

1 Article

2 The role of the cell volume-area ratio in 3 thermodynamic analysis of the cancer growth control 4 for *in vitro* experiments

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12 **Abstract:** From a thermodynamic point of view, living cell life is no more than a cyclic process. It
13 starts with the newly separated daughter cells and restarts when the next generations grow as free
14 entities. In this cycle the cell changes its entropy. In cancer the growth control is damaged. In this
15 paper we analyze the role of the volume-area ratio in cell in relation to the heat exchange between
16 cell and its environment in order to point out the effect on the cancer growth. The result holds to a
17 possible control of the cancer growth based on the heat exchanged by the cancer towards its
18 environment, and the membrane potential variation, with the consequence of controlling the ions
19 fluxes and the related biochemical reactions. This second law approach could represent a starting
20 point for a possible future support for the anticancer therapies, in order to improve their
21 effectiveness for the untreatable cancers.

22 **Keywords:** Biothermodynamics; Complex systems; Thermodynamics of biological systems;
23 Biophysical resonance.
24

25 1. Introduction

26 Living cells live in an environment which slowly changes its chemical and physical properties.
27 In mammalian animals, organs and tissues are generally endowed with a homeostatic capability,
28 which is characterized by internal thermal regulation.

29 A living system can be defined as an open system, so non-equilibrium thermodynamics, can be
30 applied just to living systems [1].

31 Cancer is no more than a living cell with a different growth behaviour in relation to the cell we
32 consider normal. At a definite phase of its life any cancer cell divides into two cells. The size of the
33 single cell at the beginning of division can vary [2]; consequently, also the relative size of the two
34 daughter cells can vary, but lower and upper bounds on cell volume have been observed [3].

35 Living cell life, cancer included, is a cyclic process, which starts with the newly separated
36 daughter cells and restarts when the next generations grow as free entities. Consequently,
37 thermodynamics of cycle processes can represent a good tool for the analysis of cancer. In
38 thermodynamics is fundamental the definition of the control volume: in this context it could be
39 useful to consider the single cell as the observed control volume, but the usual experimental setup
40 doesn't allow us to introduce such approach, so we are forced to observe a multitude of single cells
41 and consider this multitude as a complex cooperative system of single cells. This system, considered
42 as a single cell or as the cooperative system of all cells in a culture, is no more than an open system
43 from a thermodynamic point of view, so, energy and matter flow through the border of the system,
44 while biochemical and biophysical transformations occur within the system, with a related net

45 production of entropy. The environment of the system considered is composed by the suspending
46 aqueous solution of cell nutrients, the substances discarded by the cells, the gaseous atmosphere
47 above the suspending solution. Consequently, the bio-system results composed of:

- 48 • The cell membrane, which delimits the volume of the cell, controls inflows and outflows of
49 molecules;
- 50 • The cytoplasm which is an aqueous solution of molecules that fills the cell interior;
- 51 • The organelles, suspended in the cytoplasm.

52 The system may contain many substances not initially present in its environment. Many
53 enzymes found in fragments of cytoplasm membranes are often not directly accessible to system
54 environment, while, in the environment, the concentrations of some molecular species decrease in
55 time, and nutrients must flow into the system in order to allow the biochemical reactions to occur
56 and produce macromolecular cell components, with a related increase of the living cell mass and
57 volume [4]. These biochemical reactions require energy. This energy is obtained from the same
58 reactions, which involve the nutrients, with a related waste of heat towards the living cell
59 environment. The net effect of all these biochemical reactions is to reduce the entropy of the system,
60 with an increase of the entropy generation in the environment [5].

61 The analysis of the chemical species and of their reactions in the living cells have led to some
62 sequences which start with nutrient molecules and end with the formation of living cells substance
63 and waste molecules and waste heat. Too numerous reaction sequences are considered to exist.
64 Moreover, many molecules can be part of more than one sequence, providing coupling of sequences.

65 The characteristics of the living systems have been determined by using batch cultures, grown
66 at the optimum temperature for the species employed by the experimenter, even if these
67 experimental method presents disadvantages because the living system is forced to live in a
68 continually changing environment, due to supply and decrease of the environmental concentration
69 of nutrients related to the cells growth, while both the environmental concentration of the waste
70 molecules and the waste heat rise proportionally.

71 Cells exchange energy and matter through their membrane, in order to maintain their living
72 conditions. All the biophysical and biochemical processes require fluxes of energy, ions and
73 molecules which are controlled by the endogenous electric fields and accumulated in the nm-thin
74 layer of water. Cells live in an environment which slowly changes its chemical and physical
75 properties. From a thermodynamic point of view, a living system is defined as an open system, in
76 non-equilibrium thermodynamic states with its environment. The 1931 cancer cells were proved to
77 be fermentative, as consequence of a metabolic injury [6] and genetic effects. In normal cells, mitosis
78 is synchronized with cell growth in order to maintain their size during replication. This biomedical
79 result represents the bioenergetic bases to link the production of lactic acid and
80 extracellular-intratatumoral acidification to cancer growth and metastasis. Moreover, both the pH of
81 the cell cytoplasm and the extracellular environment are controlled by the living cell membrane
82 potential.

83 Differentiated cells are hyperpolarized as compared to quiescent or cycling cells, and the
84 hyperpolarization increases the efflux of some ions (Ca^{2+} , K^+ , Zn^{2+} etc.). The fundamental role of
85 inhibition of proliferation in normal as well as neoplastic cells has recently been pointed out as well
86 as the interactions between the ion channels and the other elements of the signalling network. Direct
87 cell migration is fundamental in tissue formation, but, if proliferation and invasion is out of control
88 then a new behaviour occurs and cancer emerges as a disease of abnormal growth. At a definite
89 phase of its life any cancer cell divides into two cells. The size of the single cell at the beginning of
90 division can vary; consequently, also the relative size of the two daughter cells can vary, but lower
91 and upper bounds on cell volume have been observed. All these processes are driven by fluxes of
92 energy and mass, and the cell shape results fundamental in their analysis. From a thermodynamic
93 point of view, living systems are no more than complex systems, open with a control of fluxes.

94 Considering the chemical reaction at constant pressure and temperature the Gibbs Free Energy
95 could seem the selected function for the study of the steady states of the living systems, but its
96 decrease as a criterion for occurrence of a spontaneous evolution is limited to the complex

97 phenomena which occur at constant temperature and pressure inside the living system. But, a
 98 general objective function for the analysis of the living systems is required. Moreover, considering
 99 that the system wastes energy, and mass, it generates irreversibility, so the general criterion for
 100 study its spontaneous evolution is the entropy generation related to the changes of the system.
 101 Entropy generation always increases in any spontaneous and irreversible evolution.

102 The aim of this paper is to analyze deeply the use of the heat role in the study of the cancer
 103 systems. Indeed, recently we have used the entropy generation to introduce a thermodynamic
 104 approach to the analysis of the cancer system in order to design a possible support to the present
 105 anticancer therapies. But, the thermodynamic approach introduces a new viewpoint for the analysis
 106 of the living systems. This paper wishes to explain and improve the thermodynamic formulation of
 107 this approach, with the second aim to highlight how to support the biomedical sciences in the
 108 comprehension of the possible thermodynamic support to the present anticancer therapies.

109 2. Materials and Methods

110 Life involves organizational and thermodynamic processes, which tend towards the maximum
 111 conversion of available energy [1-8]. The biochemical reactions produce or consume external
 112 metabolites, and connect internal metabolites, at constant concentrations in the cells at their steady
 113 states. In order to do so, cells must exchange energy and matter through their membrane. Indeed,
 114 many processes such as replication, transcription and translation, require fluxes of ions and
 115 molecules which are driven by the endogenous electric fields and accumulate in the nm-thin layer of
 116 water [11,14,15]. These fluxes of ions induce biochemical reactions within cells and tissues.

117 Moreover, the 1931 Nobel laureate Otto Warburg proved that cancer cells are fermentative,
 118 pointing out that this was the consequence of a metabolic injury [16]. In normal cells, mitosis is
 119 synchronized with cell growth for cells to maintain their size during replication. Even if the tumor
 120 behavior is more complex and it is probably based also on genetic structures of the cells, the results
 121 of Warburg highlights the important role of energy conversion in cells. From a thermodynamic point
 122 of view, this result can represent the starting point of the analysis; indeed, the genetic processes have
 123 the consequence of regulation for the cell behavior, but, if we consider the cell as a black box, as
 124 usually done in thermodynamics, then the genetic regulation is no more than the "mind" of the cell,
 125 without any direct consequence on the thermodynamic balances. Indeed, the thermodynamic
 126 approach evaluates the life cycle of the cell by considering only the energy and mass fluxes balances
 127 during the whole cycle of cell life, and not considering the gene activities, but evaluating only their
 128 consequences expressed by the energy conversion in cell.

129 Living cells metabolism implies flows of matter and heat into and out of the cells. What are the
 130 consequences of these fluxes and metabolic reactions? To answer to this question, we consider the
 131 Gibbs free energy variation in time, due to the mass fluxes. It results [12,17-19]:

$$132 \quad \frac{dG}{dt} = \dot{W} + \sum_{in} \dot{n}_{in} \tilde{\mu}_{in} - \sum_{out} \dot{n}_{out} \tilde{\mu}_{out} + T_0 \dot{S}_g \quad (1)$$

133 where G is the Gibbs free energy, \dot{W} is the useful power done by the cell, \dot{n} is the molar flow,
 134 $\tilde{\mu} = \mu + Ze\phi$ is the molar electrochemical potential, with μ chemical potential, Ze charge of the
 135 ion considered, and ϕ cell membrane electric potential, T_0 is the temperature of the cell
 136 environment, t is the time, *in* means inflow, *out* means outflow, and \dot{S}_g is the entropy
 137 generation rate, defined as [12,17-19]:

$$138 \quad \dot{S}_g = \frac{dS}{dt} - \frac{\dot{Q}}{T_0} - \sum_{in} \dot{n}_{in} \bar{s}_{in} + \sum_{out} \dot{n}_{out} \bar{s}_{out} + \sum_f \dot{n}_f \bar{s}_f \quad (2)$$

139 where \dot{Q} is the heat power exchanged, \bar{s} is the molar specific entropy, and f means formed. This
 140 equation highlights as these flows cause entropy variation; moreover, these fluxes imply also a great
 141 number of chemical reactions within the cells, accompanied by entropy generation. Now, we
 142 introduce the equation (2) into the equation (1) obtaining:

$$143 \quad \frac{dG}{dt} = \dot{W} - \dot{Q} + \sum_{in} \dot{n}_{in} (\tilde{\mu}_{in} - T_0 \bar{s}_{in}) - \sum_{out} \dot{n}_{out} (\tilde{\mu}_{out} - T_0 \bar{s}_{out}) + \sum_f \dot{n}_f \bar{s}_f + T_0 \frac{dS}{dt} \quad (3)$$

144 From a thermodynamic point of view, the cell life is no more than a thermodynamic stationary
 145 state, at constant environmental temperature and pressure, and this can be obtained by introducing
 146 the two conditions [12,17-19]:

$$147 \quad \begin{cases} \frac{dG}{dt} = 0 \\ \frac{dS}{dt} = 0 \end{cases} \quad (4)$$

148 which hold to the equation:

$$149 \quad \dot{Q} = \dot{W} + \sum_{in} \dot{n}_{in} (\tilde{\mu}_{in} - T_0 \bar{s}_{in}) - \sum_{out} \dot{n}_{out} (\tilde{\mu}_{out} - T_0 \bar{s}_{out}) + \sum_f \dot{n}_f \bar{s}_f \quad (5)$$

150 which expresses the fundamental role of heat fluxes between the cell and its environment.

151 The equation (5) is very difficult to be numerically evaluated, because it implies the knowledge
 152 of all the balance at the second member of the equation for each cell. So, we must find an alternative
 153 way to evaluate the cell heat flux.

154 The heat flux is no more than the heat power exchanged by the cell and its environment. We can
 155 consider that, inside the experimental setup usually used in the biophysical and biochemical
 156 analysis of cells, the heat flux is exchanged by convection with the suspending aqueous solution
 157 around any cell, so we can write:

$$158 \quad \dot{Q} = \rho_{cell} V c_{cell} \frac{dT_{cell}}{dt} = \alpha A (T_{cell} - T_0) = \alpha \frac{V}{\langle R \rangle} (T_{cell} - T_0) \quad (6)$$

159 where α is the coefficient of convection, $A = V / \langle R \rangle$ is the surface area of the cell, which changes
 160 with the phases of the development of the cell, V is the volume of the cell, $\langle R \rangle$ is the volume/area
 161 ratio, a parameter which influences the chemical reaction time and the fluxes through the cell
 162 membrane, ρ_{cell} is the cell mass density, and c_{cell} is the specific heat of the cell, and $(T_{cell} - T_0)$ is
 163 the difference of temperatures between the cell temperature and the environment temperature. The
 164 term $A = V / \langle R \rangle$ is the geometric shape of the cell in relation to convection. We must introduce
 165 this quantity because at a stage of his life a cell has a definite volume, but it can change its shape in
 166 relation to its duplication phase at that time. Indeed, in eukaryotic cells, the main control process
 167 occurs at the G1/S transition, in late S (DNA synthesis) phase, at mitosis (M) entry and at the
 168 metaphase to anaphase transition. Any process is controlled by the cyclin-dependent kinases,
 169 regulated by the oscillatory expression of G1 and G1/S-cyclins, S-cyclins, and M-cyclins. The
 170 transition between metaphase to anaphase is triggered by the anaphase-promoting
 171 complex/cyclosome (APC/C). Mitogens stimulate the entry into the cell cycle from a quiescent (G0)
 172 phase. Exit from mitosis can lead to differentiation, apoptosis, or return to quiescence [20]. These
 173 mechanisms are altered in neoplastic cells.

174 Now, from equation (6), we can obtain:

$$175 \quad \frac{d \ln(T_{cell} - T_0)}{dt} = \frac{\alpha}{\rho_{cell} c_{cell}} \frac{1}{\langle R \rangle} \quad (7)$$

176 with the result that greater is the volume-area ratio lower is the thermal exchange, being α , ρ_{cell}
 177 and c_{cell} approximately constant. It occurs because the cell adapts its volume/area rate in order to
 178 optimize the cell membrane fluxes to obtain the work it needs, and, conversely, this geometric rate
 179 controls also the heat exchange.

180 3. Results

181 The results obtained highlight the role of the volume-area ratio of the cells in relation to their
 182 heat exchange in *in vitro* experiments. This effect is fundamental when the heat exchange plays a
 183 fundamental role in the analysis of the experimental data.

184 In particular, we wish to highlight that this effect is particularly interesting in the study of the
 185 cancer growth compared with normal cell growth, because cancer presents a different metabolic
 186 cycle. This is important when we study the cancer growth control. Indeed, we can point out that the
 187 equation (9) is also the equation which links a frequency, i.e. the inverse of the time
 188 $\nu = 1/\tau = (\alpha / \rho_{cell} c_{cell} \langle R \rangle)$, to a structural and geometrical properties of the cell and its
 189 environment, in relation to the heat exchange. But, what is this frequency?

190 It is difficult to find an answer without considering the thermodynamic approach. Indeed, each
 191 system presents a proper time of answer to the external thermal perturbation. We suggest that this
 192 frequency is no more than the inverse of the cell proper time of answering to the external thermal
 193 perturbation, or the heat exchange rate. Indeed, the heat flow can be written as:

$$194 \quad \dot{Q} = \frac{Q}{\tau} = Q\nu \quad (10)$$

195 where Q is the heat wasted during the cell life. Moreover, we can consider also that the ions fluxes
 196 are controlled by the cell membrane potential, which is related to the Gibbs free energy by the
 197 relation [17-19]:

$$198 \quad dG = d\phi - 2.3 \frac{RT_0}{F} dpH \quad (11)$$

199 where R is the universal constant of gasses and F is the Faraday constant. At the stationary states,
 200 remembering the relations (4), it follows:

$$201 \quad d\phi = 2.3 \frac{RT_0}{F} dpH \quad (12)$$

202 which links the variation of the cell membrane electric potential to the variation of the pH, related to
 203 the ions fluxes. Conversely, we can try to force a variation of the pH by a variation of the cell
 204 membrane electric potential.

205 How can we try to do so? We can induce a variation in the cell membrane electric potential by
 206 using an electromagnetic wave, with a frequency just equal to the proper frequency of our system.

207 The results are summarized in the Table 1. It is possible to highlight that the electromagnetic
 208 waves induce a different behavior in the cancer cells considered; indeed, they decrease their growth
 209 if compared with the cancer cells outside of the electromagnetic field, proving that there exists a
 210 forcing phenomena of heat flux control which controls the related ions fluxes and, consequently,
 211 induce a different behavior to the biosystem.

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 213 **Table 1. Cell parameters and calculated ELF-EMF frequencies.**

214 For each cell line, thirty cells in different fields were evaluated in their size. Cell volumes were
 215 estimated and used to calculate the forcing frequencies of ELF-EMF, using data in our previous
 216 experimental paper [21].

cell line	cell size [μm^2]	cell volume [μm^3]	Volume-area ratio	mean frequency [Hz]
MCF7	$1,993 \pm 16$	$16,468 \pm 793$	8.26 ± 0.46	5.0 ± 0.7
	$1,051 \pm 13$	$17,303 \pm 1,040$	16.45 ± 1.18	
	$2,604 \pm 21$	$42,284 \pm 2,068$	16.24 ± 0.93	
SKBR3	$1,033 \pm 11$	$1,795 \pm 97$	1.74 ± 0.11	8.0 ± 2.0
	$2,066 \pm 17$	$29,048 \pm 1,301$	14.06 ± 0.74	
	$2,454 \pm 20$	$47,594 \pm 2,168$	20.21 ± 1.05	
	$1,042 \pm 11$	$1,300 \pm 80$	1.25 ± 0.09	

GTL16	1,873 ± 15	2,630 ± 140	1.40 ± 0.09	14.0 ± 3.0
	1,059 ± 12	1,260 ± 77	1.19 ± 0.09	

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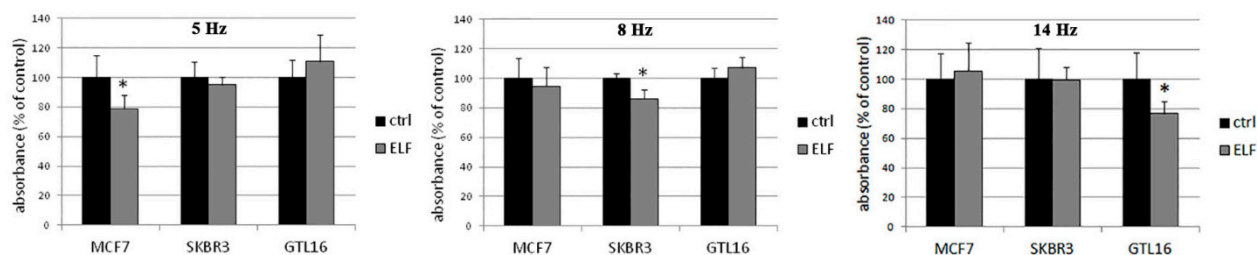
Figure 1. Decreasing of growth at the proper frequency.

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The experimental results confirm the results here obtained, in particular that the cancer cells modify their behavior only if irradiated by an electromagnetic field at the proper frequency for the cell line considered in our previous experimental paper [21].



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4. Discussion

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The results, here obtained, point out the fundamental role of the cell volume-area ratio in relation to the fluxes control.

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Indeed, there is a temperature difference between the interior of a living cell and its environment. This is a thermodynamic necessity for life. Sensible heat is exchanged between inside and outside of the cell due to this temperature difference. This heat flow contribute to entropy generation. Part of the entropy generated appears outside the cell as sensible heat. The fraction of all entropy generation that appears in this form depends on the nature and number of processes occurring in the cell. Consideration of the temperature difference between environment and cell interior allows the introduction of non-equilibrium thermodynamics for the analysis of cells behavior. Brock suggested that the stability of thermophilic organisms can be attributed to membrane structure properties of these organisms [22].

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The temperature gradient contribution to the flow of substances through the cell membranes of the cell with a consequent influence on metabolic processes [23-26]. The approach here suggested allows us to evaluate the homeostatic cellular response to external perturbations. This answer is no more than a thermo-chemical output of the cell in the environment. So, we can suggest that the thermodynamic approach holds to a model of analysis of the action and reaction in terms of membrane flux variation. This approach could represent a new approach to design possible support to the present anticancer therapies, by introducing external fields variation, at the proper answer time, in the therapeutic protocols. From the experimental results it is clear a reduction of the growth of the cancer with the consequence of improving the effects of the present therapies.

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The growth rate of cells, cancer included, at a fixed temperature is a function of both composition of the medium and chemical potentials of the component substances. This represents a sort of control to the growth, because there is a maximum rate at which each bio-chemical reaction can occur under the existing constraints. This rate is conditioned by the volume-area ratio, because it controls the ions fluxes, i.e. the fluxes of the chemical reactants.

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So, our results show a method for the design of therapies and experiments for their analysis. Indeed, the specific effect of the single frequencies has two important consequences:

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- In every cell type different parameters of electromagnetic waves impact differently the energy utilization and proliferation, with different inhibition effects on the cell growth;
- The same electromagnetic wave has distinct effects on different cells, with a selectivity behavior.

254 Recently, the Bandyopadhyay research team has demonstrated that the spontaneous
255 oscillations of neurons microtubules of the frequency of around 1 MHz oscillation of electrical dipole
256 moments of free electrons and conformational switching, cause wave interference which are the
257 origin of the characteristic shape of the electrical oscillations of the brain at the
258 electroencephalographic signal of 4-40 Hz nested gestalts, named beat frequencies [27,28]. The result
259 is no more than the link between the brain synchronization of consciousness with the quantum
260 mechanical behaviours of microtubules. This result proves that quantum vibrations microtubules
261 are entangled across neuronal networks via the gap junction, interconnecting channels which
262 physically link neurons together. This result highlights the fundamental role of microtubules, and
263 their quantum effect, with particular interest for resonance, in cell behaviour. But the proven theory
264 doesn't suggest any link between microscopic and macroscopic behaviours of cells. This result is no
265 more than a thermodynamic resonance for the whole cell. Microtubules are electrical polar
266 structures with power supply from hydrolysis (around 10-14 W cm⁻¹ per unit length of the
267 microtubule) of guanosine triphosphate to guanosine diphosphate: the related energy can excite
268 vibrations. Microtubules lose part of the energy by viscous damping of the surrounding cytosol [29].
269 Microtubules plays a fundamental role both in the organisation activities of the living cells
270 cytoskeleton and in the intracellular transport [30]. As Bandyopadhyay showed the microtubules of
271 the frequency of around 1 MHz, but the global effect of the neurons microtubules synchronisation is
272 the brain 4-40 Hz signal, lower than five orders of magnitude. In particular, Poznanski et al. pointed
273 out how intracellular capacitive effects of bound electrical charges within mitochondrial membranes
274 can influence electrotonic signals expressed as solitary waves [21] and that the outer mitochondrial
275 membrane acts as an amplifier of the ingoing waves. Moreover, they showed the fundamental role
276 of the changes of the mitochondrial membrane equilibrium potential in sustaining solitons with
277 self-regulation in their amplitude [21]. In physics and chemistry, resonance is the phenomenon in
278 which a vibrating system or external force drives another system to oscillate with greater amplitude
279 at specific frequencies. At resonant frequencies, small periodic driving forces have the ability to
280 produce large amplitude oscillations, due to the storage of vibration energy. So, the effect shown by
281 Bandyopadhyay is no more that the amplification of a resonance electromagnetic interaction
282 between external electromagnetic waves and cells microtubules, which generates also the
283 microtubules synchronisation. This is a biological resonant effect!

284 But, microtubules are in all cells and they play the same role in all human cells. So, what
285 Bandyopadhyay has pointed out, must occur in all cells, with different global effects in relation to
286 their specialized functions. In particular, the mitochondrial respiratory chain and oxidative
287 phosphorylation cause a dispersion of energy as heat, related to the energy from nutrients converted
288 in proton-motive force driving ATP synthesis, necessary to transport molecules against gradient
289 [31]. In this context, it is possible to introduce a thermodynamic approach which realize this
290 biophysical model is to consider the cells as adaptive thermal engines, able to convert energy from
291 one form to another by coupling metabolic and chemical reactions with transport processes [24-26];
292 cells irreversibly consume free energy to maintain thermal and chemical processes and to sustain the
293 transport of matter, energy and ions [24-26]. Human cells must exchange their wasted heat with a
294 constant temperature environment (the human body around them), so if a difference in the
295 metabolism and in the efficiency of the cell system occurs, as in cancer, the cell encounters a difficult
296 in maintain its optimal inside temperature for life. But, in any heat exchange there is a characteristic
297 temperature function of physical (density, specific heat, convective coefficient) and geometrical (cell
298 volume and membrane area) quantities. This time represents a specific time for any cells lines. But,
299 this time is also the resonant time, and its inverse is no more than the biological tuning fork
300 frequency. So, by inflowing an electromagnetic wave with a frequency evaluated by the biological
301 tuning fork frequency for any cell line, we can obtain a forced behaviour of the considered cell line, a
302 modification in the inside organisational process due to the synchronisation of the microtubules
303 involved in the resonant interaction, just as in the Bandyopadhyay effect. So, external short
304 frequency can produce high microtubules resonant effect, with the control of the cell behaviour.
305 Indeed, the fluxes across the cell membrane represent a fundamental quantity for cell behaviour

306 control [24-26]: the cells reach their optimal asset by a selective process of interactions with their
307 environment, with the consequent effect of the redistribution of energy and mass flows in their
308 metabolic network, enabled by regulatory proteins [24-26]. Indeed, the cell mitotic cycle is composed
309 of a sequence of processes such as DNA replication, chromosome condensation and segregation,
310 duplication and migration of the spindle pole, breakdown of the nuclear envelope, and cytokinesis
311 [20]. Moreover, the cell cycle has been highlighted to be controlled by a control system which
312 monitors DNA integrity before any transition to the next phase. In eukaryotic cells, the main control
313 process occurs at the G1/S transition, in late S (DNA synthesis) phase, at mitosis (M) entry and at the
314 metaphase to anaphase transition. Any process is controlled by the cyclin-dependent kinases,
315 regulated by the oscillatory expression of G1 and G1/S-cyclins, S-cyclins, and M-cyclins. The
316 transition between metaphase to anaphase is triggered by the anaphase-promoting
317 complex/cyclosome (APC/C). Mitogens stimulate the entry into the cell cycle from a quiescent (G0)
318 phase. Exit from mitosis can lead to differentiation, apoptosis, or return to quiescence [20]. These
319 mechanisms are altered in neoplastic cells. Cell metabolism is constrained by the maximum amount
320 of macromolecules that can be contained in the intracellular volume: any biochemical process
321 requires energy, and any energy conversion generates outflows of energy, due to the second law of
322 thermodynamics. So, the fundamental thermodynamic approach to the behaviour of the cell systems
323 is to consider inflows and outflows of energy and masses (ions included). Indeed, it was
324 experimentally pointed out that:

- 325 • Any increase in K⁺-channel expression and activity of K⁺ at the G1/S boundary is often
326 necessary for cell cycles;
- 327 • Ca²⁺fluxes, which can bind with tubulin, control the membrane potential, regulate the mitotic
328 spindle and cytokinesis, regulate DNA transcription, modulate the expression and activity of
329 the transcription factors that control expression of the G1 cyclins, producing direct effects on
330 cyclins, cyclin kinases, and the associated proteins;
- 331 • Ions-fluxes vary the membrane potential which determines pH variation inside and outside the
332 cell, with a consequent variation of the metabolic cycle [24-26].

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334 5. Conclusions

335 In this paper, we have developed the analysis of a thermodynamic approach to cancer cells,
336 with particular regards to the role of the volume-area ratio in the heat exchange and the
337 consequences to the cancer cells behavior.

338 We have pointed out the existence of a proper time of answer of any cell line to the heat
339 exchange. This time results related to the cells volume-area ratio, a geometrical parameter
340 fundamental for the considerations on the fluxes and cells membrane electric potential variation.

341 Then, starting from some previous experimental results [23-26], we have obtained also an
342 experimental proof of the present results.

343 The results highlight how the irreversibility plays a fundamental role also in biophysical
344 systems; indeed, the geometrical rate is completely related to the entropy generation as it is clear by
345 introducing the relation (6) into the relation (2). This holds to a new approach to biological physics,
346 based on the first and second law of thermodynamics.

347 The results here obtained confirm the biochemical results obtained by other scientists who used
348 an entropic approach; indeed, Luo has demonstrated that a low-frequency and low-intensity
349 electromagnetic field or ultrasound irradiation may increase the entropy production rate of a cell in
350 normal tissue than that in cancer; consequently, it reverse the direction of entropy current between
351 two kinds of cells. The modification of PH value of cells may also cause the reversal of the direction
352 of entropy flow between healthy and cancerous cells [31,32].

353

354 **Author Contributions:** UL developed the theoretical model, and the thermodynamic considerations. GG
355 developed the experiments and the data analysis.

356 **Conflicts of Interest:** The authors declare no conflict of interest.

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