06*. *International Conference on*. IEEE.
- 71. Sintubin, L., De Windt, W., Dick, J., Mast, J., van der Ha, D., Verstraete, W. and Boon, N., 2009. Lactic acid bacteria as reducing and capping agent for the fast and efficient production of silver nanoparticles. *Applied Microbiology and Biotechnology*, 84(4), pp.741-749.
- 72. Smitha, S.L., Philip, D. and Gopchandran, K.G., 2009. Green synthesis of gold nanoparticles using Cinnamomum zeylanicum leaf broth. Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy, 74(3), pp.735-739.
- 73. Lee, J.H., Hwang, E.T., Kim, B.C., Lee, S.M., Sang, B.I., Choi, Y.S., Kim, J. and Gu, M.B., 2007. Stable and continuous long-term enzymatic reaction using an enzyme–nanofiber composite. *Applied microbiology and biotechnology*, 75(6), pp.1301-1307.
- 74. Zhong, L.S., Hu, J.S., Cao, A.M., Liu, Q., Song, W.G. and Wan, L.J., 2007. 3D flowerlike ceria micro/nanocomposite structure and its application for water treatment and CO removal. *Chemistry of Materials*, 19(7), pp.1648-1655.

- 75. Raut, R.W., Kolekar, N.S., Lakkakula, J.R., Mendhulkar, V.D. and Kashid, S.B., 2010. Extracellular synthesis of silver nanoparticles using dried leaves of *Pongamia pinnata* (L) pierre. *Nano-Micro Letters*, 2(2), pp.106-113.
- 76. Srivastava, N.K. and Majumder, C.B., 2008. Novel biofiltration methods for the treatment of heavy metals from industrial wastewater. *Journal of hazardous materials*, 151(1), pp.1-8.
- 77. Olenin, A.Y. and Lisichkin, G.V., 2011. Metal nanoparticles in condensed media: preparation and the bulk and surface structural dynamics. *Russian Chemical Reviews*, 80(7), p.605.
- 78. Wang, P., Lombi, E., Zhao, F.J. and Kopittke, P.M., 2016. Nanotechnology: a new opportunity in plant sciences. *Trends in plant science*, 21(8), pp.699-712.
- 79. Robinson, T., McMullan, G., Marchant, R. and Nigam, P., 2001. Remediation of dyes in textile effluent: a critical review on current treatment technologies with a proposed alternative. *Bioresource technology*, 77(3), pp.247-255.
- 80. Li, S., Shen, Y., Xie, A., Yu, X., Zhang, X., Yang, L. and Li, C., 2007. Rapid, room-temperature synthesis of amorphous selenium/protein composites using *Capsicum annuum* L. extract. *Nanotechnology*, 18(40), p.405101.
- 81. Bhati-Kushwaha, H. and Malik, C.P., 2012. Biomimeting of Silver Nanoparticles from the Dried Extract of *Tridax procumbens* L.(Stem and Leaf). *Journal of Plant Science Research*, 28(2).
- 82. Nanda, A. and Saravanan, M., 2009. Biosynthesis of silver nanoparticles from *Staphylococcus* aureus and its antimicrobial activity against MRSA and MRSE. *Nanomedicine: Nanotechnology, Biology and Medicine*, 5(4), pp.452-456.
- 83. Ingale, A.G. and Chaudhari, A.N., 2013. Biogenic synthesis of nanoparticles and potential applications: an eco-friendly approach. *J Nanomed Nanotechol*, 4(165), pp.1-7.
- 84. Gnanadesigan, M., Anand, M., Ravikumar, S., Maruthupandy, M., Ali, M.S., Vijayakumar, V. and Kumaraguru, A.K., 2012. Antibacterial potential of biosynthesised silver nanoparticles using *Avicennia marina* mangrove plant. *Applied Nanoscience*, 2(2), pp.143-147.

- 85. Husseiny, M.I., El-Aziz, M.A., Badr, Y. and Mahmoud, M.A., 2007. Biosynthesis of gold nanoparticles using *Pseudomonas aeruginosa*. *Spectrochimica Acta Part A: Molecular and Biomolecular Spectroscopy*, 67(3-4), pp.1003-1006.
- 86. Murugadoss, G., Rajamannan, B. and Madhusudhanana, U., 2009. Synthesis and characterization of water-soluble ZnS: Mn<sup>2+</sup> nanocrystals. *Chalcogenide Letters*, 6(5).
- 87. Li, H., Li, W., Zhang, Y., Wang, T., Wang, B., Xu, W., Jiang, L., Song, W., Shu, C. and Wang, C., 2011. Chrysanthemum-like α-FeOOH microspheres produced by a simple green method and their outstanding ability in heavy metal ion removal. *Journal of Materials Chemistry*, 21(22), pp.7878-7881.
- 88. Klaus-Joerger, T., Joerger, R., Olsson, E. and Granqvist, C.G., 2001. Bacteria as workers in the living factory: metal-accumulating bacteria and their potential for materials science. *TRENDS in Biotechnology*, 19(1), pp.15-20.
- 89. Priya, M.M., Selvi, B.K. and Paul, J.A., 2011. Green synthesis of silver nanoparticles from the leaf extracts of *Euphorbia hirta* and *Nerium indicum*. *Digest Journal of Nanomaterials & Biostructures* (*DJNB*), 6(2).
- 90. Joglekar, S., Kodam, K., Dhaygude, M. and Hudlikar, M., 2011. Novel route for rapid biosynthesis of lead nanoparticles using aqueous extract of *Jatropha curcas* L. latex. *Materials Letters*, 65(19-20), pp.3170-3172.
- 91. Jampílek, J. and Kráľová, K., 2015. Application of nanotechnology in agriculture and food industry, its prospects and risks. *Ecological Chemistry and Engineering S*, 22(3), pp.321-361.
- 92. Rathore, K.S., Patidar, D., Janu, Y., Saxena, N.S., Sharma, K. and Sharma, T.P., 2008. Structural and optical characterization of chemically synthesized ZnS nanoparticles. *Chalcogenide Letters*, 5(6), pp.105-110.
- 93. Gardea-Torresdey, J.L., Parsons, J.G., Gomez, E., Peralta-Videa, J., Troiani, H.E., Santiago, P. and Yacaman, M.J., 2002. Formation and growth of Au nanoparticles inside live alfalfa plants. *Nano letters*, 2(4), pp.397-401.

- 94. Narayanan, K.B. and Sakthivel, N., 2008. Coriander leaf mediated biosynthesis of gold nanoparticles. *Materials Letters*, 62(30), pp.4588-4590.
- 95. Velammal, S.P., Devi, T.A. and Amaladhas, T.P., 2016. Antioxidant, antimicrobial and cytotoxic activities of silver and gold nanoparticles synthesized using *Plumbago zeylanica* bark. *Journal of Nanostructure in Chemistry*, 6(3), pp.247-260.
- 96. Shankar, S.S., Ahmad, A., Pasricha, R. and Sastry, M., 2003. Bioreduction of chloroaurate ions by geranium leaves and its endophytic fungus yields gold nanoparticles of different shapes. *Journal of Materials Chemistry*, *13*(7), pp.1822-1826.
- 97. Mueller, N.C., Braun, J., Bruns, J., Černík, M., Rissing, P., Rickerby, D. and Nowack, B., 2012.

  Application of nanoscale zero valent iron (NZVI) for groundwater remediation in Europe. *Environmental Science and Pollution Research*, 19(2), pp.550-558.
- 98. Kim, J.H., Hong, Y.J., Park, B.K. and Kang, Y.C., 2013. Nano-sized LiNi<sup>0</sup>. 5Mn<sub>1</sub>. 5O<sub>4</sub> cathode powders with good electrochemical properties prepared by high temperature flame spray pyrolysis. *Journal of Industrial and Engineering Chemistry*, 19(4), pp.1204-1208.
- 99. Lee, J.S., Shin, D.H., Jun, J., Lee, C. and Jang, J., 2014. Fe<sub>3</sub>O<sub>4</sub>/Carbon Hybrid Nanoparticle Electrodes for High-Capacity Electrochemical Capacitors. *ChemSusChem*, 7(6), pp.1676-1683.
- 100. Turhan, T., Avcıbası, Y.G. and Sahiner, N., 2013. Versatile p (3-sulfopropyl methacrylate) hydrogel reactor for the preparation of Co, Ni nanoparticles and their use in hydrogen production. *Journal of Industrial and Engineering Chemistry*, 19(4), pp.1218-1225.
- 101. Zhong, H.X., Ma, Y.L., Cao, X.F., Chen, X.T. and Xue, Z.L., 2009. Preparation and characterization of flowerlike Y<sub>2</sub> (OH) 5NO<sub>3</sub>· 1.5 H<sub>2</sub>O and Y<sub>2</sub>O<sub>3</sub> and their efficient removal of Cr (VI) from aqueous solution. *The Journal of Physical Chemistry C*, 113(9), pp.3461-3466.
- 102. Nazeruddin, G.M., Prasad, S.R., Shaikh, Y.I., Ansari, J., Sonawane, K.D., Nayak, A.K., Deshmukh, M.B., Patil, P.S., Rathor, B.M. and Prasad, N.R., 2016. In-vitro Bio-fabrication of Multi-applicative Silver Nanoparticles using *Nicotiana tabacum* leaf extract. *Res. J. Life Sci. Bioinformat. Pharmaceut, Chem. Sci*, 2, pp.6-33.

- 103. He, S., Zhang, Y., Guo, Z. and Gu, N., 2008. Biological synthesis of gold nanowires using extract of *Rhodopseudomonas capsulata*. *Biotechnology progress*, 24(2), pp.476-480.
- 104. Ewecharoen, A., Thiravetyan, P., Wendel, E. and Bertagnolli, H., 2009. Nickel adsorption by sodium polyacrylate-grafted activated carbon. *Journal of hazardous materials*, 171(1-3), pp.335-339.
- 105. Mallikarjuna, K., Narasimha, G., Dillip, G.R., Praveen, B., Shreedhar, B., Lakshmi, C.S., Reddy, B.V.S. and Raju, B.D.P., 2011. Green synthesis of silver nanoparticles using Ocimum leaf extract and their characterization. *Digest Journal of Nanomaterials and Biostructures*, 6(1), pp.181-186.
- 106. Nair, B. and Pradeep, T., 2002. Coalescence of nanoclusters and formation of submicron crystallites assisted by Lactobacillus strains. *Crystal growth & design*, 2(4), pp.293-298.
- 107. Lu, C. and Chiu, H., 2006. Adsorption of zinc (II) from water with purified carbon nanotubes. *Chemical Engineering Science*, 61(4), pp.1138-1145.
- 108. Ahmaruzzaman, M., 2011. Industrial wastes as low-cost potential adsorbents for the treatment of wastewater laden with heavy metals. *Advances in colloid and interface science*, 166(1-2), pp.36-59.
- 109. Zhu, B., Wu, S., Xia, X., Lu, X., Zhang, X., Xia, N. and Liu, T., 2016. Effects of carbonaceous materials on microbial bioavailability of 2, 2', 4, 4'-tetrabromodiphenyl ether (BDE-47) in sediments. *Journal of hazardous materials*, 312, pp.216-223.
- 110. Wang, Y.S., Shan, X.Q., Feng, M.H., Chen, G.C., Pei, Z.G., Wen, B., Liu, T., Xie, Y.N. and Owens, G., 2009. Effects of copper, lead, and cadmium on the sorption of 2, 4, 6-trichlorophenol onto and desorption from wheat ash and two commercial humic acids. *Environmental science & technology*, 43(15), pp.5726-5731.
- 111. Yang, X., Li, Q., Wang, H., Huang, J., Lin, L., Wang, W., Sun, D., Su, Y., Opiyo, J.B., Hong, L. and Wang, Y., 2010. Green synthesis of palladium nanoparticles using broth of *Cinnamomum camphora* leaf. *Journal of Nanoparticle Research*, 12(5), pp.1589-1598.

- 112. Hong, S.H., Kwon, S.N., Bae, J.S. and Song, M.Y., 2012. Hydrogen storage characteristics of melt spun Mg–23.5 Ni–5Cu alloys mixed with LaNi5 and/or Nb2O5. *Journal of Industrial and Engineering Chemistry*, 18(1), pp.61-64.
- 113. Li, X.Q. and Zhang, W.X., 2007. Sequestration of metal cations with zerovalent iron nanoparticles a study with high resolution X-ray photoelectron spectroscopy (HR-XPS). *The Journal of Physical Chemistry C*, 111(19), pp.6939-6946.
- 114. Russell, D., 2016. AEMS Compendium. U.S. Patent Application 14/544,443.
- 115. Kalavathy, H., 2010. Studies on heavy metal removal from aqueous solutions and effluent.
- 116. Park, S.J. and Kim, Y.M., 2005. Adsorption behaviors of heavy metal ions onto electrochemically oxidized activated carbon fibers. *Materials Science and Engineering: A*, 391(1-2), pp.121-123.
- 117. Zhang, W.X., 2003. Nanoscale iron particles for environmental remediation: an overview. *Journal of nanoparticle Research*, 5(3-4), pp.323-332.
- 118. Ahmed, E., Kalathil, S., Shi, L., Alharbi, O. and Wang, P., 2018. Synthesis of ultra-small platinum, palladium and gold nanoparticles by Shewanella loihica PV-4 electrochemically active biofilms and their enhanced catalytic activities. *Journal of Saudi Chemical Society*, 22(8), pp.919-929.
- 119. Prabhu, S. and Poulose, E.K., 2012. Silver nanoparticles: mechanism of antimicrobial action, synthesis, medical applications, and toxicity effects. *International nano letters*, 2(1), p.32.
- 120. Pejman, A., Bidhendi, G.N., Ardestani, M., Saeedi, M. and Baghvand, A., 2015. A new index for assessing heavy metals contamination in sediments: a case study. *Ecological indicators*, 58, pp.365-373.
- 121. Mikula, K., Izydorczyk, G., Skrzypczak, D., Mironiuk, M., Moustakas, K., Witek-Krowiak, A. and Chojnacka, K., 2020. Controlled release micronutrient fertilizers for precision agriculture—A review. Science of The Total Environment, 712, p.136365.
- 122. Kalia, A., Sharma, S.P., Kaur, H. and Kaur, H., 2020. Novel nanocomposite-based controlled-release fertilizer and pesticide formulations: prospects and challenges. In Multifunctional Hybrid Nanomaterials for Sustainable Agri-Food and Ecosystems (pp. 99-134). Elsevier.

- 123. Muhammad, Z., Inayat, N. and Majeed, A., 2020. Application of Nanoparticles in Agriculture as Fertilizers and Pesticides: Challenges and Opportunities. In New Frontiers in Stress Management for Durable Agriculture (pp. 281-293). Springer, Singapore.
- 124. Fu, L., Wang, Z., Dhankher, O.P. and Xing, B., 2020. Nanotechnology as a new sustainable approach for controlling crop diseases and increasing agricultural production. Journal of Experimental Botany, 71(2), pp.507-519.
- 125. Pandey, G., 2020. Agri-Nanotechnology for Sustainable Agriculture. In Ecological and Practical Applications for Sustainable Agriculture (pp. 229-249). Springer, Singapore.
- 126. Iqbal, M.A., 2019. Nano-Fertilizers for Sustainable Crop Production under Changing Climate: A Global Perspective. In Sustainable Crop Production. IntechOpen.
- 127. Rathnayaka, R.M.N.N., Iqbal, Y.B. and Rifnas, L.M., 2018. Influence of urea and nano-nitrogen fertilizers on the growth and yield of rice (*Oryza sativa* L.) cultivar 'Bg 250'. Influence of Urea and Nano-Nitrogen Fertilizers on the Growth and Yield of Rice (*Oryza sativa* L.) Cultivar 'Bg 250', 5(2), pp.7-7.
- 128. Usman, M., Farooq, M., Wakeel, A., Nawaz, A., Cheema, S.A., ur Rehman, H., Ashraf, I. and Sanaullah, M., 2020. Nanotechnology in agriculture: Current status, challenges and future opportunities. Science of The Total Environment, p.137778.
- 129. Rehmanullah, Z.M., Inayat, N. and Majeed, A., 2020. Application of Nanoparticles in Agriculture as Fertilizers. New Frontiers in Stress Management for Durable Agriculture, p.281.
- 130. Shukla, Y.M., 2019. Nanofertilizers: A Recent Approach in Crop Production. In Nanotechnology for Agriculture: Crop Production & Protection (pp. 25-58). Springer, Singapore.
- 131. Guha, T., Gopal, G., Kundu, R. and Mukherjee, A., 2020. Nanocomposites for delivering agrochemicals: A comprehensive review. Journal of Agricultural and Food Chemistry, 68(12), pp.3691-3702.
- 132. Heinisch, M., Jácome, J. and Miricescu, D., 2019. Current Experience with Application of Metal-based Nanofertilizers. In MATEC Web of Conferences (Vol. 290, p. 03006). EDP Sciences.

- 133. Singh, S.K., Kasana, R.C., Yadav, R.S. and Pathak, R., 2020. Current Status of Biologically Produced Nanoparticles in Agriculture. In Biogenic Nano-Particles and their Use in Agroecosystems (pp. 393-406). Springer, Singapore.
- 134. Masood, F. and Malik, A., 2013. Current Aspects of Metal Resistant Bacteria in Bioremediation: From Genes to Ecosystem. In *Management of Microbial Resources in the Environment* (pp. 289-311). Springer, Dordrecht.
- 135. Wu, W., Huang, H., Ling, Z., Yu, Z., Jiang, Y., Liu, P. and Li, X., 2016. Genome sequencing reveals mechanisms for heavy metal resistance and polycyclic aromatic hydrocarbon degradation in Delftia lacustris strain LZ-C. *Ecotoxicology*, 25(1), pp.234-247.
- 136. Diaz, M.R., Swart, P.K., Eberli, G.P., Oehlert, A.M., Devlin, Q., Saeid, A. and Altabet, M.A., 2015. Geochemical evidence of microbial activity within ooids. *Sedimentology*, 62(7), pp.2090-2112.
- 137. Engel, J., 2017. Biominerals and Their Function in Different Organisms. In *A Critical Survey of Biomineralization* (pp. 7-11). Springer, Cham.
- 138. Neumann, W., Gulati, A. and Nolan, E.M., 2017. Metal homeostasis in infectious disease: recent advances in bacterial metallophores and the human metal-withholding response. *Current opinion in chemical biology*, 37, pp.10-18.
- 139. Kang, F., Qu, X., Alvarez, P.J. and Zhu, D., 2017. Extracellular saccharide-mediated reduction of Au<sup>3+</sup> to gold nanoparticles: new insights for heavy metals biomineralization on microbial surfaces. *Environmental science & technology*, 51(5), pp.2776-2785.
- 140. Waseem, H., Jameel, S., Ali, J., Saleem Ur Rehman, H., Tauseef, I., Farooq, U., Jamal, A. and Ali, M., 2019. Contributions and challenges of high throughput qPCR for determining antimicrobial resistance in the environment: a critical review. *Molecules*, 24(1), p.163.
- 141. Ghosh, S., Patil, S., Ahire, M., Kitture, R., Kale, S., Pardesi, K., Cameotra, S.S., Bellare, J., Dhavale, D.D., Jabgunde, A. and Chopade, B.A., 2012. Synthesis of silver nanoparticles using Dioscorea bulbifera tuber extract and evaluation of its synergistic potential in combination with antimicrobial agents. *International journal of nanomedicine*, 7, p.483.

142. Jang, H.M., Lee, J., Kim, Y.B., Jeon, J.H., Shin, J., Park, M.R. and Kim, Y.M., 2018. Fate of antibiotic resistance genes and metal resistance genes during thermophilic aerobic digestion of sewage sludge. *Bioresource technology*, 249, pp.635-643.