

Article

Effect of roasting time and cryogenic milling on the physicochemical characteristics of dried ginseng powder

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Abstract: This study aimed to decrease the particle size of ginseng by roasting and cryogenic milling to increase its water solubility and physiological activity. The samples were roasted for different times (9–21 min) and generated in different sizes (10–50, and > 50 μm). All roasted samples revealed significantly smaller particle sizes than did non-roasted samples based on Sauter mean diameter (D [3,2], $p < 0.05$). Further, the particle sizes of roasted samples decreased till roasting up to 15 min. In terms of the water solubility index (WSI), antioxidant activity, total polyphenol content (TPC), and total polysaccharides according to particle size, 10–20 μm -sized samples showed the highest values when compared with >50 μm -sized samples. Based on roasting time, WSI values of all samples roasted for up to 15 min were higher than those of the control (not roasted) ($p < 0.05$). Antioxidant activity and TPC also increased with increasing roasting time. Total polysaccharide content was the highest upon roasting for 15 min except for the 10–20 μm sample. Ginsenoside content of roasted samples >20 μm size was higher than that of the control (not roasted) except after 15 min of roasting. Therefore, roasting and cryogenic milling are effective in producing ginseng root powder.

Keywords: cryogenic milling, ginseng, particle size, physiological activity, roasting, water solubility

1. Introduction

Ginseng (*Panax ginseng* C.A. Meyer) has been known as general herbal medicine and functional food in south and southeast Asia for over 2,000 years [1]. Ginseng root is more widely used than the leaf and stem of ginseng because of residual pesticides [2]. Ginseng roots also have many bioactive constituents such as ginsenosides, acidic polysaccharides, and phenolic compounds [3–5]. In particular, major ginsenosides (Rg1, Re, Rf, Rb1, Rc, Rb2, Rb3, and Rd) and minor ones ((Rg2(S), Rg2(R), Rg3(S), Rg3(R), and Rh1(S))) in ginseng roots have many functions including anti-cancer, antioxidation, anti-stress, blood pressure adjustment, and immune system enhancement effects [6–8]. Currently, ginseng products mostly exist in the form of extracts and powders [9]. Ginseng powder forms are more effective than extract forms, because it contains all the bioactive components with hydrophilic, hydrophobic, and amphiphilic properties [9]. However, ginseng powder has problems of low water solubility, which need to be improved.

Roasting is an appropriate pretreatment solution to improve the low solubility of food products. When roasting raw materials, various components are changed by reactions such as decomposition, synthesis, and condensation [10]. This facilitates the dissolution of components contained in food by high temperature [10]. Further, the roasting treatment breaks down plant cell wall and increases intracellular space, resulting in easy extraction of bioactive components and increased water-soluble contents [11–13]. Thus, roasting increases the ginsenoside, browning substance (melanoid), acidic polysaccharide, and antioxidant contents [14–17]. As along with roasting, this study used a cryogenic (very-low temperature) milling machine to prepare ginseng powder, which is not common in commercial ginseng grinding. Cryogenic milling is a pulverization method that uses a liquid nitrogen bath at -196°C .



Cryogenic milling reduces particle sizes by weakening the cell wall and increases the water solubility index when compared with effects of high temperature milling [18]. Manohar *et al* (2001) reported that the particle size of *Curcuma domestica* subjected to cryogenic milling (8.56 μm) was smaller when compared with effects of hammer milling (88.2 μm) [19]. Cryogenic milling also prevents discoloration and oxidation by high heat [20].

Therefore, this study aimed to investigate the effect of roasting and cryogenic milling on increasing the water solubility and bioactive constituents in ginseng root. Experiments were also performed to compare the physicochemical properties such as particle size, appearance, color, water solubility index (WSI), DPPH scavenging activity, total polyphenol content (TPC), total/acidic polysaccharide content, and ginsenoside content based on particle size and roasting.

2. Materials and Methods

2.1. Ginseng preparation and pretreatment

Four-year-old dried ginseng roots (*Panax ginseng*) were purchased from Bio hongsam (Geumsan, Korea). Ginseng roots were cut to approximately 0.5 cm thickness and the cut ginseng roots (125 g) were roasted for 9, 12, 15, 18, and 21 min at 180°C using a roaster (CBR-101; Gene Café, Ansan, Korea).

2.2. Preparation of roasted ginseng powder

Roasted ginseng roots were ground using a cryogenic milling machine (SPEX 6875D Freezer/Mill; Spex sampleprep, Metuchen, NJ, USA). The milling machine was pre-cooled for 5 min to reach -196°C before grinding. The ginseng was placed in a vial and milled at 15 cycles per second (CPS, the number of cycles in which the steel impactor moved forward and backward per second) for 3 min. The milling machine was also cooled for 1 min between repetitions of the milling cycle.

2.3. Treatments

Figure 1 presents the cryogenic milling conditions of dried ginseng root. The cryogenic milling condition was set in three stages including the cycles, rate (CPS), and time. The conditions with 1 cycle were based on 5 min of precooling, 3 min of milling time, 1 min of cooling time, and 15 CPS milling rate. The ranges of each stage were set to 2-12 cycles, 5-15 CPSs, and 0.5-5 min. After cryogenic milling of ginseng treatments under different conditions, the treatments were classified according to particle size based on the Sauter mean diameter, D [3, 2] (Table 1). In the range of each size, one treatment was selected as a representative condition. The samples were roasted and then ground by cryogenic milling using the selected condition. Roasting as pretreatment step was performed for 9, 12, 15, 18, and 21 min at 180°C (Table 2).

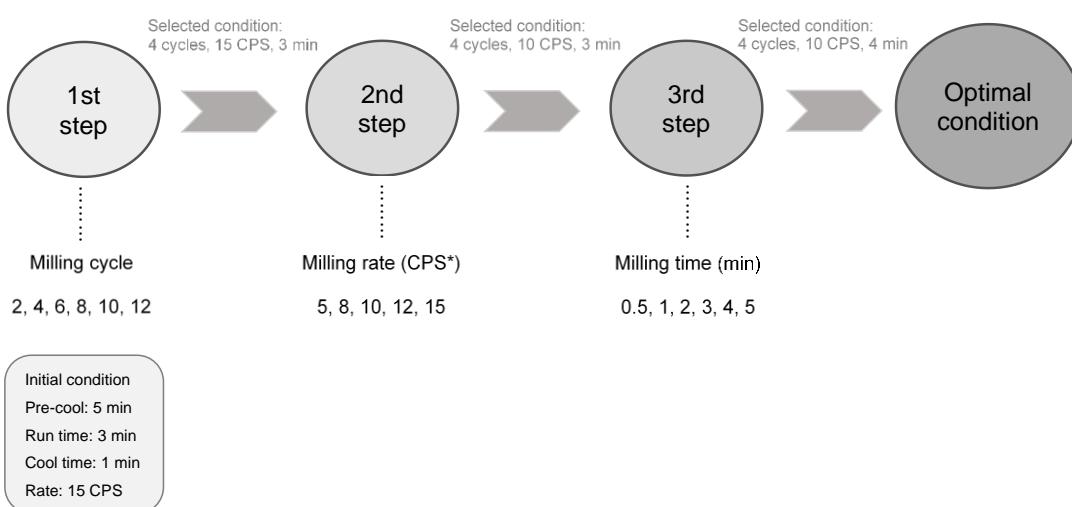


Fig. 1. Cryogenic milling conditions for dried ginseng root. The condition of 1 cycle was 3 min-milling at a rate of 15 CPS. CPS* indicates the number of back-and-forth cycles per second completed by the impactor.

2.4. Particle size

Particle size was measured using a laser diffraction particle size analyzer (Mastersizer 3000; Malvern, Worcestershire, UK). The measurement conditions of particle size were presented in a dry condition as follows: refractive index of particles, 1.520; absorption index of particles, 0.01; dispersion medium, dry; dispersion refractive index, 1; particle shape, non-spherical. After placing the ginseng powder on the specimen inlet, the sample was automatically injected using the flow rate (50% power) and the feed rate (50% power). The particle size of ginseng powders was analyzed at least 5 times.

2.5. Color

A petri dish with a size of 35×10 mm was completely filled with ginseng powders. Color parameters of ginseng powder were measured using a CR-400 colorimeter (Konica minolta sensing, Tokyo, Japan) after calibration with a standard plate (L^* , 96.67; a^* , 0.21; and b^* , 1.82). All samples were measured 5 times.

2.6. WSI

Average WSI of all ginseng powders was analyzed in triplicates as described by Lee *et al* (2013) [21]. Ginseng powder (0.5 g) was dissolved in 30 mL of distilled water, and the solution was shaken for 60 min at 80°C in a water bath (BF-30SB; Biofree, Seoul, Korea). The solution was then centrifuged using a centrifugal separator (1736R; GYROZEN, Daejeon, Korea) for 30 min at 10,000 rpm. The supernatant was filtered through a Whatman No. 2 paper (Sigma, St. Louis, MO, USA). Next, the filtered solution was dried for 24 h at 60°C using a hot air dryer (LD-918TH; Lequip, Hawsung, Korea). WSI was calculated using the following equation:

$$\text{WSI (\%)} = (\text{Weight of dried ginseng solution} / \text{Weight of ginseng powder}) \times 100$$

2.7. Antioxidant extract preparation

Each 1 g of ginseng sample was extracted with 25 mL of 70% ethanol using a water bath (BF-30SB; Biofree) for 3 h at 80°C. The extracted ginseng solution was filtered through a Whatman No. 2 paper (Sigma) and the filtrate was concentrated using an EYELA rotary evaporator N-1000 (Sunileyela, Seongnam, Korea). The concentrate was then used as a sample after lyophilization using a freeze-dryer (MCFD8512; Ilshinbiobase, Yangju, Korea).

2.8. DPPH free radical scavenging activity

Antioxidant activity was measured by modifying the Blois method of removing 2,2-diphenyl-1-picryl-hydrazyl (DPPH) (Sigma, St. Louis, MO, USA) free radical [22]. The ginseng solution (160 μ L) diluted with HPLC-grade methanol was mixed with 40 μ L of DPPH reagent (Ab 0.95-0.99). In the same way, instead of DPPH reagent, methanol (40 μ L) was added to the ginseng solution (160 μ L) for calibration of sample color. The control was prepared by mixing methanol (160 μ L) and DPPH (40 μ L). All the samples were reacted at room temperature (22-24 °C) in dark for 10 min. Absorbance of the samples was analyzed at 517 nm using a Multiskan™ GO plate reader (Thermo Scientific, Waltham, MA, USA). The scavenging activity of DPPH free radicals was calculated by the following equation. Next, the antioxidant activity was converted to an IC50 value representing the amount of ginseng that removed 50% of DPPH free radical.

$$\text{DPPH free radical scavenging activity (\%)} = [1 - \{(A1 - A2) / A3\}] \times 100$$

A1: Absorbance of DPPH solution (40 μ L) + ginseng solution (160 μ L)

A2: Absorbance of methanol (40 μ L) + ginseng solution (160 μ L)

A3: Absorbance of DPPH (40 μ L) + methanol (160 μ L)

2.9. Ginseng extract

To manufacture ginseng extract, 0.5 g of ginseng powder was dissolved in distilled water (50 mL) using a water bath (BF-30SB; Biofree) for 2 h at 95°C. Ginseng solutions were then centrifuged using a centrifugal separator (1736R; GYROZEN) for 10 min at 10,000 rpm and filtered. Then, samples were adjusted to a volumetric flask (50 mL) for the measurement of physicochemical characteristics (total polyphenol and total acidic polysaccharide contents).

2.9.1. Total polyphenol content (TPC)

TPC was measured using the Folin-Denis method (Ough and Amerine 1980) [23]. Ginseng extract (2 mL) was vortexed with 2 mL of Folin-Ciocalteu reagent for 3 min, and 10% Na₂CO₃ (2 mL) was added to the mixture. The final mixtures were reacted for 1 h at room temperature in dark, and the sample absorbance was analyzed at 700 nm using a Multiskan™ GO plate reader (Thermo Scientific). The TPC was calculated based on a standard curve prepared using gallic acid (Sigma-Aldrich, St. Louis, MO, USA).

2.9.2. Total polysaccharide content

Total polysaccharide content was analyzed by a method using phenol and sulfuric acid reagents [24]. The extracted ginseng solution (0.6 mL) was mixed with 5% phenol (0.3 mL) using a vortex mixer, and 1.5 mL concentrated sulfuric acid was added. The mixed solution was shaken for 30 min at 85°C in a water bath (BF-30SB; Biofree), and cooled for 5 min at room temperature in the dark. Absorbance of the final mixture (0.2 mL) was analyzed at 490 nm using a Multiskan™ GO plate reader (Thermo Scientific). The total polysaccharide content was obtained using a standard curve prepared with D-glucose (Samchun, Seoul, Korea).

2.9.3. Acidic polysaccharide content

The acidic polysaccharide content was analyzed using carbazole and sulfuric acid reagents [25]. The extracted ginseng solution (0.5 mL) was mixed with 0.25 mL of carbazole-ethanol reagent (Alfa Aesar, Haverhill, MA, USA), and then with 3 mL of H₂SO₄. The mixed solution was shaken for 5 min at 80°C in a water bath (BF-30SB; Biofree), and cooled for 15 min in dark at room temperature. Absorbance of the final mixture (0.2 mL) was then analyzed at 525 nm. The acidic polysaccharide content of ginseng was determined using a standard curve prepared with D-galacturonic acid (Sigma, St. Louis, MO, USA).

2.10. Ginsenoside content

After mixing 2 g of ginseng powder with 70% methanol (25 mL), crude ginsenoside was extracted using a Funnel Shaker (RS-1; Jeio Tech, Dajeon, Korea). The extract was adjusted to 50 mL using a volumetric flask. The ginseng solution was then centrifuged at 3,500 rpm for 5 min and filtered with a 0.22 µm syringe filter (Millipore, Burlington, MA, USA). The filtered ginseng solution was subjected to high-performance liquid chromatography (1260 Infinity II LC System; Agilent, Santa Clara, CA, USA) with a C18 HPLC column (Kinetex, 5 µm, 250×4.6 mm; Phenomenex, Seoul, Korea). The temperature condition of the column was set to 30°C and the flow rate was 1.0 mL/min. The ginsenoside content was measured by solvent gradient analysis with water and acetonitrile, and the content of acetonitrile was set as follows: (0-5 min, 20%; 20 min, 23%; 25 min, 30%; 45 min, 40%; 55-65 min, 50%; and 70-75 min, 20%). The total operating time was 75 min. The injection volume of the sample was 10 µL. The sample absorbance was measured at 203 nm.

2.11. Statistical analysis

All analyses of ginseng powder were conducted at least 3 times. All result values were indicated as the average value ± standard deviation using the software SPSS 22.0 (SPSS Inc., USA). Verification of significant differences was performed using Duncan's multiple range test ($p < 0.05$) after one-way analysis of variance (ANOVA).

3. Results and Discussion

3.1 Particle size

In our preliminary study, the optimal particle size of ginseng was obtained with wet milling and cryogenic milling, respectively. The particle size of ginseng powder ground using wet milling was 114 µm based on the D [3,2] value (data not shown). In contrast, the particle size of ginseng powder using cryogenic milling was significantly decreased to 14 µm based on the D [3,2] value ($p < 0.05$). This result was in agreement with a previous study showing that the particle sizes of ginseng roots ground using wet milling and dry milling were 139.3 µm and 15 µm, respectively, under optimum conditions [26]. Therefore, cryogenic milling was used in this study since particle size reduction was found to be effective. Cryogenic milling is a dry milling process, involving grinding at ultra-low temperatures. It increases the hardness and decreases the extensibility of the ginseng cell wall, thus

making it fragile, which results in rapid particle size reduction [18]. In this study, ginseng powders of different sizes were prepared using different conditions of cycle, rate (CPS), and milling time in a cryogenic milling machine (Table 1). Then, the five representative samples (1, 6, 11, 14, and 15) were chosen for each size group from 15 ginseng samples with different sizes, and the chosen samples were roasted for different times. In this study, roasting temperature was 180°C, which was the optimal temperature in preliminary experiments conducted at various temperatures (160, 170, 180, 190, and 200°C). Therefore, roasting was performed for 9, 12, 15, 18, and 21 min according to the roasting time based on 180°C. Table 2 presents the particle sizes of ginseng powder samples with different roasting time. All roasted treatments demonstrated significant size reduction compared with the control (not roasted) based on the D [3,2] value ($p < 0.05$). This result was in agreement with the previous study that roasted ginseng powder (below 30.0 μm) had a smaller particle size than that of non-roasted ginseng powder (1,000-1,250 μm) [18]. Kim *et al.* (2004) reported that roasting weakens the starch-protein interactions and destroys the starch granules, eventually reducing the particle size [26]. As the roasting time increased, a decreasing pattern of particle sizes was observed up to 15 min. Im *et al* (2010) also showed that the initial particle size (177.2 μm) was decreased with increasing milling times (10, 20, 30, and 40 min), which finally decreased to 92.2 μm after 40 min of grinding [27]. However, when the all roasted treatments were roasted for over 18 min, the particle size was found to be increased again in this study. These results were consistent with previous studies showing that flocculation of powder occurred as the particle size decreased [28,29]. The flocculation seemed to arise due to Van der Waals forces produced by increased surface area and energy [30]. However, the particle size of all roasted ginseng samples was smaller than the control based on the D [3,2] value ($p < 0.05$). Therefore, roasting seemed to be an effective pretreatment for size reduction.

Table 1. Treatment of dried ginseng powder by particle size with various cryogenic milling conditions

Size (μm)	Treatment	Treatment conditions			Particle size (μm)				
		Cycle	CPS	Time	D [3,2]	D [4,3]	Dx (10)	Dx (50)	Dx (90)
10-20	1	4	10	4	14.0 \pm 0.2 ^c	68.20 \pm 7.10 ^{ib}	5.80 \pm 0.2 ^{ib}	28.50 \pm 0.9 ^{gc}	145.9 \pm 8.50 ^{ib}
	2	4	10	3	15.0 \pm 0.7 ^{ib}	63.80 \pm 4.50 ^{ib}	6.10 \pm 0.3 ^{ia}	30.80 \pm 1.2 ^{gb}	150.9 \pm 8.10 ^{ib}
	3	4	10	5	15.5 \pm 0.6 ^{ia}	89.40 \pm 10.9 ^{ha}	6.20 \pm 0.2 ^{ia}	32.10 \pm 1.7 ^{ga}	193.5 \pm 16.5 ^{ha}
20-30	4	4	10	2	21.0 \pm 2.2 ^{ie}	114.4 \pm 5.50 ^{fgBC}	8.10 \pm 0.9 ^{ie}	47.60 \pm 1.4 ^{fc}	308.4 \pm 15.7 ^{ic}
	5	4	8	3	22.5 \pm 0.4 ^{hd}	105.1 \pm 3.70 ^{gc}	8.90 \pm 0.3 ^{hd}	47.80 \pm 0.8 ^{fc}	263.8 \pm 11.1 ^{gd}
	6	4	12	3	24.9 \pm 0.4 ^{gc}	113.7 \pm 4.20 ^{fgBC}	9.60 \pm 0.1 ^{gc}	57.20 \pm 0.8 ^{eb}	299.2 \pm 4.70 ^{icd}
	7	6	15	3	28.7 \pm 1.0 ^{ea}	131.5 \pm 18.0 ^{ea}	11.8 \pm 0.3 ^{eb}	53.70 \pm 3.5 ^{ea}	377.9 \pm 78.2 ^{da}
	8	4	15	3	30.1 \pm 1.1 ^{fb}	124.0 \pm 8.20 ^{fb}	13.8 \pm 0.2 ^{da}	60.50 \pm 2.6 ^{ea}	345.5 \pm 40.4 ^{eb}
30-40	9	8	15	3	31.7 \pm 1.2 ^{ec}	162.8 \pm 17.2 ^{ec}	12.9 \pm 0.3 ^{eb}	60.80 \pm 4.7 ^{ec}	522.3 \pm 54.0 ^{dc}
	10	10	15	3	31.9 \pm 1.6 ^{ec}	177.0 \pm 14.4 ^{db}	11.9 \pm 0.3 ^{fd}	61.10 \pm 4.3 ^{ec}	567.3 \pm 45.4 ^{cb}
	11	4	10	1	34.5 \pm 0.8 ^{db}	215.5 \pm 11.2 ^{ba}	12.4 \pm 0.3 ^{ec}	113.8 \pm 9.1 ^{ba}	591.7 \pm 24.5 ^{bcB}
40-50	12	12	15	3	40.0 \pm 2.8 ^{ca}	226.0 \pm 3.00 ^{ba}	15.2 \pm 0.5 ^{ca}	86.70 \pm 6.3 ^{dB}	664.2 \pm 33.0 ^{aA}
	13	4	5	3	40.6 \pm 1.0 ^{ca}	217.6 \pm 6.40 ^{ba}	15.0 \pm 0.3 ^{ca}	111.3 \pm 4.4 ^{ba}	602.4 \pm 15.8 ^{bb}
	14	2	15	3	46.8 \pm 1.0 ^b	198.8 \pm 7.00 ^c	19.0 \pm 0.4 ^b	104.0 \pm 1.4 ^c	529.4 \pm 22.8 ^d
>50	15	4	10	0.5	64.1 \pm 5.3 ^a	289.0 \pm 5.00 ^a	25.0 \pm 2.1 ^a	207.4 \pm 6.3 ^a	690.4 \pm 10.6 ^a

^{a-j}Means with different superscripts within whole particle size group are significantly different ($p < 0.05$). ^{A-E}Means with different superscripts within the similar size group are significantly different ($p < 0.05$). Particle size (μm) set up based on D[3,2]. Fifteen treatments were generated according to cycle, CPS (the number of back-and-forth cycles per second), and time conditions, and the larger the number of treatments, the larger is the particle size. One cycle is based on pulverization for 3 min at a rate of 15 CPS oscillating 15 times per second.

Table 2. Particle size of dried ginseng powder with various roasting time conditions

Size (μm)	Roasting ¹⁾ Time (min)	Particle size (μm)			
		D [3,2]	D [4,3]	Dx (10)	Dx (50)
10-20	Control	14.00 \pm 0.20 ^{eA}	68.20 \pm 7.10 ^{eA}	5.80 \pm 0.20 ^{eA}	28.50 \pm 0.90 ^{eA}
	9	8.22 \pm 0.33 ^{eC}	45.78 \pm 4.61 ^{eB}	3.28 \pm 0.11 ^{eC}	18.67 \pm 1.70 ^{eB}
	12	7.58 \pm 0.27 ^{dD}	30.83 \pm 2.86 ^{eC}	3.04 \pm 0.10 ^{eD}	15.87 \pm 1.13 ^{eC}
	15	5.60 \pm 0.12 ^{eF}	20.94 \pm 0.88 ^{dD}	2.19 \pm 0.05 ^{eE}	10.29 \pm 0.63 ^{eE}
	18	7.06 \pm 0.10 ^{eE}	26.03 \pm 2.02 ^{eCD}	2.90 \pm 0.07 ^{eD}	13.93 \pm 0.30 ^{eD}
	21	9.27 \pm 0.10 ^{dB}	29.38 \pm 3.99 ^{dC}	3.89 \pm 0.05 ^{dB}	17.77 \pm 1.67 ^{bB}
20-30	Control	24.90 \pm 0.40 ^{dA}	113.70 \pm 4.20 ^{dA}	9.60 \pm 0.10 ^{dA}	57.20 \pm 0.80 ^{dA}
	9	20.34 \pm 1.52 ^{dB}	112.50 \pm 5.45 ^{dA}	7.87 \pm 0.15 ^{dB}	48.94 \pm 1.14 ^{dB}
	12	16.92 \pm 0.63 ^{dD}	92.02 \pm 13.90 ^{dB}	7.10 \pm 0.34 ^{cC}	35.08 \pm 0.30 ^{dC}
	15	9.80 \pm 0.44 ^{dE}	52.40 \pm 11.92 ^{cD}	3.87 \pm 0.13 ^{dE}	22.56 \pm 2.38 ^{dD}
	18	17.38 \pm 0.39 ^{dCD}	68.33 \pm 5.95 ^{dC}	6.51 \pm 0.69 ^{dD}	35.58 \pm 6.31 ^{dC}
	21	17.62 \pm 0.48 ^{cC}	97.50 \pm 8.48 ^{cB}	7.07 \pm 0.24 ^{cC}	37.62 \pm 1.48 ^{bC}
30-40	Control	34.50 \pm 0.80 ^{cA}	215.50 \pm 11.20 ^{bA}	12.40 \pm 0.30 ^{cA}	113.80 \pm 9.10 ^{bA}
	9	30.60 \pm 0.84 ^{cB}	181.00 \pm 10.86 ^{cB}	11.88 \pm 0.44 ^{cB}	87.08 \pm 1.99 ^{cB}
	12	16.45 \pm 0.13 ^{cD}	137.40 \pm 9.10 ^{cD}	6.38 \pm 0.12 ^{dD}	48.70 \pm 1.21 ^{cD}
	15	12.66 \pm 0.59 ^{cE}	62.40 \pm 2.49 ^{cE}	4.92 \pm 0.16 ^{cE}	31.21 \pm 2.28 ^{cE}
	18	23.58 \pm 0.99 ^{cC}	143.75 \pm 3.86 ^{cCD}	9.26 \pm 0.48 ^{cC}	57.28 \pm 2.61 ^{cC}
	21	24.22 \pm 0.79 ^{bC}	155.00 \pm 3.56 ^{bC}	9.27 \pm 0.46 ^{bC}	62.08 \pm 1.65 ^{bC}
40-50	Control	46.80 \pm 1.00 ^{bA}	198.80 \pm 7.00 ^{cA}	19.00 \pm 0.40 ^{bA}	104.00 \pm 1.40 ^{cB}
	9	37.98 \pm 2.72 ^{bB}	208.67 \pm 6.81 ^{bA}	14.18 \pm 1.18 ^{bB}	127.25 \pm 11.76 ^{bA}
	12	28.80 \pm 0.73 ^{bC}	156.60 \pm 3.78 ^{bB}	11.14 \pm 0.46 ^{bC}	75.54 \pm 1.84 ^{bC}
	15	14.88 \pm 0.53 ^{bE}	90.56 \pm 11.83 ^{bC}	5.98 \pm 0.28 ^{bE}	41.40 \pm 0.88 ^{bD}
	18	26.83 \pm 0.68 ^{bD}	156.00 \pm 3.87 ^{bB}	10.36 \pm 0.46 ^{bD}	73.58 \pm 3.24 ^{bC}
	21	26.46 \pm 0.50 ^{bD}	157.40 \pm 4.28 ^{bB}	10.05 \pm 0.32 ^{bD}	70.20 \pm 1.90 ^{bC}
>50	Control	64.10 \pm 5.30 ^{aA}	289.00 \pm 5.00 ^{aB}	25.00 \pm 2.10 ^{aA}	207.40 \pm 6.30 ^{aAB}
	9	48.33 \pm 0.96 ^{aB}	282.25 \pm 10.14 ^{aB}	18.80 \pm 0.75 ^{aB}	193.50 \pm 7.33 ^{aB}
	12	45.83 \pm 0.51 ^{aB}	287.67 \pm 0.58 ^{aB}	16.53 \pm 0.38 ^{aB}	197.75 \pm 7.14 ^{aAB}
	15	25.58 \pm 0.10 ^{aC}	219.40 \pm 4.28 ^{aC}	9.73 \pm 0.12 ^{aC}	96.60 \pm 3.32 ^{aC}
	18	43.26 \pm 1.31 ^{aB}	293.80 \pm 14.82 ^{aB}	15.76 \pm 0.64 ^{aB}	192.00 \pm 19.7 ^{aB}
	21	45.96 \pm 6.74 ^{aB}	338.57 \pm 57.82 ^{aA}	16.97 \pm 3.84 ^{aB}	265.71 \pm 95.90 ^{aA}

^{a-e}Means with different superscripts within the same time group are significantly different ($p < 0.05$).

^{A-F}Means with different superscripts within the same size group are significantly different ($p < 0.05$).

¹⁾After cryogenic milling of ginseng samples with different conditions, the samples were classified with by their particle sizes. One treatment condition in each size range was then selected, and samples selected by particle size were roasted and pulverized by cryogenic milling.

3.2 Appearance

The appearances of dried raw ginseng are shown in Figure 2 to compare the changes before and after roasting. As the roasting time increased, the color of ginseng seemed to be gradually changed to black, and started to be burned in case of roasting over 18 min. After dried ginseng was pulverized by cryogenic milling, Figure 3 presents the appearance of ginseng powder before and after roasting. According to the particle sizes, the appearance of the control group (not roasted) was rougher as the particle size was large. However, the roasted powders seemed to be softer than the control. According to the roasting times, the color of ginseng powders appeared to be gradually darkened with

increasing roasting time. The color of ginseng powder darkened as the particle size increased and the roasting time increased.

	Roasting time (min)					
	Control	9	12	15	18	21
Ginseng						

Fig. 2. Appearance of dried ginseng root roasted for different times.

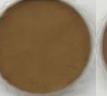
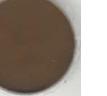
Size (μm)						
	10-20	20-30	30-40	40-50	>50	
10-20						
20-30						
30-40						
40-50						
>50						

Fig. 3. Appearance of dried ginseng root powder pretreated with different roasting times.

3.3 Color

Color changes in ginseng powder were observed according to particle size and roasting time (Table 3). When comparing with the control alone as the particle size increased, the color values showed lower lightness (L^*) and higher redness (a^*) and yellowness (b^*). All color values of controls were significantly different except for the a^* and b^* values of the samples with sizes $>50 \mu\text{m}$ ($p < 0.05$). Kim *et al.* (2012) reported that in general, for powders, L^* values increased as the particle size decreased, whereas the a^* and b^* values varied according to the sample [31]. The smaller the particle size, the higher is the scattered light, and thus the higher is the L^* [32]. With the pretreatment of roasting, L^* values of roasted ginseng powders were significantly lower than those of the controls ($p < 0.05$), whereas the a^* values were significantly higher than those of the controls ($p < 0.05$). Further, b^* values were lower than those of the controls when roasting was done for over 12 min except for 10-20 μm sized samples ($p < 0.05$). As the roasting time increased in roasted treatments, the L^* and b^* values of all samples were significantly decreased ($p < 0.05$). The a^* values tended to increase until roasting for 12 min. However, the a^* values were significantly decreased when samples were roasted for over 15 min with an exception of 10-20 μm sized samples ($p < 0.05$). This result was in agreement

with a previous study showing that L^* and b^* values tended to decrease as the roasting time (5.27, 6.19, 6.39, and 8.01 min) increased at 170°C [33]. Further, the a^* values tended to increase at the early stage of roasting and decreased again as the roasting time increased [33]. These color changes were caused by the Maillard reaction of non-enzymatic browning and caramelization [33]. Particle size and roasting time was found to affect the color parameters of ginseng powder.

Table 3. Color of dried ginseng root powder with various roasting temperature conditions

Size (μm)	Roasting ¹⁾ Time (min)	Color					
		L^*		a^*		b^*	
10-20	Control	82.13	\pm	0.54 ^{aA}	0.71	\pm	0.22 ^{dC}
	9	65.65	\pm	0.04 ^{aB}	5.26	\pm	0.00 ^{eB}
	12	57.65	\pm	0.05 ^{aC}	6.01	\pm	0.02 ^{eA}
	15	52.35	\pm	0.08 ^{aD}	5.44	\pm	0.03 ^{dB}
	18	45.95	\pm	0.07 ^{aE}	5.87	\pm	0.01 ^{aA}
	21	44.53	\pm	0.05 ^{aF}	5.42	\pm	0.01 ^{aB}
20-30	Control	80.44	\pm	0.25 ^{bA}	1.07	\pm	0.04 ^{cF}
	9	60.99	\pm	0.06 ^{cB}	5.94	\pm	0.01 ^{dC}
	12	54.27	\pm	0.13 ^{bC}	6.29	\pm	0.01 ^{dA}
	15	44.59	\pm	0.09 ^{bD}	6.04	\pm	0.05 ^{bB}
	18	44.03	\pm	0.02 ^{bE}	5.55	\pm	0.02 ^{cD}
	21	40.31	\pm	0.01 ^{bF}	5.27	\pm	0.00 ^{bE}
30-40	Control	78.77	\pm	0.77 ^{cA}	1.41	\pm	0.09 ^{bF}
	9	61.23	\pm	0.05 ^{bB}	6.01	\pm	0.01 ^{cC}
	12	51.71	\pm	0.14 ^{cC}	6.66	\pm	0.02 ^{bA}
	15	43.83	\pm	0.16 ^{cD}	6.32	\pm	0.02 ^{aB}
	18	42.03	\pm	0.02 ^{cE}	5.71	\pm	0.02 ^{bD}
	21	38.19	\pm	0.13 ^{cF}	4.99	\pm	0.01 ^{cE}
40-50	Control	77.57	\pm	0.04 ^{dA}	1.83	\pm	0.02 ^{aF}
	9	59.63	\pm	0.08 ^{dB}	6.49	\pm	0.01 ^{bB}
	12	51.65	\pm	0.14 ^{cC}	6.56	\pm	0.03 ^{aA}
	15	43.36	\pm	0.03 ^{dD}	5.91	\pm	0.03 ^{cC}
	18	40.78	\pm	0.14 ^{dE}	5.54	\pm	0.02 ^{cD}
	21	37.72	\pm	0.07 ^{dF}	4.99	\pm	0.02 ^{cE}
>50	Control	76.31	\pm	0.10 ^{eA}	1.91	\pm	0.05 ^{aE}
	9	56.92	\pm	0.08 ^{eB}	6.83	\pm	0.03 ^{aA}
	12	49.10	\pm	0.12 ^{eC}	6.86	\pm	0.00 ^{aA}
	15	41.77	\pm	0.06 ^{eD}	5.95	\pm	0.04 ^{cB}
	18	40.08	\pm	0.06 ^{eE}	5.34	\pm	0.00 ^{dC}
	21	37.35	\pm	0.18 ^{eF}	4.97	\pm	0.01 ^{dD}

^{a-e}Means with different superscripts within the same time group are significantly different ($p < 0.05$).

^{A-F}Means with different superscripts within the same size group are significantly different ($p < 0.05$).

¹⁾After cryogenic milling of ginseng samples with different conditions, the samples were classified by their particle sizes. One treatment condition in each size range was then selected, and samples selected by particle size were roasted and pulverized by cryogenic milling.

3.4 WSI

Figure 4 presents the WSI of dried ginseng powder with different roasting times. WSI values tended to increase as the particle size became smaller except for treatments involving roasting at over 18 min. This result was consistent with the findings of a previous study that the WSI value of 30.0 μm -sized ginseng powder was increased by 17% when compared with the 1,000-1,250 μm -sized

ginseng powder [18]. Ultrafine grinding removed the cellulose barrier in the plant material, resulting in a smaller particle size, increased water holding capacity, and ultimately increased WSI [18,34]. In addition to particle size, roasting time also had a significant effect on WSI. WSI values of all treatments except for 18 min and 21 min were higher than those of the control ($p < 0.05$). This result was explained by dextrinization wherein the starch structure was destroyed by high heat [35]. The starch chain split from amylopectin (insoluble starch) into amylose (soluble starch), thus increasing the amount of soluble materials [35]. In addition, another study also reported that glycosidic linkages were broken down at high temperature (approximately 80°C), resulting in the release of glucose, known as soluble dietary fiber [36]. However, in this study, the WSI values were unexpectedly decreased in case of roasting over 18 min. Another study has also reported a similar result, showing that when ginseng was roasted for 30 min at 200°C, the soluble solids content increased until 10 min, but decreased rapidly in case of roasting for over 10 min [15]. The reason of decreased WSI can be attributed to the burning caused by long heat treatment over 18 min (Figure 2).

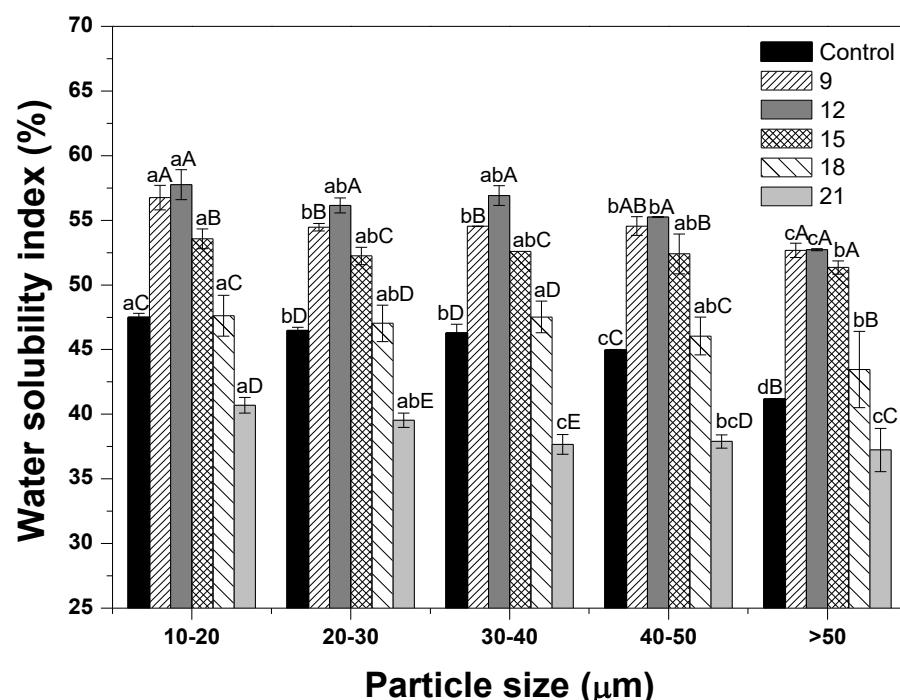


Fig. 4. Water solubility index of dried ginseng powders with different particle sizes according to roasting time. After cryogenic milling of ginseng samples with different conditions, the samples were classified by their particle sizes. One treatment condition in each size range was then selected, and samples selected by particle size were then roasted and pulverized by cryogenic milling. ^{a-d}Means with different superscripts within the same time group are significantly different ($p < 0.05$). ^{A-E}Means with different superscripts within the same size group are significantly different ($p < 0.05$).

3.5 Antioxidant activity

The antioxidant activity of ginseng powder is shown in Figure 5. As the particle size decreased, an increasing pattern of antioxidant activity was observed in control samples alone. This result indicated that the ultrafine grinding increased the antioxidant activity by increasing the specific surface area and capillary effect as the particle size decreased (normal grinding, 69.8%; fine grinding, 70.7%; and ultrafine grinding, 83.8%) [37-39]. For the roasting effect, the antioxidant activity values of all roasted samples showed no trend according to particle size, whereas that of roasted samples was significantly higher than that of the control ($p < 0.05$). The roasting process appeared to increase melanoidins by reacting reducing sugars and nitrogen oxides through the Maillard reaction, leading to an increase in antioxidant activity [17]. This result was also concordant with the findings of a study

by Song *et al.* (2013) showing that roasting treatment (110°C, 30 min) resulted in 2 times higher antioxidant activity than did the unroasted treatment [40]. Further, antioxidant activity tended to increase with an increase in roasting time except for the ginseng powder sample sized 10-20 μm . However, it did not show a significant difference. Seong *et al.* (2018) reported a result consistent with findings of this study and showed an increasing tendency of antioxidant activity as the roasting time increased [41]. However, no significant difference was observed with roasting for 20 and 30 min at 170°C ($p > 0.05$) [41]. Thus, these results were observed in a narrow range of roasting time in this study.

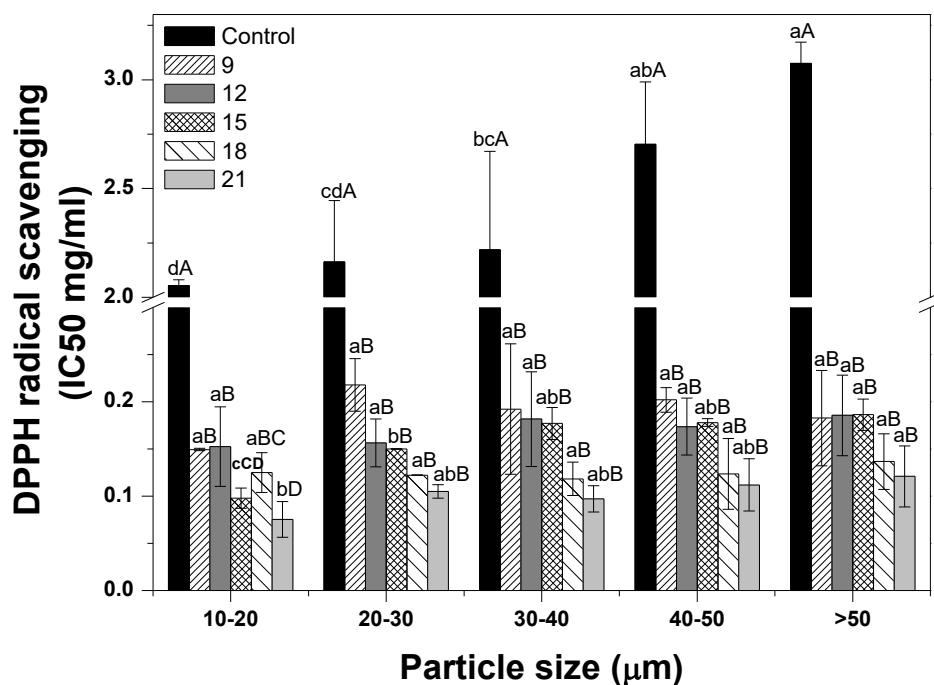


Fig. 5. DPPH radical scavenging activity of dried ginseng powders with different particle sizes according to roasting time. After cryogenic milling of ginseng samples with different conditions, the samples were classified by their particle sizes. One treatment condition in each size range was then selected, and samples selected by particle size were then roasted and pulverized by cryogenic milling.

3.6 TPC

TPC values of ginseng powder were also analyzed (Figure 6). TPC values showed an increasing pattern when compared with control samples. Further, TPC values of roasted ginseng powder with 10–20 μm -size tended to increase when compared with those of >50 μm samples. These results were also in agreement with the findings of a previous study that showed that TPCs displayed an increasing pattern as the particle size decreased in five treatments with various particle sizes (M1, 199.17 μm ; M2, 178.27 μm ; M3, 85.48 μm ; M4, 27.04 μm ; and M5 20.97 μm) [42]. This phenomenon was explained by the particle size-extraction relationship, indicating that phenolic compounds were easily released due to an increase in the specific surface area as the particle size decreased [37]. According to the roasting treatment, all roasted treatments showed significantly higher TPC values than did the control ($p < 0.05$). TPC values tended to increase as the roasting time increased. Roasting produced a Maillard reaction product with phenol structure and increased TPC values [43]. This can also be explained by the fact that the treatment changes phenolic acids to a free form, which is easier to release from the food matrix owing to cell disruption [44]. These results were similar to those of a

previous study showing that the TPC values of corn kernels increased as the roasting time (10, 20, 30, 40, and 50 min) increased at 180°C [45].

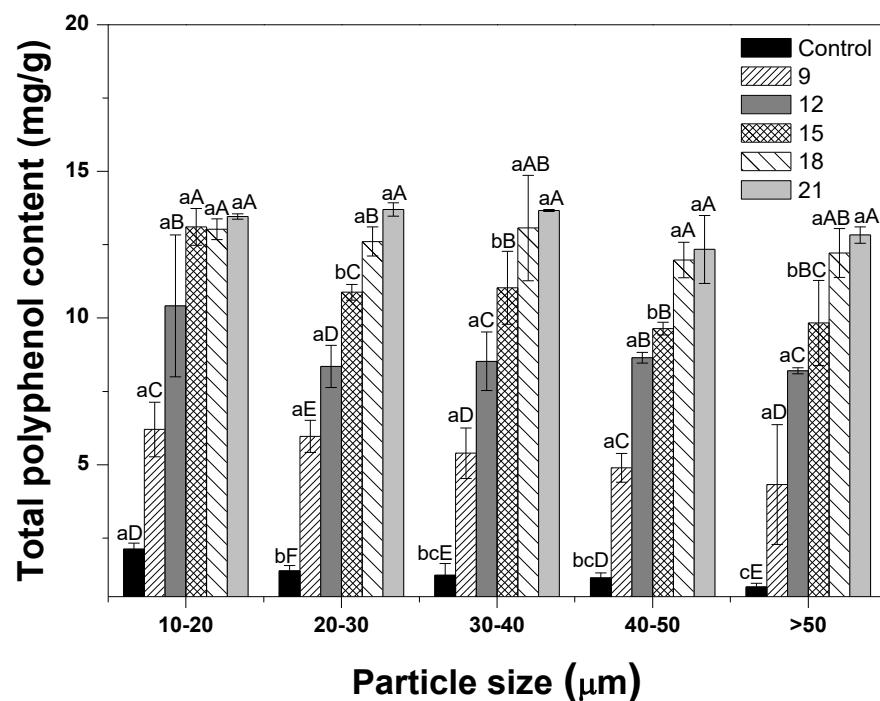


Fig. 6. Total polyphenol content of dried ginseng powders with different particle sizes according to roasting time. After cryogenic milling of ginseng samples with different conditions, the samples were classified by their particle sizes. One treatment condition in each size range was then selected, and samples selected by particle size were then roasted and pulverized by cryogenic milling. ^{a-c}Means with different superscripts within the same time group are significantly different ($p < 0.05$). ^{A-E}Means with different superscripts within the same size group are significantly different ($p < 0.05$).

3.7 Polysaccharide content

Figure 7a and 7b present the total and acidic polysaccharide contents, respectively. The total polysaccharide contents of 10-20 μm -sized treatments revealed significantly higher values than the >50 μm -sized treatments, except for the samples roasted for 21 min ($p < 0.05$). Cho *et al* (2010) also reported that ginseng samples with sizes smaller than 150 μm had higher total polysaccharide yields comparing to samples with higher sizes [46]. However, the acidic polysaccharide contents did not show an increasing pattern as the particle size decreased. These results were also concordant with a previous study showing that acidic polysaccharides of 10-40 mesh treatment (mean diameter: >500 μm) showed a higher value than those of 40-100 mesh treatment (mean diameter: 162.9 μm) [47]. This phenomenon appeared to be caused by water holding capacity [47]. Water penetration and adsorption are facilitated by structural changes (specific surface increase, ginseng starch damage) arising from decreasing particle size [47]. As a result, the flocculated powder due to an increase of the water holding capacity decreased the diffusion of components from the extractive solvent [47]. For comparison of the roasting treatments, the total polysaccharides of samples sized as 10-20 μm showed the highest value when roasting for 12 min, and the other samples with different sizes showed high contents when roasting for 15 min. However, the total polysaccharides of all treatments tended to gradually decrease with roasting for over 18 min (relatively long heat treatment). In another study, when comparing the light (3-3.5 min), medium (4-5 min), and dark (7-8.5 min) roasting treatments, the total polysaccharide contents of Arabica coffee beans was found to be decreased as the roasting time increased [48]. This was because the long roasting process destroyed or chemically altered the monosaccharides [48]. However, acidic polysaccharides in all roasting treatments showed

higher values than did the control (not-roasted) despite the absence of any trend according to roasting time ($p < 0.05$). Roasting was considered to decompose the starch components, and facilitate the extraction of acidic polysaccharides [49]. This result suggested that the reduction of total polysaccharides (acidic and neutral) was mainly caused by neutral polysaccharide decomposition [50]. Acidic polysaccharides are known to have higher biological activities than do neutral polysaccharides in ginseng [50], and thus, roasting can be considered an effective pretreatment.

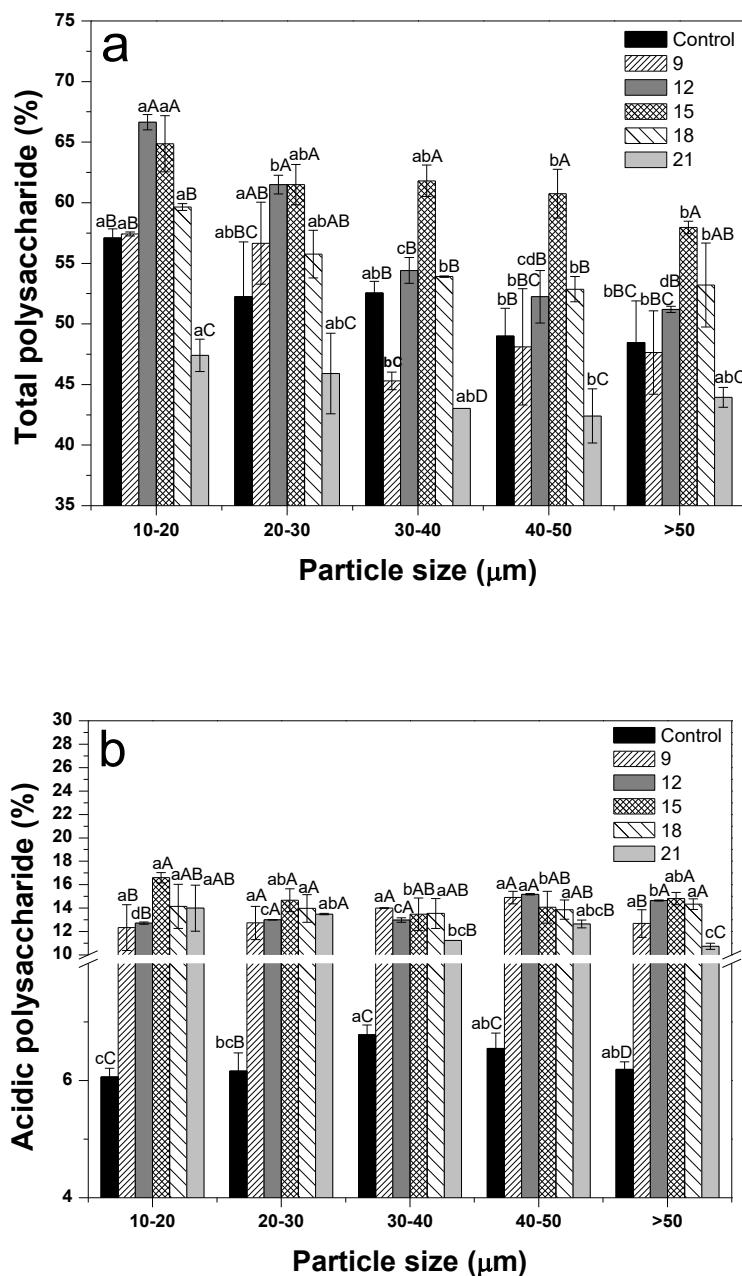


Fig. 7. Polysaccharides of dried ginseng powders with different particle size according to roasting time. After cryogenic milling of ginseng samples with different conditions, the samples were classified by their particle sizes. One treatment condition in each size range was then selected, and samples selected by particle size were then roasted and pulverized by cryogenic milling. ^{a-c}Means with different superscripts within the same time group are significantly different ($p < 0.05$). ^{A-E}Means with different superscripts within the same size group are significantly different ($p < 0.05$).

3.8 Ginsenoside content

Ginsenoside content values showed no tendency of changing according to particle size (Table 4). According to the roasting pretreatment, roasted samples with over 20 μm size had higher contents than the control except for the samples roasted for 15 min. Seong *et al.* (2018) reported that the ginsenoside contents of ginseng samples roasted for 20-30 min at 170°C were higher than those in the control (not roasted) [41]. The ginseng samples with 10-30 μm size revealed the highest ginsenoside contents in case of roasting for 18 min. However, samples with over 30 μm size indicated the highest contents when roasting for 12 min ($p < 0.05$). These results indicated that the roasting time to obtain the optimum ginsenoside content varied according to particle size.

Rg1, Re, and Rb1, known as the main components of ginseng, also indicated a pattern according to roasting time. Rg1 in all size treatments showed the highest contents with roasting for 9 min ($p < 0.05$). Re showed the highest content with roasting for 12 min except for samples with 10-20 μm size. Finally, Rb1 of the 10-30 μm -sized samples showed the highest contents with roasting for 18 min, whereas samples with over 30 μm size and relatively larger size showed the highest contents with roasting for 12 min ($p < 0.05$). Therefore, the ginsenoside content was observed to change with particle size and roasting time.

Table 4. Ginsenosides content of dried ginseng powder with different roasting time conditions

Roasting ¹⁾		Ginsenosides content (mg/g)							
Size (μm)	Time (min)	Rg1	Re	Rf	Rb1	Rc	Rb2	Rb3	Rd
10-20	Control	3.68 \pm 0.03 ^C	3.41 \pm 0.02 ^D	1.46 \pm 0.08 ^B	4.26 \pm 0.03 ^C	1.14 \pm 0.04 ^C	0.97 \pm 0.03 ^C	0.19 \pm 0.02 ^C	0.31 \pm 0.01 ^C
	9	4.62 \pm 0.02 ^A	2.82 \pm 0.01 ^E	1.01 \pm 0.01 ^C	4.18 \pm 0.04 ^C	1.01 \pm 0.03 ^D	0.91 \pm 0.01 ^C	0.15 \pm 0.00 ^C	0.35 \pm 0.00 ^C
	12	3.91 \pm 0.13 ^B	7.56 \pm 0.21 ^B	1.77 \pm 0.00 ^A	8.34 \pm 0.30 ^B	3.35 \pm 0.13 ^B	2.99 \pm 0.11 ^B	0.51 \pm 0.02 ^B	0.91 \pm 0.03 ^B
	15	2.30 \pm 0.14 ^E	1.84 \pm 0.10 ^F	0.73 \pm 0.01 ^D	3.04 \pm 0.03 ^D	1.05 \pm 0.03 ^{CD}	0.92 \pm 0.01 ^C	0.17 \pm 0.01 ^C	0.18 \pm 0.01 ^D
	18	4.00 \pm 0.09 ^B	8.01 \pm 0.17 ^A	1.85 \pm 0.04 ^A	9.88 \pm 0.82 ^A	4.07 \pm 0.10 ^A	3.60 \pm 0.11 ^A	0.66 \pm 0.03 ^A	1.02 \pm 0.05 ^A
	21	3.35 \pm 0.05 ^D	6.88 \pm 0.14 ^C	0.27 \pm 0.08 ^E	ND ^E	ND ^E	ND ^D	ND ^D	0.21 \pm 0.02 ^D
20-30	Time (min)	Rg2(S)	Rg2(R)	Rg3(S)	Rg3(R)	Rh1(S)	Rh2(S)	Sum	
	Control	0.17 \pm 0.01 ^D	0.09 \pm 0.01 ^C	ND ^B	ND ^B	0.10 \pm 0.02 ^B	ND	15.77 \pm 0.02 ^{bC}	
	9	0.15 \pm 0.01 ^D	0.14 \pm 0.00 ^{BC}	ND ^B	ND ^B	0.03 \pm 0.00 ^D	ND	15.39 \pm 0.09 ^{bC}	
	12	0.73 \pm 0.02 ^B	0.48 \pm 0.02 ^A	ND ^B	ND ^B	0.04 \pm 0.00 ^D	ND	30.64 \pm 1.03 ^{cB}	
	15	0.26 \pm 0.04 ^C	0.10 \pm 0.04 ^C	0.19 \pm 0.01 ^A	0.07 \pm 0.03 ^A	0.22 \pm 0.00 ^A	ND	11.06 \pm 0.03 ^{cD}	
	18	0.89 \pm 0.02 ^A	0.34 \pm 0.30 ^{AB}	ND ^B	ND ^B	0.07 \pm 0.00 ^C	ND	34.39 \pm 1.10 ^{aA}	
	21	ND ^E	0.19 \pm 0.04 ^{BC}	ND ^B	ND ^B	ND ^E	ND	10.73 \pm 0.00 ^{dD}	
Size (μm)	Time (min)	Rg1	Re	Rf	Rb1	Rc	Rb2	Rb3	Rd
30	Control	3.28 \pm 0.04 ^B	2.68 \pm 0.02 ^C	0.81 \pm 0.70 ^B	3.60 \pm 0.06 ^D	0.90 \pm 0.02 ^D	0.78 \pm 0.02 ^D	0.19 \pm 0.01 ^C	0.28 \pm 0.02 ^D
	9	4.59 \pm 0.07 ^A	2.37 \pm 0.03 ^{CD}	0.96 \pm 0.03 ^B	4.24 \pm 0.06 ^C	1.12 \pm 0.01 ^C	1.03 \pm 0.01 ^D	0.17 \pm 0.00 ^C	0.38 \pm 0.01 ^C
	12	3.10 \pm 0.05 ^B	8.88 \pm 0.15 ^A	1.65 \pm 0.06 ^A	9.62 \pm 0.15 ^A	4.14 \pm 0.06 ^A	3.64 \pm 0.07 ^A	0.62 \pm 0.01 ^A	1.14 \pm 0.01 ^A
	15	2.17 \pm 0.07 ^C	2.08 \pm 0.08 ^D	0.73 \pm 0.03 ^B	3.08 \pm 0.16 ^E	1.05 \pm 0.00 ^C	0.84 \pm 0.02 ^D	0.14 \pm 0.01 ^C	0.23 \pm 0.02 ^E
	18	2.98 \pm 0.37 ^B	8.71 \pm 0.24 ^A	1.67 \pm 0.17 ^A	9.71 \pm 0.34 ^A	4.09 \pm 0.10 ^A	3.29 \pm 0.41 ^B	0.60 \pm 0.09 ^A	1.11 \pm 0.02 ^A

21	2.11±0.03 ^C	5.47±0.35 ^B	1.60±0.10 ^A	6.49±0.04 ^B	3.14±0.01 ^B	2.88±0.00 ^C	0.51±0.02 ^B	0.87±0.05 ^B		
Size (μm)	Rg2(S)	Rg2(R)	Rg3(S)	Rg3(R)	Rh1(S)	Rh2(S)	Sum			
Control	0.06±0.02 ^E	0.06±0.01 ^E	ND ^C	ND ^B	0.10±0.01 ^C	ND	12.75±0.07 ^{cD}			
9	0.15±0.00 ^D	0.14±0.00 ^D	ND ^C	ND ^B	0.04±0.00 ^E	ND	15.18±0.20 ^{bC}			
12	0.93±0.01 ^B	0.63±0.01 ^A	ND ^C	ND ^B	0.04±0.00 ^E	ND	34.39±0.55 ^{bA}			
15	0.32±0.02 ^C	0.16±0.01 ^D	0.19±0.02 ^B	0.05±0.01 ^A	0.22±0.03 ^B	ND	11.26±0.03 ^{bcE}			
18	0.97±0.02 ^B	0.56±0.07 ^B	ND ^C	ND ^B	0.07±0.01 ^D	ND	34.41±0.97 ^{aA}			
21	1.21±0.08 ^A	0.39±0.03 ^C	0.46±0.03 ^A	ND ^B	0.26±0.02 ^A	ND	25.80±0.15 ^{aB}			
Size (μm)	Time (min)	Rg1	Re	Rf	Rb1	Rc	Rb2	Rb3	Rd	
Control	3.93±0.08 ^B	3.24±0.04 ^D	1.13±0.31 ^C	4.60±0.28 ^D	1.57±0.16 ^D	1.33±0.17 ^C	0.19±0.04 ^D	0.42±0.07 ^E		
9	5.20±0.08 ^A	2.94±0.03 ^E	1.04±0.01 ^C	5.16±0.09 ^C	1.42±0.03 ^D	1.29±0.04 ^C	0.21±0.01 ^D	0.49±0.01 ^D		
12	3.20±0.02 ^C	9.94±0.03 ^A	1.75±0.07 ^A	11.14±0.02 ^A	4.54±0.06 ^A	4.00±0.04 ^A	0.68±0.01 ^B	1.30±0.01 ^A		
15	2.14±0.05 ^F	2.26±0.07 ^F	0.68±0.00 ^D	3.39±0.07 ^E	1.26±0.09 ^E	1.06±0.06 ^D	0.15±0.04 ^E	0.26±0.03 ^F		
18	3.02±0.10 ^D	8.59±0.22 ^B	1.81±0.06 ^A	10.01±0.37 ^B	4.31±0.13 ^B	4.03±0.11 ^A	0.74±0.03 ^A	1.14±0.02 ^B		
21	1.79±0.02 ^E	4.70±0.08 ^C	1.51±0.03 ^B	5.40±0.06 ^C	2.53±0.05 ^C	2.29±0.03 ^B	0.43±0.00 ^C	0.74±0.02 ^C		
30-40	Time (min)	Rg2(S)	Rg2(R)	Rg3(S)	Rg3(R)	Rh1(S)	Rh2(S)	Sum		
Control	0.20±0.08 ^E	0.11±0.12 ^E	ND ^C	ND ^B	0.05±0.09 ^C	ND	16.77±0.10 ^{aE}			
9	0.20±0.00 ^E	0.20±0.00 ^D	ND ^C	ND ^B	0.05±0.00 ^C	ND	18.20±0.25 ^{aD}			
12	1.04±0.01 ^C	0.73±0.01 ^A	ND ^C	ND ^B	0.05±0.00 ^C	ND	38.36±0.21 ^{aA}			
15	0.32±0.01 ^D	0.09±0.00 ^E	0.18±0.00 ^B	0.05±0.00 ^A	0.20±0.04 ^B	ND	12.05±0.03 ^{aF}			
18	1.13±0.04 ^B	0.56±0.01 ^B	0.02±0.01 ^C	ND ^B	0.10±0.01 ^C	ND	35.46±1.01 ^{aB}			
21	1.28±0.03 ^A	0.38±0.01 ^C	0.67±0.01 ^A	ND ^B	0.33±0.01 ^A	ND	22.05±0.31 ^{cC}			
40-50	Size (μm)	Time (min)	Rg1	Re	Rf	Rb1	Rc	Rb2	Rb3	Rd
Control	3.14±0.02 ^C	2.70±0.01 ^E	1.05±0.03 ^C	3.23±0.14 ^E	0.73±0.06 ^E	0.66±0.04 ^F	0.17±0.01 ^C	0.25±0.02 ^E		
9	4.93±0.03 ^A	3.41±0.05 ^D	1.08±0.01 ^C	5.38±0.06 ^D	1.35±0.02 ^C	1.20±0.05 ^D	0.19±0.01 ^C	0.49±0.01 ^D		
12	3.67±0.06 ^B	8.68±0.11 ^A	1.61±0.04 ^B	10.22±0.14 ^A	4.01±0.05 ^A	3.55±0.06 ^A	0.58±0.00 ^A	1.23±0.04 ^A		
15	2.16±0.03 ^F	1.85±0.01 ^F	0.76±0.01 ^D	2.95±0.05 ^F	1.04±0.06 ^D	0.91±0.03 ^E	0.18±0.07 ^C	0.19±0.01 ^F		
18	2.94±0.02 ^D	8.46±0.06 ^B	1.78±0.02 ^A	9.53±0.16 ^B	4.03±0.02 ^A	3.44±0.03 ^B	0.62±0.01 ^A	1.08±0.02 ^B		
21	2.36±0.03 ^E	5.67±0.06 ^C	1.64±0.02 ^B	6.93±0.06 ^C	2.63±0.02 ^B	2.29±0.06 ^C	0.45±0.01 ^B	0.72±0.03 ^C		
	Time (min)	Rg2(S)	Rg2(R)	Rg3(S)	Rg3(R)	Rh1(S)	Rh2(S)	Sum		
Control	0.07±0.01 ^E	0.02±0.01 ^E	ND ^D	ND ^B	0.11±0.01 ^B	ND	12.13±0.03 ^{cE}			
9	0.21±0.00 ^D	0.25±0.00 ^C	NDD	ND ^B	0.05±0.00 ^C	ND	18.52±0.22 ^{aD}			
12	0.86±0.01 ^C	0.67±0.01 ^A	NDD	ND ^B	0.05±0.00 ^C	ND	35.14±0.50 ^{bA}			

15	0.27±0.08 ^D	0.16±0.04 ^D	0.19±0.02 ^B	0.05±0.05 ^A	0.23±0.06 ^A	ND	10.93±0.04 ^{eF}		
18	1.12±0.03 ^B	0.65±0.01 ^A	0.03±0.00 ^C	ND ^B	0.12±0.00 ^B	ND	33.73±0.42 ^{aB}		
21	1.19±0.01 ^A	0.47±0.01 ^B	0.38±0.01 ^A	ND ^B	0.25±0.00 ^A	ND	24.99±0.28 ^{bC}		
Size (μm)									
	Time (min)	Rg1	Re	Rf	Rb1	Rc	Rb2	Rb3	Rd
Control		3.81±0.10 ^B	3.27±0.05 ^D	1.50±0.05 ^B	4.20±0.11 ^E	1.10±0.05 ^E	0.94±0.05 ^E	0.20±0.03 ^C	0.29±0.03 ^E
9		4.47±0.16 ^A	3.53±0.08 ^D	1.13±0.04 ^C	5.33±0.19 ^D	1.53±0.06 ^D	1.36±0.05 ^D	0.24±0.01 ^C	0.47±0.01 ^D
12		2.91±0.07 ^C	8.32±0.32 ^A	1.52±0.03 ^B	9.65±0.33 ^A	4.15±0.18 ^A	3.77±0.15 ^A	0.62±0.03 ^A	1.18±0.07 ^A
15		2.22±0.01 ^E	1.98±0.05 ^E	0.81±0.00 ^D	3.14±0.03 ^F	1.14±0.07 ^E	0.96±0.03 ^E	0.20±0.06 ^C	0.27±0.03 ^E
18		2.74±0.03 ^D	7.55±0.09 ^B	1.66±0.02 ^A	8.91±0.05 ^B	3.73±0.02 ^B	3.27±0.03 ^B	0.62±0.01 ^A	0.97±0.02 ^B
21		1.96±0.06 ^F	4.76±0.22 ^C	1.50±0.03 ^B	5.86±0.17 ^C	2.48±0.06 ^C	2.22±0.05 ^C	0.42±0.00 ^B	0.72±0.02 ^C
>50	Time (min)	Rg2(S)	Rg2(R)	Rg3(S)	Rg3(R)	Rh1(S)	Rh2(S)	Sum	
	Control	0.16±0.03 ^E	0.09±0.01 ^F	ND ^D	ND ^D	0.11±0.02 ^C	ND	15.66±0.04 ^{bE}	
	9	0.29±0.02 ^D	0.25±0.01 ^D	ND ^D	ND ^D	0.05±0.00 ^D	ND	18.66±0.62 ^{aD}	
	12	0.91±0.03 ^C	0.55±0.01 ^B	0.01±0.02 ^D	ND ^D	0.04±0.00 ^D	ND	33.64±1.22 ^{bA}	
	15	0.33±0.04 ^D	0.17±0.02 ^E	0.26±0.02 ^B	0.12±0.00 ^A	0.21±0.00 ^B	ND	11.82±0.02 ^{abF}	
	18	1.09±0.00 ^B	0.57±0.00 ^A	0.06±0.00 ^C	0.06±0.00 ^B	0.11±0.00 ^C	ND	31.36±0.22 ^{bb}	
	21	1.21±0.03 ^A	0.40±0.01 ^C	0.50±0.01 ^A	0.03±0.02 ^C	0.27±0.02 ^A	ND	22.32±0.61 ^c	

^{a-c}Means with different superscripts within the same time group are significantly different ($p < 0.05$).

^{A-F}Means with different superscripts within the same size group are significantly different ($p < 0.05$).

¹⁾After cryogenic milling of ginseng samples with different conditions, the samples were classified by their particle sizes. One treatment condition in each size range was then selected, and samples selected by particle size were roasted and pulverized by cryogenic milling.

4. Conclusions

The purpose of this study was to decrease the particle size of ginseng by roasting and cryogenic milling and eventually increase the water solubility and physiological activity of ginseng.

Ginseng samples were roasted for different times of 9-21 min and ground to different sizes (10-50, and >50 μm) by cryogenic milling. All roasted samples indicated significantly smaller particle sizes compared with the control (not roasted) based on D [3,2] ($p < 0.05$). The particle sizes of the roasted samples showed a decreasing pattern until 15 min of roasting. However, with roasting for over 18 min, the particle size of the ginseng powder was increased due to flocculation. WSI values indicated an increasing pattern as the particle size decreased except for samples roasted at over 18 min. As for the results of antioxidant activity and TPC, samples with 10-20 μm size demonstrated higher values than those for over 50 μm -sized samples. The total polysaccharide contents of 10-20 μm -sized samples revealed significantly higher values than for over 50 μm -sized samples with an exception of samples roasted for 21 min ($p < 0.05$). However, acidic polysaccharides did not show any trend according to particle size. Ginsenoside contents showed no tendency according to particle size. Based on roasting time, the WSI showed higher values up to 15 min of roasting compared to the control (not roasted) ($p < 0.05$). Antioxidant activity and TPC also showed an increasing trend with increasing roasting time. Total polysaccharides showed the highest values when the samples were roasted for 15 min except for the particles with 10-20 μm size. Acidic polysaccharide values showed no tendency depending on roasting time, but showed significantly higher values than did the control

(not roasted). Ginsenoside did not show any pattern depending on roasting time, but the ginsenoside contents of roasted samples with over 20 μm size were higher than the control except in samples roasted for 15 min. This study proves that roasting and cryogenic milling for certain times have a positive effect such as reduction of particle size, increased WSI, and increased physiological activity in the manufacture of ginseng powder.

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