

1 Article

2 Nanoemulsion-Enabled Oral Delivery of Novel 3 Anticancer ω -3 Fatty Acid Derivatives

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20

21 **Abstract:** Lipid-based drugs are emerging as an interesting class of novel anticancer drugs with the
22 potential to target specific cancer cell metabolic pathway linked to their proliferation and
23 invasiveness. In particular, ω -3 polyunsaturated fatty acids (PUFA) derivatives such as epoxides
24 and their bioisosteres have demonstrated the potential to suppress growth and promote apoptosis
25 in triple-negative human breast cancer cells MDA-MB-231. In this study 16-(4'-chloro-3'-
26 trifluorophenyl)carbamoylamino]hexadecanoic acid (ClFPh-CHA), an anticancer lipid derived
27 from ω -3,17,18-epoxyeicosanoic acid, was formulated as a stable nanoemulsion with size around
28 150 nm and narrow droplet size distribution (PDI<0.200) through phase-inversion emulsification
29 process followed by high pressure homogenization in view of an oral administration. The ClFPh-
30 CHA-loaded nanoemulsions were able to significantly decrease the relative tumor volume in mice
31 bearing an intramammary tumor xenograft at all doses tested (2.5, 10 and 40 mg/kg) after 32 days
32 of daily oral administration. Furthermore, absolute tumor weight was decreased to 50% of untreated
33 control at 10 and 40 mg/kg, while intraperitoneal administration could achieve a significant
34 reduction only at the highest dose of 40 mg/kg. Results suggest that oral administration of ClFPh-
35 CHA formulated as a nanoemulsion has a sufficient bioavailability to provide an anticancer effect
36 in mice and that the activity is at least equal if not superior to that obtained by a conventional
37 parenteral administration of equivalent doses of the same drug.

38 **Keywords:** nanoemulsion; oral delivery; ω -3 polyunsaturated fatty acid derivative; MDA-MB-231;
39 triple-negative breast cancer

40 1. Introduction

41 Breast cancer is the most common type of cancer among women worldwide after non-melanoma
42 skin cancer, affecting more than 200,000 women annually and killing more than 40,000 women each
43 year. Breast cancer also affects men, but it is rare, accounting for only 1% of all cases of cancer. Quite
44 worryingly, statistics indicate an increasing incidence in both developed and developing countries.
45 There are several types of breast cancer, and their characteristics affect both treatment options and
46 therapeutic outcomes. Triple-negative breast cancer (TNBC), *i.e.* tumors that lack expression of

47 estrogen receptor (ER), progesterone receptor (PR), as well as human epidermal growth factor
48 receptor type 2 (HER2), are aggressive and highly metastatic, and do not respond to endocrine or
49 monoclonal antibodies-based therapies. Thus, patients with TNBC have limited treatment options
50 and subsequently, a poorer prognosis [1]. It is known that during the development of breast cancer
51 several regulatory mechanisms are in imbalance, while processes such as chronic inflammation and
52 pro-oxidative reactions are stimulated. The imbalance between free radicals and antioxidant species
53 induced by exogenous (diet and smoking) or endogenous (estrogens) factors promotes oxidative
54 stress during the onset, promotion and progression of breast cancer [2-4]. The development of
55 compounds with novel mechanisms of action against tumor cells are required to develop anticancer
56 therapeutics able to induce further remissions in cancer patients who are resistant to established
57 agents.

58 Many studies have recently evidenced a relationship between polyunsaturated fatty acids and
59 cancer. In fact, ω -6 polyunsaturated fatty acids (PUFAs) are converted by tumor cells to potent
60 eicosanoid promoters of tumor cell proliferation through the over-expression of cyclooxygenase,
61 lipoxygenase or cytochrome P450 enzymes. In contrast, several eicosanoid metabolites from the
62 biotransformation of ω -3 PUFAs impair particular tumorigenic pathways, indicating that the intake
63 of ω -3 polyunsaturated fatty acids might decrease cancer risk, in particular for breast, prostate and
64 colon cancer [5,6]. These anticancer effects, attributed to ω -3 PUFAs metabolites such as
65 prostaglandin E3, resolvins, 15-hydroxyeicosatetraenoic acids (HETE), epoxides of docosahexaenoic
66 (DHA), and 17,18-epoxide of eicosapentaenoic acid (ω -3,17,18-epoxyeicosanoic acid), have been the
67 trigger to develop synthetic ω -3 PUFA epoxides derivatives and to study their anti-proliferative and
68 pro-apoptotic effect in breast cancer models [7-10]. From this work, the synthetic ω -3 polyunsaturated
69 fatty acid derivative 16-(4'-chloro-3'-trifluorophenyl)carbamoylamino]hexadecanoic acid (ClFPh-
70 CHA, Figure 1) was identified as a promising anticancer agent [10].

71 However, one of the main problems with this class of compounds is their solubilization, as most
72 of them show very poor water solubility making their administration challenging. Formulation is a
73 promising approach to overcome the active compound physico-chemical issues to improve
74 bioavailability and enable their clinical use. In this direction, an interesting approach emerging as a
75 possible solution for the administration of such challenging compounds is the use of nanoemulsions
76 [11,12].

77 Nanoemulsions (NEs) are colloidal dispersions, which consist in emulsions having the dispersed
78 droplets in the nanometric scale. Droplet size falls typically between those of microemulsions and
79 conventional emulsions, with a size range of 20-500 nm. In contrast to microemulsions that are
80 thermodynamically stable, nanoemulsions are subjected the same thermodynamic instability of
81 macroemulsions. However, nanoemulsions are more stable than conventional emulsions, since their
82 small droplet size makes them less susceptible to creaming or sedimentation phenomena and
83 provides them with high kinetic stability. Another important aspect is that they can be produced
84 using more practical surfactant concentrations *e.g.* 5%, when compared to ~ 50% which is typically
85 used to prepare microemulsions [13]. Accordingly, submicron and nanoemulsions have gained
86 increasing attention as drug delivery systems for oral, parenteral, transdermal and topical (*e.g.*
87 dermo-cosmetic, vitamins and anti-aging agents) applications [14-17]. Recent progress in the control
88 of size distribution and in the understanding of stabilization mechanisms of nanoemulsion has also
89 contributed to renewed attention to these particular emulsion systems [18,19].

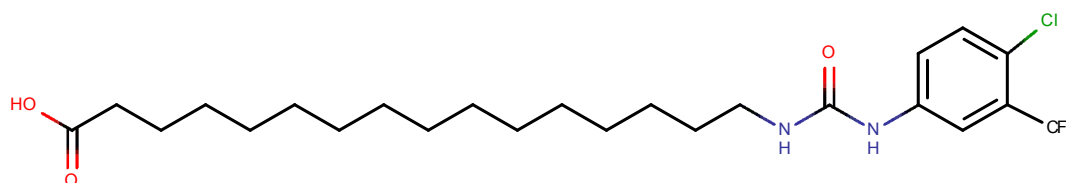
90 In particular, in the case of oral administration nanoemulsions have displayed some specific
91 advantages: 1) NEs, because of drug encapsulation, provide a platform to protect active
92 pharmaceutical substances from enzymes, low pH and other environmental conditions in the
93 gastrointestinal (GI) tract; 2) oral delivery via nanoemulsion results in rapid absorption and steady
94 state levels are reached within 30 minutes, which suggests substantial absorption even from P-
95 glycoprotein-rich distal ileal regions; 3) NEs have been shown to be able to increase the oral
96 bioavailability of lipophilic drugs (such as curcumin, some antibiotics, fatty acids, etc.); 4) NEs can
97 also minimize side effects with a reduction of dose with the same effect as a non-encapsulated drug.
98

99 The aim of the present work was to design, develop and characterize a nanoemulsion loaded
100 with the novel synthetic fatty acid ClPh-CHA that induces apoptosis in TNBC cell lines. The loaded
101 nanoemulsion was then used for oral delivery in an animal tumor model, and its anticancer activity
102 was compared to that of the same compound administered intraperitoneally.

103 2. Materials and Methods

104 2.1. Materials

105 Sorbitan monooleate (Span™ 80) (HLB= 4.3) and Polysorbate 80 (Tween™ 80) (HLB = 15.0) were
106 both supplied by Croda (Wetherill Park, NSW, Australia). The oil phase in the emulsion was
107 constituted by pharmaceutical grade medium chain triglycerides (Labrafac™ lipophile WL 1349,
108 Gatefossè, Saint-Priest, France). Ultrapure Milli-Q water filtered through 2 µm filters was used in all
109 experiments (Arium® pro - Water purification system, Sartorius, Dandenong South, VIC, Australia).
110 The ω-3 17,18-epoxyeicosanoic acid analogue [16-(4'-chloro-3'-
111 trifluorophenyl)carbamoylamino]hexadecanoic acid (ClPh-CHA, Figure 1) was synthesized
112 according to a procedure previously reported by Rawling et al. [10]. HPLC grade acetonitrile and
113 methanol for chromatography were obtained from Honeywell Burdick and Jackson (Muskegon, MI,
114 USA). Ammonium formate AR grade was obtained from Asia Pacific Specialty Chems Ltd. (Seven
115 Hills, NSW, Australia). Chemicals were used as received without any further purification.



116
117 **Figure 1.** Structure of the ω-3 polyunsaturated fatty acid derivative 16-(4'-chloro-3'-
118 trifluorophenyl)carbamoylamino]hexadecanoic acid (ClPh-CHA).

119 2.2. Methods

120 2.2.1. Study of the Impact of Process Parameters on Nanoemulsion Preparation

121 To determine the influence of different process and composition variables on the nanoemulsion
122 preparation, several conditions parameters were investigated.

123 (a) Emulsification technique

124 Two major types of emulsification were used: direct emulsification and phase-inversion
125 emulsification. In these experiments, one phase was initially placed in a stirred vessel and then the
126 second phase was gradually added to the mixing vessel. Both phases contained a constant
127 concentration (10% w/w) of the surfactant. Depending on the phase initially placed in the vessel and
128 also on the distribution of the surfactants between the phases six possible process approaches were
129 identified, three for direct and three for phase-inversion emulsification. In the direct emulsification
130 approach the dispersed phase (oil) is added to the continuous phase (water). Depending on how the
131 surfactant mixture is incorporated into the emulsion, three combinations were examined: D1. Oil
132 phase containing the low-HLB surfactant (Span 80) was gradually added to the water phase
133 containing the high-HLB surfactant (Tween 80); D2. Oil phase containing half quantity of both
134 surfactants was added to the water phase containing the rest of both surfactants; D3. Oil phase
135 containing the high-HLB surfactant (Tween 80) was gradually added to the water phase containing
136 the low-HLB surfactant (Span 80).

137 In phase-inversion emulsification, the continuous phase (water) was added to the dispersed phase
138 (oil). Therefore, the application of such a method involves a phase inversion from an initial W/O
139 emulsion to the desired (O/W) emulsion. Depending on how the surfactant mixture is incorporated
140 into the emulsion, three combinations were examined: P1. Water phase containing the high-HLB

141 surfactant (Tween 80) was gradually added to the oil phase containing the low-HLB surfactant (Span
142 80); P2. Water was added to the oil phase containing both surfactants; P3. Water phase containing the
143 low-HLB surfactant (Span 80) was gradually added to the oil phase containing the high-HLB
144 surfactant (Tween 80).

145
146 (b) Temperature

147 The emulsification was carried out by pre-heating separately the two phases at various
148 temperatures, i.e. 25 ± 2 °C, 45 ± 2 °C, 70 ± 2 °C and 85 ± 2 °C, applying the phase-inversion technique
149 P2 and keeping the surfactants concentration at 10% w/w.

150
151 (c) Surfactant concentration

152 The concentration of oil (medium chain triglycerides) and emulsion phases temperatures in all
153 emulsions were kept constant (at 10.0 w/w% and 85 ± 2 °C, respectively), while the overall surfactant
154 concentration (Span 80 + Tween 80) was varied from 8.0 to 12.0% w/w, keeping constant their relative
155 ratio to achieve an HLB of 11.0. The nanoemulsion were prepared with the phase-inversion technique
156 P2, as described above.

157
158 (d) Number of homogenization cycles

159 The previously characterized nanoemulsion prepared according to optimized parameters
160 (phase-inversion emulsification P2, 10% w/w surfactant concentration and 85°C phases temperature)
161 were then processed with a high-pressure homogenizer, where they were subjected to 1, 3, 5, 10 and
162 15 homogenization cycles at a constant pressure of 1,500 bar. The stability of these formulations was
163 evaluated at predetermined storage times (up to 90 days) for macroscopic appearance, droplet size
164 and polydispersity index (PDI).

165

166 2.2.2. Preparation of ClPh-CHA-loaded Nanoemulsion Preparation

167 In order to load the ω -3 PUFA derivative ClPh-CHA into the optimized nanoemulsion, the
168 emulsification process was performed as follows. Nanoemulsions were prepared by the phase-
169 inversion method in which the water phase was continuously added into the oil phase containing
170 both surfactants. Initially both surfactants (Span 80 and Tween 80, 3.74 and 6.26% w/w of the final
171 preparation, respectively) were dissolved in the oil phase (10% w/w). The synthetic ω -3 PUFA
172 derivative ClPh-CHA was added in oil phase at the final concentration the 10 mg/mL. This ratio of
173 surfactants was chosen to achieve an HLB value of 11.0, required HLB value necessary to obtain an
174 oil-in-water emulsion using medium chain triglycerides as oil phase. The oil and water phases were
175 heated separately at 85 ± 2 °C. Then, the water phase (80.0% w/w) was slowly added in the oil phase
176 containing the blend of surfactants. The components were mixed initially using magnetic stirring at
177 600 rpm (model RCT, IKA®, Staufen, Germany) and then processed by high-pressure homogenization
178 (Emuliflex-C5, Avestin, Ottawa, ON, Canada) until the emulsion temperature decreased to room
179 temperature (25 ± 2 °C).

180 Nanoemulsions prepared according to the determined optimal emulsification parameters were
181 characterized for stability in terms of droplet size distribution (z-average and PDI) during 3 months
182 storage.

183

184 2.2.3. Nanoemulsion Characterization

185 (a) Macroscopic Appearance

186 All prepared emulsions were evaluated macroscopically by visual examination. Emulsions were
187 carefully inspected, immediately and 24 hours after preparation, in order to observe any evident
188 macroscopic instability such as creaming or phase separation.

189

190 (b) Particle Size Distribution Analysis

191 Size distribution measurements were performed using a Zetasizer Nano ZS90 (Malvern
192 Pananalytical, Malvern, UK). This system determines particle size from 2 nm up to 3.0 µm of diameter
193 in liquid dispersions by measuring the intensity of light scattered due to the Brownian motion of the
194 particles. Before analysis, samples were diluted 1:10 using distilled water. Dynamic Light Scattering
195 was performed using laser wavelength 633 nm and 90° scattering angle at 25°C. Instrument software
196 (Zetasizer Software ver. 7.11, Malvern Pananalytical, Malvern, UK) automatically determined the
197 most appropriate measurement duration. The hydrodynamic diameter and distribution of emulsion
198 droplets were calculated by the cumulants fit and expressed as z-average and polydispersity index
199 (PDI) according to ISO recommendations (ISO 13321:1996E "Particle Size Analysis"). All samples
200 were evaluated in triplicate.

201
202 (c) Stability Test

203 Emulsions were maintained stored at room temperature (25± 2°C) for up to 90 days, monitoring
204 their macroscopic appearance. Characterization of samples in terms of preparation particle size
205 distribution (z-average and PDI) was conducted at predetermined intervals after their preparation
206 and compared with the initial values.

207
208 (d) Determination of Encapsulation Efficiency

209 The encapsulation efficiency (EE%) of compound ClPh-CHA into nanoemulsion was
210 determined indirectly by the ultrafiltration method according to Equation (1).

$$EE\% = \frac{(NE\ Total - NE\ Aqueous)}{NE\ Total} \times 100, \quad (1)$$

211 The percentage of encapsulated substance was calculated as the difference between the total
212 amount of compound in the formulation, which was measured after the dissolution of the
213 nanoemulsion in methanol (NE Total), and the amount of compound present in the aqueous phase
214 (NE Aqueous) calculated from its concentration measured in the aqueous ultrafiltrate obtained from
215 the nanoemulsion centrifugation (5 min, 2,700×g, Centrifuge 5417R, Eppendorf, Macquarie Park,
216 NSW, Australia) using centrifugal filter tubes (Ultrafree® CL, cut-off 10,000 MW, Millipore,
217 Burlington, MA, USA), divided by the total amount of compound in the nanoemulsion (NE Total)
218 multiplied by 100. Analysis of compound ClPh-CHA was performed by the validated HPLC- MS
219 method reported below.

220

221 2.2.4. HPLC/MS Analytical Method for ClPh-CHA Quantification

222 HPLC analysis was performed on an Agilent Technologies 6490 Triple Quadrupole LC/MS
223 system (Agilent Technologies, Singapore) fitted with a Waters Sunfire C18 column (2.1x100mm,
224 3.5µm particle size, Ireland). The samples were prepared by dissolving an aliquot of the compound
225 in methanol. An isocratic solvent mixture of 75% acetonitrile and 25% water was employed with a
226 flow rate of 0.5 mL/min for the first 6 minutes. A linear gradient to 100% acetonitrile over 2.5 minutes
227 was used, held in those conditions for 1 minute and finally restored to 75% acetonitrile over 1 minute
228 and held for 1.5 minutes before the next injection. The total run time was 12 minutes. Automated
229 sample injections (10 µL) were used, and the samples stored at 8°C in the autosampler. Agilent Jet
230 Stream Electrospray ionization was used in negative ionization mode with high purity nitrogen (BOC
231 gases, Sydney, NSW, Australia) as the sheath gas at 40 psi, 2900°C. The capillary voltage was 3500V
232 and the nozzle voltage was 1500V. Selected ion monitoring was used. The data was acquired using
233 Agilent MassHunter Workstation software (version B.06.00 SP1, Agilent Technologies, Santa Clara,
234 CA, USA) and interpreted with the use of Microsoft Excel 2007 (Microsoft Corp., Redmond, WA,
235 USA).

236 A calibration curve was constructed for ClPh-CHA to quantify its concentration in the
237 nanoemulsion formulation. A stock solution in methanol was prepared from an accurately weighed
238 sample of the compound and the calibration solutions were obtained by subsequent dilution of the
239 stock using methanol. Linearity was confirmed in the range 0.5 to 100 ng/ml ($R^2 = 0.9996$).

240 2.2.5. *In Vivo* Studies

241 Female Balb/c nu/nu mice (6 or 7 weeks of age) were obtained from Animal Resources Centre
242 (Perth, WA, Australia). Mice were housed in sterile cages in a temperature-controlled animal house
243 at the University of Sydney in accordance with the University Animal Welfare guidelines and an
244 approved animal ethics protocol (University of Sydney Animal Ethics Committee, Approval L24/2-
245 2012/3/5680). Mice were acclimatized and monitored in the animal house for one week prior to
246 treatments. All mouse manipulations were carried out under aseptic conditions in a biosafety laminar
247 flow hood.

248 Human breast adenocarcinoma MDA-MB-231 cells (ATCC, Manassas, VA, USA) cultivated at
249 37°C in a humidified atmosphere of 5% CO₂ in air in DMEM supplemented with 10% fetal bovine
250 serum (Thermo-Fisher Scientific, Scoresby, VIC, Australia) and 1% penicillin/streptomycin
251 (Invitrogen, Carlsbad, CA, USA) were harvested using Trypsin/EDTA during exponential growth
252 and washed twice in ice-cold PBS (pH 7.4). Cells were resuspended in ice-cold PBS combined 1:1 with
253 Matrigel (BD Bioscience, Franklin Lakes, NJ, USA) and injected into the left inguinal mammary gland
254 (4x10⁴ cells/0.1 mL). Mice were randomly assigned to either the control (CTL, n=5-8) or treatment
255 groups (n=5-8). Treatment with the ω-3 17,18-epoxyeicosanoic acid isostere CIFPh-CHA by
256 intraperitoneal (IP) or oral administration began 3-4 days after tumor cell inoculation. For
257 intraperitoneal administration CIFPh-CHA was injected in mice at doses of 2.5, 10 or 40 mg/kg after
258 dissolution in corn oil containing 8% DMSO (50 μL volume) once-a-day (6 days per week) for 38 days.
259 In the case of oral administration, CIFPh-CHA-loaded nanoemulsion was administered to
260 xenografted mice by oral gavage using 20G-38 stainless steel gavage needles at doses 2.5, 10 or 40
261 mg/kg (80 μL volume) once-a-day (6 days per week) and for 32 days. A control group of animals
262 receiving vehicle only were included for each experiment.

263 Mouse body weights were recorded 6 days a week. Commencing eight days after tumor cell
264 inoculation, tumor volumes were monitored by measuring the major longitudinal diameter (length,
265 L) and the major transverse diameter (width, W) of the tumor mass using a pair of calipers every 3-4
266 days. The tumor volume was calculated according to Equation (2), as described previously [20].

$$V = \frac{(L \times W^2)}{2}, \quad (2)$$

267 On the final day of CIFPh-CHA administration, following euthanasia with CO₂, all primary
268 tumors were harvested and weighed.

269 2.2.6. Statistics

270 All data are presented as mean ± standard deviation of at least three independent measures
271 (n≥3), if not otherwise stated. ANOVA and Fishers LSD *post-hoc* test was used to identify significant
272 differences between control and CIFPh-CHA treatment groups.

273 3. Results

274 3.1. Optimization of Nanoemulsion Production Process

275 The initial screening of the two major emulsification techniques evidenced that only phase-
276 inversion emulsification technique allowed for the production of stable nanoemulsions. Previous
277 studies already evidenced that the PIT method is suitable to obtain O/A microemulsions (MEs) or
278 nanoemulsions (NEs) with good technological properties using low percentages of non-ionic
279 surfactants [21,22]. In this specific case, of the six compositions and phase combinations investigated,
280 only the phase inversion emulsification process P2, in which the water phase was added to the oily
281 phase containing both surfactants, provided a stable nanoemulsion with an average droplet size and
282 PDI over 3 months of 130.9 ± 19.8 nm and 0.146 ± 0.02, respectively. In all other cases (D1, D2, D3, P1
283 and P3) phase separation occurred as soon as 24 hours after preparation.

284 Once evidenced that only one emulsification process provided a stable nanoemulsion, the
285 temperature of emulsification of the two phases was investigated, as temperature affects several

286 aspects of the emulsification process, such as solubility of surfactants, viscosity of the phases and
 287 interfacial tension [23,24]. The particle size and stability of emulsions obtained at five different
 288 temperatures are presented in Table 1.

289 **Table 1.** Influence of emulsification temperature on droplets size distribution and stability.

Time	1 day		30 days		60 days		90 days	
T (°C)	Size (nm)	PDI	Size (nm)	PDI	Size (nm)	PDI	Size (nm)	PDI
25	Creaming		-	-	-	-	-	-
45	101.5±1.1	0.17±0.02	125.9±2.4	0.20±0.00	113.8±1.2	0.23±0.01	104.5±0.4	0.17±0.01
70	133.1±8.0	0.29±0.04	135.4±8.3	0.29±0.01	243.4±29.7	0.28±0.08	554.5±11.2	0.44±0.01
85	137.1±3.0	0.16±0.01	144.6±3.1	0.16±0.01	136.9±0.8	0.13±0.02	150.5±0.3	0.24±0.01

290 The nanoemulsion could not be prepared at 25°C as creaming of the formulation was
 291 immediately evident after 24 hours. Nanometric size emulsions could be obtained for temperatures
 292 from 45 to 85°C with sizes slightly above 100 nm and relatively narrow distribution (PDI<0.30).
 293 Storage did not affect the properties of the nanoemulsion for up to 30 days. However, after 60 and 90
 294 days a significant and progressive increase in droplet particle size and in polydispersity was
 295 evidenced for the nanoemulsion prepared at 70°C, suggesting an instability due to a progressive
 296 coalescence of dispersed phase droplets. As a consequence, even though no apparent instability was
 297 evident for the preparation obtained at 45°C, it was considered that the most robust formulation
 298 could be obtained by performing the emulsification process at 85°C.

299 Subsequently, using the phase inversion emulsification procedure P2 at 85°C, the effect of the
 300 total surfactant concentration on the characteristics of the nanoemulsions was evaluated. Table 2
 301 presents these results in terms of nanoemulsion droplet size, PDI and stability over 3 months.

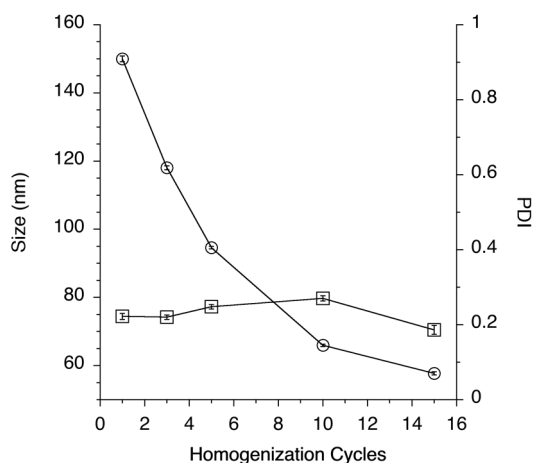
302 **Table 2.** Influence of the total concentration of surfactants on droplet size distribution and stability.

Time	1 day		30 days		60 days		90 days	
Surf. (%) ¹	Size (nm)	PDI	Size (nm)	PDI	Size (nm)	PDI	Size (nm)	PDI
8	162.4±7.7	0.28±0.02	Coalescence		-	-	-	-
9	150.0±1.1	0.20±0.02	156.5±3.2	0.15±0.02	164.4±3.2	0.25±0.04	151.1±1.8	0.19±0.01
10	137.1±3.0	0.16±0.01	144.6±3.1	0.16±0.01	136.9±0.8	0.13±0.02	150.5±0.3	0.24±0.01
11	120.3±2.3	0.10±0.03	130.7±6.2	0.14±0.02	125.0±1.3	0.13±0.02	122.53±1.1	0.17±0.01
12	122.3±1.5	0.10±0.02	129.0±3.3	0.12±0.02	126.4±1.1	0.18±0.01	124.9±0.5	0.17±0.01

303 ¹Overall percentage by weight of Span 80 and Tween 80

304 Data evidenced that with the exception of a surfactant concentration of 8% w/w, in all other cases
 305 it was possible to obtain nanoemulsions with small particle size that were stable for at least 90 days.
 306 The formulation containing an overall surfactant concentration of 10% w/w was selected as the
 307 minimum required to provide a particle size below 150 nm and PDI below 0.2 that was stable over
 308 90 days, suggesting a narrow droplet distribution maintained for all the three months of the stability
 309 study.

310 Finally, the effect of the number of cycles through the high-pressure homogenizer on particle
 311 size distribution was evaluated (Figure 2).



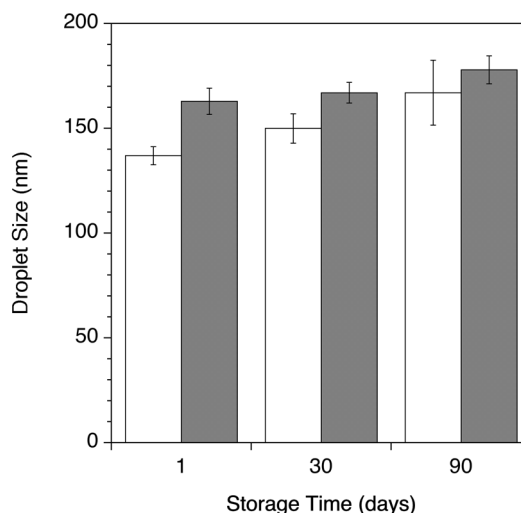
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313 **Figure 2.** Effect of the number of homogenization cycles on nanoemulsion droplet size (empty circles) and
314 PDI (empty squares).

315 Figure 2 shows that increasing the number of homogenization cycles progressively decreased
316 particle size to a value slightly lower than 60 nm. At the same time PDI of nanoemulsion processed
317 with HPH at 1,500 bar was kept fairly low with values ranging between 0.186 and 0.270. The
318 formulation processed with HPH were also found to be stable for at least 3 months at room
319 temperature, with no relevant trend evidenced for average droplet size and PDI (data not shown).
320 However, due to the unknown chemical stability of ClFPh-CHA to the extreme conditions of the
321 high-pressure homogenization process it was decided to avoid prolonged homogenization times and
322 to limit the number of cycles to 3.

323 3.2. ClFPh-CHA-loaded Nanoemulsion

324 As a result of the investigation of the parameters affecting the nanoemulsion preparation
325 process, O/W nano-sized emulsions were prepared with and without ClFPh-CHA at 85°C, with an
326 overall concentration of surfactants of 10% w/w providing a resultant HLB of 11 and concluding the
327 process P2 followed by 3 passages the high-pressure homogenizer. Compound ClFPh-CHA when
328 added to the formulation, was dissolved in the oil phase to provide a final concentration of 10mg/mL.
329 This concentration was selected to allow the ClFPh-CHA loaded nanoemulsion to be delivered to
330 mice in volumes of 100µL or less. Encapsulation efficiency of $99.9 \pm 2.3\%$ suggests a complete drug
331 encapsulation into the oil dispersed phase, as expected for a highly lipophilic compound.
332 Nanoemulsions with and without compound showed a milky bluish aspect with low viscosity. Nano-
333 sized emulsions showed no macroscopic instability phenomena, such as creaming, phase separation,
334 or drug precipitation during the storage period of 90 days at 25°C. Average droplet size during the
335 stability testing is shown in Figure 3.



336

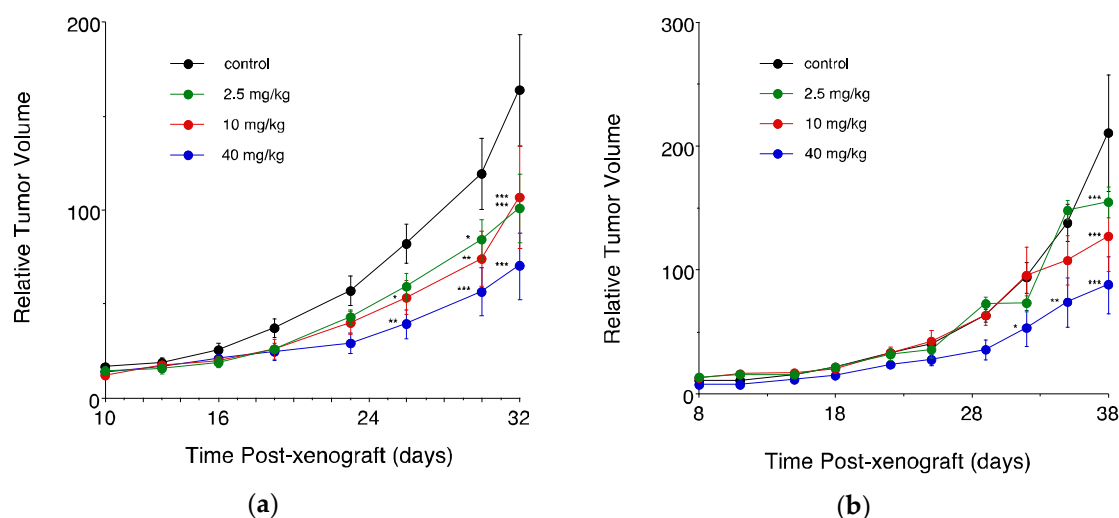
337 **Figure 3.** Nanoemulsion droplet size evaluated during storage for CIFPh-CHA-loaded (gray bars) and
 338 blank nanoemulsions (white bars).

339 The average droplet size of the drug-loaded nanoemulsion was greater than the blank
 340 nanoemulsion at all time points. Furthermore, droplet size did show a slight trend to increase for
 341 both drug-loaded and blank emulsion. Nevertheless, PDIs below 0.2 measured at all time during the
 342 storage time help to exclude critical phenomena of instability during storage (data not shown).

343 3.3. *In vivo* studies in mice bearing an intramammary tumor xenograft

344 The anticancer efficacy of CIFPh-CHA was assessed in xenografted nu/nu Balb/c mice (Figure
 345 4). As shown in Figure 4a, a significant dose-dependent decrease in tumor volume after
 346 administration of the nanoemulsion by oral gavage was evident starting from 26 days of daily
 347 treatment post-xenograft. After 32 days significant results were obtained for all the doses tested, with
 348 the lower doses (2.5 and 10 mg/kg) reducing tumor volume to ~60% of control, and the higher dosage
 349 tested (40 mg/kg) reducing volume to 42% of control. Importantly, control and drug treated mice
 350 gained body weight throughout the study and no deaths were recorded, which indicates that CIFPh-
 351 CHA and the nanoemulsion were well tolerated.

352

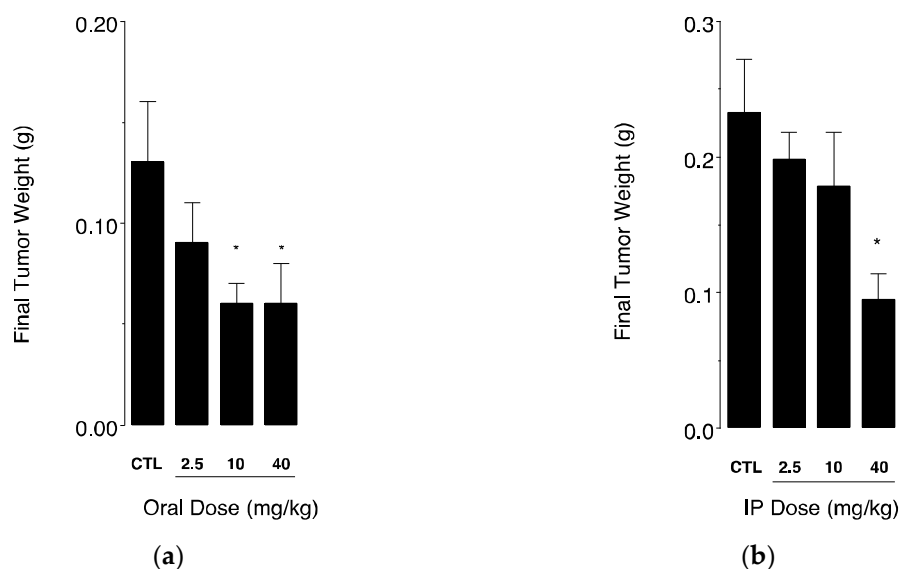


353 **Figure 4.** Dose-dependent effects of CIFPh-CHA on the *in vivo* growth of MDA-MB-231 cell xenografts.
 354 CIFPh-CHA was delivered orally as a nanoemulsion (a) or by intraperitoneal injection as a solution (b).

355 Control groups in each experiment received vehicle only. Different from control: ***P<0.001, **P<0.01,
356 *P<0.05. (Panel (b) reprinted with permission from [10]. Copyright (YEAR) American Chemical Society).

357 As shown by relative tumor volume obtained for the IP administration of the drug solution
358 (Figure 4b), a significant difference from the untreated control could be obtained only starting from
359 day 32 and exclusively for the higher dosage. A significant tumor reduction was obtained for all the
360 doses tested only by day 38 post-xenograft.

361 When considering the final weights of excised tumors (Figure 5), oral administration of the drug-
362 loaded nanoemulsion was able to significantly reduce tumor mass to ~50% of untreated control at
363 doses of 10 and 40 mg/kg (Figure 5a).



364 **Figure 5.** Final weights of excised tumours were determined at necropsy for (a) mice receiving
365 increasing doses of ClFPh-CHA-loaded nanoemulsion orally and (b) mice receiving increasing doses
366 of ClFPh-CHA in solution by intraperitoneal injection (32 days and 38 days of daily treatment for oral
367 and IP administration, respectively). Values were compared to tumor weights of control groups of
368 mice that received vehicle only (CTL). Different from control: *P<0.05.

369 Interestingly, the same compound administered intraperitoneally produced a significant
370 reduction of tumor mass to ~40% of untreated control only at the highest dose (Figure 5b), while the
371 lower doses of 2.5 and 10 mg/kg failed to produce significant reductions in tumor mass. In addition,
372 this result was obtained over a longer treatment period (38 days IP injection vs. 32 days oral gavage).
373 Combined, these results suggest that oral delivery of ClFPh-CHA in the nanoemulsion is superior to
374 IP delivery as a solution.

375 4. Discussion

376 Lipids and their metabolites have been identified as key mediators of cell growth and death
377 pathways, in healthy as well as in cancer cells. Due to this role, lipid-based drug discovery in cancer
378 research has the aim to find new targets and develop specific and tolerable drugs able to address the
379 proliferation, apoptosis and angiogenesis dysregulations at the base of aggressive characteristics of
380 tumor cells [25-27]. We have developed a drug discovery approach based on the consideration that
381 eicosanoids metabolites of ω -6 PUFA arachidonic acid promote tumorigenesis, while ω -3 PUFA and
382 certain metabolites have tumor suppressive actions. Indeed, long chain ω -3 monounsaturated fatty
383 acids were demonstrated to effectively decrease proliferation and invasiveness of breast cancer cells
384 over-expressing COX-2, possibly via a marked decrease of ω -6 PUFA-derived eicosanoid
385 prostaglandin E₂ [7]. In another study, the ω -3 17,18-epoxide of eicosapentaenoic acid, but not its
386 regioisomers, selectively showed promising anticancer properties by inducing cell cycle arrest and

387 growth suppression through p38 MAPK activation and down-regulation of cyclin D1 [7,9]. Hence, a
388 series of synthetic analogues of ω -3,17,18-epoxyeicosanoic acid were designed as potential drug
389 candidates with antiproliferative and pro-apoptotic properties [8]. Among these series of analogues,
390 ClFPh-CHA, an arylurea analogue of 17,18-epoxy-EPA, emerged as a inhibitor of breast cancer
391 proliferation *in vitro* and *in vivo* after intraperitoneal administration [10]. Even though parenteral
392 administration is still the main approach for cancer therapy, a major trend in recent years has been to
393 develop formulations to allow for oral delivery of anticancer agents in order to switch to a therapy
394 preferred by patients and providing improved quality of life. Furthermore, oral therapy is
395 economically more convenient, avoiding the need for hospitalization, and in many studies showed
396 less severe side effects, with patients feeling less sick and more capable to attend daily activities [28].
397 However, the oral delivery of many anticancer drugs presents a number of challenges because of
398 inadequate aqueous solubility, chemical and/or enzymatic degradation in gastrointestinal fluids,
399 extensive biotransformation in the intestinal epithelium and in the liver, and poor absorption due to
400 poor intestinal permeability or efflux phenomena [29]. ClFPh-CHA has an extremely low aqueous
401 solubility (~150 ng/mL) and as a consequence is either a class II (low solubility, high permeability) or,
402 more likely, a class IV (low solubility, low permeability) substance according to the
403 Biopharmaceutical Classification System (BCS); this predicts dissolution rate-limited intestinal
404 absorption and variable absorption from the GI tract [30,31]. ClFPh-CHA required a formulation that
405 enabled oral delivery and to assure its bioavailability via this administration route. Among possible
406 options, nano-sized drug delivery systems encapsulating lipophilic drugs have been demonstrated
407 to be able to protect the drug from GI degradation, enhance drug absorption through the intestinal
408 barriers and ultimately enhance bioavailability by modifying the pharmacokinetic profile of the
409 encapsulated drug [32]. In particular lipid-based nanomedicines such as liposomes [33], solid lipid
410 nanoparticles [34] and polymer/lipid hybrid systems [35,36] have shown potential through a number
411 of mechanisms ranging from mucoadhesion and apparent drug dissolution in GI tract fluids, to the
412 modification of enterocyte-based transport and metabolism and even providing selective lymphatic
413 uptake of the drug [37].

414 In the present work a straightforward approach was selected with the development of a
415 nanoemulsion able to encapsulate ClFPh-CHA and presenting particle size below 200 nm and narrow
416 particle size distribution. The formation of nano-sized o/w emulsions was attributed to the kinetics
417 of emulsification during the catastrophic phase-inversion process and the consequent rearrangement
418 of the interfacial surfactant molecules resulting in curvature change and fine dispersion of the
419 dispersed phase. It was decided use two different surfactants (Tween 80 and Span 80), because it has
420 been suggested that a single surfactant is unlikely to produce the desired stability [23]. The
421 combination of the composition of the initial phases, temperature and surfactant concentration
422 produced a stable nanoemulsion suitable for loading with a relatively high drug concentration. High
423 pressure homogenization was selected as a convenient scale-up method to further refine the
424 nanoparticle droplet size distribution, since this could have an impact on GI tract lipolysis and the
425 enhancement of bioavailability [38].

426 More significantly, when administered orally every day to mice bearing breast cancer
427 xenografts, the nanoemulsion showed a significant reduction in tumor volume after 26 days for the
428 two higher doses tested (10 and 40 mg/kg) and after 30 days also for the lowest (2.5 mg/kg) when
429 compared to tumors developed in untreated mice. This preliminary *in vivo* finding demonstrated that
430 the drug, already shown to be an effective pro-apoptotic compound after intraperitoneal injection, is
431 absorbed when administered orally as a nanoemulsion and presumably achieves blood
432 concentrations sufficient to decrease MDA-MB-231 cells proliferation. Furthermore, these cells are
433 human triple-negative breast cancers cells, *i.e.* not expressing the molecular targets used in the
434 treatment of other breast cancers. Triple-negative cancers are more difficult to treat and their
435 prognosis is extremely poor [1]. Further studies should determine the absolute bioavailability and
436 the mechanism of absorption of ClFPh-CHA. However, considering the highly lipophilic nature of
437 the compound it is possible that its inclusion in the nanoemulsion favored its lymphatic uptake [39].
438 In fact, other lipid-based nanocarriers such as liposomes and solid lipid nanoparticles have been

439 shown to be suitable carriers for lymphatic delivery [40]. The drug, absorbed through this physiologic
440 intestinal lipid transport system, would reach the systemic circulation through lymphatic vessels and
441 draining lymph nodes, bypassing the liver and first-pass metabolism, ultimately enhancing its oral
442 bioavailability.

443 Interestingly, when the tumor weight was taken into account, ClFPh-CHA-loaded
444 nanoemulsion administered orally produced a significant reduction of tumor mass at a dose of 10
445 mg/kg, while a dose of 40 mg/kg was required for a similar reduction in tumor mass when
446 administered intraperitoneally. This finding suggests that the efficacy of the drug could be greater
447 after oral administration. One explanation could be that the bioavailability is suboptimal after
448 intraperitoneal injection since the injection of the corn oil/DMSO solution could lead to drug
449 precipitation when in contact with peritoneal fluids. In fact, an abdominal precipitate was evident in
450 ClFPh-CHA-treated rats at necropsy after administration of the 40 mg/kg dose. Drug precipitation
451 would lead to delayed absorption due to slow drug crystals dissolution and low drug concentrations
452 at tumor site.
453

454 5. Conclusions

455 Nanotechnologies have been often applied the administration of existing anticancer drugs, with
456 the aim to modify their pharmacokinetics, improve their efficacy and reduce drug-related adverse
457 effects. This reformulation strategy often improves the clinical utility of old drugs offering important
458 benefits such as reduction of critical toxicities [41]. In this work, pharmaceutical nanotechnologies
459 were applied to the formulation of a promising lipid-based anticancer drugs designed to provide a
460 valid alternative to existing treatment to aggressive forms of breast cancer. A nanoemulsion
461 formulation approach was selected to enable the innovative anticancer drug oral delivery, despite its
462 limited aqueous solubility. *In vivo* preliminary results indicate that ω -3 17,18-epoxyeicosanoic acid
463 bioisosteres can be formulated as oil-in-water nanoemulsion providing sufficient absorption and
464 bioavailability to hinder tumor proliferation to an extent matching, if not exceeding, the anticancer
465 activity obtained via a conventional parenteral administration.

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