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Optimization and entropy production: application to Carnot-like refrigeration machines

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Abstract: Several optimization models of irreversible reverse cycle machines have been developed based on different optimization criteria in the literature, most of them using linear heat transfer laws at the source and sink. This raises the issue how close to actual operation conditions they are, since the heat transfer law on the phase-change processes is dependent on ΔT^3 . This paper addresses this issue by proposing a general model for study and optimization of thermal machines with two heat reservoirs applied to a Carnot-like refrigerator, with non-linear heat transfer laws and internal and external irreversibility. The optimization was performed using First and Second Law of Thermodynamics and Lagrange multipliers method. Thus, several constraints were imposed to the system, also different objective functions were considered, allowing finding the optimum operating conditions, as well as the limited variation ranges of the system parameters. Results show that the nature of the heat transfer laws affects the optimum values of system parameters for obtaining maximum performances and also their magnitude. Sensitivity studies with respect to system several parameters are presented. The results contribute to the understanding of the system limits in operation under different constraints and allow choosing the most convenient variables in given circumstances.

Keywords: Carnot machine; refrigeration; entropy production; thermodynamics; optimization

1. Introduction

The second half of the last century saw the emergence of a new branch of Thermodynamics, called Thermodynamics in Finite Time, due to the process duration that was considered by the heat transfer rate [1,2]. The corresponding approach emphasized a maximum power regime [3] for an endo-reversible, but exo-irreversible engine. The engine efficiency at maximum power regime is less than the Carnot cycle one, and it introduced the particular form of the “nice radical”.

Since then, quite a lot of works analyzing the exo-irreversible engines have been developed [4,5]. Generally, they have considered only the external irreversibility of the thermal machine, which is due to the heat transfer at finite temperature difference between the sources and the working fluid. Also, mainly the linear form of the heat transfer law was used.

The same approach applied to the study of heat pumps shows no more optimum for the useful effect (heat flow delivered at the hot end), because the temperature of the working fluid has no upper limit imposed on the hot end [6]. The same happens for the refrigerating machine, where there is no lower limit to the cold end (as for thermal engine).

To the first reported work (to our knowledge) [6], other papers followed [7-9] dealing with heat pump or refrigerating machine optimization, but introducing an additional constraint (mostly mechanical power or useful effect imposed). The optimization is not reserved only to vapor mechanical compression machines, but it has been extended since that time to the trithermal and

quadrithermal ones [10–12]. However, one notes that most of the reported studies deal with thermostats at sources and sinks and linear heat transfer laws.

Proposals have been made to extend the analysis by using other forms of the heat transfer law [13–15].

Further progress has been made by introducing an irreversibility factor (ratio), most often constant [16], that accounts for the internal irreversibility of the machine (endo-irreversibility). A step further [17] represents the internal irreversibility considered in the model as internal entropy production rate, \dot{S}_i . This method is preferred to the one of ratios, as we still see in recent work [18]. It provides a more general approach compared to the use of the irreversibility factor [16,19–20]. Also, it aims the entropic analysis of machines and processes and is more and more important as an instrument getting modeling closer to real operation [21–27].

The present work proposes a general model of study and optimization of irreversible Carnot-like refrigerating machines and extends the previous studies [28–31]. In terms of optimization, it is well known that several objective functions can be chosen: maximum of coefficient of performance (COP), minimum of energy consumption, minimum of total entropy production, or economic objectives [32]. Environmental concerns have also become predominant [33], and recently led to the introduction of a new objective called ECOP (Ecological COP) [34]. Two objective functions will be considered here for each of the two studied cases: *maximum refrigeration load* and *minimum total entropy production rate* when COP is imposed, respectively, *maximum COP* and *minimum total entropy production rate*, for imposed refrigeration load.

The mathematical approach includes the most general laws of science, namely the First and Second Laws of Thermodynamics. The equations of heat transfer at the hot and cold side of the machine are modeled by non-linear functions of the temperature difference between the reservoirs and the cycle working fluid. The internal irreversibility is introduced by the method of internal entropy production rate, above mentioned. Three variation laws with temperature for the internal entropy production are considered in an attempt to approach analytical treatment of actual operation.

The purpose of this work is to study the influence of the non-linearity of the heat transfer law on the performance. This will *extend the model validity* beyond the convective and radiative heat transfer laws [13]. The results highlight the existence of optimal operating regimes of the reverse cycle machine subjected to dimensional and operating constraints. The sensitivity study with respect to the model parameters provides interesting results related to the *limitation of the variation range of the variables* of the model, and *different operating regime* of the refrigeration machine.

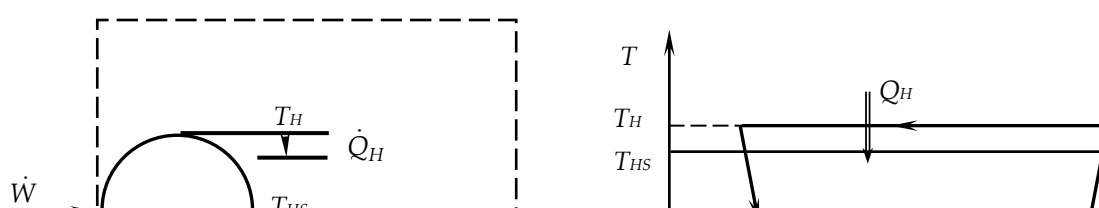
The new and important results reported here are presently

2. Materials and Methods

2.1. Proposed application study

The model of study and optimization of Carnot-like machines is applied here to an irreversible refrigeration machine. Its architecture and the corresponding irreversible cycle consisting of two isothermal processes and two adiabatic irreversible ones are illustrated in Figure 1. The irreversibility is present in the T-S diagram by the entropy production on the adiabatic processes (compression and expansion), which is mainly due to internal losses. On the isothermal processes, it is marked by the different heat transfer compared to the corresponding reversible processes (also more reduced at the source and larger at the sink).

Usually, the internal irreversibility of the machine, introduced in the model as a parameter, is represented by the corresponding internal entropy production term, \dot{S}_i . It takes into account the internal irreversibility generated on each cycle process, summing the throttling losses, friction ones, etc. To this one an external irreversibility is added, being generated by finite temperature heat transfer processes between the working fluid and the heat reservoirs.



(a) (b)

Figure 1. Irreversible Carnot-like refrigeration machine: (a) Scheme; (b) Irreversible cycle in T - S diagram.

As shown in Figure 1, the two heat reservoirs are thermostats, of constant temperatures T_{HS} and T_{LS} , the first one being actually the environment. The refrigeration machine is considered operating in steady state regime.

2.2. Mathematical model

The model of study and optimization of reverse cycle machines is developed by using the most general laws in Thermodynamics, the First and Second Law, to which generalized forms of heat transfer laws applied to the source and sink are added.

By taking account of the sign convention adopted here, that considers the heat transfer rate positive when entering the cycle, and negative for leaving it, and the consumed work transfer rate (power request) as negative, the expression of the First Law of Thermodynamics can be written as:

$$\dot{W} = \dot{Q}_H + \dot{Q}_L, \quad (1)$$

with \dot{W} – mechanical power supplied; \dot{Q}_H – heat transfer rate rejected by the working fluid at the source (hot heat exchanger); \dot{Q}_L – refrigeration load.

When looking the machine irreversibility, the Second Law of Thermodynamics takes different expressions, as applied to two possible systems. Thus, once considering only the internal irreversibilities (endo-irreversible machine) by the internal entropy production term, it comes:

$$\frac{\dot{Q}_H}{T_H} + \frac{\dot{Q}_L}{T_L} + \dot{S}_i = 0, \quad (2)$$

with T_H – temperature of the working fluid at the machine hot temperature side; T_L – temperature of the working fluid at the machine low temperature side; \dot{S}_i – entropy production due to internal irreversibilities of the machine.

Then, when the whole system is considered (cycle and the two heat reservoirs), the total entropy production term appears in the Second Law expression:

$$\frac{\dot{Q}_H}{T_{HS}} + \frac{\dot{Q}_L}{T_{LS}} + \dot{S}_T = 0, \quad (3)$$

with T_{HS} – source temperature at the machine hot part; T_{LS} – source temperature at the machine cold part; \dot{S}_T – total entropy production (internal and external).

For the heat transfer rates present in the above equations, two forms of non-linear heat transfer laws between the heat reservoirs and working fluid are considered:

$$\dot{Q}_H = K_{H,gen} (T_{HS} - T_H)^{n_H} (-1)^{n_H-1} ; \quad \dot{Q}_L = K_{L,gen} (T_{LS} - T_L)^{n_L}, \quad (4)$$

$$\dot{Q}_H = K_{H,gen} (T_{HS}^{n_H} - T_H^{n_H}) ; \quad \dot{Q}_L = K_{L,gen} (T_{LS}^{n_L} - T_L^{n_L}), \quad (5)$$

where n_i is the exponent of the heat transfer law (H – hot sink or high temperature side, L – cold source or low temperature side).

In practical application these non-linear forms correspond to phase-change heat transfer (eq. (4), with $n_i = 3$), respectively radiation heat transfer (eq. (5) with $n_i = 4$). One note that making $n_i = 1$, one gets the linear approximation, so called, convective heat transfer law or Newton's law.

For the refrigeration machines performance evaluation, according to its definition, the coefficient of performance is:

$$COP = \frac{\dot{Q}_L}{|\dot{W}|}. \quad (6)$$

By combining the two expressions of the Second Law, given by eq. (2) and (3), the total entropy production rate results as:

$$\dot{S}_T = \dot{S}_i + \dot{Q}_H \left(\frac{1}{T_H} - \frac{1}{T_{HS}} \right) + \dot{Q}_L \left(\frac{1}{T_L} - \frac{1}{T_{LS}} \right). \quad (7)$$

The three positive terms contained in the above expression accounts for different type of irreversibility, namely (1) the internal one by \dot{S}_i , and (2) the external one due to finite temperature heat transfer processes between the heat reservoirs and working fluid.

Without insight in the internal irreversibility mechanism of machine processes, it is difficult to establish a variation law for the internal entropy production. Thus, in order to provide generality to the model, the following approaches were considered [35,36]:

a) constant

$$\dot{S}_i = \text{const}, \quad (8)$$

b) linear variation law with working fluid temperature difference:

$$\dot{S}_i = \text{const} (T_H - T_L), \quad (9)$$

c) logarithmic variation law with working fluid temperature ratio:

$$\dot{S}_i = \text{const} \left(\ln \frac{T_H}{T_L} \right). \quad (10)$$

2.3. Dimensionless model

The same concern for developing a general model for the optimization of refrigeration machines and easily adapted to different characteristics and constraints of the studied case, led to the adoption of the non-dimensional form of the equations. Therefore, the following terms and expressions will be further used:

- Dimensionless temperatures are expressed relative to the reference temperature T_{HS} which is also the ambient one:

$$\theta_{SL} = \frac{T_{LS}}{T_{HS}}, \quad \theta_H = \frac{T_H}{T_{HS}}, \quad \theta_L = \frac{T_L}{T_{HS}}. \quad (11)$$

- Dimensionless energy fluxes are expressed relative to the product $\dot{m}c_p T_{HS}$, where $\dot{m}c_p$ is the heat rate capacity of the working gas corresponding to the end of the compression process;
 - at the sink or machine cold part (positive):

$$\tilde{q}_L = \frac{\dot{Q}_L}{\dot{m}c_p T_{HS}} = CTQ_{L,gen} (\theta_{LS} - \theta_L)^{n_L}, \quad (11)$$

- at the source or machine hot part (always negative)

$$\tilde{q}_H = \frac{\dot{Q}_H}{\dot{m}c_p T_{HS}} = CTQ_{H,gen} (1 - \theta_H)^{n_H} (-1)^{n_H-1}, \quad (12)$$

- power supply:

$$\tilde{w} = \frac{|\dot{W}|}{\dot{m}c_p T_{HS}} = \frac{\tilde{q}_L}{COP}. \quad (13)$$

- Non-linear thermal conductances are expressed relative to the product $\dot{m}c_p T_{HS}^{(n_H-1)}$:
 - total one:

$$CTQ_{T,gen} = \frac{(UA)_T T_{HS}^{(n_H-1)}}{\dot{m}c_p}, \quad (14)$$

- of the two heat exchangers:

$$CTQ_{L,gen} = \frac{(UA)_L T_{HS}^{(n_H-1)}}{\dot{m}c_p}, \quad (15)$$

$$CTQ_{H,gen} = \frac{(UA)_H T_{HS}^{(n_H-1)}}{\dot{m}c_p}. \quad (16)$$

- Internal and total entropy production terms use the same heat rate capacity, $\dot{m}c_p$, as reference:

$$\tilde{s}_i = \frac{\dot{S}_i}{\dot{m}c_p}, \quad \tilde{s}_T = \frac{\dot{S}_T}{\dot{m}c_p}. \quad (17)-(18)$$

- The dimensionless constraint relative to a fixed total heat exchanger conductance will have as equivalent:

$$CTQ_{T,gen} = CTQ_{H,gen} + CTQ_{L,gen}, \quad (19)$$

which will be considered as optimization constraint finite parameter, while $CTQ_{H,gen}$ and $CTQ_{L,gen}$ are system optimization variables.

According to the adopted sign convention and the refrigeration machine scheme, the temperatures hierarchy is $T_F < T_{LS} < T_{HS} < T_H$, that imposes the following constraints:

$$\begin{aligned} \theta_L < \theta_{LS} < \theta_{HS} < \theta_H, \quad \theta_{HS} = 1, \\ \tilde{q}_H < 0, \quad \tilde{q}_L > 0, \quad \tilde{w} < 0, \quad \tilde{s}_i > 0. \end{aligned} \quad (20)$$

These constraints will always be valid for any refrigerator being studied, regardless of the presence of any other restrictions imposed by the user or the machine design.

2.4. Optimization method and studied cases

The dimensionless variables that will be used in the optimization procedure are θ_L and $CTQ_{L,gen}$, θ_H and $CTQ_{H,gen}$, while the considered parameters are: θ_{LS} , CTQ_T , \tilde{s}_i , and the imposed performance characteristics.

The optimization is achieved for two cases where different performances are imposed. For a refrigeration machine, the most important performances are the *COP* and the refrigeration load.

The first studied case corresponds to an *imposed value of the COP*, together with an *imposed internal entropy production rate*. The objective functions could be: maximum refrigeration load, minimum total dissipation (minimum total entropy production), or minimum power consumption.

Among them, results will be reported here when maximum refrigerating load is sought, respectively, minimum total entropy production rate.

The second optimization is done for an *imposed refrigerating load*, together with an *imposed internal entropy production rate*. Thus, other objective functions will be sought, such as maximum *COP*, minimum total dissipation, or minimum power consumption. Again, only the first two objective functions previously cited will be considered in this analysis, for the sake of comprehensible presentation but the method is similarly applied.

The optimization procedure uses the *Lagrange Multipliers Method* that will lead to a system of equations for each objective function, and its solution will provide optimal values of the control variables of the system, θ_L , $CTQ_{L,gen}$, θ_H , $CTQ_{H,gen}$, leading to the corresponding system performances achieved in the considered constraint operation. Thus, the Lagrangian function corresponding to the present study is expressed as:

$$L = FO + \lambda_1 C_1 + \lambda_2 C_2, \quad (21)$$

where: *FO* is one of the considered objective functions stated above; λ_1 and λ_2 are the corresponding Lagrange multipliers; C_1 and C_2 are the problem constraints represented by the First Law of Thermodynamics including the imposed performance characteristic constraint and Second Law of Thermodynamics applied to the cycle including the internal entropy production restriction.

The optimal values of the model variables are obtained by solving the following system of equations:

$$\left. \frac{\partial L}{\partial var} \right|_{1-m-var} = 0, \quad \frac{\partial L}{\partial \lambda_1} = 0, \quad \frac{\partial L}{\partial \lambda_2} = 0. \quad (22)$$

2.4.1. Imposed coefficient of performance

The First Law of Thermodynamics combined with eq. (13) and eqs. (11)–(12) that takes into account the expressions of generalized heat transfer laws provides:

$$CTQ_{H,gen}(1 - \theta_H)^{n_H}(-1)^{n_H-1} + CTQ_{L,gen}(\theta_{LS} - \theta_L)^{n_L} \frac{1+COP}{COP} = 0. \quad (23)$$

By generalizing the above expression, the **first constraint equation (C1)** of the model results as:

$$CTQ_{H,gen}f + CTQ_{L,gen} \frac{1+COP}{COP} \xi = 0, \quad (24)$$

where the following notations of the new functions were used:

$$\begin{aligned} f &= (1 - \theta_H)^{n_H}(-1)^{n_H-1} \\ \xi &= (\theta_{LS} - \theta_L)^{n_L} \end{aligned} \quad (25)$$

One notes that *f* is a function of θ_H only, while ξ is a function of θ_L only.

The Second Law of Thermodynamics combined with eqs. (11)–(12) and applied to the cycle gives:

$$\frac{CTQ_{H,gen}(1 - \theta_H)^{n_H}(-1)^{n_H-1}}{\theta_H} + \frac{CTQ_{L,gen}(\theta_{LS} - \theta_L)^{n_L}}{\theta_L} + \tilde{s}_i = 0, \quad (26)$$

and considering the two function introduced by eq. (25), the **second constraint (C2)** of the system results as:

$$CTQ_{H,gen} \frac{f}{\theta_H} + CTQ_{L,gen} \frac{\xi}{\theta_L} + g = 0. \quad (27)$$

One notes that g is a function of both θ_H and θ_L according to the three variation laws adopted for the entropy production rate dependence on temperature, eq. (8)–(10).

1. Refrigeration load as Objective Function

The first objective function considered in the optimization procedure is the refrigeration load:

$$FO = \tilde{q}_L = CTQ_{L,gen} (\theta_{LS} - \theta_L)^{n_L} = CTQ_{L,gen} \xi. \quad (28)$$

The Lagrangian is expressed in this case as:

$$CTQ_{L,gen} \xi + \lambda_1 \left[(CTQ_{T,gen} - CTQ_{L,gen}) f + CTQ_{L,gen} \xi (1 + COP) / COP \right] + \lambda_2 \left[(CTQ_{T,gen} - CTQ_{L,gen}) f / \theta_H + CTQ_{L,gen} \xi / \theta_L + g \right] = 0. \quad (29)$$

The derivatives indicated in (22) are calculated for the considered variables ($CTQ_{L,gen}$, θ_H , θ_L) and the two multipliers (λ_1 , λ_2), as follows:

$$\frac{\partial L}{\partial CTQ_{L,gen}} = 0: \quad \xi + \lambda_1 \left[-f + \xi (1 + COP) / COP \right] + \lambda_2 \left[-f / \theta_H + \xi / \theta_L \right] = 0, \quad (30)$$

$$\frac{\partial L}{\partial \theta_H} = 0: \quad \lambda_1 (CTQ_{T,gen} - CTQ_{L,gen}) f' + \lambda_2 \left[(CTQ_{T,gen} - CTQ_{L,gen}) \frac{\theta_H f' - f}{\theta_H^2} + g'_H \right] = 0, \quad (31)$$

where f' and g'_H are the derivatives of function f and g respectively, with respect to θ_H ;

$$\frac{\partial L}{\partial \theta_L} = 0: \quad CTQ_{L,gen} \xi' + \lambda_1 CTQ_{L,gen} \frac{1 + COP}{COP} \xi' + \lambda_2 \left(CTQ_{L,gen} \frac{\theta_L \xi' - \xi}{\theta_L^2} + g'_L \right) = 0, \quad (32)$$

where ξ' and g'_L are the derivatives of function ξ and g respectively, with respect to θ_L ;

$$\frac{\partial L}{\partial \lambda_1} = 0: \quad (CTQ_{T,gen} - CTQ_{L,gen}) f + CTQ_{L,gen} \frac{1 + COP}{COP} \xi = 0, \quad (33)$$

$$\frac{\partial L}{\partial \lambda_2} = 0: \quad (CTQ_{T,gen} - CTQ_{L,gen}) \frac{f}{\theta_H} + CTQ_{L,gen} \frac{\xi}{\theta_L} + g = 0. \quad (34)$$

By combining eqs. (30)–(32) and after terms rearrangement one equation results that associated with eqs. (33), (34) and (19) leads to a system that is solved by Newton-Raphson method with respect to $CTQ_{L,gen}$, θ_H and θ_L , since $CTQ_{H,gen}$ is expressed by the last equation of the system.

$$\begin{aligned} \frac{f^2}{\theta_H^2 f'} - \frac{f g'_H}{CTQ_{H,gen} f'} &= \frac{\xi^2}{\theta_L^2 \xi'} - \frac{\xi g'_L}{CTQ_{L,gen} \xi'} \\ CTQ_{H,gen} f + \frac{1 + COP}{COP} CTQ_{L,gen} \xi &= 0 \\ CTQ_{H,gen} \frac{f}{\theta_H} + CTQ_{L,gen} \frac{\xi}{\theta_L} + g &= 0 \\ CTQ_{H,gen} &= CTQ_{T,gen} - CTQ_{L,gen} \end{aligned} \quad (35)$$

Note that the system has to be solved for each of the three non-dimension functions g corresponding to the internal entropy production rate, eqs. (8)–(10).

2. Total entropy production rate as Objective Function

One considers here a second objective function, the total entropy production rate, expressed as:

$$FO = \tilde{s}_T = -\tilde{q}_H - \frac{\tilde{q}_L}{\theta_{LS}} = -CTQ_{H,gen}f - CTQ_{L,gen}\frac{\xi}{\theta_{LS}}. \quad (36)$$

The Lagrangian becomes in this case:

$$\begin{aligned} & -\left(CTQ_{T,gen} - CTQ_{L,gen}\right)f - CTQ_{L,gen}\frac{\xi}{\theta_{LS}} + \\ & + \lambda_1 \left[\left(CTQ_{T,gen} - CTQ_{L,gen}\right)f + CTQ_{L,gen}\xi(1+COP)/COP \right] + \\ & + \lambda_2 \left[\left(CTQ_{T,gen} - CTQ_{L,gen}\right)f/\theta_H + CTQ_{L,gen}\xi/\theta_L + g \right] = 0 \end{aligned} \quad (37)$$

The derivatives corresponding to the three variables ($CTQ_{L,gen}$, θ_H , θ_L) and the two multipliers (λ_1 , λ_2) are:

$$\frac{\partial L}{\partial CTQ_{L,gen}} = 0: \quad f - \frac{\xi}{\theta_{LS}} + \lambda_1 \left(-f + \frac{1+COP}{COP}\xi \right) + \lambda_2 \left(-\frac{f}{\theta_H} + \frac{\xi}{\theta_L} \right) = 0, \quad (38)$$

$$\frac{\partial L}{\partial \theta_H} = 0: \quad \left(CTQ_{T,gen} - CTQ_{L,gen}\right)f'(\lambda_1 - 1) + \lambda_2 \left[\left(CTQ_{T,gen} - CTQ_{L,gen}\right)\frac{\theta_H f' - f}{\theta_H^2} + g'_H \right] = 0, \quad (39)$$

$$\frac{\partial L}{\partial \theta_L} = 0: \quad \frac{CTQ_{L,gen}\xi'}{\theta_{SL}} + \lambda_1 CTQ_{L,gen}\frac{1+COP}{COP}\xi' + \lambda_2 \left[CTQ_{L,gen}\frac{\theta_L \xi' - \xi}{\theta_L^2} + g'_L \right] = 0, \quad (40)$$

where f' , g'_H , ξ' and g'_L have the same meaning as in eq. (31) and (32);

$$\frac{\partial L}{\partial \lambda_1} = 0: \quad \left(CTQ_{T,gen} - CTQ_{L,gen}\right)f + CTQ_{L,gen}\frac{1+COP}{COP}\xi = 0, \quad (41)$$

$$\frac{\partial L}{\partial \lambda_2} = 0: \quad \left(CTQ_{T,gen} - CTQ_{L,gen}\right)\frac{f}{\theta_H} + CTQ_{L,gen}\frac{\xi}{\theta_L} + g = 0. \quad (42)$$

Similar to the previous case, the corresponding equation system results as:

$$\begin{aligned} & \frac{f^2}{\theta_H^2 f'} - \frac{f g'_H}{CTQ_{H,gen} f'} = \frac{\xi^2}{\theta_L^2 \xi'} - \frac{\xi g'_L}{CTQ_{L,gen} \xi'} \\ & CTQ_{H,gen}f + \frac{1+COP}{COP}CTQ_{L,gen}\xi = 0 \\ & CTQ_{H,gen}\frac{f}{\theta_H} + CTQ_{L,gen}\frac{\xi}{\theta_L} + g = 0 \\ & CTQ_{H,gen} = CTQ_{T,gen} - CTQ_{L,gen} \end{aligned} \quad (43)$$

The comparison of the systems from eq. (35) and (43) shows that they are identical, meaning that no matter the objective function is considered, the same system of equations results. Thus, the optima of the two objective functions (maxima and minima, respectively) correspond to the same values of the model variables resulting from the solution of the equation system. The interpretation of this result lies on the fact that the total entropy production given by eq. (3) and written in non-dimensional form becomes:

$$\tilde{s}_T = -\tilde{q}_H - \frac{\tilde{q}_L}{\theta_{LS}}. \quad (44)$$

By considering eq. (1) in non-dimensional form and eq. (13), eq. (44) is re-written as:

$$\tilde{s}_T = -\tilde{q}_L \left(\frac{1}{COP} - 1 + \frac{1}{\theta_{LS}} \right). \quad (45)$$

As COP and θ_{SL} are imposed for the system operation, the optimum for \tilde{q}_L will lead to optimum for \tilde{s}_T in the same time. More specific, maximum of \tilde{q}_L will lead to minimum of \tilde{s}_T , since the minimum of entropy production is directly correlated to minimum of consumed power, which at its turn is linked by eq. (13) to maximum of \tilde{q}_L for imposed COP .

2.4.2. Imposed refrigeration load

The same optimization method is applied in this case, but the terms in equations are expressed function on the imposed non-dimensional refrigeration load, \tilde{q}_L , instead of COP . Thus, the first constraint equation (C1) of the model, given by eq. (24) in the previous case, becomes:

$$\tilde{q}_L - CTQ_{L,gen} \xi = 0, \quad (46)$$

while the second constraint equation (C2) remains the same as in the previous case of imposed COP , given by eq. (27).

The objective function FO in eq. (21) is either the COP , or the non-dimensional entropy production \tilde{s}_T . The Lagrangian is computed and a similar system of equations as eq. (35) or (44) is obtained and solved with respect to the same variables $CTQ_{L,gen}$, θ_H , θ_L :

$$\begin{aligned} \frac{f^2}{\theta_H^2 f'} - \frac{f g'_H}{CTQ_{H,gen} f'} &= \frac{\xi^2}{\theta_L^2 \xi'} - \frac{\xi g'_L}{CTQ_{L,gen} \xi'} \\ \tilde{q}_L - CTQ_{L,gen} \xi &= 0 \\ CTQ_{H,gen} \frac{f}{\theta_H} + CTQ_{L,gen} \frac{\xi}{\theta_L} + g &= 0 \\ CTQ_{H,gen} &= CTQ_{T,gen} - CTQ_{L,gen} \end{aligned} \quad (47)$$

3. Results

The optimization method described above leads to a system of non-linear equations that is numerically solved in Matlab by applying Newton-Raphson method. The system parameters set to constant values in the numerical simulation are: total coefficient of heat transfer $CTQ_{T,gen} = 2$, non-dimensional low source temperature $\theta_{LS} = 0.87$ (corresponding to a temperature of the cold source of -5°C in case of an ambient temperature of 35°C) and the internal entropy production \tilde{s}_i for which different cases have been simulated, as given by eq. (8)-(10), where the “const” term was set to a constant value of 0.005. Since the obtained results are similar for the three variation profiles, for the sake of comprehensibility, only the results for constant internal entropy production will be further presented. The other constant values considered correspond to the imposed constraints, such as coefficient of performance COP or non-dimensional refrigeration load \tilde{q}_L .

The system is solved with respect to the unknown solutions $CTQ_{L,gen}$, θ_L and θ_H . The obtained physical solutions allow computing the system performances in terms of \tilde{q}_L or COP and the other non-dimensional energy rates \tilde{q}_H , \tilde{w} . Also, the total entropy production \tilde{s}_T is computed.

The results are presented for the two considered cases, namely imposed COP and imposed refrigeration load, respectively.

3.1. Imposed coefficient of performance

For the numerical simulation, its value has been set to $COP = 1.5$ and for the internal entropy production $\tilde{s}_i = 0.005$ was considered. After solving the system of equations, two physical solutions θ_L and θ_H are obtained. These two solutions correspond to two different operating regimes, namely one of maximum refrigerating load, denoted by subscript 1, and the other leading to minimum total entropy production, denoted by subscript 2 in the figures. As it was analytically proved, the extrema in both regimes correspond to the same optimum values of system parameters. Results are further presented for the case of linear and non-linear heat transfer laws.

3.1.1. Linear heat transfer laws

In Figure 2a non-dimensional energy rates and total entropy production are presented for the two operating regimes function on the control variable θ_L . The upper part of the plots corresponds to the maximum refrigerating load regime, while the lower one corresponds to the minimum total dissipation regime. One might notice that the most convenient operating parameters are those located in the left part of the maximum values, since here the consumed power and total dissipation are lower for lower cold source temperatures. The non-dimensional refrigerating load is plotted versus θ_L generating loop curves as shown in Figure 2b. One chosen value for the refrigerant temperature θ_L might indicate the operation at two refrigerating loads, on the upper part and the lower one respectively.

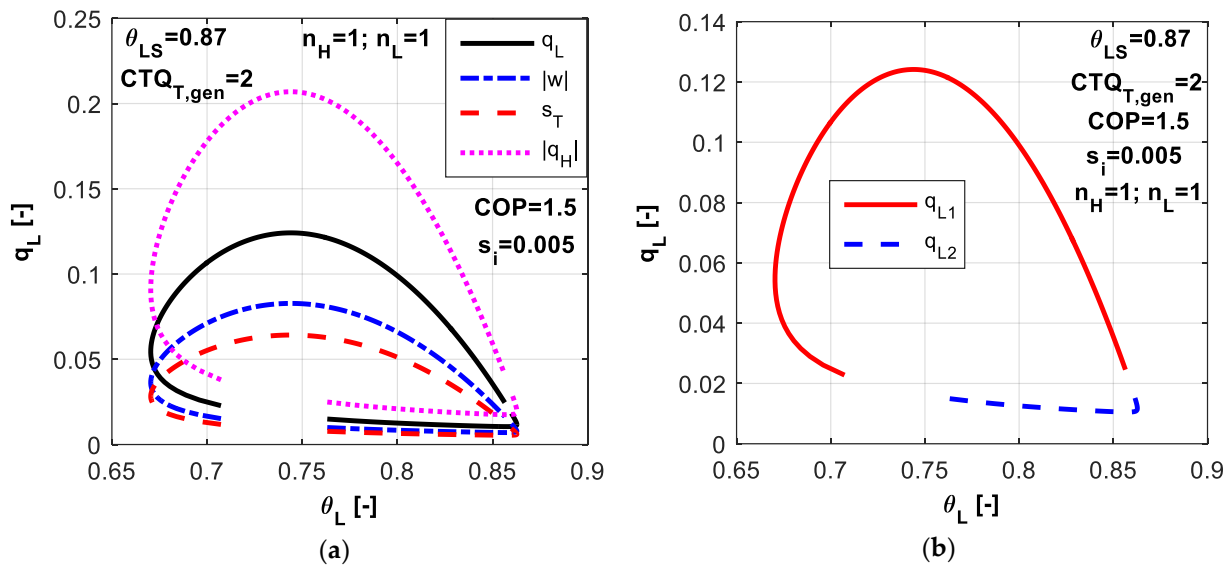


Figure 2. Imposed COP and \tilde{s}_i constant: (a) Non-dimensional energy rates and total entropy production for the two operating regimes; (b) Non-dimensional refrigerating load for the two operating regimes.

Also, one desired value of the refrigerating load could be associated to a refrigerating temperature closer to the reference one (ambient) as seen in the right-hand-side of the plot, or contrary to a lower temperature value in the left-hand-side. The choice of the operating regime depends on the system settings and the other variables that are behind (θ_H , $CTQ_{L,gen}$, etc).

Figure 3 presents a T - s like diagram. The possible range values are emphasized for the refrigerant temperatures at the hot and cold reservoirs. Also one might notice that the second solution of the system of equations (θ_{L2} , θ_{H2}) corresponds to the lower dissipation rates, but also to the lower energy rates. Thus, the second solution might not be of interest from the operation regime point of view.

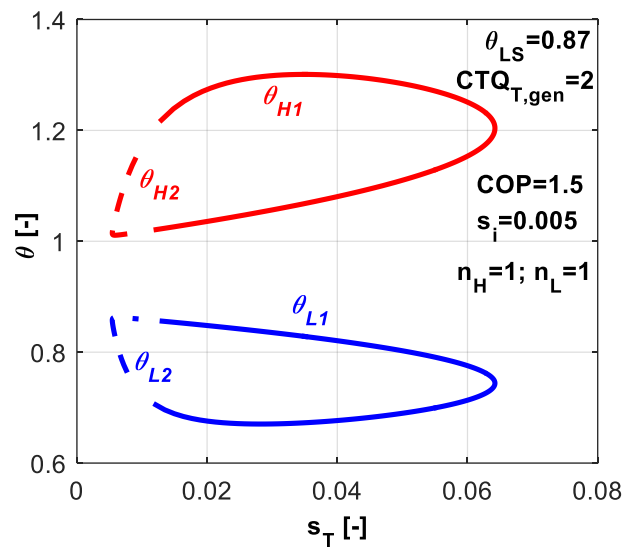


Figure 3. The two solutions of the non-dimensional refrigerant temperatures function on \tilde{s}_T for the case with imposed COP and \tilde{s}_i constant.

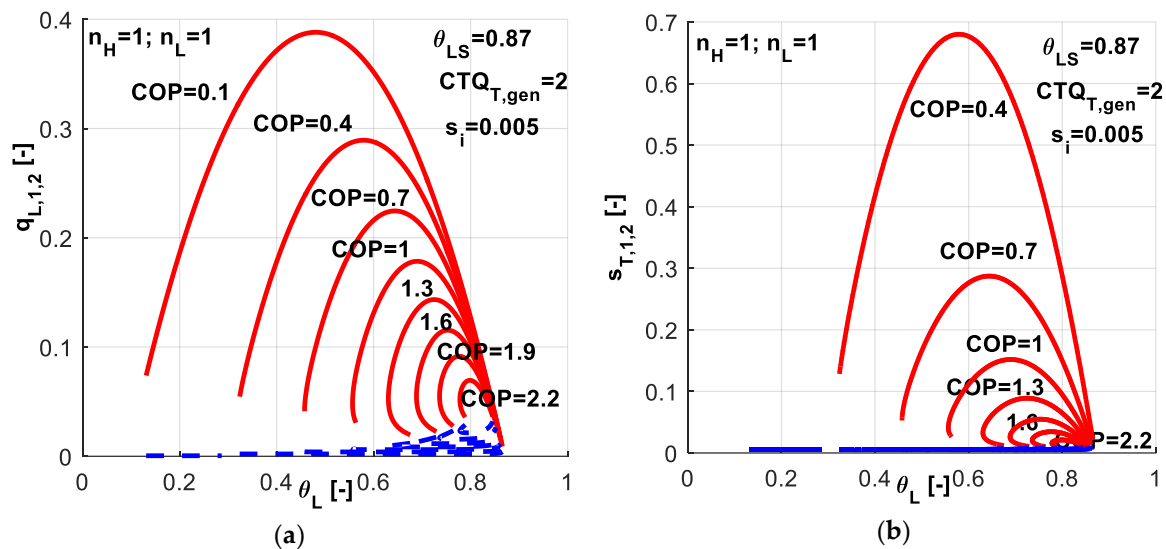


Figure 4. Imposed COP and \tilde{s}_i constant - sensitivity study with respect to COP (a) Non-dimensional refrigeration load for the two regimes; (b) Non-dimensional total entropy production for the two regimes.

A sensitivity study with respect to COP was performed. It revealed the possible range for system variables and the associated performances. For the maximum refrigeration load regime, as higher COP values constraint the operation, as lower the refrigerating loads are and the range for refrigerant temperature becomes narrower, as emphasized by red upper curves in Figure 4a. For the minimum total dissipation regime (blue down curves in Figure 4a), the higher COP is imposed, the higher refrigerating load is. This particular behavior of the system is explained by the refrigerant temperature variation and heat exchanger inventory distribution in terms of $CTQ_{L,gen}$ and $CTQ_{H,gen}$ associated to this solution of the equations system. Also one might notice that when the system operates at an imposed COP value, there is a certain refrigerant temperature θ_L leading to a maximum value of the refrigerating load. The corresponding total entropy production is presented in Figure 4b.

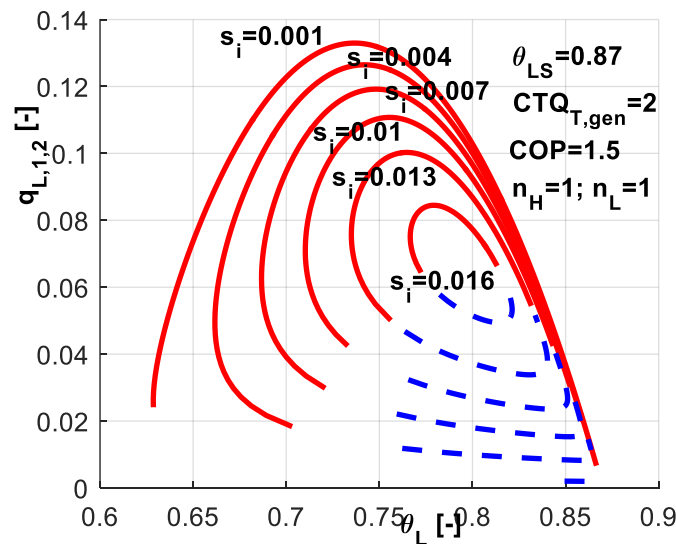


Figure 5. Imposed COP and \tilde{s}_i constant - sensitivity study with respect to \tilde{s}_i . Non-dimensional refrigeration load for the two regimes.

The effect of internal entropy rate \tilde{s}_i on system variables and performances was studied, too. Its value is closely related to the built machine. Figure 5 reveals that higher \tilde{s}_i values are accompanied by a narrower range for system variables and obviously lower performances. It is interesting to notice that when operating at higher values of the refrigerant temperature θ_L , the effect of \tilde{s}_i is more reduced as emphasized by the curves tangent region in the right-hand-side of the plot. Contrary, its effect is very important in the region of lower refrigerant temperatures – left-hand-side of the curves.

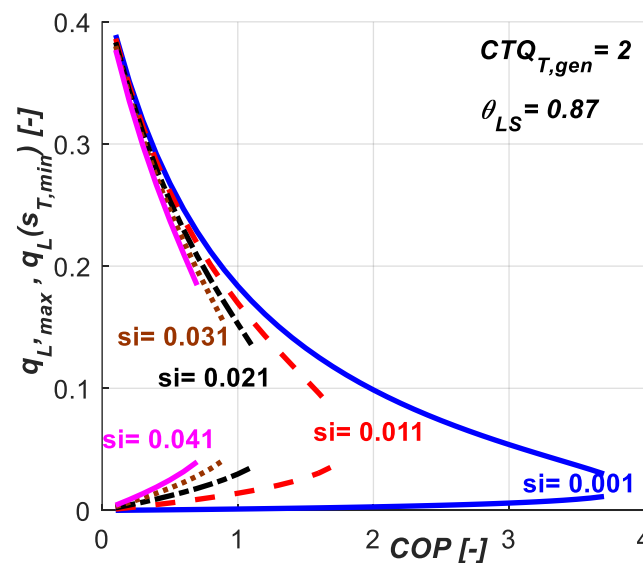


Figure 6. Maximum refrigeration load for a given set of COP and \tilde{s}_i constant values.

Figure 6 presents the maximum values for the refrigeration load that could be achieved for a given set of COP and \tilde{s}_i values. It also emphasizes the effect of \tilde{s}_i on the possible operating range of the system under these constraints. Lower \tilde{s}_i values let the system operates on a wider range of COP values. Increasing \tilde{s}_i values limits the system performances in terms of achievable COP and, diminishes the maximum refrigeration load attainable limit (the upper curves) for a given COP value.

3.1.2. Non-linear heat transfer laws

In Figure 7 one might analyze the effect of different nature of heat transfer laws at the source and sink on the non-dimensional refrigerating load (Figure 7a) and total entropy production (Figure 7b). Symmetric and non-symmetric combinations of exponents n_H and n_L have been chosen. The value of 3 corresponds to a phase change process. The nature of the considered heat transfer laws affects the optimum values of system parameters for obtaining maximum system performances and also the magnitude of system performances. As revealed by Figure 7a, a phase change process at the cold source ($n_L = 3$) involves lower refrigerating loads and shifts the maximum achievable one towards lower temperature values; the peak of the curves is shifted towards left with respect to the linear case ($n_H = n_L = 1$). By constraining the operation at a COP value of 0.4 leads to a maximum achievable non-dimensional refrigerating load of about 0.28 in the case of linear heat transfer laws, and 0.06 respectively for the case of phase change processes at both reservoirs. The justification relies on the temperature difference between the two heat sources and the refrigerant that also depends on the heat transfer law, becoming lower for the phase change processes.

We also notice that the nature of the heat transfer law influences the range for possible values of the system variables, especially in the case of phase-change processes.

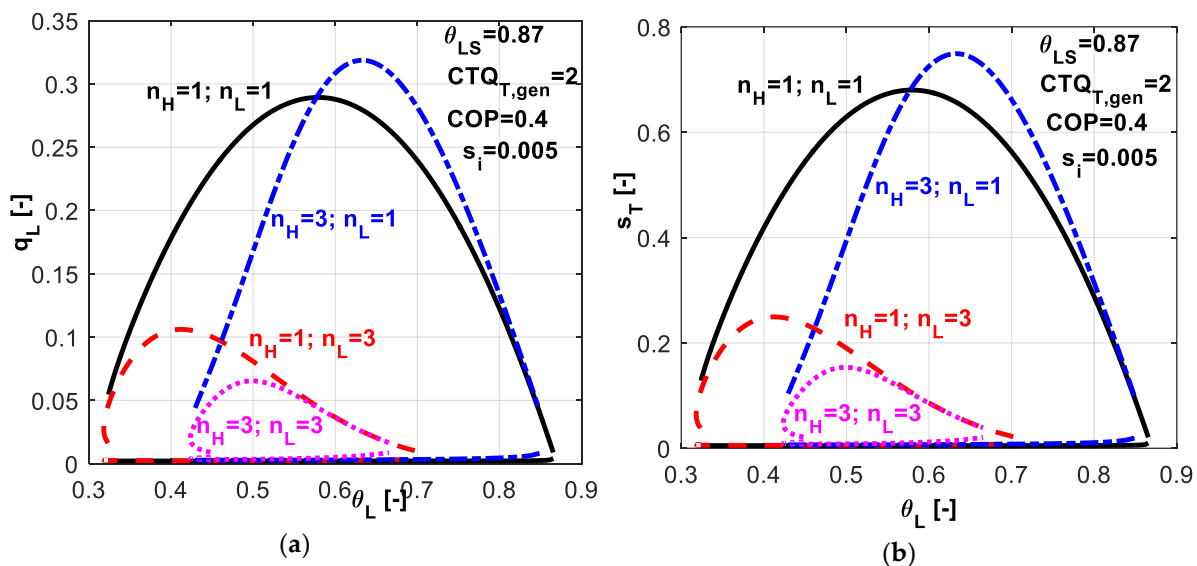


Figure 7. Imposed COP and \tilde{s}_i constant - sensitivity study with respect to heat transfer laws nature at both reservoirs (a) Non-dimensional refrigerating load; (b) Non-dimensional total entropy production.

A sensitivity study with respect to COP values associated to different combinations of heat transfer laws emphasize the results obtained in Figure 8. One might notice that the different considered combinations of exponents n_H and n_L lead to different possible ranges for COP as a constraint. Maintaining the same settings for the machine as previously ($CTQ_{T,gen} = 2$, $\theta_{LS} = 0.87$, $\tilde{s}_i = 0.005$), one observes that:

- in the case of linear heat transfer laws at both reservoirs, the COP values could be imposed up to a maximum value of 2.2 (Figure 4);
- when a phase change process is considered at the cold source (Figure 8a-b), the maximum possible COP is about 0.9;

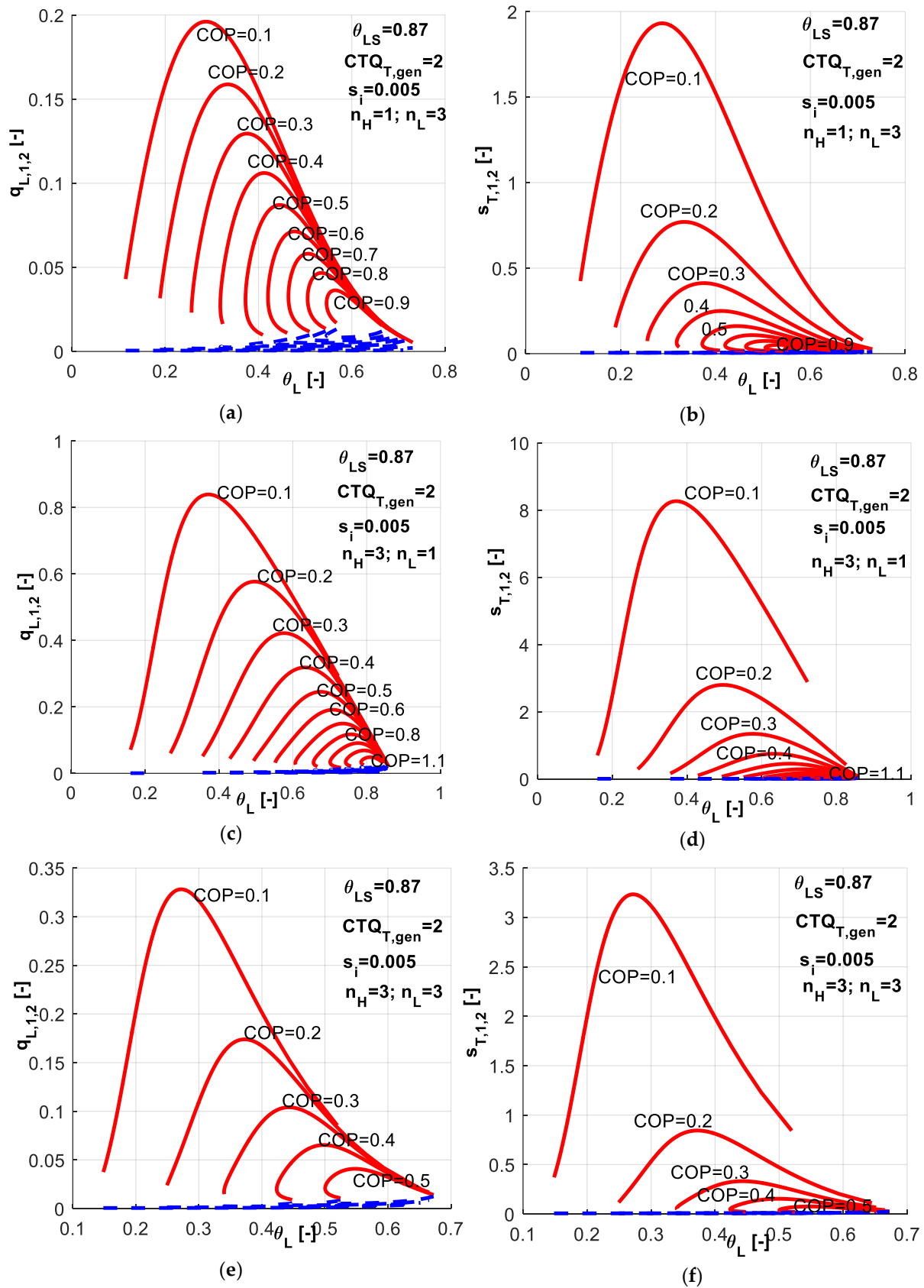


Figure 8. COP and \bar{s}_i constant constraints - sensitivity study with respect to heat transfer laws nature at both reservoirs for different combinations (n_H , n_L).

- if the phase change process is considered at the hot sink (Figure 8c-d), the maximum possible COP is about 1.1;

- while in the case of phase change process at both reservoirs (Figure 8e-f), the COP is limited to 0.5. In fact, this is the reason for which a value of 0.4 was considered for COP in the simulations presented in Figure 7.

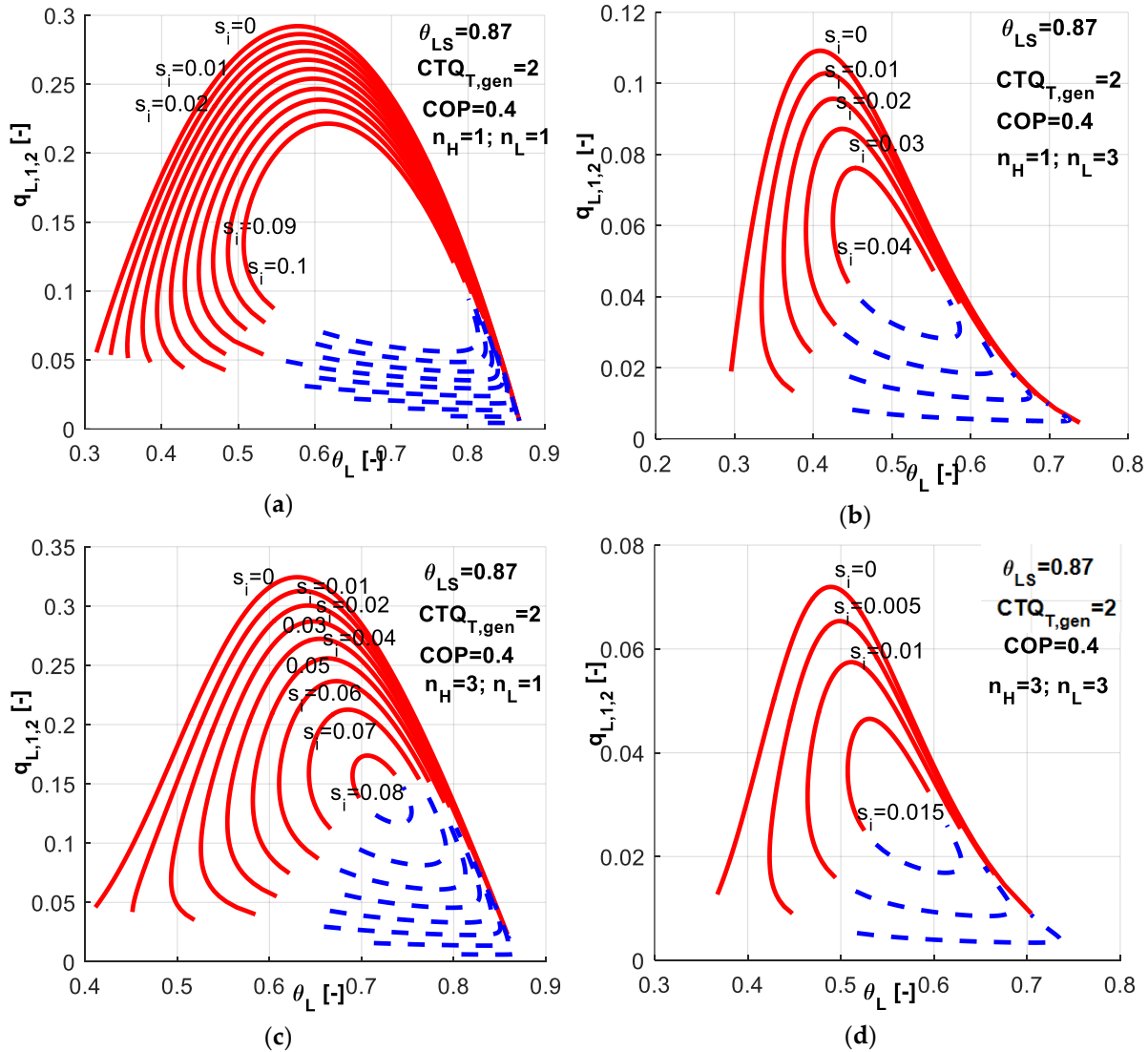


Figure 9. Imposed COP and \tilde{s}_i constant – non-dimensional refrigerating load for different combinations of heat transfer laws at both reservoirs (n_H, n_L).

Figure 9 presents the combined effect of heat transfer laws and \tilde{s}_i constraint values on the non-dimensional refrigerating load and possible range values of refrigerant cold temperature. The first observation regards the \tilde{s}_i constraint values. Linear heat transfer laws allow the system to operate at higher internal irreversibilities (Figure 9a) when needed, while a phase change process occurring at a heat source limits the operation to lower internal irreversibilities (Figure 9b-c). The most restrictive case is the one at which phase change processes occur at both reservoirs (Figure 9d), for which the maximum acceptable constraint in terms of \tilde{s}_i is about 0.015 under the other given settings ($CTQ_{T,gen}, \theta_{LS}$). The second observation refers to the range of achievable θ_L values. The limit case of endoreversible operation ($\tilde{s}=0$) is obviously the most generous in all cases, while the most restrictive one is the case of phase change processes occurring at both reservoirs. For a given value of $\tilde{s}_i=0.04$ for example, the linear case allows θ_L to be obtained from 0.4 to 0.9 (Figure 9a). When phase change occurs at the cold source (Figure 9b), θ_L is limited to the range 0.42-0.58, while if the phase

change occurs at the hot sink (Figure 9c), the range is 0.55-0.81. The four combinations of heat transfer laws (Figure 9d) do not allow at all the system operation in this case.

The above results contribute to the understanding of the system limits in operation under different constraints and allow the engineer to choose the most convenient variables in certain circumstances.

3.2. Imposed refrigeration load

Similar to the previous case, a numerical simulation was carried out when imposing the refrigeration load instead of COP . The considered imposed values for the non-dimensional parameters are $\tilde{q}_L = 0.1$ and $\tilde{s}_i = 0.005$. The other parameters have the same values, namely $\theta_{LS} = 0.87$ and $CTQ_{T,gen} = 2$.

This time the system of equations reveals only one physical solution θ_L and θ_H . This solution corresponds to maximum COP operating regime and in the same time to minimum total entropy production one, as proved before (the extrema in both regimes correspond to the same optimum values of system parameters). Results are further presented for the case of linear and non-linear heat transfer laws.

3.2.1. Linear heat transfer laws

For the case of linear heat transfer laws, the results are plotted in Figure 10. One might notice a large range for the refrigerant cold non-dimensional temperature θ_L and the existence of an optimum value corresponding to minimum non-dimensional power consumption \tilde{w} and minimum total entropy production \tilde{s}_T (Figure 10a). This optimum value leads also to maximum COP as revealed by Figure 10b. Numerically, this point tends to equal distribution of CTQ between sources, namely $CTQ_{L,gen} = CTQ_{H,gen}$. The results are similar to the previous case where the imposed value of COP leads to an optimum value of θ_L corresponding to maximum \tilde{q}_L .

In Figure 11 the non-dimensional temperatures θ_L and θ_H are plotted versus dimensionless total entropy production \tilde{s}_T . Obviously, the cycle operating between minimum difference in refrigerant temperatures is accompanied by the minimum total dissipation (left extrema in Figure 11). As this difference increases, the total dissipation increases.

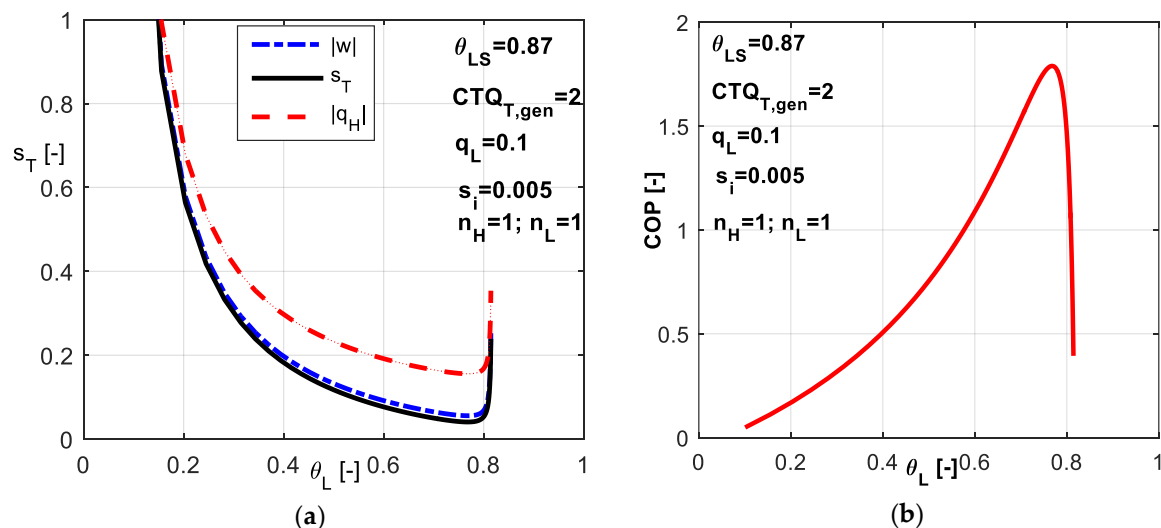


Figure 10. Imposed \tilde{q}_L and \tilde{s}_i constant (a) Non-dimensional energy rates and total entropy production; (b) COP variation with respect to θ_L .

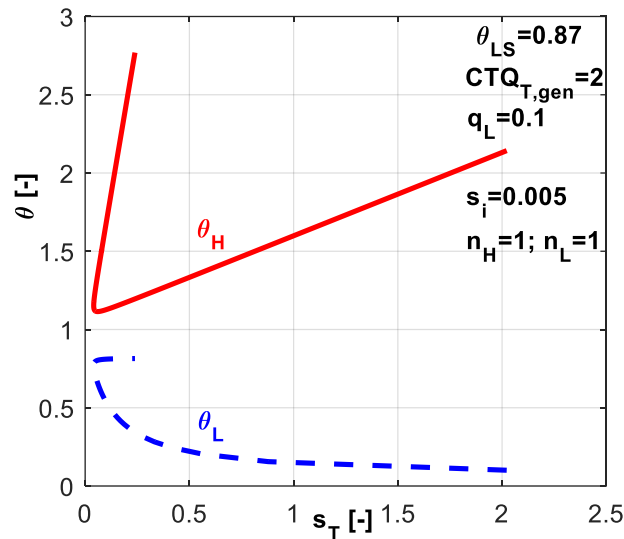


Figure 11. Non-dimensional refrigerant temperatures at hot and cold reservoirs for the case of imposed \tilde{q}_L and \tilde{s}_i constant.

A sensitivity study with respect to the refrigerating load emphasizes the variation range for dimensionless cold refrigerant temperature θ_L limiting the values of COP (Figure 12a). The behavior is similar to the one presented for the previous case in Figure 4a limiting the values of \tilde{q}_L when imposing COP . For each imposed \tilde{q}_L value, an optimum θ_L value exists for which the system COP is maximum. Higher values of constraint refrigeration load \tilde{q}_L diminish the COP values and shift the maximum COP towards lower cold refrigerant temperature values θ_L . For the chosen set of system parameters, a dimensionless refrigeration load of 0.3 is the maximum attainable in the range of 0.35-0.62 for θ_L values. The internal entropy production effects are emphasized in Figure 12b. The most important effect is noticed on limiting the COP values, rather than limiting θ_L .

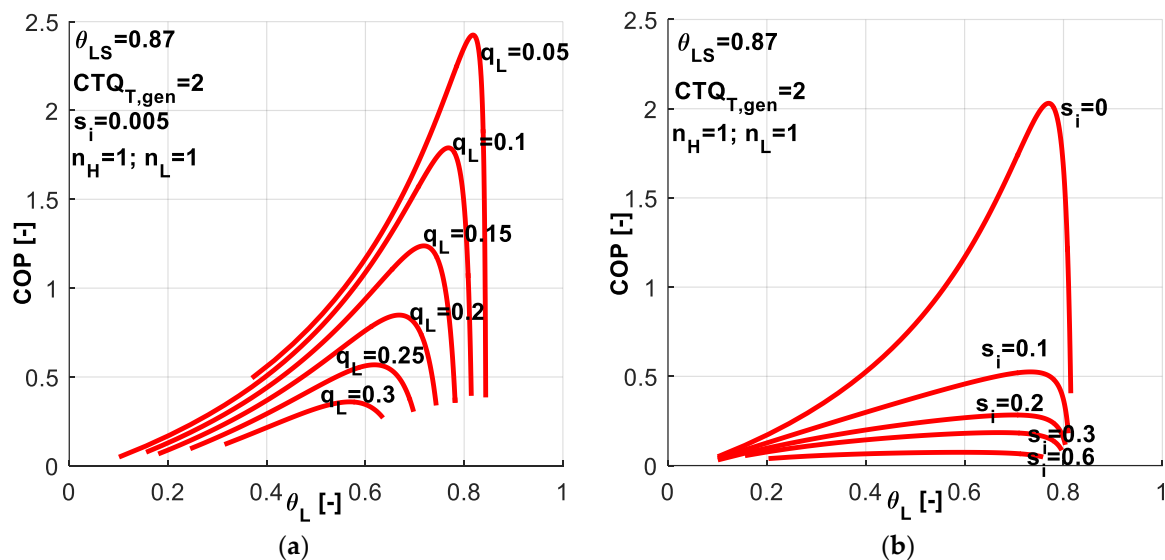


Figure 12. COP variation for the case with imposed \tilde{q}_L and \tilde{s}_i constant (a) Sensitivity with respect to dimensionless refrigerating load; (b) Sensitivity with respect to dimensionless internal entropy production.

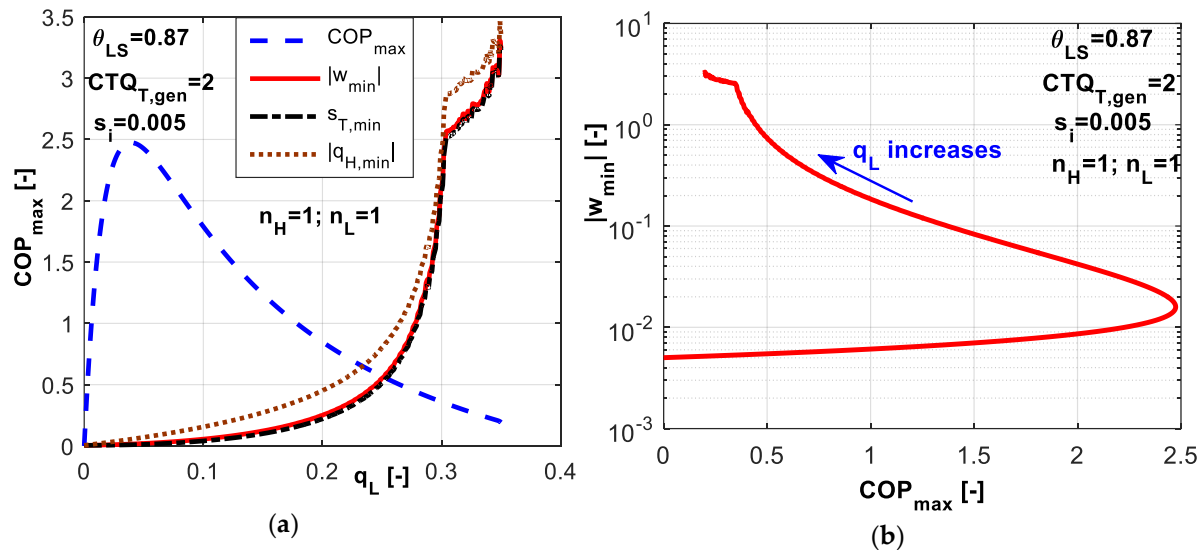


Figure 13. \tilde{q}_L influence: (a) Dimensionless energy rates and COP variation – sensitivity with respect to \tilde{q}_L ; (b) Dimensionless consumed power versus maximum achievable COP .

Figure 13 presents the variation of maximum COP values and the corresponding minimum energy rates and total entropy dissipation with dimensionless refrigerating load. A maximum value among the COP maxima is obtained for $\tilde{q}_L = 0.04$ in the given conditions. These types of plots could be used to set the system operating regimes for maximum performances under some imposed constraint. Figure 13b emphasizes the variation of dimensionless consumed mechanical power with maximum COP , at logarithmic scale, for increasing \tilde{q}_L values. This plot puts into evidence that the chosen operating regime is a trade-off between COP and refrigerating load \tilde{q}_L , for high value of \tilde{q}_L . The lower part of the curve corresponds to low values of \tilde{q}_L and \tilde{w} , and the value of maximum COP appears as a transition point.

3.2.2. Non-linear heat transfer laws

The effect of different combinations of heat transfer laws on the system COP is emphasized in Figure 14. The most favorable combination from COP point of view is the case of linear heat transfer laws at both sides, while the most constrainable is the one considering phase change processes at both sides. These two combinations constraint the COP values and also the range for allowable θ_L values.

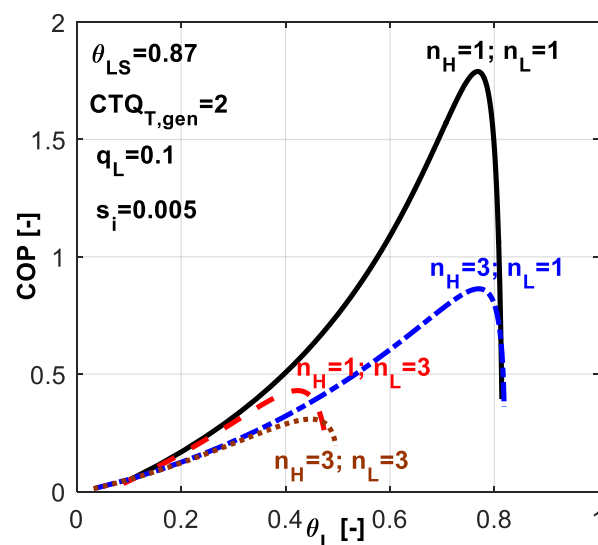


Figure 14. COP variation for the case of imposed \tilde{q}_L and \tilde{s}_i constant - sensitivity study with respect to heat transfer laws nature at the source and sink.

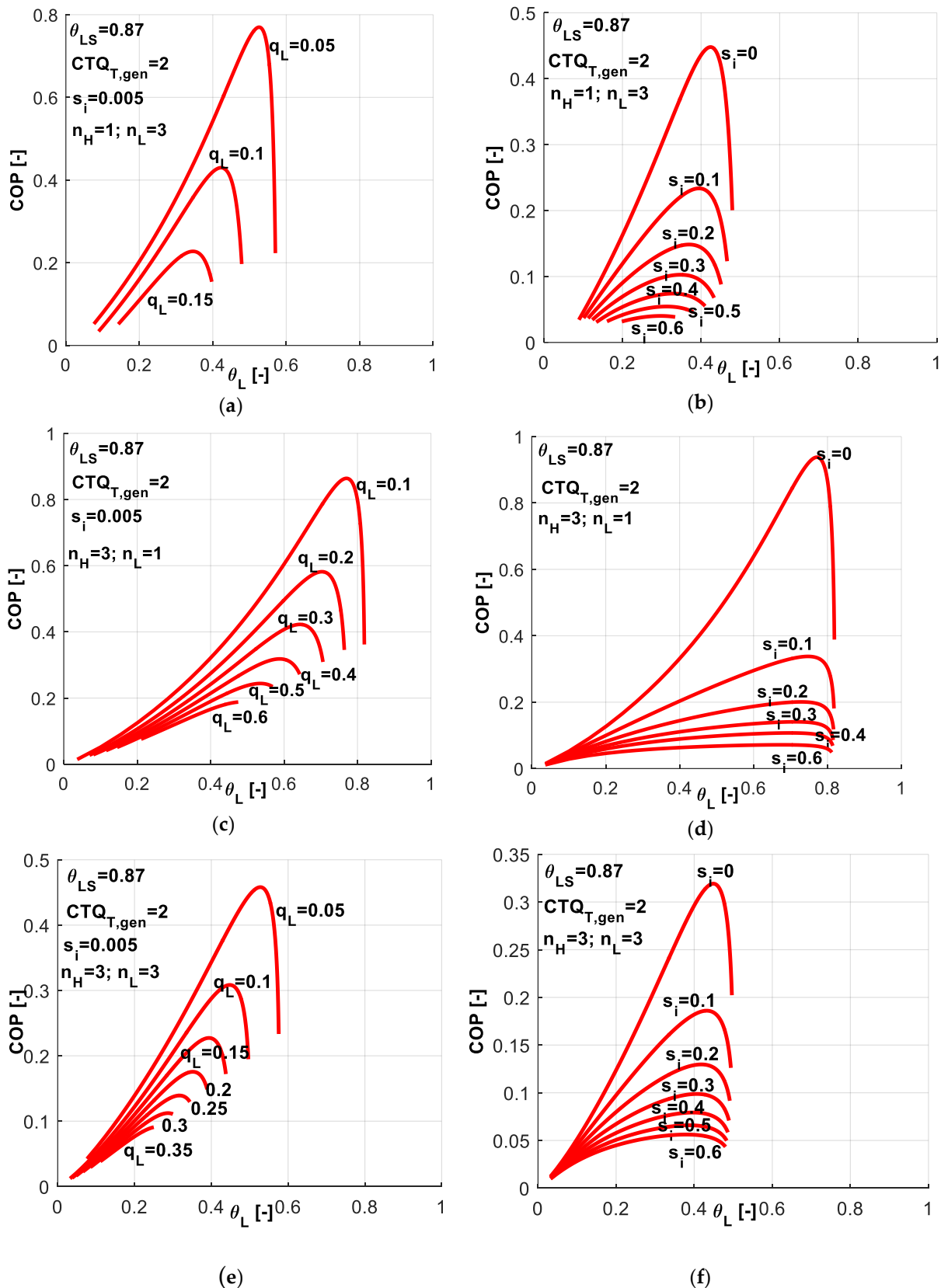


Figure 15. COP variation for the case of imposed \tilde{q}_L and \tilde{s}_i constant, for different combinations of heat transfer laws - sensitivity study with respect to \tilde{q}_L and \tilde{s}_i .

Figure 15 presents a sensitivity study with respect to \tilde{q}_L and \tilde{s}_i values associated to different combinations of heat transfer laws. This study is similar to the one presented in Figure 8 when imposing COP instead of \tilde{q}_L . One might notice that the different considered combinations of

exponents n_H and n_L lead to different possible ranges for \tilde{q}_L as a constraint and for θ_L as control variable. Maintaining the same settings for the machine in all cases ($CTQ_{T,gen} = 2$, $\theta_{LS} = 0.87$, $\tilde{s}_i = 0.005$), one observes that:

- in the case of linear heat transfer laws at both reservoirs, the \tilde{q}_L values could be imposed up to a maximum value of about 0.3 (Figure 12a); the limits in θ_L are quite large and the θ_L values are naturally decreasing as \tilde{q}_L decreases;
- when a phase change process is considered at the cold source (Figure 15a-b), the maximum possible \tilde{q}_L is about 0.15, half of the above-mentioned value; also the limits for θ_L values are much reduced;
- if the phase change process is considered at the hot sink (Figure 15c-d), the maximum possible \tilde{q}_L is about 0.6, so twice with respect to the linear case; the limits for θ_L values are comparable to those obtained for the linear case;
- while in the case of phase change process at both reservoirs (Figure 15e-f), the \tilde{q}_L is limited to 0.35, and thus the effects of the two phase change processes are cancelling each other when comparing to the linear laws; θ_L values are more reduced in this case.

One might deduce from these results that when a system is supposed to undergo a deep cooling process, the nature of the heat transfer law is very important. In this regard, Figure 16 reveals the effect of chosen heat transfer law for such applications, when the required dimensionless refrigerating load is more important, here $\tilde{q}_L = 0.3$.

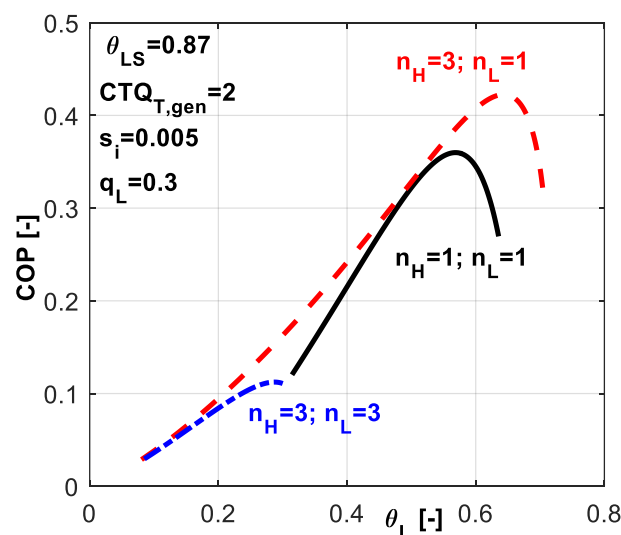


Figure 16. COP variation for the case of imposed \tilde{q}_L and \tilde{s}_i constant - sensitivity study with respect to heat transfer laws nature at the reservoirs.

One might notice that for this application, the best alternative is to choose a linear heat transfer law at the sink and a phase change process at the hot source ($n_H = 3$, $n_L = 1$). This combination leads to maximum COP for a required cold temperature and also it allows the system operation over a wider range of θ_L values. From Figure 15c one may also notice that this combination ensures the system operation at heavier refrigerating loads.

4. Conclusions-perspectives

A general model for the study and optimization of irreversible refrigeration Carnot-like machines was presented. The study started with machines optimization under different constraints and ended with important aspects related to the intrinsic phenomena affecting the systems operation.

The results confirm that the First and Second Law of Thermodynamics are very useful tools in optimization problems under constraints.

The system limits in operation under different constraints were emphasized from the point of view of possible values for system control variables (such as θ_L), for imposed constraint (COP or \tilde{q}_L) and achievable performances (\tilde{q}_L or COP , \tilde{s}_T , other dimensionless energy rates). Moreover, the results are useful in deciding the most convenient values and heat transfer laws in particular circumstances and for specific applications.

The above results contribute to the understanding of the system limits in operation under different constraints and allow the engineer to choose the most convenient variables in given circumstances.

Optimal variables are found for the best performances that the system could achieve under specified constraints. Also the limits of the system operation are determined.

Further development of the model for other constraint (imposed \tilde{w}) is made. Also, a comparison of the direct and reverse machine models and results are under consideration. It seems very promising and gives a new perspective on their optimization by an unified approach.

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