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Controlling Spontaneous Emission from Perovskite Nanocrystals with Metal-Emitter-Metal Nanostructures

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Abstract: We show the enhancement in the intensity and emission rate of perovskite cesium lead bromide (CsPbBr_3) and formamidinium lead bromide (FAPbBr_3) nanocrystals in the presence of single and double gold layer cavities, which we refer to as Metal-Emitter (ME) and Metal-Emitter-Metal (MEM) nanostructures. Up to 1.9-fold photoluminescence intensity and up to 5.4-fold emission rate enhancements were obtained for FAPbBr_3 nanocrystals confined by double gold layers, which are attributed to plasmonic confinement from the gold layers. The experimentally obtained values are validated by analytical calculations and electromagnetic simulations. Such an effective method of manipulation of the spontaneous emission rate by simple plasmonic nanostructures can be utilized in sensing and detection applications.

Keywords: perovskite; nanocrystal; Purcell effect

1. Introduction

Metal halide perovskites for quantum emitter have gained interest over the years due to their low cost and solution-processable characteristics [1]. Importantly, the defect-tolerant characteristics of halide perovskites are responsible for their high absorption coefficients and high photoluminescence quantum yield, enabling potential large-scale applications in photovoltaic (PV) devices [2], scintillators [3-6], light-emitting diodes (LEDs), lasers, photodetectors, and field-effect transistors (FETs) [7]. Moderate exciton binding energy in the several meV range, large oscillator strength, and tuneable bandgap achieved via tailoring of chemical composition and structural diversity are the other advantages of halide perovskite, notably in nonlinear optics applications [8] and strong light-matter interaction between quantum emitter and optical cavity.

Coupling electromagnetic modes with quantum emitter [9, 10] results in the emission rate modification known as the Purcell effect, which is relevant for realizing single-photon nano-emitters [11]. The light-confining structures can be realized via different platforms, such as photonic crystals [12, 13], disorder [14, 15], and plasmonic nanostructures [16]. In this work, we investigate the spontaneous emission rate modification of halide perovskite nanocrystals (NC) coupled with simple plasmonic cavities. The quantum emitters investigated in this work are cesium lead bromide (CsPbBr_3) and formamidinium lead bromide (FAPbBr_3) NC, while the plasmonic cavity is defined in metal-emitter (ME) and metal-emitter-metal (MEM) structures in which the NC are sandwiched between single and double gold films, respectively. We obtained 1.2-fold and 1.9-fold intensity enhancement for CsPbBr_3 and FAPbBr_3 NC respectively; whereas the emission rate enhancements

are 4.3-fold and 5.4-fold for CsPbBr₃ and FAPbBr₃ NC respectively. Manipulation of the spontaneous emission rate by simple plasmonic nanostructures can be utilized in sensing and detection applications [17, 18]. Furthermore, such simple structures with large emission enhancement factors will improve the brightness for light-emitting devices (LED) applications and improve the signal modulations in telecommunication technologies.

2. Materials and Methods

The schematic of the ME and MEM structures are shown in Figure 1(a). The 100-nm thick bottom gold layer was physically deposited on glass substrate by electron beam evaporation. To avoid non-radiative losses, a thin polystyrene (PS) layer was deposited in between gold and perovskite layer. Prior to polystyrene deposition, the gold surface was treated by oxygen plasma with low power and pressure of 450mBar for 30-50s. PS layer was spin coated from PS in toluene solution (1.5% w/v) at 8000rpm to thickness of 15nm. Cesium lead bromide (CsPbBr₃) NC was synthesized according to a hot-injection protocol described in the literature [5, 19]. In a typical synthesis, 0.7 g lead (II) bromide (PbBr₂), oleic acid (0.5 mL), oleylamine (0.5 mL) and octadecene (5 mL) were loaded into a 50 mL flask and dried under vacuum at 100 °C for 0.5 h. Subsequently, nitrogen (N₂) was introduced and the mixture was heated up to 170 °C until PbBr₂ was completely dissolved. A hot Cs-oleate precursor (0.5 mL) was quickly injected into the flask under vigorous stirring. After 10 s of reaction, the flask was quickly immersed into an ice bath to quench the reaction. The CsPbBr₃ NC precipitate was collected by centrifugation at 8000 rpm for 10 min and subsequently spin coated on the PS surface to thicknesses of 50nm and 100nm. Formamidinium lead bromide (FAPbBr₃) NC was synthesized and spin coated by methods described in Chin et al [20].

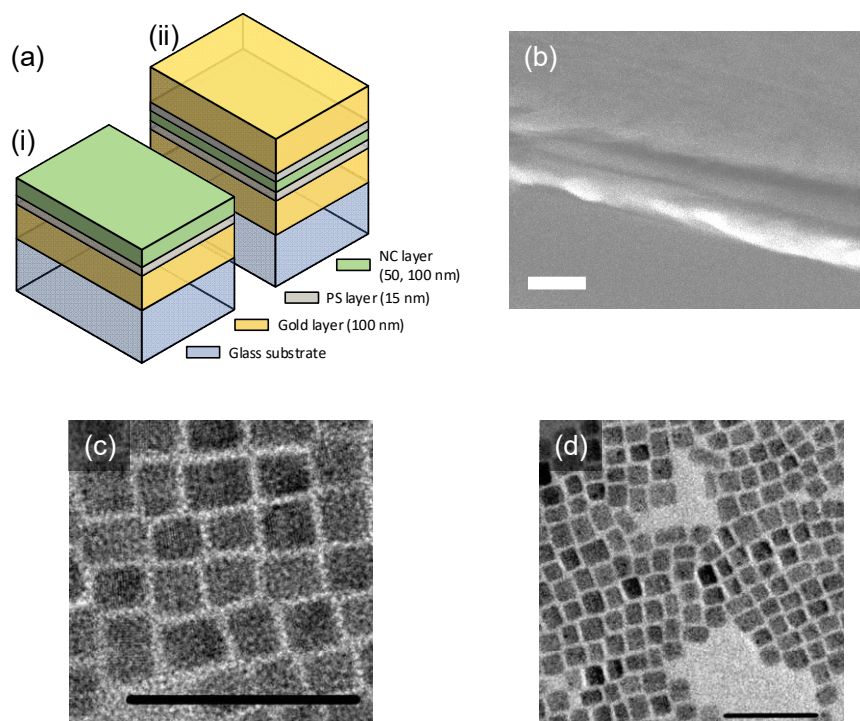


Figure 1.(a) Schematic of i) single- and ii) double-Au-PS layers with CsPbBr₃ or FAPbBr₃ NC ; (b) Scanning electron micrograph of the cross section of 50-nm CsPbBr₃ NC, double 15-nm PS films, and double-Au layers. The white scale bar indicates 100 nm; TEM pictures of (c) CsPbBr₃ and (d) FAPbBr₃ NC, with black scale bar indicating 50nm.

PL measurements were performed at room temperature using free-space excitation and collection through a visible–near-infrared microscope objective (Nikon 20×, NA = 0.40). The sample

was excited via a 10 MHz picosecond pulsed diode laser (Master Oscillator Fibre Amplifier, Picoquant, excitation wavelength at 355 nm, pulse width 50 ps, and power of 40 μ W). PL spectra were detected using AvaSpec-HERO spectrometer. The emission was then filtered by a linear variable filter at 505 and 545 nm and detected by a single-photon avalanche photodiode connected to a time-correlated single-photon counting acquisition module (Edinburgh Instruments, TCC900). The TRPL decay curves were fitted with two exponential functions. The average lifetime was calculated from the decay time components and amplitudes of the fitted TRPL decay curves.

For double gold structure, another gold layer was evaporated on another PS layer on top of the perovskite NC layer to 100nm thickness. The cross-sectional scanning electron micrograph (SEM) of the double gold structure is shown in Figure 1(b). The transmission electron micrograph (TEM) pictures of the CsPbBr₃ and FAPbBr₃ NC are shown in Figures 1(c) and 1(d) respectively.

3. Results and Discussion

Figure 2 presents the photoluminescence (PL) spectra of CsPbBr₃ and FAPbBr₃ NC in different scenarios. We varied the thickness of the NC layer to 50nm and 100 nm for the two NC and normalized the PL spectra to those of the reference NC layer on glass substrate (dashed black curves). The emission wavelengths are found at 505nm (for CsPbBr₃ NC) and 545 nm (for FAPbBr₃ NC). The normalized PL spectra of NC in metal-emitter situation are shown in red curves, showing intensity enhancement of ~ 1.13 and ~ 1.41 times for CsPbBr₃ and FAPbBr₃ NC of 50-nm thickness, respectively. Here, 15-nm thick PS layer is introduced between the gold film and the NC layer to minimize non-radiative loss. Stronger resonance is expected when a second gold layer is introduced. Normalized PL spectra of NC in metal-emitter-metal situation are shown in blue curves, showing PL intensity enhancements of ~ 1.23 (for CsPbBr₃ NC) and ~ 1.91 (for FAPbBr₃ NC) of 50-nm thickness.

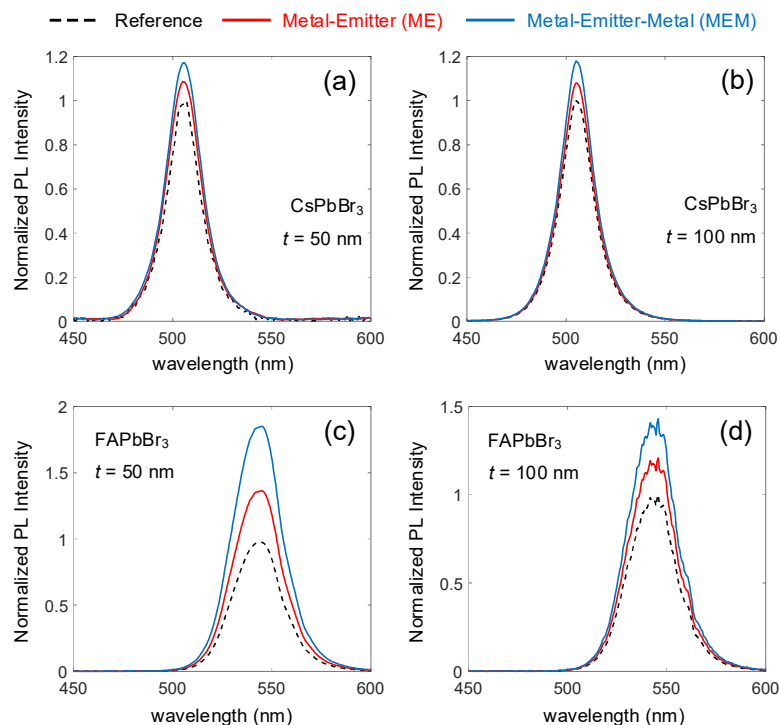


Figure 2. Photoluminescence (PL) spectra of NC-PS-Au layers at room temperature (RT) with CsPbBr₃ NC of (a) 50- and (b) 100-nm thick; and FAPbBr₃ NC of (c) 50- and (d) 100-nm thick. The black dashed, red solid, and blue solid lines correspond to NC reference, single-Au-PS, and double-Au-PS layers, respectively.

As both the gold film(s) and the NC layer form the plasmonic cavity, with light confinement within the NC layer, the role of the NC layer is thus expected on the resonance condition. This is illustrated when the NC layer thickness is increased to 100 nm, presented in Figure 2(b) and Figure 2(d) for CsPbBr₃ and FAPbBr₃ NC, respectively. While the PL intensity enhancement remains

relatively unchanged for CsPbBr₃ NC, it decreases from ~1.41 to ~1.21 (for metal-emitter configuration) and from ~1.91 to ~1.42 (for metal-emitter-metal configuration) for FAPbBr₃ NC. By the same reasoning, the resonance condition is also dependent on the thickness of the PS spacer layer. This means that the PS layer serves not only to minimize non-radiative loss, but also to modify the resonance condition of the plasmonic cavity. The PL characteristics of NC layer without PS spacer layer is shown in Figure S1 (Supplementary Information), indicating strong decrease in emission intensity, which has also been observed before [21]. Time-resolved photoluminescence characteristics for all scenarios are presented in Figure 3, showing spontaneous emission rate modification as the NC are coupled in metal-emitter or metal-emitter-metal configurations. Since both intensity and emission rate are enhanced, as shown in Table 1, this indicates Purcell effect taking place in both types of NC. To verify the role of PS spacer layer in Purcell effect, we present in Figure S2 (Supplementary Information) the time-resolved PL measurements of 50-nm thick CsPbBr₃ NC on gold film without PS spacer layer. We found that the fast decay component in this situation is very close to the instrument response function (IRF) of about 0.3 ns [22]. Coupled with the intensity decrease in Figure S1, it can be seen that the fast decay rate is not attributed to Purcell effect, but to non-radiative loss instead. Therefore, while there is a reduction of non-radiative recombination by PS, the plasmonic/dielectric cavity are still the key driving force which enhances the emission rate.

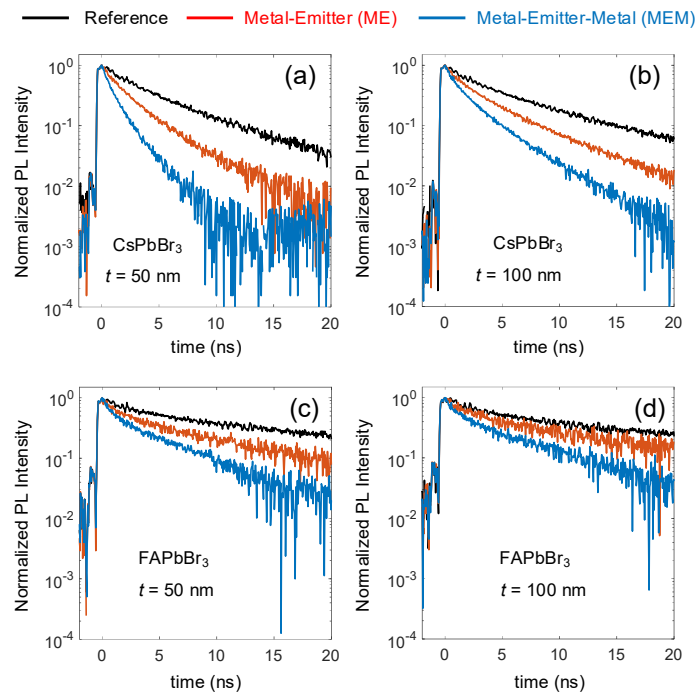


Figure 3. Time-resolved emission measurements. Decay curves of reference (black), single-Au-PS- (red), and double-Au-PS-NC samples (blue) measured for (a) 50- and (b) 100-nm-thick CsPbBr₃ NC and (c) 50- and (d) 100-nm-thick FAPbBr₃ NC. For measurements of CsPbBr₃ and FAPbBr₃ NC, we monitor 505- and 545-nm emission, respectively.

To further understand the experimental results, we turn to analytical calculation and then validate it with finite element simulation (FEM) in COMSOL. The metal-emitter and metal-emitter-metal configurations can be modelled as a set of thin-film stacks with different optical permittivities and the NC as dipoles within the NC layer, as denoted by the red dots, as shown in Figure 4(a). The orientation of the electric dipole is such that it gives emission in the XY plane. Using the Fresnel reflections, the spontaneous emission rate modification of a dipole in metal-emitter configuration can be calculated as [23-26]

$$F_p = \frac{\Gamma_T}{\Gamma_0} = 1 - \frac{3}{2} Q_e \operatorname{Im} \int_0^\infty R_p^{1,2,3}(k_2, t) \exp(-2l_1 d) \frac{u^3}{l_1} du, \quad (1)$$

where F_p is the spontaneous emission rate enhancement factor, Q_e is the quantum efficiency, d is the dipole distance from the interface of metal layer, t is the thickness of the metal layer, k is the propagation constant, $l_m = -i\sqrt{\epsilon_m/\epsilon_1 - u^2}$, $R_p^{1,2,3}$ is the total p -polarized reflection from the thin film stack

$$R_p^{1,2,3}(k_2, t) = \frac{R_p^{1,2} + (R_p^{2,3} \cdot e^{-2l_2 k_2 t})}{1 + R_p^{1,2} \cdot (R_p^{2,3} \cdot e^{-2l_2 k_2 t})}, \quad (2)$$

with $R_p^{m,n} = \frac{\epsilon_m l_n - \epsilon_n l_m}{\epsilon_m l_n + \epsilon_n l_m}$ denoting p -polarized Fresnel reflection between m th and n th layer. The subscripts 1, 2, and 3 refers to perovskite NC layer, gold layer, and glass substrate, respectively. For an electric dipole in metal-emitter-metal configuration, the emission rate modification is

$$F_p = \frac{\Gamma_T}{\Gamma_0} = 1 - \left[Q_e \cdot \left(1 - \frac{3}{2} \operatorname{Im} \int_0^\infty \frac{(1 - R_p^{1,2,3}(k_2, t) e^{-2l_1 k_1 d})(1 - R_p^{1,2,3}(k_2, t) e^{-2l_1 k_1 (L-d)})}{(1 - (R_p^{1,2,3}(k_2, t))^2 e^{-2l_1 k_1 L})} \frac{u^3}{l_1} du \right) \right], \quad (3)$$

where the subscripts 1, 2, and 3 refers to perovskite NC layer, gold layer, and air, respectively.

Table 1. Intensity and emission rate enhancements in metal-emitter and metal-emitter-metal configurations

Perovskite NC	Intensity enhancement		Emission rate enhancement	
	single	double	single	double
CsPbBr ₃ (50nm)	1.13 ± 0.11	1.23 ± 0.12	2.00 ± 0.57	4.28 ± 0.57
CsPbBr ₃ (100nm)	1.08 ± 0.11	1.18 ± 0.11	1.76 ± 0.62	2.92 ± 0.63
FAPbBr ₃ (50nm)	1.41 ± 0.14	1.91 ± 0.19	2.42 ± 0.66	5.38 ± 0.67
FAPbBr ₃ (100nm)	1.21 ± 0.12	1.42 ± 0.14	1.75 ± 0.66	4.18 ± 0.67

The graphs next to the schematics in Figure 4(a) illustrate the change in spontaneous emission rate enhancement over dipole distance from metal surface. At the centre of the perovskite film ($d = 0.5L$), the spontaneous emission rate enhancement factor (F_p) for double gold layer structure is approximately two times of that of the single gold case, in range of $L \lesssim 0.5\lambda$. The quantum efficiency of the perovskite NC is assumed to be $Q_e = 0.4$ [19, 20], while their permittivities were obtained from spectroscopic ellipsometry, as shown in Figure 4(b).

In the FEM [22, 27], the emission rate enhancement is calculated as the dipole power ratio between that within the plasmonic layer structures and that in the perovskite film only. The calculated $|E|$ -field mappings at resonances for ME and MEM structures with CsPbBr₃ NC film are shown in Figure 4(c) and Figure 4(d), respectively. The NC layer is denoted by the green dashed lines, while the gold films are indicated by the yellow dashed lines. The optical modes are confined differently in the two configurations. The E -field for the ME case is localized at the vicinity of the electric dipole, while the E -field for the MEM case is rather extended in the transverse direction, mimicking that of a waveguide. This translates to stronger light-matter interaction in the MEM case, which gives higher emission rate modification.

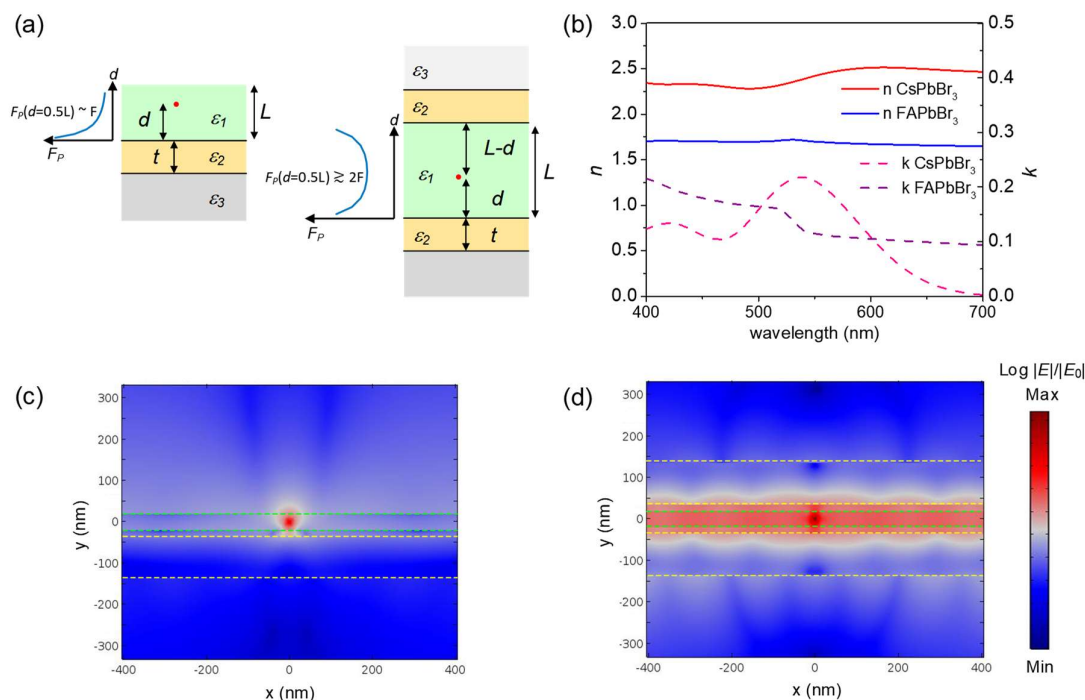


Figure 4. Analytical calculation and electromagnetic simulation approaches to determine the emission rate enhancement: (a) Geometry of single and double finite films for analytical calculation and electromagnetic simulation of fluorescence near an interface. Some variables in the scheme are explained in the text. The graphs illustrate the change in spontaneous emission rate enhancement factor (F_P) over dipole distance from metal surface; (b) Experimental optical constants obtained from variable angle spectroscopic ellipsometry for CsPbBr₃ and FAPbBr₃ NC film; Electric field intensity enhancement mapping for a point dipole in the middle 50-nm-thick CsPbBr₃ film (c) on the top of one Au-PS layer and (d) sandwiched between two Au-PS layers at 540nm and 606 nm resonance wavelength, respectively. The structures are indicated by dashed lines.

The comparison among the calculated, simulated, and measured emission rates for the two NC with 50-nm and 100-nm thicknesses is presented in Figure 5. For the calculation and FEM simulation in this work, the emission rate is averaged over more than 10 equally spaced points over the entire perovskite NC layer thickness. We found that our analytical results and FEM simulations are in good agreement with our experimental results. The difference is due to the boundary problem between PS and NC layers in both simulations. For FAPbBr₃ NC, up to 1.9-fold PL intensity and up to 5.4-fold emission rate enhancements were observed for the case of 50nm thick NC layer in double gold layer structure. We note that the intensity enhancement of FAPbBr₃ NC is higher than that of CsPbBr₃ NC. This can be attributed to the interband transition of gold at 539 nm [28, 29], suggesting that the gold layer behaves more as dielectric at wavelengths shorter than its interband transition. Indeed, this appears to be the case for CsPbBr₃ NC whose emission wavelength is at 505 nm (< 539 nm). However, the light is still confined by virtue of total internal reflection. This is because the real refractive index of gold is 0.887 at 505 nm [28]. Considering the real refractive index of CsPbBr₃ NC is 2.3 at 505nm, the index contrast of ~ 1.4 is large enough for ensuring optical confinement within the NC layer. Taking into account that the emission linewidth of CsPbBr₃ NC of ~ 20 nm, the enhancement is attributed more to dielectric confinement rather than plasmonic confinement. This is in contrast with FAPbBr₃ NC, whose emission wavelength (545 nm) is longer than the interband transition of gold. Considering its emission linewidth of ~ 30 nm, the light confinement is dominated more by plasmonic mechanism rather than by total internal reflection. This also agrees with more pronounced enhancements in FAPbBr₃ NC compared to CsPbBr₃ NC.

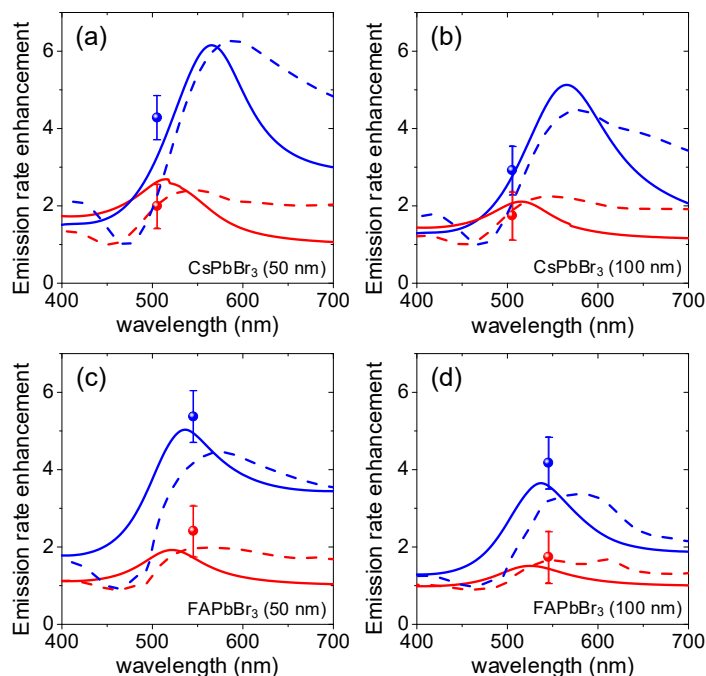


Figure 5. Emission rate enhancement comparison between measurements and calculations. Experimental values (spheres) and calculations of emission rate enhancement from the analytical (solid lines) and the electromagnetic simulations (dashed lines) of fluorescence near the interface from single-Au-PS (red) and double-Au-PS (blue) layers with (a) 50nm-thick CsPbBr₃ NC film, (b) 100-nm thick CsPbBr₃ NC film, (c) 50nm-thick FAPbBr₃ NC film, (d) 100-nm thick FAPbBr₃ NC film.

4. Conclusions

The intensity and emission rate characteristics of CsPbBr₃ and FAPbBr₃ NC have been investigated in two metal-emitter configurations. Owing to their emission wavelengths with respect to the interband transition of gold, the enhancements in the two NC are attributed to different mechanisms. For CsPbBr₃ NC, whose emission wavelength is shorter than gold interband transition, the gold behaves more as dielectric and the light is confined by virtue of strong index contrast. For FAPbBr₃ NC, whose emission wavelength is longer than gold interband transition, the light is confined by plasmonic mechanisms. These confinement mechanisms are the reason for the big differences in the intensity and emission enhancements of the two NC. By treating the metal-emitter and metal-emitter-metal configurations as thin film stacks, we arrived at the analytical formulation to elucidate the emission enhancement, which was also validated by our FEM simulation. We found good agreement between our calculation and our experimental results. In addition, we have shown that a simple plasmonic cavity based on single or double gold films can be used as a platform to study light-matter interaction, particularly in solution-processable perovskite NC, which can be used for sensing application [17, 18].

Supplementary Materials: Figure S1: Photoluminescence (PL) spectra of CsPbBr₃ NC-Au layers (without PS) at room temperature (RT); Figure S2: Time-resolved emission of CsPbBr₃ NC film with Au, Table S1: Key parameters of time resolved PL spectrum.

Author Contributions: Conceptualization, M.D.B.; formal analysis, M.D.B., L.T., L.Y.M.T.; investigation, M.D.B., L.T.; resources, K.L., G.W.P.A. and C.X.Y.; data curation, M.D.B., L.T.; writing—original draft preparation, M.D.B., L.T.; writing—review and editing, L. Y.M. T, G.W.P.A.; visualization, M.D.B., L.T., L.Y.M.T.; supervision M.D.B., H.W. and Z.D.H. All authors have read and agreed to the published version of the manuscript.

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Conflicts of Interest: The authors declare no conflict of interest.

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