

Article

Characterization of polyurethane foam waste for reuse in eco-efficient building materials

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Abstract: In the European Union, the demand for polyurethane is continually growing. In 2017, the estimated production value of polyurethane was 700,400T, of which 27.3% is taken to landfill, which causes an environmental problem. In this paper the behaviour of various polyurethane foams from the waste of different types of industries will be analysed with the aim of assessing their potential use in construction materials. In order to this, the wastes were chemically tested by means of CHNS, TGA, and leaching tests. They were tested microstructurally by means of SEM. The processing parameters of the waste was calculated after finding out its granulometry and its physical properties i.e. density and water absorption capacity. In addition, the possibility of incorporating these wastes in plaster matrices was studied by determining its rendering in an operational context, finding out its mechanical resistance to flexion and compression at 7 days, its reaction to fire as well as its weight per unit of area and its thermal behaviour. The results show that in all cases, the waste is inert and does not undergo leaching. The generation process of the waste determines the foam's microstructure in addition to its physical-chemical properties that directly affect building materials in which they are included, thus offering different ways in which they can be applied.

Keywords: polymer waste; polyurethane foam; leaching test; microstructure

1. Introduction

According to the latest report published by Plastic Europe-the Facts 2017 [1], the demand for plastic in Europe in 2016 was 49.9 MTn, 3.1% higher as regards 2014. Of this demand, 7.5% is polyurethane, which implies an annual demand of 3.78 MTn in 2017. Of the 3.78 MTn, approximately 70% is in the form of foam (1.40 MTn of flexible foam, 1.22 MTn rigid foam) 30% being that of polyurethane elastomers and other products. Of the 2.62 MTn of PU foam, approximately 27% of waste is generated (700.400 Tn), of which, 31.1% is recycled (220.000 Tn), 41.3% is incinerated (294.278 Tn) and the remaining 27.3% is taken to landfill (193.120 Tn). The sectors according to the weighting for demand are: Construction and building (24.5%); automotive (19.5%) refrigeration (21.3%) and other sectors within textiles, usage in technology etc. (34.7%). The majority of polyurethane products such as low- and high-density foam are thermostable [2]. The material is characterized by its lattice structure, maintaining its shape and resistance under high-pressure and high-temperature conditions that end up degrading, thus after it has been manufactured, and after it reaches gelation point, the material cannot be melted in order to be remodeled into other products. As a consequence of this, the recycling process of thermostable polyurethanes is complex and unprofitable (chemical, mechanical and thermochemical recycling) [3]. As regards recovery techniques based on incineration in order to regain energy, there are environmental disadvantages due to the emission of atmospheric contaminants such as HCB dioxins and the emission of fine particles [4]. These disadvantages along

with the large amounts of waste taken to landfill, give rise to thinking of alternative recovery of this type of waste. During the past decade, European obligations to control the environmental impact of waste incineration (Directive 2000/76/CE) [5] and of landfill of waste (Directive 2008/98/CE) [6] have led to the increase cost of these waste treatment options. These costs will increase as more strict controls are introduced, as taxes on landfill and on incineration increase, further encouraging reuse.

There are several papers where the option of reusing polyurethane foam waste, combined with pitch binders, has been researched and PU foam waste used as a dry aggregate in different cement or gypsum matrices. Studies on cement and PU mortars have shown that there is a positive influence of these recycled aggregates on their manufacture which insures excellent durability even with regard to other traditional aggregates [7]. Previous research has led us to think that this polymer is able to reduce the amount of sand in cement mortars by substituting sand with PU by between 13-33% [8], 25-50% [9] or even 25-100% [10], all of these taking into account substitution in volume. The choice of the volume of substitution depends on the characteristics that are desired to be achieved in the final product. Products that are considerably more flexible and hydrophobic than other conventional materials [11] could potentially be obtained. In reference to research on gypsum material with polyurethane, the results are shown that establish the compatibility of the PU waste with a gypsum-based aggregate by combining different amounts of the PU waste in order to obtain a new cladding material for façades with thermal insulation properties [12]. Other research has advanced the design of this material, incorporating it in prefabricated gypsum materials that is extremely light-weight and has thermal and sound insulation properties. Laboratory tests have been carried out to improve the mechanical properties of these materials by means of the inclusions of additives and fibres [13]. Nevertheless, there is still a long way to go in order to optimise the properties of these materials. Fundamentally, in terms of its fire reaction properties and acoustic improvements. One of the key parameters for this is a thorough study of the waste's physical, chemical and micro structural characteristics, which can vary depending on its provenance.

The aim of this research is centred on the analysis of the properties of five polyurethane wastes from different industries with a view to assess its potential reuse in prefabricated, gypsum-based construction materials. The intention by doing so, is to provide an alternative to the current practice of incineration and recycling options in accordance with the criteria established in the European Parliament Directive 2008/98/CE and the European Council 19 November 2008 Directive on waste.

2. Materials and Methods

In order to determine the viability of using polyurethane waste cells to outline the possibility of its use in new building materials, five types were selected from different industries and chemical characterization tests were carried out using elemental analysis (CHNS), thermal gravimetric analysis and waste leaching test were also carried out. In order for them to be incorporated into new materials as a dry aggregate, the waste must be previously processed. After this, the granulometry is determined and the processing parameters are calculated as well as their physical characterization analysis which determines their real and apparent density, their ability to absorb water. Finally, they are microstructurally characterized using Scanning Electron Microscopy (SEM).

3.1. Materials

Five polyurethane foams from different industries were analysed (Figure 1):

(P), (B), from the refrigeration insulation industry in Cuenca, Spain.

(I) from the refrigeration insulation industry in Zaragoza, Spain.

(A) from the automotive industry in France

(SG) from scrap vehicles, Madrid (Spain)

The definition of each polymeric waste is:

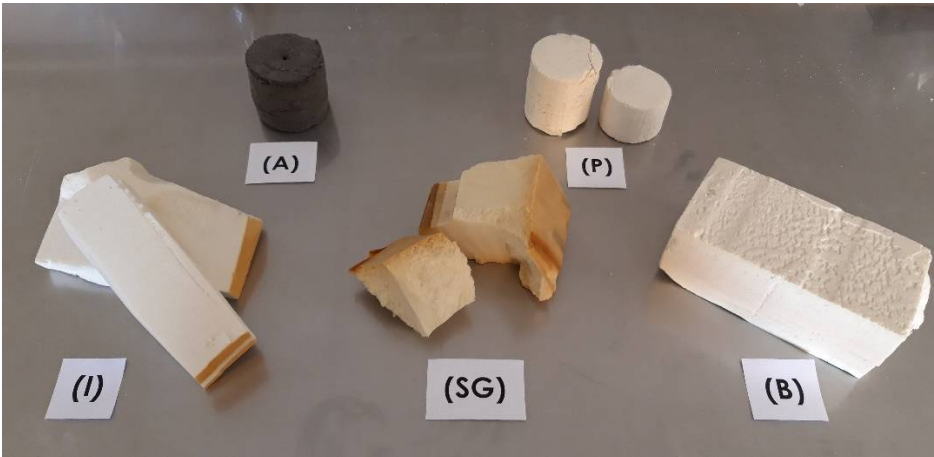
(P): Rigid yellow polyurethane foam waste, in powder, compressed into pellet form (Pellets). The waste is generated in the manufacture of insulation panels for the refrigeration sector. The waste is produced by trimming edges during the production stage.

(B): Rigid yellow polyurethane foam waste, in the form of plates (Block). Waste generated in the manufacture of insulation panels for the refrigeration sector. The waste comes from rejected panels and remnants of panels used in factory tests.

(I): Rigid yellow polyurethane foam waste in the form of plates (Block). The waste is generated in the manufacture of insulation panels for the refrigeration sector and comes from factory waste.

(A): Semi-rigid grey polyurethane foam waste, which comes in pieces and powder form, it is compressed into a pellet shape. The waste is generated in the manufacture of automobiles.

(SG): Semi-rigid polyurethane foam waste; they are remains of car seats from scrapped cars.



| Company | Origin | Presentation | Nomenclature |
|--------------|--|--------------|--------------|
| PAP | Insulating panels for the refrigeration sector | PELLETS | P |
| PAP | Insulating panels for the refrigeration sector | BLOCK | B |
| ITALPANNELLI | Insulating panels for the refrigeration sector | BLOCK | I |
| ANTOLÍN | Manufacture of automobiles | PELLETS | A |
| SIGRAUTO | Car seats | BLOCK | SG |

Figure 1. Polyurethane foams from different industries.

109 3.2. Methods

110 Elemental Analysis (CHNS)

111 This technique is used for the quantitative determination of carbon (C), hydrogen (H), nitrogen (N)
112 and sulphur (S) in all sample types, to obtain the oxide content, measured as a percentage of the
113 weight. The equipment used is a LECO Analyzer CHNS-932 and VTF-900. The analysis technique is
114 fully automated, and is based on the combustion of the samples under optimum conditions ($T = 950$ -
115 $1,100^{\circ}\text{C}$ in pure oxygen atmosphere) to convert the aforementioned elements into simple gases (CO_2 ,
116 N_2 , H_2O and SO_2) to achieve a quantitative determination of C, N, H and S content.

117 Waste Leaching test

118 This test was carried out according to the UNE-EN 12457-2 Standard [14]. A sample of waste is placed
119 in a bottle with a certain amount of water, and placed in a stirring device for 24 hours. Once the
120 mixture is stirred and filtered, the eluate (liquid to be tested) is obtained. Tests like pH, Electrical
121 conductivity (EC), Salt and TDS (total dissolved solids) were obtained. It is necessary to make a blank
122 with distilled water, before testing the eluate.

123 Thermogravimetric analysis (TGA)

124 This technique measures the change in mass of a sample, while being heated at a constant speed. In
125 this case, it is used to find out the degradation temperature of the waste, thus outlining the working
126 temperature of the material. The waste samples used in this test were previously processed. The
127 equipment used, is a Q600 thermal analyser TA Instruments (TGA/DSC), which simultaneously
128 provides a true measurement on the same sample from room temperature to 1500°C of heat flow
129 (DSC) and weight change (TGA). It has a dual balance mechanism, a twin conductor, and horizontal
130 purge gas system with mass flow control and gas switching capability. This equipment is joined via
131 a TG interface to an F-TIR spectrometer, which also facilitates the simultaneous analysis by infrared
132 spectroscopy of the gases produced in the decomposition of the substances studied.

133 Density

134 To calculate the real density of the polymer waste, test principles for natural stone, according to the
135 UNE-EN 1936: 2007 Standard [15] were applied, using the pycnometer method. It is necessary to
136 crush a sample of raw material until a fineness of particle capable of passing through a 0.063 mm
137 sieve is achieved. A 10 g sample of material in isopropyl alcohol is placed into the pycnometer and
138 then weighed. The pycnometer is cleaned, filled with isopropyl alcohol again and reweighed. The
139 real density (in Kg / m^3) is calculated by means of the ratio between the mass of the dry and crushed
140 test piece, and the volume of liquid displaced by the mass.

141 To determine the apparent density of the different wastes used in this research, its weight / volume
142 ratio was calculated. The procedure consists of filling a container with a specific volume and it is
143 weighed, then the weight / volume equation is applied. In this case, a 1 litre capacity container was
144 used as a reference, which was filled with crushed waste and then weighed.

145 Water absorption capacity

146 In this test, that which is established in standard UNE-EN 13755:2008 [16] was adapted. The test
147 consists of placing the material to be tested (dry and with constant mass), into a container filled with
148 water until fully covered, for 24 hours. At the end of that period of time, the material is weighed and
149 is placed back into the water for another 24 hours. The material is weighed yet again and if it can be
150 observed that there is a constant weight that does not differ from the previous day, the material has
151 reached saturation.

Then the following equation is applied:

$$Ab = (\text{Saturated weight} - \text{Dry weight}) / \text{Dry weight} \times 10088$$

Laser granulometry

The different foams were crushed, and their granulometric size was determined through laser granulometry diffraction using a HELOS 12K SYMPATEC analyser. The samples were analysed for 15 s in an isopropyl alcohol suspension.

Processing parameters

The processing parameters were defined by determining cutting time, crushing time and the energy of the crushing.

It is necessary to process the polyurethane foam to be used in the tests (B), (SG) and (I) wastes are split into smaller pieces. These pieces are placed in a RETSCH SM100 Mill, where they undergo a crushing and sieving process (Figure 2). The pellets (P and A) are directly placed into the crusher.



Figure 2. Previous Processing of Polyurethane foam waste

Scanning Electron Microscopy (SEM)

This technique, allows for the microscopic structure of the different polyurethanes (closed-cell open-cell) to be discovered. For this test, the waste samples did not undergo processing.

The equipment used, is a Microscope FEI Quanta-600, which allows for the observation and characterization of samples by obtaining high-resolution imaging of organic and inorganic materials at high magnifications. The equipment can be used in high vacuum, acting as a traditional scanning electron microscope (SEM), and can work in environmental mode (ESEM). The latter mode allows observation without coating or metallizing the sample, which makes it a non-destructive technique. The equipment is also used alongside two sets of X-ray microanalysers, the EDX and WDX Oxford that allows for elemental analysis in a timely manner, or the compositional mapping of specific areas of the materials studied.

3. Results and Discussion

3.1. Elementary analysis (CNHS)

Table 1 shows the results of the carbon, nitrogen and sulphur components of each of the wastes that were analysed. As was expected, carbon was the majority component of all the polymer wastes. In each case they had a similar percentage of carbon. The analysis also confirmed the existence of hydrogen and nitrogen in smaller proportions with respect to the whole of the waste. As regards the other components that each waste has, it could be asserted that as it is a case of polyurethane, there is a significant amount of oxygen [17] and other components associated with the possible impurities each foam may contain. For example, when it comes to foams that come from the scrapped vehicle seats (SG), it can contain metals linked to elements from the actual seat such as copper or aluminium, which will later be identified in the scanning electron microscopy test. On no occasion was the presence of sulphur detected.

Table 1. Results of Results CNHS Analysis of different PU waste

| Waste | Chemical element (%) | | | | |
|-------|----------------------|------|------|------|--------|
| | C | H | N | S | Others |
| P | 64.48 | 5.63 | 6.74 | 0.00 | 23.15 |
| B | 62.06 | 5.07 | 6.58 | 0.00 | 26.29 |
| SG | 64.67 | 7.75 | 4.80 | 0.00 | 22.78 |
| A | 63.74 | 6.15 | 6.04 | 0.00 | 24.07 |
| I | 63.34 | 5.58 | 7.28 | 0.00 | 23.80 |

3.2. Thermogravimetric Analysis (TGA)

The Tga test results show the % loss of weight of the different wastes when the temperature increases.

Wastes (P) and (B) come from the same company and have the same isocyanate polyol component composition of. The difference between them both is the presence of metal impurities that (P) has with respect to (B), which shows as being totally clean. This difference is noted in the loss of mass, which in the case of (P) occurs at 280°C, and that does not occur in waste (B). In both cases the first degradation occurs at around 200°C, polymer decomposition occurs from 325°C to 550°C (Figure 3).

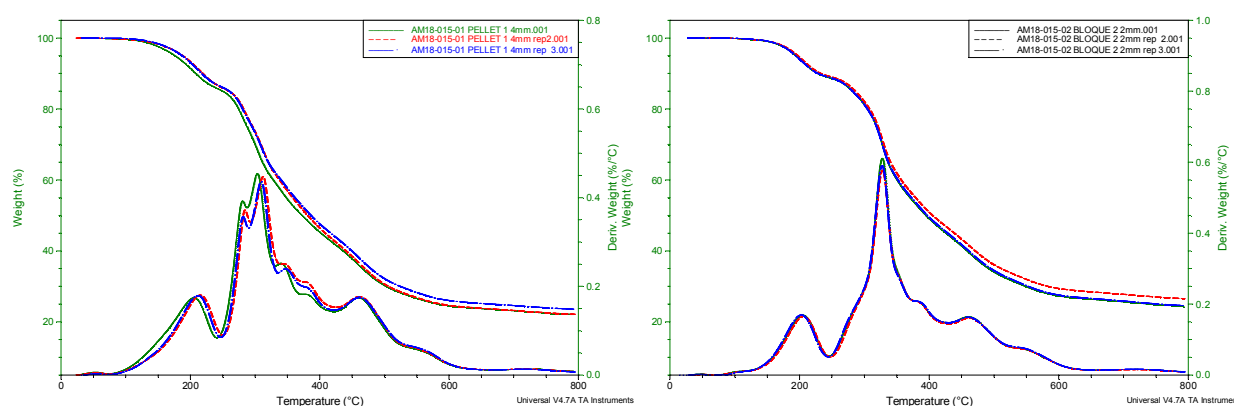


Figure 3. Tga of the polyurethanes (P)-Fig.3a. and (B)-Fig.3b. that come from the insulation industry for refrigeration from the Paneles Aislantes Peninsulares Factory

In the case of foam (I) (Figure 4), it shows a very similar behaviour to that of foam (B). Both of them come from the insulation for refrigeration industry and have a similar initial mass loss at 238°C and the total decomposition of the polymer occurs between 345°C-450°C.

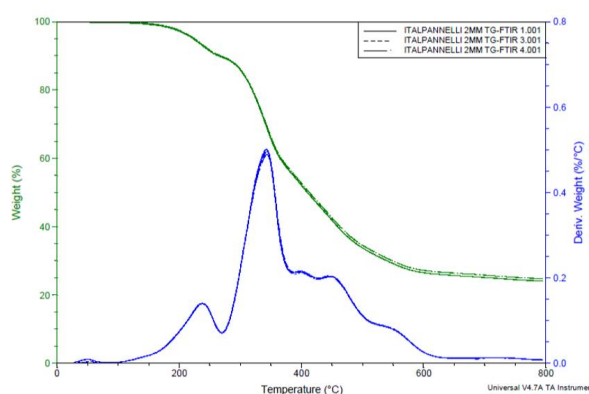


Figure 4. TGA of polyurethane (I) that come from the insulation industry for refrigeration from the Italpanelli factory

In the case of flexible foam (SG), a minimal loss of mass was observed at 280 °C, probably due to the metal impurities that car seats contain. The loss of mass corresponding to the polymer's decomposition occurs between 400°C and 550°C (Figure 5a). In the case of foam (A), three different mass losses occur (Figure 5b). The first of these losses occurs at 320°C with a significant degradation of the material. The second loss occurs at 400°C which corresponds to the polymers degradation and the last stage occurs at 500°C, which corresponds to the loss of other components in this foam.

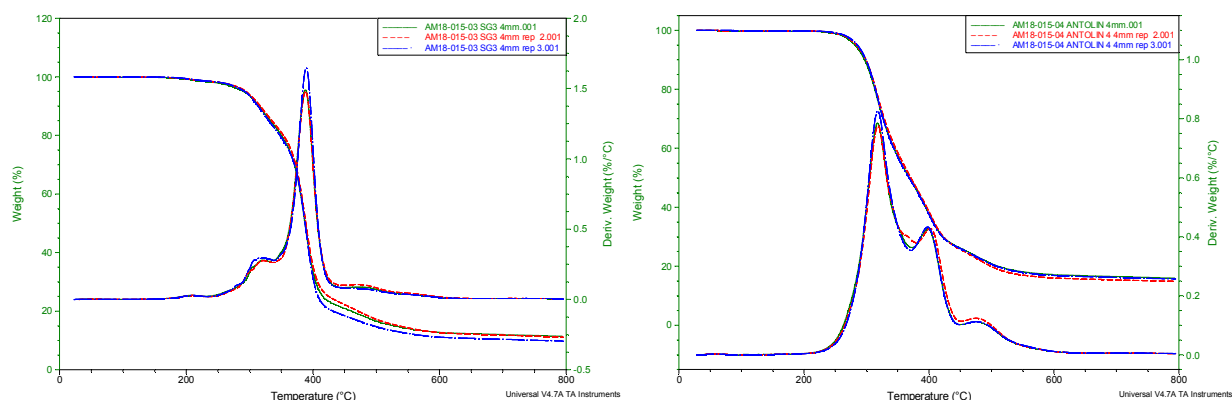


Figure 5. The Tga of polyurethanes (SG) - Fig. 5a. and Fig.5b that come from the insulation industry for refrigeration, from the Paneles Aislantes Peninsulares factory.

These results indicate that the thermal behaviour of the material is acceptable [18]. Thus, this thermal analysis technique is a highly useful tool for studying the reuse of these polymers, with no chemical or physical changes detected.

3.3. Scanning Electron Microscopy (SEM)

Generally speaking, polymeric cellular materials can be defined by a two-phased structure in which the gaseous phase stemming from a foaming agent, whether physical or chemical, was dispersed throughout a solid polymeric matrix [19]. Foam is a specific type of cellular material that is generated by the expansion of a material in liquid form. This is the case of wastes type (B), (I) and (SG) that were analysed in this research. However, there are processes subsequent to foaming that cause a loss of cellular structure of the polymer. This is the case for wastes (P) and (A) that are obtained by means of a milling process that generates particles of an extremely fine nature that are also compressed. This causes a structure with layers of polymer that are very different from that of the wastes obtained differently and have sheets of foamed polyurethane. (Figures 6a-6d).

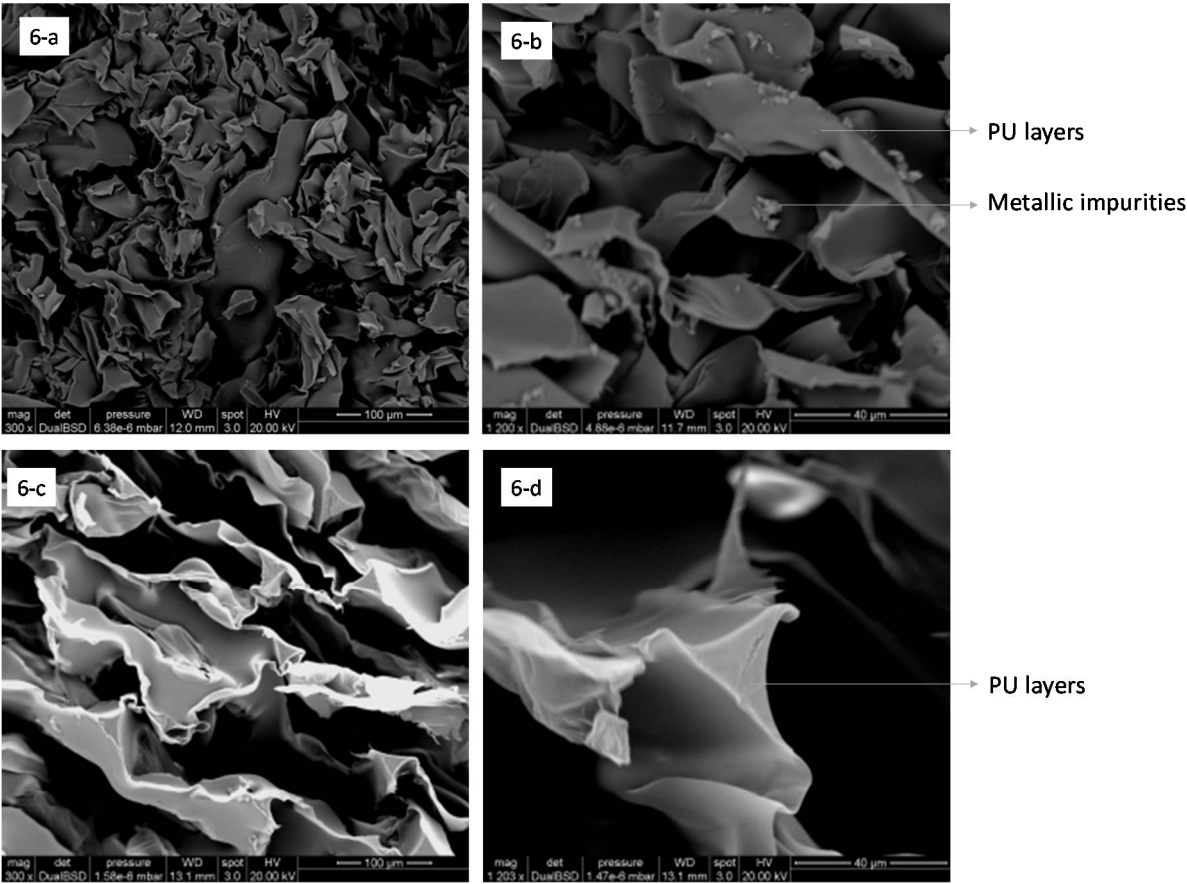


Figure 6. Microstructure of the PU waste (P)-Fig.6a-6b. y (A)-Fig.6c-6d by SEM

The cellular materials are classified according to their cellular structure and the cell connectivity. In the case of foam type (B), the structure is an intermediate structure and it can be seen that a portion of the cellular structures is formed by an open-cell structure while the other portion of the cellular structure is formed of a closed-cell structure (Figure 7a-7b). In this case, the walls are of a 10 μm thickness and the cells have a diameter of between 10 and 40 8 μm. Foam type (I) has a closed-cell structure in which the gas is occluded in the interior of the cells. The cells are largely homogeneous in terms of the size of which they are comprised, between 100-200 μm (Figure 7c-7d). Flexible foam (SG) has an open-cell structure where gas can freely circulate between the cells since they are interconnected with each other, which can cause an improvement in the acoustic properties [20]. (Figure 7e-7f). In this case, a characteristic that is typical of this type of foam can be observed, that is

the presence of pores in the cells' interconnecting walls, as well as the presence of metal impurities.

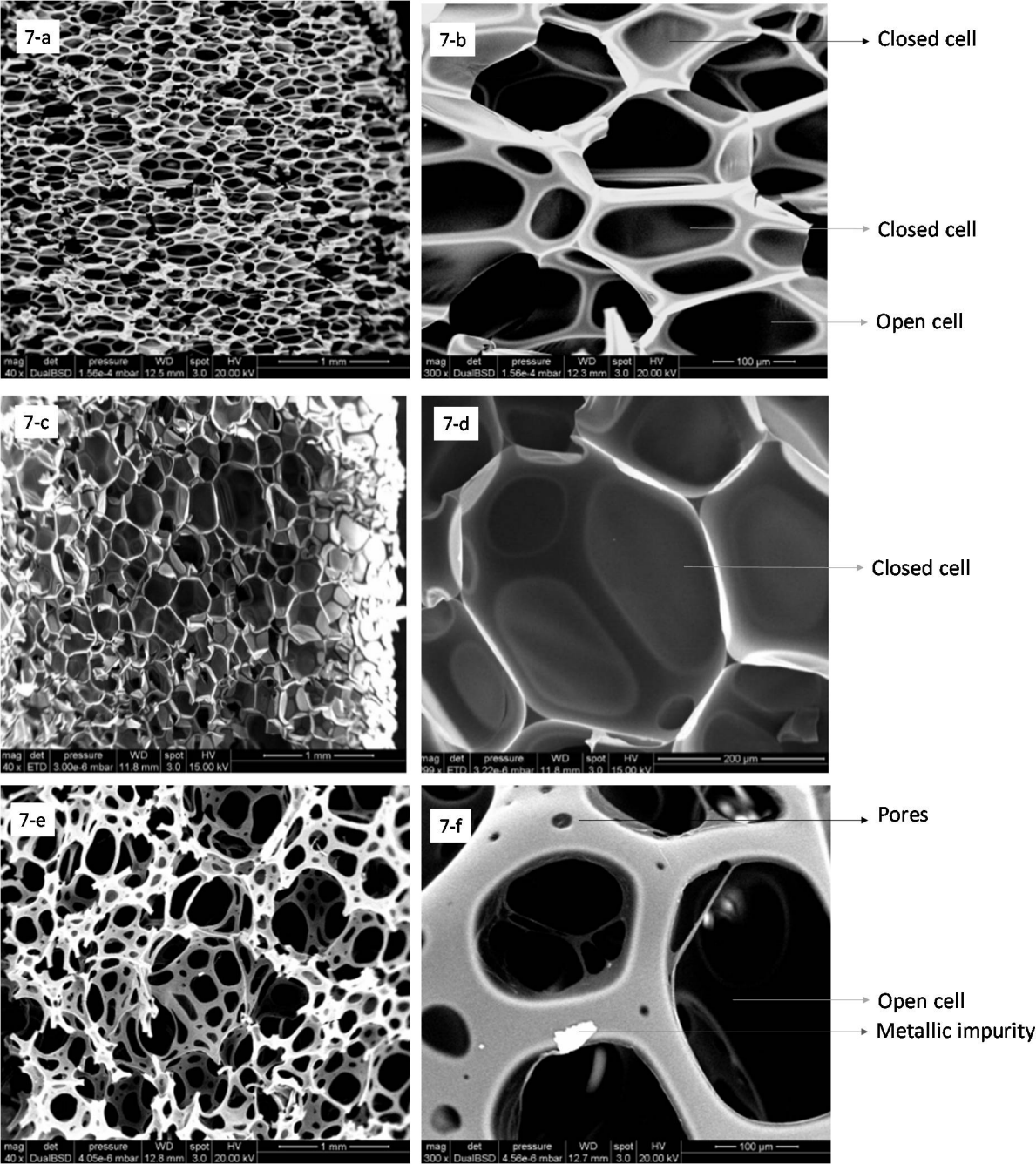


Figure 7. Microstructure of the PU waste (B)-Fig. 7a-7b; (I)-Fig. 7c-7d and (A)-Fig. 7d-7e by SEM

3.4. Waste Leaching test

One of the processes that must be monitored when using waste materials as raw materials in construction, is leaching. It is common for the materials to be in contact with water or dampness, which could cause a leaching process [21]. In view of the results obtained from the leaching test (Table 2), it can be observed that the electrical conductivity does not exceed that maximum value permitted (3000 $\mu\text{S}/\text{cm}$). As for the maximum amount of the total of dissolved solids (500 mg/L), on no occasion was this amount exceeded. The values were always lower than 100 mg/L. The Ph levels were also within the permitted range (5,5-9). It can therefore be established that the wastes analysed do not display any contaminating behaviour when in contact with water or dampness thus they can be used in construction materials.

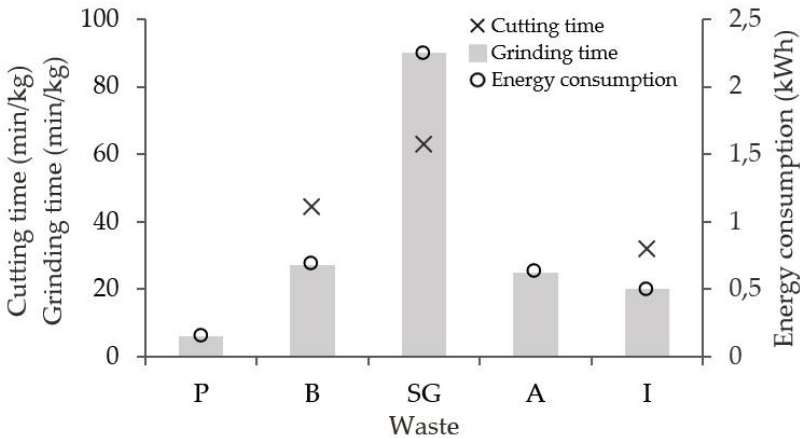
Table 2. Electrical connectivity, total of dissolved solids, salt and the pH of the different wastes

| Waste | EC (μs/cm) | TDS (mg/L) | Salt (mg/L) | pH |
|-----------------|------------|------------|--------------|-----|
| Distilled water | 1.8 | 1.24 | Out of scale | 6.6 |
| P | 38.2 | 19.7 | 14.3 | 6.6 |
| B | 63.4 | 40.5 | 28.3 | 7.5 |
| SG | 149.2 | 95.4 | 69.0 | 7.9 |
| A | 27.9 | 21.5 | 15.6 | 7.7 |
| I | 32.8 | 20.7 | 15.3 | 6.8 |

3.5. Processing parameters

In Graph 1 the preparation (cutting) and crushing times are outlined, as well as the energy used in the processing of 1 Kg of waste.

In this paper, wastes in slab form (I), board (B) as a whole (SG) and in pellet form (P), (A) were studied. In the case of wastes (I), (B) and (SG) it was necessary to cut then prior to placing them into the shredder. It must be taken into consideration that the process is carried out on a laboratory scale in which the shredder has a limited input capacity. In this case, the waste that needed the least amount of cutting time was waste (I) explained by the fact that it is has a compact closed-cell structure and is more rigid than the other foams. The foam that needed the most amount of time was type (SG), 60 minutes shredding per Kg of the sample, as this is a flexible, highly pliable and difficult to handle foam. As regards machine shredding time, the values proportionally vary to the prior cutting time. In waste type (P) the duration of shredding took the least amount of time (6 minutes). This is followed by type (I) with 20 minutes. The longest time is for waste (SG), there was additional difficulty in working with the waste due to the machine’s sieve becoming blocked due to the nature of this type of foam. The use of energy.

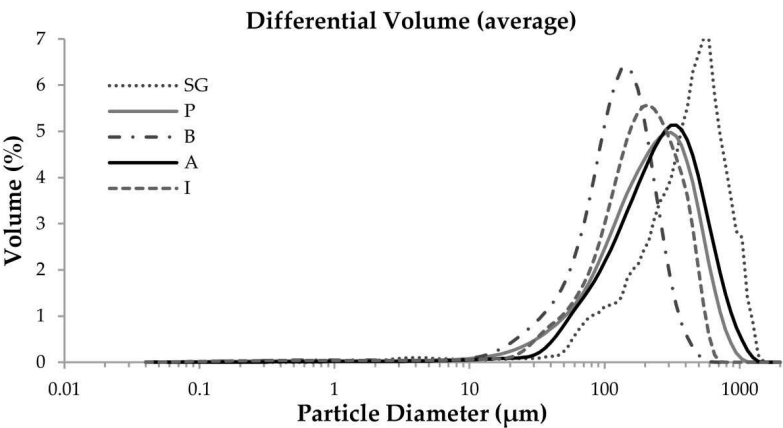


Graph 1. Cutting time, grinding time and energy consumption of different PU wastes

3.6. Laser diffraction granulometry

The granulometric study focussed on sizes less than 1mm. Graph 2 shows the granulometric results of the different wastes in sizes smaller than 1 mm. In view of the results, it can be observed that wastes in powder form (P), (A) have a very similar particle size with an average diameter of 229μm and 271μm respectively. The waste in the form of a slab (I) has a diameter of 194μm, noticeably greater than the board shaped waste (B). The most significant difference can be observed in waste (SG) with an average particle size of 401 μm and parts that can reach 772μm. The particle size distribution in

the different wastes, alongside the real density values will determine the final mechanical properties of the construction material [22]. Such is the case of plasters with dry aggregate of waste polyurethane that will be studied in the following section.



Graph 2. Granulometric curve (Volume %) of different PU wastes

3.7. Determination of apparent and real density. Water absorption capacity

Another parameter that strongly determines the material’s final properties and as a consequence, how it can be applied is density. On looking at the results of Table 3, it can be observed that wastes with a higher apparent density are those that have been compressed into pellet form (P), (A), which indicates that in a lessor volume there is space for a larger amount of the material. This factor is due to the material’s extreme fineness and to the polyurethane being arranged in layers. This was previously observed microstructurally by SEM imagery. The polymer with the lowest apparent density was the flexible waste (SG) with 72% less. As can be observed, the apparent densities are, in all cases, very low. This indicates that there was a problem related with the storage of this type of waste as well as its transportation and keeping since a low weight of the materials occupies a large volume [23]. The water absorption capacity of each waste varies according to the foam’s structure and morphology. Thus, the flexible foam (SG) is the foam that showed the greatest absorption capacity. This is probably due to the highly porous nature of its cells, which is further accentuated by the presence of pores in between the cell walls. As it is a case of an open cell structure (Fig. 7e-7f), the water enters the interior of the foam more easily. Both waste (B) and waste (I) showed lower absorption capacity caused by a semi-closed and closed cell structure respectively.

Table 3. Results of apparent density, real density and total water absorption of different PU waste

| Waste | Apparent density (Kg/m3) | Real density (Kg/m3) | Total absorption (%) |
|-------|-----------------------------|-------------------------|-------------------------|
| P | 141.7 | 1052.7 | 2.0 |
| B | 45.5 | 1370.9 | 28.0 |
| SG | 39.8 | 1211.1 | 645.0 |
| A | 86.1 | 1378.6 | 333.5 |
| I | 56.0 | 1105.0 | 49.0 |

3.8. The possibilities of the uses of the wastes studied

One of the ultimate aims of this research is to improve the use of these types of foams and expanding this use in the industries that generate polyurethane, in this way improve the ratio of the volume of reused PU. One option is to incorporate the processed waste as a dry aggregate in plaster matrices. Different substitutions have been made in previous studies, of plaster type A1 for rigid PU foam waste [24, 25]. The conclusion was reached that the optimal ratio of components in volume could be

(1/1.5), that is, 1part gypsum with 1.5 parts of polyurethane foam waste. This study used all of the wastes characterised in this paper.

The following test results will now be shown: mechanical resistance to compression and flexion at 28 days [26] (Table 4) and fire reaction by means of non-combustion test [27] and the gross heat of combustion test [28] (Table 5). Of the samples that shown, the best behaviour, the weight per unit of area and thermal conductivity, [29] was calculated compared with a standard plaster.

Table 4. Mechanical resistance to compression and flexion at 28 days in samples (1/1.5) with different PU waste

| Sample | 1/1.5 (P) | 1/1.5 (SG) | 1/1.5 (A) | 1/1.5 (B) | 1/1.5 (I) |
|--------------------------------|-----------|------------|-----------|-----------|-----------|
| Compression strength at 7 days | 2.00 | 3.71 | 3.70 | 3.95 | 4.01 |
| Flexion strength at 7 days | 1.15 | 1.71 | 1.91 | 2.23 | 2.81 |

Table 5. Results of non-combustion test and gross heat of combustion test in samples (1/1.5) with different PU waste

| Sample | 1/1.5 (P) | 1/1.5 (SG) | 1/1.5 (A) | 1/1.5 (B) | 1/1.5 (I) |
|----------------------------------|-----------|------------|-----------|-----------|-----------|
| Temperature increase (°C) | 71.15 | * | * | 16.6 | 19.5 |
| Flaming time (s) | 339 | * | * | NONE | NONE |
| Loss of mass (%) | 37.89 | * | * | 26.63 | 27.72 |
| Superior Calorific Power (MJ/Kg) | - | - | - | 1.048 | 1.596 |

*Failed

In all cases, the mechanical resistance obtained met the requirements outlined in the regulations with over 2 MPa of resistance to compression and over 1MP of flexion resistance. As regards the fire reaction properties, initially, the samples underwent the non-combustion test. The thermal behaviour of the samples, confirmed by the non-combustibility test, gives us an idea of their fire retardance properties. The results of the non-combustibility test (Table 5) confirmed that the samples that included polyurethane in their composition, and specifically, the 1/1.5-(B) sample and 1/1.5 (I), did not have flaming times of less than 20 seconds with a temperature increase of below 50 °C and losses of less than 50% of their mass. This result indicated that even if the contribution of the materials to fire reaction is taken into account, their composition corresponded to Euroclass A1 (non-combustible) in accordance with the European fire reaction classification of building materials for homogeneous products [30]. In order to check this classification, the Superior Calorific Power was calculated showing a value of below 2 MJ/Kg. Therefore, these materials can be classified as non-combustible. The rest of the mixtures with other wastes did not meet the minimum standards required in the non-combustible test. They will be need to be tested with regulation EN- 13823 and EN ISO 11925-2 in order to check classifications A2 or lower. It was noted that in the SEM tests, these wastes had impurities due to metal contamination or due to adhesives, which would be the reason why they did not reach the minimal requirements established in order to be classified as A1.

Mixtures 1/1.5-(B) sample and 1/1.5 (I) were tested in order to determine their weight per unit of surface and their thermal conductivity which are two important properties when it comes to determining their characteristics when in use. The values from both tests are shown in Table 6.

Table 6. Results of thermal conductivity and weight per unit of surface of sample 1/1.5(B) & 1/1.5 (I)

| Parameter | Standard Plaster | Gypsum- PU 1/1.5-(B) | Gypsum- PU 1/1.5-(I) |
|------------------------------|------------------|----------------------|----------------------|
| Thermal conductivity (W/m*k) | 0.30 | 0.20 | 0.19 |
| Weight (Kg/m2) | 8.33 | 5.88 | 5.60 |

In view of the results, it can be observed that the materials composed of Gypsum-PU-(B) had a 30% and 33% reduction in surface weight with respect to the standard plaster, which is explained by a lower real density of waste type (B) 1370.9 kg/m³ and 1105.0 kg/m³ of waste (I) with respect to that of the plaster that it substituted, which is 2650 Kg/m³. Given that the thermal conductivity depends on density and on the characteristics of the actual PU waste [31], the values were reduced by up to 36% when it comes to mixtures 1/1.5 (I) with respect to the conventional plaster which gives rise to an improvement in the material's thermal insulation.

4. Conclusions

Five PU wastes from different sectors and industries were chosen in order, for there to be a wider scope for PU to be reused and for it to be easier for the project to be replicated allowing polyurethane waste to begin to be used in different sectors.

- All of the polymers degrade at above 200 ° C. In the case of polyurethane SG, degradation occurs at a higher temperature (400 ° C).
- None of the PU wastes have a leaching capacity and they are all considered to be suitable for use in new construction materials.
- The wastes that had been compacted had the best processing times, with the same prior cutting time and low energy use. This characteristic that these types of foams display along with the fact that they have the greatest apparent density create an advantage with respect to the other wastes with regard to PU being productively reused in building materials. Both in terms of transportation (the material's generating factory) and in the collection of the waste and the rendering of the mixture.
- It was observed that the polyurethane that underwent a milling process had a high level of fineness with average particle sizes being around 250 µm and had greater levels of apparent density in respect of the rest of the wastes. The flexible foam had a larger average particle size of approximately 400 µm. In this case, the apparent density is lower compared to the rest of the foams.
- The microstructure of the polyurethanes is different depending on the industry from whence they came. In the case of board and slab shaped wastes that come from the refrigeration industry, the structure is hexagonal semi-closed celled. Open and closed cells can be observed in the images from the SEM. The waste from the refrigeration industry is in slab form (I) and has a closed-cell structure. In both cases, adequate thermal behaviour was predicted, which could be used in improving thermal insulation when they are included in construction material.
- It is observed that SG waste has a structure suited to be used as a possible acoustic absorber, because of its open pore structure.

- The wastes that result from milling processes have a structure that is in the form of overlapping layers with no defined hexagonal structure. In this case, the wastes had some metal impurities in their structure associated with the actual milling process.
- As a final aim of this research, there is the possibility of including the wastes in plaster matrices in ratio with volume (1/1.5) thus obtaining adequate mechanical resistance to compression of over 2 MPa, obtaining a reduction in thermal conductivity by 33% and a reduction in the weight of the material by 31% was analysed. As regards the non-combustibility test and calorific value test, only the rigid PU foam wastes (B), and (I) met the standards to have an A1 classification, which is ideal for interior cladding materials for buildings. There was worse fire reaction behaviour in samples (P), (A) and (SG) due to the impurities that they contain. Nevertheless, it must be determined whether the classification in these two cases would be that of A2 or worse and alternative ways for the material to be applied in different areas of a build will need to be found.

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