

Article

Tests to Ensure the Minimum Methane Concentration for Gas Engines to Limit Atmospheric Emissions

Marek Borowski ^{1,*}, Piotr Życzkowski ², Rafał Łuczak ³, Michał Karch ⁴ and Jianwei Cheng ⁵

¹ AGH University of Science and Technology, Kraków, Poland; borowski@agh.edu.pl

² AGH University of Science and Technology, Kraków, Poland; piotr.zyczkowski@agh.edu.pl

³ AGH University of Science and Technology, Kraków, Poland; rluczak@agh.edu.pl

⁴ AGH University of Science and Technology, Kraków, Poland; karch@agh.edu.pl

⁵ China University of Mining & Technology (CUMT), Xuzhou, China; Cheng.Jianwei@cumt.edu.cn

* Correspondence: borowski@agh.edu.pl; Tel.: +48-12-6172068 (M.B.)

Abstract: During the extraction of hard coal in Polish conditions, methane is emitted, which is referred to as mine gas. As a result of the desorption of methane, a greenhouse gas is released from coal seams. In order to reduce atmospheric emissions, methane from coal seams is captured by a methane drainage system. On the other hand, methane, which has been separated into underground mining excavations, is discharged into the atmosphere with a stream of ventilation air. For many years, Polish hard coal mines have been capturing methane to ensure the safety of the crew and the continuity of mining operations. As a greenhouse gas, methane has a significant potential, as it is more effective at absorbing and re-emitting radiation than carbon dioxide. The increase in the amount of methane in the atmosphere is a significant factor influencing global warming, however, it is not as strong as the increase in carbon dioxide. Therefore, in Polish mines, the methane-air mixture captured in the methane drainage system is not emitted to the atmosphere, but burned as fuel in systems, including cogeneration systems, to generate electricity, heat and cold. However, in order for such use to be possible, the methane-air mixture must meet appropriate quality and quantity requirements. The article presents an analysis of changes in selected parameters of the captured methane-air mixture from one of the hard coal mines in the Upper Silesian Coal Basin in Poland. The paper analyses the changes in concentration and size of the captured methane stream through the methane capturing system. The gas captured by the methane drainage system, as an energy source, can be used in cogeneration, when the methane concentration is greater than 40%. Considering the variability of CH₄ concentration in the captured mixture, it was also indicated which pure methane stream must be added to the gas mixture in order for this gas to be used as a fuel for gas engines. The balance of power of produced electric energy in gas engines is presented. Possible solutions ensuring constant concentration of the captured methane-air mixture are also presented.

Keywords: methane emissions; coal seam mining; gas engines; greenhouse gas

1. Introduction

Currently, methane is estimated to have a greenhouse gas potential 28–34 times higher than carbon dioxide [1]. Hard coal mining is responsible for 8% of global anthropogenic methane emissions [2]. It is estimated that methane emissions from mining plants, converted into CO₂, total more than 1 million tonnes of CO₂ per year [3]. The use of methane from coal seams not only reduces its emissions to the atmosphere, but can also generate economic profit. More and more precise methods of estimating methane emissions to the atmosphere are being developed, both from operating mines [4–6] and from closed mines [7]. The vast majority of methane emissions to the atmosphere in operating mines comes from ventilation air (Ventilation Air Methane - VAM) and is

estimated to be around 70% [8–12]. Although concentrations of methane in ventilation air are insignificant and usually range from 0 to 0.7% and most commonly amount to 0.3% [8, 13–15], due to their large air streams they are of great importance in the context of emissions. Most VAM technologies require a methane concentration of 0.5% as the minimum concentration for stable operation [9]. Solutions that operate at a concentration of 0.25% (TFRR - Thermal Flow Reversal Reactor) are already in use, while effective and economically justified operation requires concentrations higher than 0.6% and the recommended concentration is 1%. Hence, a mixture of methane from draining with VAM is used to increase the concentration of methane [16]. Many papers describe the solutions used to utilize methane from ventilation air, such as [3, 17–19].

However, much greater possibilities exist for the use of methane from methane draining. Mining methane with a concentration in the range of 30 to 100% can be used as [3]: fuel in blast furnaces, heating furnaces and boilers and industrial burners, in engines or turbines for power generation. It can also be injected into cold gas pipelines, for the production of fertilizers or as fuel for vehicles. Analyses show that the use of gas engines is more cost-effective than injecting gas into pipelines [20]. This results in high demands as to the purity of methane and therefore high purification costs before it is injected into the pipelines. Detailed methods of methane use from both methane drainage and ventilation air were presented in the paper [21]. In Polish hard coal mines, for many years there has been a gradual development of underground methane drainage and economic use of the captured methane in power and heat installations. Regardless of the method used, the concentration of methane is important, but also the continuity of supply, the quality of the mixture and fluctuations in concentration. The calorific value of a mixture should not vary by more than 10% when used in gas turbines [10]. For proper operation, gas engines require a minimum methane concentration of 25–30%, while gas turbines require a minimum concentration of 40% [22]. Some authors stipulate higher minimum concentration values for gas engines - 35% [23]. In general, a methane concentration of at least 30% is required. In recent years, however, combustion engines have started to appear on the market, which may use mine gas with a methane concentration below 30% [3]. Under Polish conditions, the utilization of mine methane with the use of gas engines is developing particularly intensively. Currently, there are more than 40 installations based on gas engines [24]. Regardless of the manner in which mine methane from methane drainage is utilised, when selecting the power plant size, the variability of inflow and purity of gas associated with routine coal mining should be taken into account, and if necessary, the standards of methane drainage should be raised to ensure that the quality of the gas is safe and compliant with relevant quality regulations. The quality and stream of the extracted methane-air mixture can be increased in various ways, e.g. by intensifying the methane capture in mining operations [24].

2. Measurement methodology and measuring system

The combined energy and cooling system in the analysed mine is used mainly for the purposes of central air-conditioning of the hard coal mine located in the Upper Silesian Coal Basin in southern Poland. It uses a comprehensive combination of electricity, heat and cold as well as compressed air production using methane obtained from methane drainage. A simplified diagram of methane use together with marked measurement points of operating parameters (pressure, flowrate, temperature, moisture and quality of methane-air mixture) is presented in Figure 1.

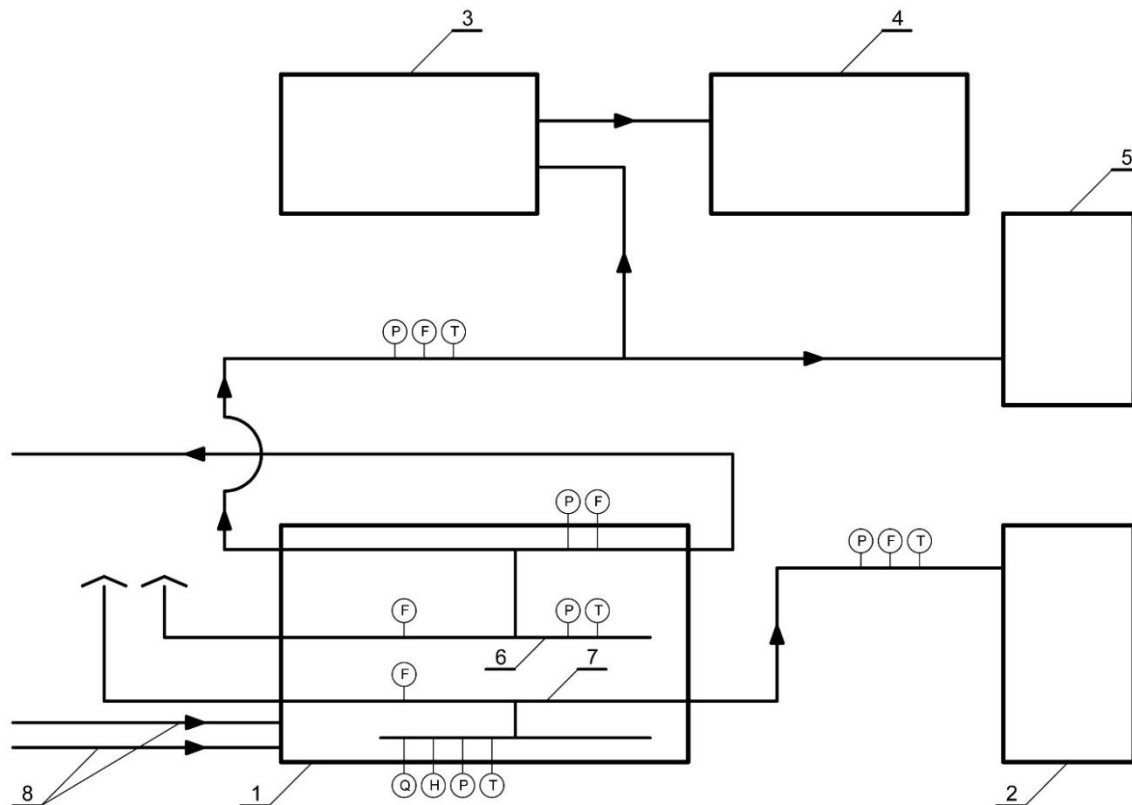


Figure 1. Flow chart of methane utilization: 1 - methane drainage station, 2 - compressor hall, 3 - reduction station, 4 - heat and power plant, 5 - gas engines, 6 - pressure collector II stage, 7 - pressure collector I stage, 8 - pipeline with methane-air mixture from hard coal mine, F - gas flowrate measurement, H - gas humidity measurement, P - gas pressure measurement, Q - gas quality measurement, T - gas temperature measurement.

The measurement of gas flowrate, temperature and pressure is performed before each of the recipients of the mixture. In addition, the quality of gas (composition) and its moisture content are measured at the methane drainage station. The composition of the mixture is determined by the ABB NGC8203+ chromatograph, which detects the following components: methane, propane, ethane, CO, CO₂, N₂, O₂, H₂S and H₂. Information on the exact composition of the mixture is essential for the proper operation of gas engines, for which methane concentration of approximately 40% is required. The turbine flow meter CGT-02 G4000 PN16 with a measuring range of 320÷6500 m³/h and a measuring accuracy of 1% is used to measure the flowrate of gas transmitted to gas engines. Temperature and moisture measurement is provided by a temperature and moisture converter for EE30EX explosion hazard zone with relative moisture accuracy of ±1% RH (0÷90% RH) and ±2% RH (90÷100% RH) and by a temperature sensor Pt1000 class A. On the other hand, the pressure of the methane-air mixture is determined using a pressure converter PC-28 with a measuring range of 4 kPa÷60 MPa.

The system of using methane derived from methane drainage is based on 4 gas engines (2 x TBG 632 V16, 2 x TCG 2032 V16) with a total electrical power of 14.3 MWe. Figure 2 shows the gas engines used [25, 26].

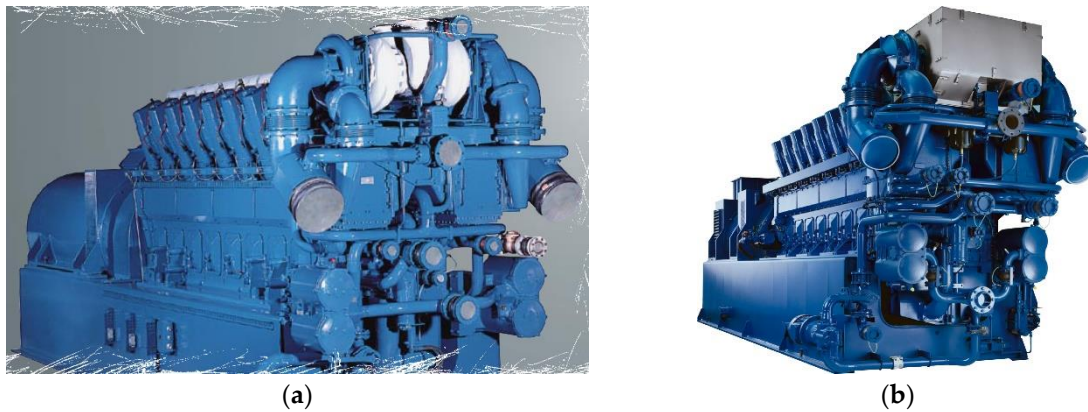


Figure 2. Gas engine: (a) TBG 632 V16; (b) TCG 2032 V16 [25, 26].

3. Analysis of the operating parameters of gas engines in the context of methane capture

Due to the specific nature of mining operations, there is a variability in the amount of captured methane and its concentration, which directly affects the operation of gas engines. Figure 3 shows the monthly operating parameters of the system while maintaining a sufficiently high methane concentration. Methane concentration in the captured methane-air mixture and electricity production in March is shown in Figure 3.

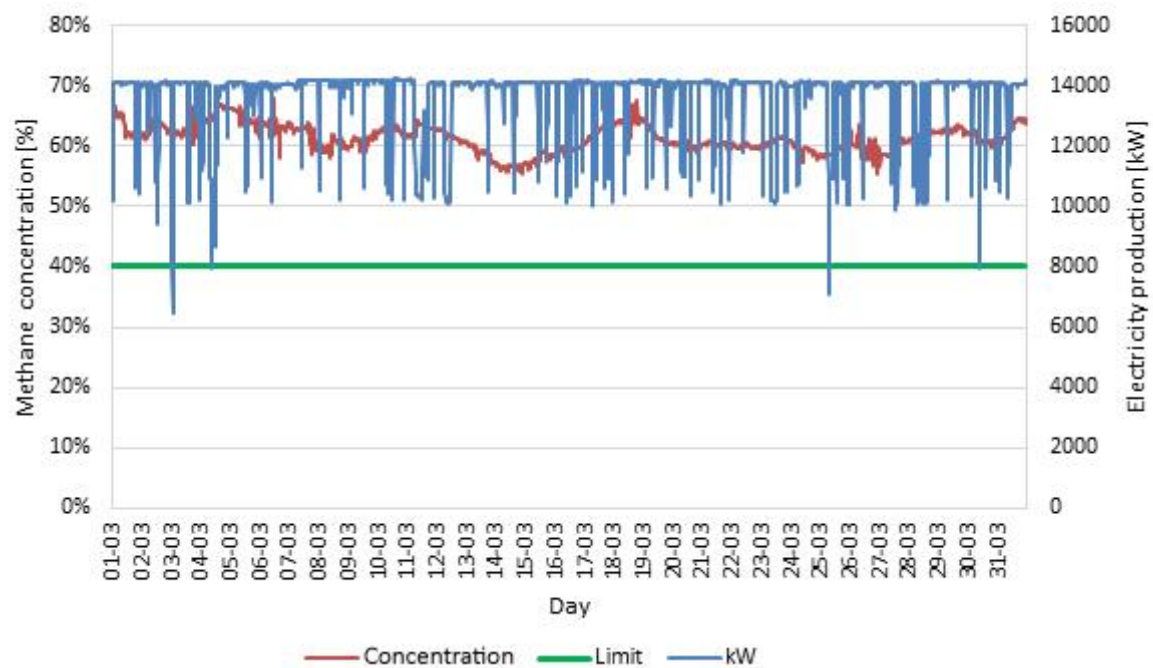


Figure 3. Methane concentration and electricity production in March.

Figure 4 shows the variation of methane concentration and flowrate for the same month. The figure also shows the possible natural gas flowrate required to achieve the required mixture concentration of 40%.

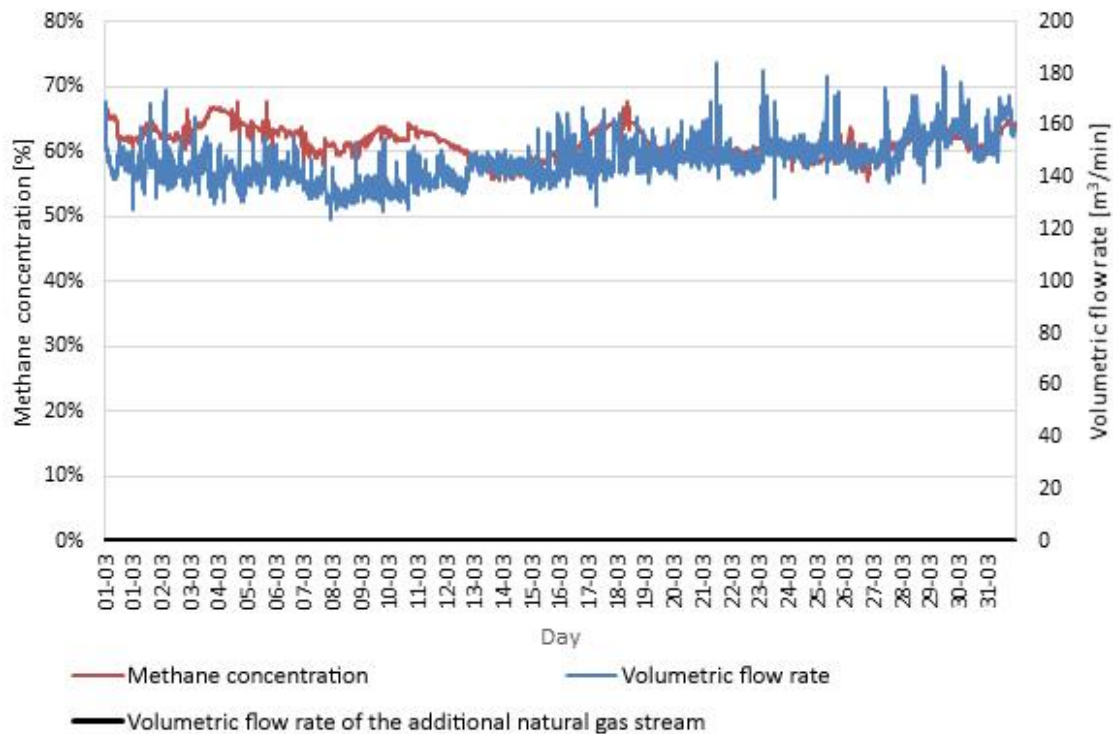


Figure 4. Methane concentration and flowrate in March.

In the analysed month, both the concentration of captured methane and its flowrate were at a sufficiently high level, which ensured stable operation of gas engines and electricity production of up to 14 MW. Similar results were obtained in the following month - April. Figure 5 shows the concentration and production of electricity, while Figure 6 shows the concentration and flowrate of the captured methane.

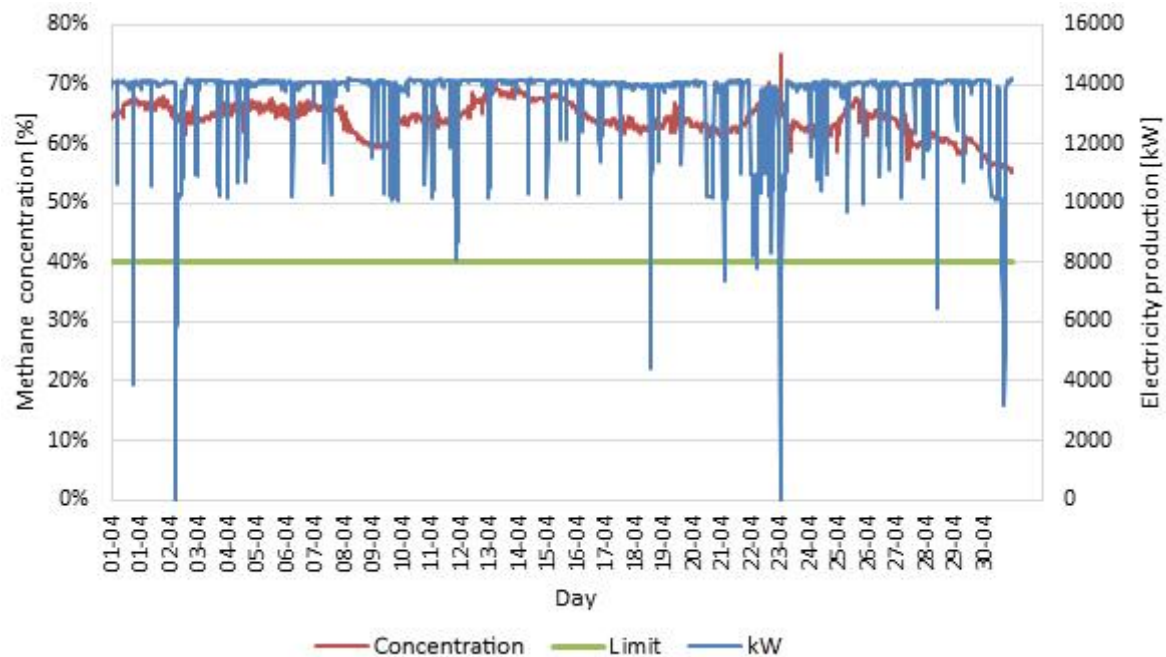


Figure 5. Methane concentration and electricity production in April.

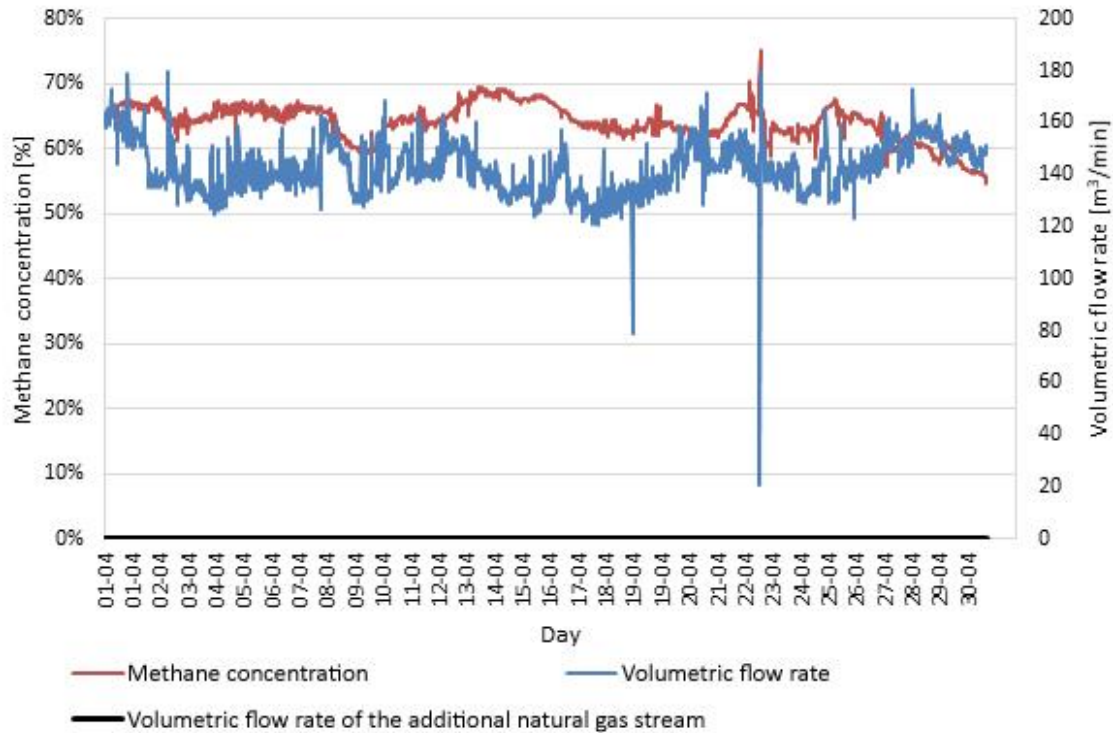


Figure 6. Methane concentration and flowrate in April.

The first problems with the preservation of methane concentration appeared in September - Fig. 7. Most importantly, a decrease in concentration can be observed during the whole month, first to a value of about 50% and from the middle of the month close to the limit value of 40%.

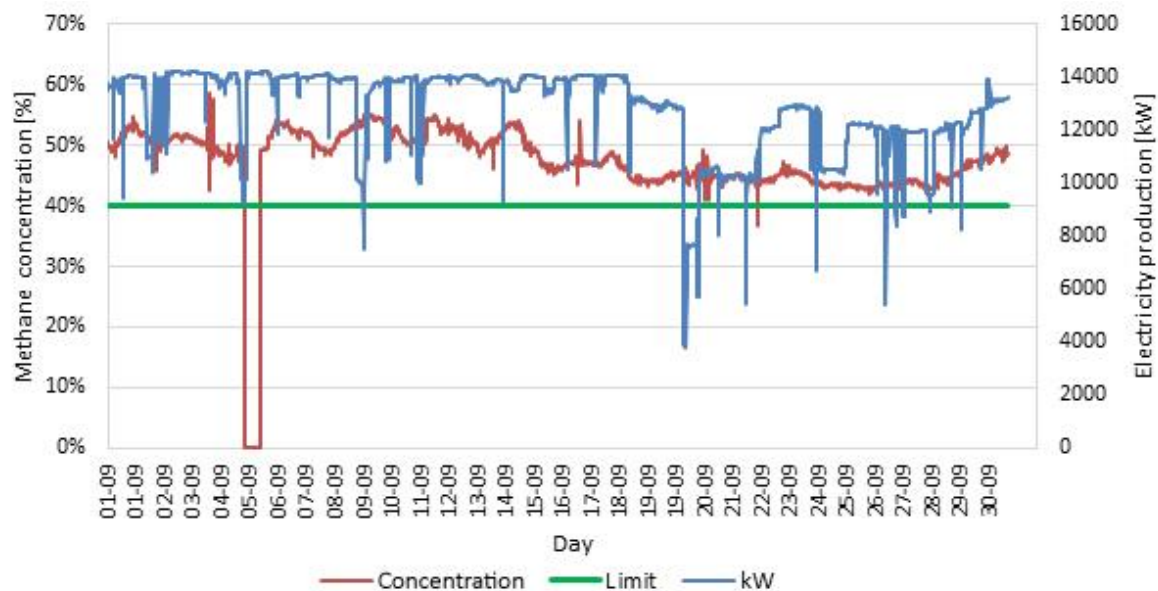


Figure 7. Methane concentration and electricity production in September.

Particularly large fluctuations can be observed in the stream of the captured gas - Fig. 8. Such fluctuations have a negative impact on electricity production and can cause the failure of gas engines.

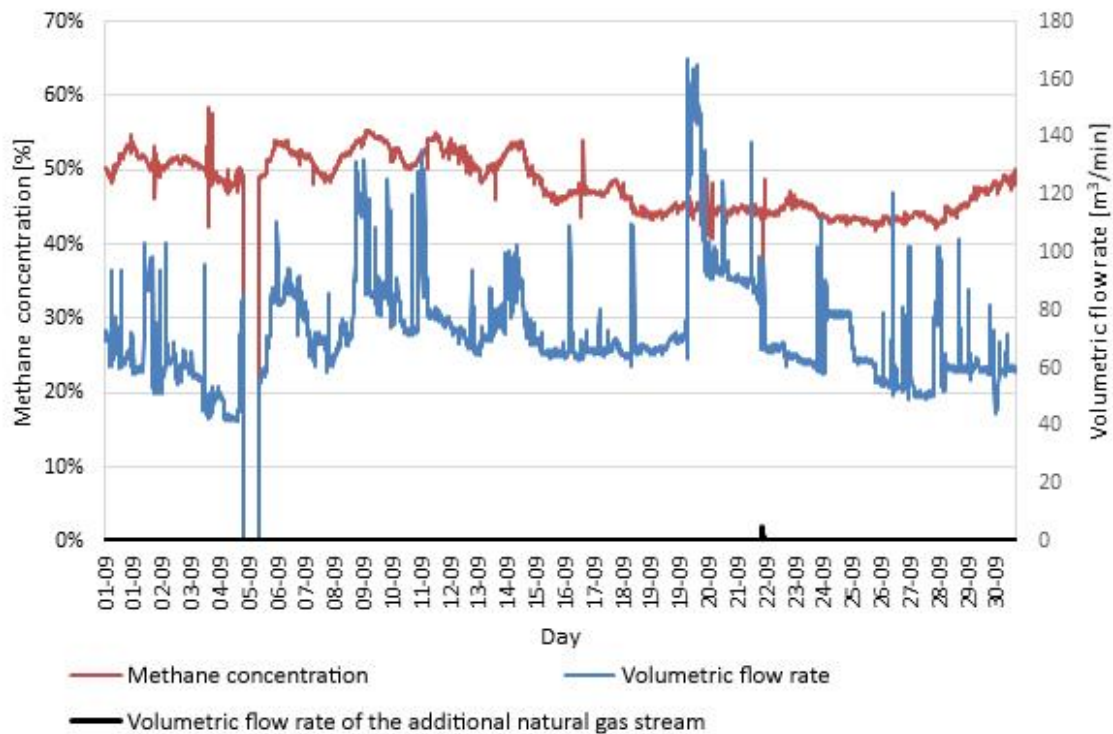


Figure 8. Methane concentration and flowrate in September.

The most unfavourable parameters of the system operation were observed in December - Fig. 9. Methane concentration was much lower than 40%, which significantly reduced the volume of electricity production.

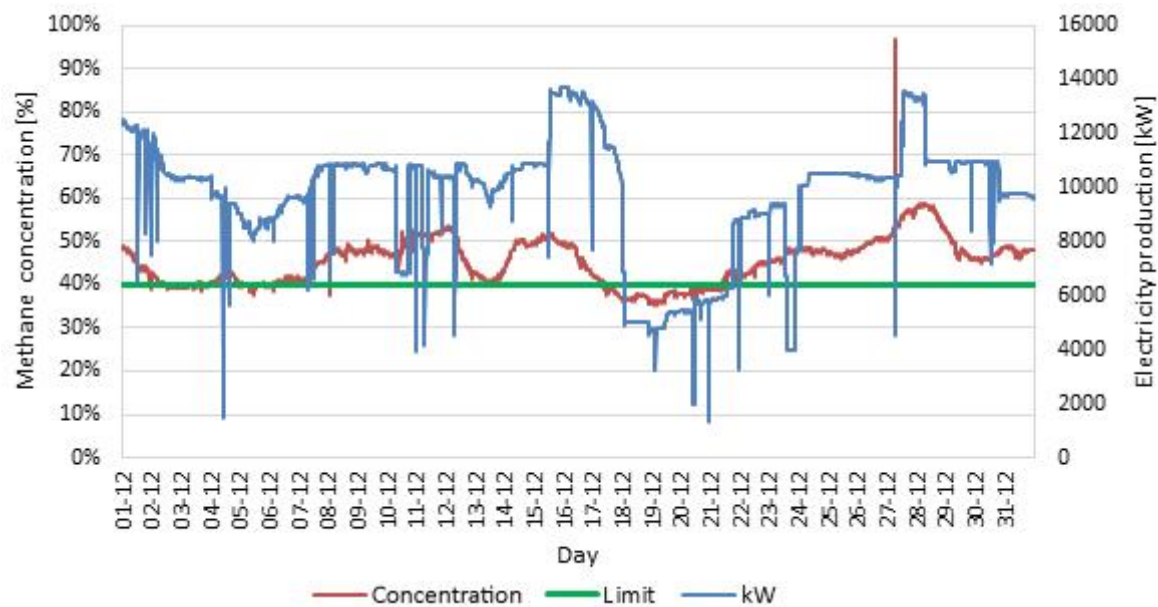


Figure 9. Methane concentration and electricity production in December.

Low concentrations and fluctuations in the gas stream cause unstable operation of gas engines. In order to obtain the required concentration of 40%, an additional natural gas stream can be used - Figure 10. The volume of the additional natural gas stream is shown in Figure 11.

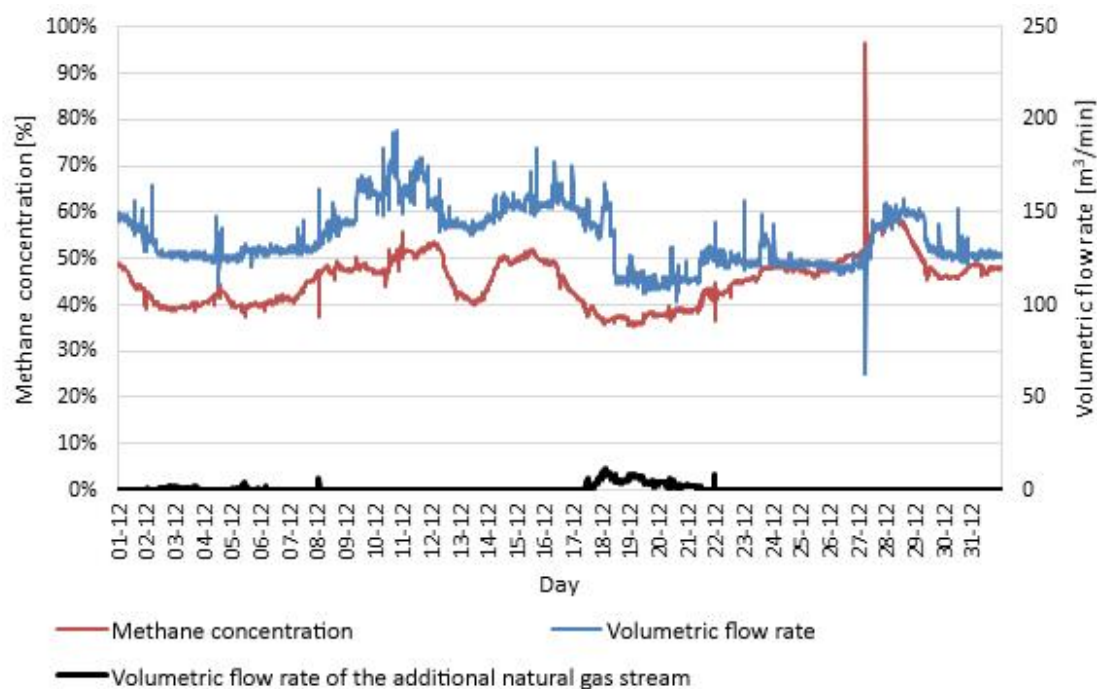


Figure 10. Methane concentration and flowrate in December.

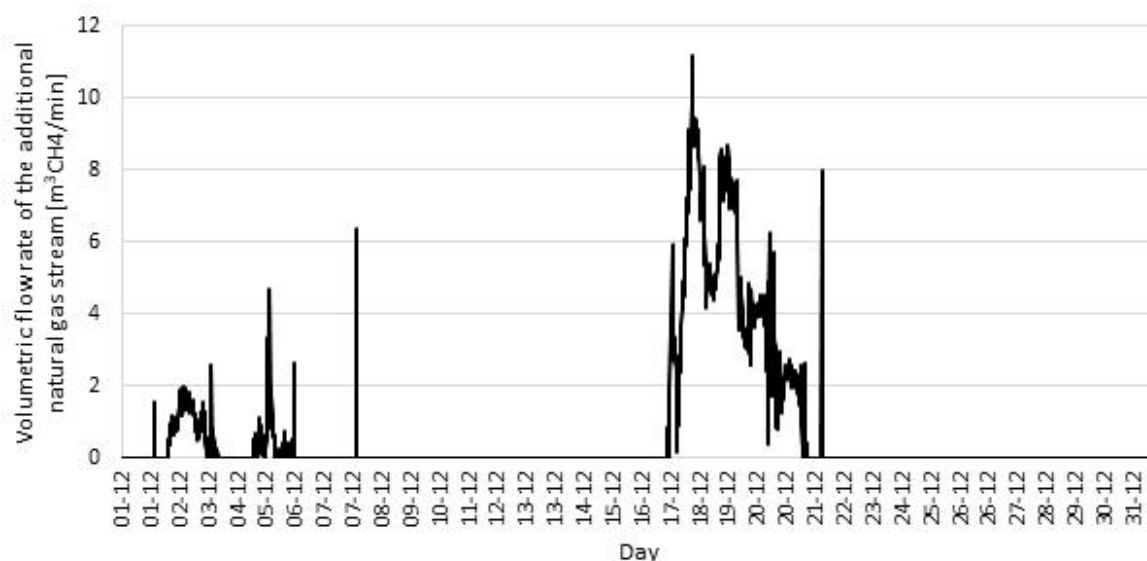


Figure 11. The flowrate of natural gas required to supplement the gas from methane drainage to reach a concentration of 40% in December.

Analysing the operation of gas engines throughout all year, it was found that the natural gas stream needed to stabilise the methane concentration would amount to a maximum of 16 $\text{m}^3\text{CH}_4/\text{min}$. Fluctuations in the concentration and the stream of gas in September and subsequent months resulted from mining operations. Its share in the total methane capture of the mine gradually decreased from 29% in October, through 26% in November to 0% in December. In order to maintain stable operation, it is necessary for the methane drainage system to capture at least 50 $\text{m}^3\text{CH}_4/\text{min}$ with a minimum concentration of 50%. It is not possible to supplement the stream by capturing methane before and after exploitation because these are too small amounts of methane streams to be obtained. It is therefore necessary to consider other methods of stabilising the concentration and stream of the captured methane-air mixture.

4. Analysis of the possibility of obtaining a higher methane concentration for gas engines

Stabilisation of the parameters of the methane-air mixture can be achieved by means of a number of methods. However, in order to avoid high costs of such processes, it is necessary to focus on improving the standards of methane capture from the rock mass. The basic methods of improving the quality of gas include:

- Purifying mine gas
- Mixing mine gas with CNG gas
- Mixing mine gas with natural gas from a low-pressure ground tank
- Capturing and using an additional gas stream of high methane concentration from tanks from already excavated areas

4.1. Purifying mine gas

During the purification of the gas from the methane drainage process, nitrogen, oxygen, carbon dioxide, carbon monoxide or hydrogen sulphide are removed from its composition. The basic process used for this purpose is pressure swing adsorption (PSA). A typical solution is to produce gas with a methane concentration of approximately 95% [27]. Various configurations of the PSA cycle have been developed, usually consisting of multi-stage processes. The number of components present in the methane drainage mixture requires a combination of processes to obtain a high concentration of methane. Pressure swing adsorption has been described in detail in many studies, mainly in the field of chemistry, often in relation to biogas quality improvement - e.g. [28–30]. One of the modifications of the pressure swing adsorption is the VPSA method - Vacuum Pressure Swing Adsorption, which consists of several stages: filling the sorption column with an methane-air mixture, adsorption of components, washing out the non-absorbed elements with an enriched product, two-stage, low pressure desorption of methane. A diagram of such an installation is shown in Figure 12.

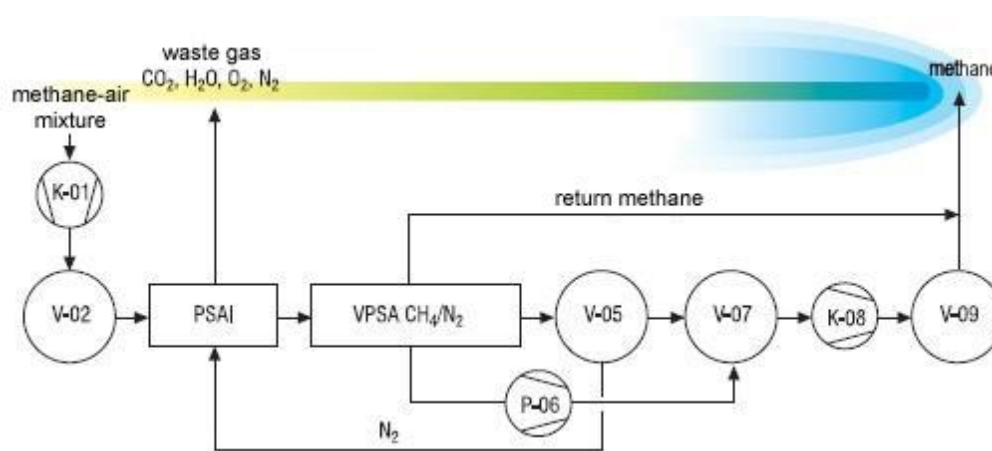


Figure 12. Diagram of the methane-air mixture enrichment installation: K-01 - gas compressor, V-02 - buffer, V-07 - CH₄ buffer, PSA I - purification PSA module, VPSA CH₄/N₂ - methane enrichment PSA module, V-05 - N₂ buffer, P-06 - vacuum pump, K-08 - CH₄ compressor, V-09 - CH₄ buffer [31].

The conducted studies allowed to obtain gas enriched with 96–98% methane from the air-methane mixture with methane content of 55.2% [31]. The PSA method is still being improved by making various modifications, such as using new sorbents or combining several methods of gas purification into one system. Such combination seems to achieve the best results. Osaka Gas combined the PSA technology and membrane separation, which achieved a methane concentration of more than 99% for a biogas [32]. A simplified diagram of such an installation is shown in Figure 13.

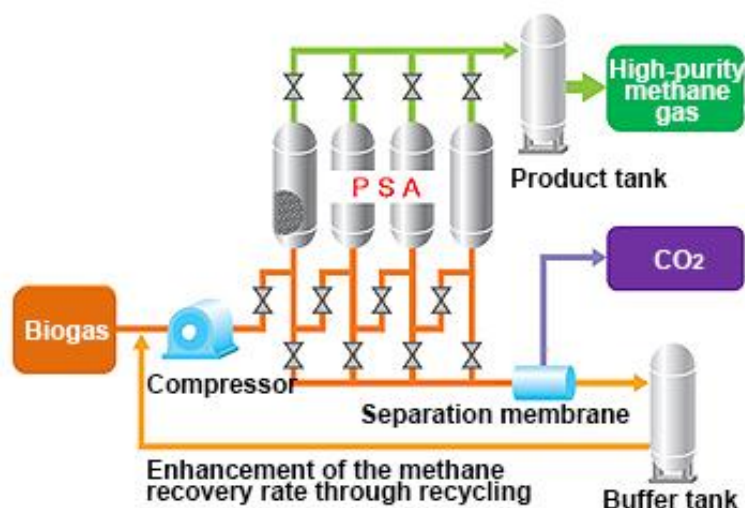


Figure 13. Diagram of a hybrid biogas purifying installation with the use of PSA technology and membrane separation [33].

Other methods used to purify gas from methane drainage are [27]:

- Molecular Sieve Adsorption - a PSA variant with a regulated molecular sieve
- Cryogenic separation - uses multiple heat exchangers to liquefy the high-pressure supply gas stream. It is characterised by a high degree of purification (up to 98%)

Methods of purification are characterised by very high efficiency and are already used in many branches of industry, also for the purification of gas from mine methane draining. However, due to a very high investment cost, the use of such a solution of gas purification for gas engines is in this case not economically justified.

4.2. Figures, Tables and Schemes

The maintenance of the required methane concentration can be achieved by using the concentrated natural gas (CNG) system to mix CNG with the methane-drained gas. The CNG would be added to obtain a mixture containing at least 50% of methane. Analyses indicate that such a process would be carried out in the period from a few to a dozen or so weeks during the year. Compression of the CNG requires the use of suitable devices to draw heat from the combustion of a part of the compressed gas or from the air. The location of the mixing station and gas tanks is very important. The best place to locate a station for mixing the CNG with the methane drainage gas is the vicinity of a station, where the capture of methane from the rock mass occurs. Before approving the assumed location on the premises of the mining plant, the categories and the scope of explosion hazard zones should be determined in accordance with the standard: PN-EN 1127-1:2001 "Explosive atmospheres. Explosion prevention and protection" and classify hazardous areas in accordance with the standard: PN-EN 60079-10:2003 "Electrical equipment in potentially explosive atmospheres. Classification of hazardous areas". Consideration should be given to existing land development and explosion hazard zones.

The applied solution essentially secures the maintenance of the required methane concentration for the undisturbed operation of gas engines and makes itself independent from the obtained concentrations and gas streams from the underground capture in hard coal mines. In order to eliminate commercial gas production, consideration should be given to the supply of gas from the mine's available intakes, which should be preceded by an appropriate analysis.

4.3. Mixing mine gas with natural gas from a low-pressure ground tank

In order to become independent from changes in atmospheric pressure, it is proposed to build a permanent natural gas tank. The gas from the methane tank would be extracted in cases where the

concentration falls below the required level. The system will automatically add gas to the methane drainage mixture, in order to correct the levels (concentrations) of methane. A 1500 m³ tank located close to the methane drainage station will be sufficient. As a low-pressure tank, it does not require any special equipment to expand the gas. Such solutions are widely used in biogas installations as well as in mines [34]. If such a solution was applied, the required works would include:

- Construction of reinforced concrete foundations for the tank
- Installation of a two-layer biogas tank with a capacity of 1500 m³
- Construction of technological, electrical, control and measurement and automation installations as well as systems of dewatering biogas pipelines
- Construction of a container biogas distribution and measurement station on the reinforced concrete foundations
- Performance of leakage tests and final tests

The application of such a solution is much cheaper than the purification technology. Additionally, the tank can store methane from methane drainage in periods of increased capture and use it to make up for gas shortages in periods of reduced capture, or in case of methane concentration drop. It eliminates the costs associated with the purchase of natural gas.

4.4. Using the capture of an additional gas stream of high methane concentration from already excavated areas

To ensure a stable capture of the methane-air mixture, it is also possible to use the capture of methane from excavated coal seams. The idea of this solution is shown in Figure 14. The proposed tank would have a capacity of approximately 3 million m³ and would be used to store the surplus of methane from methane drainage. The system requires the use of an automatic regulation system allowing to correct the levels (concentrations) of gas streams captured at the methane drainage station.

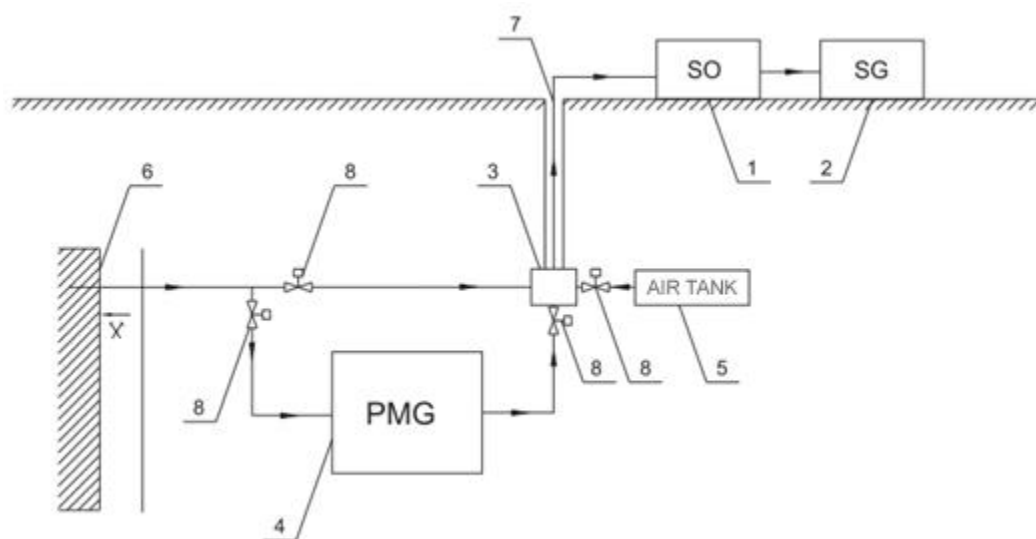


Figure 14. Diagram of technology for the stabilisation of parameters of gas captured from methane drainage with the use of underground tanks: 1 - methane drainage station, 2 - gas engines, 3 - mixer, 4 - methane tank, 5 - air tank, 6 - operation area, 7 - pipeline, 8 - valves with PMG sensors - underground gas tank, SO - methane capture station, SG - gas engine station [35]

From a theoretical point of view, such a system is feasible to implement. The main problem is to ensure the tightness of the rock mass acting as a tank. Since all underground sources of methane capture are a system of 'connected vessels', it is doubtful whether the capture is reliable in terms of methane concentration and stream. The use of such tanks could only be carried out periodically. Control of methane inflow from tanks with high methane concentrations will not allow to maintain

the required amount and concentration in the gas fed to gas engines, which is caused by difficulties in regulating the methane drainage network.

5. Conclusions

Methane, as a greenhouse gas, contributes to global warming. Methane emissions from fossil fuel extraction appear to be still underestimated and could be up to 50% higher than previously thought. It is estimated to be between 120 and 150 million tonnes more, representing 20–25% of total methane emissions from all sources (natural and man-made). Therefore, reducing methane emissions from anthropogenic sources such as mineral extraction and improving the quality of the infrastructure for methane capturing and processing is a very effective way of a short-term reduction of emissions of this important greenhouse gas.

The analysis of the operation of the methane capture station with respect to appropriate concentrations and streams of captured gas from methane drainage and the review of possible solutions improving the quality of the captured gas allowed to present the following conclusions, that in order to maintain the methane concentration above the assumed value of 40%, the flowrate of gas from the capture of methane in the amount of not less than 50 m³CH₄/min with a minimum concentration of 50% should be ensured. It ensures that the required methane concentration in the fuel mixture for gas engines is maintained. Other sources of methane capture i.e. before or after exploitation sources shall be no more than complementary and shall not constitute the primary source of the methane-air mixture for gas engines.

The analysis of methane capture in particular months shows that the methane drainage station captured gas at the level of approx. 150 m³/min in the period from January to July. At that time, the stream of captured methane from the areas of exploitation was about 75 m³/min, which constitutes half of the mixture at the methane drainage station. From August to November, a decrease of about half of the captured gas is observed. Analysing data from methane capture, it is difficult to indicate the reason for such a state, except for the reduction of methane stream captured from exploitation.

The use of a methane-air mixture purification system to obtain high methane concentrations is not economically justified due to very high investment costs.

The use of systems of mixing compressed natural gas CNG with the gas from methane drainage of the mine essentially ensure that the required methane concentration is maintained for the undisturbed operation of gas engines. In order to eliminate commercial gas production, consideration should be given to the supply of gas from the mine's available intakes.

The use of underground methane tanks requires the identification of conditions, and the execution of a project, including the determination of the amount of methane and the time of its exploitation. It is possible for a certain period of time, however, the difficulty in obtaining the required methane concentration eliminates this system of supplying the methane-air mixture to the methane drainage station.

The optimal solution seems to be a system with a tank collecting methane from the methane drainage system at the methane drainage station. Due to its limited capacity, it ensures the operation of engines for several days. However, this solution is technically simple and inexpensive compared to other solutions. Additionally, it allows the use of the methane captured by methane drainage installations, which eliminates the costs associated with the purchase of gas.

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