1 Article

2 Design and analysis of domestic micro-cogeneration

³ potential for an ORC system adapted to a Solar

4 **Domestic Hot Water System**

5 Daniel Leal-Chávez¹, Ricardo Beltrán-Chacón^{1*}, Paola Cardenas-Terrazas¹, Saúl Islas², Nicolás 6 Velázquez²

- ¹ Centro de Investigación en Materiales Avanzados, S.C., CIMAV, Miguel de Cervantes 120, Complejo
 ⁸ Industrial Chihuahua, Chih. C.P. 31136, México
- 9 ² Instituto de ingeniería, Universidad Autónoma de Baja California, Av. Álvaro Obregón S/N, Colonia
 10 Nueva, Mexicali, Baja California, C.P. 21100, México.

11 * Correspondence: ricardo.beltran@cimav.edu.mx; Tel.: +52 614 439 4884

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13 Abstract: This paper proposes the configuration of an Organic Rankine Cycle (ORC) coupled to a 14 solar domestic hot water system (SDHWS), with the purpose of analyzing the cogeneration 15 capacity of the system. A simulation of the SDHWS was conducted at different temperatures, 16 observing its performance to determine the amounts of useable heat generated by the solar 17 collector; thus, from an energy balance, the amount of heat that may be used by the ORC could be 18 determined. The working fluid that would be suitable for the temperatures and pressures given in 19 the system were selected. The best fluid for the given conditions of superheated vapor at 120 °C 20 and 604 kPa and a condensation temperature of 60 °C and 115 kPa was acetone. The main 21 parameters for the expander thermodynamic design that may be used in such ORC were obtained 22 with the possibility of generating 443 kWh of annual electric energy, with 6.65 % global efficiency 23 of solar to electric power, or an overall efficiency of the cogeneration system of 56.35 % with a solar 24 collector of 2.84 m².

- Keywords: Solar Domestic cogeneration, Organic Rankine Cycle, Acetone, Evacuated tube solarcollector
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28 1. Introduction

29 The growing demand of energy and the great dependence on fossil fuels require more efficient 30 processes and technologies that allow a transition into a system with sustainable energy sources.

In general terms, the objective is that buildings have zero net energy consumption through an efficient use of energy and the adoption of renewable sources of energy and other technologies [1]. Photovoltaic, solar, thermal and biomass technologies can contribute to reduce consumption for heating of spaces in the Mediterranean region [2]. Solar energy systems allow consumption to be reduced to almost zero net energy in buildings. For combined heat and power (CHP) systems, return periods are between 5.5-6.5 years, while for photovoltaic systems it takes 7 years [3].

The Organic Rankine Cycle technology has intensified in recent years and is being used as themain alternative to convert energy at low temperature in power [4].

Solar water heaters generally capture the greater amount of solar energy during seasons withlow hot water consumption, such as summer. In these circumstances, a lot of solar energy is wasted.

- 41 Extending the utilization of energy captured by solar collectors through a second application 42 would favor a technological alternative for electric production through renewable energies and a
- 43 reduction of CO₂ emissions associated with conventional fossil fuel-based generation techniques.
- 44 An option to improve the utilization factor of solar systems is integrating ORC systems to 45 generate electricity in a standard domestic use. The technical feasibility of a system of 2.9 kWe/19.51

46 kWth comprising commercial items showed return values of 3.8 years [5]. A 2 kWe/18kWth Organic
47 Rankine cycle system was developed to provide electricity and hot water in a residential building. It
48 was found that the system could provide 67 % of the energy required for hot water [6].

The Organic Rankine Cycle (ORC) is a proven technique to convert unused low temperature heat into useable mechanical or electric energy [7] and is characterized by its simplicity and flexibility at a relatively low operating temperature [8].

52 This technique has been widely used in recent years with different heat sources, such as: 53 geothermal systems, heat recovery, biomass and solar systems. There are many applications and 54 configurations at different temperatures and capacities [9]. In recent years, there have been many 55 studies on ORC, ranging from optimization, through component and working fluid selection, to 56 techno-economic and experimental studies [10].

57 Currently, there are a number of domestic-sized solar ORC systems [11–15]; their main 58 operation characteristics are shown in Table 1. However, none of them has shown a configuration 59 using direct vapor generation with evacuated tube collectors. These configurations generally use an 60 indirect heat exchanger between the solar heating system and the ORC system (see figure 1). The 61 aim is using the solar system only for pre-heating the working fluid and later to evaporate the fluid 62 with a conventional heat source under controlled conditions.



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Figure 1. a) Proposed Configuration, b) Typical Configuration.

Output Isentropic Global Author power Efficiency (%) Efficiency (%) Helvaci and Khan 2016 [11] 58.66 135.96 W 3.81 Taccani et al. 2016 [12] 670 We 62 8 Vittorini et al. 2018 [13] 700 W Not shown 3* J. Freeman et al. 2017 [14] 75* 7.3 ~ 1 kWe 60* 4.98 Cioccolanti, et.al 2017 [15] 2 kWe *assumed by the author

Table 1. Performance of different ORC.

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67 There are different studies in the field of optimization of residential solar ORC configurations. 68 A system with flat plate collectors with a heat source of 3.542 kW [11] uses a heat exchanger for 69 condensation and a subsystem for waste heating to heat the water in a hot water tank. A 70 configuration of 100 m² of a parabolic channel with ORC [12] uses a heat exchanger to transfer heat 71 from the parabolic channel and operate as an evaporator for the ORC. A configuration of an 72 evaporator SDHWS used as an intermediate heat exchanger to heat the working fluid of the ORC 73 system [13] additionally uses a system of indirect heating for the DHWS and 3 flat plate solar 74 thermal collectors. A configuration with a solar area of 15 m² and thermal energy storage [14] which 75 maintains and uses energy to mitigate solar energy variations in the ORC. A solar system with a 76 linear Fresnel collector with a total surface area of 148 m², a thermal storage with phase change and 77 an ORC [15] uses the tank indirectly for phase change as the evaporator from the ORC.

78 The proposed system presents a fundamental difference to other applications reported in the 79 literature in that the heat transfer is carried out directly from the evacuated tube solar collectors,

80 using it as an evaporator. This improves the cycle efficiency due to the elimination of heat losses in 81 the intermediate heat exchanger. When intermediate heat exchangers are added, each heat 82 exchanger requires a temperature difference to be effective. Therefore, each added heat exchanger 83 decreases the maximum source temperature. In other words, the heat transfer increases the 84 generation of entropy of the system, thus reducing the ability to produce work.

This configuration is simpler because it eliminates the need for using two separate systems (solar system and ORC system), thus avoiding the need of two pumps to recirculate the fluid in both systems.

88 The integration of an ORC system into a solar domestic hot water system (SDHWS) is presented 89 to achieve a domestic micro-cogeneration, taking into consideration the pressures and temperatures 90 at which these two systems may work properly. The energy of the fluid that does not produce work 91 in the expander is utilized in the hot water tank for simultaneous production of hot water and

92 electricity.

93 The use of a variable flow pump is proposed for the vapor generation to occur directly in the 94 solar collector, ensuring a constant vapor pressure and temperature at the collector outlet, while for 95 the condensation zone of the system, a heat exchanger integrated within the SDHWS hot water tank 96 will be used (see Figure 2).

97 There are a number of studies related to working fluid selection for the ORC [16–20]; however, 98 they generally focus on operation temperatures. In contrast, in this study were considered first the 99 (relatively low) operation pressures that the SDHWS may withstand, and later their influence on the 100 fluid properties and the thermal performance system were evaluated.

For the basic engineering design of this system is proposed a dimensioning method, including (A) an analysis of the thermal energy potential available in the SDHWS for a secondary use (ORC); (B) identification of operation conditions (the most adequate temperature and mass flow for an efficient operation of the SDHWS and ORC integrated system); and (C) the identification of working fluids that may allow obtaining a better performance under these conditions.

For this proposal was conducted a year-long simulation in TRNSYS of a conventional solar hot water system with an evacuated tube collector. The results of the simulation of the main components in the system were validated with experimental data available in the literature. In this analysis were considered the year-long climatic characteristics, the hourly solar energy utilization in the collector and the hot water needs for an average-sized household.

111 The results of the simulation in TRNSYS show the energy flows and their destination, 112 classifying them into thermal losses, useable heat for hot water and potential useable (unutilized) 113 heat.

114 From this study, it is possible to analyze the system performance when working at different 115 temperature levels and to estimate the cogeneration potential.

116 2. Description of the System

A cogeneration system is proposed for integration into solar water heating systems [21] as shown in Figure 2 and Table 2. It comprises a solar collector (evaporator) (2) which is interconnected to a two-position, three-way valve (9). In one position, the valve (9) allows the working fluid to flow towards the condenser (5), and in the other valve position the working fluid is directed towards an expansion device (3). The working fluid that comes out of the expansion device (3) is directed towards the condenser (5). The condenser (5) is a heat exchanger submerged inside the hot water tank (4).

124 The system uses an auxiliary heat exchanger (7), in which a two-position, three-way valve (8) 125 directs the working fluid towards such heat exchanger when heat removal in the condenser (5) is not 126 sufficient. When the system changes from heating mode to ORC mode, the fluid pump (1) reduces

127 the flow, causing part of the working fluid to be stored in a receptor tank (6).



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Figure 2. Scheme of ORC adapted to SDHWS.

130 **Table 2.** Basic specifications of the system components.

Solar collector		Cooling coil	
Aperture area	2.84 m ²	Length	18.6 m
Fluid capacity	0.7901	Area	0.929 m ²
Flow rate	0.0333 l/s	Coil material	cooper
Max flow rate	0.25 l/s	Coil internal diameter	1.27 cm
Max operating pressure	800 kPa	Coil external diameter	1.59 cm
Hot water tank		Expander	
Capacity	2701	Max. Output Power	245W
Height	1.5 m	Nominal expander power	194 W
Internal diameter	0.45 m	Nominal Volumetric flow	850 l/h
Insulation thickness	0.047 m	Expansion ratio	4.85
Insulation conductivity	0.036 W/mK	Isentropic efficiency	0.85
Pump		Design point of the ORC	
Required power for ORC	1.5 W	Irradiance	950 W/m ²
Flow rate	15.4 l/h	Collector thermal efficiency	62.50 %
Input pressure	115 kPa	Output collector power	1687 W
Output pressure	604 kPa	04 kPa Expander output 194 W	
Fluid temperature	60 °C	Global efficiency	7.20 %

131 One characteristic of the proposed system is that it may work at two operation modes; either

132 water heating only or combined heat and power production, depending on the requirements.

133 2.1. Operation Mode: Water Heating

134 The system configuration for exclusive production of hot water requires that the working fluid 135 at the collector output is directed towards the heat exchanger inside the hot water tank. Through the

136 heat exchanger/condenser, the working fluid transfers heat indirectly (without direct contact) to the

137 water in the hot water tank. To achieve an efficient operation, the working fluid mass flow is 138 controlled so that the working fluid temperature at the collector output is higher than the preset 139 average tank water temperature. The working fluid which leaves the condenser out of the hot water 140 tank is then directed towards the fluid pump and then to the evaporator inlet. This cycle is operated 141 as long as the working fluid temperature at the solar collector output is higher than the water 142 temperature in the hot water tank, or if the preset temperature (60 °C) is reached. In this mode of 143 operation, the expander is not used, the fluid instead flows directly from the collector to the 144 submerged heat exchanger.

In this mode of operation, there is no phase change of the working fluid within the cycle, so the pump only circulates the working fluid, and thus it does not impose a high pressure change across the cycle.

148 2.2. Operation Mode: Organic Rankine Cycle

149 The contribution of thermal energy to the water in the hot water tank when there is low hot 150 water demand causes a progressive increase of water temperature in the hot water tank. Once the 151 hot water tank temperature reaches a preset operation temperature (60 °C), the valve (9) (Figure 2) 152 changes position and directs the working fluid towards the volumetric expander. Simultaneously, 153 the pump turns off momentarily to drain the working fluid partially into the receptor tank (6) which 154 stores the unused working fluid when working in the Rankine cycle. After this, the pump starts to 155 operate with a mass flow that allows producing superheated vapor (at 120 °C) at the evaporator 156 outlet.

Variations in the solar resource and the useable heat produced require that the flow is regulated depending on the solar irradiance to produce steadily superheated vapor at the evaporator outlet. The expander may start working at the moment when there is a pressure differential between the hot side (collector outlet) and the cold side (hot water tank and/or condenser) of the system and convert some useful heat into useable mechanical work. This way, some of the solar energy captured by the solar collector which is not required to heat water (once it has reached a preset temperature) is used to produce mechanical work.

164 The working fluid at high temperature at the expander output is used to heat the water in the 165 hot water tank when the conditions of the hot water tank temperature allow it, thus achieving a 166 cogeneration of electrical and thermal energy.

167 As heat is transferred into the hot water tank, its temperature increases, and the temperature 168 difference between the working fluid and the hot water tank decreases; this eventually reduces the 169 condenser cooling capacity to remove the necessary energy to condense the working fluid. Under 170 these circumstances, it is necessary to have an auxiliary heat exchanger that allows the removal of 171 the necessary energy to condensate the working fluid before it reaches the fluid pump. To achieve 172 this energy removal, the valve (8) changes position so it directs the working fluid towards the 173 auxiliary heat exchanger where it is subcooled and stored in the receptor tank, completing the 174 Rankine cycle.

When all the energy provided by the collector is required for water heating, or when there is not enough irradiance for working fluid evaporation, the valve (9) changes position so the fluid flows directly towards the condenser in the hot water tank, thus homogenizing pressure in the whole system.

Heat transfer is conducted directly from the solar collector towards the ORC working fluid, andthe fluid energy not extracted by the expander is conducted towards the hot water tank for thesimultaneous production of hot water and electricity.

182 **3.** Thermodynamic Model of the SDHWS

183 The SDHWS was analyzed in TRNSYS, describing a conventional evacuated tube collector solar184 hot water system.

- 185 TRNSYS is a software environment used to conduct dynamic simulations of the performance of
- 186 systems in transient state (i. e. solar energy applications, HVAC, building thermal analysis, etc.).

187 These simulations allow predicting the performance of a system, by analyzing performance under 188 dynamic conditions, allowing a proper system sizing. Table 3 shows the principal types used and 189 Figure 3 shows the scheme used for the study of TRNSYS simulation.

190	Fable 3 Description of principal SDHWS components in TRNSVS [22]
170	Table 5. Description of principal 3D1103 Components in TRINS13 [22]

Туре	Component	Description
		This component models the thermal performance of an
71 Ev	Erroguotod Tubo Color	evacuated tube collector type using theory, defining the
	Collector	efficiency of an evacuated tube by the Solar Ratings and
	Collector	Certification Commission (SRCC) and accounting for the
		incidence angle modifiers (IAMs)
	Storage tank; Fixed	Models a stratified liquid storage tank (vertically cylindrical tank
60	Inlets, Non-Uniform	with one inlet and one outlet flow). This includes the internal
	Losses	heat exchanger.
3	Pump	This pump model computes a mass flow rate using a variable
	rump	control function.
		The on/off differential controller generates a control function
	ON/OFF Differential	with a value either 1 or 0. The control signal value is chosen as a
2	Controller	function of the difference between upper and lower
	Controller	temperatures Th and Tl, compared with two dead band
		temperature differences DTl and DT2.
31	Pipe or Duct	This component models the thermal behavior of fluid flow in a
51	Tipe of Duct	pipe or duct using variable size segments of fluid.
109	Data Reader and	This component reads weather data at regular time intervals
107	Radiation Processor	from a data file.
		Type11 model uses a tee piece in which two inlet liquid streams
11	Tee Piece	are mixed together into a single liquid outlet stream, which is
		subject to external control.
11	Tempering Valve	This instance of the Type11 model uses a temperature controlled
11		liquid flow diverter.
	Time Dependent	This function is used to generate a water consumption profile
14	Forcing Function	that is repeated day to day. This includes water usage in specific
		amounts and times according to typical consumption [23].







Figure 3. SDHWS scheme in TRNSYS simulation studio.

193 The weather data location used in this study was Chihuahua, Mexico with a 1/64 hour time 194 interval for the simulation in order to obtain consistent results.

195 The objective of this analysis is characterizing the system performance at different operation 196 temperatures and to identify the parameters that influence heat gain. The model refers only to the 197 operation of the SDHWS, without considering the operation within the ORC. The focus of this 198 simulation is to analyze the collector useful energy and the domestic hot water energy requirements 199 in order to estimate the energy that could be used within the ORC system.

200 The results allow for a quantification of the heat transfer rate, the extra heat that could be 201 harnessed for a secondary use, the mass flows and physical conditions of working fluids that will be 202 used in the system.

203 The collector converts solar irradiance into useable heat and transfers it to the working fluid 204 that flows through the collector manifold towards the cooling coil submerged in the hot water tank 205 where it heats water for domestic use. The pump keeps working until the top of the hot water tank 206 reaches an operation temperature of ~75 °C as recommended by the manufacturer [24]. If the hot 207 water does not reach the desired temperature upon its utilization, it may be heated by an auxiliary 208 conventional heater.

209 The operation and simulation conditions for the main components (i.e. solar collector, hot water 210 tank, and internal heat exchanger) were validated with experimental data available in the literature 211 [27-29].

212 3.1 Solar Collector

213 Solar collectors are a special type of heat exchangers that transform solar radiation into useable 214 heat which is transferred into a fluid that circulates through such collector [26].

215 The results obtained in the field tests conducted by Kottwitza [25] were considered for the 216 validation of the mathematical model used in this study, while Q_{model} shows the results obtained

217 through the model used by type 71, as shown in Figure 4.





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Figure 4. Comparison of experimental tests vs numerical model of the solar collector.

As it may be seen in Figure 4, there are very close approximations to the values obtained in the collector experimental data, with a maximum error of 1.44 % at the outlet power.

222 3.2. Hot Water Tank

The hot water tank is an important component of a solar hot water system since it may considerably improve its efficiency and profitability, it allows a better use of the solar system, and equates the useful heat of the solar resource and energy demand [27]. Solar energy is a time-dependent energy resource just like the energy requirement, but both occur at times different than the solar energy supply; therefore, energy storage is necessary if solar energy is to be used to meet a major part of the energy needs.

229 For the validation of the hot water tank model and submerged cooling coil, the experimental 230 data in the literature were compared to data obtained through simulation. The results obtained by 231 [27] who used a commercial 270 l hot water tank were considered. This tank uses fiberglass 232 insulation with a 0.047 m thickness, and a 0.036 W/mK thermal conductivity. The hot water tank 233 has a 1.5 m height and 0.45 m internal diameter. In such experiment, the water was heated and 234 mixed at a temperature of 54.0 °C in an environment of 20 ± 0.03 °C during 48 hours. Readings were 235 carried out every 3 min, with temperature sensors spaced every 15 cm in a vertical array to obtain 236 stratification readings. The heat loss coefficients for each node are considered for the validation of 237 this model, as shown in the Table 4.

238	Fable 4 . U coefficient weighing by nodes in the hot water tank

Cool down test	U area-weighted average value (W/m2K)
Node 1	0.66
Node 2	1.13
Node 3	1.13
Node 4	1.13
Node 5	1.11
Node 6	0.95
Node 7	0.79
Node 8	1.16
Node 9	2.54

From the data of the experimental cool down test conducted by [27], a loss of 14,580 kJ was obtained during the 48 hours. The result of the simulation of type 60 was 14,835 kJ with an error of 1.74 %, which shows that the mathematical model of the hot water tank is very accurate.

From the same loss calculation taking a single average loss coefficient from the insulation characteristics, $U = 0.86 \text{ W/m}^2\text{K}$, a significantly reduced heat loss is obtained, with a 19% difference with respect of the experiment.

245 3.3. Cooling Coil

The hot water tank sub-system for water heating in this study consists of a cooling coil submerged in the storage tank, where the high temperature working fluid coming from the solar collector exchanges heat with the water in the hot water tank.

The data presented by R. Farrington [28] were used for the validation of the cooling coil. This cooling coil has a total length of 18.6 m and an area of 0.929 m²; it is made of plain copper tube with an internal diameter of 1.27 cm and an external diameter of 1.59 cm. The dimensions and positions of the coil in the storage tank are shown in Figure 5.



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Figure 5. Scheme for heat exchanger submerged in the hot water tank [28].

The experimental data presented by R. Farrington [28] were taken into account to validate the heat exchanger simulation.



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Figure 6. Comparison of experimental data vs TRNSYS of submerged cooling coil.

The curves displayed by the model developed in TRNSYS by type 60 show a very good adjustment in respect to the experimental data obtained by [28], with an average difference of 1.6 % between the three models (Figure 6).

262 4. Energy Balance

It is a common practice to use computer simulation to predict the performance of a solar domestic hot water system. This process is based on the precise specification of the system characteristics, with the purpose of integrating all these sub-systems and to analyze the behavior that would be obtained through the interaction of all these components in a time-variable transitory state according to the weather changes in the site.

268 4.1 Operation of the Solar SDHWS

The typical meteorological year (TMY) of Chihuahua, Mexico was considered for the simulation of the SDHWS; and the solar energy captured by the collector and the hot water needs were assumed for an average-sized household in Mexico.

The required heat for the year-long supply of hot water is determined after analyzing the requirements, as well as the main devices in the system and the collector performance. In this study are considered the requirements of an average-sized household in Mexico of 3.7 (~4) members

according to [29]. A daily average hot water consumption of 236 l is considered (Table 5), at an

average temperature of 45 °C [30].

277 **Table 5.** Typical Residential Hot Water Consumption [30].

Usage	Amount of hot water (1)	Daily Total Consumption per Household (l)
1 x Handwashing	3	(12 Uses) 36
1 x Shower	35	(4 Uses) 140
1 x Dishwashing	20	(2 Uses) 40
Cleaning	3 per person per day	12
Cooking	2 per person per day	8

Considering table 5, the hourly consumption is adjusted to the typical household consumption reported by ASHRAE [23]. A daily consumption of 236 l at 45 °C was considered, representing per

280 year 86,140 l and 4,464 kWh of thermal energy. The use of this amount of hot water will be

281 distributed as shown in Figure 7 by the use of type 14 on TRNSYS





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Three energy inputs were considered to obtain this required heat in addition to the heat lossesin the system: the solar collector, the auxiliary system and the network water supply.

In the system, using the type 31, a piping section from the hot water tank to the tap of 10 m and 1.27 cm in diameter was considered, located outdoors with a thermal insulation of 13 mm. For the connection between the solar collector and the hot water tank, 5 m piping with a 19 mm thickness was used. For the thermal conductivity of the insulated copper tube, the data reported by Marini et al. [31] of k = 0.031 W/mK was used. The results obtained show values similar to those reported by ASHRAE [23] of UA = 0.346 W/mK, U = 8.48 W/m²K based on the internal diameter of the pipe.

The system should maintain the water in the hot water tank at a minimum temperature of 60 °C to avoid the growth of bacteria such as Legionella [23,30]. Therefore, there will be heat losses at the hot water tank which also should be provided by the water heating system. These temperatures may vary depending on the internal temperature of the hot water tank and the ambient temperature.

Taking these data into consideration, as well as the behavior of the types of the components described above, the simulation was carried out in TRNSYS obtaining the energy balance shown in Figure 8. The energy inputs for each inlet as well as the heat losses for each part of the system can be identified.



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Figure 8. Energy balance in the domestic hot water system.

Under these conditions, the solar collector and the water supply input at ambient temperature provides 96 % of the required energy, with a total energy of 5,673 kWh, whereas the auxiliary heater must provide the remaining 4 % with 221 kWh to meet the hot water requirement. There are partial losses through the environment of 1,858 kWh, taking into consideration the losses of the solar collector, the hot water tank, and the supply pipes.

However, the solar collector may supply a greater amount of energy if the re-circulation pump
 is kept operating during sunny hours, that is, not turning off the recirculation pump when the water
 reaches the set temperature point. This may be attractive if this useable heat is used in another

310 process, such as an ORC.

311 4.2 Collector Performance at Different Temperatures

312 As the temperature difference between the working fluid and the ambient temperature 313 increases, the collector efficiency decreases. However, there is a higher gain of enthalpy in the 314 working fluid through a lower volumetric flow.

Taking into consideration that one of the objectives of this study is to identify the thermal potential of a solar collector for a secondary use additional to water heating, it is necessary to analyze the collector thermal performance at different mass flows and temperatures. Identifying

these conditions will allow an estimation of the different working fluids performance under these circumstances. For cogeneration applications, these conditions involve superheated vapor at the

320 evaporator (collector) and subcooled liquid at the condenser.

For a higher efficiency in the ORC is required a higher pressure difference; therefore, the aim is to have the highest permissible pressure in the system. In this case of study, a maximum permissible pressure of 800 kPa is considered in the collector with a stagnation temperature of 228 °C according to the manufacturer data sheet [24]. In the condenser the objective is to condense with the hot water tank operation temperature at a relatively low pressure. This way, different scenarios may be analyzed considering different operation temperatures in order to select the best operating condition.

The condensation temperature parameter has a direct impact on the system performance. Therefore, when defining the condensation temperature, it should be considered that setting a temperature lower than 60 °C implies that certain amount of energy should be removed from the system through an external heat exchanger. These may compromise the amount of useable energy for heating water and increase the need for auxiliary heating.

333 On the other hand, if there is an exceedingly high temperature, the heat losses in the piping and 334 hot water tank may increase, decreasing the collector efficiency. Thus, it is necessary to find a point 335 where the need for the required useable energy may be met and at the same time operate at 336 temperatures and pressures that allow working efficiently in an ORC.

Figure 9 shows the thermal power delivered by the collector at different mass flows and inlet temperatures. As the inlet temperature in the collector increases, it transfers less energy to the fluid due to increasing heat losses. It also shows that the heat transfer rate drops when increasing the outlet temperature and decreases exponentially when flow rates are less than 1 kg/s. However, in order to have good efficiency within the Rankine cycle, it is necessary to have the greatest possible temperature difference. Under these conditions the best point of operation is found when the collector is able to meet the needs for hot water and is obtained the greatest amount of useable heat.



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Figure 9. Solar collector performance at different mass flow rates and input temperatures.

In Figure 9 it may be noted that when operating at lower input temperatures is obtained a higher thermal power. For the Rankine cycle, there is an attempt to obtain the lowest temperature cycle, to keep the heat losses as low as possible, so the analysis is focused on inlet temperatures of 40, 50 and 60 °C.

When requiring a constant temperature and operating continuously in a transitory state, it is necessary to have a control system for its regulation. In order to do this is considered a variable flow pump, which is able to control the system optimally. The considered expander must work with a variable mass flow as well, where the expander would have a constant pressure and temperature supply, and by having the expander of a fixed built-in volume ratio, the angular speed would increase or decrease depending on operating conditions.

For the collector simulation, the input fluid was evaluated at different initial temperatures. A variable flow pump with a feedback controller was considered to obtain a constant output temperature at different irradiances.

Figure 10 shows schematically the amount of energy delivered by the cooling coil under normaloperation conditions (energy from the solar collector) and the energy that the collector could give at

361 different inlet temperatures, maintaining a 60 °C temperature increase, and the extra heat that could

362 be harnessed for a secondary use in each case.







Figure 10. Useable heat during the year at different temperatures.

Figure 11 shows the thermal energy produced annually with each established temperature difference and the required mass flow.





Figure 11. Energy produced at different operation temperatures.

369 **5. ORC Efficiencies**

To select the operation temperatures in the cycle, it is necessary to analyze the performance of the heat source (solar collector), as well as the ORC when operating at such temperatures. It is desired a point of operation that allows the solar collector to have enough efficiency to supply the energy required by the SDHWS and for the ORC to have a good efficiency.

The collector efficiency decreases approximately 8 % when increasing the outlet temperature in the system by 60 °C. In contrast, the Carnot efficiency increases to 12 % for the same increase in temperature (Figure 12).





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Figure 12. Efficiencies at different operation temperatures.

To condense at a temperature lower than 60 °C in the Rankine cycle, it is necessary to remove energy from the system, which implies a waste of the solar energy captured by the collector; therefore, the total system efficiency decreases. Condensing at temperatures ≥ 60 °C allows the utilization of a greater amount of heat for the generation of electric power without compromising the energy required to keep the water at 60 °C.



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Figure 13. Energy and efficiency of the cycle at different operation temperatures.

With an input temperature of 60 °C and an output temperature of 120 °C allows obtaining a useful energy of 3,960 kWh and a Carnot efficiency of around 15 % (Figure 13).

388 Type 71 simulates solar collectors for heating a working fluid without phase change. Collector 389 thermal efficiency and thermal losses depend mainly on the fluid average temperature. The energy 390 balance in the solar collector when operating in ORC mode, where phase change occurs, is 391 performed considering the latent heat of vaporization.

When analyzing solar collectors with phase change, the filling factor is considered. The filling factor is the ratio of volume occupied by liquid with respect to the total volume of the evaporator. A filling factor near to one indicates that most of the volume is occupied by liquid. A filling factor close to zero indicates that most of the volume is occupied by vapor, which can involve overheating of the fluid and increase the average temperature affecting the calculation of thermal efficiency.

397 An experimental study [29] conducted with a filling factor of 55 % shows a decrease in the 398 thermal efficiency of only 4.5 % compared to a collector without phase change. This shows that the 399 thermal efficiency of the solar collector depends mainly on the average temperature even when the 400 heat transfer coefficients with phase change are very different.

In this study, the vapor superheated temperature (120 °C) is close to saturation temperature
(119.6 °C). Therefore, it is considered that the variation of mean temperature over thermal efficiency
is negligible. Thus, several authors have used different configurations to use collectors as
evaporators, obtaining good efficiencies [32–34].

405 5.1. Working Fluid Selection

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406 A great number of organic fluids with a boiling temperature lower than water may be used to 407 operate with low temperatures and pressures in an ORC. A proper selection of the working fluid in 408 an ORC is extremely important since it will not only affect the cycle efficiency, but also the size of 409 components, the expander design and the system stability.

Firstly, for a proper selection of the working fluid, a practical pressure and temperature are defined within the cycle limits. A high pressure ratio favors the cycle efficiency. In this study, the high pressure is limited by the solar collector maximum operating pressure. It is important to also consider the isentropic efficiencies that would be obtained in the pump and the expander when operating with such fluid.

415 Another consideration is that when working near the fluid critical pressure, any change of 416 temperature means great changes in pressure, which makes for an unstable system [20], therefore, 417 when selecting the working fluid, it is important to work at a certain distance from the critical point.

418 In a study conducted by Qiu G. [19], it is mentioned that the main criteria for the proper 419 selection of the working fluid, in decreasing order of relevance, are the following:

- Not banned by relevant national standards.
 - High enthalpy change in the expander; this translates into a higher work output.
- Working fluids should be easy to manage at ambient temperature, expecting the ambient temperature not to be higher than 100 °C.
 - Isentropic or "dry" fluids are more adequate to avoid superheating in the cycle.
 - High latent heat; this makes components in the system smaller.
- 426
 427
 Preferential physical properties, such as: chemical stability, high thermal conductivity, low viscosity and operating at low pressures.
 - High safety requirements such as low toxicity, low corrosion, non-flammable and non-explosive.
 - Accessibility and low cost.

Through the EES software [35], 29 different organic fluids were analyzed, chosen from Bao et al, Qui and Rayegan et al. [18–20]. It was found that only 16 may be superheated at 120 °C and 800 kPa and subcooled at 60 °C and 100 kPa (Figure 14). Also 13 fluids (not shown) could not be used in the cycle because the saturation pressures obtained at the required temperatures were not feasible in the SDHWS.





436

437

Figure 14. Saturation pressure of working fluids for the ORC conditions.

438 5.1.1. Enthalpy Change

The main characteristic that must be taken into account is a high enthalpy change in the expander, because this means that most of vapor enthalpy may be transformed into useable mechanical work. Therefore, the isentropic enthalpy change in the cycle is evaluated with the set temperatures and a pressure not higher than 800 kPa.

For fluids with a saturation pressure higher than 800 kPa at 120 °C, the enthalpy change is considered limited at this pressure (shown in orange in Figures 14-17). A similar consideration is made for the cooling down conditions at 60 °C, in order to maintain the operation conditions within the permissible range.

The power produced by the expander is the product of enthalpy change and the mass flow.
Therefore, by having a greater enthalpy change, a smaller device (or lower volumetric flow rate) is
required.

450 Figure 15 shows the difference of enthalpy for the different fluids analyzed. It is observed that

451 the 5 fluids with a higher enthalpy difference between selected temperatures and pressures are:

452 Acetone, Benzene, Toluene, Cyclohexane, and Isohexane.



453

454

Figure 15. Enthalpy change of working fluids (at a condensation temperature of 60 °C).

455 5.1.2. Thermal Efficiency

Another important point to consider for the proper fluid selection is the thermal efficiency of the ideal cycle; this is the ratio between the enthalpy change at the expander, and the increase of enthalpy at the cycle, as shown in equation 1.

$$\eta_{ORC,ideal} = \frac{h_3 - h_4}{h_3 - h_1}$$
(1)

459 For a particular saturation temperature difference between the evaporation and condensation 460 states, there will be an enthalpy change. According to equation 1, the thermal efficiency depends not 461 only on the enthalpy change during the expansion but also on the enthalpy change required for 462 evaporation, i.e. the latent heat of vaporization. For each saturation temperature, different working 463 fluids have different latent heat of vaporization due to their intrinsic properties. Therefore, two 464 working fluids can present the same enthalpy change for a given temperature difference, but require 465 a different latent heat of vaporization. Under these conditions, the fluid with a bigger latent heat of 466 vaporization will present a lower thermal efficiency.

467 The thermal efficiencies of the analyzed fluids in Figure 16 show that the 5 fluids with a higher 468 enthalpy change are those with a higher thermal efficiency. It is important to note that not always

the fluids with a higher enthalpy change during the expansion will have a higher thermal efficiency.

470 Some examples of this are Isopentane, Neopentane, and R123, just to mention a few.



471

472 **Figure 16.** Thermal efficiency of different working fluids (at a condensation temperature of 60 °C).

473 5.1.3. Expansion Ratio

474 For the dimensioning of the expander, it is necessary to know the expansion ratio of the fluid. It

475 may be observed that the fluids with a higher expansion ratio are those with a higher enthalpy

476 change in the cycle, but not in the same order (Figure 17).



477



479 It may be observed that there is no direct relation between the fluid properties and the thermal 480 performance parameters; therefore, it is important to take note of them to make the right choice. 481 From the expansion ratio, it is possible to choose the type of expander that could be used within the 482 cycle. In the next section, the performance of the cycle with the selected working fluid is shown, as 483 well as other parameters required for the selection and suitable design of the expansion device. The 484 selected expander type is a volumetric expansion device because it is more appropriate for the ORC 485 at small capacities [36] due to its required lower flow rates, higher pressure ratios and rotational 486 speeds which are lower in comparison to velocity expander type. The basic expander selection is 487 described in section 6.

488 5.2. ORC with Acetone as Working Fluid

489 At low pressures, the best candidates were hydrocarbons, due to a higher isentropic enthalpy 490 change. Therefore, the pressure in the cycle may be more relevant than the operation temperature at 491 the moment of selecting a working fluid for an ORC.

Acetone was selected as the most suitable fluid to operate in the cycle with the given conditions.
In Figure 18, the Rankine cycle is shown, indicating that the saturation pressure is 604 kPa at 120 °C
and 115 kPa at 60 °C at the expander output.

Some studies have used acetone as working fluid in heat pipe collectors, getting better
performance than with other working fluids [37–41]. These results confirm that acetone is an
adequate working fluid to operate with phase changes in solar collectors.



498

499

Figure 18. Rankine cycle with acetone as working fluid.

500 Among the advantages of using acetone in a SDHWS are: it prevents freezing, scaling, fouling 501 and corrosion. This fluid has a low GWP (Global Warming Potential), which is a widely used 502 parameter to show how much heat is trapped in the atmosphere by a greenhouse gas, being 0.5 503 which is considerably low when compared to the use of typical refrigerants (i.e. R134a GWP = 1300; 504 R410a GWP =1700). In addition to this, acetone is considered not to have Ozone Depletion Potential 505 (ODP), therefore it is a great candidate for this type of applications [42]. The auto-ignition 506 temperature for acetone is 465 °C [43], since this temperature cannot be reached within the cycle, the 507 flammability or explosiveness of the fluid is avoided. Flammability is also prevented because open 508 flames and sparks are not present in the cycle under typical operating conditions.

Acetone has been experimentally used as a working fluid [1,28,42,44,45]. Other procedures to select working fluids for ORC systems indicate that the acetone achieves a high performance for ORC at 130 °C [20].

512 5.3. Analysis of Energy Potential Delivered to the Expander

To analyze the cogeneration system, the available energy generated by the solar collector was considered (after meeting the energy requirements of the hot water tank at the set point temperature of 60 °C), as well as the conditions in which this energy was given (irradiance, time, ambient temperature, etc.). In the simulation study, the working fluid was used within type 71, considering the enthalpy required to take the working fluid from subcooled liquid at 60 °C to superheated steam at 120 °C and 604 kPa in order to obtain the mass flows required on the process. The energy generated by the expander is calculated by:

$$E_{elect} = \sum_{i=1}^{n} (h_4 - h_3) \dot{m}_i \cdot \Delta t_i \cdot \eta_{exp}$$
⁽²⁾

520 The electrical energy generated by the system is calculated considering *n* as the number of 521 intervals (when it is possible to use the cogeneration system annually), and Δt the time step where 522 the mass flow is calculated, taking a 1/64 hour time interval.

523 Once the working fluid has been selected and the heat required to change the state from 524 subcooled liquid at 60 °C to superheated vapor at 120 °C has been defined, the required mass flows 525 are determined through an energy balance. Consequently, it is possible to calculate the volumetric 526 flow that would be obtained under the different operation conditions as shown in Figure 19. With 527 these parameters defined, it is possible to select the most adequate expansion device.







Figure 19. Thermal power delivered by the solar collector at different flow rates.

530 The collector simulation allows for an estimation of the thermal power, the average useable heat

and the irradiance levels at which the higher energy (2,346 kWh/m²) and power that can be obtained

532 (Figure 20), in order to optimize the efficiency of the system for these conditions.







Figure 20. Power and energy given at different irradiance levels during the year.

535 In table 6, the main parameters required for the expander thermodynamic design are listed.

536 **Table 6.** Main parameters for the expander design.

Working Fluid: Acetone			
	Inlet	Outlet	Difference
Pressure	604 kPa	115 kPa	$P_1/P_2 = 5.25$
Temperature	120 °C	60 °C	$\Delta T = 60 \ ^{\circ}C$
Enthalpy	892 kJ/kg	815 kJ/kg	∆h = 77 kJ/kg
Viscosity	10.18x10 ⁻⁶ Pa·s	23.12x10 ⁻⁵ Pa·s	$\Delta \mu = 22.x10^{-5} \text{ Pa} \cdot \text{s}$
Density	12.68 kg/m ³	2.612 kg/m ³	$Q_1/Q_2 = 4.85$
Operation Range			

Volumetric Flow	V [*] min=180 l/h	V [.] max = 900 l/h	Vinominal = 850 l/h
Inlet Heat	Q_{min} = 310 W	Q _{max} = 1830 W	Q _{nominal} = 1690 W

537 6. Expander

Once the ORC configuration (Figure 2) and the system operation conditions (Table 6) have been
established and the working fluid has been defined, it is possible to select the best suited expander to
work within the cycle.

541 The expansion device is one of the 4 main components of an ORC system with the most 542 influence over the cycle performance and the total system cost. Adequate selection of expander 543 depends on the operation conditions, working fluid, size of the system, mass flow, chemical 544 composition of the fluid, temperature difference and pressure allowed within the system [46].

545 An expander is a type of rotary mechanism which converts the kinetic energy and enthalpy of a 546 fluid into mechanical energy, directing the flow to the rotors of this device. The expansion device is 547 essential in the performance of an ORC system, and there are two main types of expansion devices, 548 velocity type: axial and radial turbines; volumetric type (positive displacement): piston expander, 549 screw expander, scroll, rotating vane and gerotor (trochoid). Velocity types are generally used in 550 systems of high scale above 50 kW. However, in recent years, they have been used in small-scale 551 ORC systems, but these have not performed adequately because its rotational speed increases 552 exponentially, which causes a decrease in the power output, in addition to expensive cost and low 553 commercial accessibility [47]. On the other hand, the volume expanders are more appropriate for the 554 ORC of small scale because they generally operate with low rate flows, higher pressure ratios and 555 lower rotational speeds than the velocity type. In addition, volume expanders can operate with 556 two-phase working fluid since this situation that can occur at the end of the expansion in certain 557 working conditions which makes them very attractive for this type of systems. They also have a high 558 reliability, compact structure, few moving parts, low noise level and low vibration [48].

Figure 21 shows the compilation of different types of expanders that have been used in different power ranges for micro-ORC [49]. The number of studies considered by the authors is shown in parentheses. It may be observed that the more adequate volumetric expansion type for the power considered in this study is the "trochoid" (gerotor) type. One of the advantages of this type of devices is that due to their configuration, there is a low relative speed between the rotor and stator, which generates lower friction losses with respect to other type of positive displacement devices [50].





Figure 21. Power Range Produced by Different Expansion Devices [49].

568 Currently, there are limited literature that reports this type of device used as an expander for an 569 ORC, however, an ORC system has shown isentropic efficiency of 85 % and a global efficiency of 7.7 570 % with a source temperature of 162.2 °C [50]. The expander isentropic efficiency $\eta_{exp} = 85$ % was 571 selected from these results. The maximum capacity of electric power generation of the proposed 572 system is estimated considering the constant efficiency. However, it is important to note that the 573 expansion isentropic efficiency can vary with time and conditions

574 In the analysis of the ideal Rankine cycle, the pump and the expander are considered isentropic 575 while the evaporator and condenser do not involve any work. The power of the pump is defined as:

$$W_{pump,in} = \dot{m}(h_2 - h_1) \tag{3}$$

576 And the power of the expander is:

$$W_{exp,out} = \dot{m}(h_3 - h_4) \tag{4}$$

577 However, one of the major deviations within the ideal Rankine cycle are the irreversibilities 578 occurring within the pump and the expander. To obtain more accurate results, it is necessary to take 579 into consideration the work involved within the cycle and also the isentropic efficiencies.

580 For this particular study, was considered a pump efficiency, $\eta_{pump} = 65 \%$ [11,14,15,51,52]. The 581 power required by the pump can be calculated by:

$$W_{pump} = \frac{\Delta h_{2-1} \cdot \dot{m}}{\eta_{pump}} \tag{5}$$

582 The expander would be considered to achieve 85 %, so the power of the expander can be 583 calculated by:

$$W_{exp} = \Delta h_{4-3} \cdot \dot{m} \cdot \eta_{exp} \tag{6}$$

584 When considering equation 5 and 6, it is possible to calculate the amount of power generated by 585 the expander as well as the power required by the pump to maintain the flow of the working fluid in 586 the cycle. To obtain the maximum power of the expander is considered the maximum flow from 587 Figure 19 in equation 6, which gives a value of 245 W.

588 The efficiency of the ORC system can be calculated by considering these isentropic efficiencies, 589 the constant temperatures and pressures in the cycle, obtaining 84.4 % of the ideal ORC, as given 590 below:

$$\eta_{ORC} = \frac{W_{exp} - W_{pump}}{\dot{m}(h_4 - h_3)} \tag{7}$$

591 The efficiency from solar to electric power can be obtained as follows:

$$\eta_{solar} = \frac{\mathbf{E}_{elect}}{\dot{Q}_u} \tag{8}$$

592 The CHP efficiency is given by:

$$\eta_{CHP} = \frac{\dot{W}_{net} + \dot{W}_{heat}}{\dot{Q}_u} \tag{9}$$

The proposed system has 2,346 kWh/m² annual irradiation, and a 2.84 m² collector surface area; it is possible to generate a useful electric energy of 443 kWh, with a global efficiency of solar to electric power of 6.65 % (equation 8). Also, the overall efficiency of the cogeneration system is 56.35 % (equation 9) when considering CHP generation

597 7. Conclusions

598 In this paper the application of an ORC to a SDHWS was presented and analyzed with the 599 purpose of observing the cycle feasibility, the system energy balances and the proper working fluid 600 selection.

601 The collector performance was analyzed operating at different temperatures and it was 602 observed that as the operation temperature increases, the collector efficiency decreases, but the 603 theoretical efficiency of the ORC increases. Thus, a point of operation was determined to meet the 604 needs of both systems.

Different working fluids were analyzed below the maximum collector permissible pressures, showing that hydrocarbons (HCs) provide a better performance than other organic fluids at low pressures (100-1000 kPa). A multiple criteria analysis was conducted (i.e. enthalpy change, thermal efficiency and expansion ratio), showing a similar performance, with some variations. This study found that acetone is the most suitable working fluid because of its higher enthalpy change in the operating conditions.

611 This design method allows evaluating the useable heat generated by a solar system under 612 different operation conditions, selecting the most suitable working fluids for the cycle and therefore 613 selecting the best design conditions for an ORC cycle expander.

It is technically feasible to implement a CHP system, with the proposed configuration, with a better utilization of the heat unutilized by the SDHWS in a second application through a low temperature ORC. With a solar collector area of 2.84 m² it is possible to generate 443 kWh of electric power annually with a solar to electric efficiency of 6.65 %, or an overall cogeneration system efficiency of 56.35 %.

619 Patents

620 Patents related to this work:

- 621 Cogeneration System for Integration into Solar Water Heating Systems, U.S. Patent, Application
 622 Number: 16,208,666.
- Dispositivo de cogeneración para integración en sistemas de calentadores solares de agua,
 IMPI: MX/E/2017/094793.

625 Author Contributions

626 Cogeneration System proposal and evaluation, D. L., Collector and thermo tank performance
627 evaluation, R. B., DHWS simulation, P. C., Working fluid selection, S. I., Analysis of energy potential
628 delivered to the expander, N. V.

629 Funding

630 The open access publishing fees for this article have been covered by the Premio PRODETES
631 2017 - SENER – World Bank - Global Environment Facility.

632 Acknowledgments

Author D.A. Leal-Chavez thanks the National Council for Science and Technology
(CONACYT) and the Centro de Investigación en Materiales Avanzados for their support for the
completion of doctorate studies (CVU 483872).

636 Conflicts of Interest,

637 The authors declare no conflicts of interest.

Nomenclature

E	Total Energy
h	specific enthalpy (kJ/kg)
ṁ	mass flow rate (kg/s)
Р	pressure (kPa)
Pr	Prandlt number
Q	heat (W)

Qʻu	Useful heat gain (W)	
S	entropy (kJ/kg)	
t	time (s)	
Т	temperature (°C)	
U	overall heat transfer coefficient	(W/m^2K)
V	volumetric flow rate (m ³ /s)	
W	Work (W)	
Subscripts		
CHP	combined heat and power	
cond	condensation	
elect	electric	
evap	evaporation	
exp	expander	
ORC	Organic Rankine Cycle	
Greek Symbols		
Δ	difference	
η	efficiency	
μ	dynamic viscosity	
Q	density	

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