

1 Article

2 Impact of Mechanical Degradation on Polymer 3 Injectivity in Porous Media

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9

10 **Abstract:** Polymer flooding is an established enhanced oil recovery (EOR) method, still many
11 aspects of polymer flooding are not well understood. This study investigates the influence of
12 mechanical degradation on flow properties of polymers in porous media. Mechanical degradation
13 due to high shear forces may occur in the injection well and at the entrance to the porous media.
14 The polymers that give high viscosity yields at a sustainable economic cost are typically large, MW
15 > 10 MDa, and have wide molecular weight distributions. Both MW and the distributions are
16 altered by mechanical degradation, leading to changes in the flow rheology of the polymer. The
17 polymer solutions were subjected to different degrees of pre-shearing and pre-filtering before
18 injected into Bentheimer outcrop sandstone cores. Rheology studies of injected and produced
19 polymer solutions were performed and interpreted together with in-situ rheology data. The core
20 floods showed a predominant shear thickening behavior at high flow velocities which is due to
21 successive contraction/expansion flow in pores. When pre-sheared, shear thickening was reduced
22 but with no significant reduction in in-situ viscosity at lower flow rates. This may be explained by
23 reduction in the extensional viscosity. Furthermore, the results show that successive degradation
24 occurred which suggests that the assumption of the highest point of shear which determines
25 mechanical degradation in a porous media does not hold for all field relevant conditions.

26 **Keywords:** HPAM Polymer; Rheology; Viscosity; Injectivity; Mechanical Degradation; Polymer
27 Flooding
28

29 1. Introduction

30 Polymer flooding has been widely implemented as a mobility control technique in secondary or
31 tertiary enhanced oil recovery (EOR) [1]. It is used to reduce the viscosity contrast between injected
32 water and displaced oil [2,3]. Commonly, two types of polymers have been utilized in EOR
33 applications, synthetic polymers primarily partially hydrolyzed polyacrylamide (HPAM) and
34 biopolymers mainly xanthan. Both polymers are used as viscosifying agents [4]. Wang, *et al.* [5]
35 reported the success of polymer flooding in increasing oil recovery factor of Daqing oil field in
36 China. Among many other successful field polymer flooding projects reported by Standnes and
37 Skjevrak [6], HPAM is the most commonly used EOR polymer.

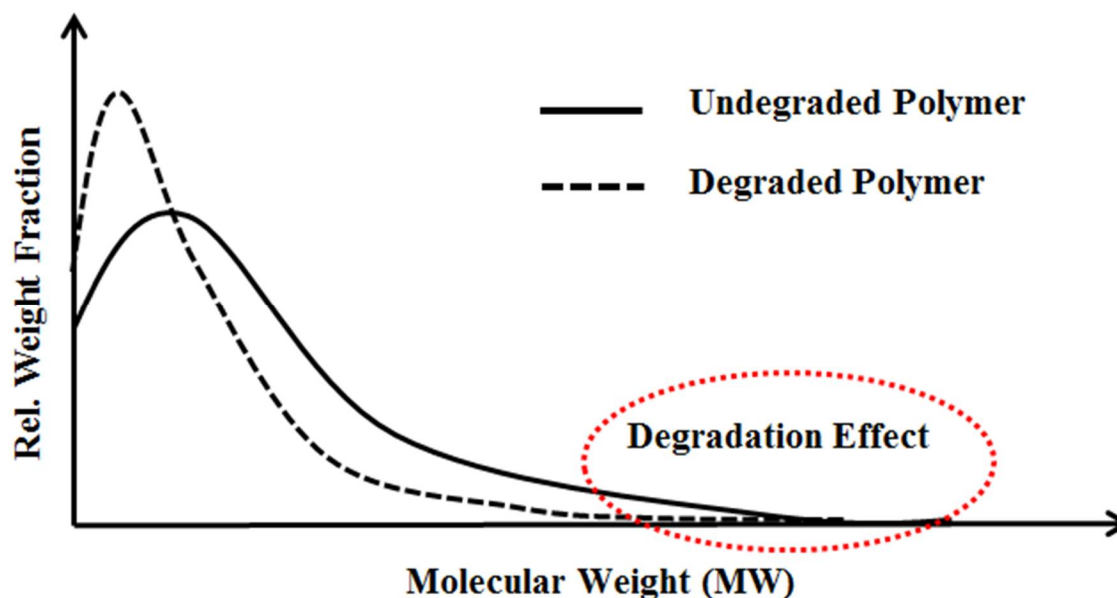
38 HPAM transported in oil reservoirs will experience different flow velocities due to high flow
39 rates at the injector and also due to local pore size variations. This results in expansion and
40 contraction of polymer flow inside the porous media. It would accordingly exhibit different flow
41 regimes with respect to shear rates. These different flow regimes have been widely discussed earlier
42 e.g., by Chauveteau [7], Southwick and Manke [8], Stavland, *et al.* [9], Zamani, *et al.* [10], and Skauge, *et al.* [11]. Due to its viscoelastic nature, its in-situ viscosity is a contribution of viscous and elastic
43 properties e.g., shear and extensional viscosity, respectively [12].

44 In a pure shear flow such as the flow in viscometer, HPAM exhibits shear thinning behavior
45 that can be described by a power law equation [13]. Shear thinning behavior is an ideal injectivity
46

47 characteristic of EOR polymers, where viscosity decreases with the increase in shear rate. High flow
48 velocities are inherent in wellbore area[14]. They cause an increase in polymer apparent viscosity in
49 porous medium (shear thickening) in contrast to dominant bulk thinning behavior measured in a
50 viscometer. In porous medium, at low flow velocities, shear viscosity is dominant while the flow is
51 dominated by extensional viscosity at high velocities [15]. During the extensional flow, polymer coils
52 experience high extensional stresses that induce flow resistance which gives the substantial rise on
53 apparent viscosity [16,17]. This theoretically interpreted by coil transition theory [7]. Although, this
54 theory has been debated later by transient network theory which explains the origin of shear
55 thickening regarding disentanglement timescale [18,19]. Regardless the theoretical interpretations of
56 shear thickening, it has been experimentally observed even at very low concentrations of HPAM, see
57 e.g., [20-23]. It increases linearly with the flow velocity after the onset of shear thickening [24]. The
58 onset of shear thickening has been given high attention in literature and even more than the
59 magnitude of shear thickening effect on viscosity [22]. The onset of shear thickening is a function of
60 many parameters such polymer molecular weight, concentration, degree of hydrolysis, salinity,
61 temperature and rock permeability [9,15,23,25].

62 In conjunction with shear thickening at high flow velocities discussed above, HPAM solution is
63 also prone to mechanical degradation [3,26]. Mechanical degradation of polymer can be described as
64 an irreversible process that leads to the breakage of polymer molecules due to high mechanical
65 stresses induced by high flow velocities or elongational deformations [16,26-28]. The breakage of
66 polymer chain induces a significant loss on polymer viscosity. Consequently, it reduces its
67 displacement efficiency and propagation within the reservoirs [28]. Mechanical degradation is a
68 function of flow velocity, pore geometry, pore tortuosity, polymer-fluid and polymer-rock
69 interactions and physicochemical properties of the polymer. It would be high for high flow rate,
70 high molecular weight polymer, high brine salinity and low formation permeability [26,29].
71 Mechanical degradation occurs at the entry point of the sand face and therefore is independent of
72 path length [7,28,30,31].

73 HPAM mechanical degradation could be minimized if the polymer is submitted for a certain
74 amount of mechanical degradation prior injection into reservoir [2,3,16]. This approach is
75 well-known as polymer pre-shearing process. This was discussed by Seright, *et al.* [32] in which
76 mechanical degradation of HPAM solution occurs at the high end of polymer molecular weight
77 distribution (MWD) as shown in Figure. 1. This is because of high molecular weight molecules have
78 large size which could offer more resistance to flow. Therefore, large elongational stresses causing
79 breakdown of molecules resulting in degradation. During preshearing process, high molecular
80 weight species will breakdown into some combination of lower molecular weight fragments leading
81 to a new MWD. The new MWD of degraded polymer translated into lower MWD. Hence, HPAM
82 viscoelastic properties that depend on high molecular weight species are more affected compared to
83 shear viscosity that depends on average molecular weight which relatively less altered by
84 preshearing process [26]. Moreover, preshearing process results in better filterability[33] by
85 removing polymer aggregates or micro gel that responsible for pore blockage. This eliminates the
86 high apparent resistance factors that may appear at low flow rates cause injectivity issues [29]. Chain
87 scission mechanisms associated with polymer degradation or shearing were extensively discussed
88 by Odell, *et al.* [34] and Muller, *et al.* [35]. The amount of pre-shearing should be optimized to avoid
89 the loss of polymer viscosity and improve its injectivity characteristics [36].



90
 91 **Figure. 1.** Schematic illustration of the effect of mechanical degradation on polymer molecular
 92 weight distribution (MWD) based on the observations reported by Seright, Maerker and Holzwarth
 93 [32]. The peak of degraded solution shift to lower molecular weight. MWD was reduced for
 94 degraded solution.

95 To represent the flow of polymer near wellbore area and as it advances within the reservoir
 96 using linear cores is challenging due to different states of velocity regimes. For example, unsteady
 97 state flow condition presents at wellbore area while a steady state condition applied in the lab for
 98 core flood studies. The experimental design of core flood has to consider filtration and degradation
 99 effects on polymer as reported by some earlier studies, e.g., Martin [37],[38]. The effect of prefiltering
 100 process in which the polymer subjected to low flow rate before injection has insignificant impact on
 101 polymer in-situ viscosity [39]. This is also observed on the experimental study performed by Skauge,
 102 Kvilhaug and Skauge [23]. They investigated the role of polymer phase behavior on in-situ viscosity.
 103 In which they found the molecular weight of polymer is the key factor dominating in-situ rheology
 104 of semi-dilute polymers. Jouenne, *et al.* [40] performed degradation studies using a blender,
 105 capillaries and porous media. The kinetics of degradation fit a master curve as a function of
 106 normalized time, regardless of the media in which the degradation occurred. Moreover, the polymer
 107 will not be further degraded beyond the point of highest strain which determined the steady state
 108 value of degradation. Until the steady state was reached, degradation increased with the number of
 109 passes of the point of highest strain. This was reached at less than 6 mm in synthetic porous media.

110 The present study aims to incorporate more realistic investigation of HPAM rheology. It
 111 experimentally investigates the effect of mechanical degradation on HPAM in-situ rheology in linear
 112 core floods combined with polymer rheological analysis. The purpose was to investigate how
 113 different polymer pretreatment methods (e.g., prefiltering, preshearing and reinjecting of polymer)
 114 influence the injectivity and polymer degradation.

115 2. Materials and Methods

116 2.1. Polymer Preparation

117 Hydrolyzed polyacrylamide (HPAM) polymers with 30% degree of hydrolysis have been used
 118 for this study with different molecular weights and concentrations as shown in Table 1. It is
 119 well-known that HPAM polymers have a broad spectrum of molecular weight distribution (e.g.,
 120 Polydispersity Index > 1)[3,41]. These polymers in average contain 90% active material as reported
 121 from the supplier. They were obtained as white granular powders from SNF Floerger, France. The
 122 polymer solutions were prepared by dissolving these polymers into 1 wt.% NaCl brine. Initially,

123 approximately 5000 ppm stock solution has been carefully prepared and then diluted into required
 124 concentrations. The preparation of stock solution was achieved by adding 3.0 g of polymer powder
 125 slowly into the shoulder of the vortex of 540.0 g brine while maintaining vigorous stirring using
 126 magnetic stirrer till the vortex disappear. Then, the stirring speed was decreased to 150 rpm and the
 127 polymer solution left under slow mixing overnight. The prepared solution was kept at 5 °C inside a
 128 fridge and was used within two weeks of preparation.

129 **Table 1.** Physicochemical properties of polymers

Solution	Polymer (Flopaam) Type	Molecular Weight (10 ⁶ g/mol = MDa)	Polymer Concentration (mg/l = ppm)	Viscosity (cP = mPa.s)
A	3630s	18	500	7.45
B	3430s	12	1000	13.54

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 131 Polymer phase behavior for both solutions A and B requires determination of critical overlap
 132 concentration (C*). For example, C* could be used to determine the solution conformational state
 133 whether it is a dilute, semi-dilute or concentrated solution. According to the values determined by
 134 Skauge, Kvilhaug and Skauge [23], both polymers solutions A and B are within semi-dilute region
 135 with different viscoelasticity.

136 2.2. Polymer Characterization

137 Shear viscosity of polymer solutions was measured at 22 °C by using Malvern Kinexus Pro
 138 Rheometer. This rheometer is equipped with different geometries. Cone-plate geometry (CP 2/50)
 139 was used for solutions with viscosity greater than 10 cP and double gap geometry (DG 25) was used
 140 for solutions with viscosity less than 10 cP.

141 2.3. Brine

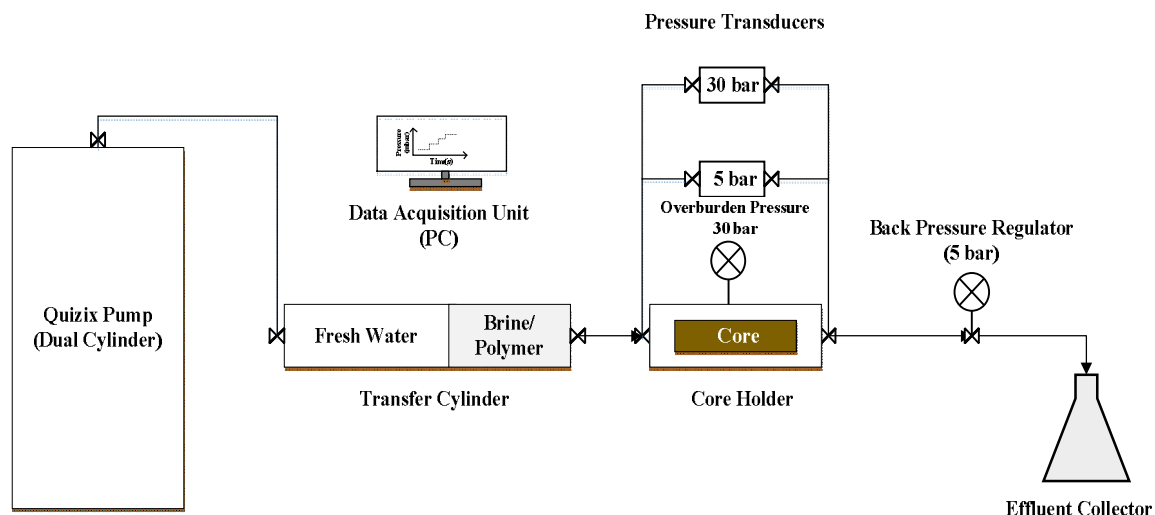
142 Brine solution containing 1 wt.% NaCl was prepared by dissolving NaCl powder obtained from
 143 Sigma-Aldrich into deionized water. Then the solution was filtered by using a 0.45 µm cellulose
 144 nitrate filter. The filtered brine was used in the preparation of polymer solutions and measurement
 145 of core petrophysical properties (e.g., porosity and permeability).

146 2.4. Porous Medium

147 Cylindrical cores of Bentheimer sandstone were used as porous medium with an average
 148 length and diameter of 10 and 3.8 cm, respectively. Also, short cores with 5 cm length were used for
 149 prefiltering and preshearing processes. Bentheimer sandstone is considered to be homogenous since
 150 it mainly contains quartz mineral [42]. The average core porosity and permeability were found to be
 151 23 % and 2.5 Darcy, respectively. More details for each core are given within the results.

152 2.5. Experimental Procedures

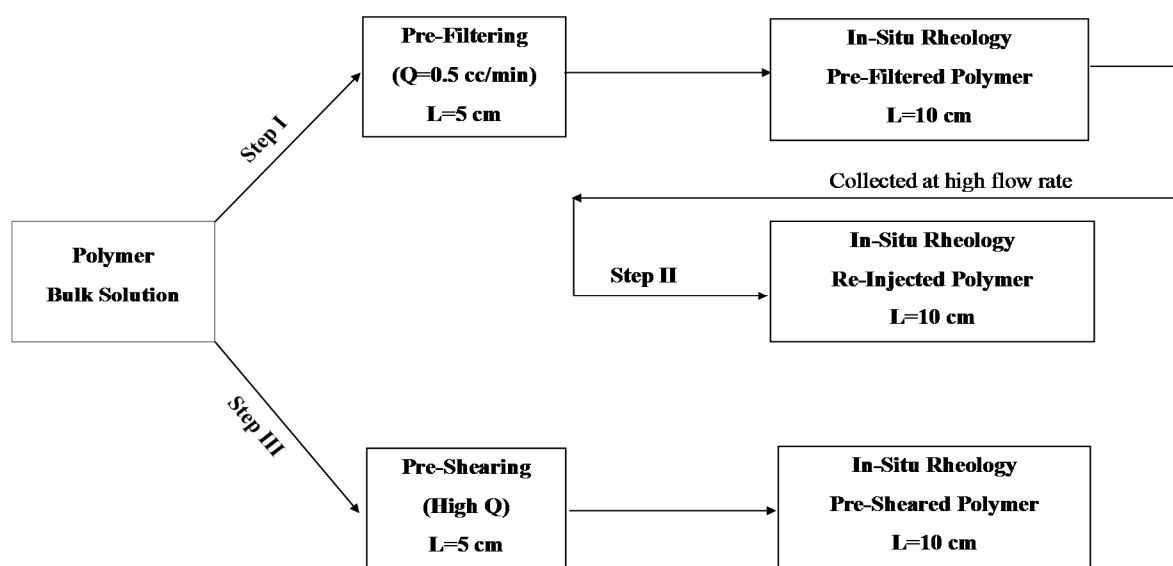
153 The core flooding experiments were carried out under room temperature (22 °C). Figure 2
 154 illustrates the experimental set-up which consists of dual piston Quizix pump, transfer cylinder, core
 155 holder, backpressure regulator and effluent collector which were mounted in series. Differential
 156 pressure transducers with maximum range of 5 and 30 bar were mounted between the inlet and
 157 outlet of the core holder to monitor pressure drop across the core during injection. Backpressure
 158 regulator was mounted at the outlet of the core to apply a pressure of 5 bar to dissolve any air that
 159 maybe found within porous medium before flooding. Backpressure regulator was removed during
 160 polymer injection to avoid polymer degradation.



161
162 **Figure 2.** Schematic diagram of core flooding apparatus

163 Before injecting polymer solution into the cores for in-situ rheology, the polymer solutions were
164 pretreated differently as illustrated in Figure 3. Polymer pretreatment methods are described below:

- 165 • Step.I, **pre-filtering** the polymer solution through short cores ($L = 5$ cm) at low flow rate ($Q = 0.5$
166 cc/min, $v_D = 0.6$ m/day). The pre-filtered polymer solution will be then injected into longer
167 core ($L = 10$ cm) for in-situ rheology measurements. This step was carried out to avoid any
168 microgel in the solution and filter out any possible large MW species. This step represents
169 industrialized polymers which are used in field applications. It is also considered as a baseline
170 for comparison with polymer solutions which were treated differently based on Steps II and III.
171
- 172 • Step.II, **re-injecting** the effluent that was collected from Step.I through long core ($L = 10$ cm) at
173 highest flow rate (e.g., the highest flow rate was achieved when the difference between
174 overburden pressure and pump pressure is 10 bar). This injection rate was ($Q = 12.0$ and 15.0
175 cc/min, $v_D = 15.5$ and 19.3 m/day) for solution A and B, respectively. This step was carried out to
176 investigate the effect of core length on the extent of degradation mechanisms as the degradation
177 performed on long core compared to short core in Step III. Also re-injected solution represents the
178 flow of industrial polymer (prefiltered polymer) deep in reservoir that has already experienced
179 filtration and degradation effects.
180
- 181 • Step.III, **pre-shearing** the polymer solutions through short cores ($L = 5$ cm) at highest flow rate
182 that obtained from Step.II. Then, the pre-sheared solution will be injected into longer core ($L =$
183 10 cm) for in-situ rheology measurements. In this step, large MW species in the solution are
184 possibly filtered and mechanically sheared to lower MW species.



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Figure 3. Flow chart of polymer pre-treatment methods and injection into porous medium

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After saturating the core plugs with brine and porosity measurement, brine was injected into the cores at different flow rates to determine the absolute permeability. Absolute permeability (K_{abs}) was obtained by Darcy's law (Eq. 1):

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$$K_{abs} = \frac{Q \cdot \eta \cdot L}{\Delta P \cdot A} \quad (1)$$

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where, Q is injection flow rate, η is fluid viscosity, ΔP is pressure drop across the core, L and A are core length and cross-sectional area, respectively. Note that, interstitial velocity (v) was obtained from Darcy velocity (v_D) where ϕ is the porosity of porous media.

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$$v = \frac{Q}{A \cdot \phi} = \frac{v_D}{\phi} \quad (2)$$

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Polymer solution, after pretreatment, was injected into the core using Quizix pumps at low flow rate ($Q = 0.5$ cc/min, $v_D = 0.6$ m/day) for at least 2 PV to satisfy porous medium polymer adsorption level and achieving stable differential pressure. Then, polymer flow rate was varied in a stepwise manner from highest to lowest rate. Each rate step was continued until stabilized pressure drop across the core was achieved. Resistance factor (RF) was calculated as following [43]:

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$$RF = \frac{\Delta P_p}{\Delta P_w} \quad (3)$$

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where, ΔP_p is the pressure drop of polymer during polymer flow and ΔP_w is the pressure drop of brine before polymer flow in porous medium.

Samples of effluents were collected at different flow rates to measure their shear viscosity by rheometer and compare with initial solution viscosity. The following equation was used to express the change in shear viscosity by taking shear viscosity at shear rate of 10 s^{-1} [40]:

207

$$\text{Deg}(\%) = \frac{\eta_i - \eta_e}{\eta_i - \eta_w} \cdot 100\% \quad (4)$$

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where, η_i is injected solution viscosity, η_e is effluent viscosity and η_w is brine viscosity which was measured to be 1.04 cP.

After terminating the polymer injection, core's permeability to brine was re-measured after injecting 5 PV of brine at high flow rates proceeded by two steps of tapering. Tapering was performed by injecting diluted effluent at low flow rate ($Q = 1.0$ cc/min, $v_D = 1.3$ m/day) with 50 and

213 25 % of initial effluent concentration (e.g., in case of solution A, effluent of this polymer was
 214 collected and diluted into 250 and 125 ppm). Residual resistance factor (RRF) which is the
 215 permeability to brine before polymer flooding to permeability of brine after polymer flooding was
 216 calculated using Eq.5 [43] :

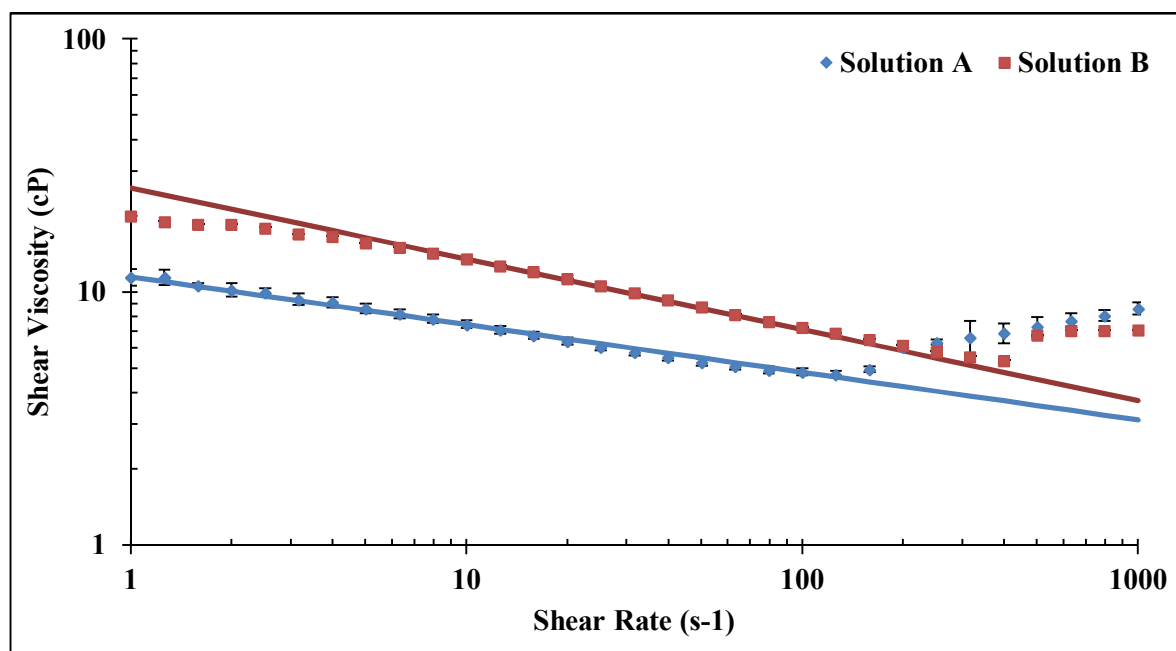
$$217 \quad RRF = \frac{K_{wi}}{K_{wf}} \quad (5)$$

218 where, K_{wi} is the absolute permeability to brine before polymer flow and K_{wf} is the absolute
 219 permeability to brine after polymer flow after tapering.

220 3. Results and Discussion

221 3.1. Shear Viscosity

222 Shear viscosity measurements were carried out for shear rates 1 – 1000 s^{-1} as shown in Figure 4.
 223 Rheometer measurements were very well matching with power law model by setting $n = 0.81$ and
 224 0.72 and $k = 11.49$ and 25.80 for bulk solutions A and B, respectively. Both solutions show
 225 predominantly shear thinning behavior, while solution B show indications of a Newtonian plateau
 226 for shear rates $< 2 s^{-1}$. The viscosity increase observed at high shear rates is mainly due to turbulence
 227 flow caused by high rotational speed (is also observed for brine) that cause an artifact on
 228 measurements [44]. Shear viscosity is 7.45 and 13.54 cP for bulk solutions A and B respectively, using
 229 the reservoir flow relevant shear rate of 10 s^{-1} as a reference.
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 232 **Figure 4.** Shear viscosity of bulk solutions A and B in 1 wt. % NaCl at 22 °C. Solid lines represent
 233 power law model. Error bars for Solution B are smaller than the size of each point (± 0.20 cP).

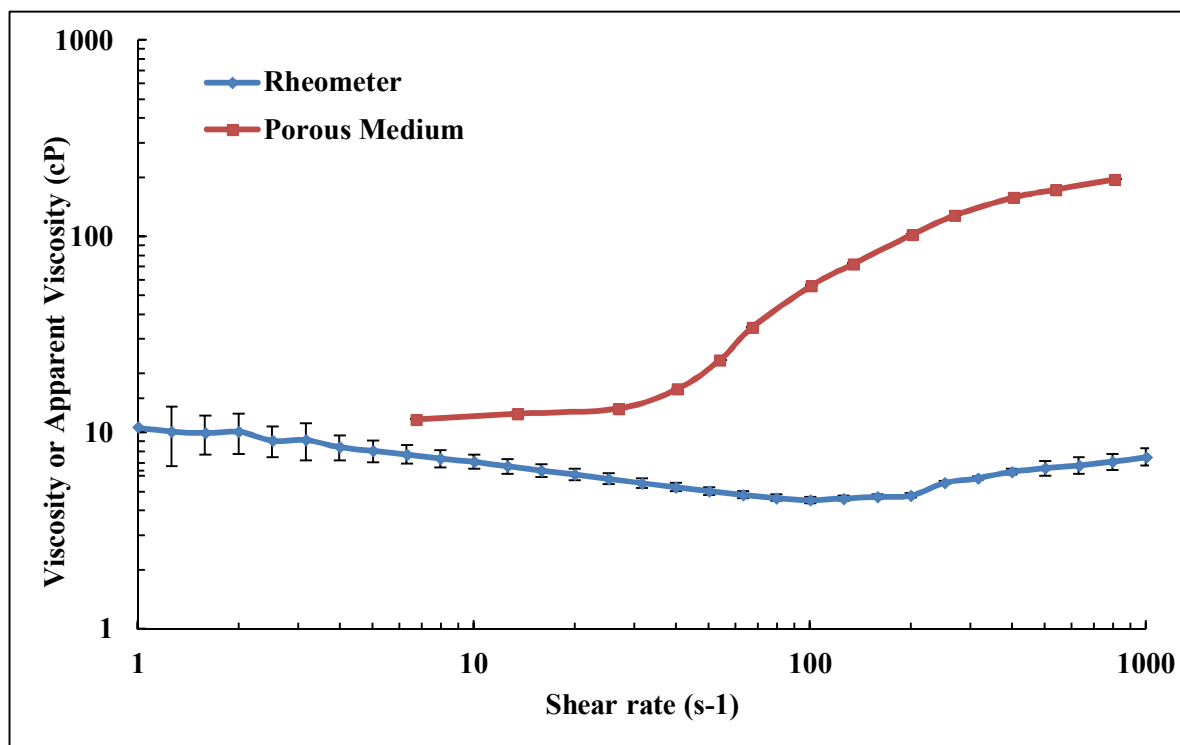
234 3.2. Apparent Viscosity in Porous Medium versus Shear Viscosity in Rheometer

235 In order to compare the flow of polymer in porous medium with the flow in rheometer, it is
 236 necessary to translate the flow velocity in porous medium to shear rate. Determining the exact
 237 equivalent shear rate in porous medium is not possible due to its distribution of pore size, tortuosity
 238 and complexity. The following equation is conventionally used to estimate an effective shear rate in
 239 porous medium [7] :

$$240 \quad \dot{\gamma} = \alpha \frac{4 v_D}{\sqrt{8 K_w \phi}} \quad (6)$$

241)

242 where v_D is Darcy fluid velocity, ϕ porosity, K_w absolute permeability, α shape factor which is an
 243 empirical parameter. Here we have applied an α value of 2.5 for consolidated sandstone
 244 (Bentheimer) [3]. The comparison of the flow in rheometer versus porous medium for solution A is
 245 shown in Figure 5 using above equation. At lower shear rates ($\dot{\gamma} < 30 \text{ s}^{-1}$) apparent viscosity
 246 approaches the upper Newtonian plateau observed from bulk flow shear viscosity. However, at
 247 moderate and high shear rates, apparent viscosity in porous medium diverges from viscosity
 248 measured in rheometer. This is clearly seen in which Newtonian and shear thickening behaviors
 249 observed in porous medium, while shear thinning behavior is shown by rheometer. However, the
 250 two curves are not expected to show the same trend as flow in porous medium is different from
 251 shear flow in a rheometer.

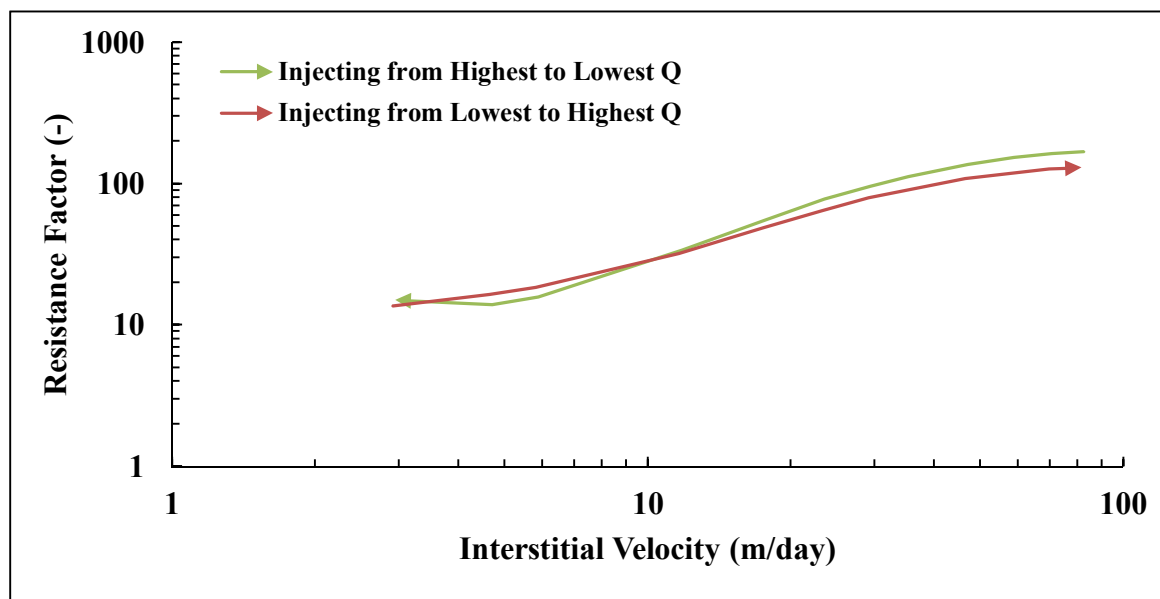


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253 **Figure 5.** Viscosity of pre-filtered solution A (Concentration = 500 ppm, MW = 18 MDa) measured in
 254 porous medium and in bulk.

255 3.3. Apparent Viscosity

256 The apparent viscosity of polymer flowing in porous medium is represented by the resistance
 257 factor (RF). RF was calculated for various flow rates in stepwise manner and are plotted versus
 258 interstitial velocity (see Figure 6). The experiment was performed on twin plugs (see Table 2) with
 259 the same polymer. An apparent Newtonian plateau is observed at low flow velocity. Above a critical
 260 flow velocity, i.e. 3 – 5 m/day, shear thickening is observed. It was found that the injection scheme
 261 from lowest to highest flow rate has lower RF at higher velocities; the difference between the
 262 schemes was 23 % at the highest velocity. The difference is relatively smaller at lower velocities. The
 263 pressure was stable for the highest rate for both schemes and the reason for the discrepancy is not
 264 clear. It is, however, reproducible and may be due to difference in hydrodynamic retention for a core
 265 saturated at high rate versus one saturated at a low rate. Since adsorption measurements were not
 266 performed for the two cases, no firm conclusions can be made on this matter. The injection scheme
 267 from highest to lowest flow rates was adopted for this study. This may also be more representative
 268 for flow velocities experienced in a reservoir where it is subjected to high velocities in the near
 269 wellbore area.



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Figure 6. Resistance factor versus interstitial velocity for pre-sheared solution A. The polymer was sheared at $Q = 12$ cc/min and RF measured at different injection schemes.

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Table 2. Properties of cores used for hysteresis investigation of polymer injection scheme.

	L (cm)	D (cm)	ϕ (-)	K_{wi} (Darcy)
Injection from lowest to highest Q	9.41	3.75	0.22	2.17
Injection from highest to lowest Q	9.44	3.78	0.22	2.18

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3.4. Backpressure Regulator Effects

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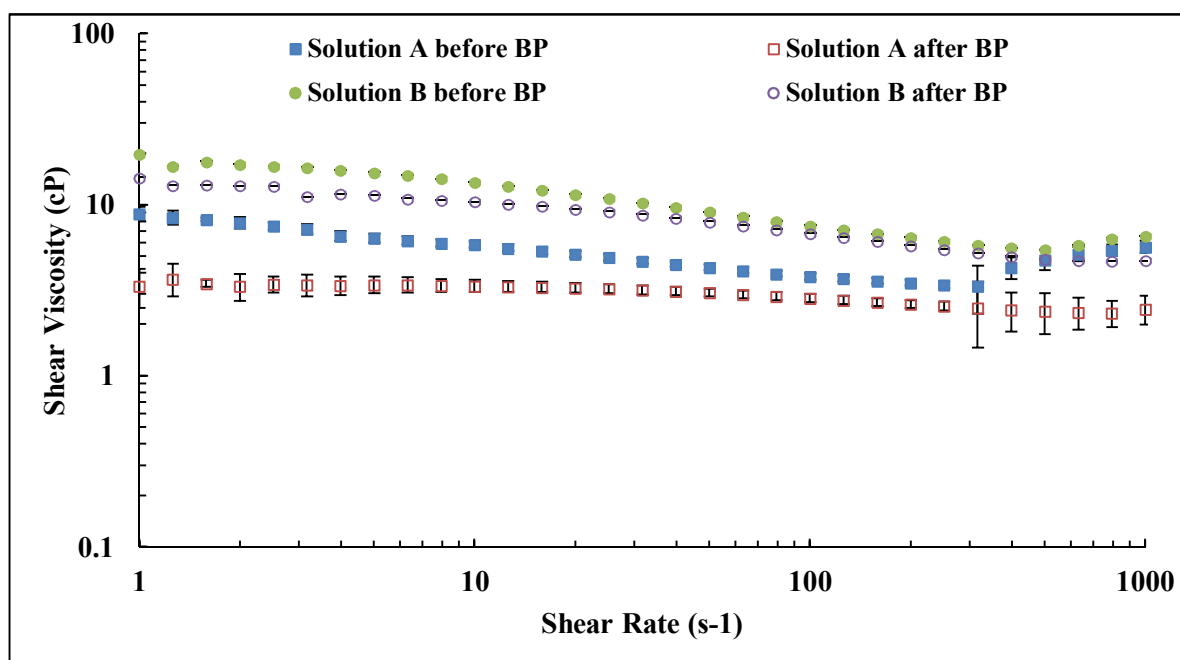
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Typical core flood setup consists of a backpressure regulator which is used to stabilize the pressure across the core and removes any air within the system. However, it was found backpressure regulator can induce a mechanical degradation of polymer as can be seen from the reduction of shear viscosity as presented in Figure 7 and tabulated data in Table 3. High molecular weight polymer with low concentration (solution A) experience high degradation compared to lower molecular weight polymer (Solution B). Solution A lost more than 50% of its original viscosity after passing backpressure regulator at high flow rates. The flow rates applied for the investigation were $Q = 12$ and 15 cc/min for Solution A and B, respectively. Note that, backpressure regulator was not used during investigation of polymer in-situ rheology to avoid mechanical degradation of polymer.



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Figure 7. Shear viscosity vs shear rate of polymers injected through backpressure regulator at high flow rates

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Table 3. Effect of backpressure regulator on shear viscosity of solution A and B. Shear viscosity measured at $\dot{\gamma} = 10 \text{ s}^{-1}$.

Shear Viscosity (cP)	Solution A		Solution B	
	Before BP	After BP	Before BP	After BP
	5.85	3.36	13.53	10.43
Deg (%)	0.00	51.76	0.00	24.82

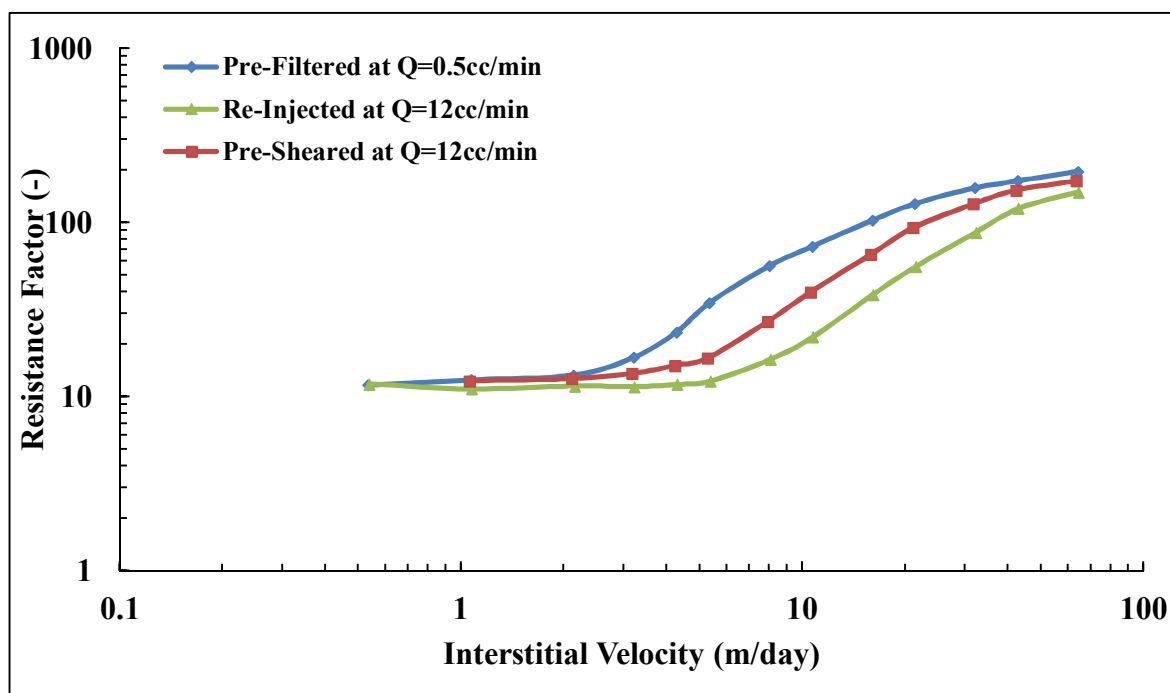
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3.5. In-Situ Viscosity of Solution A

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Three core floods were performed to determine the influence of polymer pre-treatment on in-situ viscosity. The three pre-treatment methods are described in Figure 3. Core data are given in Table 4. Resistance factor as a function of flow velocity for solution A (Concentration = 500 ppm, MW 18 MDa) are presented in Figure 8. All three core floods show apparent Newtonian behavior at low flow velocities followed by shear thickening behavior at higher flow velocities. At low flow velocities all the solutions approach an RF value of ~12 regardless of their pre-treatment procedure. However, after the onset of shear thickening, the solutions show distinctly different RF curves. The pre-filtered solution exhibits the highest RF compared to the pre-sheared and reinjected solutions. The RF correlates to the onset of shear thickening. Pre-filtered solution has an onset of shear thickening at $v_c = 2.5 \text{ m/day}$ which is lower than for pre-sheared and reinjected solutions which have $v_c = 4.0$ and 6.7 m/day , respectively. The onset of shear thickening was measured apparently from the RF curve that represents the point of departure from Newtonian to shear thickening behavior. The shift in the onset of shear thickening to higher velocities indicates that the polymer has experienced degradation and that the molecular weight distribution is altered [25,26].

In addition to the effect on the onset of shear thickening, mechanical degradation markedly affects the degree of shear thickening. The degree of shear thickening is represented by the slope of shear thickening part of the RF curve, i.e. the change of RF with respect to flow velocity. The higher slope was found for pre-filtered solution, $m = 7.7 \text{ (m/day)}^{-1}$ compared to pre-sheared and reinjected solutions ($m = 4.3$ and 3.0 (m/day)^{-1} , respectively). Reinjected solution experienced further degradation as reinjected into porous media. This indicates further degradation occurred with increasing core length.



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Figure 8. Resistance factor (RF) versus interstitial velocity for solution A.

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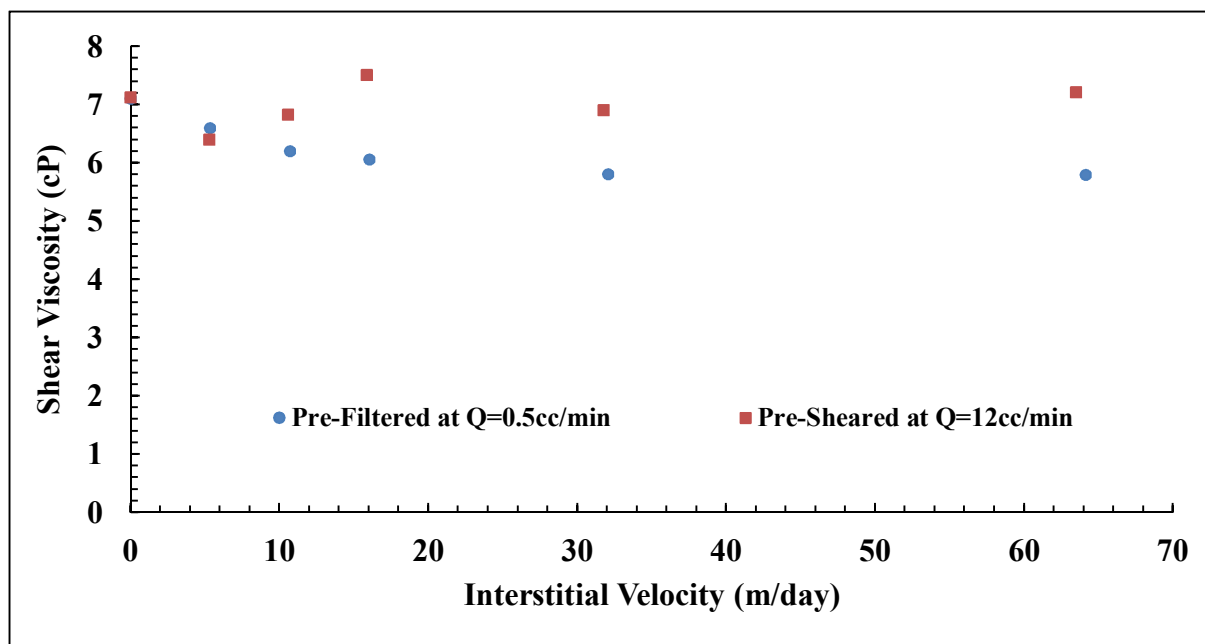
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During the polymer injection, shear viscosity of effluents were measured as shown in Figure 9. The viscosity at $v = 0$ m/day means the injected viscosity of solution after passing pre-treatment processes (η_i). The highest flow velocity ($v = 64.1$ m/day) for the pre-filtered solution showed a shear degradation of 22 %. However, no significant shear degradation was observed for pre-sheared solutions. The injected viscosity of the reinjected solution is lower than that of injected pre-filtered and pre-sheared solutions which could explain the lower RF observed in the porous medium. Still, this is not sufficient to explain the difference in RF curves observed for pre-sheared and pre-filtered solutions that have similar injected viscosity but show different flow behavior in the porous medium. This is further evidence that in-situ viscosity cannot be predicted from bulk shear viscosity measurements, even for flooding experiments with the same polymer, concentration, brine and temperature. However, this is only true for flow velocities above the onset of shear thickening. Below, at typical reservoir flow velocities, the in-situ viscosities are very similar. The pre-treatment at high shear reduces the extensional viscosity [25,26,45-47]. This confirms that, the high RF values at high flow velocities are due to extensional viscosity that is negligible at low velocities, in which shear viscosity is more dominant.



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331 **Figure 9.** Effluents shear viscosity of solution A measured at $\dot{\gamma} = 10 \text{ s}^{-1}$. Error bars are smaller than
 332 the size of each point ($\pm 0.20 \text{ cP}$).

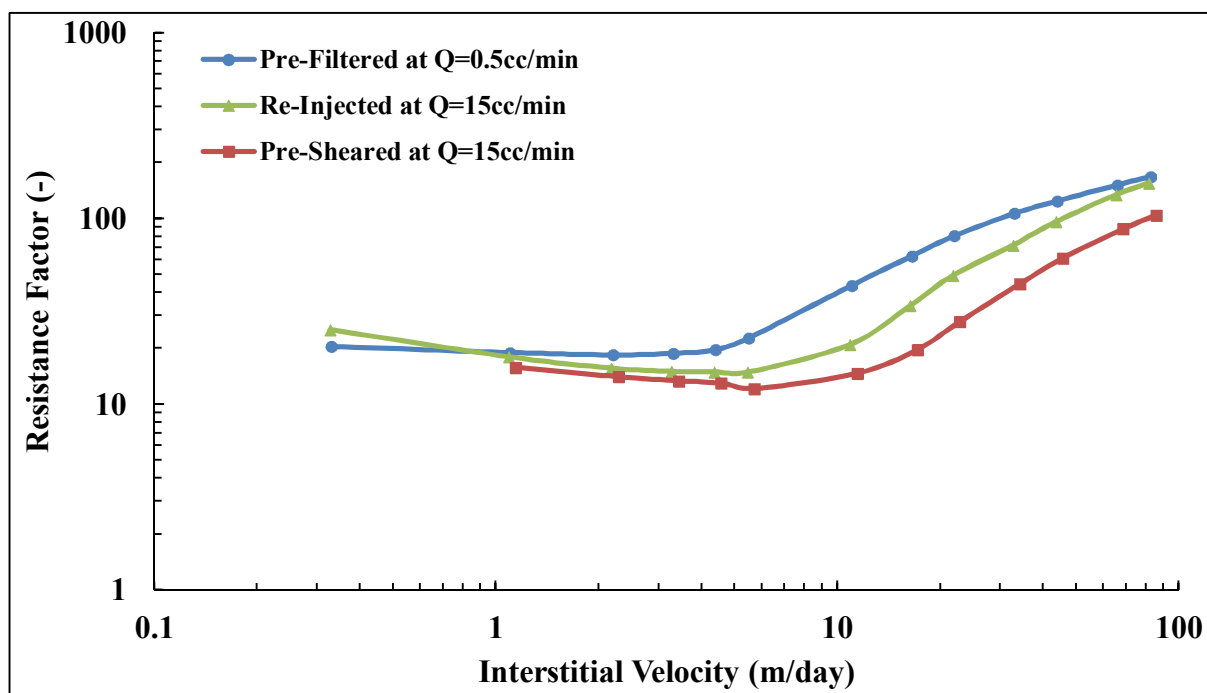
333 **Table 4.** Core and solution properties for injected solution A (Concentration = 500 ppm, MW = 18
 334 MDa)

	L (cm)	D (cm)	ϕ (-)	K_{wi} (Darcy)	K_{wf} (Darcy)	RRF (-)	η_i (cP)	η_e (cP)	v_c (m/day)	m (m/day) ⁻¹
Pre-filtered at Q=0.5cc/min	9.82	3.77	0.24	2.57	1.40	1.84	7.11	5.79	2.51	7.68
Re-injected at Q=12cc/min	9.78	3.77	0.24	2.39	1.28	1.86	5.79	-	6.71	3.00
Pre-sheared at Q=12cc/min	9.72	3.77	0.24	2.25	0.82	2.75	7.13	7.21	4.00	4.32

335

336 3.6. In-Situ Viscosity of Solution B

337 The same procedures were applied for solution B (Concentration = 1000 ppm, MW = 12 MDa)
 338 which has a shear viscosity of 13.54 cP at 10 s^{-1} . Figure 10 shows RF curves for solution B at different
 339 preparation procedures. Core data are given in Table 5. At low flow velocities, a weak shear thinning
 340 behavior was observed, which was not observed for solution A. This is most likely because solution
 341 B has higher concentration than polymer A for which the degree of entanglements is higher in
 342 solution B. Another observation is that all three solutions approach similar RF values at low
 343 velocities, which shows that the degree of pre-shearing and re-injection of polymer at high rates do
 344 not significantly change the in-situ viscosity at typical reservoir flow rates. This is in sharp contrast
 345 to higher flow rates where significant differences are observed. It was found that the pre-filtered
 346 solution possesses an earlier onset of shear thickening ($v_c = 4.1 \text{ m/day}$) compared with pre-sheared
 347 and reinjected solutions ($v_c = 12.0$ and 7.7 m/day ; respectively). Furthermore, the degree of shear
 348 thickening was higher for pre-filtered solution, $m = 3.5 \text{ (m/day)}^{-1}$ compared to pre-sheared and
 349 reinjected solutions ($m = 1.5$ and 2.3 (m/day)^{-1} ; respectively).



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Figure 10. Resistance factor (RF) versus interstitial velocity for solution B (Concentration = 1000 ppm, MW = 12 MDa).

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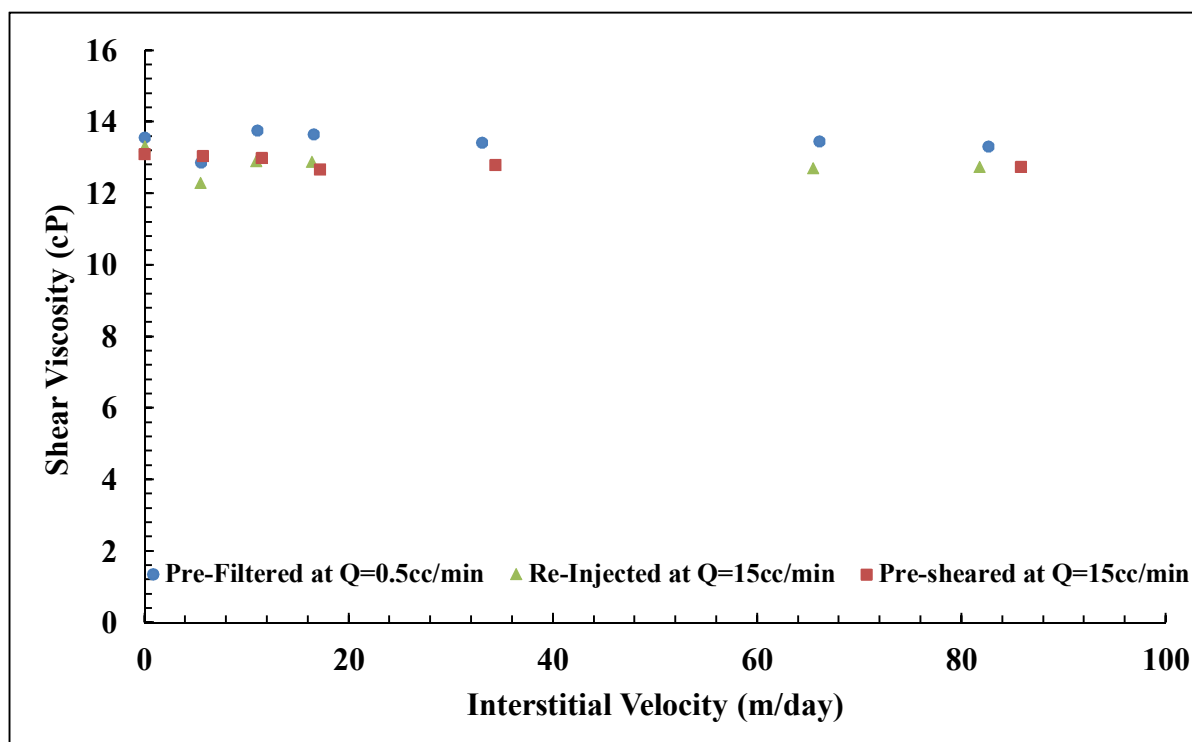
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Effluent viscosities do not show significant degradation for any of the three core floods as shown in Figure 11. This corresponds well with the RF curves that show similar viscosity at typical reservoir flow velocities, i.e. 0.1 to 1.0 m/day. However, the effluents collected at higher flow velocities, such as 30 to 70 m/day also show similar values, while the RF values are significantly different. This shows that the difference in shear thickening at high velocities is due to extensional viscosity which is not reflected in the measurements of shear viscosity.



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Figure 11. Effluent shear viscosity of solution B measured at $\dot{\gamma} = 10 \text{ s}^{-1}$. Error bars are smaller than the size of each point ($\pm 0.20 \text{ cP}$).

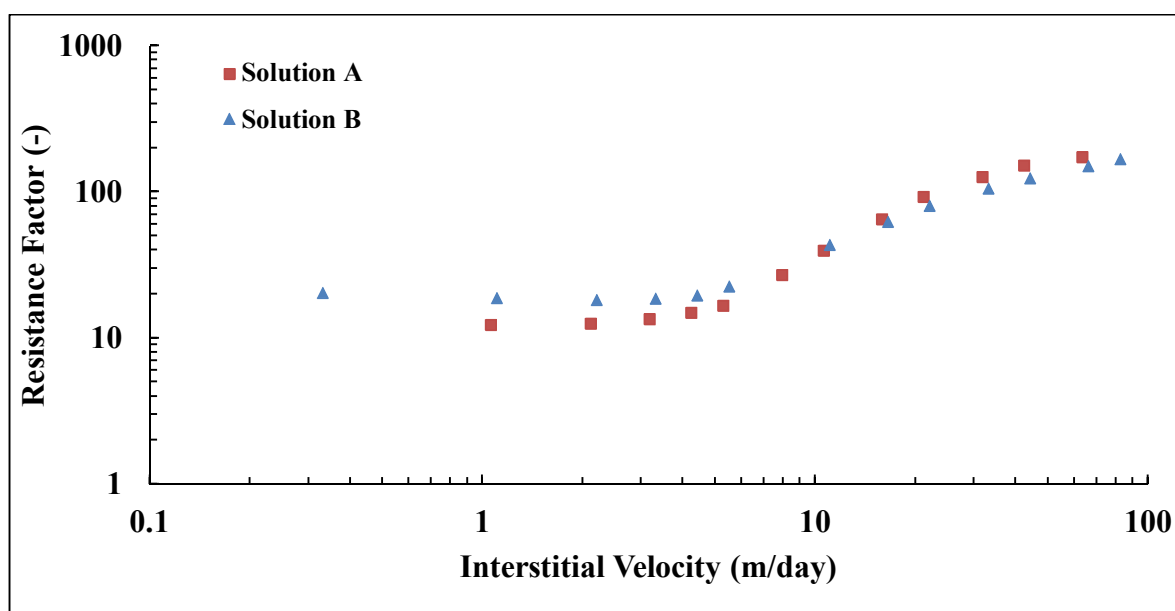
362 **Table 5.** Core and solution properties for injected solution B (Concentration = 1000 ppm, MW = 12
 363 MDa)

	L (cm)	D (cm)	ϕ (-)	K_{wi} (Darcy)	K_{wf} (Darcy)	RRF (-)	η_i (cP)	η_e (cP)	v_c (m/day)	m (m/day) ⁻¹
Pre-filtered at (Q=0.5cc/min)	9.82	3.79	0.23	2.16	0.96	2.24	13.57	13.31	4.06	3.50
Re-injected at (Q=15cc/min)	9.57	3.79	0.23	2.08	1.24	1.68	13.31	12.75	7.69	2.28
Pre-sheared at (Q=15cc/min)	10.27	3.77	0.23	2.80	1.54	1.82	13.10	12.75	11.99	1.46

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365 3.7. Influence of degradation on polymer molecular weight

366 Figure 12 shows the effect of pre-shearing on onset of shear thickening for high molecular
 367 weight polymer (solution A) to that of lower molecular weight polymer (solution B). Shift in onset of
 368 shear thickening has been an indication of reduction of molecular weight distribution [9,23,25]. It is
 369 generally difficult to quantify the reduction of MWD due to the difficulties in determining MW of
 370 high molecular weight polymers by methods such as size exclusion chromatography (SEC) and
 371 asymmetric flow field-flow fractionation (AF4) [48-50]. To characterize the reduction by core floods
 372 would usually require a large number of core floods. In this experiment, the onset of shear
 373 thickening and the shape of the RF curves are similar after the onset of shear thickening for
 374 pre-sheared solution A and pre-filtered solution B (Figure 12). This indicates the MWD of solution A
 375 was shifted to lower distribution similar to that of prefiltered solution B. This supports the
 376 observation given by Puls, *et al.* [51] in which pre-shearing process reduces HPAM molecular
 377 weight distribution as they observed by using size exclusion chromatography (SEC) for determining
 378 the MW of solution.
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381 **Figure 12.** Influence of pre-shearing on solution A (MW 18 MDa) compared to pre-filtered solution
 382 B (MW 12 MDa).

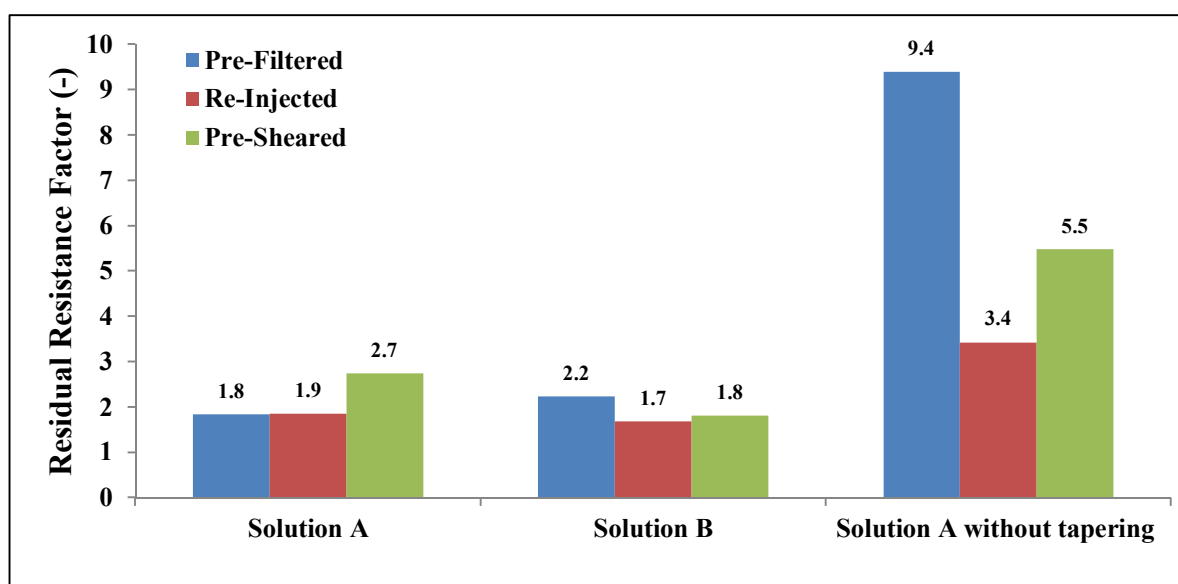
383 3.8. Permeability Reduction

384 HPAM flowing in a porous medium will adsorb on rock surfaces and be trapped within narrow
 385 pores, resulting in polymer retention. Polymer retention consequently causes permeability reduction

386 which can be experimentally evaluated by residual resistance factor (RRF). Lake [2] reported that
 387 RRF can be reduced if the polymer is pre-sheared before injection. RRF is a function of polymer
 388 molecular weight, degree of hydrolysis, flow velocity and pores structures. Other authors, such as
 389 Yerramill, *et al.* [52] and Morris and Jackson [45] reported that RRF increases with increasing
 390 polymer concentration.

391 RRF values obtained for solutions A and B in this experimental study are shown in Figure 13.
 392 The RRF for pre-filtered solutions A and B were 1.8 and 2.2, respectively. In this context, the
 393 difference in RRF is regarded as small, particularly when considering the difference in concentration
 394 and molecular weight for solutions A and B (Concentration = 500 and 1000 ppm, MW = 18 and 12
 395 MDa, respectively). Similarly, there were small differences in RRF between degraded (pre-sheared
 396 and re-injected) and non-degraded solutions (pre-filtered).

397 How to measure “true” RRF has been debated [53]. In this experiment, the influence of tapering
 398 on RRF was quantified. RRF values were reduced from 9.4 to 1.8 for pre-filtered solution A before
 399 and after tapering; respectively. Recall that tapering was performed to flush-out as much as possible
 400 of retained polymer. Tapering was performed in two steps before water post-flush by injecting
 401 diluted polymer effluent with 50 and 25 % of initial effluent polymer concentration.
 402



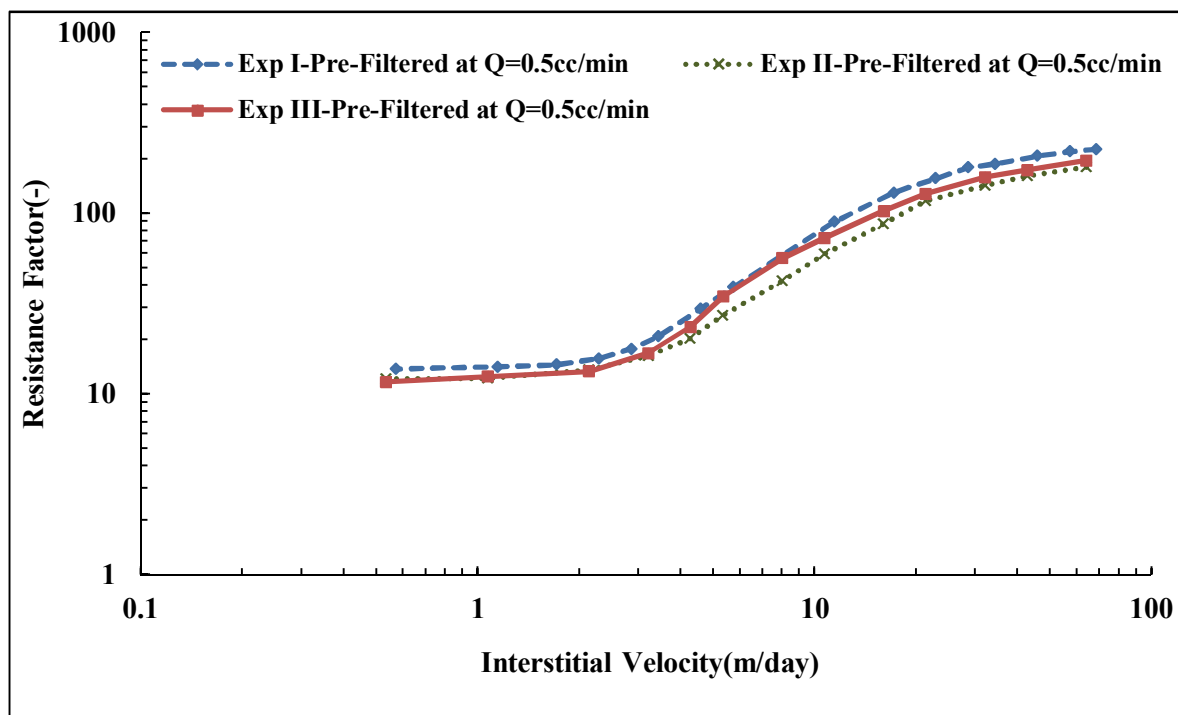
403

404 **Figure 13.** Residual resistance factor (RRF) after tapering for both solutions and without tapering for
 405 solution A.

406 3.9. Reproducibility of Experiments

407 The shift in onset of shear thickening to higher velocities and the reduction in slope for
 408 pre-sheared and re-injected solutions are consistent with a mechanical degradation and alteration of
 409 polymer molecular weight distribution. Figure 8 shows a successive degradation of high molecular
 410 weight polymer as it is reinjected into porous media while it is in contrast with lower molecular
 411 weight solution shown in Figure 10. This was not clear if the polymer might be exposed to further
 412 degradation during reinjection as reported by Sorbie and Roberts [54] and Al Hashmi, Al Maamari,
 413 Al Shabib, Masnsoor, Zaitoun and Al Sharji [31].

414 To further investigate the difference on the extent of mechanical degradation found between
 415 preshearing and reinjection in which the core length has been varied. Similar sets of experiments
 416 were performed which confirm that resistance factor of prefiltered solutions for both polymers are
 417 reproducible as shown in Figure 14. This confirms that RRF is reproducible for prefiltered solution
 418 which indicates the cores were fairly homogenous in which they have quite similar absolute
 419 properties (e.g., porosity and permeability as shown in Table 6).
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Figure 14. Reproduced resistance factor vs. interstitial velocity of pre-filtered solution A.

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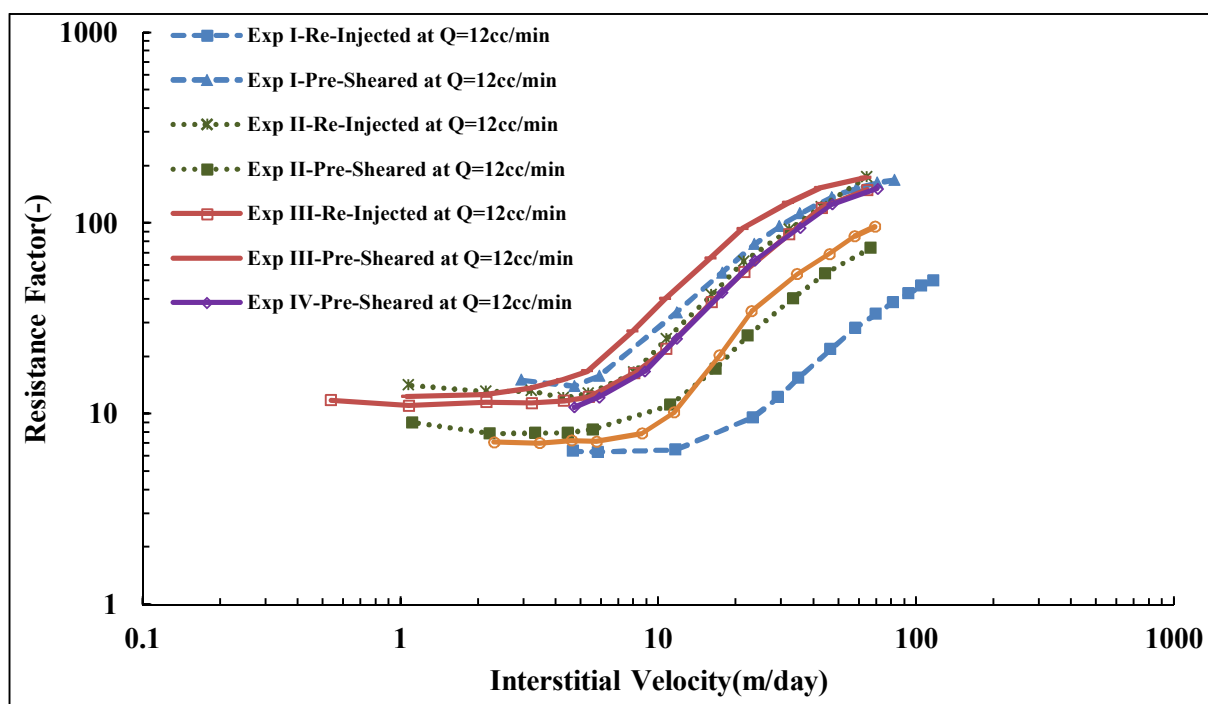
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However, some disparity found on reproducing resistance factor of presheared and reinjected solutions. It was found in some experiments reinjected solution experience further degradation compared to presheared solution as in Exp I and III in Figure 15. The high degradation found of reinjected solution on Exp I is attributed to the associated effect with the use of backpressure regulator which was not used in the rest of the experiments. The match between RF curves of presheared solution in Exp II and V demonstrates that preshearing process can filter and preshear the polymer. A disparity was found on reproducing resistance factor of presheared solution as well. This could be expected due to the fact that pretreated polymer could experience different fields of shear rate due to different topology of the cores regardless of similarity on the cores' apparent petrophysical properties (see Table 6). Preshearing process could further or similar degrade the polymer as reinjection process does and vice versa. This indicates the core length or exposure time has an effect on degradation mechanism, although, some previous studies[40] reported the polymer degradation has less dependency in on core length. This could be true in synthetic porous medium but not in realistic porous medium such core plugs. However, the dependence of polymer successive degradation on characteristic length to approach a steady state value in a realistic porous medium is ambiguous and still not defined.



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Figure 15. Reproduced resistance factor vs. interstitial velocity of presheared and reinjected solution A.

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Table 6. Core and solution properties for reproducibility study performed with solution A (Concentration = 500 ppm, MW = 18 MDa)

	L (cm)	D (cm)	ϕ (-)	K_{wi} (Darcy)	K_{wf} (Darcy)	RRF (-)	η_i (cP)	η_e (cP)
Exp I-Pre-Filtered at Q=0.5cc/min	9.37	3.76	0.23	2.59	0.28	9.40	7.34	5.26
Exp I- Re-Injected at Q=12cc/min	9.58	3.77	0.22	2.28	0.67	3.43	5.26	4.84
Exp I- Pre-Sheared at Q=12cc/min	9.44	3.78	0.22	2.18	0.40	5.48	-	-
Exp II- Pre-Filtered at Q=0.5cc/min	9.54	3.77	0.24	2.48	1.35	1.84	6.81	6.62
Exp II- Re-Injected at Q=12cc/min	9.81	3.77	0.24	2.41	1.17	2.06	6.62	6.18
Exp II- Pre-Sheared at Q=12cc/min	9.74	3.77	0.23	2.19	0.98	2.24	4.06	4.55
Exp III- Pre-Filtered at Q=0.5cc/min	9.82	3.77	0.24	2.57	1.40	1.84	7.11	5.79
Exp III-Re-Injected at Q=12cc/min	9.78	3.77	0.24	2.39	1.28	1.86	5.79	-
Exp III-Pre-Sheared at Q=12cc/min	9.72	3.77	0.24	2.25	0.82	2.75	7.13	7.21
Exp IV -Pre-Sheared at Q=12cc/min	9.88	3.78	0.22	2.31	1.12	2.06	6.51	6.30
Exp V-Pre-Sheared at Q=12cc/min and Pre-Filtered at Q=0.5cc/min	10.00	3.79	0.22	2.72	1.15	2.36	5.93	5.37

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3.10. Preshearing at Very High Flow Rates

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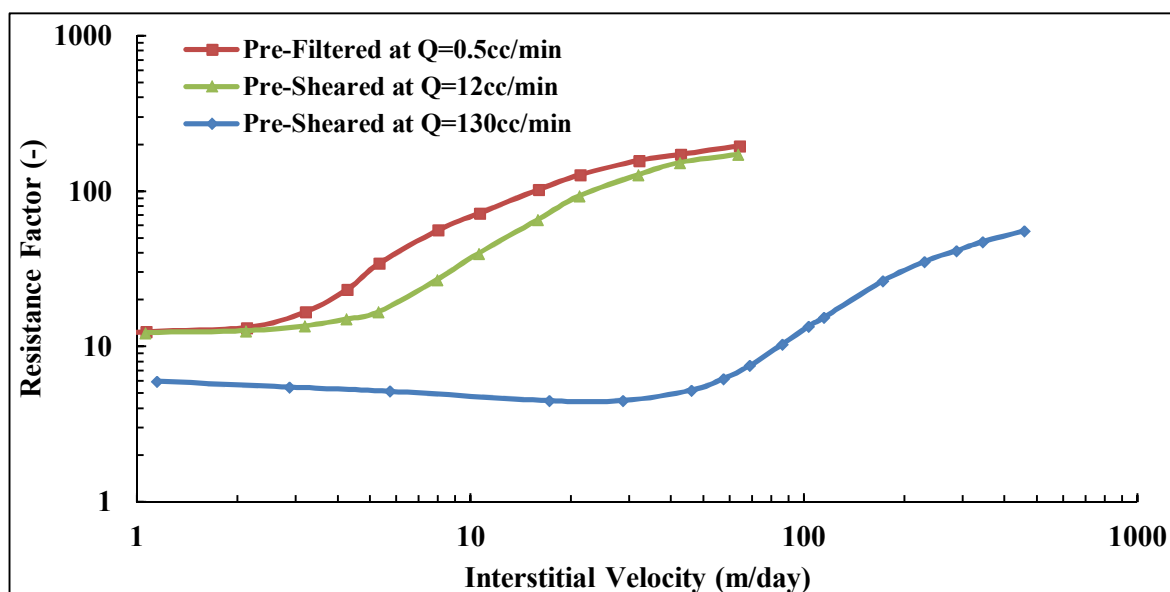
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The preshearing process described on Figure 3 was based on the highest shear rate that was achieved from the first step which was limited by experimental setup. In this experiment, the polymer (Solution A) was presheared at extremely high flow rate ($Q = 130$ cc/min, $v_D = 166.8$ m/day) in short core and its shear viscosity degraded by 32 % compared to original bulk viscosity of 6.16 cP. It can be seen in Figure 16 a significant reduction on its in-situ viscosity e.g., the reduction is more than 50% compared to prefiltered and presheared polymers at lower flow rate. Moreover, a significant alteration can be observed on viscoelastic properties of presheared polymer at high flow rate. Its onset of shear thickening is shifted to much higher velocities and degree of shear thickening

453 reduced significantly. Furthermore, it was found that residual resistance factor can be reduced if
 454 the polymer is submitted to high degradation as can be seen in the Table 7. The core properties of
 455 prefiltered and presheared polymer at $Q = 12$ cc/min were presented earlier in Table 4.
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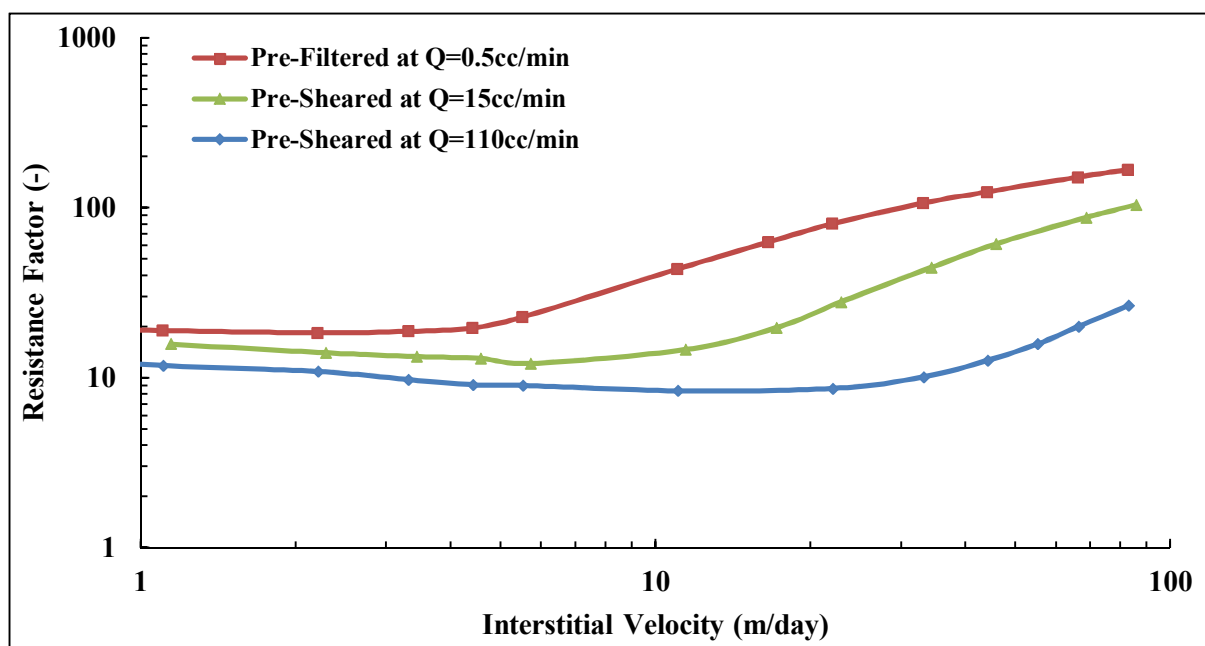


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458 **Figure 16.** Resistance factor of presheared solution A at different flow rates versus prefiltered
 459 solution.

460 Similar observations were found for solution B with lower molecular weight that was
 461 presheared at high flow rate ($Q=110$ cc/min, $v_D = 141.2$ m/day). Note that the flow rate applied for
 462 preshearing this solution was lower than solution A due to higher concentration of solution B. Its
 463 shear viscosity degraded by 16% compared to bulk viscosity of solution B which was 13.25 cP. This
 464 results in a significant loss of its in-situ viscosity as well compared to prefiltered and presheared
 465 solutions at lower flow rate as can be seen in Figure 17. It is viscoelastic parameters reduced and the
 466 onset of shear thickening shifted to much higher velocities while the degree of shear thickening is
 467 reduced. Core properties and viscoelastic parameters for prefiltered and presheared solution B at
 468 low and high flow rates are given in Table 5 and 7; respectively.

469 It can be also seen from Figure 16 and 17 that both solutions responded differently to the high
 470 flow rate applied for each of them. High molecular weight polymer suffers more degradation
 471 compared to lower solution as discussed earlier in this paper. Although it was presheared at lower
 472 rate compared to solution B. Hence, the applied flow rate for preshearing process has to be
 473 optimized to avoid the loss of polymer viscosity while improve its viscoelastic properties that results
 474 in better injectivity.



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Figure 17. Resistance factor of presheared solution B at different flow rates versus prefiltered solution.

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Table 7. Core and solution properties for presheared solution A and B at very high flow rates.

	L (cm)	D (cm)	ϕ (-)	K_{wi} (Darcy)	K_{wf} (Darcy)	RRF (-)	η_i (cP)	η_e (cP)	v_c (m/day)	m (m/day) ⁻¹
Solution A										
Pre-sheared at Q=130cc/min	8.87	3.78	0.22	2.37	1.49	1.6	4.52	4.47	53.61	0.18
Solution B										
Pre-sheared at Q=110cc/min	9.71	3.79	0.23	1.87	1.09	1.7	11.30	11.26	33.5	0.37

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4. Conclusions

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In-situ rheology of HPAM polymers has been investigated for solutions that have been subjected to different conditions prior to in-situ measurements. Three conditions were evaluated: 1) injection of solutions injected through a short Bentheimer core at low rate (prefiltering), 2) prefiltered solutions subsequently injected at high rate through a Bentheimer core (reinjecting) and 3) solutions injected at high rate through a short Bentheimer core (preshearing). The following conclusions could be made:

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- Initial studies show that backpressure regulator can induce a mechanical degradation of polymer that reduces its viscoelastic properties and may lead to erroneous conclusions from laboratory experiments. All experiments reported have been made without a backpressure regulator.
- HPAM flowing in porous medium at low velocities (e.g., reservoir velocities) show predominantly Newtonian behavior followed by shear thickening at higher flow velocities (e.g., at wellbore area). Both flow behaviors are absent in rheometer measurements that demonstrate a predominantly shear thinning behavior at comparable flow rates. This conclusion is limited to the polymer concentration used (weak semi-dilute region).
- At high flow velocities similar to those experienced in the near wellbore area, polymer flow history plays a substantial role for HPAM viscoelasticity. Solutions exposed to high rates showed delay in onset of shear thickening and reduction in apparent viscosity compared to the solutions exposed to low rates.

- 499 • All solutions, regardless of previous exposure to high or low rates, showed similar apparent
 500 viscosity and predominantly Newtonian behavior at low velocities (i.e., reservoir velocities).
 501 • These results show that mechanical degradation is beneficial for the polymer types and
 502 concentrations investigated here since injection pressures are reduced and reservoir apparent
 503 viscosities are maintained.
 504 • Polymer flow history (pretreatment) has little impact on residual resistance factors (RRF). RRF
 505 is found to be more influenced by fluid exchange process (tapering).
 506 • Mechanical degradation during polymer reinjection is coupled to several parameters such as
 507 characteristics of the porous media, flow rate, geometry (inlet/outlet effect), and polymer
 508 exposure time to the porous media. Results indicate that more exposure to porous media may
 509 lead to increased mechanical degradation. However, understanding the effect of exposure time
 510 on mechanical degradation requires more investigations.

511 Author Contributions

512 Conceptualization, T.S. and A.S.; Formal analysis, T.S.; Methodology, B.S.; Supervision, A.S.;
 513 Writing – original draft, B.A.; Writing – review & editing, B.A., T.S., B.S., and A.S.
 514

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520 Conflicts of Interest

521 The authors declare no conflict of interest
 522

523 Nomenclature

A = cross-sectional area (cm²)
 AF4 = Asymmetrical flow field-flow fractionation
 BP = backpressure
 C* = critical overlap concentration (ppm)
 D = core diameter (cm)
 EOR = enhanced oil recovery
 HPAM = partially hydrolyzed polyacrylamide
 k = viscosity constant
 K_{abs} = absolute permeability to brine (Darcy)
 K_{wf} = absolute permeability to brine after polymer flow (Darcy)
 K_{wi} = absolute permeability to brine before polymer flow (Darcy)
 L = core length (cm)
 MW = molecular weight (MDa)
 MWD = molecular weight distribution, dimensionless
 n = power law index, dimensionless
 PV = pore volume, dimensionless
 Q = flow rate (cc/min)
 RF = resistance factor, dimensionless
 RRF = residual resistance factor, dimensionless
 SEC = size exclusion chromatography
 v = interstitial velocity (m/day)

v_c = onset of shear thickening (m/day)
 v_D = Darcy velocity (m/day)
 ΔP_p = pressure drop during polymer flow (bar)
 ΔP_w = pressure drop during water flow (bar)
 η_e = effluent viscosity (cP)
 η_i = injected solution viscosity (cP)
 η_w = brine viscosity (cP)
 ϕ = porosity, dimensionless
 $\dot{\gamma}$ = shear rate (s^{-1})

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