

1 Article

# 2 Structural, Magnetic and Mössbauer Studies of Ga 3 and Zr Doped $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$ and $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$ 4 ( $x=0.00, 0.25, 0.50, 0.75, 1.00$ )

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9 **Abstract:** The substitution of Zr and Ga in  $\text{Dy}_2\text{Fe}_{17}$  was found to have an important effect on their  
10 structure and magnetic properties. The Rietveld analysis confirmed that the crystalline system  
11 were  $\text{Th}_2\text{Ni}_{17}$  structure. Lattice parameters  $a$  (Å) and  $c$  (Å), unit cell volume (Å<sup>3</sup>), bonding distance  
12 (Å) were calculated by using Rietveld analysis. The unit cell volume of  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$  and  
13  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$  increase linearly with the Zr and Ga substitution. The substitution of Zr and Ga  
14 are limited up to  $x=1$  to avoid the decrease in saturation magnetization. The Curie temperature  
15 ( $T_c$ ) of  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$  and  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$  are found to be Zr content dependent. The maximum  
16 curie temperature was observed 510K.  $x = 0.75$  for  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$  and 505.1 K,  $x = 0.5$  for  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$   
17 which are 102 K and 97 K higher than  $\text{Dy}_2\text{Fe}_{17}$ . The room temperature Mössbauer analysis  
18 shows the decrease average hyperfine field and increases the isomer shift with Zr doping.

19 **Keywords:** permanent magnetic materials; 2:17 intermetallic; X-ray diffraction; rietveld analysis;  
20 Mössbauer Spectroscopy; curie temperature; soft magnet; doping  
21

## 22 1. Introduction

23 Intermetallic compounds (rare-earth elements (R) and 3d-transition elements (T)) possess  
24 outstanding magnetic properties because of their high saturation magnetization ( $M_s$ ).  $\text{R}_2\text{Fe}_{17}$   
25 compounds were studied in late the 1970s and the values of  $(\text{BH})_{\text{max}}$  and Coercivity ( $H_c$ ) were found  
26 about 26 MGOe and 15 kOe respectively but they have very low Curie temperature (~473 K for  
27  $\text{Gd}_2\text{Fe}_{17}$  and ~300 K for  $\text{Dy}_2\text{Fe}_{17}$ ) and low magnetic anisotropies [1]. Many researches have been done  
28 to improve the curie temperatures of  $\text{Dy}_2\text{Fe}_{17}$  [2–4] either by replacing the Fe atoms with non-magnetic  
29 atoms (Al, Si, Ga) [5] or by doping refractory atoms (Ti, V, Mo, Nb, W, Zr) in Fe site [6–10]. It is also  
30 found that substitution of magnetic atoms such as (Co, Ni, Cr, Mn, Ni) [11], non-metals such as C, N,  
31 and H in  $\text{R}_2\text{Fe}_{17}$  lattice also increase the curie temperature of  $\text{R}_2\text{Fe}_{17}$  compounds. [12–15]. The  
32 substitution of non-magnetic atoms at Fe sites has been reported to increase ferromagnetic coupling  
33 which in turn increases the Curie temperature [16–17]) and magneto-crystalline anisotropy [18]. The  
34 introduction of nitrogen on interstitial sites of  $\text{R}_2\text{Fe}_{17}\text{N}_x$  also increased the Curie temperature. This  
35 happened because of volume expansion and increase of the distance of iron atoms which changes the  
36 magnetic properties of  $\text{R}_2\text{Fe}_{17}$  compounds [19,20]. Betancourt *et al* (2003) [21] reported that Zr and Nb  
37 substitution can also improve magnetic properties. The curie temperatures of  $\text{Ce}_2\text{Fe}_{17}$  [22],  $\text{Gd}_2\text{Fe}_{17}$  [5]  
38  $\text{Dy}_2\text{Fe}_{17}$  [8],  $\text{Pr}_2\text{Fe}_{17}$  [23], etc., are improved by the addition of elements like Si, Cr, Mn, and Ga. The  
39 substitution of the non-magnetic atom into the  $\text{R}_2\text{Fe}_{17}$  brings the concomitant decrease in  
40 magnetization which affects the energy product of the permanent magnets [18]. Furthermore, it is  
41 possible to increase the magnetic moment *via* Fe-Zr 3d band hybridization which can bring band  
42 narrowing or increase exchange splitting by moving the  $3d\uparrow$  states below the Fermi level or allow  
43 charge transfer out of the 3d band, provided the spin-down density of states exceed the spin-up  
44 density [24]

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46 The present work focusses on replacing the Fe atom by elements, Zr and Ga into the Dy<sub>2</sub>Fe<sub>17</sub>.  
47 The substitution of the non-magnetic atom was limited to up to x=1 to prevent a significant reduction  
48 in saturation magnetization. Doping of Zr and Ga atoms can increase the lattice parameter *a*, *c* and  
49 unit cell volume and can also improve Fe-Fe exchange interaction.

## 50 2. Experimental

51 The raw materials of Dy, Fe, Zr and Ga were used with ~99.9% of purity and purchased from  
52 Sigma Aldrich. The samples Dy<sub>2</sub>Fe<sub>17-x</sub>Zr<sub>x</sub> and Dy<sub>2</sub>Fe<sub>16</sub>Ga<sub>1-x</sub>Zr<sub>x</sub> (*x* = 0.00, 0.25, 0.50, 0.75, 1.00) were  
53 prepared via arc melting technique in a high purity argon atmosphere. The ingots were melted  
54 several times to ensure homogeneity. The prepared alloys were made into fine powders and the X-  
55 ray powder data sets were collected using Bruker (D8 Advance) diffractometer with the  
56 monochromatic incident beam of CuK<sub>α</sub> ( $\lambda \sim 1.5406 \text{ \AA}$ ) radiation with the 2 $\theta$  range from 20° to 70° with  
57 the step size of 0.042°. The XRD analysis was performed by the well-known refinement Rietveld  
58 method [25] using the JANA2006 [26] software package. The purpose of this refinement process is to  
59 minimize the difference between the theoretically modeled profile and the observed ones as a  
60 function,  $\sum_{hkl} W_{hkl} (|F_o| - |F_c|)^2$  where *W<sub>hkl</sub>* is the weight assigned to each observation and *F<sub>o</sub>* and  
61 *F<sub>c</sub>* are the observed and calculated structure factors.

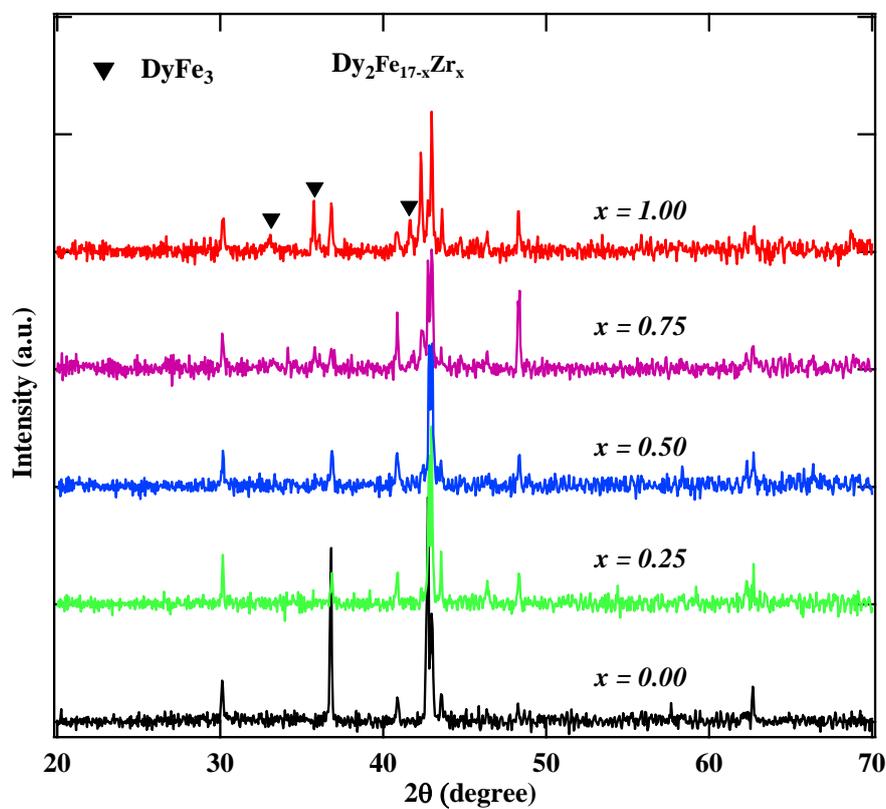
62 Magnetic properties of the powder sample were investigated at room temperature (RT) using  
63 vibrating sample magnetometer (VSM) in the maximum field of 1.2T. To minimize the effect of the  
64 demagnetizing field, the samples were compacted at 3000psi and cut into rectangular parallelepiped  
65 with the ratio of length to a width larger than three times and embedded in epoxy. A modified  
66 thermogravimetric analyzer (DuPont 910) equipped with a permanent magnet was used to determine  
67 the Curie temperature of composite samples.

68 The Mössbauer spectra of the samples were obtained at RT using a 25 mCi <sup>57</sup>Co source in Rh foil  
69 mounted on a constant acceleration drive system (SEE Co. Minneapolis, USA) in transmission  
70 geometry. The velocity scale of the Mössbauer spectrometer was calibrated by measuring the  
71 hyperfine field of  $\alpha$ -Fe foil, at room temperature. The Mössbauer spectra were analyzed using the  
72 WMoss software from SEE Co. The spectra were least-square fitted with the hyperfine field (HF),  
73 isomer shift (IS) and quadrupole shift (QS) as variables.

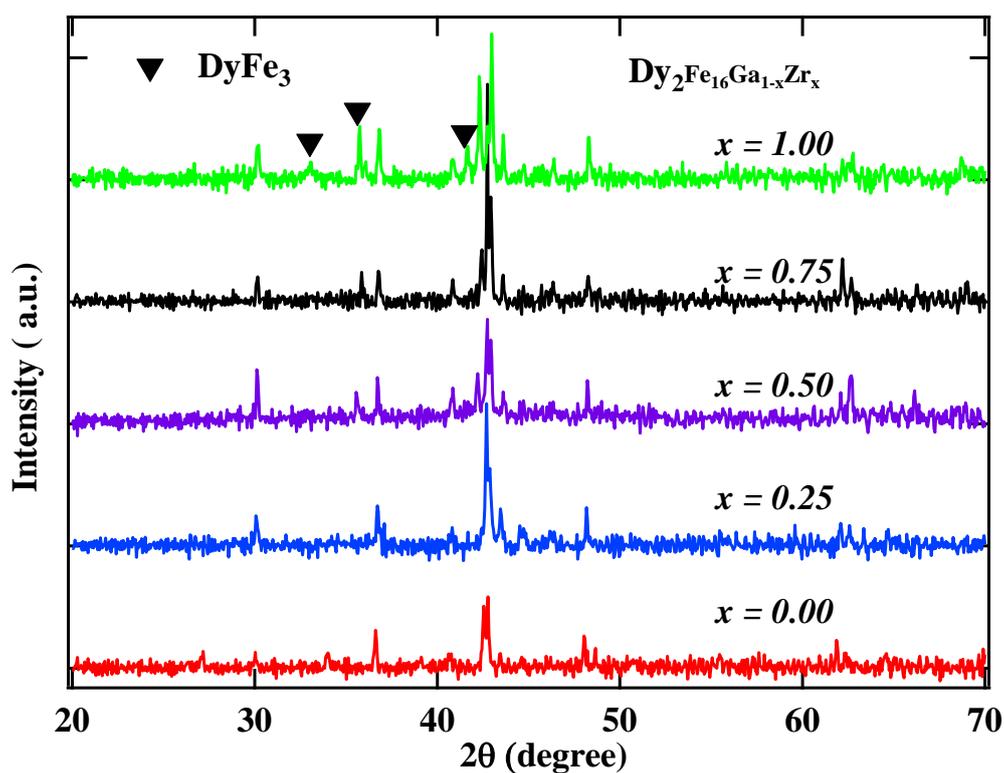
## 74 3. Results and Discussions

75 **Figure 1(a)** and **(b)** represent the raw XRD powder profile of Dy<sub>2</sub>Fe<sub>17-x</sub>Zr<sub>x</sub> and Dy<sub>2</sub>Fe<sub>16</sub>Ga<sub>1-x</sub>Zr<sub>x</sub> (*x*  
76 = 0.00, 0.25, 0.50, 0.75, 1.00). **Figure 2** represents the representative Rietveld refinement profile for  
77 Dy<sub>2</sub>Fe<sub>16</sub>Ga<sub>0.75</sub>Zr<sub>0.25</sub>. The Rietveld [25] fittings reveal a nice matching of the observed and calculated  
78 profiles for both systems. A small additional phase DyFe<sub>3</sub> was observed for x=1 Zr content. During  
79 the refinement process, the scale, structural parameters, lattice parameters, peak shift, preferred  
80 orientation, background profile functions, thermal parameters, surface roughness were refined until  
81 the observed XRD profile matches well with the calculated profile. The initial crystal structure  
82 parameters were used as given by Liao et al. [27]. In the hexagonal setting, Dy was fixed at the 2*b* (0,  
83 0, 0.25) and 2*d* (0.333, 0.667, 0.75) site and Fe was fixed at 4*f*(0.333 0.667 0.105), 6*g*(0.5 0 0), 12*j*(0.333  
84 0.969 0.25), and 12*k*(0.167 0.333 0.985). The profile was constructed using a pseudo-Voigt function.  
85 Profile asymmetry was introduced by employing the multi-term Simpson rule integration devised  
86 by Howard [28]. A surface roughness correction was also applied using the Pitschke, Hermann, and  
87 Matter [29] model. **Table 1** shows the refined structural parameters for both for Dy<sub>2</sub>Fe<sub>16-x</sub>Zr<sub>x</sub> and  
88 Dy<sub>2</sub>Fe<sub>16</sub>Ga<sub>1-x</sub>Zr<sub>x</sub>. From **Table 1**, the refined results confirmed that the unit cell volume ( $\text{\AA}^3$ ) increases  
89 with x-content. The plot for lattice parameters *a*, *c* and unit cell volume (*V*) are shown in **Figure 3 (a)**  
90 and **3(b)**. This increase in volume maybe because of bigger atom Ga (*r*<sub>ionic</sub> = 136 pm) and Zr (*r*<sub>ionic</sub> = 206  
91 pm) sitting at the smaller iron (*r*<sub>ionic</sub> = 126pm) atom site. The increasing lattice parameters confirmed  
92 that both Zr and Ga addition to the host lattice has been done properly. **Table 2** represents, the site  
93 occupancy table for Dy<sub>2</sub>Fe<sub>17-x</sub>Zr<sub>x</sub> and Dy<sub>2</sub>Fe<sub>16</sub>Ga<sub>1-x</sub>Zr<sub>x</sub>, which shows the Zr atom occupies all 4*f*, 6*g*, 12*j*

94 and 12k sites but the 12j and 12k sites are mostly impacted with the Zr substitution (**Figure 4(a)**), the  
 95 site occupancy of Zr and Ga for  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$  are plotted in **Figure 4(b)** and the values are presented  
 96 in **Table 2**. From these results, both Zr and Ga atoms prefer to be in 12j and 12k sites with a minimum  
 97 affinity for 4f and 6g sites [30,31] and the increase in Zr occupancy and the decrease in Ga occupancy  
 98 at 12j and 12k site shows that Zr atom is replaced by Ga atom in  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$ .



(a)



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(b)

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**Figure 1.** X-ray diffraction patterns of (a)  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$  (b)  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$  compounds.

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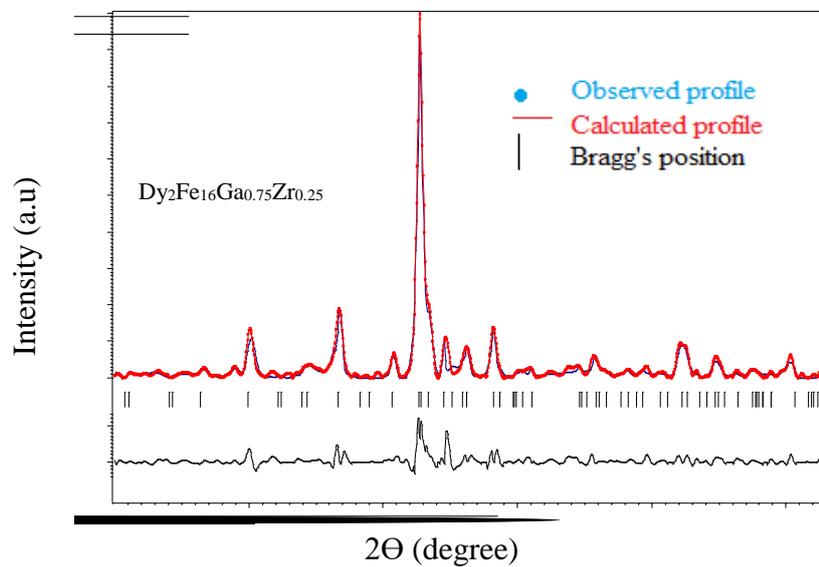
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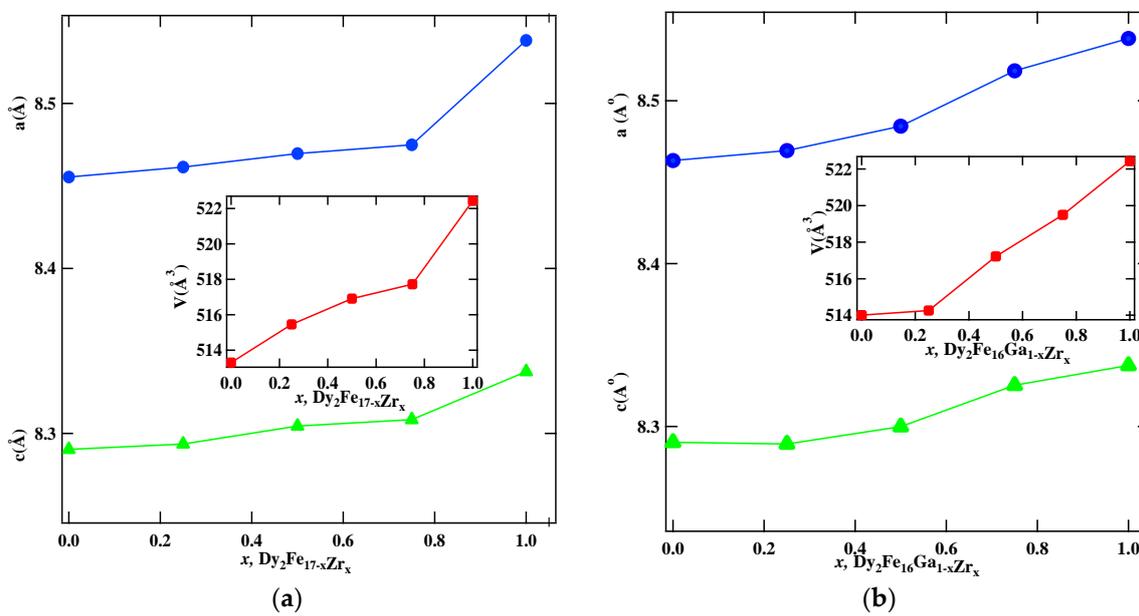
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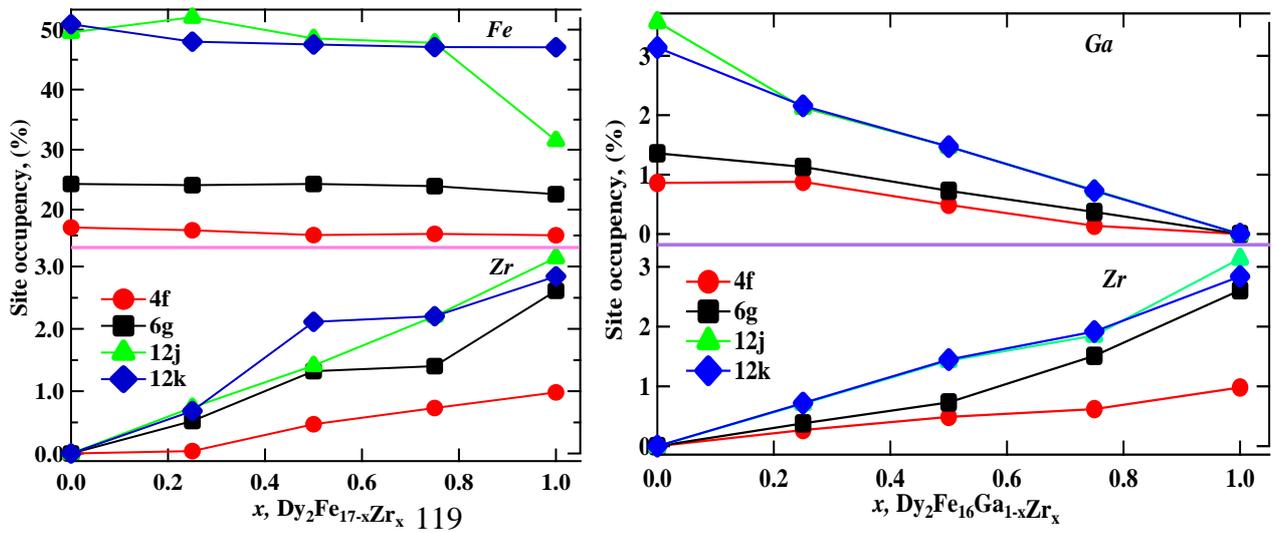
**Figure 2.** Rietveld refinement profile for  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{0.75}\text{Zr}_{0.25}$ 

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**Figure 3.** Lattice parameters  $a$  and  $c$  and unit cell volume of (a)  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$  and (b)  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$  as a function of Zr content.



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(a)

(b)

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**Figure 4.** The percentage occupancy of Fe and Zr at iron sites in (a)  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$  (b) Fe and Zr at Ga sites in  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$  as a function of Zr content.

123  
124**Table 1.** Lattice parameters and unit cell volume of  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$  and  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$  obtained from Rietveld refinements.

		<b>a (Å)</b>	<b>c (Å)</b>	<b>V (Å<sup>3</sup>)</b>	<b>R<sub>obs</sub> (%)</b>	<b>wR<sub>obs</sub> (%)</b>	<b>R<sub>p</sub> (%)</b>	<b>wR<sub>p</sub> (%)</b>
$\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$	0	8.45541(11)	8.29036(4)	513.303(18)	4.17	5.95	10.65	14.49
	0.25	8.46147(19)	8.29361(13)	515.456(21)	3.13	4.16	12.32	15.43
	0.5	8.46963(16)	8.30442(12)	516.913(16)	4.16	4.33	11.91	17.21
	0.75	8.4749(14)	8.30836(11)	517.725(21)	4.52	5.11	9.32	10.12
	1	8.53823(17)	8.33743(19)	522.429(13)	5.01	6.99	9.96	10.21
$\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$	0	8.46324(3)	8.29025(3)	514.013(26)	5.97	6.17	10.21	12.13
	0.25	8.46943(15)	8.28922(7)	514.265(28)	5.65	6.97	13.64	16.98
	0.5	8.48437(12)	8.29985(5)	517.221(23)	4.52	5.11	9.32	10.12
	0.75	8.51833(7)	8.32536(7)	519.498(13)	5.21	6.12	12.21	11.21
	1	8.53823(17)	8.33743(19)	522.429(13)	5.01	6.99	9.96	10.21

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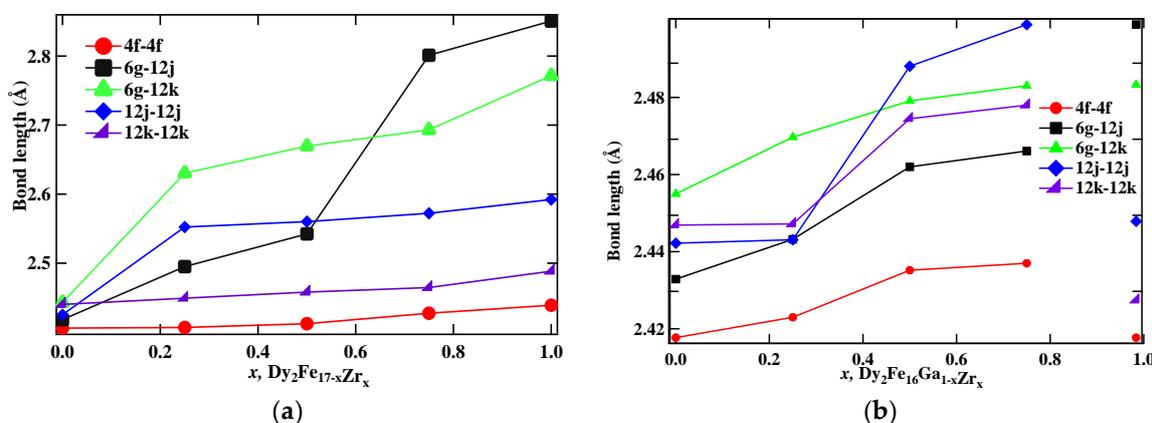
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**Table 2.** The site occupancy table for  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$  and  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$ .

	<b>x</b>	<b>Fe(4f)</b>	<b>Fe(6g)</b>	<b>Fe(12j)</b>	<b>Fe(12k)</b>	<b>Zr(4f)</b>	<b>Zr(6g)</b>	<b>Zr(12j)</b>	<b>Zr(12k)</b>
Site occupancy for $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$ (%)	0.00	16.9(59)	24.3(6)	49.5(95)	50.9(31)	0	0	0	0
	0.25	16.5(2)	24.1(7)	52.0(1)	48.0(4)	0.04(3)	0.52(6)	0.74(1)	0.68(13)
	0.50	15.7(1)	24.3(4)	48.5(12)	47.5(7)	0.47(14)	1.32(15)	1.41(3)	2.11(17)
	0.75	15.9(2)	23.9(5)	47.7(1)	47.1(17)	0.73(8)	1.40(3)	2.20(6)	2.20(15)
	1.00	15.6(4)	22.6(4)	31.5(1)	47.0(3)	0.98(7)	2.61(6)	3.14(3)	2.84(11)
	<b>x</b>	<b>Ga(4f)</b>	<b>Ga(6g)</b>	<b>Ga(12j)</b>	<b>Ga(12k)</b>	<b>Zr(4f)</b>	<b>Zr(6g)</b>	<b>Zr(12j)</b>	<b>Zr(12k)</b>
Site occupancy for $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$ (%)	0.00	0.86(21)	1.36(15)	3.57(16)	3.14(23)	0	0	0	0
	0.25	0.88(43)	1.13(43)	2.13(12)	2.16(11)	0.27(12)	0.38(14)	0.71(36)	0.72(41)
	0.50	0.49(31)	0.73(42)	1.47(13)	1.47(32)	0.49(41)	0.73(32)	1.43(32)	1.45(11)
	0.75	0.14(22)	0.37(62)	0.74(16)	0.73(31)	0.62(21)	1.51(21)	1.85(22)	1.92(13)
	1.00	0	0	0	0	0.98(7)	2.61(6)	3.14(3)	2.84(11)

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128 **Table 3** represents the distance between the Fe-Fe neighbor sites, Ms and Tc for  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$  and  
 129  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$  and the values are plotted in **Figure 5(a)** and **(b)**. It can be observed from **Table 3** that  
 130 the average bond distance between each site changes with the concentration of both Zr and Ga ( $4f-4f$   
 131  $\sim 2.42$  Å), ( $12k-12k \sim 2.46$  Å) and ( $12j-12j \sim 2.52$  Å).  $12j-12j$  sites show both Zr and Ga prefers to stay in  
 132 the same sites.

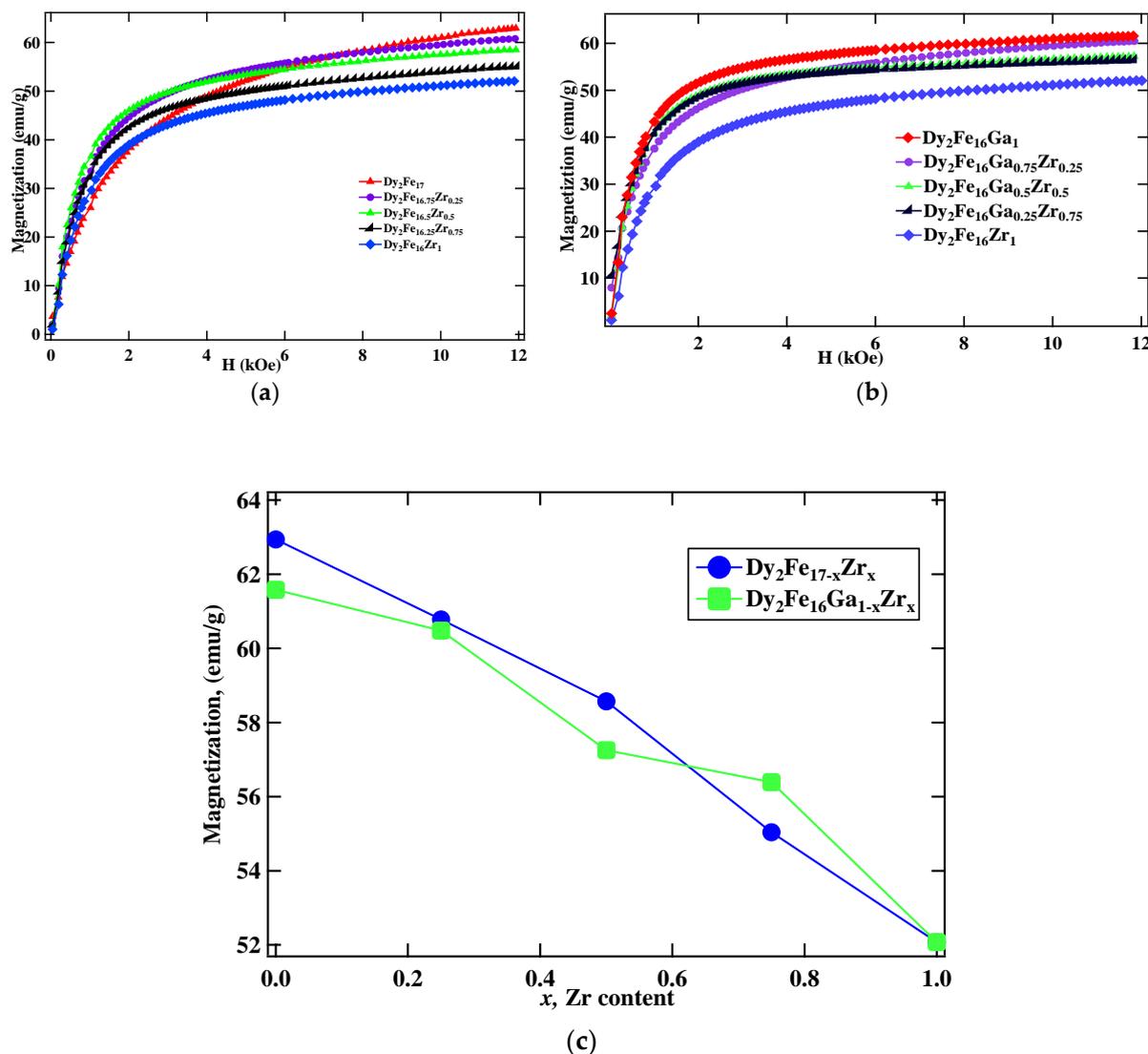


133 **Figure 5.** Dependence of bond lengths on  $x$  of (a)  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$  (b)  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$ .

134 **Table 3.** Distance between the Fe-Fe neighbor sites, Ms and Tc for  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$  and  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$ .

$x$	Bond length $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$ (Å)					Bond length $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$ (Å)					$\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$		$\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$	
	$4f-4f$	$6g-12j$	$6g-12k$	$12j-12j$	$12k-12k$	$4f-4f$	$6g-12j$	$6g-12k$	$12j-12j$	$12k-12k$	Ms (emu/g)	Tc (K)	Ms	Tc (K)
0.00	2.406	2.418	2.443	2.425	2.44	2.418	2.433	2.455	2.442	2.44	62.94	408	61.58	488
0.25	2.407	2.495	2.63	2.552	2.449	2.423	2.443	2.469	2.443	60.78	456	60.48	500.1	
0.50	2.412	2.542	2.669	2.56	2.458	2.435	2.462	2.479	2.488	58.57	503.9	57.25	505.1	
0.75	2.427	2.801	2.693	2.572	2.464	2.437	2.466	2.483	2.499	55.04	510	56.39	502.5	
1.00	2.439	2.851	2.772	2.592	2.488	2.439	2.85	2.771	2.592	52.07	425	52.07	425	

135 The room temperature magnetic properties obtained from  $M$  vs.  $H$  measurement for  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$   
 136 and  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$  are plotted in **Figure 6(a)** and **(b)** respectively. The  $M$  vs.  $H$  plot shows that the  
 137 saturation magnetization ( $M_s$ ) decreases with an increase in the Zr substitution in both  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$   
 138 and  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$ . The “Law of Approach” to saturation magnetization was used to determine the  
 139 saturation magnetization,  $M_s$ . The “Law of approach” describes the relationship between  
 140 magnetization  $M$  on the applied magnetic field for  $H$  greater than coercive field  $H_c$ . The  
 141 magnetization near  $M_s$  can be written as  $M = M_s \left( 1 - \frac{a}{H} - \frac{b}{H^2} \right) + \kappa H$  [32] where  $M$  is the  
 142 magnetization,  $H$  is the applied magnetic field, and  $M_s$  is the saturation magnetization attained at  
 143 high field. The term  $\kappa H$  represents the field-induced increase in the spontaneous magnetization of  
 144 the domains. This term is very small at a temperature well below the Curie temperature and could  
 145 be neglected. The term “ $a$ ” is generally interpreted as due to microstress and ignored in the high field  
 146 region, and “ $b$ ” as due to crystal anisotropy. Where magneto-crystalline is a dominant term, a plot of  
 147  $M$  vs.  $1/H^2$  in the high field region gives a straight line, the intercept of which (with the  $M$ -axis) gives  
 148 the  $M_s$  and the slope of which gives the magneto-crystalline anisotropy constant. An interesting  
 149 variation in  $M_s$  is noticed with the Zr atom doping. The decrease in saturation magnetization in  
 150  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$  with the increase in Zr content is attributed to the substitution of non-magnetic Zr atom  
 151 for Fe atom. This decrease in the saturation magnetization is the result of the dilution effect *i.e.*  
 152 replacing Fe atom ( $2.2 \mu_B$ ) with non-magnetic Zr atom. Similar reduction in  $M_s$  has been reported  
 153 earlier for  $\text{Dy}_2\text{Fe}_{17-x}\text{Ga}_x$  [18],  $\text{Ce}_2\text{Fe}_{17-x}\text{Ga}_x$  [33],  $\text{Sm}_2\text{Fe}_{17-x}\text{Ga}_x$  [34] compounds. From **Figure 6(c)**, it is  
 154 observed that the saturation magnetization decreases linearly from  $62.94 \mu_B/\text{Fe}$  to  $52.07 \mu_B/\text{Fe}$  per Zr  
 155 atom for  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$  however magnetization decreases from  $61.58 \mu_B/\text{Fe}$  to  $52.07 \mu_B/\text{Fe}$  in  
 156  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$ . The dropdown of magnetization in both cases could be due to the reduction of the  
 157 magnetic moment of iron by Zr atoms.

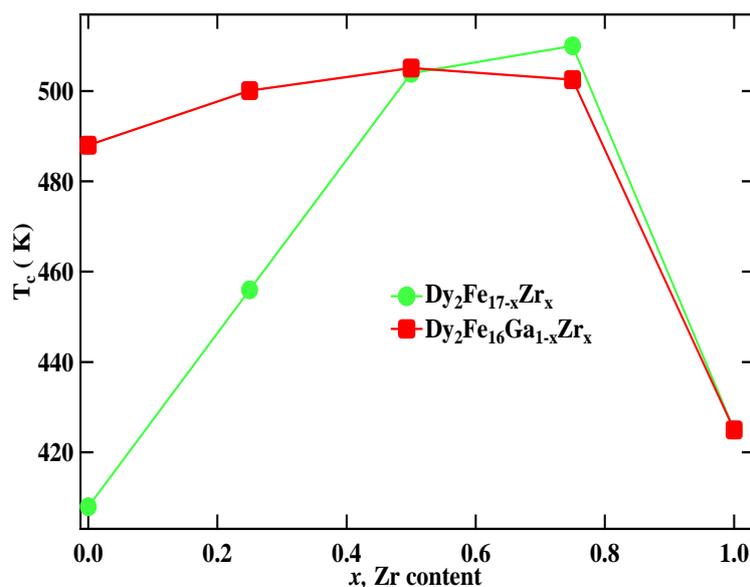


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**Figure 6.** RT M vs. H plot of (a) Dy<sub>2</sub>Fe<sub>17-x</sub>Zr<sub>x</sub> (b) Dy<sub>2</sub>Fe<sub>16</sub>Ga<sub>1-x</sub>Zr<sub>x</sub> (c) Magnetization (Ms) vs. x for Dy<sub>2</sub>Fe<sub>17-x</sub>Zr<sub>x</sub> and Dy<sub>2</sub>Fe<sub>16</sub>Ga<sub>1-x</sub>Zr<sub>x</sub>.

161 The Curie Temperature (T<sub>c</sub>) of Dy<sub>2</sub>Fe<sub>17-x</sub>Zr<sub>x</sub> and Dy<sub>2</sub>Fe<sub>16</sub>Ga<sub>1-x</sub>Zr<sub>x</sub> as a function of Zr content is  
162 shown in **Figure 7** and corresponding data are listed in **Table 3**. It is observed that for Dy<sub>2</sub>Fe<sub>17-x</sub>Zr<sub>x</sub>  
163 Curie temperature increases from 408 K ( $x = 0.00$ ) to a maximum of 510 K ( $x = 0.75$ ) and then decreases  
164 to 425 K ( $x = 1.00$ ). The achieved curie temperature of Dy<sub>2</sub>Fe<sub>16.25</sub>Zr<sub>0.75</sub> is 102 K higher than that of  
165 Dy<sub>2</sub>Fe<sub>17</sub>. It is observed from **Figure 5(a) and (b)** that the bond length increases with the increase in the  
166 Zr content. In general, the curie temperature in rare-earth intermetallic is due to three kinds of  
167 exchange interactions namely the 3d-3d exchange interactions *i.e.* between the magnetic moment of  
168 the Fe sub-lattice ( $J_{FeFe}$ ), 4f-4f exchange interaction *i.e.* the interaction between the magnetic moment  
169 within the R sub-lattice ( $J_{RR}$ ), and the inter sub-lattice 3d-4f exchange interaction ( $J_{RFe}$ ). It is reported  
170 that the Curie temperature (T<sub>c</sub>) increases with an increase in the  $J_{FeFe}$  [35]. Thus, the increase or  
171 decrease in T<sub>c</sub> in the R<sub>2</sub>Fe<sub>17</sub> intermetallic may be due to the increase or decrease in the interaction

172 parameter  $J_{\text{FeFe}}$ . The interactions between the rare-earth spins (4f-4f) are assumed to be weak and  
 173 negligible in comparison with the other two types of interactions. The low  $T_c$  observed in parent  
 174  $\text{Dy}_2\text{Fe}_{17}$  compound is believed to be due to the short Fe-Fe interatomic distances found at the 4f(6c)  
 175 sites in the hexagonal (rhombohedral) structure which couple antiferromagnetically [36] since their  
 176 separation is less than 2.45 Å **Figure 5(a) and (b)**, needed for ferromagnetic ordering [37]. In the  
 177 compounds with hexagonal structure the Fe(4f)- Fe(4f) interactions are strongly negative whereas the  
 178 Fe(6g)- Fe(12j), Fe(6g)-Fe(12k) and Fe(12k)-Fe(12k) interactions are weakly negatives. Thus, the  
 179 increase in  $T_c$  is considered due to the increase in strength of the Fe-Fe exchange coupling that occurs  
 180 from the increase in Fe-Fe bond lengths for Fe-rich  $\text{R}_2\text{Fe}_{17}$  compounds. The strength of Fe-Fe  
 181 exchange interaction highly depends on interatomic Fe-Fe distance [38,39]. Accordingly, the  
 182 exchange interactions between iron atoms situated at distances smaller (greater) than 2.45–2.50 Å are  
 183 negative (positive). In the  $\text{R}_2\text{Fe}_{17}$  majority of Fe-Fe distances favor the negative interaction [18]. The  
 184 negative exchange interaction can be reduced either by volume expansion or by reducing the number  
 185 of Fe-Fe pairs with negative exchange interactions. It is noted that the increase in  $T_c$  has been reported  
 186 earlier with Ga substitution ( $x=1$ ) [18] however with the simultaneous substitution of non-magnetic Ga  
 187 and Zr atoms enhances the curie temperature of the intermetallic that we have reported without  
 188 significantly lowering the saturation magnetization. It is found that, for  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$ , Curie  
 189 temperature increases slowly from 488 K ( $x = 0.00$ ) to maximum of 505.1 K ( $x = 0.50$ ) and then  
 190 decreases to 425 K ( $x = 1.00$ ). The maximum  $T_c$  observed in  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{0.5}\text{Zr}_{0.5}$  (505.1 K) is 17.1 K greater  
 191 than of  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_1$  and 97 K higher than the  $\text{Dy}_2\text{Fe}_{17}$  which is indicating that double substitution of  
 192 Zr and Ga does slow enhancement in  $T_c$  over the single substitution.

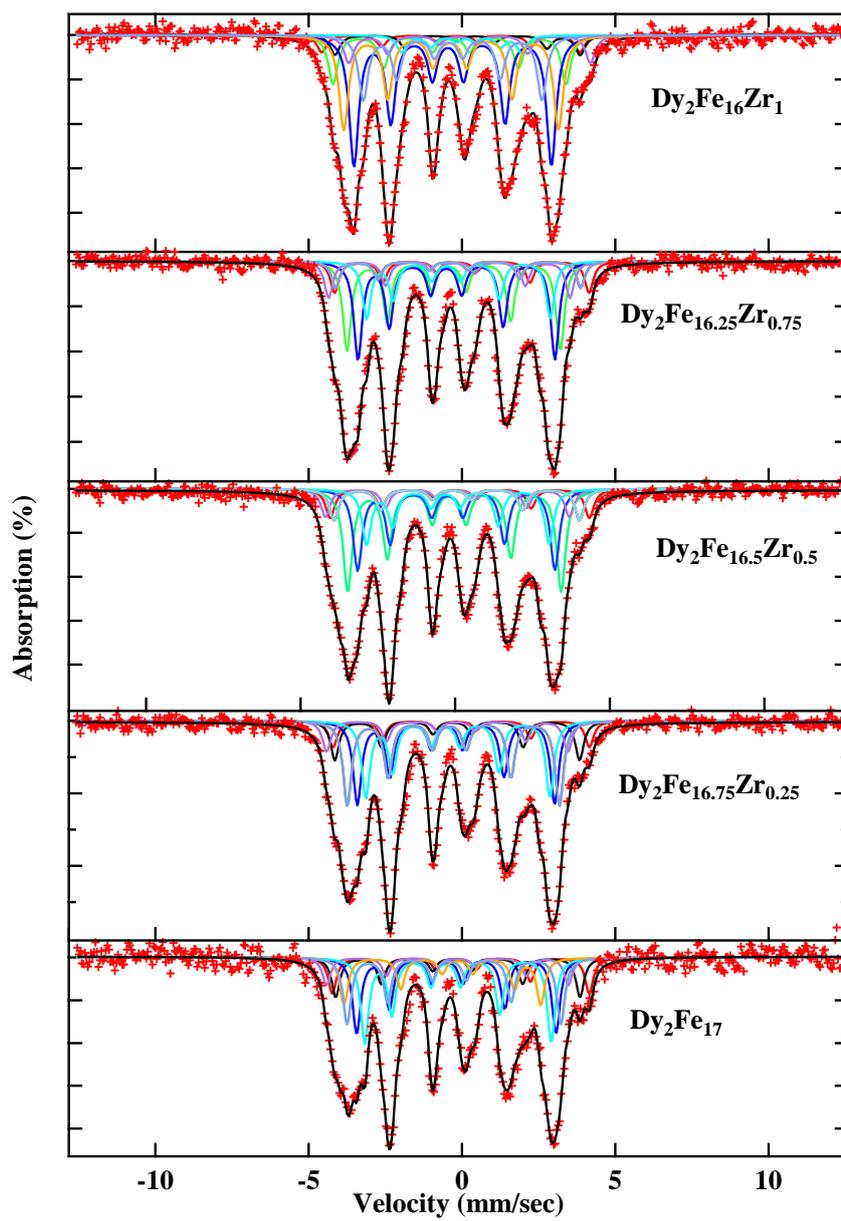


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**Figure 7.** Comparative study for the  $T_c$  vs.  $x$  for  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$  and  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$ .

195 The Mössbauer spectra are plotted in **Figure 8(a) and (b)** for  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$  and  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$   
 196 ( $x=0.00, 0.25, 0.50, 0.75, 1.00$ ) respectively. The resulting spectra were fitted using the WMOSS  
 197 program. Since the  $\text{Dy}_2\text{Fe}_{17}$  compounds have a basal magnetization, either seven or eight magnetic  
 198 sextets are required to fit their Mössbauer spectra. The fitted data are shown in **Fig. 8(a) and (b)**  
 199 was carried out with eight magnetic sextets assigned to 4f, 6g, 12j, and 12k sites in  $\text{Dy}_2\text{Fe}_{17}$ . Doublets were  
 200 used for additional phases for the sample at  $x=1$  for the paramagnetic phase ( $\text{DyFe}_3$ ) observed during  
 201 Mössbauer fitting.



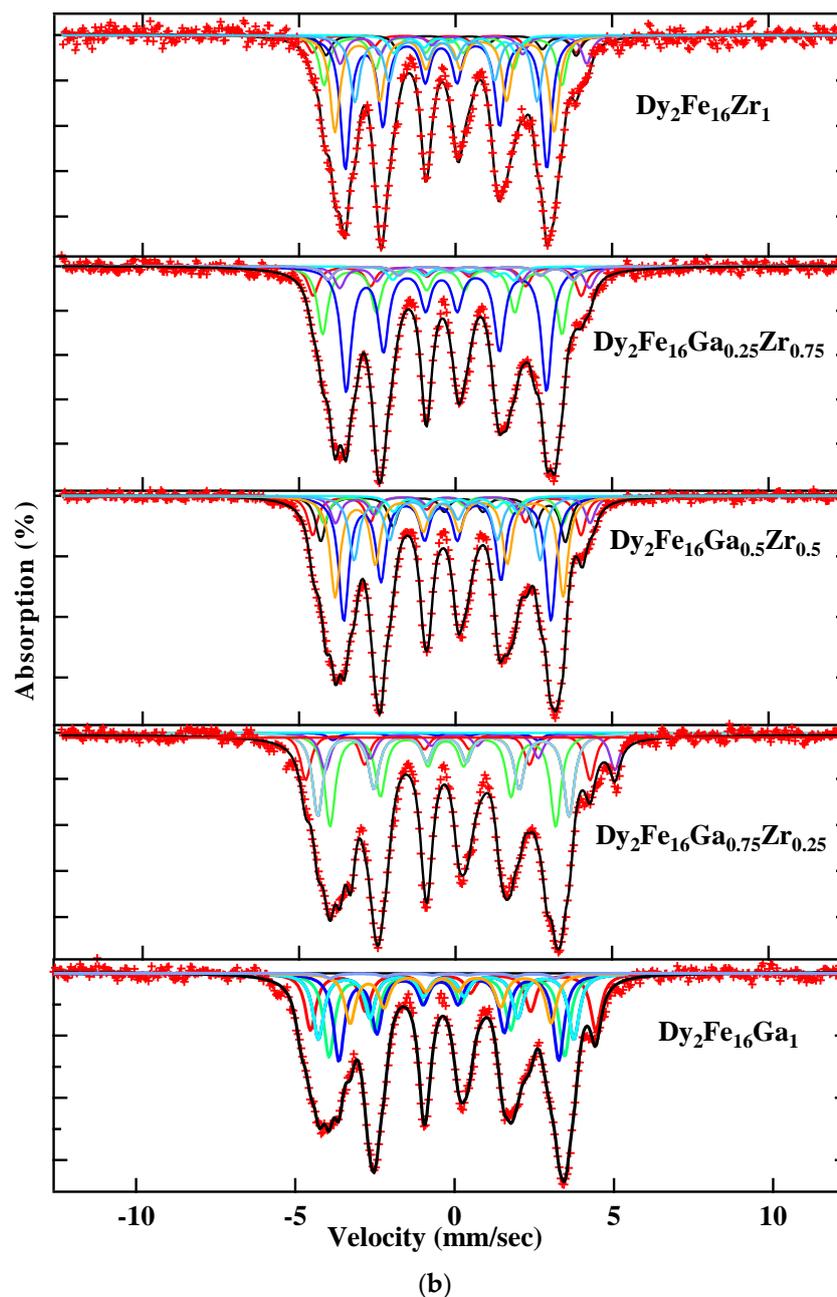
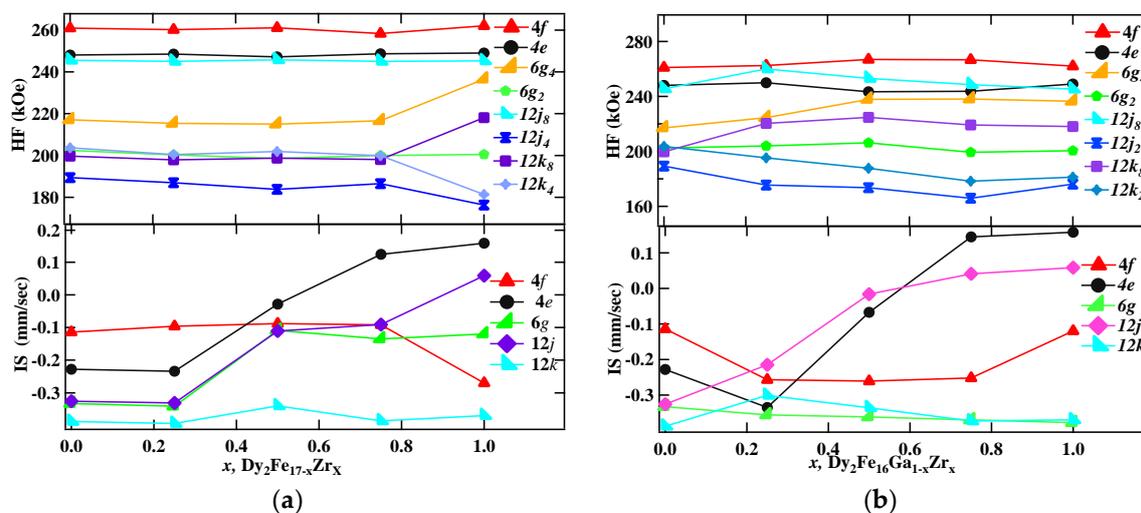


Figure 8. RT Mössbauer spectra for (a)  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$ , (b)  $\text{Dy}_2\text{Fe}_{16}\text{Ga}_{1-x}\text{Zr}_x$ .

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203 The hyperfine parameters derived from the fitting are listed in **Table 4(a)** and **(b)** and the  
 204 hyperfine field (HF) and isomer shifts (IS) are plotted in **Figure 9(a)** and **(b)**. There exists a direct  
 205 correlation between hyperfine field values of a site to its near neighbor (NN) iron sites. In case of  
 206  $\text{Th}_2\text{Ni}_{17}$  structure,  $12k$  site has 9 NN Fe sites (1 ( $4f$ ), 2 ( $6g$ ), 4 ( $12j$ ), 2( $12k$ )),  $12j$  has 10 NN Fe sites (2 ( $4f$ ),  
 207 2 ( $6g$ ), 2 ( $12j$ ), 4 ( $12k$ )),  $6g$  has 10 NN Fe sites (2 ( $4f$ ), 0 ( $6g$ ), 4 ( $12j$ ), 4( $12k$ )), and  $4f$  site has 11 NN Fe sites  
 208 (1 ( $4f$ ), 3 ( $6g$ ), 6 ( $12j$ ), 3 ( $12k$ )). Following the NN distribution, the observed HF values are in  
 209  $4f(6c) > 12j(18f) > 6g(9d) > 12k(18h)$  sequence, which is similar to the sequence observed in other  $\text{R}_2\text{Fe}_{17}$   
 210 compounds [40,41]. It is obvious that  $4f$  ( $6c$ ) site has the maximum hyperfine field, since it has the  
 211 maximum number of Fe nearest neighbors, whereas, the  $18h$  ( $12k$ ) site has the minimum number of  
 212 Fe neighbors and consequently has the least HF value. Although  $6g(9d)$  and  $12j(18f)$  sites have the  
 213 same number of Fe neighbors, the former has comparatively smaller Fe-Fe distances and hence a  
 214 larger hyperfine field. The hyperfine field ( $B_{\text{hf}}$ ) decreases with increasing Zr content. The decrease in  
 215 the hyperfine fields could be from the decrease in magnetic moments of iron in  $\text{Dy}_2\text{Fe}_{17-x}\text{Zr}_x$ . It is  
 216 found that the 9.23% of total area occupied by the  $\text{DyFe}_3$  paramagnetic phase for  $x = 1$  Zr content.

217 This observation supports the formation of additional paramagnetic phase DyFe<sub>3</sub> as seen in the XRD  
 218 plot in Figure 1.



219 **Figure 9.** RT Mössbauer hyperfine parameters plots for (a) Dy<sub>2</sub>Fe<sub>17-x</sub>Zr<sub>x</sub> (b) Dy<sub>2</sub>Fe<sub>16</sub>Ga<sub>1-x</sub>Zr<sub>x</sub>.

220 **Table 4.** (a) The RT hyperfine parameters, HF (kOe), IS (mm/sec), QS (mm/sec) and area (%) of the  
 221 Dy<sub>2</sub>Fe<sub>17-x</sub>Zr<sub>x</sub>.

	$x$	4f	4e	6g <sub>4</sub>	6g <sub>2</sub>	12j <sub>8</sub>	12j <sub>4</sub>	12k <sub>8</sub>	12k <sub>4</sub>	Doublet
HF (kOe)	0.00	261.0	248.0	217.0	202.4	245.6	189.0	199.7	203.7	
	0.25	260.2	248.5	215.3	200.3	245.0	187.0	197.9	200.4	
	0.50	261.1	247.1	214.9	198.6	245.8	184.0	198.7	201.9	
	0.75	258.4	248.7	216.6	200.0	245.0	186.0	198.0	199.9	
	1.00	262.1	249.0	236.5	200.5	245.3	176.0	218.1	181.3	34.0
IS (mm/s)	0.00	-0.114	-0.228	-0.333	-0.33	-0.326	-0.33	-0.39	-0.39	
	0.25	-0.096	-0.234	-0.341	-0.34	-0.331	-0.33	-0.39	-0.39	
	0.50	-0.088	-0.028	-0.109	-0.11	-0.111	-0.11	-0.34	-0.34	
	0.75	-0.092	0.124	-0.135	-0.14	-0.091	-0.09	-0.39	-0.39	
	1.00	-0.27	0.158	-0.12	-0.12	0.059	0.06	-0.37	-0.37	0.056
QS (mm/s)	0.00	0.152	0.191	0.152	0.303	-0.295	0.4	0.083	-0.58	
	0.25	0.152	0.16	0.128	0.296	-0.291	0.4	-0.03	-0.58	
	0.50	0.119	0.163	0.175	0.289	-0.276	0.4	0.121	-0.5	
	0.75	0.178	0.301	0.157	0.335	-0.199	0.4	-0.03	-0.58	
	1.00	0.123	0.25	-0.048	0.155	-0.25	0.35	0.042	-0.58	-0.39
Area (%)	0.00	4.0	1.91	15.3	17.5	4.3	19.5	10.9	13.1	
	0.25	5.5	8.2	17.7	17.4	6.3	16.0	10.4	13.3	
	0.50	6.2	6.7	21.9	17.3	6.0	11.7	9.60	15.1	
	0.75	5.25	4.0	21.0	19.0	7.6	14.0	14.81	13.5	
	1.00	4.2	4.9	11.2	31.9	6.2	4.21	23.1	15.2	9.23

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228**Table 4. (b)** The RT hyperfine parameters, HF (kOe), IS (mm/sec), QS (mm/sec) and area (%) of the  $Dy_2Fe_{16}Ga_{1-x}Zr_x$ .

	$x$	$4f$	$4e$	$6g_4$	$6g_2$	$12j_8$	$12j_4$	$12k_8$	$12k_4$	Doublet
HF (kOe)	0.000	278.7	249.7	230.2	214.9	260	218.2	196.5	220	
	0.25	282.9	249.9	224.5	204	288.1	164.8	220.3	195.3	
	0.5	267.6	267.9	229	204	254.2	123.4	226.8	185.3	
	0.75	266.7	243.7	238.1	199.4	248.6	120.7	219.2	178.3	
	1.00	262.1	248.9	236.5	200.4	245.3	122	218.1	181.3	34
IS (mm/s)	0.00	-0.126	-0.301	-0.309	-0.309	-0.422	-0.422	-0.254	-0.254	
	0.25	-0.257	-0.335	-0.356	-0.356	0.215	0.215	-0.301	-0.301	
	0.50	-0.227	-0.509	-0.367	-0.367	0.068	0.068	-0.327	-0.327	
	0.75	-0.252	0.145	-0.37	-0.37	0.041	0.041	-0.373	-0.373	
	1.00	-0.27	0.158	-0.378	-0.378	0.058	0.058	-0.37	-0.37	0.0557
QD (mm/s)	0.00	0.158	0.0107	0.133	0.259	-0.418	-0.48	0.259	-0.58	
	0.25	0.084	-0.102	-0.093	-0.568	0.5	-0.477	0.284	0.178	
	0.50	-0.069	-0.7	-0.196	0.201	0.5	-0.134	0.249	0.118	
	0.75	-0.029	-0.6	-0.102	0.156	0.5	-0.214	0.057	0.079	
	1.00	-0.181	-0.6	-0.049	0.155	0.39	-0.142	0.042	0.136	-0.388
Area (%)	0.00	13.7	15.9	20	21	10.5	5.5	11.9	3.9	
	0.25	5.2	16.7	18.8	4.62	5.5	2.6	19	20.2	
	0.50	7.4	7.8	18.7	17.73	6.17	5.5	22.1	14.1	
	0.75	6.75	3.8	28.2	4.3	3	26.8	12	14.4	
	1.00	4.2	4.87	11.2	31.9	6.16	4.2	23.1	15.2	9.23

229 On average, the isomer shifts (IS) of all sites increases with the increase in Zr content. The  
 230 increase in the isomer shift (IS) could be related to the volume expansion resulting from the Zr  
 231 substitution for Fe. The volume expansion lowers the *s*-electron density at the Fe nuclei which leads  
 232 to an increase in isomer shift (IS). At higher Zr content ( $x = 1.00$ ), a decrease in isomer shift (IS) was  
 233 observed. This decrease in the isomer shift may be due to the combined effect of volume expansion  
 234 and the increase in electron density. The decrease in hyperfine fields on  $Dy_2Fe_{16}Ga_{1-x}Zr_x$  is attributed  
 235 due to the decrease in magnetization because of the competition between the positive effect on  
 236 magnetic moments due to bond length and the negative effect caused by magnetic dilution of the  
 237 substituted non-magnetic atom. The average isomer shift (IS) of all sites increases with increasing Zr  
 238 content except the 12*k* site. The increase in the isomer shift (IS) could be related to the volume  
 239 expansion resulting from the Zr substitution for Ga. The volume expansion lowers the electron  
 240 density at the Fe nuclei which leads to an increase in isomer shift (IS).

#### 241 4. Conclusions

242 The effect of substitution of Ga and Zr in intermetallic compounds  $Dy_2Fe_{17}$  prepared via arc  
 243 melting are carefully studied. The intermetallic compounds are  $\alpha$ -Fe free and are found to crystallize  
 244 in hexagonal  $Th_2Ni_{17}$  type structure. Paramagnetic phase  $DyFe_3$  is observed for  $x=1$  and confirmed  
 245 using Mössbauer spectroscopy. From the Rietveld refinement, it is observed that most of Zr atoms  
 246 occupying 12*j* and 12*k* sites of Fe for  $Dy_2Fe_{17-x}Zr_x$  and  $Dy_2Fe_{16}Ga_{1-x}Zr_x$  intermetallic compound. Lattice  
 247 parameters *a*, *c* and unit cell volume are increasing with the substitution of Zr and Ga. The Fe-Fe bond  
 248 length increases with Zr substitution. However, the maximum change in Fe-Fe bond length is  
 249 observed for 6*g*-12*j*, 6*g*-12*k* and 12*j*-12*j* sites. This change in bond length found to be affected by the  
 250 strength of Fe-Fe exchange interaction. The maximum *T<sub>c</sub>* is observed in  $Dy_2Fe_{16}Ga_{0.5}Zr_{0.5}$  (505.1 K) for  
 251 double substitution of Ga and Zr however for single substitution, maximum *T<sub>c</sub>* is observed as 510K  
 252 in  $Dy_2Fe_{16.25}Zr_{0.75}$ . This indicates that double substitution of Zr and Ga does slow enhancement in *T<sub>c</sub>*  
 253 with the single substitution. Ga and Zr dopant and their content can significantly improve *T<sub>c</sub>* and *M<sub>s</sub>*  
 254 thus ensuring their potential use in high-temperature permanent magnet applications. The hyperfine  
 255 field values decrease for both  $Dy_2Fe_{17-x}Zr_x$  and  $Dy_2Fe_{16}Ga_{1-x}Zr_x$  with Zr doping due to the decrease in

256 the magnetic moment of Fe atoms. The isomer shift (IS) showed an increase in nature due to the  
257 increase in volume effect.

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