



## 24 1. Introduction

25 Steel production has increased continuously over the years with industrial developments. In  
26 2009, global steel production was reported to be 1.25 billion tons, but increased to 1.6 billion  
27 tons in 2015 [1]. As of 2015, Korea was producing 70 million tons of steel annually, ranking  
28 the 5<sup>th</sup> in the world for steel production and the 1<sup>st</sup> for steel product consumption per person.  
29 Typically, approximately 130~300 kg of slag is produced for every 1 ton of steel [2] so that  
30 more than 400 million tons of steel slag is produced globally, with Korea accounting for  
31 approximately 24 million tons. Blast furnace slag, a high value construction material, has been  
32 efficiently utilized through recycling of byproduct but still used for road asphalt, embankments,  
33 and simple reclamation due to unstable substances. These include free CaO and MgO, which  
34 cause volume expansion in cementitious material. In the previous researches, EOS (Electric  
35 Arc Furnace Oxidizing Slag) was reported to contain smaller amounts of these unstable  
36 materials [3] and used for aggregates for road construction in some countries [4].

37 Further, many studies are being performed for a feasible use of EOS as a fine aggregate for  
38 concrete with shortages of natural aggregates. It was evaluated that the addition of EOS to  
39 concrete showed an increase in strength of concrete [5–7] but was still pointed out that the  
40 particle grading of EOS as a fine aggregate is uneven due to the crushing process employed  
41 during EOS production [8]. The researches on durability evaluation were also performed for  
42 EOS concrete, which showed improved durability performance compared to normal concrete  
43 [9–11]. In the previous works, chloride penetration tests performed on EOS concrete and  
44 improved resistance to chloride ingress was evaluated [12]. The fundamental studies on  
45 physical and chemical characteristics of EOS were investigated considering production  
46 processes as well [13], indicating that the quality of concrete or mortar with EOS varies with

47 used EOS which has different engineering properties with treat process and product region.

48 In the present work, the physical and chemical properties of EOS from South Korea are  
49 investigated firstly, and the engineering properties in cement mortar containing EOS for fine  
50 aggregate are evaluated with several tests like compressive strength, pore characteristics, and  
51 length change measurements. Its applicable feasibility to construction material is discussed  
52 with evaluation of engineering properties.

53

## 54 **2. Fundamntal properties of EOS as fine aggregate for construction material**

### 55 2.1 Electric furnace slag

56 Electric furnace slag is an industrial by-product which is produced during the smelting process  
57 of scrap iron using electricity for molten steel. It can be categorized into two classes, oxidizing  
58 and reducing slag, depending on how the impurities are removed during the steel production,  
59 resulting furnace oxidizing slag-EOS and electric arc furnace reduction slag-ERS, respectively.  
60 When electric furnace slag is produced, small amounts of free CaO and free MgO are generated  
61 in the form of unstable products and these can cause cracks in concrete when incorporated into  
62 concrete [14]. EOS has a large content of Fe but contains small amounts of unstable by-  
63 products. Hence, it has significant potential for use as a concrete material. On the other hand,  
64 ERS has a higher concentration of unstable by-products, which makes its limitedly application  
65 to construction material. In the open-air storage yard, EOS is often mixed with ERS and this  
66 make quality control of EOS difficult.

67

68

69 2.2 Korean Industrial Standard (KS) F 4571 [15]

70 In 2007, Korean Industrial Standards (KS) specifications were proposed for the use of EOS as  
 71 a fine aggregate for concrete. These specifications were amended in 2012 for including the use  
 72 of EOS as a coarse aggregate but it has been used very limitedly due to high transportation  
 73 costs since EOS has been mainly produced in limited locations in Korea. Table 1 lists the  
 74 physical properties and chemical compositions for EOS in the KS specification.

75

76 Table 1. Chemical composition of EOS per KS F 4571 specification

Parameter	Value	
Chemical constituent	CaO	Concentration not more than 40.0%
	MgO	Concentration not more than 10.0%
	FeO	Concentration not more than 50.0%
	Basicity	Not more than 2.0
Absorption (%)	Not more than 2.0	
Density (kg/cm <sup>3</sup> )	3.1~4.5	

77

### 78 3. Experimental program for EOS aggregate

#### 79 3.1 Evaluation of EOS properties as fine aggregate

80 EOS samples were obtained from steel companies (region-Dangjin, H company), and their  
 81 properties were evaluated for use feasibility as a fine aggregate. The tests included sieve  
 82 analysis test referred to KS F 2502 [16], and density/ absorption rate test in KS F 2504[17].  
 83 They are adopted to determine the physical properties of EOS samples. In addition, pH

84 measurements and X-ray diffraction (XRD) analysis were performed to evaluate the chemical  
 85 properties of EOS samples.

86

87 3.2 Evaluation of mortar mixed with EOS aggregate

88 3.2.1 Outline of test program for cement mortar with EOS and ERS

89 Cement mortar was prepared using EOS aggregate. Table 2 lists the mixing proportions in  
 90 accordance with the KS L ISO 679 standard [18]. For the samples of EOS mortar test, EOS  
 91 content was set as 0%, 50%, and 100%. In addition, MIP (Mercury Intrusion Porosimetry) test  
 92 was performed for pore distribution evaluation on the mortar samples. For the compressive  
 93 strength evaluation, the test specimens with 40× 40 × 160 mm were prepared. The pore  
 94 characteristics and strength were evaluated at 3 days, 7 days, and 28 days.

95 Rate of length change tests were also performed on for cement mortar with EOS and ERS. Free  
 96 CaO amount in binder plays an important role in volume swelling, so that CaO measurement  
 97 was carried out. In table 2, mixing proportions for cement mortar were listed with several  
 98 replacement ratios of EOS and ERS.

99

100 Table 2. Mixing proportions used to produce mortar samples

	W/C	Unit weight (kg/cm <sup>3</sup> )				
	(%)	Cement	Water	Sand	EOS	ERS
OPC				1350	-	-
EOS50				675	908.5	-

EOS100	50	450	225	-	1,817	-
EOS9/ERS1				-	1635	182
EOS8/ERS2				-	1453	364

101 EOS50: 50% EOS is used as the fine aggregate, with the remaining 50% being sand.

102 EOS100: 100% EOS is used as the fine aggregate.

103 EOS9/ERS1: 90% EOS and 10% ERS are used as the fine aggregate.

104 EOS8/ERS2: 80% EOS and 20% ERS are used as the fine aggregate.

105

### 106 3.2.2 MIP measurements

107 MIP was used to measure pore distribution and porosity in the hardened mortar samples. The  
 108 measurements were performed using a Micromeritics Auto Pore IV 9505 analyzer at a  
 109 maximum pressure of 33,000 psi (228 MPa). The test particles were extracted from the center  
 110 of the test mortar specimen. This particle size was approximately 5 mm and subsequently  
 111 immersed in an acetone solution for 24 hours, then dried for 2 hours before measurement.

112

### 113 3.2.3 Rate of length change test

114 The test for length change was performed referred to KS L 5107 [19] which covers sample  
 115 preparation and determination of the volume change due to CaO and MgO in the sample. The  
 116 expandability of the EOS mortar evaluated by replacing EOS with ERS from 10% and 20.0%,  
 117 which contains swelling components as listed in Table 1. The test sample with dimensions of  
 118  $25.4 \times 25.4 \times 254$  mm was cut and cured for 24 hours at 20 °C. The initial length of the test  
 119 sample was measured, then it was placed in autoclave and the temperature of the autoclave was  
 120 elevated to the state that the steam pressure reached  $2 \pm 0.07$  MPa after 45~75 min. The heat

121 was turned off after 3 hours, once the pressure had reached  $2 \pm 0.07$  MPa. After 90 min, the  
122 autoclave was allowed to cool, so that the pressure decreased to less than 0.07 MPa. The  
123 ventilation valve was opened for decreasing the inside pressure to atmospheric pressure. Next  
124 the test piece was immediately immersed in water heated to over  $90$  °C , then cold water was  
125 added around the test sample and it was cooled at  $23$  °C for 15 min. After its surface was dried,  
126 the length was measured. The process was repeated after 12 , 24 , 36 , and 48 hours to determine  
127 the rate of length change.

128

#### 129 *3.2.4 Measurement of free CaO content*

130 The ethylene glycol method defined in ASTM STP 985 [20] was used to measure the free CaO  
131 content in EOS sample. The ethylene glycol method can quickly and quantitatively analyze  
132 small amounts of free CaO in cement, so that it has been widely adopted in the previous  
133 researches [21, 22]. For the test, a sample was first extracted based on the KS F 2501 (aggregate  
134 sample extraction method) [23]. The sample was dried for 1 hour in a dryer at  $80$  °C and  
135 crushed in a disk mill for powder type with  $100$   $\mu$ m. In the case of the ethylene glycol method,  
136 the mixing rate must be adjusted based on the concentrations of the materials, that is to say, test  
137 sample should be 1 g if the free CaO content is within 3%. First, 1 g of the sample was extracted  
138 considering the free CaO content in each material. The sample was then placed in a 100-mL  
139 Erlenmeyer flask, and 50 mL of ethylene glycol was added. The sample was mixed with the  
140 ethylene glycol for 30 minutes using a water bath heated to  $60$  °C. The treated sample was then  
141 suction-filtered using two sheets of No. 5B filter paper in a Buchner funnel, and the sample  
142 remaining in the flask was washed using 30 mL of ethylene glycol. The filtrate was placed in  
143 an absorbent Erlenmeyer flask, and approximately 2~3 drops of Bromocresol green solution

144 were added. Subsequently, a standard N/10-HCl solution was added to the flask, and the free  
145 CaO content was determined based on the consumption rate of the solution.

146

## 147 **4. Results and discussion**

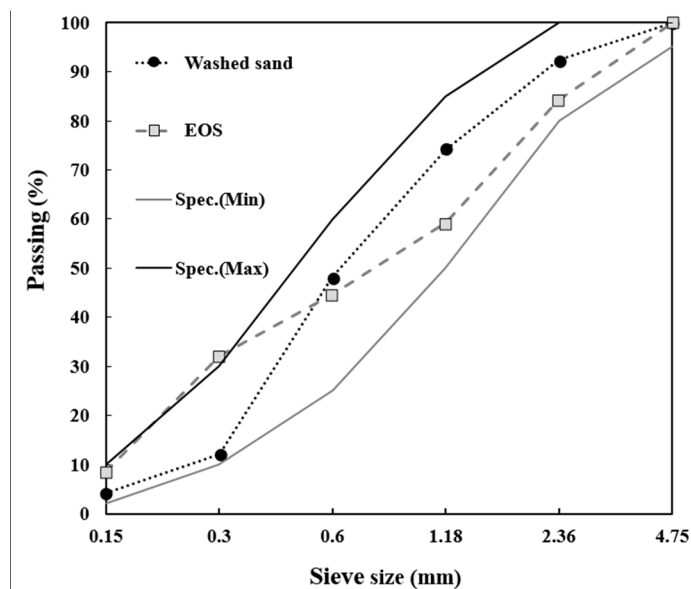
### 148 4.1 Evaluation of EOS properties

#### 149 *4.1.1 Physical properties and chemical composition*

150 Figure 1 shows the particle distribution curves of the EOS and washed sand samples used in  
151 the present work. Table 3 and Table 4 show the physical and chemical (XRF measurement  
152 result) properties of the materials used, respectively. The FMs (Fineness Modulus) of the EOS  
153 and washed sand samples were evaluated to 2.71 and 2.76, respectively, which shows similar  
154 level of FM. Their chemical compositions vary considerably. The main components of EOS  
155 are CaO (26.1%) and Fe<sub>2</sub>O<sub>3</sub> (36.8%) and its density is 3.58 kg/cm<sup>3</sup>. On the other hand, the  
156 primary component of the washed sand sample was SiO<sub>2</sub> (86.2%) with 2.59 kg/cm<sup>3</sup> of density.

157





158

159 Fig. 1 Grading curves for fine aggregates used in the present study.

160 Table 3. Physical properties of fine aggregates used in the present study

Parameter		Washed sand	EOS
Density (kg/cm <sup>3</sup> )	KS F 2504	2.59	3.58
Fineness modulus	KS F 2502	2.76	2.71
Absorption (%)	KS F 2504	1.56	1.72

161

162 Table 4. Chemical compositions of cement, EOS, and washed sand samples (%)

	CaO	SiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	MgO	Fe <sub>2</sub> O <sub>3</sub>	MnO	Other
Cement	62.6	21.9	4.8	2.6	3.4	-	4.7
EOS	26.1	15.5	11.9	3.4	36.8	6	0.3
Sand	0.5	86.2	5.8	0.2	0.5	-	6.8

163

164

#### 165 4.1.2 pH measurements

166 To determine the rate of change in pH with addition of EOS and washed sand, 10 g of each  
 167 material was mixed with 20 mL of distilled water, and a pH meter was used to measure the  
 168 initial pH value. The mixture was then stirred at 100 rpm at 20 °C and pH was measured after  
 169 30 minutes, 1 hour, 2 hour, and 3 hours. Figure 2 shows the stirrer used for the pH  
 170 measurements and the results of the pH measurements for the EOS and washed sand samples  
 171 are shown in Figure 3. It is observed that, in the EOS sample, the pH increases from 6.7 to 10.1  
 172 after 30 min. of stirring with almost constant pH thereafter. On the other hand, in the washed  
 173 sand, the pH increases slightly after 30 min. and keeps constant to 3 hours. EOS sample is  
 174 highly alkaline due to the presence of soluble calcium salts which can emit  $\text{Ca}^+$  ions in solution.

175



Fig. 2. Experimental setup with magnetic stirrer

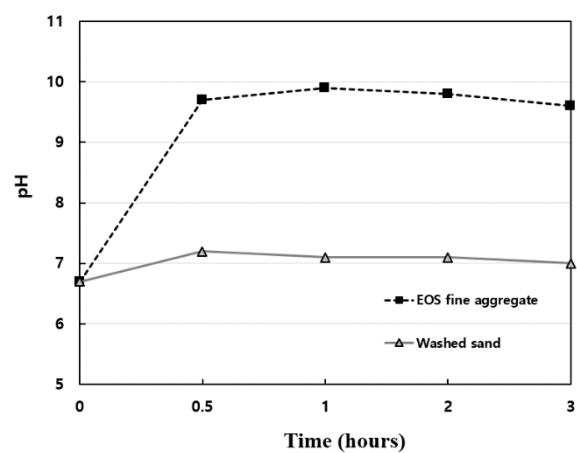


Fig. 3. Results of pH measurement

176

#### 177 4.1.3 XRD analysis results

178 Figure 4 shows the XRD analysis results for the EOS sample. It is found that the EOS sample

179 contains 30.2 % of monticellite ( $\text{MgCa}(\text{SiO}_2)$ ), 23.9% of magnesium iron aluminum oxide,  
 180 21.8% of wustite ( $\text{FeO}$ ), 18,2% of larnite ( $\text{Ca}_2(\text{SiO}_4)$ ), 3.3% of quartz ( $\text{SiO}_2$ ), and 2.5% of  
 181 hercynite ( $\text{AlFeO}_4$ ). Its main components are  $\text{CaO}$ ,  $\text{SiO}_2$ , and  $\text{FeO}$ , with a large amount of  
 182 larnite, which has the crystal structure of  $\beta\text{-C}_2\text{S}$ , the main mineral of cement, also being present.

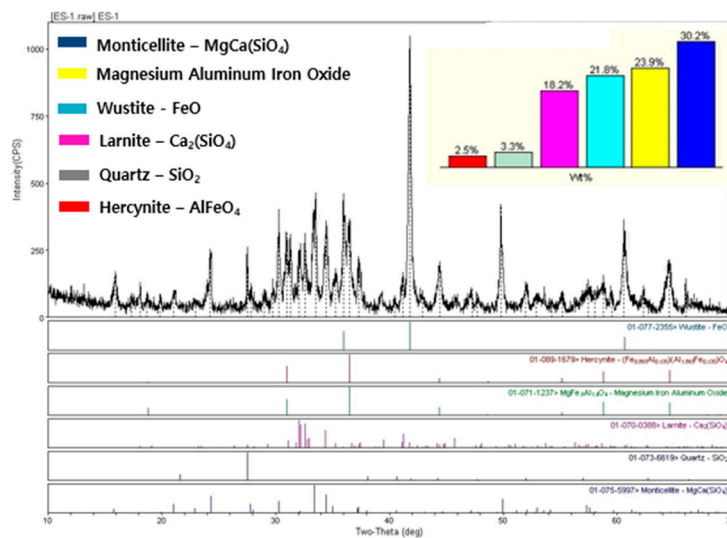


Fig. 4. Results of XRD analysis of EOS sample.

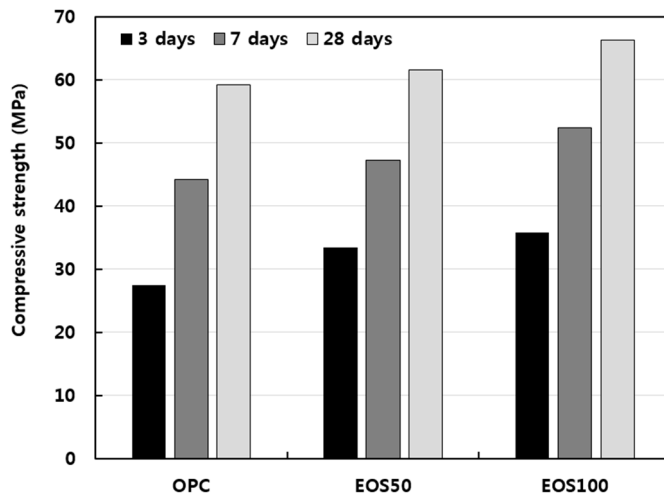
183

## 184 4.2 Evaluation of mortar with EOS

### 185 4.2.1 Compression strength

186 Figure 5 shows the compression strength test results of mortar with different EOS contents with  
 187 ages. As the amount of EOS increases, the compression strength also increases. The strength  
 188 of OPC at 28 days is 59 MPa, while that of EOS50 is 62 MPa and that of EOS100 is 66 MPa.  
 189 It shows that the strength of EOS50 is approximately 5 MPa higher than that of OPC at 3 days,  
 190 3 MPa higher at 7 days, and 2 MPa higher at 28 days. In the comparison with EOS100, the  
 191 strength of EOS100 was 8 MPa higher than that of OPC at 3 and 7 days, and 7 MPa higher at

192 28 days.

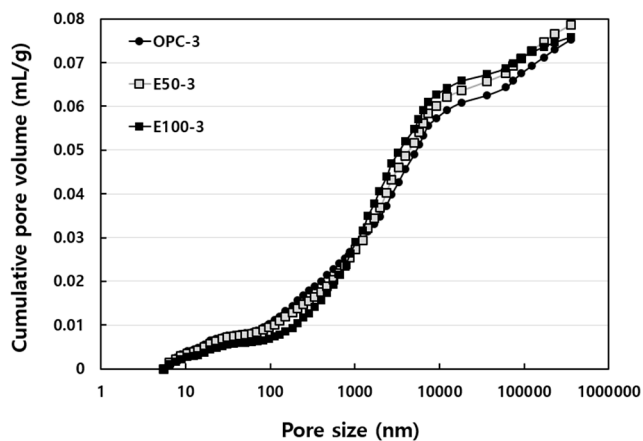


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194 Fig. 5 Compressive strengths of mortar samples.

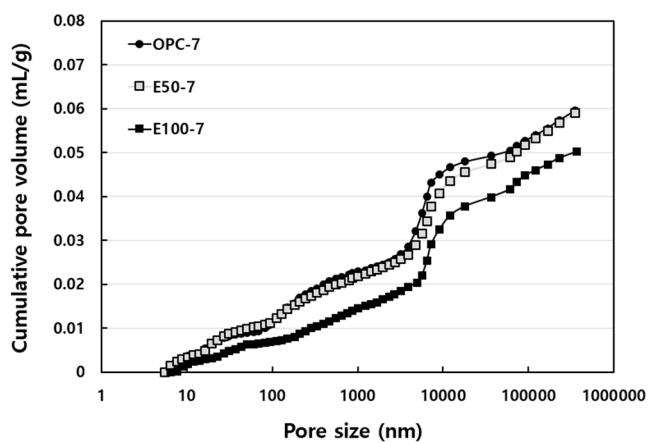
#### 195 4.2.2 MIP measurements

196 Figures 6, 7, and 8 show the cumulative pore volumes of the mortar samples with ages. Usually  
197 the cumulative pore volume decreases with aging due to hydration. EOS50 has a higher pore  
198 volume at 3 days than OPC, however, it was lower than that of OPC after 7 days. EOS100  
199 exhibits a low cumulative pore volume overall, which shows porosity decreases with an  
200 increase in the EOS content. Figure 9 compares the cumulative pore volume and compression  
201 strength in mortar samples. The decrease in the cumulative pore volume is evaluated with  
202 increasing compression strength, which is typical relationship between porosity and strength  
203 development.



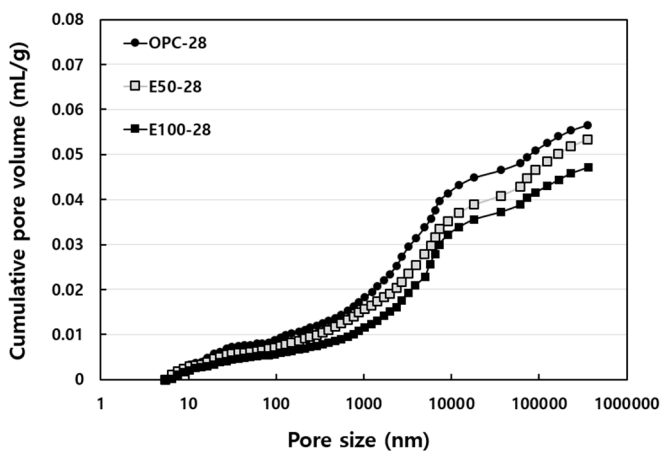
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205 Fig 6. MIP results at 3 days.



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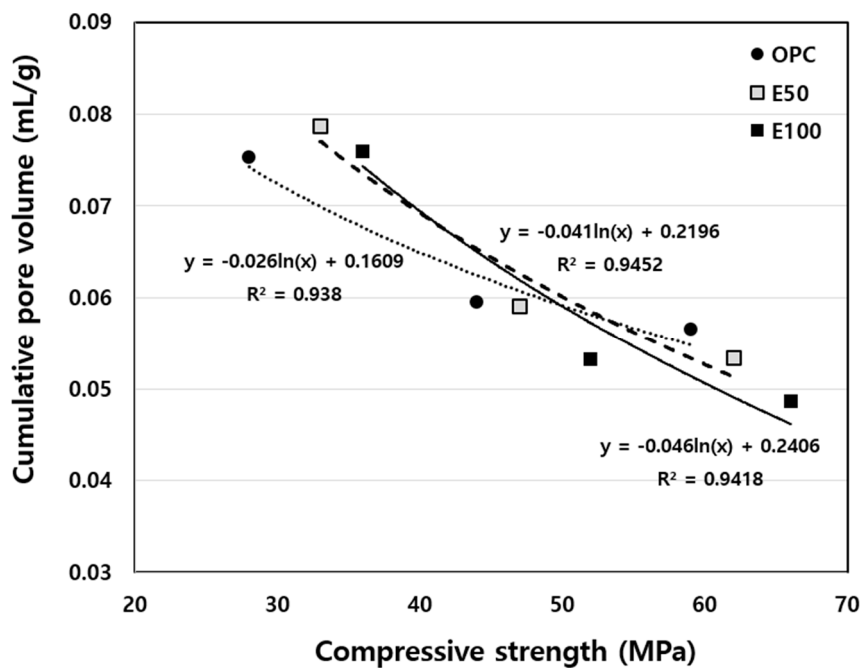
207 Fig 7. MIPs results at 7 days.



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209 Fig 8. MIP results at 28 days.

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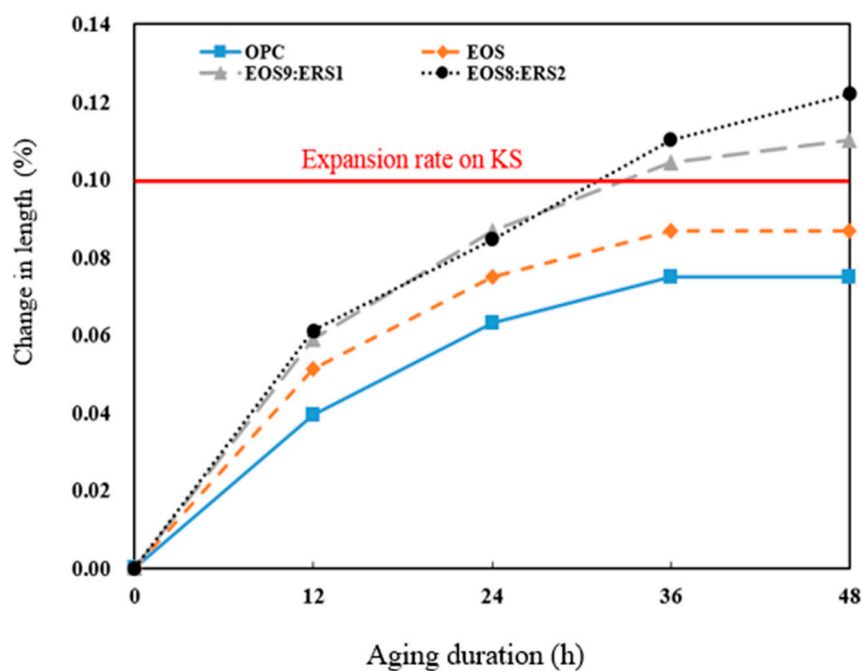
212 Fig. 9. Relationship between cumulative pore volume and compressive strength.

#### 213 4.2.3 Rate of length change

214 Figure 10 shows the results of the rate of length change measurements for mortar. The rate of  
 215 length change for OPC and EOS is measured to be less than 0.1%, which shows satisfactory  
 216 results with the requirements in the KS Specification. While the rates of length change for  
 217 EOS9/ERS1 and EOS8/ERS2 at 24 hours satisfy the standard, those after 36 hours are greater  
 218 than 0.1% of expansion. The expansion increases with increasing ERS amount. It is thought  
 219 that the excessive expansion is attributable to the free CaO present in ERS. In Table 5, the  
 220 basicity ( $\text{CaO}/\text{SiO}_2$ ) and free CaO content are summarized. The amount of free CaO in EOS is  
 221 0.491%, while it increases to 0.560% and 0.700% in EOS9/ERS1 and EOS8/ERS2,  
 222 respectively. The basicity is approximately 3.08 for ERS, 2.08 for EOS9/ERS1, and 2.36 for

223 EOS8/ERS2. In the mixed conditions (EOS9:ERS1, EOS8:ERS2), the basicity results does not  
 224 meet with the requirement in KS standard which is below 2.0. The free CaO content which  
 225 causes swelling has a critical effect on the rate of length change.

226



227

228 Fig. 10. Results of rate of length change measurements for mortar when using autoclave.

229

230 Table 5. Results of basicity and free CaO content measurements.

Sample	CaO/SiO <sub>2</sub>	Free CaO
EOS	1.68	0.491
ERS	3.08	1.738
EOS9/ERS1	2.08	0.560
EOS8/ERS2	2.36	0.700

231

## 232 5. Conclusions

233 In the work, the engineering performance in EOS and cement mortar containing EOS is  
234 evaluated for utilization as fine aggregate. The conclusions on the study can be summarized as  
235 follows:

- 236 1. The XRD analysis of EOS sample shows that its primary components are CaO, SiO<sub>2</sub>,  
237 and FeO. It also contains 18.2% larnite, which has a crystal structure of  $\beta$ -C<sub>2</sub>S with  
238 similar mineral of cement. Moreover, pH measurement shows that EOS sample has  
239 high-alkalinity. This provides a soluble calcium salts causing a large number of Ca<sup>+</sup> ions  
240 given the chemical properties of the EOS.
- 241 2. For cement mortar with EOS, the tests on compressive strength and porosity are  
242 performed. It reveals that cumulative pore volume decreases and the compressive  
243 strength increases as more addition of EOS as the fine aggregate. The changes in the  
244 compression strength and cumulative pore volume vary with ages. EOS contains a  
245 greater proportion of fine particles than sand, which can be the reason for the lower  
246 porosity due to packing effect. The change in the compressive strength with ages in the  
247 EOS mortar is much affected by the large amount of larnite ( $\beta$ -C<sub>2</sub>S) present in the EOS.
- 248 3. The mortars with EOS and washed sand as their fine aggregates satisfy the rate of  
249 length change requirement below 0.1% in the KS Specification. However, all the  
250 results from the mortar containing ERS show higher changing rate over 0.1% after 36  
251 hours. The expansion of the samples is attributed to the free CaO present in ERS. The  
252 free CaO contents of the EOS9/ERS1 and EOS8/ERS2 mortars are 0.56% and 0.70%  
253 respectively, while their basicity values (CaO/SiO<sub>2</sub>) are 2.09 and 2.36, respectively.  
254 The values fall outside 2.0 of the acceptable limits in the KS specifications. The results



255 suggest that more consideration must be taken for the usage of mixed ERS and EOS as  
256 a fine aggregate.

257

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