

Article

Unveiling the differential anti-oxidant activity of maslinic acid in murine melanoma cells and in rat embryonic healthy cells following treatment with hydrogen peroxide

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Abstract: Maslinic acid (MA) is a natural triterpene from *Olea europaea* whose pharmacological functions have been showed. The objective of this study was to examine MA effect on cell viability (by MTT assay), reactive oxygen species (ROS levels, by flow cytometry) and key anti-oxidant enzyme activities (by spectrophotometry) in murine skin melanoma (B16F10) cells compared to healthy cells (A10). MA induced cytotoxic effects in cancer cells (IC₅₀ 42 µM) whereas no effect was found in A10 cells treated with MA (up to 210 µM). In order to produce a stress situation in cells, 0.15 mM of H₂O₂ were added. Under stressful conditions, MA protected both cell lines against oxidative damage, decreasing intracellular ROS, being higher in B16F10 than in A10 cells. The treatment with H₂O₂ and without MA produced different responses in anti-oxidant enzymes activities depending on cell line. In A10 cells, all enzymes were up-regulated, but in B16F10 cells only superoxide dismutase, glutathione S-transferase and glutathione peroxidase increased their activities. MA restored the enzyme activities to similar levels than control group in both cell lines, highlighting that in A10 cells the highest MA doses induced values lower than control. Overall, these findings demonstrate the great anti-oxidant capacity of MA.

Keywords: anti-oxidant activity; anti-oxidant enzymes, anti-proliferative activity; maslinic acid; melanoma; *Olea europaea*; ROS levels.

1. Introduction

Oxidative stress is defined as an imbalance between pro-oxidant and anti-oxidant molecules that modify the normal cell physiology due to proteins, lipids, carbohydrates and nucleic acids damage [1]. Many cellular processes depend on the variations in the levels of ROS and NADPH that take place during their development and that, fundamentally, are determined by the activity of the different production systems of this reduced coenzyme, especially those belonging to the pentose phosphate pathway (PPP), glucose-6-phosphate dehydrogenase (G6PDH) and 6-phosphogluconate dehydrogenase (6PGDH) and, also, to NADP-dependent isocitrate dehydrogenase (ICDH-NADP). The cellular redox state will be key to interpreting the behaviour of most of these key cellular

processes for the vital development of organisms, such as cell differentiation [2-4], cellular growth [5-8], nutrition cell [6, 9-11] and cell aging [12].

Normal cellular metabolism is well established as the source of endogenous pro-oxidants and the most important are reactive oxygen species (ROS) and reactive nitrogen species (RNS) [13]. Under these normal circumstances, ROS are neutralized by an anti-oxidant defense system consisting of enzymes such as, catalase (CAT), which reduces hydrogen peroxide; superoxide dismutase (SOD), which detoxifies superoxide radical; glutathione peroxidase (GPX), which reduces hydrogen peroxide and other organic peroxides; S-transferase glutathione (GST), which detoxifies harmful molecules; glutathione reductase (GR), which regenerate glutathione (GSH) from its oxidized form (GSSG) by NADPH-dependent pathway; G6PDH which produces NADPH for GR mechanism and also is a direct scavenger of free radicals [14, 15] and numerous non enzymatic anti-oxidant molecules [16]. However, although a certain level of DNA damage cannot be avoided, oxidative stress can occur when the balance is upset, either by excessive production of ROS, by deficient anti-oxidant defences, or by a combination of both [17]. In cancer cells, oxidative stress has been linked to the regulation of numerous cellular processes, including DNA damage, proliferation, cell adhesion, migration, and regulation of cell survival or cell death [15].

Hydrogen peroxide (H_2O_2) is originated from $O_2^{\bullet-}$ by SOD. It is not a free radical as such, but it is a reactive form, since it has the capacity to generate the hydroxyl radical in the presence of metals such as iron [1]. H_2O_2 is a key metabolite in oxidative stress being an important representative ROS that arises during the aerobic respiration process and from other cellular sources [18]. As a messenger molecule, H_2O_2 diffuses through cells and tissues to initiate immediate cellular effects, such as cell shape changes, initiation of proliferation and recruitment of immune cells. If not controlled, an excess of H_2O_2 can provoke a condition of decontrolled oxidative stress in the cells producing irreparable damage [19]. Therefore, cell death and survival processes can be modulated controlling intracellular H_2O_2 levels by using anti-oxidants. The appeal of natural products has stemmed from the cost-effectiveness, fewer side effects of natural products and for being an inexhaustible source. As such, products like maslinic acid (MA) have received heightened interest.

MA, $C_{30}H_{48}O_4$ (2α , 3β -dihydroxyolean-12-en-28-oic acid) is natural pentacyclic triterpene, also known as crataegolic acid and derived from its structural analogue, oleanolic acid (Figure 1). It is a compound formed by 30 carbon atoms grouped into five cycles that, as substituents, have two methyl groups on the C-4 and C-21 carbons; a methyl groups at the C-8, C-10 and C-15 carbons; two hydroxyl groups on the C-2 and C-3 carbons; a carboxyl group at the C-28 carbon and a double bond on the C-12 and C-13 [20-23].

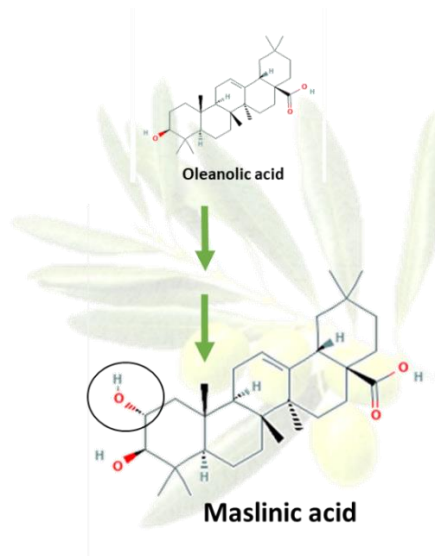


Figure 1. Schedule chemical structures of oleanolic acid and maslinic acids from pomace olive (*Olea europaea* L.) (Pubchem).

Maslinic acid has been isolated from various plants, being especially abundant in *Crataegus oxyacantha* and in the surface wax of fruits and leaves of *Olea europaea* and the solid waste from olive oil extraction [24]. Furthermore, MA is among the main triterpenes present in olives and olive oil. Its concentration in the oil depends on the type of olive oil and the variety of olive tree [23].

MA has a high pharmacological interest, highlighting its anti-tumour effect in certain types of cancer besides its anti-inflammatory, anti-oxidant and anti-diabetogenic, anti-hypertensive, anti-viral and cardioprotective effects, among others [1, 23]. Among the bioactivities attributed to MA, its anti-oxidant effect is the most contradictory. Montilla et al. [25] observed that MA reduced the plasma levels of endogenous lipid peroxides and the susceptibility of plasma to lipid peroxidation in a model of oxidative status induced by CCl_4 . Similarly, MA isolated from the flowers of *Punica granatum* prevented the CuSO_4 induced oxidation of rabbit plasma LDL [26]. Allouche et al. [27] performed a study about the anti-oxidant properties of several pentacyclic triterpenic diols and acids on LDL particles isolated from human plasma. Particularly, MA not only retarded the initiation and decreased the rate of CuSO_4 induced LDL oxidation, but also showed peroxyl radical scavenging activity and a slight metal (copper) chelating effect [27].

Further research has been done in macrophages, which play a role in the defensive system of the organism in response to activation by a pathogen [28]. Results of this study indicated that MA did not exert any direct inhibitory effects on the formation of NO and superoxide, but did reduce the generation of H_2O_2 (IC_{50} of $46.3 \mu\text{M}$), in a way that was similar to that of catalase [29]. In a previous study, we concluded that MA produced an increase in the ROS level under stress conditions caused by absence of FBS in melanoma cell [15]. Furthermore, in this same study, MA had an anti-oxidant effect at lower assayed levels; however, at higher dosages MA induced cellular damage by apoptosis.

The objective of this study was to examine the anti-oxidant effect of MA in murine skin melanoma (B16F10) cells and in a healthy cell line derived from the thoracic aorta of embryonic rat (A10), by analyzing the proliferation, ROS production and the activity of the main anti-oxidants enzymes under cellular stress conditions induced by H_2O_2 .

2. Results

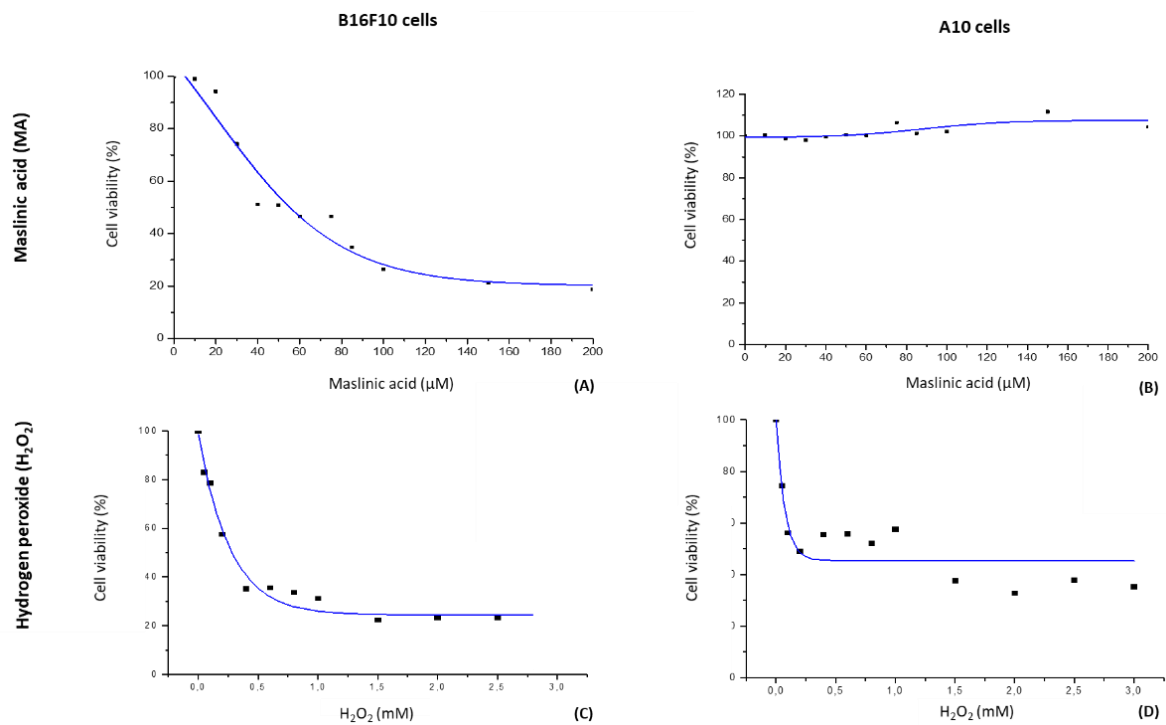
2.1. MA inhibits proliferation of B16F10 cells by a dose-dependent mechanism

We examined the effect of MA on the proliferation of B16F10 melanoma and A10 cell lines using MTT assay (Figure 2 A and B). B16F10 and A10 cells were exposed to 0 to $212 \mu\text{M}$ of MA for 24 h and cell survival was then compared with untreated controls, as determined by formazan dye uptake. The percentage of living cells (viable formazan-accumulating cells) decreased concomitantly with dose in cancer cells (B16F10) while in healthy cells (A10), MA did not produce any cytotoxic effect, even at the highest doses used ($210 \mu\text{M}$). In B16F10 cells, we noticed that 50% growth inhibition values (IC_{50}) in response to MA were of $42 \mu\text{M}$ after 24 h of addition of this compound. Based on this value, the rest of studies were performed with an exposition of cells to $10.6 \mu\text{M}$ ($\text{IC}_{50/4}$), $21.2 \mu\text{M}$ ($\text{IC}_{50/2}$), $42.3 \mu\text{M}$ (IC_{50}) and $84.6 \mu\text{M}$ ($2 \cdot \text{IC}_{50}$) of MA for 24 h.

2.2. H_2O_2 modifies cell viability

We examined the effect of H_2O_2 on cell viability of B16F10 and A10 cells in order to know the optimal dose capable of producing a stress without reaching the cell death (Figure 2 C and D). Cells were exposed to 0 to 3 mM of H_2O_2 for 24 h and cell survival was then compared with untreated controls, as determined by formazan dye uptake. The percentage of living cells decreased concomitantly with dose in both cell lines until the concentration of 1.5 mM from which H_2O_2 had no cell viability effect. In both cell lines, we noticed an IC_{50} of 0.2 mM. In no case, with the doses used and the incubation time employed, any mortality higher than 80% was reached. Based on these results, the concentration of 0.15 mM of H_2O_2 was used to perform the rest of studies.

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Figure 2. Cytotoxicity curves of maslinic acid (MA) on B16F10 murine melanoma cells (A) and A10 rat embryonic healthy cells (B) and hydrogen peroxide (H₂O₂) on B16F10 murine melanoma cells (C) and A10 rat embryonic healthy cells (D). Cell proliferation was determined by MTT assay. Values are expressed as means ± SEM (n = 9).

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2.3. Maslinic acid influences on mitochondrial-membrane potential

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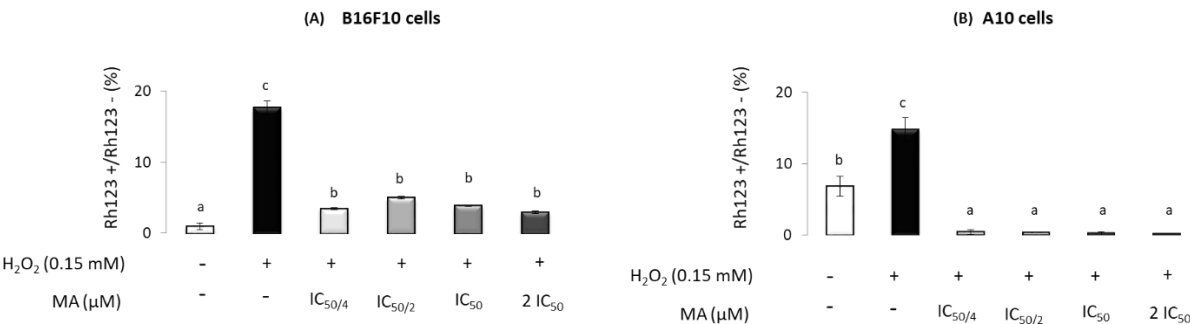
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We performed an assay of ROS production by DHR. Firstly, we tested ROS levels that occur over time in the presence of 0.150 mM of H₂O₂ and using different MA levels (IC_{50/4}, IC_{50/2}, IC₅₀ and 2·IC₅₀) based on our cytotoxicity results. The results are showed in Figure 3. After 24 h of H₂O₂ treatment, ROS levels increased significantly in both cell lines, compared to the controls. The increment observed by H₂O₂ addition was offset by MA supplement, which produced a decreased in the intracellular ROS levels at any MA levels used (IC_{50/4}, IC_{50/2}, IC₅₀ and 2·IC₅₀) in B16F10 and A10 cells. Moreover, in A10, ROS levels decreased below the value of control cells for all concentrations of MA tested.

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Figure 3. Positive fluorescent Rh123 on B16F10 cells (A) and A10 cells (B) with (+) or without (-) H₂O₂ and MA treatment at different dosages, IC_{50/4}, IC_{50/2}, IC₅₀, and 2·IC₅₀ (10.6, 21.6; 42.3 and 84.6 μM, respectively). Values are expressed as mean ± SEM (n = 9). Different letters indicate significant differences (p < 0.05).

2.4. MA exerts as anti-oxidant activity modulating enzymatic defense system

An evaluation of the activities of the anti-oxidant enzymes superoxide dismutase (SOD), glutathione-S-transferase (GST), catalase (CAT), glucose 6-phosphate dehydrogenase (G6PDH), glutathione peroxidase (GPX) and glutathione reductase (GR) was performed under our experimental conditions.

Results obtained for SOD in B16F10 cells showed that H₂O₂ increased its activity and MA (at any level) decreased, but without significant differences induced by different MA concentrations. In A10, H₂O₂ increased the SOD activity, increment that was mitigated with MA addition. Moreover, 42.3 and 84.6 μM of MA (IC₅₀ and 2·IC₅₀, respectively) resulted on SOD activity levels lower than those found in absence of H₂O₂ and MA (Figure 4 A and B).

In B16F10 cells, GST activity increased in presence of H₂O₂, whereas the incubation with IC_{50/4} and IC_{50/2} of MA decreased the levels making them equal to the control without H₂O₂ and MA. The use of IC₅₀ and 2·IC₅₀ MA concentrations induced GST activities lower than control. Similar results were found in A10 for GST activity, with the exception of MA at IC_{50/2} that also induced lower levels than control (Figure 4 c and d).

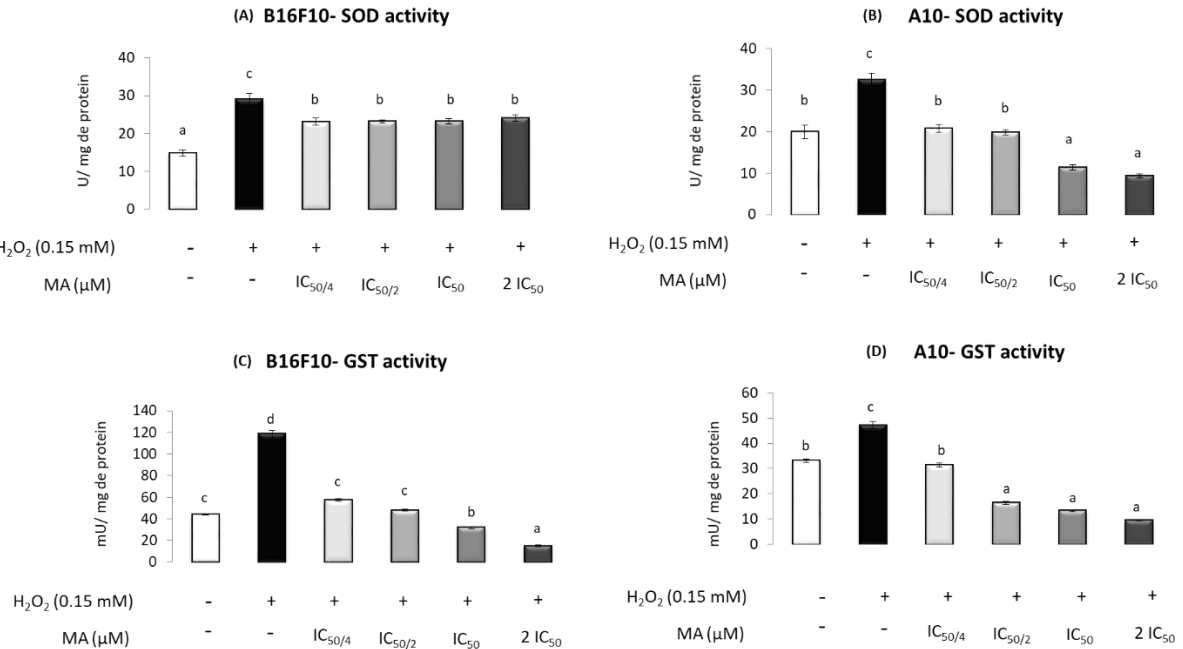


Figure 4. Effect of maslinic acid on the specific activity of Superoxide dismutase (SOD) in cancer cells, B16F10 5(A), and normal cells, A10 (B), and glutathione S-transferase (GST) in cancer cells, B16F10 (C), and normal cells, A10 (D), subjected to the presence of hydrogen peroxide. Enzyme activities (U or mU × mg protein⁻¹) are expressed as mean ± SEM (n = 9). Different letters indicate significant differences (P < 0.05).

In B16F10 cells, CAT activity decreased in presence of H₂O₂ compared to the control, whereas MA increased the activity reaching values higher or similar than control when IC_{50/4} or IC_{50/2} were used. In A10 cells, H₂O₂ produced an increased in the CAT activity and its activity was not recovered until the highest doses of MA were used (IC₅₀ and 2·IC₅₀) (Figure 5 A and B).

Regarding G6PDH, the results indicated its activity decreased in presence of H₂O₂ in B16F10 and MA in all concentrations tested had similar values to the control without H₂O₂ and MA. In A10 cells, no changes were induced by H₂O₂ in G6PDH, whereas MA decreased its activity at any tested level (Figure 5 C and D).

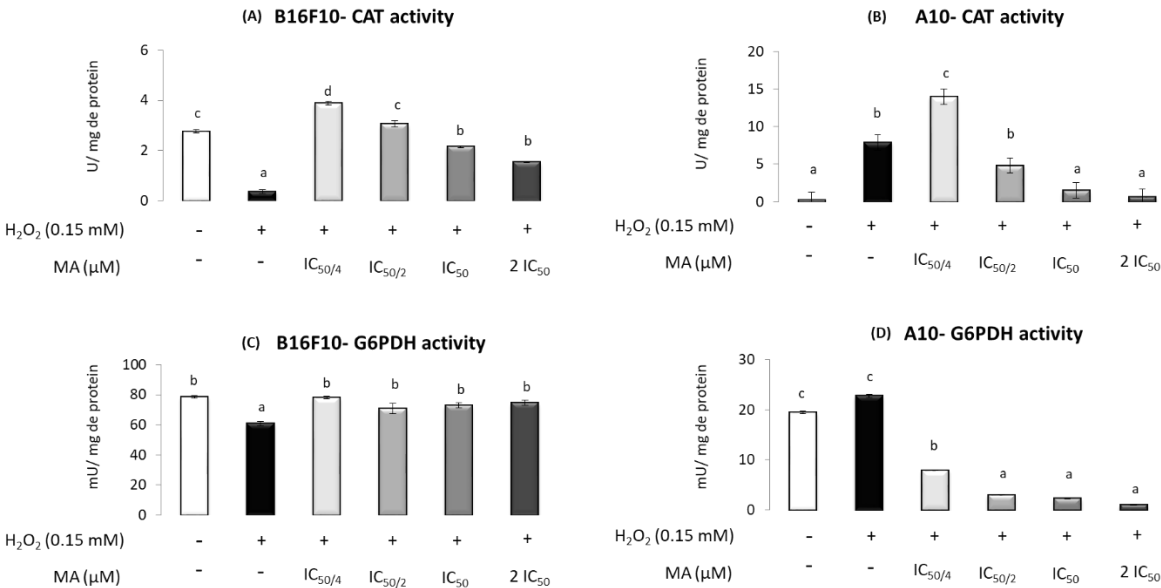


Figure 5. Effect of maslinic acid on the specific activity of catalase (CAT) in cancer cells, B16F10 (A), and normal cells, A10 (B), and glucose 6-phosphate dehydrogenase (G6PDH) in cancer cells, B16F10 (C), and normal cells, A10 (D), subjected to the presence of hydrogen peroxide. Enzyme activities (U or mU × mg protein⁻¹) are expressed as mean ± SEM (n = 9). Different letters indicate significant differences (P < 0.05).

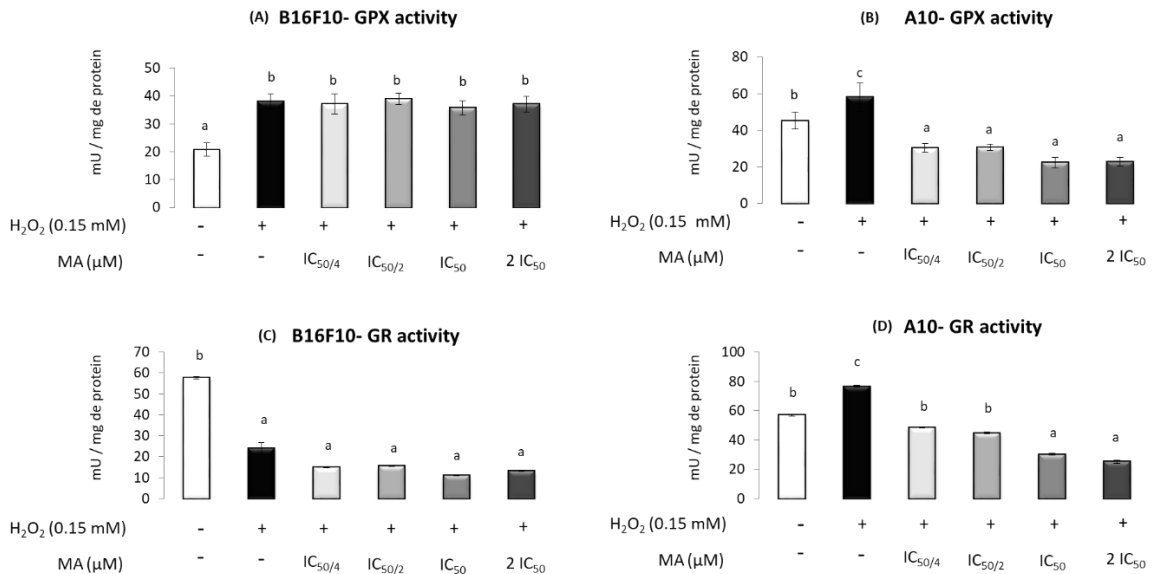


Figure 6. Effect of maslinic acid on the specific activity of glutathione peroxidase (GPX) in cancer cells, B16F10 (A), and normal cells, A10 (B), and glutathione reductase (GR) in cancer cells, B16F10 (C), and normal cells, A10 (D), subjected to the presence of hydrogen peroxide. Enzyme activity (mU × mg protein⁻¹) are expressed as mean ± SEM (n = 9). Different letters indicate significant differences (P < 0.05).

GPX increased in presence of H₂O₂ but any change was observed when MA was administrated in B16F10. In A10 cells, all concentrations of MA decreased GPX activity increased by the H₂O₂ even below the control without H₂O₂ and MA (Figure 6 A and B).

In B16F10, GR activity was decreased by H₂O₂ and no changes were observed at any MA level, maintaining lower levels of activity compared to the control. In A10, H₂O₂ increased GR activity, increment that was mitigated with MA addition. Moreover, 42.3 and 84.6 μ M of MA induced values of GR activity lower than those found in absence of H₂O₂ and MA (Figure 6 C and D).

3. Discussion

Olive oil from *Olea europaea* is a common component in the diet of the Mediterranean countries. It has been showed that this natural source of nutrients presents very important beneficial properties for health due to their chemical composition. The pentacyclic triterpenes are especially abundant in the surface wax of fruits and leaves of tree. In this sense, maslinic acid (MA) in the main triterpene included in this family of molecules with demonstrated properties that include anti-cancer, anti-inflammatory, anti-viral, anti-allergic, anti-diabetogenic effects, among others [23, 30, 31]. Due to the large amount of bioactivities, these molecules are appreciated as chemopreventive agents in pathophysiological diseases such as cancer [23, 27], cardiovascular and neurodegenerative diseases, such as atherosclerosis, also as well in biological process like aging [25].

The effect cytotoxic of MA has been studied in different cell lines, including both cancer and healthy cells. In this sense, it has been showed that this compound has a variable effect depending on the type of cells and experimental conditions. In the present study, MA showed cytotoxic effects only in B16F10 cells and this effect was important, as the IC₅₀ of MA was 42 μ M. In another study performed in melanoma cells by Kim et al. [32], the authors observed a lower MA cytotoxicity effect. So, they concluded that IC₅₀ for MA in SK-MEL-3 was also 42 μ M, but in their study the incubation period was 48h *versus* the 24h used in the present study. Other results obtained in our research group showed that in colon cancer cells, the IC₅₀ value for MA was 30 μ M in HT29 cells after 72h of incubation [31, 33, 34], showing a higher cytotoxic effect in Caco-2 cells in which the IC₅₀ of MA was 10.82 μ M, also after 72 h [35]. Other authors observed IC₅₀ values for MA ranging between 32 μ M and 64 μ M in different cancer cells such as lung (A549), ovary (SK-OV-3), colon (HCT15) and glioma (XF498) after 48h of incubation [32]. In several bladder cancer cell lines incubated with MA during 48h, other authors found IC₅₀ values between 20 and 300 μ M [36]. Notwithstanding, compared to melanoma cells, non-cytotoxic effect was found in A10 healthy cells in the present study. These results are relevant since demonstrate the selective cytotoxic effect of this compound. Studies focused on the selective effect of MA are scarce. Reyes et al. [22] observed that epithelial intestinal cells incubated with 30 μ M of MA during 72h (IC₅₀ value for HT29) exhibited a survival rate of 78% in IEC-6 cells and 68% in IEC-18.

An intracellular redox balance is crucial to ensure the viability, growth and diversity of cell functions [17]. An excess of ROS is related to pathological processes producing oxidative damage when the anti-oxidant defense system is not able to counteract it. Hence, there is a growing interest of the scientific community and industry to obtain substances of natural origin aimed at preventing these pathologies and alterations caused by ROS. In this context, it is important to characterize the protective biochemical functions of natural anti-oxidants and to study their intracellular pathway signalling. A large number of plant constituents, such as MA, have anti-oxidant properties [37].

The present study examined the anti-oxidant effects of MA in skin melanoma cancer cells (B16F10) and thoracic aorta of embryonic rat cells (A10). The major findings were that MA improves oxidative stress caused by a H₂O₂ excess decreasing the intracellular ROS levels in both cell lines. H₂O₂ was used in this study because it is known that when is presented in excess, is one of the major compounds that can damage the cells [38]. Other authors have showed clearly the influence of H₂O₂ on the viability and ROS production of different cell types that were subsequently treated with other anti-oxidant natural compounds that reversed the oxidative damage caused by H₂O₂ [39, 40]. Furthermore, similar studies than us with others ROS producing molecules (i.e. CCl₄) observed that MA counteracted the lipid peroxidation generated in the nucleus [25]. Similarly, MA prevented the CuSO₄ induced oxidation in rabbit plasma LDL [41]. Moreover, Yang et al. [42] evaluated the anti-

oxidant effects of MA derivatives that showed radical-scavenging activity and inhibition in NO production in RAW 264.7 cells.

Regarding MA effect on anti-oxidant enzymes in B16F10 cell line, as expected, SOD, GST and GPX activities were increased in response to the H₂O₂ addition. However, H₂O₂ decreased CAT, G6PDH and GR activities. H₂O₂ excess is responsible of the cellular damages that include imbalance in membranes and biomolecule alterations. This fact supposes an energetic extra cost for the maintenance of cellular homeostasis in order to repair the damaged produced. Both, cellular damages and energy requirements could affect to G6PDH activity, either for the impaired glucose transporters or enhanced glycolysis pathway, decreasing the available glucose for the pentose phosphate pathway. The regulation of NADPH levels is essential to understand the behaviour of numerous physiological processes being especially important for growth and cell differentiation [2], but also for anti-oxidant processes, among others. NADPH is mainly generated by G6PDH enzyme and is required for the GPX and CAT activity involved in H₂O₂ removal pathway. So, CAT is protected of its inactivation by NADPH. Moreover, this reduction equivalent is used by GR to regenerate the oxidized glutathione to its reduced form required for GPX activity [43, 44]. The lower G6PDH activity observed when H₂O₂ was added could result in a decrease in available NADPH levels which might justified the reduction of CAT and GR activities. Similar results have been observed in previous studies in B16F10 cells submitted to stress conditions induced by FBS absence [15]. Mokthari et al. [15] reported low activity levels in CAT that were produced by the low NADPH levels due to the imbalance in the cellular homeostasis.

It has been established that MA is a compound with anti-oxidant capacity [15, 25, 27]. When MA was added to the B16F10 cells, the changes induced by H₂O₂ excess in SOD and GST activities were counteracted. In this sense, scavenger MA effect reduced ROS levels and subsequently the need for acting these anti-oxidant defences. Moreover, the observed decreased ROS levels due to MA would involve a lower cellular damage, which would reverse the possible mechanisms responsible for the decrease G6PDH when H₂O₂ is added, raising its activity until the control values. This recovered G6PDH activity would produce the required NADPH level to prevent and reverse the down-regulation of CAT [43]. Finally, a dose-response effect was observed both in GPX and CAT, highlighting the anti-oxidant behaviour of MA in these cells.

Results found in this work, in A10 healthy cells, revealed that the all enzyme activities increased in stressful conditions originated by the H₂O₂ addition, except in G6PDH whose activity was slightly higher. MA in these cells restored the anti-oxidant enzyme activities reaching values similar to the control group in cells treated with the lowest MA levels (IC_{50/4}). Furthermore, when cells were treated with the two highest doses of MA (IC₅₀ and 2·IC₅₀), the activity of all anti-oxidant enzymes, with the exception of CAT, decreased below control levels. This fact, confirms the relevant anti-oxidant effect of MA [15, 27, 43].

A summary of the biological properties of the MA in relation to its cytotoxic and proliferative effect as well as its antioxidant activity in cancer and healthy cells can be seen in Figure 7.

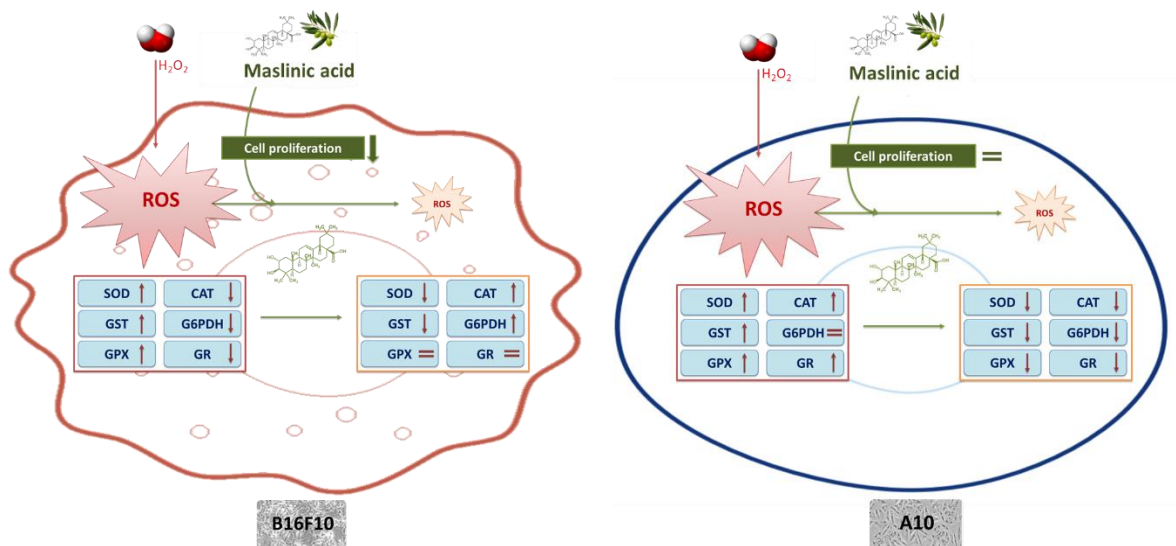


Figure 7. Scheme of the metabolic changes, in the proliferation processes and in the activity of the antioxidant system enzymes, that occur in cancer (B16F10) and healthy (A10) cells in response to treatment with H₂O₂ and MA.

4. Materials and Methods

4.1. Compounds

Maslinic acid was obtained from olive pomace and kindly donated by Biomaslinic S.A. Granada, Spain. The extract is a chemically pure white powder composed of 98% maslinic acid and stable when stored at 4 °C. It was dissolved before use at 10 mg/mL in 50% dimethyl sulfoxide (DMSO) and 50% phosphate buffer solution (PBS). This solution was diluted in cell culture medium for assay purposes.

4.2. Cell lines and cultures

Mouse melanoma cells, B16F10 is a variant of the murine melanoma cell B16. These cells show a higher metastatic potential than B16 cells [45]. A10 is a cell line derived from the thoracic aorta of embryonic mouse and it is used as a study model of smooth muscle cells (SMC). This cell type shows a great proliferative capacity and may be subcultured several times, allowing rapid attainment of cell mass) [46]. Both cell lines used were provided by the cell bank of the University of Granada (Spain). Cell lines were cultured in Dulbecco's modified Eagle's medium (DMEM) containing glucose (4.5 g/L) and L-glutamine (2 mM) from the commercial PAA home, 10% heat-inactivated foetal bovine serum (FBS) and 0.5% gentamicin to B16F10 cells and 10,000 units/mL penicillin, and 10 mg/mL streptomycin to A10 cells. The cell lines were maintained in a humidified atmosphere with 5% CO₂ at 37°C. Cells were passaged at preconfluent densities by the use of a solution containing 0.05% trypsin and 0.5 mM EDTA. Cells were seeded in the culture dishes at the desired density. After 24h hours, when the cells are attached to the dish, the cells were incubated with 0.15 mM of hydrogen peroxide (H₂O₂), in order to produce a stress situation into cells. Following, cells were incubated with several concentrations of MA as indicated below.

4.3. MTT assay

MTT assay was performed as described by Mokhtari et al. [15]. Briefly, 200 μ L cell suspension of B16F10 or A10 ($1.5 \cdot 10^3$ cells/well) were cultured in 96 well plates. Subsequent to the adherence of the cells, different MA dilutions on a scale of 10 to 210 μ M were added separately. Incubation times were 24 h for all cases. MTT was dissolved in the medium and added to the wells at a final concentration of 0.5 mg/mL. Following 2 h of incubation, the generated formazan was dissolved in DMSO. Absorbance was measured at 570 nm in a multiplate reader (Bio-tek®). Absorbance was proportional to the number of viable cells. The MA concentration leading to 50% inhibition of cell proliferation (IC_{50}) was determined. The results are expressed as percentage of live cells compared with the control considered as 100% cell viability. Cell viability in B16F10 and A10 cells were also studied in presence of H_2O_2 by MTT assay. H_2O_2 was dissolved in the culture medium of the cell lines. The concentrations used were made extemporaneously and protected by light before use to prevent degradation of the compound. After 24 hours of incubation the medium was removed, adding fresh medium with different concentrations of H_2O_2 per well: 0; 0.05; 0.1; 0.2; 0.4; 0.6; 0.8; 1.1; 1.5; 2.0; 2.6 and 3 mM to a final volume of 200 μ L for 24h.

4.4. Flow-cytometry analysis of the mitochondrial-membrane potential

Changes in the mitochondrial-membrane potential can be examined by monitoring the cell fluorescence after double staining with rhodamine 123 (Rh123) and propidium iodide (PI). Rh123 is a membrane-permeable fluorescent cationic dye that is selectively taken up by mitochondria directly proportional to the MMP (mitochondrial-membrane permeabilization) [47]. B16F10 and A10 cells (4×10^5 cells/well) were seeded on 12-well plates with 2 mL of medium and treated with 0.15 mM for 24 h and MA at $IC_{50/4}$, $IC_{50/2}$, IC_{50} , and $2 \cdot IC_{50}$ (10.6, 21.6; 42.3 and 84.6 μ M, respectively) concentrations during 24 h more. Following the treatment, the medium was removed and a fresh medium with DHR, at a final concentration of 5 μ g/mL, was added. After 30 min of incubation, the medium was removed and the cells were washed and resuspended in PBS with 5 μ g/mL of PI. The intensity of fluorescence from Rh123 and PI was determined using an ACS flow cytometer (Coulter Corporation, Hialeah, FL, USA), at the excitation and emission wavelengths of 500 nm and 536 nm, respectively. The experiments were performed three times with two replicates per assay.

4.5. Anti-oxidant enzymes assays

In order to prepare the samples for analytic procedures, cells were homogenized in RIPA buffer. Immediately, cells were sonicated on ice for 5 min and maintained by moderate shaking at 4°C for 1 h. Every 15 min, the samples were moderately shaken in a vortex. The lysates were spun in a centrifuge at 10,000 g at 4°C for 15 min. All enzyme assays were carried out at 37°C using a Power Wave X microplate scanning spectrophotometer (Bio-Tek Instruments, USA) and run in duplicate in 96-well microplates. The optimal substrate and protein concentrations for the measurement of maximal activity for each enzyme were established by preliminary assays. The enzymatic reactions were initiated by addition of the cell extract, except for SOD where xanthine oxidase was used. The millimolar extinction coefficients used for H_2O_2 , NADH/NADPH, and DTNB [5,5-dithiobis (2-nitrobenzoic acid)] were 0.039, 6.22, and 13.6, respectively. The assay conditions were as follows:

Superoxide dismutase (SOD; EC 1.15.1.1) activity was measured by the ferricytochrome c method using xanthine/xanthine oxidase as the source of superoxide radicals. The reaction mixture consisted of 50 mM potassium phosphate buffer (pH 7.8), 0.1 mM EDTA, 0.1 mM xanthine, 0.013 mM cytochrome c, and 0.024 IU/mL xanthine oxidase. Activity is reported in units of SOD per milligram of protein. One unit of activity was defined as the amount of enzyme necessary to produce a 50% inhibition of the ferricytochrome c reduction rate [48].

Catalase (CAT; EC 1.11.1.6) activity was determined by measuring the decrease of hydrogen peroxide concentration at 240 nm according to Aebi [49]. The reaction mixture contained 50 mM potassium phosphate buffer (pH 7.0) and 10.6 mM of freshly prepared H₂O₂.

Glucose-6-phosphate dehydrogenase (G6PDH; EC 1.1.1.49) activity was determined at pH 7.6 in a medium containing 50 mM Hepes buffer, 2 mM MgCl₂, 0.8 mM NADP⁺ and glucose 6-phosphate were used as substrate. The enzyme activity was determined by measuring the reduction of NADP⁺ at 340 nm as previously described by Lupiáñez et al. [2] and Peragón et al. [50]. The change in absorbance at 340 nm was recorded and, after confirmation of no exogenous activity, the reaction started by the addition of substrate.

Glutathione S-transferase (GST; EC 2.5.1.18) activity was measured according to the method described by Habig et al. [51], using 1-chloro-2,4-dinitrobenzene as a substrate.

Glutathione peroxidase (GPX, EC 1.11.1.9) activity was determined using the method described by Flohe and Günzler [52], based on the oxidation of NADPH, which is used to regenerate the reduced glutathione (GSH) from oxidized glutathione (GS-SG) obtained by the action of glutathione peroxidase.

Glutathione reductase (GR, EC 1.8.1.7) was determined by the modified method of Carlberg and Mannervik [53]. We measured the decrease in absorbance produced by the oxidation of NADPH used by GR in the passage of oxidized glutathione (GS-SG) to reduced glutathione (GSH).

All enzyme activities (except SOD) are expressed as units or milliunits per milligram of soluble protein (specific activity). One unit of enzyme activity was defined as the amount of enzyme required to transform 1 μmol of substrate per min under the above assay conditions.

Soluble protein concentration was determined using the method of Bradford, with bovine serum albumin used as a standard.

4.6. Statistical analysis

Data are showed as mean ± standard error mean (SEM). The statistical significance among different experimental groups was determined by one-way analysis of variance (ANOVA) test. When *F* values (*P* < 0.05) were significant, means were compared using Tukey's HSD test. SPSS version 15.0 for Windows software package was used for statistical analysis.

5. Conclusions

In conclusion, the results obtained in the present study demonstrate that MA exerts a selective anti-proliferative effect against B16F10 cell line, whereas in A10 healthy cells, MA did not present cytotoxic effect. This natural compound prevent oxidative stress caused by H₂O₂ excess decreasing the ROS production levels. Moreover, depending on cell line, the anti-oxidant enzymatic response were different in presence H₂O₂ and without MA. So, in healthy A10 cells, all enzymes up-regulated their activity, but in B16F10 cells only SOD, GST and GPX increased it. In most cases, MA treatment restores the activities of enzymes to similar levels of the control group, highlighting that in A10 cells the doses of MA highest (IC₅₀ and 2-IC₅₀) presented values below that control. Overall, these findings demonstrate the great anti-oxidant capacity of maslinic acid.

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Conflicts of Interest: The authors of this work declare that this research has been carried out in the absence of any commercial or financial relationship and, therefore, does not present any conflict of interest now or in the future.

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