

Review

# Advances in Fluorescent Single-Chain Nanoparticles

Julen De-La-Cuesta<sup>1,2</sup>, Edurne González<sup>1</sup> and José A. Pomposo<sup>1,2,3,\*</sup>

<sup>1</sup> Centro de Física de Materiales (CSIC, UPV/EHU) - MPC, Materials Physics Center, Paseo Manuel de Lardizabal 5, E-20018 San Sebastian, Spain; [julen.delacuesta@ehu.es](mailto:julen.delacuesta@ehu.es) (J. D.-L.-C.); [edurne.gonzalez001@ehu.es](mailto:edurne.gonzalez001@ehu.es) (E. G.)

<sup>2</sup> Departamento de Física de Materiales, Universidad del País Vasco (UPV/EHU), Apartado 1072, E-20080 San Sebastian, Spain

<sup>3</sup> IKERBASQUE - Basque Foundation for Science, María Díaz de Haro 3, E-48013 Bilbao, Spain

\* Correspondence: [joseito.pomposo@ehu.es](mailto:joseito.pomposo@ehu.es); Tel.: +34-943-018-801; Fax: +34-943-015-800

**Abstract:** Fluorophore molecules can be monitored by fluorescence spectroscopy and microscopy which are highly useful and widely used techniques in cell biology, biochemistry and medicine (*e.g.*, biomarker analysis, immunoassays, cancer diagnosis). Several fluorescent micro- and nanoparticle systems based on block copolymer micelles and cross-linked polymer networks, quantum dots,  $\pi$ -conjugated polymers, and dendrimers have been evaluated as optical imaging systems. In this review, we highlight recent advances in the construction of fluorescent single-chain nanoparticles (SCNPs) which are valuable artificial soft nano-objects with tunable, small size (as small as 3 nm). In particular, the main methods currently available to endow SCNPs with fluorescent properties are discussed in detail, showing illustrative examples.

**Keywords:** Nanoparticles, fluorescence, optical imaging.

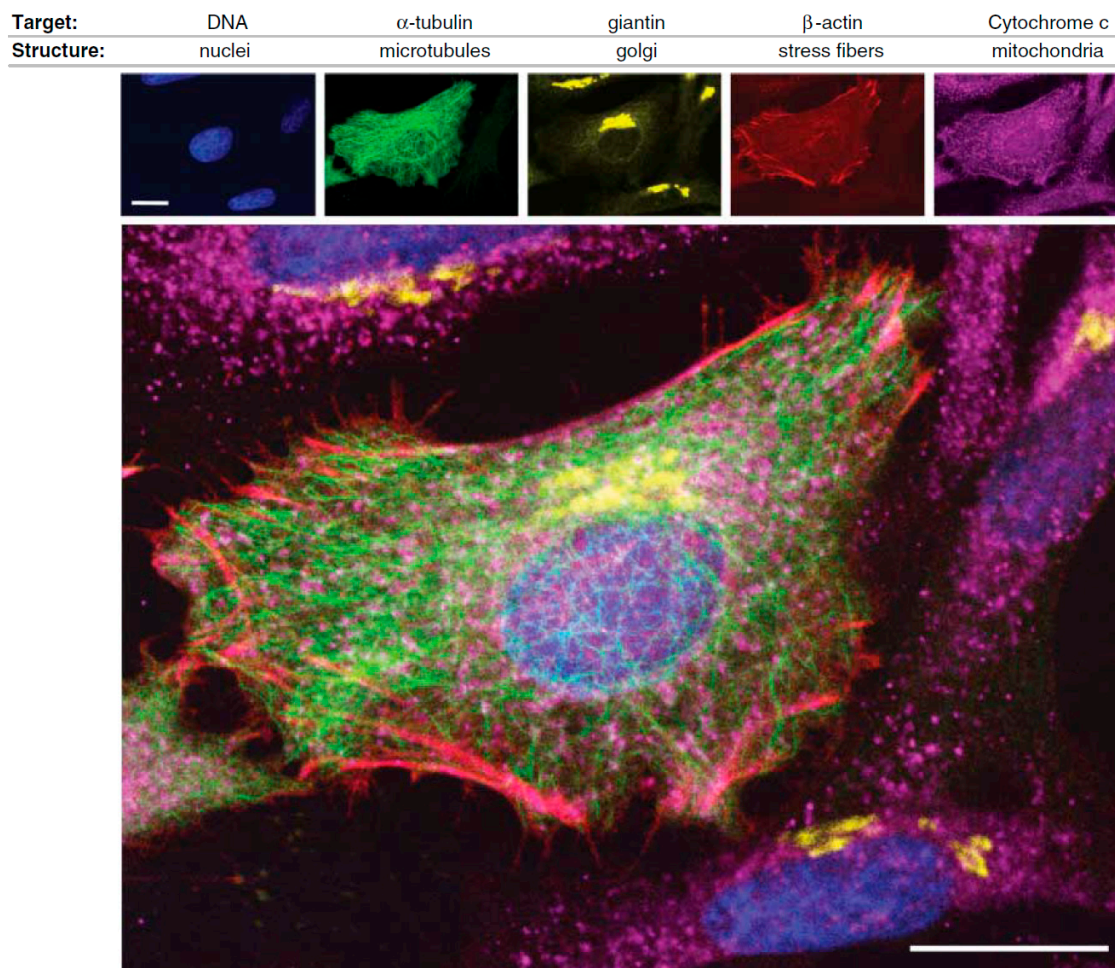
## 1. Introduction

Fluorescence has long been used in biomarker analysis, immunoassays and diagnostic imaging, including cancer diagnosis [1]. Many small organic fluorescent molecules (< 1 kD in molecular weight) which are currently commercially available suffer from several limitations concerning wavelength range, brightness (*i.e.*, extinction coefficient for absorbance  $\times$  quantum yield of fluorescence), photobleaching (*i.e.*, photostability) and fluorescence self-quenching. Also, the toxicity exhibited by some promising organic fluorophore molecules has limited their application when intended for *in vitro* and *in vivo* optical imaging.

Fluorescent micro- and nanoparticle systems have been proposed as a solution to the above mentioned limitations of small organic fluorescent dyes [2]. When compared to the low absorption coefficients of individual organic fluorophore molecules, fluorescent micro- and nanoparticles often contain multiple fluorophore entities leading to increased photoluminescent emission. Additionally, their encapsulation into the particle provides with improved stability, and often both reduced photobleaching and toxicity. Also, improved biocompatibility can be obtained by proper selection of the nature of the particle. Moreover, a unique feature of micro- and nanoparticles is the possibility to combine imaging and drug delivery characteristics in the same system to produce theranostic agents for nanomedicine applications [3]. Several fluorescent micro- and nanoparticle systems based on block copolymer micelles and cross-linked polymer networks, quantum dots (QDs),  $\pi$ -conjugated polymers, and dendrimers have been evaluated as optical imaging systems. The polymerization of monomers containing fluorescent moieties has been employed to prepare amphiphilic block copolymers (BCPs) which self-assemble in water to produce fluorescent micelles [4]. Also, decoration of non-fluorescent BCPs with fluorescent dyes has been carried out to produce red, green and blue photoluminescent (PL) emission *via* crystallization-driven self-assembly, allowing even white light PL emission [5]. Fluorescent cross-linked polymer networks have been typically prepared through direct free radical cross-linking oil-in-water emulsion polymerization [6]. QDs which are inorganic nanocrystals with excellent fluorescence properties (extinction coefficients  $\times 10$  to  $\times 100$  of those of

organic fluorophore dyes) have been fabricated by a variety of methods [7]. Also different fluorescent  $\pi$ -conjugated polymers [8] and dendrimers [9] have been synthesized and evaluated as optical imaging systems.

A nice example of the current fluorescent toolbox available for assessing protein location and function in cell biology is depicted in Figure 1, in which a variety of targeting methods and fluorophores (organic dyes, QDs, green fluorescent protein) were combined to visualize different cell structures of *HeLa* cancer cells [10].

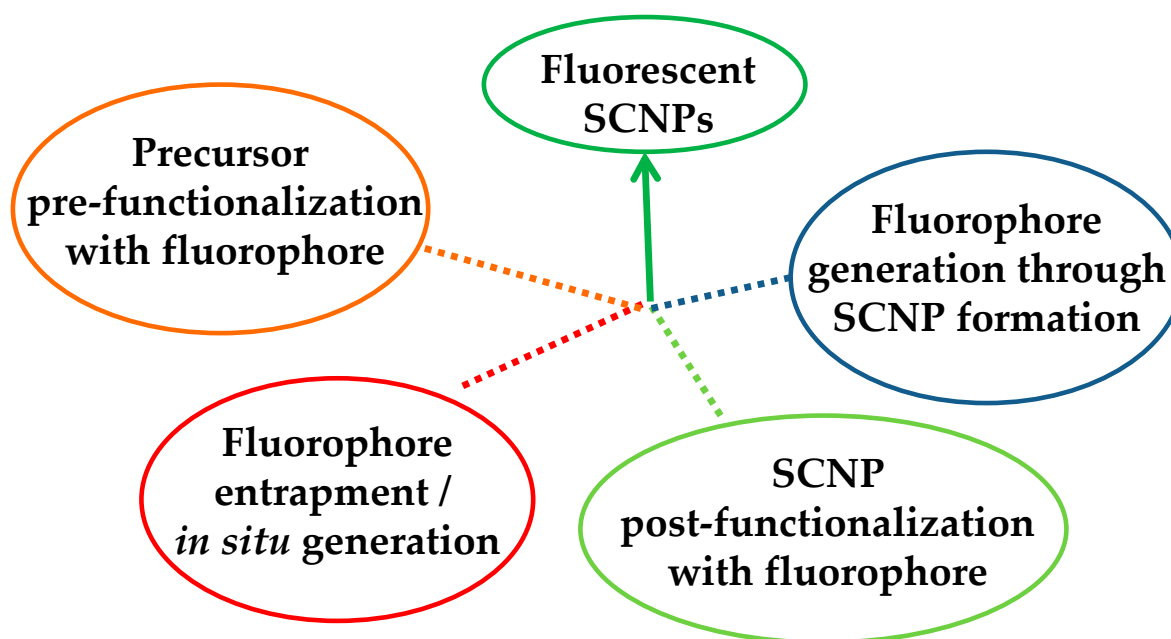


**Figure 1.** Illustration of the application of complementary fluorophores for the selective detection of different targets in *HeLa* cancer cells: DNA of nuclei (blue color),  $\alpha$ -tubulin of microtubules (green color), giantin of Golgi bodies (yellow color),  $\beta$ -actin of stress fibers (red color), and Cytochrome c of mitochondria (purple color). Scale bars: 20  $\mu$ m (reprinted from [10] with permission, Copyright American Association for the Advancement of Science, 2006).

In spite of the successful use of fluorescent micro- and nanoparticles in some optical imaging applications certain problems remain to be solved since the large size of these particles (often > 10 nm) prevents efficient traversal of intact membranes in cells. Additionally, an appropriate tuning of the ultraviolet-visible (UV-vis) absorption and photoluminescence emission wavelengths is sometimes difficult or even impossible. Moreover, *in vivo* accumulation of large particles in the body is a real problem for some of these systems.

In recent years, a new type of polymer nanoparticles of very small size (as small as 3 nm) has been developed [11–30] and several methods to endow these single-chain soft nano-objects so-called single-chain polymer nanoparticles (SCNPs) with fluorescent characteristics have been reported [11]. SCNPs are prepared through folding/collapse of individual polymer chains by means of intramolecular cross-linking driven by covalent bonds or reversible interactions [19–27]. The

molecular weight of the SCNP precursor polymer and its functionalization degree are essential parameters that control SCNP size, in addition to the nature of the interactions employed to perform the folding/collapse as well as solvent quality (good solvent, selective solvent) [12,13]. In this sense, the folding of a linear synthetic polymer to a collapsed state provides with one (or more) denser local packaging zone(s) where fluorophore molecules can be efficiently accommodated (see below for details) [14]. Concerning the morphology of SCNPs in solution, two limiting conformations can be obtained by current synthetic methods: a sparse morphology resembling that typical of intrinsically disordered proteins (IDPs) and a globular morphology as often found in enzymes [16].



**Figure 2.** Different strategies developed to endow SCNPs with fluorescent properties.

Four different ways have been opened to endow SCNPs with fluorescence properties (Figure 2): i) “precursor pre-functionalization with fluorophore”, *i.e.* functionalization of the SCNP precursor polymer with fluorophore molecules before intramolecular cross-linking, ii) “fluorophore entrapment / *in situ* generation”, *i.e.* entrapment of external fluorophore molecules into non-fluorescent SCNPs by taking advantage of the denser local packaging zone(s) of the SCNPs or *in situ* generation of the fluorophore molecule inside the SCNP, iii) “SCNP post-functionalization with fluorophore”, *i.e.* post-functionalization of the SCNPs *via* chemical reaction with appropriate, complementary reactive fluorophore molecules, and iv) “fluorophore generation through SCNP formation”, *i.e.* generation of fluorophore functional groups through intramolecular cross-linking.

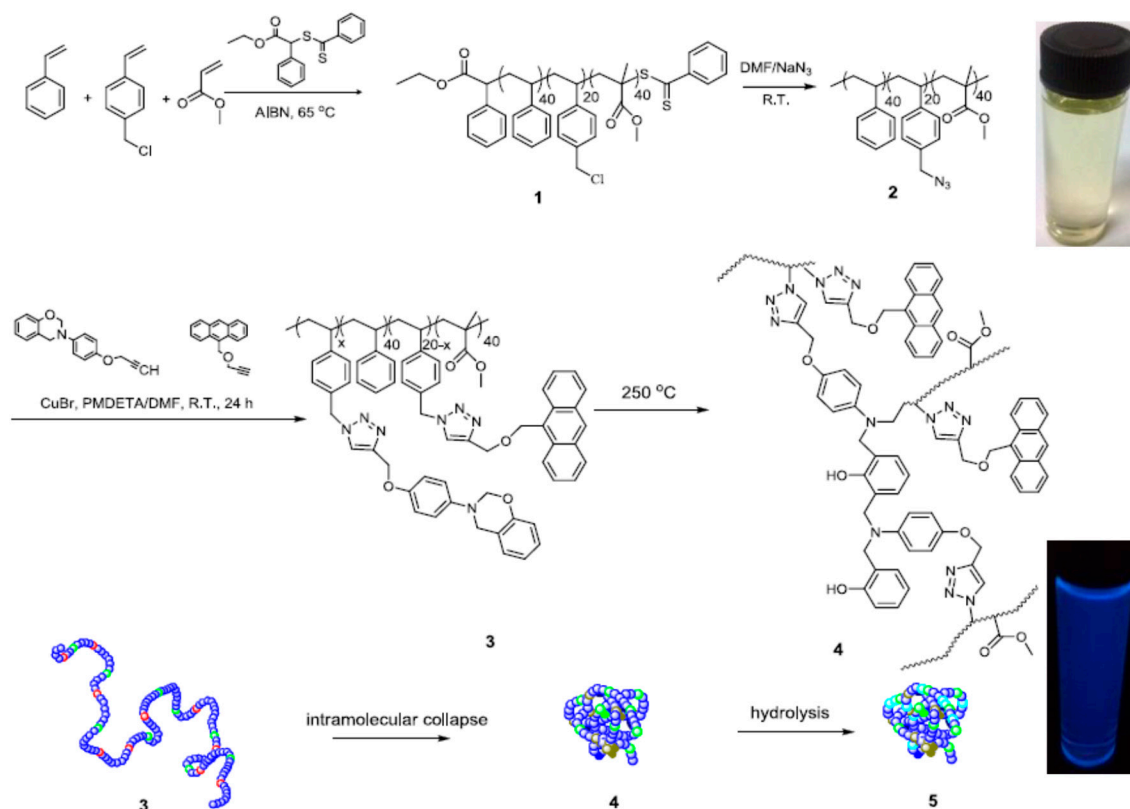
The present review summarizes the recent advances performed in last years for the construction of fluorescent SCNPs through the above methods, showing illustrative examples.

## 2. Fluorescent single-chain nanoparticles: synthesis routes

When compared to the development of other fluorophore micro- and nano-particle systems such as block copolymer micelles and cross-linked polymer networks [4-6], quantum dots [7],  $\pi$ -conjugated polymers [8] as well as dendrimers [9], the preparation of fluorescent SCNPs is still in its early infancy [11]. Nevertheless, four different routes have been established to endow SCNPs with fluorescent properties, paving the way to the potential construction of new fluorescent probes with ultra-small size (*e.g.*, 3 nm in diameter), higher brightness and better photostability than previous particle-based systems.

### 2.1. Precursor pre-functionalization with fluorophore

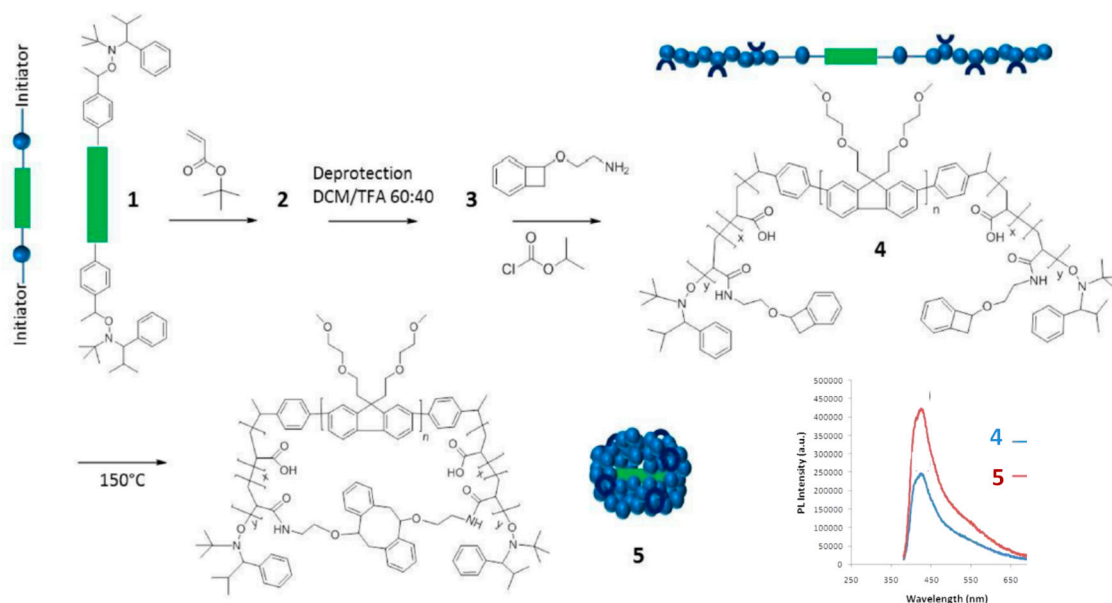
The synthesis of fluorescent SCNPs through functionalization of the linear precursor polymer with a fluorophore moiety is shown with an illustrative example in Figure 3. In this work by Wang *et al.* [31] a linear polystyrene (PS)-based terpolymer decorated with azide ( $-N_3$ ) functional groups (**2**) was first prepared from a PS-based terpolymer containing chloromethyl functional groups (**1**). Nucleophilic substitution of  $-Cl$  groups by  $-N_3$  moieties was carried out in dimethylformamide (DMF) at r.t. Subsequently, **2** was decorated with reactive and fluorescent groups (benzoxazine and



**Figure 3.** Synthesis of fluorescent SCNPs (**4**) *via* functionalization of the SCNP precursor terpolymer **2** with reactive (benzoxazine) and fluorescent (anthracene) groups (**3**) followed by intramolecular ring-opening polymerization of the benzoxazine moieties. Hydrolysis of **4** leads directly to water-soluble, fluorescent SCNPs (**5**) with a maximum emission peak centered at 412 nm when irradiated at  $\lambda = 367$  nm (reprinted from [31] with permission, Copyright Elsevier, 2014).

anthracene moieties, respectively) *via* “click chemistry” to give the functionalized precursor polymer **3**. Intramolecular cross-linking of **3** *via* benzoxazine ring-opening polymerization (ROP) was carried out at high dilution in dibenzyl ether at 250 °C to give SCNP **4**. A further hydrolysis step under acidic conditions (pH = 3 - 4) was required to render water-soluble the resulting SCNPs (**5**), which were found to show fluorescence with a maximum emission peak centered at 412 nm when irradiated at  $\lambda = 367$  nm (see Figure 3). The successful formation of SCNPs with an average hydrodynamic radius of 6 nm was confirmed by a combination of characterization techniques, including size exclusion chromatography (SEC),  $^1H$  NMR and FTIR spectroscopy, dynamic light scattering (DLS) as well as atomic force microscopy (AFM) and transmission electron microscopy (TEM) measurements.





**Figure 4.** Synthesis of water-soluble fluorescent SCNPs (5) from a precursor polymer (4) containing a polyfluorene block (depicted in green) *via* intramolecular single-chain folding/collapse at 150 °C. The values of photoluminescence (PL) quantum efficiency of 4 and 5 (both with maximum PL intensity at 438 nm) were 2.8% and 5.1%, respectively (reprinted from [32] with permission, Copyright American Chemical Society, 2013).

Another example of the “precursor pre-functionalization with fluorophore” method is provided in Figure 4. In this work by Adkins *et al.* [32], the SCNP precursor polymer 4 was an ABA-type triblock copolymer in which the center block (B) was an ethylene oxide modified polyfluorene segment, and the lateral blocks (A) were polyacrylate segments decorated with low-temperature benzocyclobutene cross-linking groups. Interestingly, the A blocks were synthesized through nitroxide-mediated living radical polymerization by using the fluorescent B block as a telechelic macroinitiator. Characterization of 4 by UV-vis and fluorescence spectroscopy revealed absorption and emission peaks at 370 and 438 nm, respectively, with a quantum efficiency of 2.8%. Upon folding/collapse of 4 *via* benzocyclobutene dimerization at high dilution, the resulting SCNPs 5 were found to display UV-vis absorption and fluorescence emission peaks also centered at 370 and 438 nm but with an increased quantum efficiency of 5.1%. This was attributed to the site-isolation effect (*i.e.*, encapsulation) experienced by the fluorescent B block upon intramolecular cross-linking of the A blocks. A similar effect was previously found by Adkins *et al.* [33] for ABA triblock copolymers containing fluorene and fluorene/thiophene center blocks which were soluble only in organic solvents (dichloromethane and cyclohexane). Remarkably, the length of the A block was found to have a significant effect in promoting this site-isolation effect.

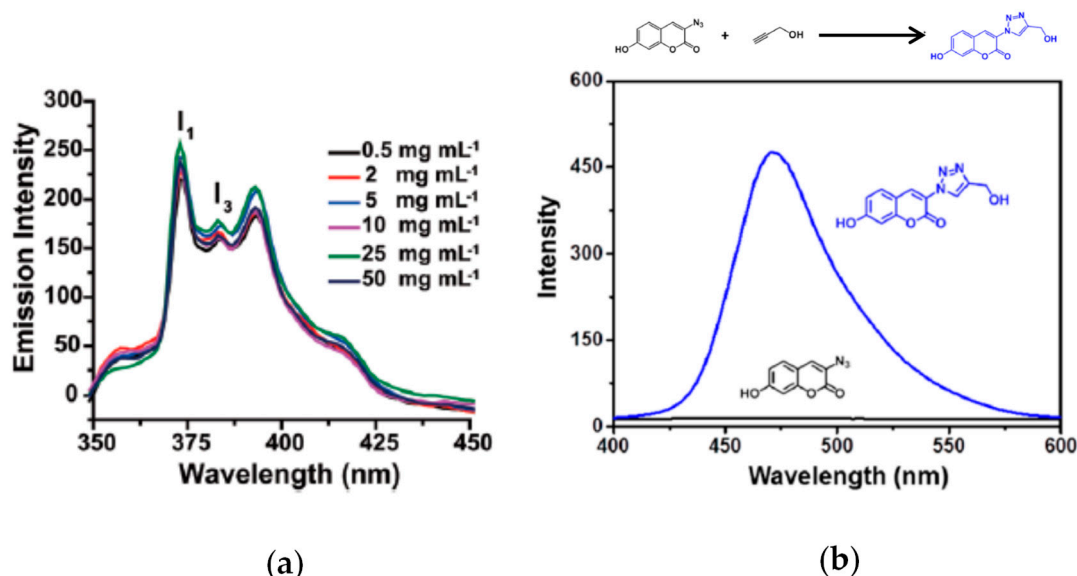
The technique of “precursor pre-functionalization with fluorophore” has been also employed by Bai *et al.* [34] to prepare organic- and aqueous-soluble fluorescent SCNPs with a diameter as small as 5 nm using a fluorescein-functionalized monomer. The SCNPs were prepared by combining the techniques of ring-opening metathesis polymerization (ROMP) and ring-closing metathesis (RCM). Remarkably, these SCNPs were brighter than fluorescein at a similar concentration, and they showed exceptional stability during photobleaching experiments using 470 nm LED in phosphate buffer at pH = 7.4.

Recently, Blasco *et al.* [35] reported on the synthesis of fluorescent SCNPs based on polythiophene precursor polymers containing photoresponsive groups. The precursor polymer showed UV-vis absorption and fluorescence emission peaks at 450 and 585 nm, whereas the SCNPs showed absorption and emission peaks at 435 and 580 nm, respectively. The shift in the position of the absorption and emission peaks, as well as the observed reduction in emission intensity (*ca.* 50%), was attributed to a change in the conformation of the rigid polythiophene chain upon folding/collapse to nanoparticle.

One clear limitation of this synthesis route to fluorescent SCNPs is the increased possibility of fluorophore-fluorophore interactions upon folding/collapse, leading to fluorescence self-quenching.

## 2.2. Fluorophore entrapment / *in situ* generation

The use of SCNPs as nanocontainers for fluorophore entrapment is another method to endow SCNPs with fluorescent properties. Sometimes, this method has been used to determine the polarity of the dense local packaging zone(s) of the SCNPs upon folding/collapse. As an example, Song *et al.* [36] applied a pyrene probe to investigate the hydrophobic interior of water-soluble SCNPs prepared through disulfide-based intrachain cross-linking (Figure 5a). It is well-known that pyrene fluorescence depends strongly on the surrounding environment, so the ratio of the intensities between the first and the third peaks ( $I_1/I_3$ ) in the pyrene emission spectrum can range from 1.9 in polar environments to 0.6 in apolar ones (Figure 5a). The disulfide-cross-linked SCNPs showed a ratio  $I_1/I_3 = 1.4$ , pointing to a moderately hydrophobic interior of the nanoparticle.



**Figure 5.** Emission spectra of pyrene sequestered into non-fluorescent SCNPs as a function of the concentration of SCNPs (a), and synthesis of a fluorescent compound into non-fluorescent Cu(I)-containing SCNPs *via* Cu(I)-catalyzed azide-alkyne cycloaddition of 3-azido-7-hydroxycoumarin and propargyl alcohol (b) (reprinted with permission from [36], Copyright Royal Society of Chemistry, 2015 and [39], Copyright American Chemical Society, 2015, respectively).

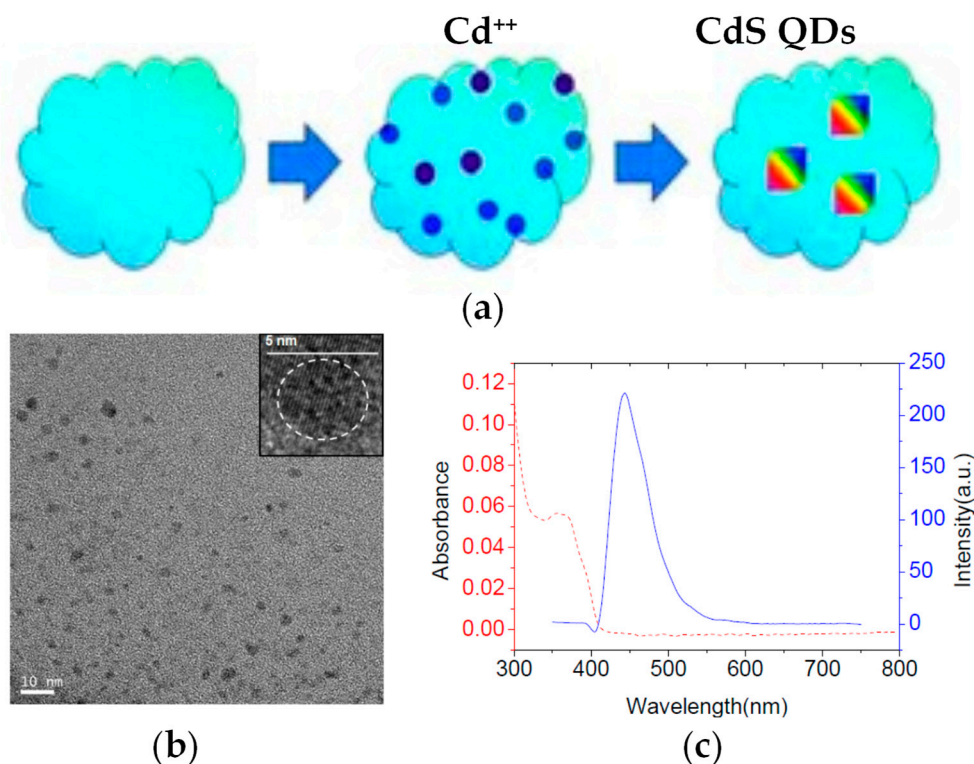
Also pyrene was used by Akagi *et al.* [37] as a fluorescent probe to quantify the hydrophobicity of SCNPs based on a hydrophilic poly( $\gamma$ -glutamic acid) polymer decorated with hydrophobic L-Phenylalanine (Phe) motifs. SCNPs with high contents of Phe pendants (35 and 42 mol %) were found to display hydrophobic nanodomains in aqueous solution, in which pyrene was solubilized. The same group reported the application of dipyrrene to evaluate the rigidity of this kind of SCNPs by fluorescence spectroscopy, recording the fluorescent intensity ratio between the excimer complex (which is present only in low rigidity environments) and the monomer [38]. The rigidity of the inner SCNP structure increased upon reduction in nanoparticle size, and through lyophilization of the samples which promotes a better close-packing by removing all water molecules from the internal structures.

*In situ* generation of fluorophore molecules inside SCNPs is another way to endow SCNPs with fluorescent properties, as demonstrated by Liu *et al.* [39] in a recent work. Amphiphilic SCNPs were prepared by these authors by post-polymerization functionalization of poly(pentafluorophenyl acrylate) as SCNP precursor polymer. Ligand-containing SCNPs capable of coordinating to Cu(I) and Pd(II) were prepared which accelerated azide-alkyne cycloaddition reactions and catalyzed depargylation reactions, respectively. As a model reaction of fluorophore molecule generation

inside SCNPs, 3-azido-7-hydroxycoumarin and propargyl alcohol were chosen as substrates for the Cu(I)-catalyzed azide-alkyne cycloaddition (CuAAC) reaction. The azidocoumarin molecule is non-fluorescent due to emission quenching by the lone pair of electrons from the azido moiety. However, after forming the triazole ring by CuAAC reaction with propargyl alcohol, the lone pair of electrons becomes localized activating the fluorescence (Figure 5b). Fluorescent spectroscopy was used to investigate the kinetics of the CuAAC reaction with different Cu(I)-containing SCNPs under dilute substrate conditions. With the most effective SCNPs, the fluorescence intensity reached a plateau corresponding to full conversion of azidocoumarin to triazole coumarin within less than 10 min. Interestingly, these reactions were found to proceed in phosphate buffer at physiological pH and at low substrate concentrations, which render this kind of SCNPs promising systems to function in complex media such as cellular environments.

Independently, Bai *et al.* [40] developed Cu-containing SCNPs as highly-efficient *in vivo* catalyst to endow cells with fluorescence by using 3-azido-7-hydroxycoumarin and 7-ethynylcoumarin as fluorogenic coumarin derivatives that “light up” upon CuAAC reaction with *p*-ethynylanisole and 2-picolylazide, respectively. These SCNPs showed the ability to enter cells and perform efficient, biocompatible click chemistry *in vivo* thus acting as intracellular nanoscale molecular synthesizers of fluorophore molecules. In fact, the conversions obtained were typically >90% in 1 h at r.t. at a concentration of SCNPs of only 1  $\mu$ M.

The application of non-fluorescent SCNPs as efficient nanoreactors for the synthesis of fluorescent QDs *in situ* was pioneered by Qian *et al.* [41] (Figure 6a). Hydrophilic SCNPs synthesized through Bergman cyclization-mediated intramolecular folding/collapse followed by hydrogenolysis on Pd/C were utilized as size-tunable nanoreactors to fabricate and encapsulate QDs in a one-pot reaction. SCNPs were prepared from linear precursors of different size (number-average molecular weight ( $M_n$ ): 30.3, 62.5 and 115.2 kDa) and functionalization (10 and 25 mol% of reactive enediyne groups). SEC measurements confirmed the successful formation of SCNPs, whereas the 3D morphology of the hydrophilic SCNPs on a mica surface was visualized by AFM. As a proof of con-



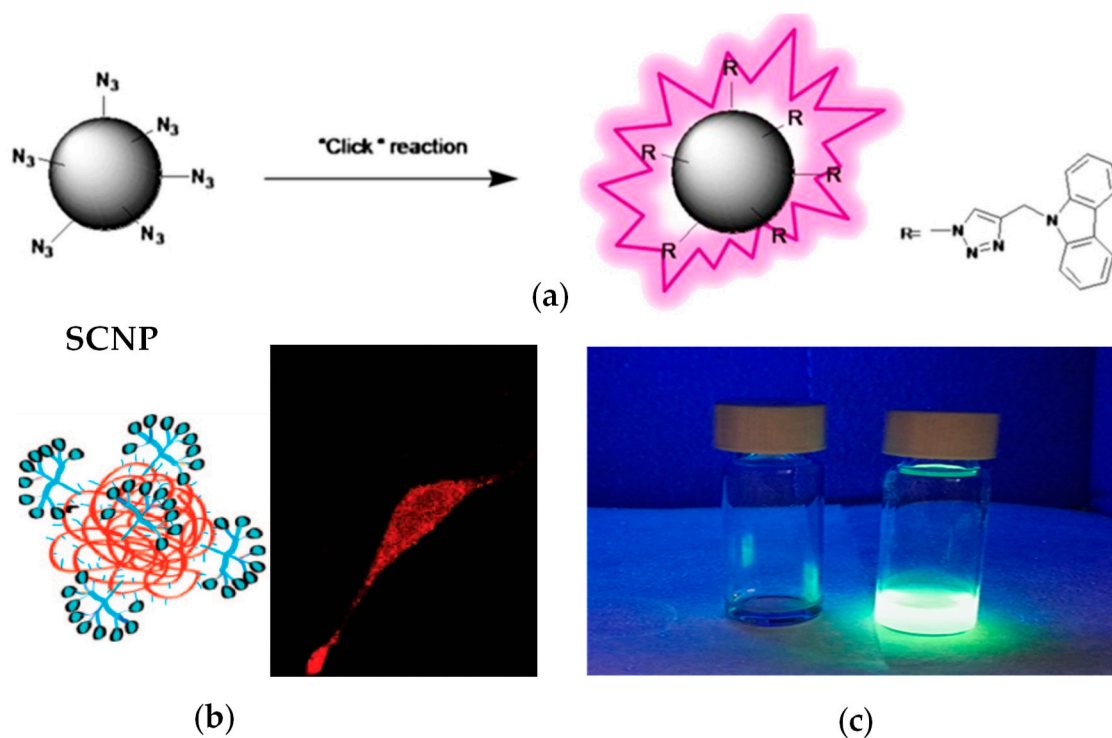
**Figure 6.** Schematic illustration of the generation of fluorescent CdS quantum dots (QDs) inside non-fluorescent SCNPs (a), TEM image of the resulting CdS QDs showing an average size of 4.7 nm (b), and photoluminescent emission spectrum (blue) of the CdS QDs displaying bright fluorescence centered at 450 nm (c) (reprinted from [41] with permission, Copyright Wiley-VCH, 2012).

cept, photoluminescent zinc sulfide (ZnS) QDs were fabricated and encapsulated in SCNPs of different size. Smaller nanoreactors produce a single QD each (4.1 nm in size) giving to higher emission intensity (quantum yield, QY = 17 % at  $\lambda = 310$  nm) while larger nanoreactors form multiple QDs each, resulting in fluorescence quenching (QY = 2 %). To further investigate the generality of the method, cadmium sulfide (CdS) QDs (4.7 nm in size, Figure 6b) were synthesized and encapsulated in the smaller nanoreactors showing bright fluorescence centered at 450 nm (Figure 6c) and excellent values of quantum yield (QY = 45 %). Moreover, the use of SCNPs as sacrificial nanoreactors for the synthesis of bright photoluminescent carbon nanodots was further reported by the same group [42].

While a permanent entrapment of QDs inside SCNPs is expected due to the relatively large size of these inorganic nanoparticles, one clear disadvantage of physical entrapment of low molecular weight fluorophore molecules into SCNPs is its reversibility.

### 2.3. SCNPs post-functionalization with fluorophore

One strategy to avoid the leaching of entrapped fluorophore molecules from SCNPs is the post-functionalization of the SCNPs *via* chemical reaction with appropriate, complementary reactive fluorophore molecules, as depicted in Figure 7 [43-45]. The first work illustrating a successful post-



**Figure 7.** Schematic illustration of the post-functionalization of a SCNPs with a fluorophore molecule (a), and examples of fluorescent SCNPs decorated with: (b) Alexa Fluor 568 dye, and internalized by a fibroblast cell as recorded by confocal fluorescent microscopy, and (c) dansylhydrazine (reprinted with permission from [43], Copyright Elsevier, 2014; [44], Copyright American Chemical Society, 2009; and [45], Copyright Royal Society of Chemistry, 2016, respectively).

functionalization of SCNPs was reported by Jiang and Thayumanavan by using amine-containing SCNPs that were efficiently decorated with pivaloyl groups (> 90 mol %) [46]. Post-functionalization of azide-containing SCNPs *via* the highly-efficient CuAAC reaction was subsequently introduced by Ruiz de Luzuriaga *et al.* [47]. This approach was followed by Li *et al.* [43] to react azide-containing SCNPs of 8 nm in size with *N*-propargyl carbazole, rendering fluorescent the initially non-fluorescent SCNPs (Figure 7a). After post-functionalization, the SCNPs showed UV-vis absorption and fluorescence emission peaks at 295 and 365 nm, respectively.



The strategy of SCNP post-functionalization *via* chemical reaction with reactive fluorophore molecules was exploited by Hamilton and Harth [44] in a seminal work to provide with molecular dendritic transporter nanoparticle vectors for efficient intracellular delivery of peptides. In this work, SCNPs were conjugated with both molecular dendritic transporter units to promote intracellular uptake by fibroblast cells, and peptides with cleavable disulfide linkers. Two commercially available fluorescent compounds (Alexa Fluor 568 dye and fluorescein) were used to selectively label the SCNP backbone and the conjugated peptides. Successful uptake and transport of the labeled SCNPs across the cellular membrane was observed by confocal fluorescence microscopy (Figure 7b).

Recently, a variety of reactive fluorophore molecules (dansylhydrazine [45], fluorescein [48], naphthalene [49]) have been successfully conjugated to different (polystyrene [45], polydimethylamino ethyl methacrylate [48], polymethyl methacrylate [49]) nanoparticles by post-functionalization (Figure 7c). This illustrates the versatility of the SCNP post-functionalization technique to endow SCNPs with fluorescent properties.

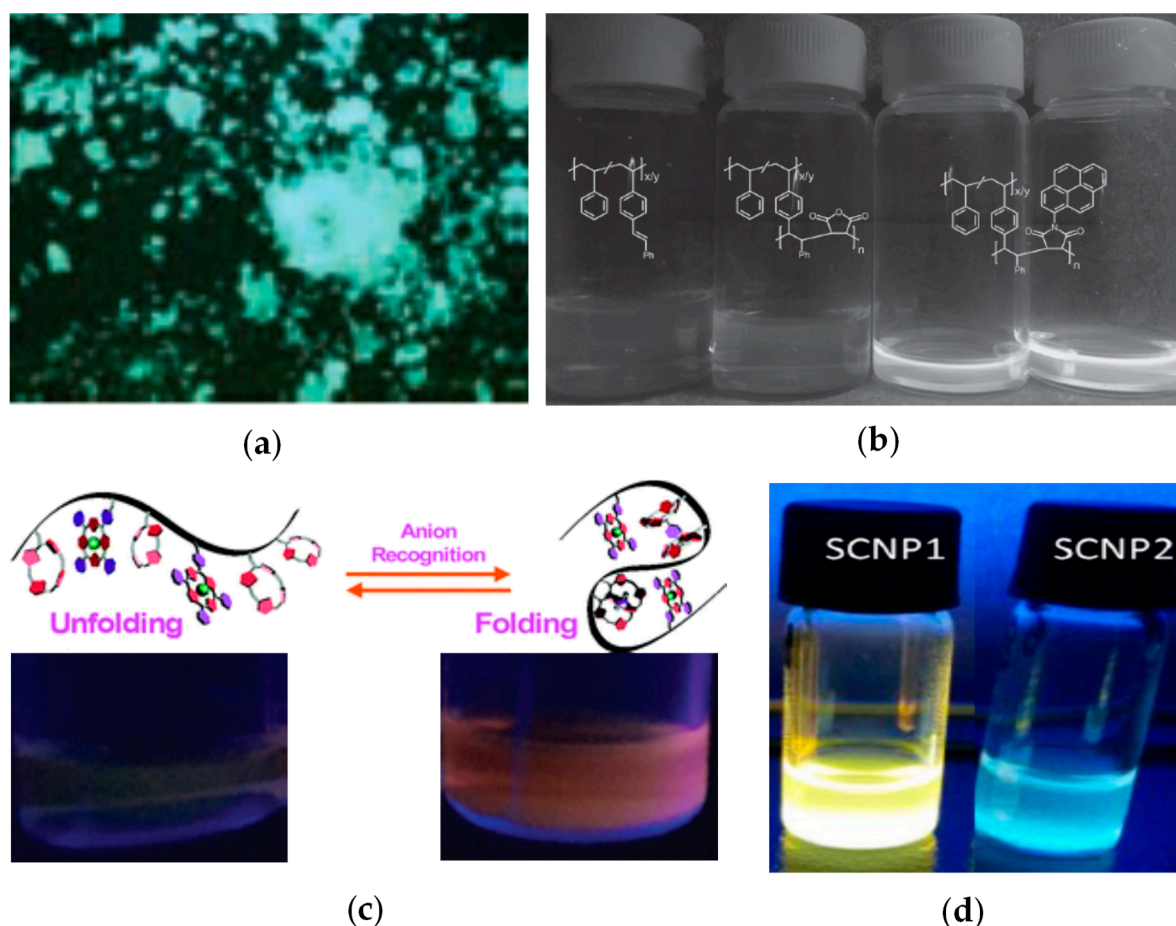
#### 2.4. Fluorophore generation through SCNP formation

The technique of fluorophore generation through SCNP formation was first reported by Oria *et al.* [50] illustrating that fluorescent SCNPs were obtained upon formation of triazole-benzene-triazole units *via* intramolecular cross-linking. Maximum photoluminescence emission at *ca.* 400 nm upon excitation at  $\lambda = 350$  nm was found when azide-containing polystyrene-based SCNPs were reacted with 1,4-diethynylbenzene *via* CuAAC reaction. The size of these SCNPs was 4.2 nm as determined by DLS measurements. No fluorescent was observed when the triazole-benzene-triazole conjugation was perturbed or absent. A fluorescent microscopy image of these SCNPs in bulk is illustrated in Figure 8a. This work was the first report of fluorescent SCNPs obtained by *in situ* generation of the fluorophore during intramolecular cross-linking.

Gillisen *et al.* [51] synthesized bipyridine-containing precursor polymers that fold intramolecularly *via*  $\pi$ - $\pi$  interactions into fluorescent SCNPs of *ca.* 15 nm in size in mixtures of tetrahydrofuran and methylcyclohexane. These SCNPs, that showed maximum photoluminescence emission at 520 nm upon excitation at  $\lambda = 382$  nm were used as fluorescent sensors of metal ions, and specifically of Cu(II), due to the strong affinity of the bipyridine units towards Cu(II) leading to strong fluorescence quenching.

Lyon *et al.* [52] prepared fluorescent SCNPs of 6 nm in size from precursor polymers containing pendant stilbene units *via* intrachain alternating copolymerization with *N*-(1-pyrene)maleimide (NPM). In absence of maleimide monomer no homopolymerization of the stilbene units to give SCNPs was observed. Hence, the role of the multifunctional NPM monomer was to promote intramolecular cross-linking of the non-fluorescent precursor polymer through alternating copolymerization with stilbene and, concurrently, to endow the resulting SCNPs with fluorescence (Figure 8b).

More recently, phosphorescent SCNPs synthesized by means of simple anion recognition have been reported by Ji *et al.* [53]. In this work, anion recognition was utilized to control the folding and unfolding of a polymethyl methacrylate-based precursors bearing pendant calix[4]pyrrole and Pt(II)-porphyrin moieties. Calix[4]pyrrole was selected as a recognized anion receptor and Pt(II)-porphyrin as a signaling agent due to their bright phosphorescence, long-lived excited-state lifetime and high stability. SCNP formation was carried out in diluted solution using a terephthalate dianion as bifunctional cross-linker, giving to nanoparticles with a diameter as small as 2 nm, and maximum phosphorescence intensity at 670 nm (Figure 8c). It is worth of mention that the precursor polymer was not phosphorescent, so the bright phosphorescence of the SCNPs in the later case was attributed to reduced contact between the Pt(II)-porphyrin moieties and molecular oxygen. Remarkably, the addition of competitive monovalent anions such as F<sup>-</sup>, Cl<sup>-</sup>, H<sub>2</sub>PO<sub>4</sub><sup>-</sup>, HSO<sub>4</sub><sup>-</sup>, Br<sup>-</sup>, and NO<sub>3</sub><sup>-</sup> gave to a decrease in the phosphorescence intensity due to SCNP unfolding. Phosphorescent SCNPs are hence promising functional nanomaterials for anion recognition.



**Figure 8.** Synthesis of fluorescent SCNPs upon intramolecular cross-linking *via* formation of fluorophore triazole-benzene-triazole units (a), fluorescent SCNPs prepared from precursor polymers containing stilbene units through intrachain alternating copolymerization with *N*-(1-pyrene)maleimide (b), phosphorescent SCNPs synthesized by means of anion recognition (c), and fluorescent SCNPs produced *via* nitrile imine-carboxylic acid ligation (SCNP1) as well as nitrile imine-mediated tetrazole-ene cycloaddition (SCNP2) (d) (reprinted with permission from [50], Copyright Wiley-VCH, 2010; [52], Copyright Wiley-VCH, 2016; [53], Copyright Royal Society of Chemistry, 2017; and [55], Copyright American Chemical Society, 2017, respectively).

Different fluorescent SCNPs based on photochemically-activated intramolecular cross-linking procedures have been developed by Barner-Kowollik and coworkers [54–57] by following the route of fluorophore generation through SCNP formation. In a first work, photoinduced nitrile imine-mediated tetrazole-ene cycloaddition (NITEC) at high dilution in tetrahydrofuran using linear precursor polymers containing protected maleimide (MAL) and tetrazole (TET) pendants was applied to the formation of fluorescent SCNPs. An increased fluorescent intensity was observed upon increasing the MAL content in the precursor polymer. Remarkably, when excited at  $\lambda = 415$  nm the maximum of the fluorescence emission band of these SCNPs was placed at 558 nm, which is a wavelength appropriate for biological applications. In a further work by the same group [55], water-soluble fluorescent SCNPs were prepared *via* two mechanisms based on tetrazole chemistry: NITEC and nitrile imine-carboxylic acid ligation (NICAL) (Figure 8d). When compared to fluorescent SCNPs prepared exclusively *via* NICAL, SCNPs collapsed *via* NITEC and NICAL showed stronger fluorescence. Recently, new polybutadiene post-functionalization techniques have been introduced by the Barner-Kowollik group [56] to produce fluorescent SCNPs based on intramolecular NITEC chemistry performed in dichloromethane. Moreover, degradable fluorescent SCNPs prepared *via* intramolecular NICAL chemistry in dichloromethane have been synthesized by this group by utilizing precursor polymers featuring self-immolative azobenzene motifs, which exhibited degradability upon treatment with sodium dithionite [57].

### 3. Conclusions

The limitations of conventional fluorophore molecules are mainly related to low brightness, unappropriated wavelength range, rapid photobleaching and fluorescence self-quenching. Sometimes, toxicity issues limited their use in biological applications. Fluorescent micro- and nanoparticle systems have been proposed as a solution to these limitations since they can contain multiple fluorophore entities leading to brighter photoluminescent emission, improved biocompatibility and stability, as well as reduced photobleaching and toxicity. Several fluorescent micro- and nanoparticle systems have been developed and evaluated as optical imaging systems (*e.g.*, block copolymer micelles and cross-linked polymer networks, QDs,  $\pi$ -conjugated polymers, and dendrimers). In spite of the successful use of some of these fluorescent micro- and nanoparticles for optical imaging applications, some problems remain to be solved since the large size of these particles (often > 10 nm) prevents efficient traversal of intact membranes in cells, appropriate tuning of UV-vis absorption and photoluminescence emission wavelengths is sometimes difficult (or even impossible), and *in vivo* accumulation of large particles in the body is a real problem for some of these systems.

In recent years, a new type of polymer nanoparticles, SCNPs, of very small size (as small as 3 nm) has been developed and several methods to endow SCNPs with fluorescent characteristics have been reported. Even if this field is still in its early infancy, fluorescent SCNPs have been prepared through four different methods: i) "precursor pre-functionalization with fluorophore", *i.e.* functionalization of the SCNPs precursor polymer with fluorophore molecules before intramolecular cross-linking, ii) "fluorophore entrapment / *in situ* generation", *i.e.* entrapment of external fluorophore molecules into non-fluorescent SCNPs by taking advantage of the denser local packaging zone(s) of the SCNPs or *in situ* generation of the fluorophore molecule inside the SCNPs, iii) "SCNP post-functionalization with fluorophore", *i.e.* post-functionalization of the SCNPs *via* chemical reaction with appropriate, complementary reactive fluorophore molecules, and iv) "fluorophore generation through SCNP formation", *i.e.* generation of fluorophore functional groups through intramolecular cross-linking. In this review, the above methods toward fluorescent SCNPs have been discussed in detail, providing with many illustrative examples.

To conclude, a bright future is expected for the application of fluorescent SCNPs to *in vitro* and *in vivo* optical imaging, as new fluorescent probes with ultra-small size (*e.g.*, 3 nm in diameter), higher brightness and better photostability than previous systems.

**Acknowledgments:** Financial support by the Spanish Ministry "Ministerio de Economía y Competitividad", MAT2015-63704-P (MINECO / FEDER, UE), the Basque Government, IT-654-13, and the Gipuzkoako Foru Aldundia, Programa Red Gipuzkoana de Ciencia, Tecnología e Innovación 2017, is acknowledged. J.D.-L.-C. is grateful to MINECO for his predoctoral grant. E. G. received funding from the "Fellows Gipuzkoa" fellowship of the Gipuzkoako Foru Aldundia.

**Author Contributions:** Julen De-La-Cuesta, Edurne González and José A. Pomposo jointly conceived, organized and wrote this review paper.

**Conflicts of Interest:** The authors declare no conflict of interest.

### References

1. Shi, J.; Kantoff, P.W.; Wooster, R.; Farokhzad, O.C. Cancer nanomedicine: progress, challenges and opportunities. *Nat. Rev. Cancer* **2017**, *17*, 20-37.
2. Robin, M.P.; O'Reilly, R.K. Strategies for preparing fluorescently labelled polymer nanoparticles. *Polym. Int.* **2015**, *64*, 174-182.
3. Chen, F.; Ehlerding, E.B.; Cai, W. Theranostic Nanoparticles. *J. Nucl. Med.* **2014**, *55*, 1919-1922.
4. Breul, A.M.; Hager, M.D.; Schubert, U.S. Fluorescent monomers as building blocks for dye labeled polymers: synthesis and application in energy conversion, biolabeling and sensors. *Chem. Soc. Rev.* **2013**, *42*, 5366-5407.
5. Hudson, Z.M.; Lunn, D.J.; Winnik, M.A.; Manners, I. Colour-tunable fluorescent multiblock micelles. *Nature Commun.* **2014**, *5*, 3372.
6. Robin, M.P.; Raymond, J.E.; O'Reilly, R.K. One-pot synthesis of super-bright fluorescent nanogel contrast agents containing a dithiomaleimide fluorophore. *Mater. Horiz.* **2015**, *2*, 54-59.



7. Michalet, X.; Pinaud, F.F.; Bentolila, L.A.; Tsay, J.M.; Doose, S.; Li, J.J.; Sundaresan, G.; Wu, A.M.; Gambhir, S.S.; Weiss, S. Quantum dots for live cells, in vivo imaging, and diagnostics. *Science* **2005**, *307*, 538-544.
8. Wu, C.; Chiu, D.T. Highly Fluorescent Semiconducting Polymer Dots for Biology and Medicine. *Angew. Chem., Int. Ed.* **2013**, *52*, 3086-3109.
9. Fuchs, S.; Otto, H.; Jehle, S.; Henklein, P.; Schlüter, A.D. Fluorescent dendrimers with a peptide cathepsin B cleavage site for drug delivery applications. *Chem. Commun.* **2005**, 1830-1832.
10. Giepmans, B.N.G.; Adams, S.R.; Ellisman, M.H.; Tsien, R.Y. The Fluorescent Toolbox for Assessing Protein Location and Function. *Science* **2006**, *312*, 217-224.
11. *Single-Chain Polymer Nanoparticles: Synthesis, Characterization, Simulations and Applications*; Pomposo, J. A., Ed.; Wiley-VCH, Weinheim, 2017.
12. Pomposo, J.A.; Rubio-Cervilla, J.; Moreno, A.J.; Lo Verso, F.; Bacova, P.; Arbe, A.; Colmenero, J. Folding Single Chains to Single-Chain Nanoparticles via Reversible Interactions: What Size Reduction Can One Expect? *Macromolecules* **2017**, *50*, 1732-1739.
13. De-La-Cuesta, J.; González, E.; Moreno, A.J.; Arbe, A.; Colmenero, J.; Pomposo, J.A. Size of Elastic Single-Chain Nanoparticles in Solution and on Surfaces. *Macromolecules* **2017**, *50*, 6323-6331.
14. Latorre-Sanchez, A.; Pomposo, J.A. Recent bioinspired applications of single-chain nanoparticles. *Polym. Int.* **2016**, *65*, 855-860.
15. Huo, M.; Wang, N.; Fang, T.; Sun, M.; Wei, Y.; Yuan, J. Single-chain polymer nanoparticles: Mimic the proteins. *Polymer* **2015**, *66*, A11-A21.
16. Pomposo, J.A. Bioinspired single-chain polymer nanoparticles. *Polym. Int.* **2014**, *63*, 589-592.
17. Perez-Baena, I.; Barroso-Bujans, F.; Gasser, U.; Arbe, A.; Moreno, A.J.; Colmenero, J.; Pomposo, J.A. Endowing Single-Chain Polymer Nanoparticles with Enzyme-Mimetic Activity. *ACS Macro Lett.* **2013**, *2*, 775-779.
18. Sanchez-Sanchez, A.; Arbe, A.; Kholbrecher, J.; Colmenero, J.; Pomposo, J. A. Efficient Synthesis of Single-Chain Globules Mimicking the Morphology and Polymerase Activity of Metalloenzymes. *Macromol. Rapid Commun.* **2015**, *36*, 1592-1597.
19. Gonzalez-Burgos, M.; Latorre-Sanchez, A.; Pomposo, J.A. Advances in Single Chain Technology. *Chem. Soc. Rev.* **2015**, *44*, 6122-6142.
20. Mavila, S.; Eivgi, O.; Berkovich, I.; Lemcoff, N.G. Intramolecular Cross-Linking Methodologies for the Synthesis of Polymer Nanoparticles. *Chem. Rev.* **2016**, *116*, 878-961.
21. Hanlon, A.M.; Lyon, C.K.; Berda, E.B. What Is Next in Single-Chain Nanoparticles? *Macromolecules* **2016**, *49*, 2-14.
22. Altintas, O.; Barner-Kowollik, C. Single-Chain Folding of Synthetic Polymers: A Critical Update. *Macromol. Rapid Commun.* **2016**, *37*, 29-46.
23. Lyon, C.K.; Prasher, A.; Hanlon, A.M.; Tuten, B.T.; Tooley, C.A.; Frank, P.G.; Berda, E.B. A Brief User's Guide to Single-Chain Nanoparticles. *Polym. Chem.* **2015**, *6*, 181-197.
24. Sanchez-Sanchez, A.; Pomposo, J.A. Single-Chain Polymer Nanoparticles via Non-Covalent and Dynamic Covalent Bonds. *Part. Part. Syst. Charact.* **2014**, *31*, 11-23.
25. Müge, A.; Elisa, H.; Meijer, E.W.; Anja, R.A.P. Dynamic Single Chain Polymeric Nanoparticles: From Structure to Function. In *Sequence-Controlled Polymers: Synthesis, Self-Assembly, and Properties*; American Chemical Society: 2014; Vol. 1170, pp 313-325.
26. Sanchez-Sanchez, A.; Perez-Baena, I.; Pomposo, J.A. Advances in Click Chemistry for Single-Chain Nanoparticle Construction. *Molecules* **2013**, *18*, 3339-3355.
27. Altintas, O.; Barner-Kowollik, C. Single Chain Folding of Synthetic Polymers by Covalent and Non-Covalent Interactions: Current Status and Future Perspectives. *Macromol. Rapid Commun.* **2012**, *33*, 958-971.
28. Aiertza, M.; Odriozola, I.; Cabanero, G.; Grande, H.-J.; Loinaz, I. Single-chain polymer nanoparticles. *Cell. Mol. Life Sci.* **2012**, *69*, 337-346.
29. Harth, E.; Horn, B.V.; Lee, V.Y.; Germack, D.S.; Gonzales, C.P.; Miller, R.D.; Hawker, C.J. A Facile Approach to Architecturally Defined Nanoparticles via Intramolecular Chain Collapse. *J. Am. Chem. Soc.* **2002**, *124*, 8653-8660.
30. Mecerreyes, D.; Lee, V.; Hawker, C.J.; Hedrick, J.L.; Wursch, A.; Volksen, W.; Magbitang, T.; Huang, E.; Miller, R.D. A Novel Approach to Functionalized Nanoparticles: Self-Crosslinking of Macromolecules in Ultradilute Solution. *Adv. Mater.* **2001**, *13*, 204-208.
31. Wang, P.; Pu, H.; Ge, J.; Jin, M.; Pan, H.; Chang, Z.; Wan, D. Fluorescence-labeled hydrophilic nanoparticles via single-chain folding. *Mater. Lett.* **2014**, *132*, 102-105.



32. Adkins, C.T.; Dobish, J.N.; Brown, S.; Harth, E. Water-Soluble Semiconducting Nanoparticles for Imaging. *ACS Macro Lett.* **2013**, *2*, 710-714.
33. Adkins, C.T.; Muchalski, H.; Harth, E. Nanoparticles with Individual Site-Isolated Semiconducting Polymers from Intramolecular Chain Collapse Processes. *Macromolecules* **2009**, *42*, 5786-5792.
34. Bai, Y.; Xing, H.; Vincil, G.A.; Lee, J.; Henderson, E.J.; Lu, Y.; Lemcoff, N.G.; Zimmerman, S.C. Practical synthesis of water-soluble organic nanoparticles with a single reactive group and a functional carrier scaffold. *Chem. Sci.* **2014**, *5*, 2862-2868.
35. Blasco, E.; Yameen, B.; Quick, A.S.; Krolla-Sidenstein, P.; Welle, A.; Wegener, M.; Barner-Kowollik, C. Designing  $\pi$ -Conjugated Polymeric Nano- and Microstructures via Light Induced Chemistry. *Macromolecules* **2016**, *48*, 8718-8728.
36. Song, C.; Li, L.; Dai, L.; Thayumanavan, S. Responsive single-chain polymer nanoparticles with host-guest features. *Polym. Chem.* **2015**, *6*, 4828-4834.
37. Akagi, T.; Piyapakorn, P.; Akashi, M. Formation of Unimer Nanoparticles by Controlling the Self-Association of Hydrophobically Modified Poly(amino acid)s. *Langmuir* **2012**, *28*, 5249-5256.
38. Piyapakorn, P.; Akagi, T.; Hachisuka, M.; Onishi, T.; Matsuoka, H.; Akashi, M. Structural Analysis of Unimer Nanoparticles Composed of Hydrophobized Poly(amino acid)s. *Macromolecules* **2013**, *46*, 6187-6194.
39. Liu, Y.; Pauloehrl, T.; Presolski, S.I.; Albertazzi, L.; Palmans, A.R.A.; Meijer, E.W. Modular Synthetic Platform for the Construction of Functional Single-Chain Polymeric Nanoparticles: From Aqueous Catalysis to Photosensitization. *J. Am. Chem. Soc.* **2015**, *137*, 13096-13105.
40. Bai, Y.; Feng, X.; Xing, H.; Xu, Y.; Kim, B.K.; Baig, N.; Zhou, T.; Gewirth, A.A.; Lu, Y.; Oldfield, E.; Zimmerman, S.C. A Highly Efficient Single-Chain Metal-Organic Nanoparticle Catalyst for Alkyne-Azide "Click" Reactions in Water and in Cells. *J. Am. Chem. Soc.* **2016**, *138*, 11077-11080.
41. Qian, G.; Zhu, B.; Wang, Y.; Deng, S.; Hu, A. Size-Tunable Polymeric Nanoreactors for One-Pot Synthesis and Encapsulation of Quantum Dots. *Macromol. Rapid Commun.* **2012**, *33*, 1393-1398.
42. Zhu, B.; Sun, S.; Wang, Y.; Deng, S.; Qian, G.; Wang, M.; Hu, A. Preparation of carbon nanodots from single chain polymeric nanoparticles and theoretical investigation of the photoluminescence mechanism. *J. Mater. Chem. C* **2013**, *1*, 580-586.
43. Li, G.; Tao, F.; Wang, L.; Li, Y.; Bai, R. A facile strategy for preparation of single-chain polymeric nanoparticles by intramolecular photo-crosslinking of azide polymers. *Polymer* **2014**, *55*, 3696-3702.
44. Hamilton, S.K.; Harth, E. Molecular Dendritic Transporter Nanoparticle Vectors Provide Efficient Intracellular Delivery of Peptides. *ACS Nano* **2009**, *3*, 402-410.
45. González-Burgos, M.; Alegría, A.; Arbe, A.; Colmenero, J.; Pomposo, J.A. An unexpected route to aldehyde-decorated single-chain nanoparticles from azides. *Polym. Chem.* **2016**, *7*, 6570-6574.
46. Jiang, J.; Thayumanavan, S. Synthesis and Characterization of Amine-Functionalized Polystyrene Nanoparticles. *Macromolecules* **2005**, *38*, 5886-5891.
47. Ruiz de Luzuriaga, A.; Ormategui, N.; Grande, H.J.; Odriozola, I.; Pomposo, J.A.; Loinaz, I. Intramolecular Click Cycloaddition: An Efficient Room-Temperature Route towards Bioconjugable Polymeric Nanoparticles. *Macromol. Rapid Commun.* **2008**, *29*, 1156-1160.
48. Zhang, Y.; Zhao, H. Surface-tunable colloidal particles stabilized by mono-tethered single-chain nanoparticles. *Polymer* **2015**, *64*, 277-284.
49. Kröger, A.P.P.; Boonen, R.J.E.A.; Paulusse, J.M.J. Well-defined single-chain polymer nanoparticles via thiol-Michael addition. *Polymer* **2017**, *120*, 119-128.
50. Oria, L.; Aguado, R.; Pomposo, J.A.; Colmenero, J. A Versatile "Click" Chemistry Precursor of Functional Polystyrene Nanoparticles. *Adv. Mater.* **2010**, *22*, 3038-3041.
51. Gillissen, M.A.J.; Voets, I.K.; Meijer, E.W.; Palmans, A.R.A. Single chain polymeric nanoparticles as compartmentalised sensors for metal ions. *Polym. Chem.* **2012**, *3*, 3166-3174.
52. Lyon, C.K.; Ellen, O.H.; Berda, E.B. Zipping Polymers into Nanoparticles via Intrachain Alternating Radical Copolymerization. *Macromol. Chem. Phys.* **2015**, *217*, 501-508.
53. Ji, X.; Guo, C.; Ke, X.-S.; Chi, X.; Sessler, J.L.; Using anion recognition to control the folding and unfolding of a single chain phosphorescent polymer. *Chem. Commun.* **2017**, *53*, 8774-8777.
54. Willenbacher, J.; Wuest, K.N.R.; Mueller, J.O.; Kaupp, M.; Wagenknecht, H.-A.; Barner-Kowollik, C. Photochemical Design of Functional Fluorescent Single-Chain Nanoparticles. *ACS Macro Lett.* **2014**, *3*, 574-579.

- 505 55. Heiler, C.; Offenloch, J.T.; Blasco, E.; Barner-Kowollik, C. Photochemically Induced Folding of Single Chain  
506 Polymer Nanoparticles in Water. *ACS Macro Lett.* **2017**, *6*, 56-61.
- 507 56. Geiselhart, C.M.; Offenloch, J.T.; Mutlu, H.; Barner-Kowollik, C. Polybutadiene Functionalization via an  
508 Efficient Avenue. *ACS Macro Lett.* **2016**, *5*, 1146-1151.
- 509 57. Offenloch, J.T.; Willenbacher, J.; Tzvetkova, P.; Heiler, C.; Mutlu, H.; Barner-Kowollik, C. Degradable  
510 fluorescent single-chain nanoparticles based on metathesis polymers. *Chem. Commun.* **2017**, *53*, 775-778.