

1 Article

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# Design of the oligonucleotide carriers: importance of 3 polyamine chain length

4 **Vadim V. Annenkov<sup>1,\*</sup>, Uma Maheswari Krishnan<sup>2</sup>, Viktor A. Pal'shin<sup>1</sup>, Stanislav N. Zelinskiy<sup>1</sup>,  
5 Gayathri Kandasamy<sup>2</sup> and Elena N. Danilovtseva<sup>1</sup>**6 <sup>1</sup> Limnological Institute of the Siberian Branch of the Russian Academy of Sciences, 3, Ulan-Batorskaya St.,  
7 P.O. Box 278, Irkutsk 664033, Russia; danilovtseva@yahoo.com (E.N.D.); acrom@mail.ru (V.A.P.)8 <sup>2</sup> Centre for Nanotechnology & Advanced Biomaterials (CeNTAB), School of Chemical and Biotechnology,  
9 SASTRA University, Thanjavur 613401, Tamil Nadu, India; umakrishnan@sastr.edu

10 \* Correspondence: annenkov@lin.irk.ru; Tel.: +7-914-8982577

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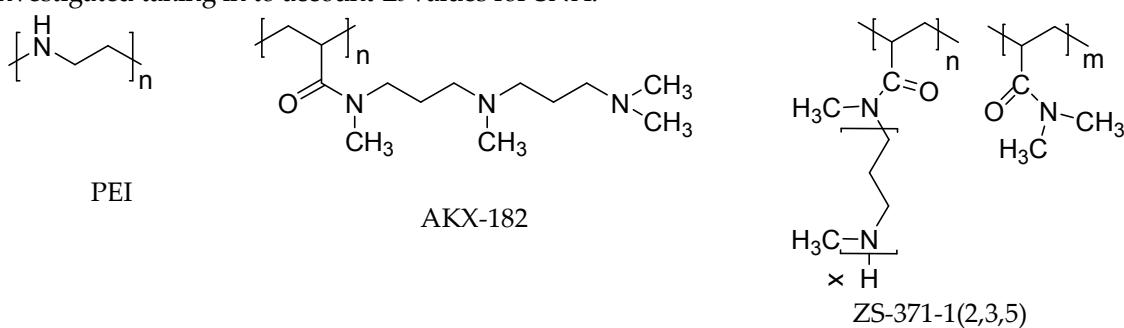
12 **Abstract:** Amine containing polymers are extensively studied as special carriers for short-chain RNA  
13 (13–25 nucleotides) which are applied as gene silencing agent in gene therapy of various diseases  
14 including cancer. Elaboration of the oligonucleotide carriers requires knowledge about peculiarities  
15 of oligonucleotide - polymeric amine interaction. Critical length of the interacting chains is the  
16 important parameter which allows to design sophisticated constructions containing oligonucleotide  
17 binding segments, solubilizing, protective and aiming parts. We studied interaction of (TCAG)<sub>n</sub>,  
18 n=1-6 DNA oligonucleotides with polyethylenimine and poly(N-(3-((3-(dimethyl-  
19 amino)propyl)(methyl)amino)propyl)-N-methylacrylamide). Critical length for oligonucleotides in  
20 interaction with polymeric amines is 8-12 units and complexation at these length can be  
21 accompanied by "all-or-nothing" effects. New dimethylacrylamide based polymers with grafted  
22 polyamine chains were obtained and studied in complexation with DNA and RNA oligonucleotides.  
23 The most effective interaction and transfection activity into A549 cancer cells was found for a sample  
24 with average number of nitrogens in polyamine chain equal to 27, i.e. for a sample in which all  
25 grafted chains are longer the critical length for polymeric amine - oligonucleotide complexation.26 **Keywords:** polymeric amines; oligonucleotides; critical length; grafted polyamines; gene delivery  
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## 1. Introduction

29 Gene silencing is a promising approach to combat various diseases including cancer [1-4]. The  
30 corresponding techniques include delivery of short nucleic acids (SNA, 13–25 nucleotides) into the  
31 cell cytoplasm. Special carriers are employed to protect SNA in circulation, to facilitate SNA  
32 penetration into cell and to escape destruction of SNA in lysosomes [5]. Amine containing polymers  
33 are extensively studied in this area and the most well-known polymer in this category is  
34 polyethylenimine (PEI) [6]. High buffer capacity of PEI at pH 6-7.5 causes escape of the  
35 polymer-SNA complex from endosomes by the "proton pump" mechanism. The other requirement  
36 for SNA carriers is low cytotoxicity which is a limitation of polymeric amines. Dedicated design of  
37 the oligonucleotide carriers requires knowledge about peculiarities of SNA - polymeric amine  
38 interaction. This interaction is a typical interpolymeric reaction [7-8] and one of the important  
39 parameters of these reactions is the critical length ( $L_c$ ) of the interacting chains. Sub-optimal length  
40 results in weak interactions while higher  $L_c$  results in relatively irreversible reactions. The  
41 importance of  $L_c$  value in SNA - polymer interactions was addressed in several articles [9-12].  
42 Synthetic polymeric amines of various length and SNA of a fixed length have been explored in  
43 several earlier reports [9,10] but the set of amine chains investigated was restricted. A recent work  
44 had investigated a set of SNA with varying chain length [11] and the minimal length of the SNA  
45 chain was determined as 14 nucleotides in reaction with virus particle of unknown chemical  
46 structure. A study of polylysine interaction with plasmid DNA [12] revealed that complexation

47 occurred at 8 lysine units and an absence of interaction was observed with samples with 3 lysine  
 48 units. The lack of information about  $L_c$  in SNA - polymeric amine interactions stems from the fact  
 49 that the main objective of studies in the field of gene therapy has focused on achieving complexation  
 50 using SNA with definite length, usually near 20 nucleotides. However, polymeric constructs for  
 51 SNA delivery often contain combinations of short amine, neutral and sometimes acidic sequences  
 52 and hence information about  $L_c$  values is required for the design of these constructs and  
 53 understanding their properties.

54 Our work has two objectives with the first being estimation of  $L_c$  for DNA oligonucleotides in  
 55 the reaction with PEI and new polymeric amine poly(N-((3-  
 56 (dimethylamino)propyl)(methyl)amino)propyl)Nmethylacrylamide) (AKX-182, Scheme 1) [13]. We  
 57 have used the DNA oligonucleotides  $(TCAG)_n$ , where  $n = 1-6$ , which contain equal amount of the  
 58 each nucleotide and cannot form stable hairpins or self-dimers. The second objective of our work is  
 59 to synthesize and study new polymers as gene delivery agents. These polymers originate from  
 60 bioinspired polymers with grafted polyamine chains [14]. Polyamines containing 3-4 nitrogen  
 61 atoms, including spermine and spermidine, play an important role in cell physiology: protein and  
 62 nucleic acid synthesis, gene expression, protection from oxidative damage and etc. [15]. The  
 63 so-called long-chain polyamines (LCPAs), more exactly oligomeric polyamines, have been found in  
 64 bacteria [16], diatom algae [17], in the siliceous sponge [18] and in haptophyte [19]. In the case of  
 65 diatoms, LCPAs are present as post-translationally grafted side chains in specific proteins namely,  
 66 silaffins [20]. These LCPAs contain up to two tens of partially methylated nitrogen atoms  
 67 separated by trimethylene or tetramethylene fragments [21]. Some polymeric amines [22,23]  
 68 capable of interacting with oligomers of silicic acid and forming stable composite nanoparticles were  
 69 considered as models of cytoplasmic silicon containing vesicles. We have elaborated [24] a method  
 70 to synthesize LCPAs as an oligomeric mixture containing polyamines with 6-30 nitrogen atoms.  
 71 Grafting of the LCPA chains on to poly(acrylic acid) resulted in polyampholytes which can control  
 72 condensation of silicic acid giving rise to silica, the structure of which was similar to biogenic silica  
 73 [25]. Neutral polymers with pendant LCPAs can interact with SNA and the obtained complexes are  
 74 promising particles for gene therapy [14]. Medical applications of polymers require homogeneous  
 75 macromolecules and in this work we synthesized poly(N,N-dimethylacrylamide) (PDMAAm)  
 76 containing grafted LCPA chains starting from narrow polymer and LCPA fractions (Scheme 1).  
 77 Influence of the length of pendant LCPA chains on the capability to interact with SNA was  
 78 investigated taking in to account  $L_c$  values for SNA.



79 **Scheme 1.** Structures of the studied polymers.

80 **2. Materials and Methods**

81 **2.1 Materials**

82 2,2'-Azobis(2-methylpropionitrile) (AIBN) (Sigma-Aldrich, St. Louis, MO, USA) was  
 83 recrystallized from ethanol prior to use. Diethyl ether (Acros, Geel, Belgium) and 1,4-dioxane  
 84 (Sigma-Aldrich) were distilled under sodium. Dimethylformamide (DMF) (Sigma-Aldrich) was  
 85 dried with CuSO<sub>4</sub> (30 min) and distilled at 5 mm Hg. Acryloyl chloride (Sigma-Aldrich) was  
 86 distilled before polymerization. Dichloromethane (Acros, Geel, Belgium) was refluxed over  
 87 phosphorus pentoxide and distilled under argon. Dimethylamine 40 wt. % aqueous solution,

88 triethylamine, 1,3-dibromopropane, N-hydroxysuccinimide, N-hydroxyphthalimide, potassium  
89 hydroxide, potassium carbonate of reagent grade (Sigma-Aldrich, Fisher, or Acros Chemicals) were  
90 used in the study. Dimethyl sulfoxide-d6 (DMSO-d6, 99.8 atom D %), deuteriochloroform (99.8 atom  
91 D %), heptafluorobutyric acid (HFBA,  $\geq$  99.0 % (GC)), acetonitrile (HPLC Far UV / gradient grade,  
92 Avantor Performance Materials B.V., Deventer, The Netherlands) and trifluoroacetic acid (TFA, 99  
93 wt. % purity) were purchased from Sigma-Aldrich (St. Louis, MO, USA), Fisher (Hampton, NH,  
94 USA), or Acros (Geel, Belgium) chemicals and used without further treatment. A 9.01 wt. % solution  
95 of dimethylamine in 1,4-dioxane was prepared via saturation of dry 1,4-dioxane with gaseous  
96 dimethylamine. The amine concentration was determined from the resulted weight gain as well as  
97 with potentiometric titration. To obtain gaseous dimethylamine, its 40 wt. % aqueous solution was  
98 added dropwise to a large excess of potassium hydroxide flakes. The evolved gas was passed  
99 through a drying column packed with KOH flakes.

100 N,N-bis[3-(methylamino)propyl]methylamine was prepared following the technique from our  
101 earlier work [26]. Oligo(N-methylazetidine) (LCPA, ZS-309 sample) was prepared from  
102 1,3-dibromopropane and N,N-bis[3-(methylamino)propyl]methylamine according to earlier work  
103 from our group [24]. AKX-182 polymer was prepared according to the protocol described earlier  
104 [13]. PEI ( $M_w$  = 30000-40000) was obtained from SERVA Fine Biochemica (Heidelberg).

105 FAM 3'-tagged DNA oligonucleotides were purchased from Evrogen JSC, Russia. Small  
106 interfering RNA (si-RNA) against vascular endothelial growth factor (VEGF) was obtained from  
107 Eurofins Genomics, USA. The sequence of the sense and anti-sense strands of the  
108 fluorophore-tagged si-RNA is as follows: sense - Cy3-GGAGUACCCUGAUGAGAUC and  
109 antisense: CCUCAUGGGACUACUCUAG-Cy3..

## 110 2.2. *Instrumentation*

111  $^1\text{H}$  nuclear magnetic resonance (NMR) spectra were obtained on a DPX 400 Bruker instrument  
112 (400.13 MHz, Billerica, MA, USA) in  $\text{CDCl}_3$  and DMSO-d<sub>6</sub>. Spectra of LCPA were recorded for  $\text{CDCl}_3$   
113 solutions, whereas those of poly(N,N-dimethylacrylamide)s grafted with LCPA were obtained for  
114 DMSO-d<sub>6</sub> solutions of samples derivatized with trifluoroacetic acid as follows: to a sample of  
115 polymer, typically of ca. 30 mg, in a glass screw cap flat-bottom vial, 500  $\mu\text{L}$  of TFA was added. The  
116 vessel was heated at about 50 °C and occasionally shaken for about 30 min to ensure complete  
117 dissolution. Then, the excessive TFA was removed to dryness at 50 °C with argon flow. The cooled  
118 residue was mixed with 600  $\mu\text{L}$  of DMSO-d<sub>6</sub> and left overnight at room temperature. The resultant  
119 solution was filtered through a cotton pad in a 1 mL polypropylene pipette tip directly to a NMR  
120 ampoule.

121 Mass spectrometric analysis was performed on an Agilent 6210 TOF LC/MS System. The  
122 samples were dissolved in acetonitrile. Water and acetonitrile with 0.1% (v/v) HFBA were used as  
123 eluting solvents A and B, respectively (Solvent A – 90%, B – 10%). The flow rate of the mobile phase  
124 was set at 0.2 mL/min, while the injection volume of sample solution was 20  $\mu\text{L}$ . The conditions for  
125 TOF MS were as follows: the mass range was m/z 100-1000 and scan time was 1 s with an interscan  
126 delay of 0.1 s; mass spectra were recorded under ESI+, V mode, centroid, normal dynamic range,  
127 capillary voltage 3500 V, desolvation temp 350°C, nitrogen flow 5 L/min.

128 The molecular masses of the new polymers were estimated via size-exclusion chromatography  
129 (SEC) using a Milichrom A02 chromatograph (JSC Econova, Novosibirsk, Russia) with 2 mm  $\times$  75  
130 mm column filled with SRT SEC-100 5  $\mu\text{m}$  phase (Sepax Technologies, Inc., Newark, NJ, USA),  
131 operated at 35 °C using phosphate buffer solution 0.15 M, pH 6.86. The flow rate of the mobile phase  
132 was set at 0.03 mL·min<sup>-1</sup> (pressure 100 psi), whereas the injection volume for 1 g·L<sup>-1</sup> of the sample  
133 solution was 1  $\mu\text{L}$ . Fractionated samples of poly(vinyl formamide) [27] were applied as standards  
134 ( $M_w/M_n$  < 1.3).

135 Atomic force microscopy (AFM) was performed using Scanning Probe Microscope CMM-2000  
136 (PROTON-MIET, ZAVOD, JSC, Russia) operated in contact mode in air at room temperature using  
137 silicon probes (nominal probe curvature radius of 10 nm). Height mode images (512x512 pixels)  
138 were collected with a scan speed between 1 and 2 Hz. The samples were placed on mica slips, water

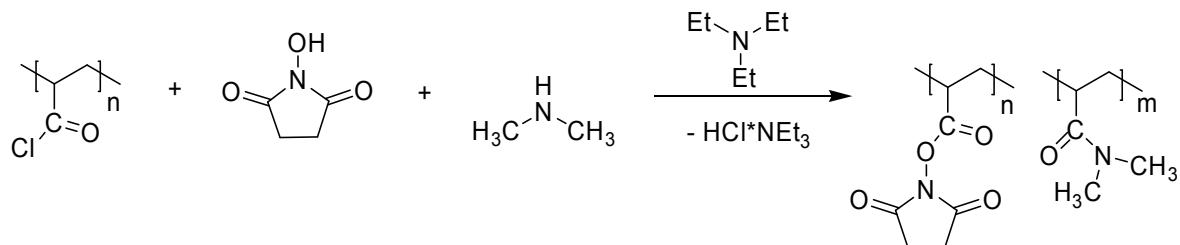
139 was removed with a filter paper after 30 min and slips were air dried. The software package  
 140 Gwyddion was used for AFM image processing.

141 2.3. *Synthesis of poly(N,N-dimethylacrylamide) with grafted LCPA chains (ZS-371-n)*

142 2.3.1. *Synthesis of Poly(acryloyl chloride) (PAC)*

143 PAC was synthesized similar to the protocol described earlier [28] by polymerization of  
 144 acryloyl chloride (6.336 g, 0.070 mol) in 25 mL of dioxane with the addition of 0.1267 g AIBN in  
 145 argon atmosphere at 60 °C for 48 h. PAC was precipitated from the reaction solution using 138 mL of  
 146 cyclohexane. The supernatant was decanted and the residue was dissolved in DMF (Solution A, see  
 147 below) for further use. With the objective to estimate yield and polymerization degree of the PAC,  
 148 the reaction mixture was poured into water (50 mL) and dialyzed against water. After freeze drying,  
 149 poly(acrylic acid) was obtained with 90% yield. According to viscometry data, [29] the  
 150 polymerization degree of the poly(acrylic acid) and, correspondingly of PAC, was found to be 220.

151 2.3.2. *Preparation of poly(N,N-dimethylacrylamide-co-N-acryloxysuccinimide) (ZS-358)*



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 153 Three initial solutions namely Solution A (The prepared PAC in 104.4 g of DMF), Solution B  
 154 (N-hydroxysuccinimide (1.61 g, 14.0 mmol) and triethylamine (1.70 g, 16.8 mmol) dissolved in 11 g  
 155 of DMF), and Solution C (Triethylamine (7.22 g, 71.4 mmole), 10.1 wt. % dimethylamine solution in  
 156 1,4-dioxane (26.55 g, corresponds to 59.48 mmol of HN(CH3)2) and 22.9 g of DMF) were combined.

157 Solution B was added to magnetically stirred solution A (cooling on an ice bath) for eight  
 158 minutes. The mixture was kept stirred with cooling for 31 minutes followed by dropwise addition of  
 159 solution C for 30 minutes. Stirring was continued with cooling for 30 minutes and then at room  
 160 temperature for two hours. The reaction vessel was left in a refrigerator for about 15 hours. The  
 161 white precipitate was filtered off using a filter funnel with sintered glass disc. The light yellowish  
 162 filtrate was concentrated to a volume of 10-15 mL under vacuum at room temperature. Addition of  
 163 110 mL of toluene precipitated a yellow sticky mass which was washed with toluene (50 mL × 1),  
 164 THF (25 mL × 2) and dried using an oil vacuum pump for four hours to yield 7.496 g of ZS-358.

165 2.3.2. *Fractional precipitation of ZS-358*

166 A solution of ZS-358 (6.434 g) in 26 mL of CH<sub>2</sub>Cl<sub>2</sub> was filtered through a glass disc (10-15 µm)  
 167 and diluted to a total volume of 215 mL with CH<sub>2</sub>Cl<sub>2</sub>. The first fraction (1F) was precipitated with  
 168 careful addition of 19.64 g of n-hexane to the stirred solution. The supernatant was decanted in about  
 169 90 minutes for the successive precipitations (Table 1).

170 2.3.3. *Preparative exclusion chromatography (SEC) fractionation of ZS-309*

171 The fractionation was performed on a glass jacketed chromatography column (0.7 × 85 cm)  
 172 packed with Sephadex G-25 (coarse, 100-300 µm). A solution of 100 mg of ZS-309 in 200 µL of 0.1 M  
 173 HCl was loaded on the wet sorbent surface followed by gravity elution at 35°C with a 25 mM acetate  
 174 buffer solution containing 0.2 M NaCl. Collected fractions were analyzed by silica gel TLC  
 175 (CH<sub>2</sub>Cl<sub>2</sub>:CH<sub>3</sub>OH: 25% aq. NH<sub>3</sub> = 2:2:1). The spots were visualized in red color by Dragendorff's  
 176 reagent [30]. Chosen fractions were combined and analyzed with LC-MS (Figure S1) with the  
 177 objective to estimate effectiveness of the fractionation. Then the solutions were evaporated to  
 178 dryness under vacuum, mixed with a solution of K<sub>2</sub>CO<sub>3</sub> (50%) in water and extracted with CH<sub>2</sub>Cl<sub>2</sub>.

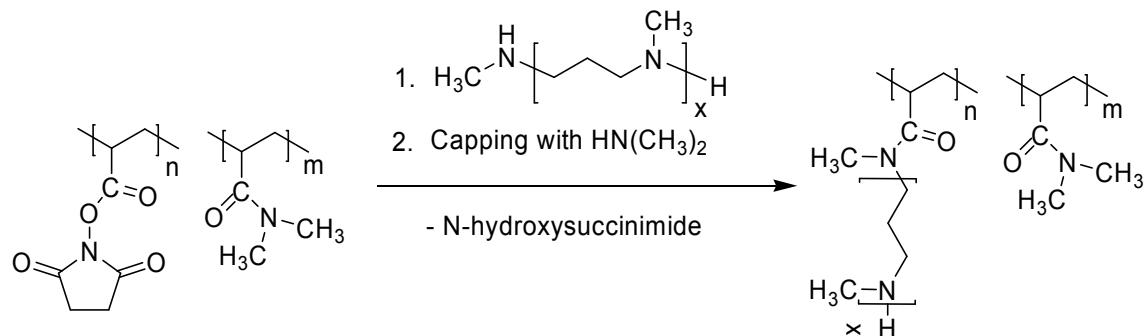
179 The extracts were dried over anhydrous potassium carbonate, evaporated and finally kept under  
 180 vacuum of an oil pump for two days. The obtained LCPA fractions contain 27.6 (ZS-309-1), 16.3  
 181 (ZS-309-2), 13.4 (ZS-309-3) nitrogen atoms in molecule as determined with  $^1\text{H}$  NMR (Figure S2) with  
 182 the use signals at 2.40 ppm (terminal methyl groups) and 2.16 ppm (methyl groups in the polyamine  
 183 chain).

184 **Table 1.** Fractionation of ZS-358.

Fraction #	Portion of n-hexane, g	Precipitation time	Weight of dried fraction, g
1F	19.64	1.5 hour	-*
2F	24.38	0.5 hour	0.6449
3F	23.79	over night	1.2384
4F	15.36	over night	0.7422
5F	18.93	over night	0.7243
6F	21.52	over night	0.8777
7F	34.39	over night	0.5444

185 \*1F was unsuitable for the further usage because of poor solubility in organic solvents. Fractions 3-7 were  
 186 precipitated from solutions kept in a refrigerator.

187 **2.3.4. Grafting of oligo(N-methylazetidine) onto poly(N,N-dimethylacrylamide-co-N-**  
 188 **acryloxysuccinimide) (ZS-371-n)**



189 **Fractions of ZS-358**

190 Preparation of ZS-371-1. ZS-309-3 (0.06602 g), a fraction of oligo(N-methylazetidine), was  
 191 dissolved in a mixture of 0.70 g of DMF and 1.02 g of 10.8%  $\text{HN}(\text{CH}_3)_2$  in 1,4-dioxane (1-st portion)  
 192 followed by addition of 0.24958 g of 3F (a ZS-358 fraction) solution in 1.83 g of DMF. The solution  
 193 was purged with argon, sealed in a glass vessel and kept at 60 °C for 24 hours. After that the 2-nd  
 194 portion of the dimethylamine solution (0.38 g) was added and heating continued for additional 24  
 195 hours. Then the mixture was rotary evaporated and further kept under vacuum of an oil pump at  
 196 room temperature for 3 hours. The sticky residue was thoroughly triturated with diethyl ether  
 197 multiple times, the ether solutions discarded and the product reprecipitated three times from  
 198 methylene chloride/methanol to ether. After vacuum drying, the polymer was dissolved in  
 199 deionized water, filtered through a 0.45  $\mu\text{m}$  cellulose acetate membrane and freeze-dried to give  
 200 0.222 g of ZS-371-1.

201 The other polymers ZS-371-(2-7) were prepared similarly, see the Table 2.

202 The content of grafted units was calculated from  $^1\text{H}$  NMR spectra (Figure S3) of  
 203 TFA-derivatized samples dissolved in  $\text{DMSO-d}_6$  [31]. The integral intensities of the  $^1\text{H}$  NMR signal  
 204 at 10.4 ppm (protons at tertiary nitrogen atoms) was compared with the integral intensities of the all  
 205 C-H protons within 0.8-4.6 ppm except for those of  $\text{DMSO}$ , keeping in mind the average length of  
 206 the grafted oligoamines.

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**Table 2.** Grafting of oligo(N-methylazetidine) onto fractions of ZS-309.

Item	ZS-371-1	ZS-371-2	ZS-371-3	ZS-371-5
Fraction of ZS-358	3F	3F	3F	4F
Amount, g	0.24958	0.51380	0.19950	0.31191
DMF for dissolving ZS-358, g	1.83	3.37	3.37	2.95
Oligo(N-methylazetidine) fraction	ZS-309-3	ZS-309-2	ZS-309	ZS-309-1
Length of the Oligo(N-methylazetidine)	13.4	16.3	15.5	27.6
Amount, g	0.06602	0.25028	0.07295	0.26000
1-st portion of 10.8% HN(CH <sub>3</sub> ) <sub>2</sub> in 1,4-dioxane, g	1.02	2.10	0.81	1.27
2-nd portion of 10.8% HN(CH <sub>3</sub> ) <sub>2</sub> in 1,4-dioxane, g	0.38	0.45	0.49	0.43
Yield of copolymer, g	0.22189	0.42537	0.16356	0.32097
M <sub>n</sub> , kDa	11.0	11.4	11.9	8.6
M <sub>w</sub> /M <sub>n</sub>	1.24	1.22	1.36	1.16
Grafting degree, %	1.8	3.0	2.3	3.4

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#### 2.4. Study of polymer - oligonucleotide interactions and in vitro activity of the polyplexes

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##### 2.4.1. Interaction with DNA oligonucleotides

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The interaction between DNA oligonucleotides and polymers was investigated by electrophoresis on agarose gel. Complexes were prepared by mixing solutions of the polymer and oligonucleotide. The samples were incubated at room temperature for 30 min and placed into the wells of the 1% agarose gel. Free oligonucleotide, as a control, was also loaded onto the gel. The gel running buffer was 40 mM Tris acetate (adjusted to pH 7.4) and 1 mM EDTA. A glycerol gel loading buffer was used (0.5% sodium dodecyl sulfate, 0.1 M EDTA (pH = 8), and 50% glycerol for 10× reagent). The gel was run at 90 V and the fluorescein-tagged oligonucleotide was visualized on a UV transilluminator.

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##### 2.4.2. RiboGreen assay

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Polyplexes were prepared in the ratio of 4:1 and diluted to 100 µL with deionized water in a 96-well plate. 100 µL of 1:1000 diluted ribogreen dye (Invitrogen, USA) was added to each well and the fluorescence intensity was measured using a multimode reader (Synergy H1, Biotek, USA) following incubation for 5 min in dark. The sample was excited at 490 nm and the emission was measured at 525 nm.

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##### 2.4.3. Study of the polymer and polyplex toxicity

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Toxicity of the polymer and polyplex was evaluated using (3-(4,5-dimethylthiazol-2-yl)-5-(3-carboxymethoxyphenyl)-2-(4-sulfophenyl)-2H-tetrazolium) MTS assay (Cell Titer 96 Aqueous one solution, Promega, USA). Four thousand A549 cells per well were seeded in a 96-well plate and incubated at 37 °C in 5 % CO<sub>2</sub>. After the cells achieved confluence, the medium was removed and washed with PBS (pH 7.4) to remove the non-adherent cells. Polyplexes were prepared by mixing 10 µM siRNA (Eurofins) with 4 mg/mL polymer solution in 1:4 ratio. Then, 5 µL of the polyplex or 2 mg/mL polymer solution were mixed with 100 µL of serum-free media and added to the cells. After 4 h, the medium was replaced with the fresh medium and incubated for 24 h or 48 h. MTS reagent (Promega, USA, 10 µL) and 100 µL of serum-free media were added to each sample well and incubated at 37 °C for 2 h. The reaction was stopped by addition of 25 µL of 10% sodium dodecyl sulfate (SDS) solution. The absorbance was measured at 490 nm using multimode reader (Synergy H1, Biotek, USA).

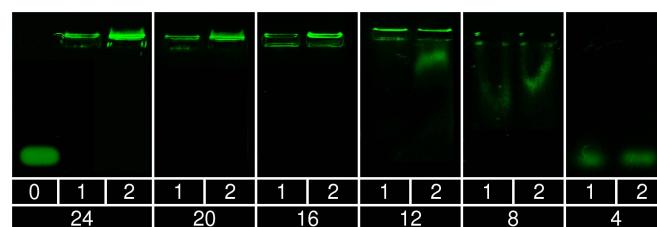
## 238 2.4.4. Study of the polyplex internalization

239 These studies were carried out with a polymer concentration of 4 mg/mL and a siRNA  
 240 concentration of 10  $\mu$ M with a polymer to siRNA ratio of 4:1. The internalisation of the polyplex in  
 241 A549 cells was evaluated using fluorescent siRNA (Eurofins) that is excited at 538 nm and emits at  
 242 640 nm. A549 cells were cultured in DMEM (Gibco, USA) on a cover slip in a 6-well plate with a  
 243 seeding density  $10^5$  cells/well. After the cells attained confluence, the medium was removed and  
 244 washed with phosphate buffered saline (PBS, pH 7.4) to remove the non-adherent cells. Then, 5  $\mu$ L  
 245 of polyplexes were added to 200  $\mu$ L of serum-free media and incubated for specific time points. The  
 246 medium was then replaced with fresh medium containing fetal bovine serum (Gibco, USA). The  
 247 cells were stained with Hoechst 33342 (Invitrogen, USA) and the images were captured using laser  
 248 scanning confocal microscopy (FV1000, Olympus, Japan).

249 **3. Results and Discussion**250 *3.1. Determination of critical length of DNA sequence in reaction with polymeric amines*

251 The influence of length of single-strand DNA oligonucleotide on the ability to interact with PEI  
 252 and AKX-182 was studied at N/P ratio (ratio of amine groups to nucleotides) equal to 20. The high  
 253 excess of polymeric amine minimizes stacking and other possible side effects. Gel electrophoresis  
 254 data (Figure 1) shows formation of slightly positive complex with 12-24-mer DNA for PEI and 16-24  
 255 mer DNA in the case of AKX-182. Decrease the DNA length to 8-12 units results in a diffuse spot  
 256 which corresponds to some negative-charged complexes. 4-mer DNA does not interact with PEI and  
 257 AKX-182. The formation of negative-charged products is unexpected, poor interaction of short  
 258 oligonucleotides with polymeric amines could result in the presence of free DNA but the observed  
 259 diffuse spots are far from free DNA position. We see here two kinds of DNA-containing particles:  
 260 neutral (or slightly positive) and negative charged. These complexes were observed at various N/P  
 261 ratios and DNA length (Table 3). Increase of the DNA length and N/P ratio results in disappearance  
 262 of negative charged complex. Nucleic acids can interact with amines by the means of ionic  
 263 interactions through phosphate groups or by hydrogen bonds through nucleobases [32,33 and  
 264 references in this review]. The latter mechanism suggests that the negatively charged free phosphate  
 265 groups compensates the positive charge of the free amine units. We can hypothesize formation of  
 266 negatively charged coordinated regions surrounded with positive charged parts of the polymer  
 267 (Scheme 2(a)). Increase of the polymeric amine content results in presence of positive charged  
 268 complex only. The formation of just only negative particles with short DNA chains is possibly rises  
 269 according to "all-or-nothing" scheme (Scheme 2(b)) which is often realized with weak associative  
 270 interactions [34,35]. Experiments with double stranded DNA (Figure 2) show similar behavior of the  
 271 DNA-polymer complexes.

272 Thus, 8-12 nucleotides represent a critical length of DNA chains at which interaction with  
 273 polymeric amines proceeds at high excess of the amine and is accompanied by "all-or-nothing"  
 274 effects.



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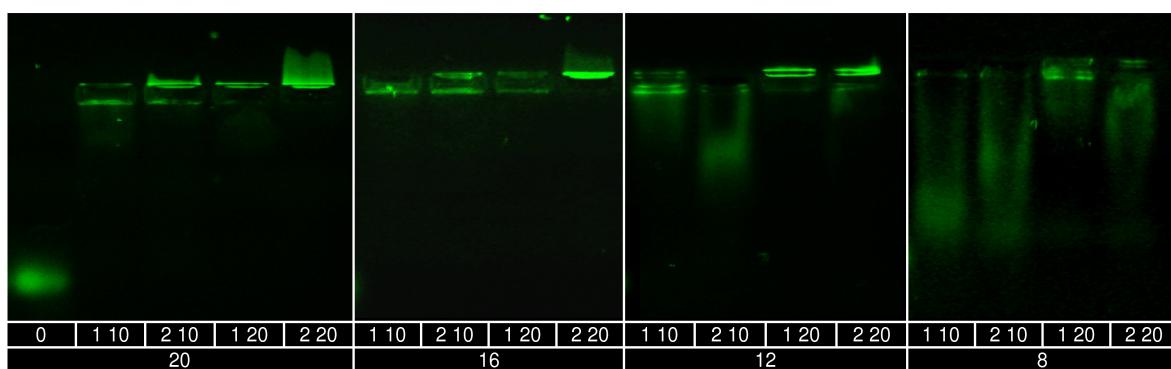
276 **Figure 1.** Gel electrophoresis data for PEI (1) and AKX-182 (2) complexes with (TCAG) $n$  DNA (last  
 277 row - length of the DNA), "0" - free DNA (similar blots were observed irrespective of DNA length).  
 278 DNA concentration was 2.5  $\mu$ M (calculating on DNA chains), N/P ratio = 20.

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**Table 3.** Types of complexes observed in DNA - polymer interactions (pH = 7.4)<sup>1</sup>

DNA length	N/P ratio									
	PEI					AKX-182				
	5	10	20	40	80	5	10	20	40	80
4	N	N	N	N	N	N	N	N	N	N
8	N	-	-,+	-,+	+	N	-	-,+	-,+	+
12	-,+	-,+	+	+	+	-	-	-,+	+	+
16	-,+	-,+	+	+	+	-,+	-,+	+	+	+
20	-,+	-,+	+	+	+	-,+	-,+	+	+	+
24	-,+	-,+	+	+	+	-,+	-,+	+	+	+

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<sup>1</sup> "N" - no interaction; "-" - a diffuse spot in negative area; "+" - neutral or slightly positive band.

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**Figure 2.** Gel electrophoresis data for PEI (1) and AKX-182 (2) complexes with duplex DNA (last row - length of the DNA), "0" - free DNA (similar blots were observed irrespective of DNA length). DNA concentration was 2.5  $\mu$ M (calculating on DNA chains), N/P ratio presented as second number in first legend row.

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### 3.2. Synthesis of poly(*N,N*-dimethylacrylamide) with grafted LCPA chains

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The desired PDMAAm with grafted LCPA chains was synthesized starting from poly(acryloyl chloride) which was converted into PDMAAm containing activated acrylic ester (oxysuccinimide) units. This copolymer was fractionated by precipitation from  $\text{CH}_2\text{Cl}_2$  with n-hexane. The mixture of LCPA oligomers was fractionated with flash-SEC and three fractions containing oligomers with average 13, 16 and 27 nitrogen atoms were obtained. These oligomers as well as non-fractionated sample were involved in the reaction with activated polymer giving rise to the target polymers (Scheme 3, Table 2). Grafting degree of the LCPA chains is 1.8-3.4,  $M_w/M_n$  ratio is 1.16-1.24 for polymers with fractionated LCPA chains which is significantly lower than the values for polymers based on poly(acryloyl chloride) without fractionation (1.37-1.58 [13]).

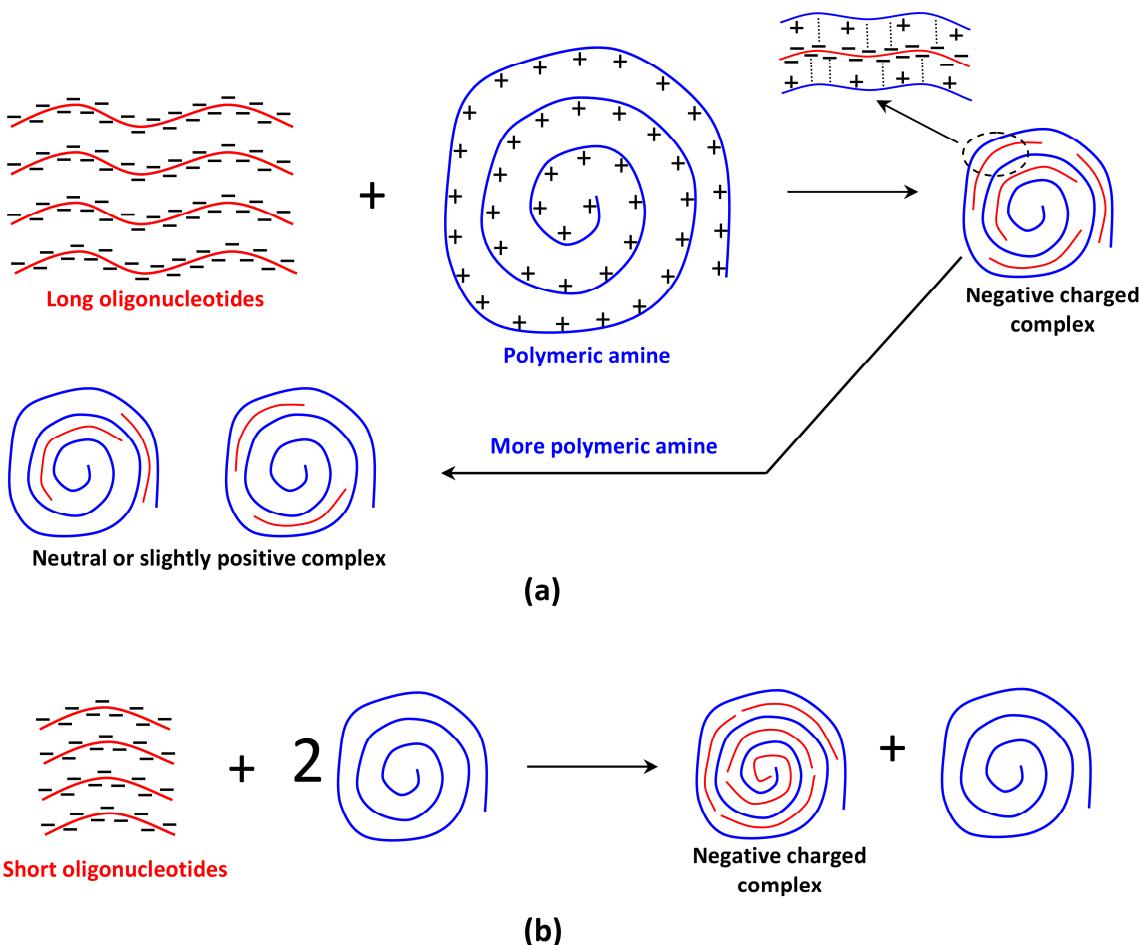
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### 3.3. Interaction of LCPA containing polymers with oligonucleotides

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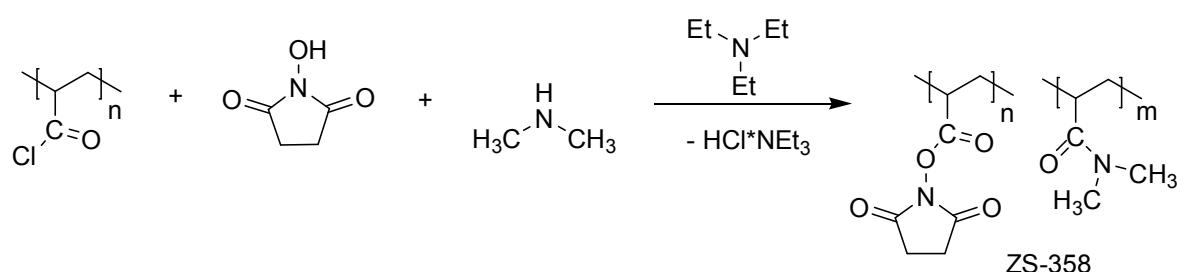
Complexation between PDMAAm with grafted LCPA and oligonucleotides was studied with gel electrophoresis (Figure 4). AKX-371-5 sample only interacts with 16-24-mer DNA N/P = 20. The other polymers give complexes at N/P > 40. This behaviour corresponds to the data with PEI and AKX-182 samples: AKX-371-5 contains long (>16 units) side polyamine chains and all these chains exceed the critical length for interaction with DNA. AKX-371-1 (LCPA chain length - 13.4) is not active in the complexation because LCPA chain is not enough for the reaction with DNA and a stacking structures from several LCPA chains are not possible due to low grafting degree. AKX-371-2 and AKX-371-3 contain a certain fraction of LCPA longer than 20 units and increase of the polymer concentration results in approximately full complexation at N/P = 150.

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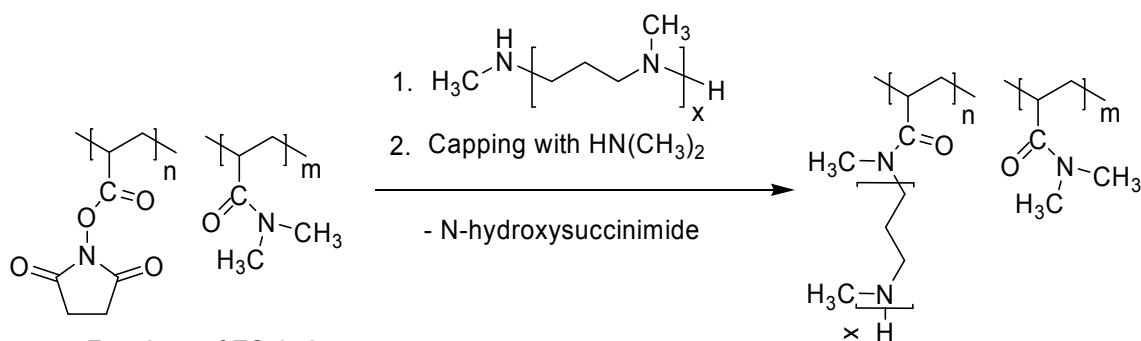


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**Scheme 2.** Interaction of long (a) and short (b) oligonucleotides with polymeric amine.

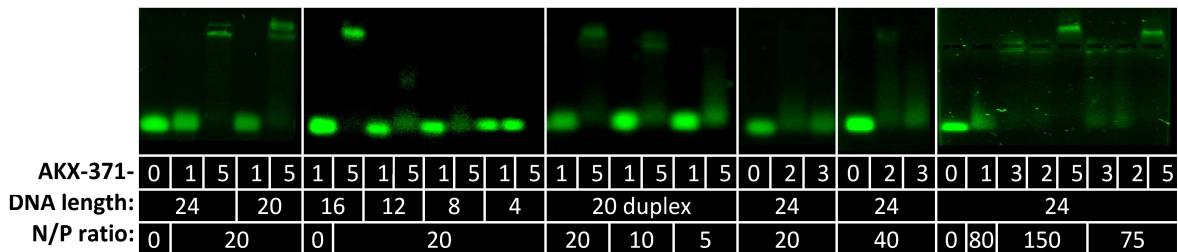
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**Scheme 3.** Synthesis of poly(N,N-dimethylacrylamide) with grafted LCRA chains

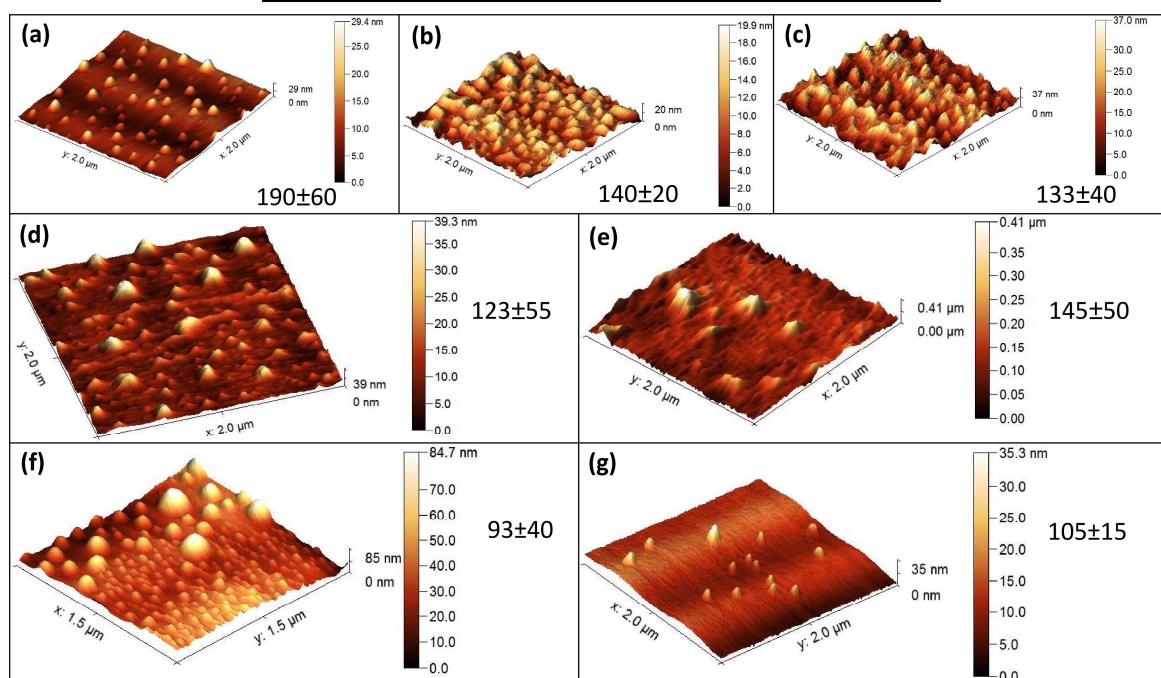


**Figure 4.** Gel electrophoresis data for AKX-371-N ( $N = 1,2,3,5$ ) complexes with DNA. "0" - free DNA (similar blots were observed irrespective of DNA length). DNA concentration was  $2.5 \mu\text{M}$  (calculating on DNA chains).

Interaction of the polymers with 19-mer double stranded si-RNA against vascular endothelial growth factor (VEGF) was studied with RiboGreen assay [36] which allows to measure concentration of free, non-complexed RNA (Table 4). The most stable complexes were obtained with ZS-371-5 polymer and the complexation proceeds to a large degree at N/P ratio 20 which corresponds to formation of partially negative charged complexes. Complexes of the LCPA containing polymers with single and double strand 20-mer DNA exist as 100-200 nm particles (Figure 5) which is appropriate for internalization into living cells [37-39].

**Table 4.** Interaction of si-RNA with LCPA containing polymers

Polymer	N/P ratio	Free RNA, %
ZS-371-1	20	6.4
ZS-371-2	20	5.6
ZS-371-3	20	11.9
ZS-371-5	20	3.8
ZS-371-1	80	8.9
ZS-371-2	150	5.8
ZS-371-3	155	7.4
ZS-371-5	180	5.3



**Figure 5.** AFM images of AKX-371-5 polymer (a), complexes of polymers AKX-371-5 (b and c), AKX-371-2 (d and e) and AKX-371-3 (f and g) with single-strand (b, d and f) and double strand (c, e and g) 20-mer DNA. N/P ratios are 180 (b and c), 150 (d and e) and 155 (f and g). AKX-371-1 did not

328 give particles appropriate for AFM study. Numbers near pictures present average size of the  
329 particles, nm.

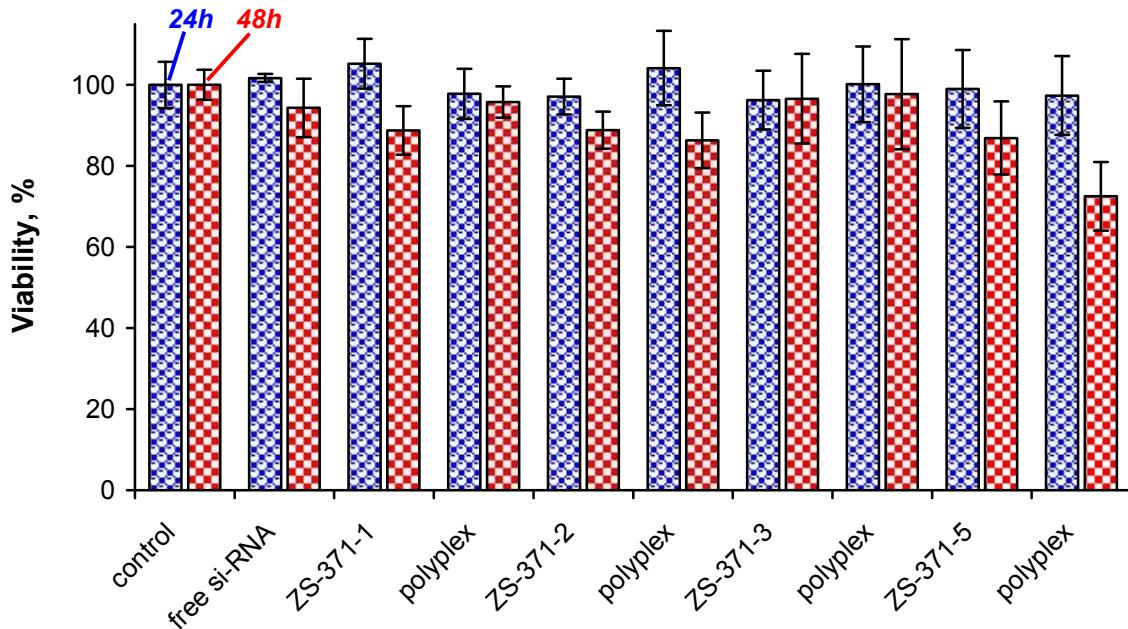
330 **3.4. *In vitro* study of transfection activity of si-RNA polyplexes based on LCPA containing polymers**

331 The new polymeric amines were studied as transfection agents with the use of 19-mer si-RNA  
332 which can silence vascular endothelial growth factor (VEGF) [40]. Viability (Figure 6) and  
333 internalization assays (Figure 7) were performed to estimate the transfection potential of the new  
334 polymers.

335 The free siRNA does not show significant reduction in viability after 24 h as well as 48 h. The  
336 free ZS-371-1 shows a slight but insignificant decrease in the viability of the A-549 cells after 48 h but  
337 its complex with anti-VEGF siRNA does not show any toxicity after both time points. This suggests  
338 that the surface charges on the uncomplexed polymer may interfere with the viability of the  
339 proliferating cells. Complexation with the siRNA neutralizes the surface charges that is manifested  
340 in the absence of any marked decrease in the cell viability after 48 h. The ZS-371-2 polymer does not  
341 alter the viability at both time points indicating its lack of toxicity. However, its polyplex with  
342 siRNA shows a slight reduction in the viability after 48 h. This suggests an effect of the VEGF  
343 silencing on different signalling pathways in the polyplex treated cells. In comparison, the ZS-371-3  
344 did not alter the cell viability after 24 h or 48 h both in the uncomplexed as well as complexed forms.  
345 This may arise due to poor internalization in to the cells or due to poor complexation or extremely  
346 high affinity complexation that restricts the release of the siRNA or due to poor ability to escape  
347 from the endosome. But, the internalization studies (Figure 6a) show that ZS-371-3 polyplex  
348 localizes rapidly within 4 h in the cells thereby ruling out the possibility of poor internalization.  
349 Further, ribogreen assay reveals that the complexation efficiency is about 92% and hence poor  
350 siRNA could not explain the observed cell viability. Therefore, it could be possible that the siRNA  
351 release or endosomal escape may be the factors that could have influenced the lack of toxicity for this  
352 formulation. The viability of cells treated with ZS-371-5 shows a small but insignificant reduction in  
353 the viability after 48 h but its complex with the anti-VEGF siRNA shows significant decrease in the  
354 cell viability after 48 h indicating the effect of VEGF silencing in the cells. Several reports [41,42] have  
355 indicated that VEGF silencing can mediate cell death and decreased proliferation by modulating the  
356 PI3K/Akt and Notch signalling pathways respectively. Internalization studies show (Figure 6b)  
357 significant internalization of si-RNA after 6h. si-RNA localizes mostly in cytoplasm (not in nucleus)  
358 which is necessary for the effective silencing. The ribogreen assay revealed that ZS-371-5 shows the  
359 maximum complexation efficiency followed by the ZS-371-2 which is explainable with the length of  
360 grafted LCPA chains. This could be one of the factors that could have influenced the cell viability  
361 assay. Additional factors such as endosomal escape and release of siRNA could also have influenced  
362 the viability that needs further validation.

363 **Conclusions**

364 We have found that critical length for oligonucleotides in interaction with polymeric amines is  
365 8-12 units and complexation at these length can be accompanied by "all-or-nothing" effects. New  
366 polymers with grafted polyamine chains were obtained and studied in complexation with DNA and  
367 RNA oligonucleotides. The most effective interaction and transfection activity was found for a  
368 sample with average number of nitrogens in polyamine chain equal to 27, i.e. for a sample which all  
369 grafted chains are longer the critical length for polymeric amine - oligonucleotide complexation.

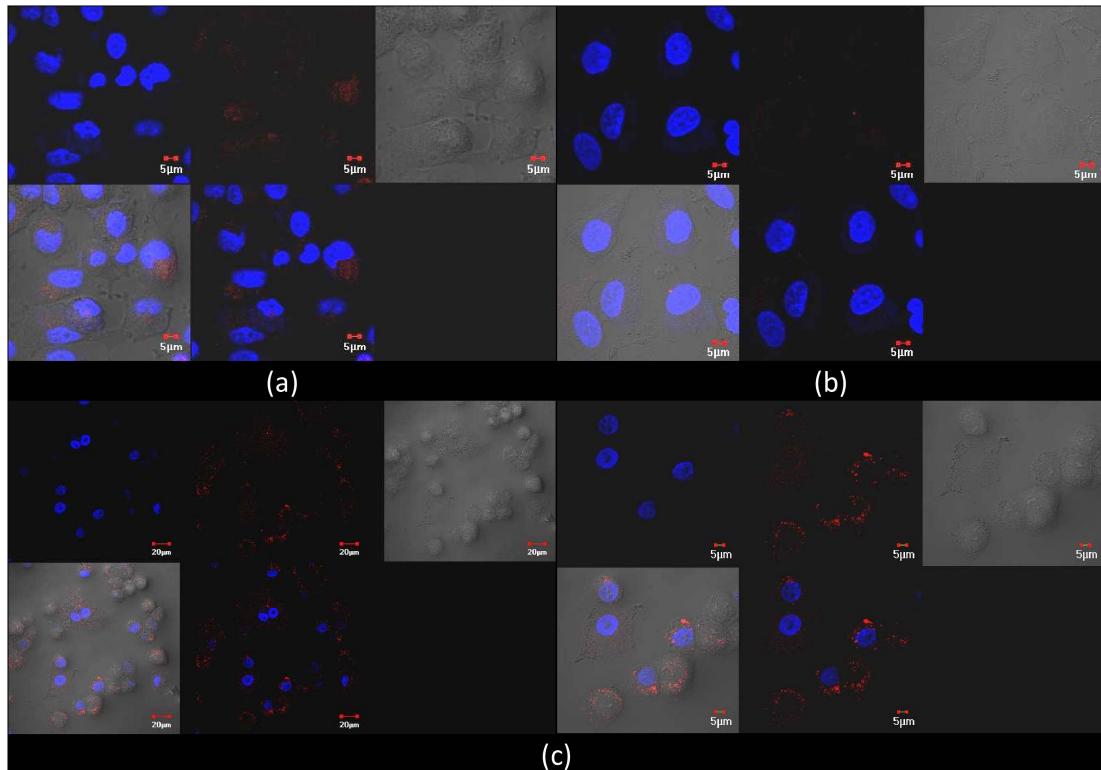


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**Figure 6.** Cell viability of A549 cancer cells after treatment with different polymers both independently as well as after complexation with anti-VEGF si-RNA.

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**Figure 7.** Internalization of polyplexes based on ZS-371-3 (a) and ZS-371-5 (b and c) polymers in A549 cancer cells after 4 (a and b) and 6 (c) h observed using laser scanning confocal microscopy. Scale bar represents 5 (a, b and c, right) and 25 (c, left)  $\mu$ m.

378 **Supplementary Materials:** The following are available online at [www.mdpi.com/link](http://www.mdpi.com/link), Figure S1. Liquid  
379 chromatography-mass spectrometry (LC-MS) analysis of ZS-309 fractions; Figure S2. 1H NMR spectra of LCPA  
380 (ZS-309) fractions.; Figure S3 1H NMR spectra of copolymers (derivatized with TFA) in DMSO-d6.

381 **Author Contributions:** Vadim V. Annenkov conceived and designed the experiments on polymer synthesis  
382 and study of polymer - DNA interactions; Uma M. Krishnan conceived and designed the experiments on  
383 polymer - si-RNA interaction and *in vitro* study of the polyplexes; Elena N. Danilovtseva performed  
384 experiments on study of polymer - DNA interactions and analyzed the data; Viktor A. Pal'shin and Stanislav N.  
385 Zelinskiy performed synthesis of new polymers; Gayathri Kandasamy performed the experiments on polymer -  
386 si-RNA interaction and *in vitro* study of the polyplexes; and Vadim V. Annenkov and Uma Maheswari  
387 Krishnan wrote the paper.

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393 **Conflicts of Interest:** The authors declare no conflict of interest.

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