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# Purification Treatment on Polluted River via Combinations of Gravel Contact Oxidation Treatment and Surface Flow Constructed Wetlands—A Case Study in Changhua County, Taiwan

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**Abstract:** Polluted rivers are a primary problem in Changhua County, Taiwan, due to rapid and massive developments in agriculture and industry. In this study, samples consisted of water influent from the Yangzi-Cuo River, Ci-Tong Jiao, Changhua County. Ecology technology, a combination of gravel contact oxidation and surface flow constructed wetlands, was employed to enhance the water quality. Ecology technology is a useful and ordinary process that focuses on original treatment of pollution through chemical, physical, and biological procedures based on the mediums of soil, plant, water, and microorganisms from pure environments. Moreover, this study aimed to minimize the river pollution index (RPI) of the Yangzi-Cuo River by using combinations of gravel contact oxidation and surface flow constructed wetlands. The inflow volume of the water samples in situ was found to be 1534–2261 cubic meters per day (CMD). The pollution in the water sample mostly involved suspended solid (SS), ammoniacal nitrogen (NH<sub>3</sub>-N), total phosphorus (TP), total nitrogen (TN), and biochemical oxygen demand (BOD), and the values of the above indexes were found to be 10.0–26.7 mg/L, 0.9–14.0 mg/L, 1.2–14.1 mg/L, 11.1–18.2 mg/L, and 0.5–3.6 mg/L, respectively. Efficiencies of pollution elimination in SS, NH<sub>3</sub>-N, TP, TN, and BOD were found to be 24.2–93.1%, 58.3–86.2%, 5.2–85.0%, 59.4–77.2%, and 46.3–76.4%, correspondingly, after purification treatment via ecology technology. Thus, ecology technology is a valid means of purification treatment for polluted rivers.

**Keywords:** ecology technology; gravel contact oxidation; surface flow constructed wetlands; water quality; river pollution index

## 1. Introduction

Household wastewater is a type of pollution generated from human beings' daily activities. Excrement from human beings and domesticated animals has generally been used to culture crops over the past five decades. Household wastewater is discharged into rivers and soil, and their self-purification functions degrade these low-grade toxic and uncomplicated wastes [1–5]. However, population centralization, the enhancement in living quality, the development of technology, and variations in living style have resulted in an increase in discharged wastewater

with more complicated water content, which such rivers and soil, with their self-purification functions, cannot handle. As a result, river and soil pollution is becoming increasingly serious. Industrial and agricultural activities are central to the economy of Changhua County, and the population of Changhua County has reached 1.3 million, consequently increasing this wastewater. Based on the estimated historical data from the Environment Protection Administration (EPA), river pollution in Taiwan has been due to industrial wastewater in the past five decades, and, after the implementation of environment protection regulations, the problem of industrial wastewater has been resolved.

However, approximately 47% of wastewater production comes from households, and such household wastewater is now the most serious issue in terms of wastewater production in Taiwan. Moreover, the popularizing rate of public sewage treatment was only 6.3% at 2008 in Changhua County, and most household wastewater, with the rain, is usually discharged directly into rivers through a drain. As a result, discharged household wastewater dramatically degrades the water quality of river. There are many purification technologies, including membrane filtration, wet pyrolysis, coagulation, and anaerobic digestion [6–9], used in waste and wastewater treatment. In addition, the EPA have started to announce policies and strategies regarding wastewater pollution treatment to the local government since 2002. The announced method included gravel contact oxidation treatment, surface flow constructed wetlands, and an aeration facility for wastewater purification [10–18]. The principle of such wastewater purification is the interaction of wastewater and natural environmental factors, such as oxygen, soil, microorganism, and plants, and the aim is to purify water quality and thus eliminate the pollution of rivers via the decrease of discharged wastewater.

The onsite wastewater treatment system (OWTS) is a novel ecological engineering designed to improve river quality in Taiwan. The efficiency of the OWTS strives for wastewater treatment and water quality purification via natural mechanisms and energy, and the mechanism is described as using fundamentally physical, chemical, and biological methods with consideration of onsite environmental conditions, including pollution category, soil quality, gradient value, flow rate, and gravel type, to achieve maximum efficiency in wastewater treatment. Metcalf and Eddy [19] declared that there were two natural wastewater treatment systems, including a soil/land treatment system and an aquatic-based treatment system, and these two systems consisted of individual methods. The soil/land treatment system contained slow rate, rapid infiltration, and overland flow methods, and the aquatic-based treatment system involved surface flow constructed wetland, natural wetland, and aquatic plant methods. However, gravel contact oxidation treatment is the most common method used in Taiwan, and this method was initially developed by the Japanese government for river quality purification [20].

The EPA collected domestic river quality data in 25 locations by conducting a project with a river quality purification technology design in 2005–2006 and then summarized the results. The variation in biochemical oxygen demand (BOD), ammonia nitrogen ( $\text{NH}_3\text{-N}$ ), total phosphorus (TP), and suspended solid (SS) presents efficiency in wastewater pollution removal via gravel contact oxidation treatment, and this can be applied as a reference for the operation of river quality natural purification treatment. As a result, this study employed gravel contact oxidation treatment to purify the water quality of the Yangzi-Cuo River in Ci-Tong Jiao, Changhua County.

## 2. Experimental and Methodology

### 2.1. Introduction of Gravel Contact Oxidation Treatment

Gravel contact oxidation treatment is a fast method compared to other natural purification technologies, and its principle is that wastewater, after it has flown rapidly through gravel, can be decomposed by the bio-membranes attached to the gravel [21]. Gravel is the main component used in gravel contact oxidation treatment, but material with a particle diameter of 0.6–1.2 mm, as a subordinate substance, can also be used. For instance, coal cinder and coke are also suitable for application in this treatment. There is usually a two-layer arrangement of gravel to enhance the

efficiency of wastewater decomposition, and the intervals of the upper and lower gravel layers are loose and tight, respectively. Furthermore, the flow rate is high to maintain an aerobic status in the upper layer, and is low to maintain an anaerobic status in the lower layer. The condition of the two gravel layers is defined as a double contact system, and the average hydraulic loading rate (HLR) is approximately 0.47 m/day. Particle diameter is the priority factor in wastewater pollution treatment, and the flow rate is rather slow if the particle size is too small to result in a high efficiency of wastewater pollution removal. Moreover, Seidel [22] investigated that gravel with a particle size less than 0.06 m is suitable for aquatic plant growth because such an environment provides a surface area that is high enough for many microorganisms to attach to, and this allows for the growth of the aquatic plant.

Based on Darcy's law [23–27], relationships regarding water flow rate in sand filters were determined through a report on the construction of the Dijon municipal water system in France. The volume flow rate ( $q$ ) is represented as a gradient of elevation ( $z$ ) in the horizontal direction of water flow ( $x$ ). Equation (1) was used to describe the relationship of  $Q$ ,  $z$ , and  $x$ :

$$q = -K_s \frac{dz}{dx} \quad (1)$$

where  $q$  is the volume flow rate;  $K_s$  symbolizes the saturated hydraulic conductivity;  $z$  represents the gradient of elevation; and  $x$  depicts the horizontal direction of water flow. Moreover, Equation (2) discussed the connection of  $q$  and the volume flow rate of underwater ( $Q$ ).

$$Q = qA \quad (2)$$

where  $Q$  stands for the volume flow rate underwater, and  $A$  describes the area of the gravel wetland. In a more detailed way, multiplication of the gradient of elevation and the width of the wetland ( $l$ ) yields an outcome of  $A$ , as depicted in Equation (3), where  $l$  is the width of the wetland.

$$A = lz \quad (3)$$

Equation (4) can be obtained after the substitution of Equations (2) and (3) into Equation (1), as shown in the following equation.

$$Q = -K_s lz \frac{dz}{dx} \quad (4)$$

After integration on both sides of the equal sign, Equation (5) can be acquired, and is illustrated as follows:

$$Q \int_0^x dx = -K_s l \int_{z_0}^z z dz \quad (5)$$

where  $z_0$  represents the initial elevation of inflow, and  $z$  stands for the current elevation of inflow at any time. A summary of Equation (5) can present the outcome of Equation (6), shown in the following equation.

$$\frac{2Qx}{K_s l} = z_0^2 - z^2 \quad (6)$$

Furthermore, Equation (5) can also depict a different functional equation, as shown in Equation (7):

$$z = \sqrt{z_0^2 - \frac{2Q}{K_s l} x} \quad (7)$$

Equation (8) mentions the hydraulic retention time (HRT) of the flow in the wetland and is as follows:

$$HRT = \frac{V}{Q} \quad (8)$$

where  $HRT$  represents the hydraulic retention time, and  $V$  is the volume of the water inside the wetland. Moreover,  $V$  is calculated by the multiplication of porosity ( $\phi$ ), the width of the wetland ( $l$ ), the elevation ( $z$ ), and the horizontal direction of the water flow ( $x$ ), as depicted in Equation (9):

$$V = \phi l z x \quad (9)$$

where  $\phi$  represents porosity.

Equation (10) can be derived from the calculation of Equation (8) as follows:

$$dHRT = \frac{dV}{dQ} \quad (10)$$

After the substitution of Equation (9) to Equation (10), a new combination can be acquired, as illustrated in Equation (11):

$$dHRT = \frac{\phi l}{Q} z dx \quad (11)$$

In addition, the substitution of Equation (7) to Equation (11) can derive Equation (12), presented in the following equation.

$$\int_0^{HRT} dHRT = \frac{\phi l}{Q} \int_0^x \left( z_0^2 - \frac{2Q}{K_s l} x \right)^{1/2} dx \quad (12)$$

Equation (13) can be obtained by summary of Equation (12) as follows:

$$HRT = \frac{\phi K_s l^2}{3Q^2} z_0^3 \left[ 1 - \left( 1 - \frac{2Q}{K_s l z_0^2} x \right)^{3/2} \right] \quad (13)$$

In the assumption situation, a first-order reaction can match the condition of wastewater purification in the wetland, and the above condition can be described in Equation (14), as depicted in the following equation.

$$C = C_0 e^{-kHRT} \quad (14)$$

where  $C_0$  represents the concentration of the inflow;  $C$  is the concentration of the outflow;  $k$  depicts the coefficient of the first-order reaction.

According to the perspective of Equations (13) and (14), a decomposition reaction of wastewater pollution can be presented, and the two factors of  $k$  and  $HRT$  are very important among those hydraulic parameters. Furthermore, the value of  $HRT$  varies according to many factors, such as saturated hydraulic conductivity ( $K_s$ ), porosity ( $\phi$ ), volume of flow, and elevation ( $z$ ). The factors of hydraulic conductivity ( $K_s$ ) and porosity ( $\phi$ ) are useful for the construction, and the operational situation of the wetland is determined by the volume of flow and elevation ( $z$ ). Decomposition reaction of pollution is altered by the growth condition of the biomembrane.

## 2.2. The Proper Design Processes of a Constructed Wetland

There are six core processes to design a constructed wetland, and those processes are listed as follows:

- (1) estimation of the amount of rainfall via collection of atmospheric data in a specific area;
- (2) evaluation of the amount of vaporization by an evaporation pan method and an on-site evaporating test;

- (3) calculation of soil infiltration data;
- (4) calculation of HRT, the volume of flow, and the design area of the constructed wetland by applying a first-order model—the  $k-C^*$  model (common used in the surface flow wetland) or the Monod kinetics model (general applied in the subsurface flow wetland) with consideration of environmental temperature and species of pollution [28];
- (5) computation of the average volume of flow via hydrographic data for adjusting the volume of flow and the design area;
- (6) collection of geographical data of the constructed wetland.

The conditions of wastewater treatments in this study are described in table 1, as shown as follows. The wastewater treatments are divided as two main methods in this study, including gravel contact oxidation and constructed wetland. Total base area and water purification treatment area are designed as 9,000 square meter and 730 square meter, respectively. The inflow source is originally exhausted from household wastewater in ci-tong jiao, Changhua County with its coordinate position of (23.070949, 120.505699). Moreover, gravity type and pumping motor diversions were applied as inflow way with the treatment capacity of 2000 CMD, and hydraulic retention time (HRT) remained as 2.59 days. The original inflow concentrations of BOD<sub>5</sub>, NH<sub>3</sub>-N, SS, and TP are 50 mg/L, 80 mg/L, 50 mg/L, and 2.6 mg/L, correspondingly, and the exhausted outflow concentrations of BOD<sub>5</sub>, NH<sub>3</sub>-N, SS, and TP are 20 mg/L, 48 mg/L, 20 mg/L, and 1.0 mg/L, respectively, after wastewater treatments. As the result, the removal efficiency of BOD<sub>5</sub>, NH<sub>3</sub>-N, SS, and TP are 60%, 40%, 60%, and 60%, respectively, and the removal capacity of BOD<sub>5</sub>, NH<sub>3</sub>-N, SS, and TP are 20 Kg/D, < 5 Kg/D, < 20 Kg/D, and < 5 Kg/D, correspondingly. Expectation of purified outflow concentration of BOD<sub>5</sub>, NH<sub>3</sub>-N, SS, and TP depict as 20 mg/L, 5 mg/L, 20 mg/L, and 5 mg/L, respectively.

### 3. Results and Discussion

#### 3.1. Removal Efficiency of SS

The initial SS concentration of inflow ranged from 12.0 to 62.0 mg/L, and the removal efficiency of the wastewater via gravel contact oxidation was 3–75%, as shown in Figure 1a. The removal efficiency of wastewater tends to be high if the inflow concentration is high. As a result, SS can be easily removed by using gravel contact oxidation treatment. Furthermore, removal efficiency changed dramatically when the inflow concentration was lower than 20 mg/L, resulting in an obscured outcome. The SS concentration of outflow was less than 20 mg/L after gravel contact oxidation treatment. Removal efficiency of the surface flow constructed wetlands also changed dramatically. Figure 1b shows that the SS concentration of outflow was 15.2 mg/L in October. Decomposition of fallen leaves from aquatic plants, deposit releases from the bottom, and biomembranes dropped in the outflow of constructed wetlands, resulting in the background concentration of SS. Moreover, treatment loading increased and SS concentration was enhanced, and the removal efficiency of the combined system ranged from 46 to 96%, as depicted in Figure 1c. Furthermore, a sedimentation basin could be used to collect sands with a particle diameter above 2 mm to prevent them from flowing to the gravel contact oxidation tank without a primary sand filter, resulting in, due to the sand accumulation, a reduction in the use capacity of the gravel contact oxidation tank.

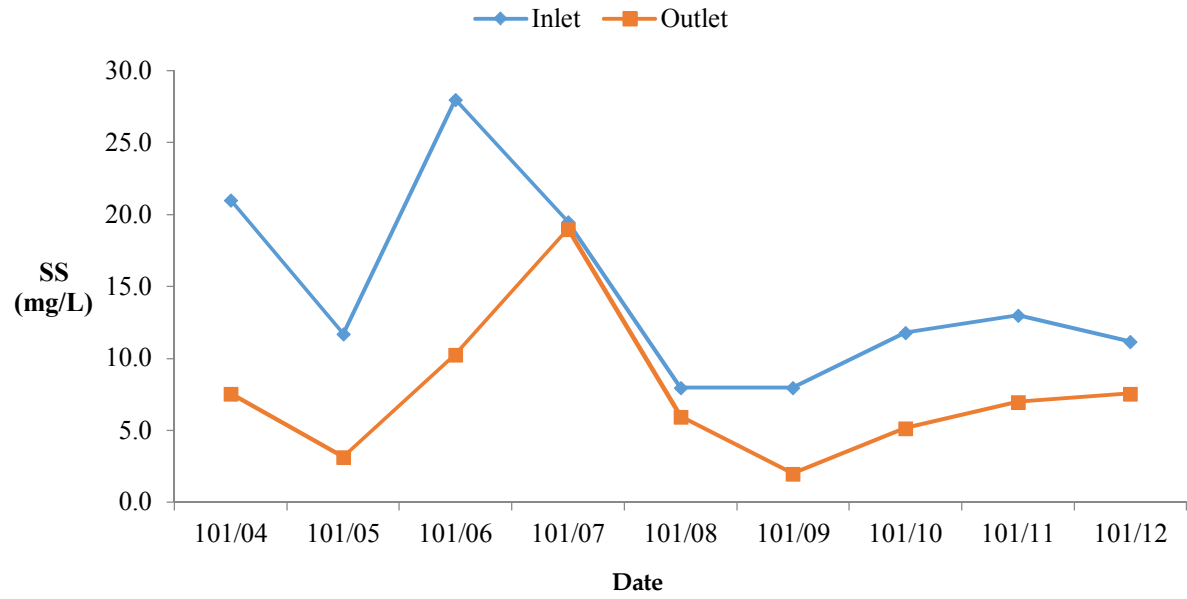


Figure 1a. SS concentration variation in inflow and outflow by using gravel contact oxygen treatment.

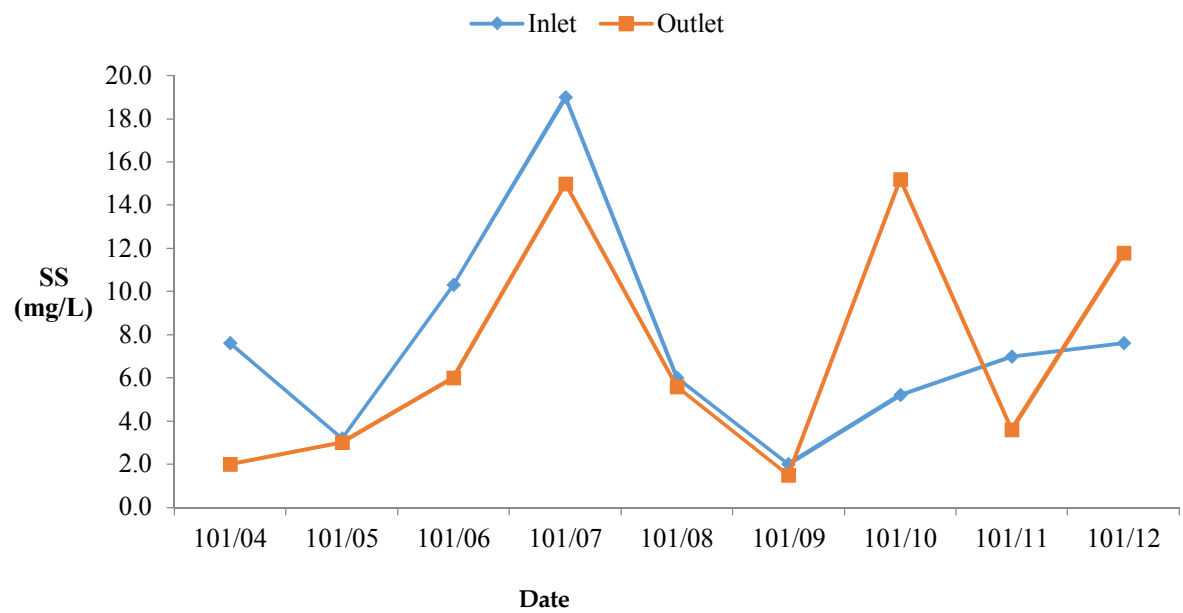
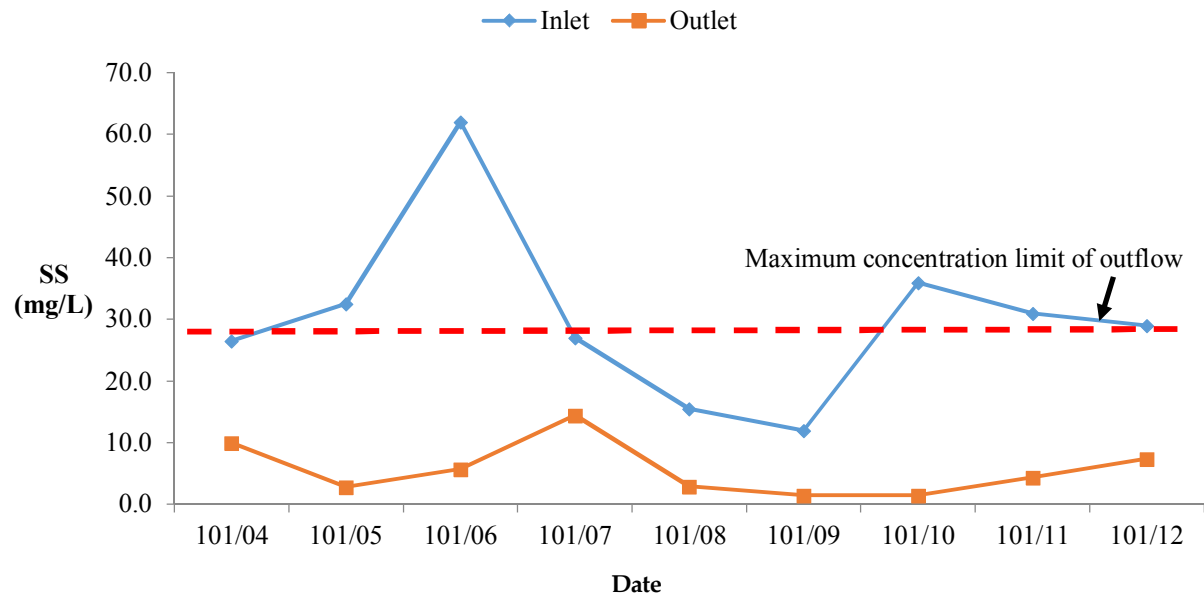


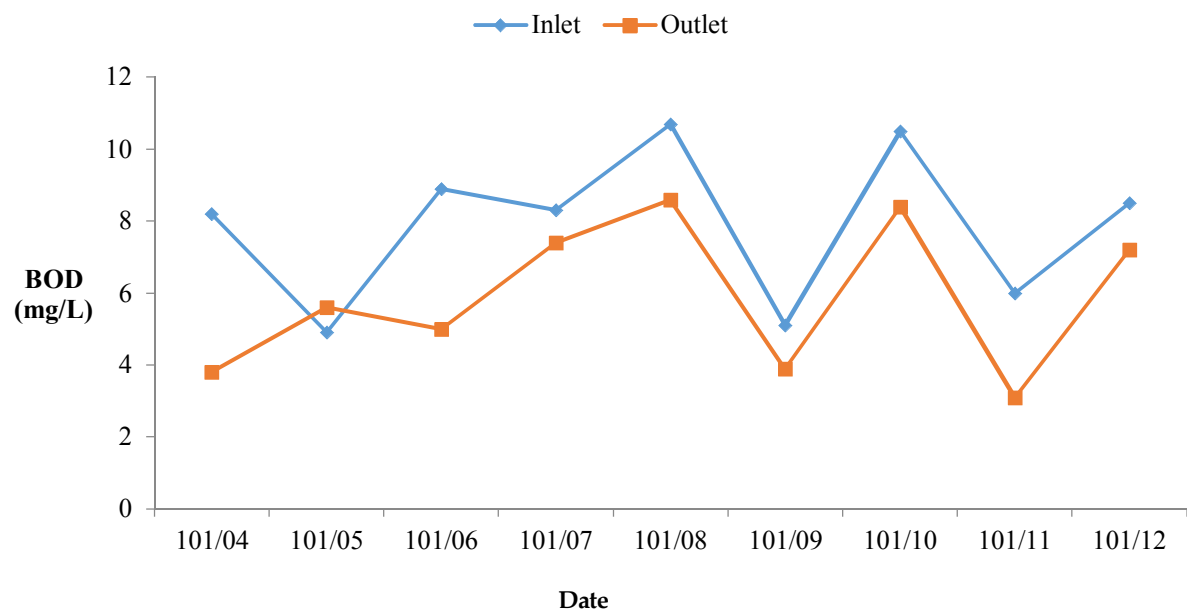
Figure 1b. SS concentration variation in inflow and outflow via the constructed wetland method.



**Figure 1c.** SS concentration variation in inflow and outflow in the combined system of gravel contact oxygen treatment and the constructed wetland method.

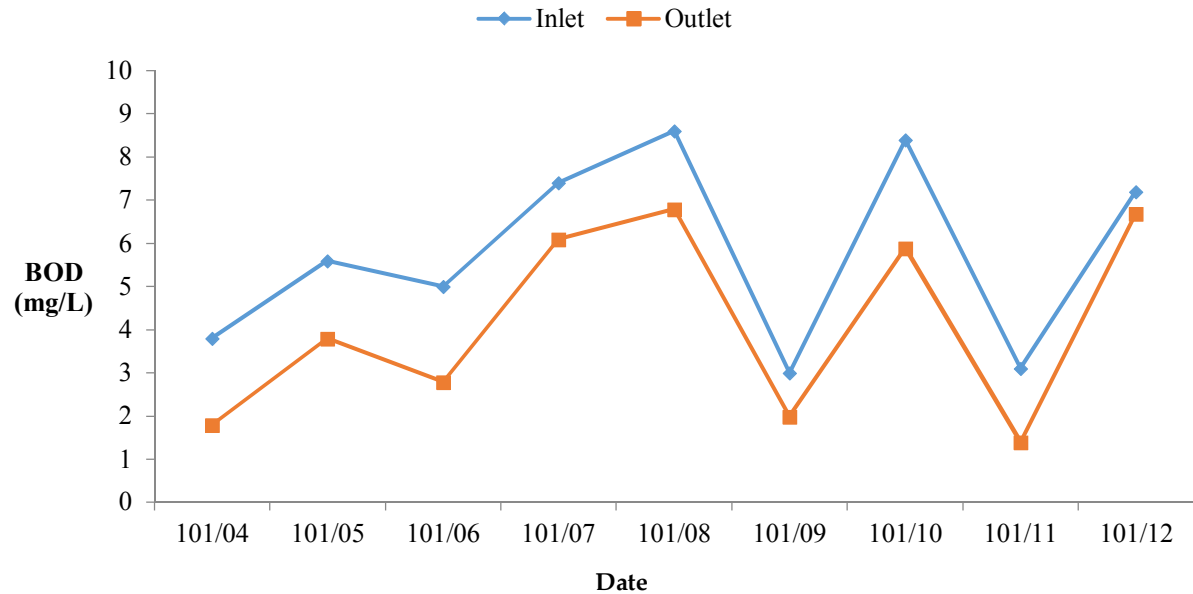
### 3.2. Removal Efficiency of BOD

The removal efficiency of the BOD was 14–54% using gravel contact oxidation treatment, as shown in Figures 2a and 2b. Figure 2b shows that the BOD concentration in May increased from 3.9 to 5.6 mg/L, and the removal efficiency of the BOD was similar using gravel contact oxidation treatment and the constructed wetland method. The BOD of the final outflow showed concentrations of 1.6–6.5 mg/L, and these results were stable and matched BOD standards. The proper BOD concentration was less than 10 mg/L. Figure 2c depicts that BOD concentrations of outflow in the combined system were lower than the maximum concentration limit of outflow, where the maximum concentration limit was 17 mg/L, after gravel contact oxygen treatment and the constructed wetland method.

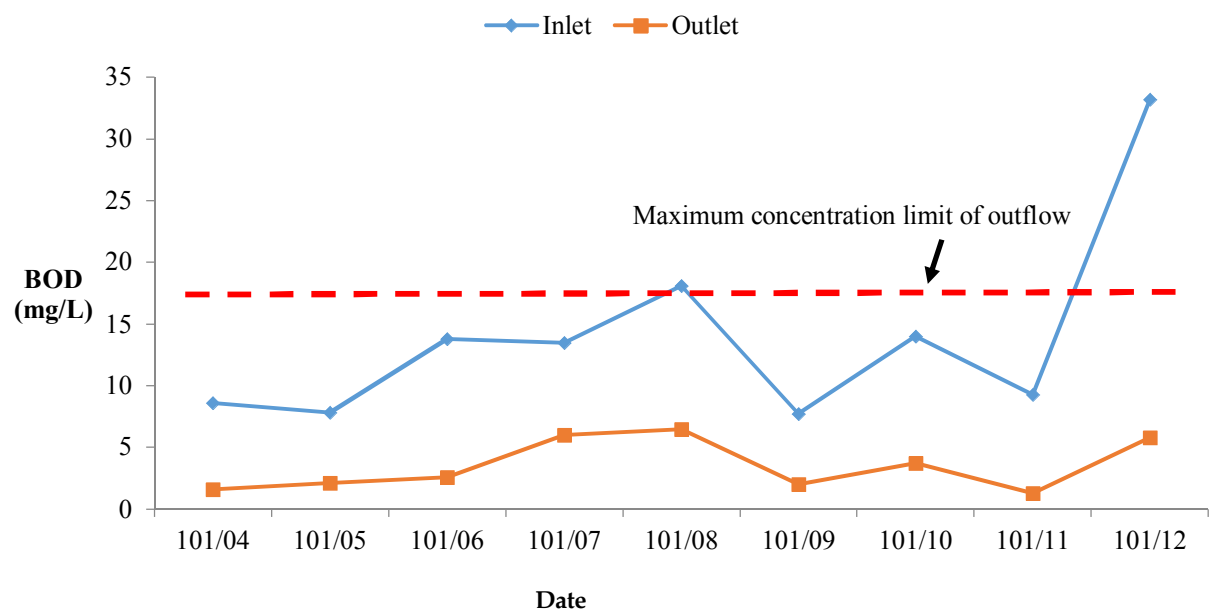


**Figure 2a.** BOD concentration variation in inflow and outflow by using gravel contact oxygen treatment.





**Figure 2b.** BOD concentration variation in inflow and outflow via the constructed wetland method.



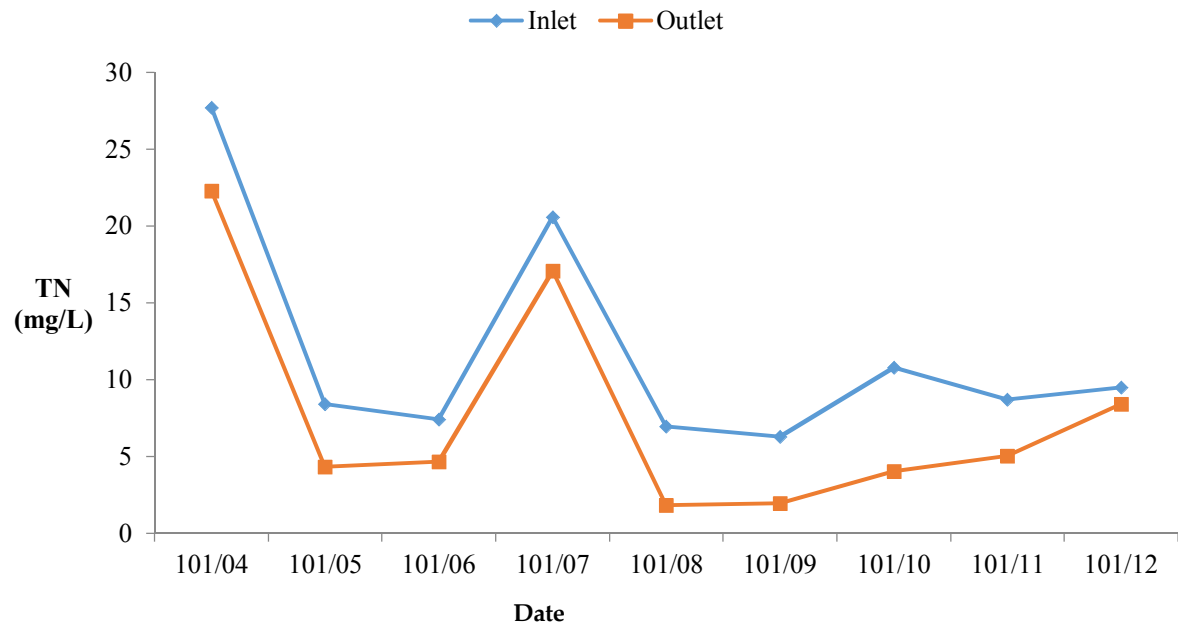
**Figure 2c.** BOD concentration variation in inflow and outflow in the combined system of gravel contact oxygen treatment and the constructed wetland method.

### 3.3. Removal Efficiency of TN

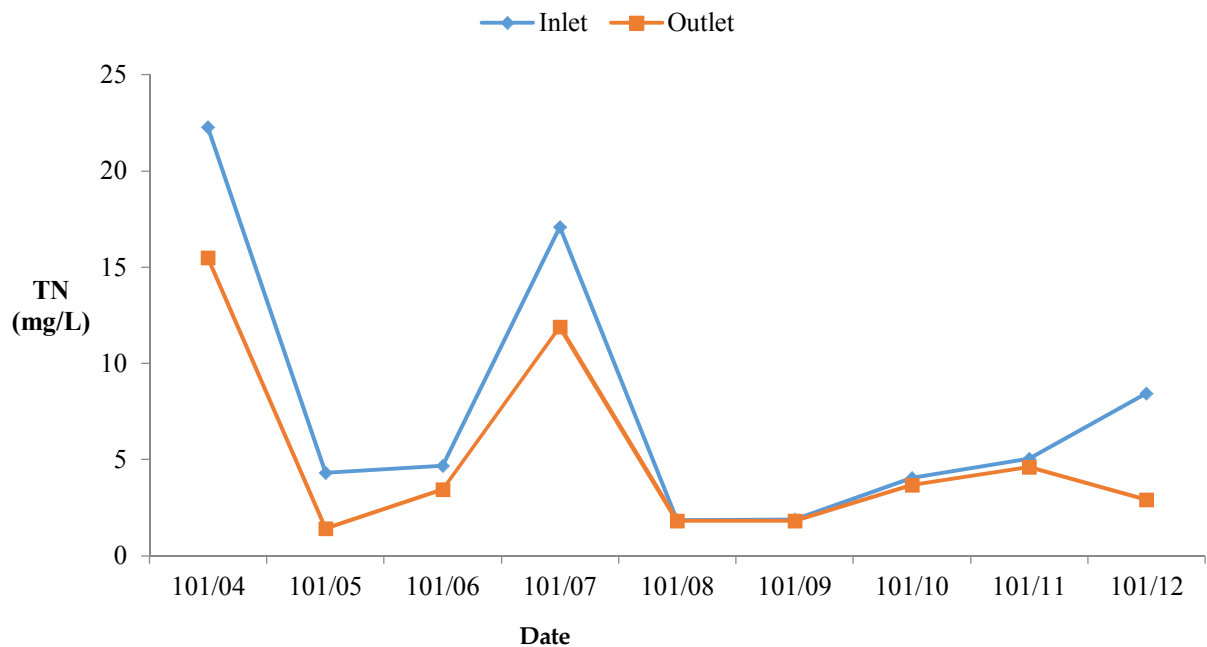
Figure 3a illustrates that the removal efficiency of TN was 17–74% via gravel contact oxidation treatment, and Figure 3b depicts that the removal efficiency of TN was 2–67% by the constructed wetland method. The removal efficiency of TN in the combined system was 57–88%. Although these removal efficiencies of TN in both June and November did not exceed 60% using gravel contact oxidation treatment or the constructed wetland method, TN concentration of outflows in June and November were 3.3 and 3.5 mg/L, respectively. Both of those values are stable and less than 5 mg/L. Figure 3c illustrates the TN concentration variation in inflow and outflow in the combined system, where the maximum concentration limit of the outflow was 7 mg/L, and most TN concentrations were lower than the maximum concentration limit, excepting the values in April and



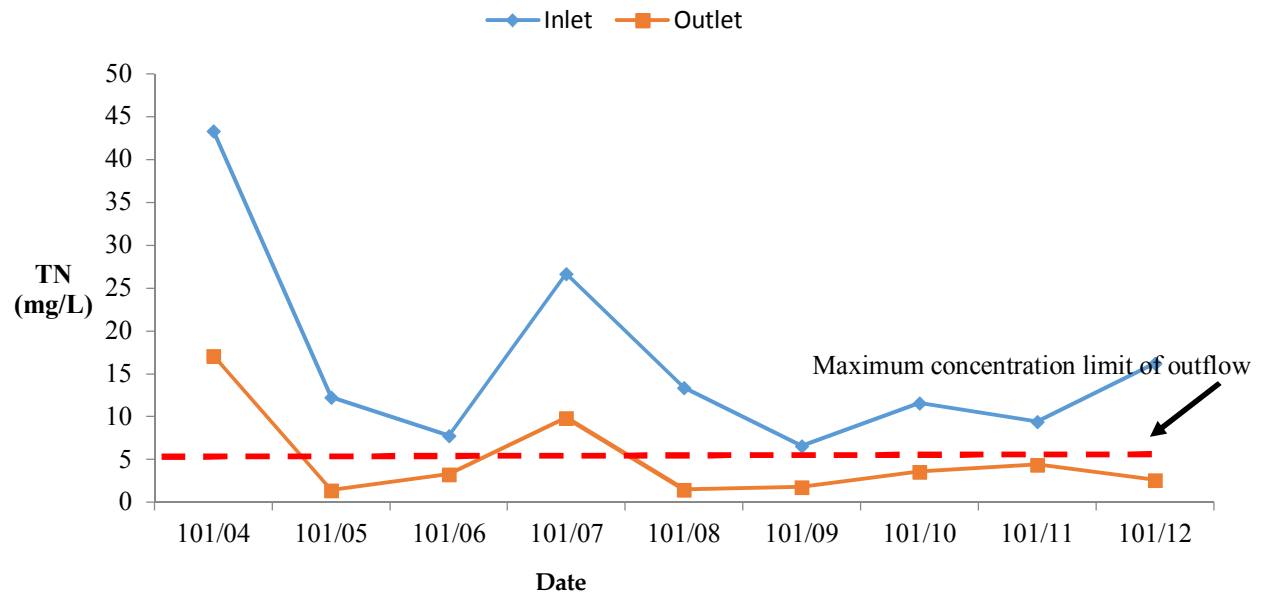
June. Although approximately 63% of TN in April can be eliminated, the TN concentration in April was the maximum value, implying that the TN concentration of outflow cannot be lower than the maximum concentration limit due to a higher loading of pollution treatment. Furthermore, there was roughly 63% of TN in June, and the TN concentration in June is the second-largest value. As a result, the loading of pollution treatment was close to 60%, and the TN removal efficiency of the combined system could reach close to 60%.



**Figure 3a.** TN concentration variation in inflow and outflow by using gravel contact oxygen treatment.



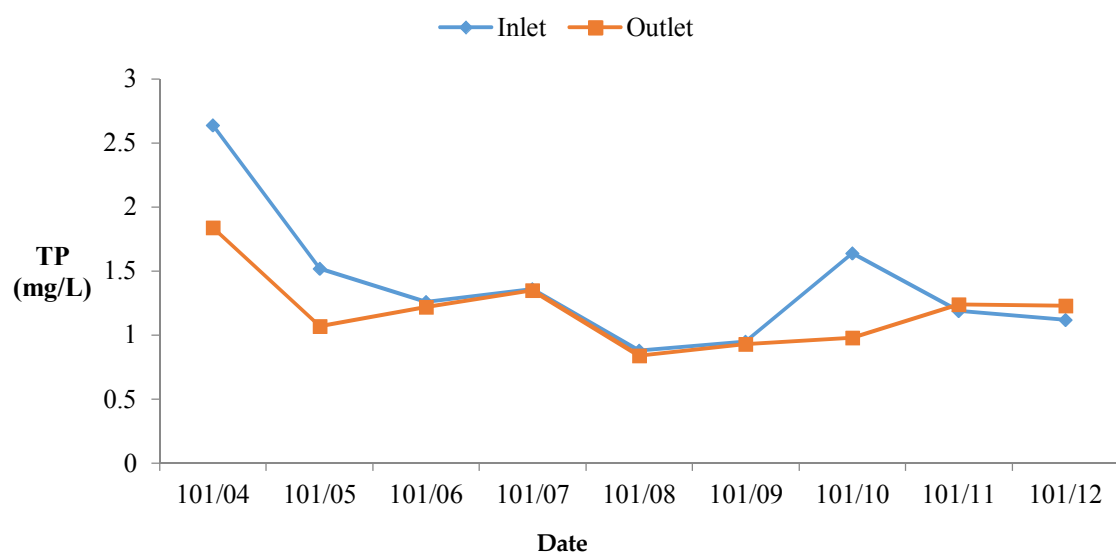
**Figure 3b.** TN concentration variation in inflow and outflow via the constructed wetland method.



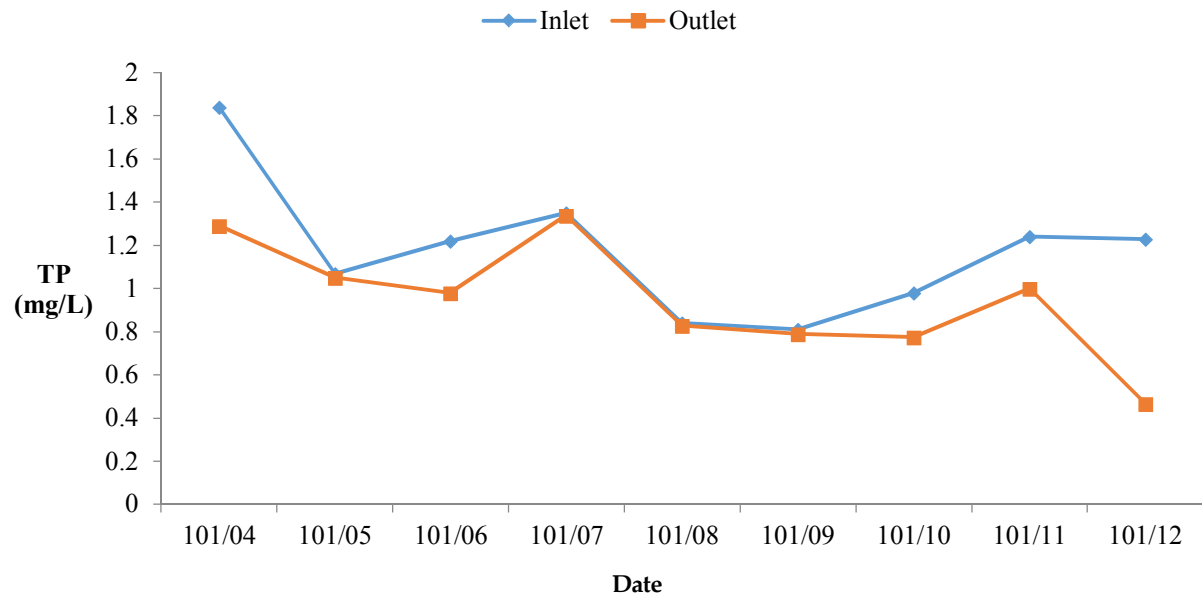
**Figure 3c.** TN concentration variation in inflow and outflow in the combined system of gravel contact oxygen treatment and the constructed wetland method.

#### 3.4. Removal Efficiency of TP

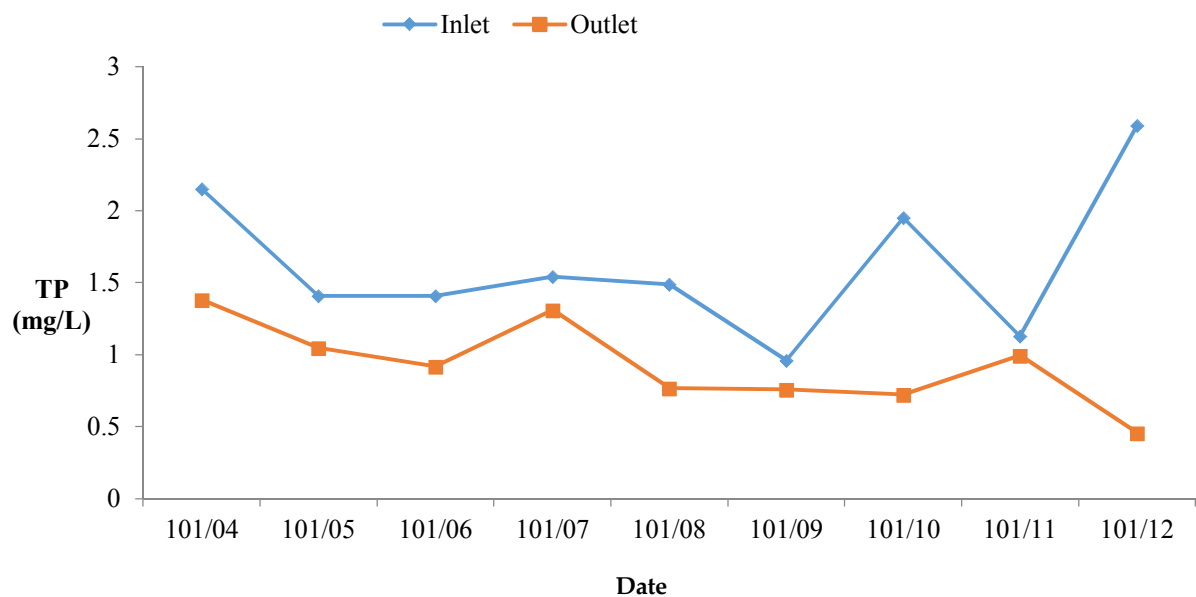
Phosphate is a nutrient of growth for microorganisms and algae, and the demand dose of phosphate as a nutrient source is only rated as 1/100 of a carbohydrate source. Moreover, Figures 4a and 4b show that the TP concentration of inflow was 2.6 mg/L, and the TP concentration of outflow was 1.4 mg/L, the maximum values of all inflow and outflow values, respectively. The removal efficiency of TP in the combined system was 12–82%, as shown in Figure 4c, and the following steps resulted in high removal efficiency values. Digestion and adsorption occurred actively in the growth of microorganisms and algae, and the removal efficiency of TP was also altered by the traits of sand samples. According to Akratos and Tsihrintzis [29], natural sands consist of Fe, Al, and Ca components. The components of natural sands were different in plastic filter samples, and phosphate ions carried electric charges to be adsorbed by positively charged ions, Fe, Al, and Ca. Furthermore, the regular disposal of spoil helped eliminate pollution.



**Figure 4a.** TP concentration variation in inflow and outflow by using gravel contact oxygen treatment.



**Figure 4b.** TP concentration variation in inflow and outflow via the constructed wetland method.



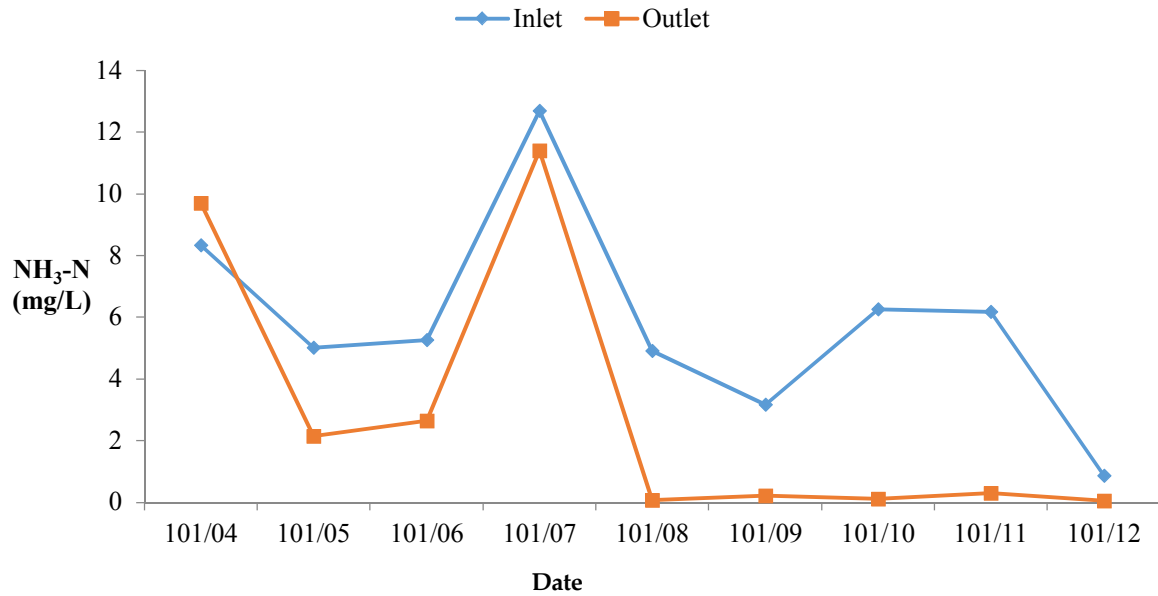
**Figure 4c.** TP concentration variation in inflow and outflow in the combined system of gravel contact oxygen treatment and the constructed wetland method.

However, the effect of reverse flushing via gravel did not obviously accompany the operational time of the gravel contact oxidation treatment and the constructed wetland method, and the sludge deposit thus worked to decrease removal efficiency. This effect is an assumption on which the reduction of the removal efficiency of TP is based. The adsorption of bottom soil and the digestion of aquatic plants for growth in the constructed wetland also enhanced the removal efficiency of TP.

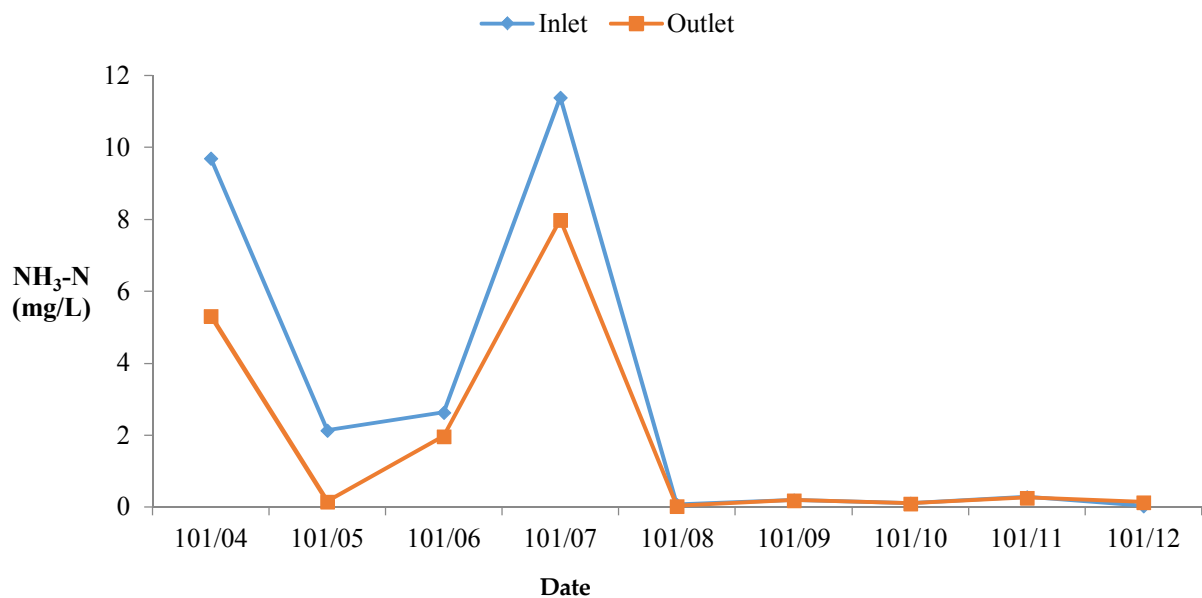
### 3.5. Removal Efficiency of $\text{NH}_3\text{-N}$

Figure 5a shows that the maximum  $\text{NH}_3\text{-N}$  concentration of inflow was 13.9 mg/L using gravel contact oxidation treatment, and the removal efficiency of  $\text{NH}_3\text{-N}$  reached a maximum value of 98%. Most  $\text{NH}_3\text{-N}$  concentrations during the whole year were lower despite higher  $\text{NH}_3\text{-N}$  concentrations in April and July, as shown in Figure 5b. Figure 5b depicts that the removal

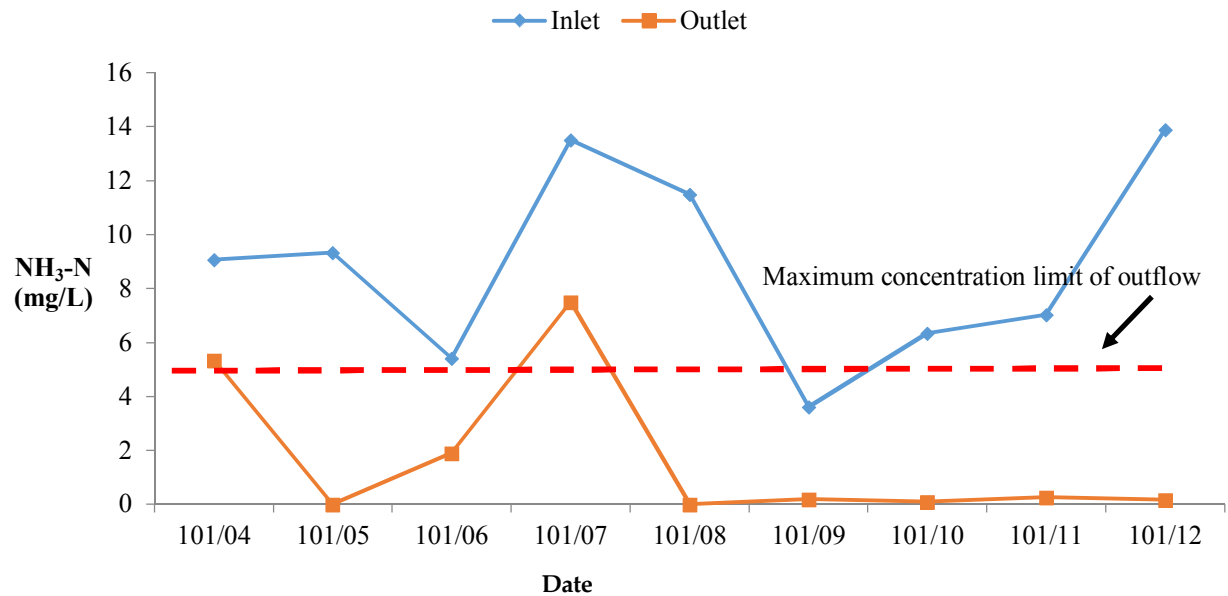
efficiency of  $\text{NH}_3\text{-N}$  reached a maximum value of 92% using the constructed wetland method. Although the  $\text{NH}_3\text{-N}$  concentration in May was lower than 0.04 mg/L and the  $\text{NH}_3\text{-N}$  concentration in August was not detected (ND), the combined removal efficiency reached 100%, as illustrated in Figure 5c. The proper reason of 100% removal efficiency was that a lot of rains diluted the concentration of wastewater due to the raining season during August to December.



**Figure 5a.**  $\text{NH}_3\text{-N}$  concentration variation in inflow and outflow by using gravel contact oxygen treatment.



**Figure 5b.**  $\text{NH}_3\text{-N}$  concentration variation in inflow and outflow via the constructed wetland method.



**Figure 5c.** NH<sub>3</sub>-N concentration variation in inflow and outflow in the combined system of gravel contact oxygen treatment and the constructed wetland method.

### 3.6. Summary of Pollution Control and Analysis of the River Pollution Index of the Inflow and Outflow

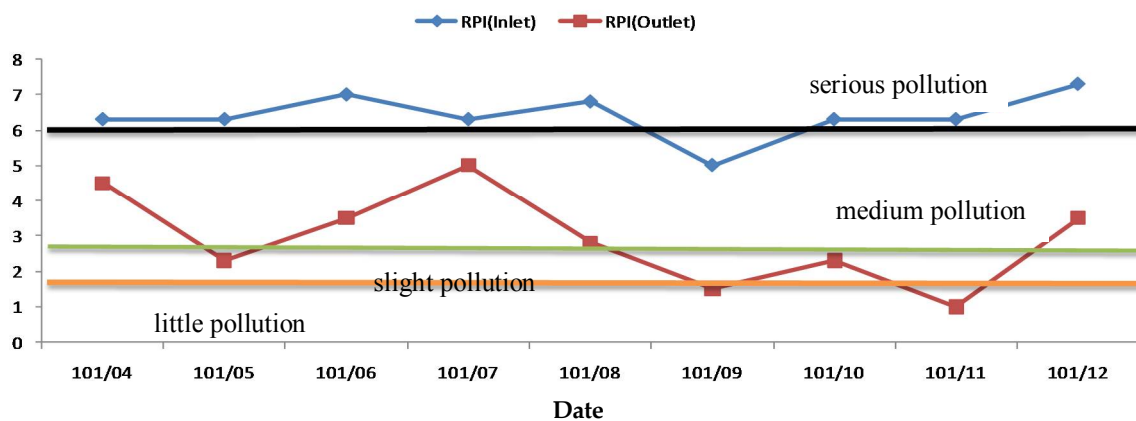
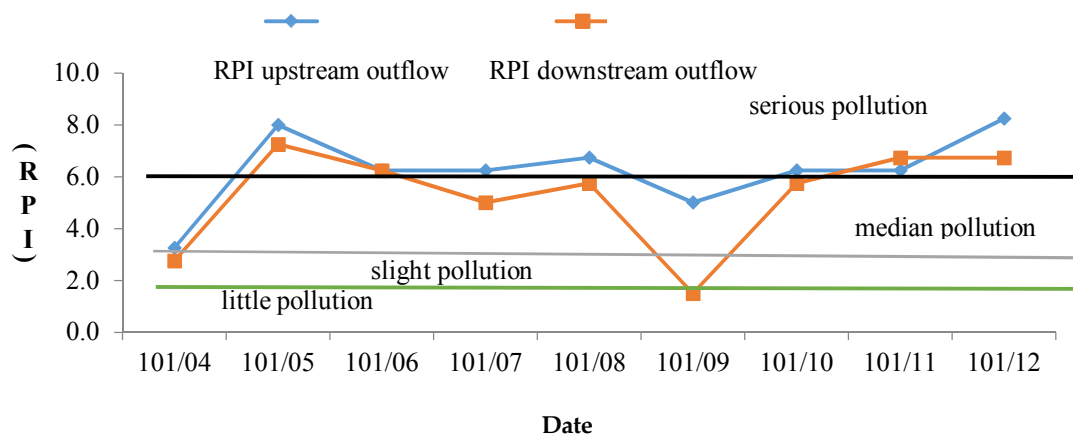
A summary of changes due to gravel contact oxidation treatment and the constructed wetland method shows that the removal efficiencies of SS, BOD, TN, TP, and NH<sub>3</sub>-N can all reach maximum values if the inflow concentration is relatively high. However, some inflow concentrations are lower, resulting in an unstable removal efficiency and unclear treatment efficiency. Improved removal efficiency occurred with the loading of high-concentration pollution. The river pollution index (RPI) is a comprehensive index for the evaluation of river quality, which is currently used by the EPA to estimate the degree of river pollution. There are four parameters that comprise the RPI: oxygen demand (DO), biochemical oxygen demand (BOD), suspended solid (SS), and ammonia nitrogen (NH<sub>3</sub>-N). The evaluation method is indicated in Table 2. The value of RPI is  $(\sum Ni)/4$ , where Ni represents the degree of different pollutions. The RPI value ranges from 1 to 10, and four degrees of water pollution are involved to differentiate the degree of pollution. This study adopted wastewater from the outflow of the Yangzi-Cuo River, Ci-Tong Jiao, as the test sample, and the water was returned to the river after gravel contact oxidation treatment and the constructed wetland method. The RPI index was used to evaluate the purification performance of wastewater. Based on the data collection regarding inflow and outflow of DO, BOD, SS, and NH<sub>3</sub>-N from 101/04–101/12, the resulting RPI indexes are presented in Table 2. It can be concluded that the water quality improved after these wastewater treatments. There are four levels pollution of RPI indexes, including little, slight, medium, and serious pollution, and the corresponding values of these pollution is located at the regions of below 2.0 mg/L, 2.0-3.0 mg/L, 3.0-6.0 mg/L, and above 6.0 mg/L. The original wastewater was classified as having serious pollution, and the post-treatment wastewater was classified as having slight pollution or medium pollution, as shown in Figure 6. The water quality in July could not reach the level of slight pollution or little pollution, because many insecticides were applied to eliminate snails in farms, resulting in residual insecticides flowing into the Yangzi-Cuo River, Ci-Tong Jiao, polluting the water. Figure 7 shows the wastewater variation in the upstream and downstream outflow. The water quality of the downstream outflow appears to have improved after wastewater treatment. Furthermore, the original inflow with an RPI index indicating serious pollution changed to an outflow with RPI indexes of slight pollution and medium pollution.

**Table 1.** Specifications on the design of the constructed wetland.

| Items                                         | Conditions                                                                                                      |
|-----------------------------------------------|-----------------------------------------------------------------------------------------------------------------|
| Designed method                               | Gravel contact oxidation and surface flow constructed wetland                                                   |
| Area                                          | a. Total base area: 9000 square meter.<br>b. Water purification treatment area: 730 square meter.               |
| Inflow source                                 | Household wastewater in Ci-Tong Jiao, Changhua County.                                                          |
| Treatment location                            | Land numbers of 1-4 and 1-5 in Ci-Tong Jiao, Changhua County.<br>Coordinates position: (23.070949, 120.505699). |
| Inflow way                                    | Gravity type and pumping motor diversions                                                                       |
| Treatment capacity                            | 2000 cubic meters per day (CMD)                                                                                 |
| Inflow concentration                          | BOD <sub>5</sub> : 50 mg/L; NH <sub>3</sub> -N: 80 mg/L; SS: 50 mg/L                                            |
| Outflow concentration                         | BOD <sub>5</sub> : 20 mg/L; NH <sub>3</sub> -N: 48 mg/L; SS: 20 mg/L                                            |
| Removal capacity                              | BOD <sub>5</sub> : 20 Kg/D; NH <sub>3</sub> -N < 5 Kg/D; SS < 20 Kg/D; TP < 5 Kg/D                              |
| Removal efficiency                            | BOD <sub>5</sub> : 60%; NH <sub>3</sub> -N: 40%; SS: 60%; TP: 60%                                               |
| HRT                                           | 2.59 days                                                                                                       |
| Expectation of purified outflow concentration | BOD <sub>5</sub> : 20 mg/L; NH <sub>3</sub> -N: 5 mg/L; SS: 20 mg/L; TP: 5 mg/L                                 |

**Table 2.** Calculation of the river pollution index (RPI) and the degree of pollution on different items.

| Item/Degree of Pollution                | Little Pollution | Slight Pollution | Medium Pollution | Serious Pollution |
|-----------------------------------------|------------------|------------------|------------------|-------------------|
| Oxygen demand (mg/L)                    | >6.5             | 3.6–6.5          | 2.0–3.5          | <2.0              |
| Biochemical oxygen demand (mg/L)        | <3.0             | 3.0–3.9          | 5.0–15           | >15               |
| Suspended solid (mg/L)                  | <20              | 20–49            | 50–100           | >100              |
| Ammonia nitrogen (mg/L)                 | <0.5             | 0.5–0.99         | 1.0–3.0          | >3.0              |
| Point on degree of different pollutions | 1                | 3                | 6                | 10                |
| RPI value                               | <2.0             | 2.0–3.0          | 3.1–6.0          | >6.0              |

**Figure 6.** RPI values of inflow and outflow.**Figure 7.** RPI values of upstream and downstream outflow.

#### 4. Conclusions

This case study focused on river quality enhancement in Changhua via cooperation with the Environmental Protection Bureau of Changhua County. This study presented the results of decreases in river pollution in the Yangzi-Cuo River, Ci-Tong Jiao, Changhua County, Taiwan, due to treatments of gravel contact oxidation and surface flow constructed wetlands. Gravel contact oxidation and surface flow constructed wetlands comprise ecology technology, aimed at using physical, chemical, and biological designs to eliminate river pollution via the mediums of soil, plant, water, and microorganisms from natural environments. Moreover, RPI reduction was applied to evaluate the efficiency of river quality enhancement. The upstream outflow belonged to serious pollution due to RPI value above 6 mg/L, and downstream outflow turned to medium and slight pollution because of RPI values of 3–6 mg/L and 2–3 mg/L, respectively. Furthermore, initial values of SS, NH<sub>3</sub>-N, TP, TN, and BOD were found to be 10.0–26.7 mg/L, 0.9–14.0 mg/L, 1.2–14.1 mg/L, 11.1–18.2 mg/L, and 0.5–3.6 mg/L, respectively, and final values of SS, NH<sub>3</sub>-N, TP, TN, and BOD were found to be 1.4–8.0 mg/L, 0.3–2.13 mg/L, 0.22–2.4 mg/L, 3.35–5.53 mg/L, and 0.5–7.4 mg/L after purification treatment by ecology technology, respectively. The efficiencies of pollution elimination on SS, NH<sub>3</sub>-N, TP, TN, and BOD were 24.2–93.1%, 58.3–86.2%, 5.2–85.0%, 59.4–77.2%, and 46.3–76.4%, respectively. Thus, this study indicates a substantial reduction in river pollution using combinations of gravel contact oxidation and surface flow constructed wetlands, and almost outlet values of those five factors do not exceed the maximum concentration limits of each pollutants' regulation, respectively. As a result, most efficiencies of pollution elimination are above 76%. Moreover, water quality of downstream outflow tends to medium and slight pollution after application of ecological engineering methods. Therefore, the study presents great results on the purification efficiency of Yangzi-Cuo River by using ecological engineering methods.

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