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Article

Half-Metallic 2D ScSi₂N₄ Phases with Non-Metal-Induced Ferromagnetism and Tunable Electronic and Optical Properties

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Abstract

Two-dimensional (2D) intrinsic half-metal materials facilitate spin filtering, low-energy dissipation, and enhanced signal integrity, making them highly desirable for next-generation nanoelectronics and quantum technologies. In this work, we constructed two novel 2D half-metallic materials, α_1 -ScSi₂N₄, and α_2 -ScSi₂N₄, with unconventional ferromagnetism originating from N atoms rather than the transition metal Sc. First-principles calculations confirm their dynamic and thermal stability, as well as their intrinsic half-metallicity. We further demonstrate that their electronic and optical properties can be effectively tuned via strain, atomic adsorption, and external electric fields. A half-metal-to-metal transition occurs under compressive strain (α_1 : 10%; α_2 : 6-10%), while H/F adsorption induces a metallic state in α_1 , and H adsorption does so in α_2 . Furthermore, α_1 becomes metallic at electric fields of -0.2 to -0.5 V/Å and 0.2 to 0.5 V/Å, while α_2 undergoes a similar transition at electric fields of -0.3 to -0.5 V/Å and 0.3 to 0.5 V/Å. Both materials exhibit strong deep-UV absorption, indicating potential in optoelectronics. Symmetry breaking, charge transfer and energy level shifting maybe the tunability mechanisms caused the half-metal-to-metal transition. These findings not only expand the family of 2D half-metals with non-metal-derived magnetism but also provide new avenues for designing tunable magnetic materials for reconfigurable electronic, spintronic, and photonic applications.

Keywords: intrinsic halfmetallicity; 2D MA₂Z₄ materials; non-metal-induced ferromagnetism; first-principles calculations

1. Introduction

Two-dimensional (2D) materials have attracted widespread interest due to their exceptional electronic, mechanical, and thermal properties. These materials exhibit high carrier mobility [1], large on/off ratios [2], excellent thermal conductivity [3], and outstanding mechanical strength [4], making them highly promising for future technological advancements. However, most stable 2D materials, such as graphene [5], hexagonal boron nitride (h-BN) [6], and molybdenum disulfide (MoS₂) [7], are intrinsically non-magnetic, which limits their applications in magnetism-related fields.

Among magnetism-related 2D materials, half-metals stand out due to their unique electronic and spin-polarized properties, making them highly valuable for applications in data storage, magnetic field sensing, and spintronic devices. [8] Half-metals exhibit characteristics of both metals and semiconductors, behaving as a metal in one spin channel while acting as a semiconductor or insulator in the other. Their ability to provide 100% spin-polarized current makes them ideal for high-efficiency spintronic applications. [9] Moreover, their atomic-scale thickness enables ultra-compact device integration, while their 2D morphology allows precise external control. However, most pristine 2D materials are either nonmagnetic or lack intrinsic half-metallicity. Even when half-

metallicity is present, the band gap is often too small to meet the essential requirements for spintronic applications. [10] These limitations have driven researchers to explore novel 2D magnetic materials through bottom-up design approaches, seeking optimal properties for next-generation spintronic devices.

Numerous studies have investigated the intrinsic properties of half-metals [11], exploring their topological characteristics [12], quantum transport phenomena [13], as well as their potential applications in nanoelectronics [14] and spintronics [15]. Extensive studies have demonstrated that strain engineering, chemical doping, and heterojunction construction represent effective approaches for modulating these materials' electronic structure and magnetic characteristics. [16–19] However, the underlying mechanisms governing electronic phase transitions in half-metals remain largely unexplored. This raises a fundamental question: Can a half-metal be transformed into a metallic state through external perturbations, and what mechanisms drive this transition?

From a theoretical perspective, electronic phase transitions in 2D half-metals involve key mechanisms such as band overlap [20], Fermi level shifting [21], charge transfer [22], and symmetry breaking [23]. Understanding these processes advances knowledge in band engineering, electron correlation, and quantum phase transitions in low-dimensional materials. Additionally, controlling the electronic structure of half-metals could lead to topological phase transitions, potentially unlocking novel quantum states relevant to quantum computing and spintronics. [24] On the application side, the ability to induce a metallic state in half-metals opens up possibilities for reconfigurable electronic devices, low-power field-effect transistors, and phase-change memory. [25] Furthermore, their tunable conductivity makes them promising candidates for transparent conducting films [26], plasmonic materials, and optoelectronic applications [27].

In 2020, Hong et al. successfully synthesized monolayer MoSi_2N_4 , a representative member of the MA_2Z_4 family, via chemical vapor deposition (CVD). [28] This material exhibited remarkable environmental stability, excellent mechanical strength, and strong optical absorption capability. [29] Inspired by this advancement, we constructed a novel monolayer structure of ScSi_2N_4 through a rational elemental substitution approach. Compared with the previously reported α_1 -phase and related heterostructures [30–32], we identified an energetically favorable new structural phase, referred to as the α_2 - ScSi_2N_4 phase. A comprehensive theoretical investigation was subsequently conducted on both α_1 - and α_2 - ScSi_2N_4 , focusing on their electronic structures, optical responses, and magnetic properties. The results provide valuable insights into the structure-property relationships in this emerging class of two-dimensional materials.

In this work, we systematically investigate the stability, magnetic properties, and tunable electronic and optical characteristics of 2D α_1 - ScSi_2N_4 and α_2 - ScSi_2N_4 using first-principles density functional theory (DFT) calculations. The structural stability of both phases is verified through phonon spectrum analysis and molecular dynamics simulations. Our results confirm that α_1 - ScSi_2N_4 and α_2 - ScSi_2N_4 exhibit halfmetallic behavior, with metallic conductivity in the spin-down channel and semiconducting behavior in the spin-up channel. By constructing different magnetic configurations, we identify ferromagnetism as the magnetic ground state and reveal that the magnetism primarily originates from nitrogen atoms. This discovery, where magnetism arises from non-transition metal elements, provides new theoretical insights into magnetic 2D materials.

Furthermore, we demonstrate that both α_1 - ScSi_2N_4 and α_2 - ScSi_2N_4 undergo a transition from half-metallicity to metallicity under three distinct external perturbations: atomic adsorption, strain engineering, and electric field application. In the α_1 - ScSi_2N_4 system, a compressive strain of 10% induces a half-metal-to-metal transition, whereas α_2 - ScSi_2N_4 exhibits a similar transition under 6%-10% compressive strain. Additionally, the half-metallicity of α_1 - ScSi_2N_4 disappears upon the adsorption of H or F atoms, while α_2 - ScSi_2N_4 undergoes a comparable transition upon H adsorption. Under the action of an external electric field, when the field strength is -0.2 to -0.5 V/Å and 0.2 to 0.5 V/Å, α_1 - ScSi_2N_4 transforms into a metallic state; when the field strength is -0.3 to -0.5 V/Å and 0.3 to 0.5 V/Å, α_2 - ScSi_2N_4 becomes a metallic state. Our analysis reveals that charge transfer, symmetry

breaking, and the Stark effect play crucial roles in these electronic transitions, offering new insights into phase changes in 2D half-metals.

Finally, we explore the optical properties of these materials and find that both α_1 -ScSi₂N₄ and α_2 -ScSi₂N₄ exhibit strong absorption peaks in the deep ultraviolet region, indicating their potential for photodetection applications. These findings highlight ScSi₂N₄ as a highly tunable material with promising applications in reconfigurable electronic and optoelectronic devices.

2. Computational Details

All structural optimizations and property evaluations were performed based on density functional theory (DFT) [33], as implemented in the Vienna Ab initio Simulation Package (VASP) [34]. The ion–electron interactions were described using the projector augmented-wave (PAW) method. [35] The exchange–correlation potential was treated with the generalized gradient approximation (GGA) [36] in the Perdew–Burke–Ernzerhof (PBE) formulation [37]. A plane-wave cutoff energy of 520 eV was adopted, and the electronic self-consistency was achieved with a total energy convergence threshold of 10^{-6} eV. Full relaxation was carried out until the Hellmann–Feynman force on each atom was less than 10^{-3} eV/Å.

To accurately simulate the monolayer structure, a vacuum spacing of ~ 30 Å was introduced along the out-of-plane (z) direction to suppress spurious interactions between periodic images. Phonon dispersion relations were calculated using the finite displacement method with a $2 \times 2 \times 1$ supercell, as implemented in the PHONOPY package. [38] Additionally, first-principles molecular dynamics (FPMD) simulations were conducted on a $3 \times 3 \times 1$ ScSi₂N₄ supercell within the canonical (NVT) ensemble at 300 K, with a time step of 2 fs and a total simulation duration of 8 ps. [39]

The VASPKIT utility [40] was employed for both preprocessing and postprocessing tasks. The frequency-dependent complex dielectric function $\epsilon(\omega)$ was extracted from the dielectric tensor computed by VASP, and the optical absorption coefficient $\alpha(\omega)$ was subsequently derived via the Kramers–Kronig transformation. The peak positions and intensities of $\alpha(\omega)$ provide direct insight into the optical band gap and interband transition characteristics of the material.

3. Results and Discussion

3.1. Structure and Stability

α_1 -ScSi₂N₄ and α_2 -ScSi₂N₄ phases belong to the emerging family of two-dimensional MA₂Z₄ materials, which consist of M-site (Group IVB, VB, or VIB elements), A-site (Group IVA), and Z-site (Group VA) atoms [Figure 1a,b]. These materials exhibit diverse electronic structure characteristics, spanning semiconductors, metals, insulators, and magnetic phases, rendering them promising for low-dimensional physics and device applications. [41–46] The α_1 -ScSi₂N₄ and α_2 -ScSi₂N₄ phases we constructed both adopt a quasi-hexagonal lattice, featuring a seven-layer atomic arrangement (N–Si–N–Sc–N–Si–N), where the Sc atomic layer is symmetrically embedded between Si–N layers, forming a stable sandwich-like monolayer structure. The ScN₂ layer resembles the structural motif of a MoS₂-type monolayer. Structural optimization reveals lattice constants of $a = b = 2.97$ Å (α_1) and $a = b = 2.99$ Å (α_2), both crystallizing in the $P\bar{6}m2$ space group (see Supplementary Table S1 for detailed structural parameters). Notably, the α_2 phase exhibits subtle differences in local atomic arrangement and interlayer coupling compared to the α_1 phase.

In the ScSi₂N₄ system, Sc and Si are sandwiched between different N layers. To analyze their bonding interactions, we calculated the crystal orbital Hamilton population (COHP). As shown in Figure 1c,d, the negative COHP value between Sc–N indicates that the Sc–N bond is in a bonding state and plays a skeletal support role in the structure. Si and N usually form strong covalent bonds. Si–N covalent bonds exist in nitride materials such as Si₃N [47], and electrons are filled in a lower energy range, which means that these bonds contribute to the structural stability of the material. Negative COHP values indicate that these N–N interactions are bonding and that they contribute to the crystal stability of the material. On the other hand, the charge density difference in Figure S1(a-b) clearly

supports the formation of strong Si-N covalent bonds and Sc-N ionic bonds. This is further confirmed by Bader charge analysis [48] (As shown in Table S2). Each Sc atom transfers about $1e^-$ to inner N atoms, and each Si transfers $1e^-$ to inner N atoms and $3e^-$ to outer N atoms, forming a polar covalent bond.

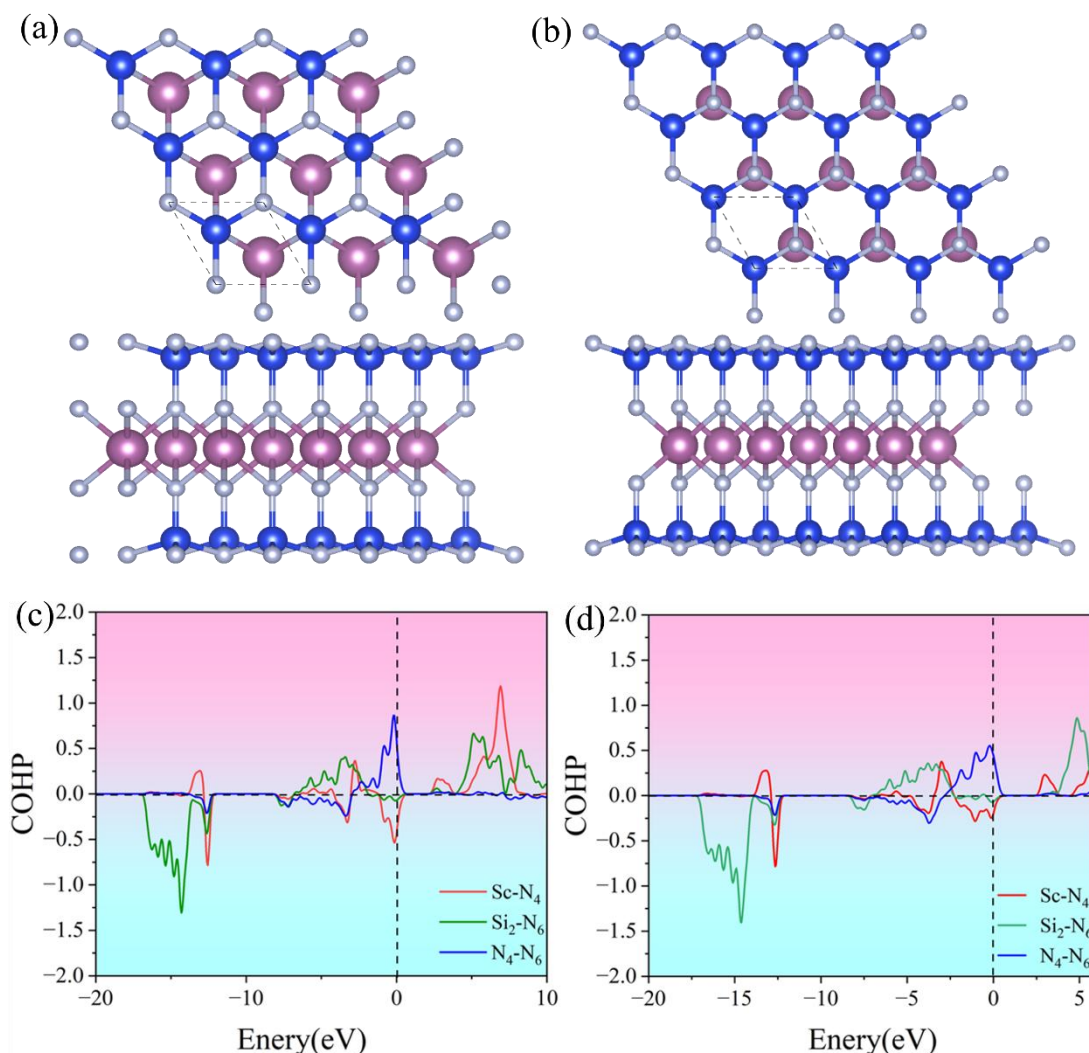


Figure 1. Structures of (a) α_1 -ScSi₂N₄ and (b) α_2 -ScSi₂N₄, COHP diagrams of (c) α_1 -ScSi₂N₄ and (d) α_2 -ScSi₂N₄. Red, green, and blue represent the bonding between Sc and N, Si and N, and N and N, respectively.

The stability of a compound is a crucial factor influencing its potential applications, as it directly impacts the feasibility of synthesis, device integration, and long-term operational reliability. Thermodynamic stability determines whether a material can be synthesized under practical conditions, while kinetic and mechanical stability ensure structural integrity under external perturbations such as temperature fluctuations and mechanical stress. Understanding these stability aspects aids in selecting appropriate synthesis routes, optimizing fabrication techniques, and predicting real-world performance in electronic, spintronic, and optoelectronic applications. Stability considerations include thermodynamic, kinetic, and mechanical aspects. The dynamic stability of ScSi₂N₄ was assessed using phonon dispersion calculations. As shown in Figure 2a,b, neither α_1 -ScSi₂N₄ nor α_2 -ScSi₂N₄ exhibits imaginary frequencies on all high-symmetry paths, which confirms their dynamic stability. In particular, the highest frequency (27-28 THz) of α_1 -ScSi₂N₄ and α_2 -ScSi₂N₄ structure indicates the robust connection between the Sc, Si, and N atoms in the two phases [49]. To further investigate the thermal stability of the ScSi₂N₄ monolayer, we performed AIMD simulations using a $2 \times 2 \times 1$ supercell at 300 K. After 8 ps, slight atomic vibrations and well-converged energy were observed in the ab initio molecular dynamics (AIMD) results for both α_1 -ScSi₂N₄ and α_2 -ScSi₂N₄,

demonstrating their thermodynamic stability. The absence of obvious structural distortions and bond breaking indicates the excellent thermal stability of the structure at 300K, as displayed in Figure 2c,d.

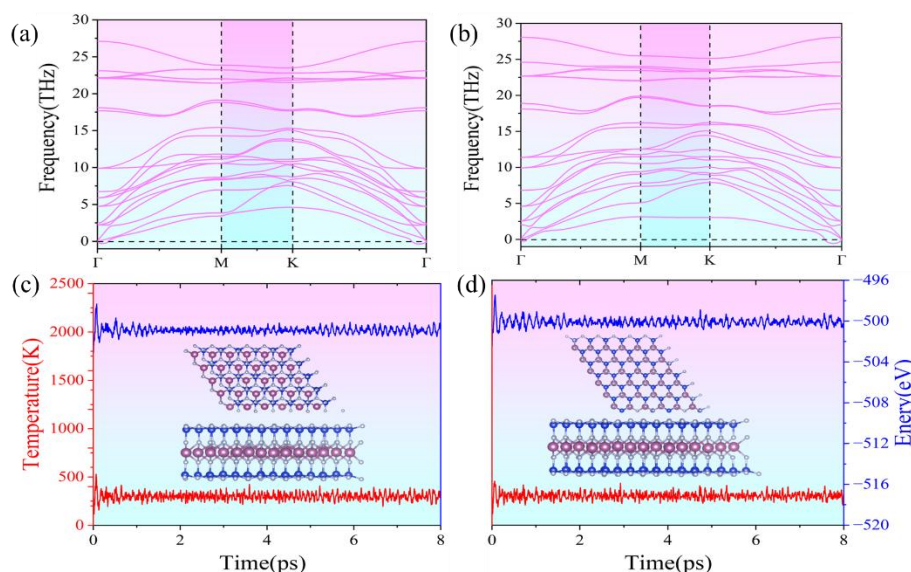


Figure 2. Phonon spectra of (a) α_1 -ScSi₂N₄ and (b) α_2 -ScSi₂N₄, molecular dynamics of (c) α_1 -ScSi₂N₄ and (d) α_2 -ScSi₂N₄ at 300K for 8ps.

Cohesive energy is a key criterion for assessing the feasibility of material synthesis. A more negative cohesive energy indicates greater material stability. The cohesive energy of α_1 -ScSi₂N₄ is -7.706 eV/atom, and that of α_2 -ScSi₂N₄ is -7.714 eV/atom. The cohesive energy is close to that of C₃N (-7.07 eV/atom) [50] and is comparable to the cohesive energy of the FeCN₂ monolayer (-7.56 eV/atom) [51]. Mechanical stability has a certain impact on the synthesis of materials on substrates and determines the structural response under strain. The calculated elastic constants of α_1 -ScSi₂N₄ (i.e., $C_{11}=348.16$ N/m, $C_{12}=152.70$ N/m, $C_{66}=97.73$ N/m) and α_2 -ScSi₂N₄ (i.e., $C_{11}=333.00$ N/m, $C_{12}=130.72$ N/m, $C_{66}=101.14$ N/m) satisfy the Born stability criterion [52,53] for a Hexagonal system (i.e., $C_{66}>0$, and $C_{11}>|C_{12}|$). The results indicate that they are mechanically stable, can maintain stability during the synthesis process, and their response under strain should be reliable. The Young's modulus and Poisson's ratio, which vary with angle, exhibit isotropy, as shown in Figure S2(a, c). The Young's modulus for α_1 -ScSi₂N₄, reflecting the material's flexibility/stiffness, is 281.19 N/m, while for α_2 -ScSi₂N₄, it is 281.67 N/m. These values are much smaller than the 340 N/m of graphene [54], but larger than the 130 N/m of MoS₂ [55], suggesting flexibility within the plane. In two-dimensional materials, the Poisson's ratio (defined as the mechanical response to external strain) is 0.44 for α_1 -ScSi₂N₄ and 0.40 for α_2 -ScSi₂N₄, as shown in Figure S2(b, d). Overall, Young's modulus and Poisson's ratio show high structural flexibility, facilitating device preparation.

3.2. Electronic Behavior

For two-dimensional materials, electronic properties are crucial as they determine the material's nature and significantly influence its practical applications. The electronic band structure of α_1 -ScSi₂N₄ and α_2 -ScSi₂N₄ are calculated from GGA+PBE, GGA+U (U=1 eV [8]), HSE06, and GGA+SOC methods as shown in Figure 3. The high symmetry paths of the two structures are both along Γ -M-K- Γ . In the spin-down channel, the valence band crosses the Fermi level, indicating metallic properties. In the spin-up channel, the band gaps are 2.29 eV for α_1 -ScSi₂N₄ and 2.53 eV for α_2 -ScSi₂N₄ at the GGA+PBE level. And the band gap values in the spin-up channel are similar at GGA+PBE, GGA+U, and GGA+SOC methods (HSE06 band structure results show a relatively large spin-up band gap). Although the band positions calculated by different methods are offset, half-metallicity is significant and well preserved in both α_1 -ScSi₂N₄ and α_2 -ScSi₂N₄ at various calculation methods.

Half-metal has great application prospects in spintronic devices, but a wide half-metal energy gap is the key to achieving this property. We usually call the absolute minimum of CBM and VBM in the spin direction of the semiconductor the half-metal gap (E_{HM}). [56] The HSE06 band structure results further confirm that the α_1 and α_2 -ScSi₂N₄ monolayers are intrinsically half-metal materials. According to this criterion, the equilibrium E_{HM} of the α_1 -ScSi₂N₄ monolayer is 0.84 eV, which is larger than YSi₂N₄ (0.248 eV) [56]. The equilibrium E_{HM} of the α_2 -ScSi₂N₄ monolayer is 0.71 eV, which prevents spin flipping. These properties have significant implications for the electronic transport properties, thermoelectric performance, spintronic characteristics, and topological features of the materials. As the data using the PBE functional gives consistent qualitative analysis results, in the subsequent calculation, we use the GGA+PBE method for all calculations.

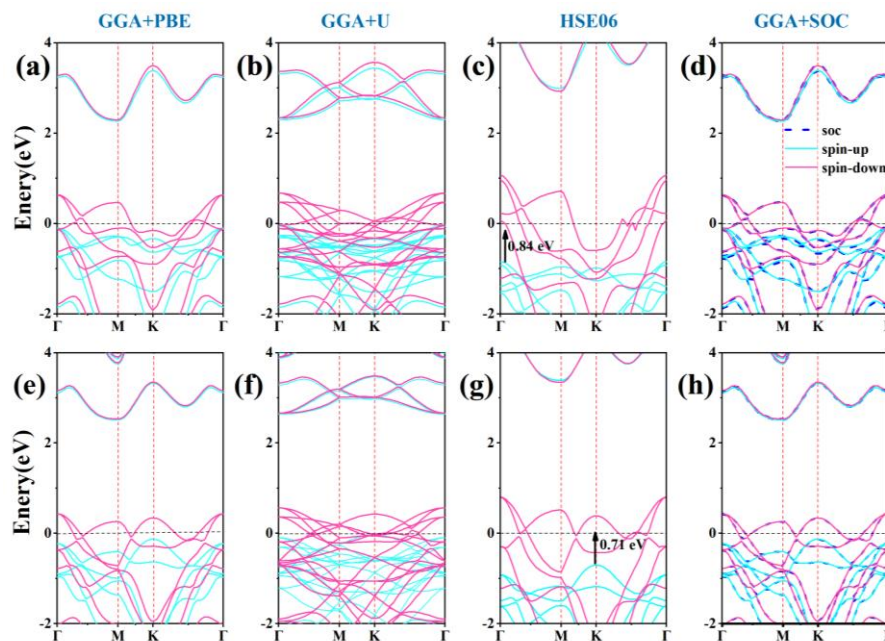


Figure 3. Electronic band structure of (a-d, the upside line) α_1 -ScSi₂N₄ and (e-h, the downside line) α_2 -ScSi₂N₄ calculated from GGA+PBE, GGA+U, HSE06, and GGA+SOC methods.

Understanding the origin of half-metallicity is essential for further exploration of the material's magnetic and spintronic properties. To elucidate the underlying mechanism, we computed the orbital-resolved electronic band structures and spin-polarized density of states (DOS) for both α_1 - and α_2 -ScSi₂N₄ monolayers, as depicted in Figure 4. The results reveal that the electronic states near the Fermi level are primarily derived from N and Sc atoms. Specifically, in both phases, the spin-down components of N-p orbitals (px, py, and pz) intersect the Fermi level, whereas the corresponding spin-up orbitals lie entirely below it, indicating complete occupancy of spin-up states and partial occupancy of spin-down states.

For α_1 -ScSi₂N₄, the conduction band minimum (CBM) in the spin-down channel is located at the M point and mainly originates from Sc-d orbitals, while the valence band maximum (VBM) appears at the Γ point and is dominated by N-px states. In α_2 -ScSi₂N₄, the CBM remains at the M point with significant Sc-d character, whereas the VBM at the Γ point is primarily contributed by N-py orbitals. The spin-resolved DOS shown in Figure 4c,f further supports that the unpaired states at the Fermi level predominantly arise from the N-p orbitals, with negligible contributions from Sc and Si atoms.

These findings clearly demonstrate that the half-metallic behavior is chiefly governed by the spin asymmetry of nitrogen p orbitals, with spin-up states being fully occupied and spin-down states partially unoccupied. Moreover, the presence of spin-down band crossings at the Fermi level suggests a robust spin polarization, implying asymmetric electron distribution between spin channels. Such a feature confirms the half-metallic nature and provides a foundation for subsequent investigations into the magnetic characteristics of the ScSi₂N₄ monolayers.

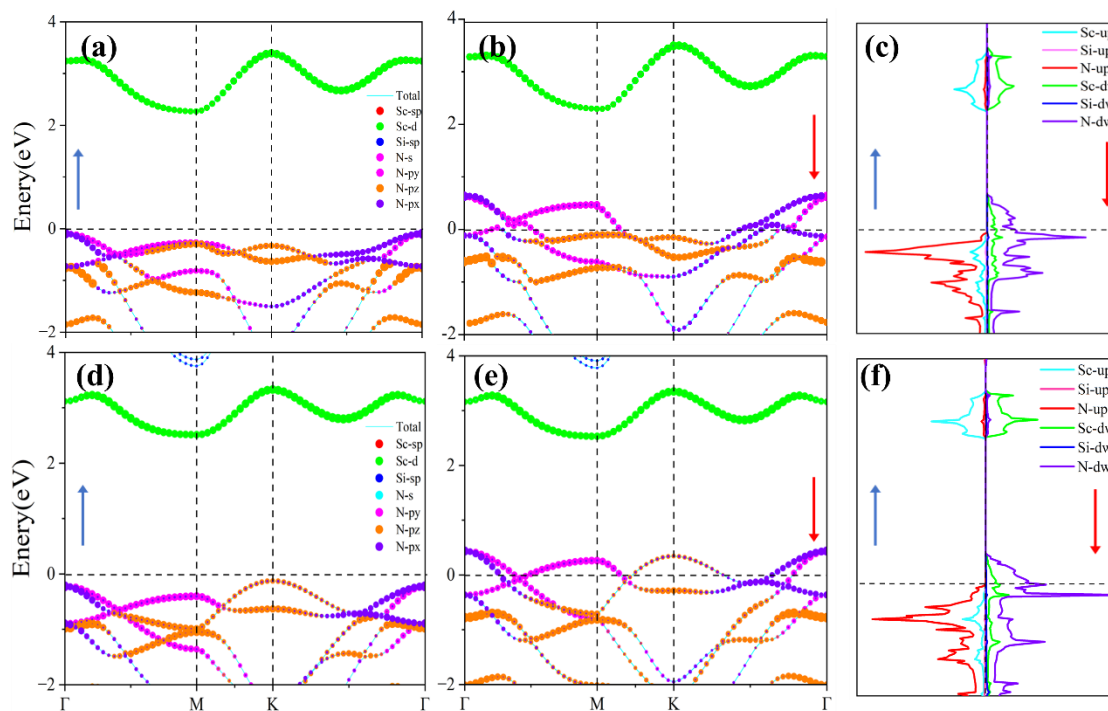


Figure 4. Orbital projection band structure and spin projection density of states (DOS) of α_1 -ScSi₂N₄ (a, b, c) and α_2 -ScSi₂N₄ (d, e, f). The \uparrow or \downarrow represents spin-up or spin-down, respectively.

3.3. Magnetic Properties

The unique electronic properties of ScSi₂N₄ monolayers inspired us to explore their magnetic properties. We established four magnetic state configurations to explore the magnetic ground state of α_1 -ScSi₂N₄ and α_2 -ScSi₂N₄. (As shown in the supporting information Figure S3), Calculations found that both α_1 -ScSi₂N₄ and α_2 -ScSi₂N₄ are ferromagnetic ground states. Surprisingly, the ferromagnetism in ScSi₂N₄ primarily originates from N atoms, rather than the transition metal Sc. In α_1 -ScSi₂N₄, the local magnetic moment of Sc is approximately 0.05 μ_B , while two types of N atoms exhibit magnetic moments of 0.34 μ_B (type 1 N: near Sc atom) and 0.05 μ_B (type 2 N: far from Sc atom), respectively. The magnetic moment of Sc atoms in α_2 -ScSi₂N₄ is about 0.04 μ_B , the local magnetic moment of type 1 N atoms is about 0.3 μ_B , and the local magnetic moment of type 2 N atoms is 0.05 μ_B . These observations raise two key questions: (1) why does magnetism originate from non-metallic atoms, and (2) why do the N atoms in α_1 -ScSi₂N₄ and α_2 -ScSi₂N₄ exhibit different magnetic moments?

To gain deeper insight into the origin of magnetism, we analyzed the spin-orbit coupling (SOC) effect. If the bands near the Fermi level were primarily formed by the d orbitals of transition metal atoms, SOC would significantly alter these band structures, potentially introducing energy gaps, as seen in the 29 meV gap in the half-metallic VCl₃ monolayer [57]. However, for the ScSi₂N₄ monolayer, since the bands near the Fermi level are mainly composed of the p orbitals of non-metallic N atoms, the SOC of the N-p orbitals is weaker than that of the TM-d orbitals. To verify this, we recalculated the electronic band structure using the GGA+SOC method (as shown in Figure 3d,h). The results show that the energy bands near the Fermi level are almost the same as those calculated by the GGA-PBE method. Since the GGA-PBE method cannot accurately describe the strongly correlated system containing unfilled d subshells, we used the GGA+U method to further verify the half-metallic properties of ScSi₂N₄ (see Figure 3b,f). The half-metallic properties of ScSi₂N₄ are well preserved except for a small shift in the energy band position. In addition, the magnetic moment calculation results show that the magnetic moment of each ScSi₂N₄ unit is about 1 μ_B . These results prove that the magnetic moment is mainly contributed by the N-p orbital rather than the Sc-d orbitals.

We further examined the spin-polarized charge distribution (Figure 5a,b) and observed that the spin density is more delocalized in the intermediate region between the metal atoms, which is

consistent with the magnetic moment properties in YN_2 [58]. In conventional transition metal magnetic materials, the magnetic moment is mainly contributed by transition metal atoms. In ScSi_2N_4 monolayers, the Pauli electronegativity of N atoms (3.04) is much greater than the Pauli electronegativity of Sc atoms (about 1.36) and the electronegativity of Si (about 1.90). Therefore, Sc ($3d^14s^2$) and Si ($3s^23p^2$) atoms will provide electrons to N ($2s^22p^3$) atoms. Based on the Bader charge transfer analysis, each Sc atom transfers about one electron to adjacent N atoms. Each Si atom also transfers one electron to the N atoms adjacent to Sc. In short, both Sc and Si atoms transfer electrons to inner N atoms rather than outer N atoms. This results in the valence electrons of Sc atoms transferring to the N atoms, forming ionic bonds. Therefore, the Sc atom remains in the $3d^24s^0$ state, in which the two occupied 3d electrons are spin-antiparallel and exhibit a nearly zero magnetic moment.

This mechanism is analogous to the situation in TaN_2 [59], where the strong electronegativity of N atoms destroys the magnetic ground state of metal atoms. Similarly, in TiC_2 [60], the electronic states near the Fermi level of transition metals are removed. At the same time, the 3p states of two inner N atoms share three foreign electrons, so that two N atoms carry nine 3p electrons. If we use the “nearly free electron gas” [61,62] model to describe these valence electrons, nine 3p electrons will present the electronic layer structure as shown in Figure 5c. This results in a magnetic moment of approximately $1\mu\text{B}$ per ScSi_2N_4 unit, which aligns with the results obtained using various methods, including GGA+PBE, GGA+U, and HSE06.

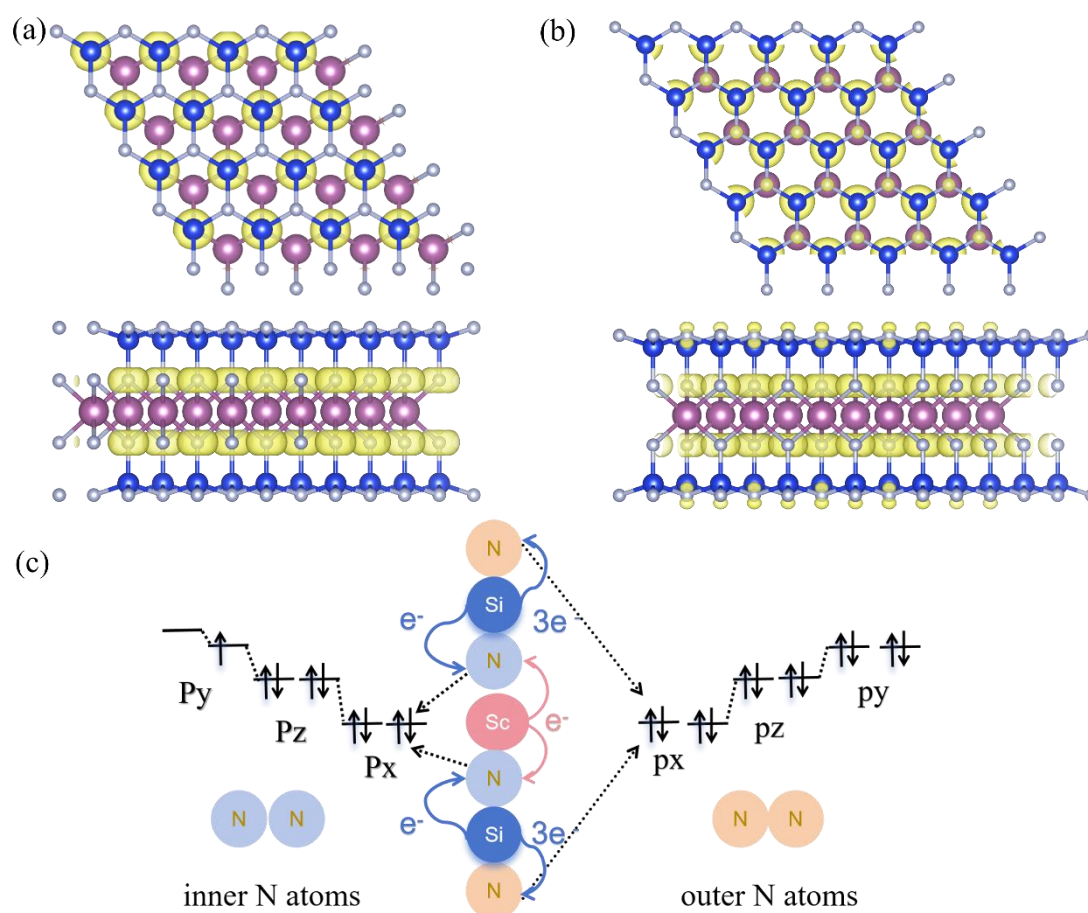


Figure 5. (a) Spin charge density diagram of $\alpha_1\text{-ScSi}_2\text{N}_4$ and (b) $\alpha_2\text{-ScSi}_2\text{N}_4$. (c) Magnetic exchange mechanism diagram.

3.4. Optical Properties

As half-metals, $\alpha_1\text{-ScSi}_2\text{N}_4$ and $\alpha_2\text{-ScSi}_2\text{N}_4$ exhibit large band gaps in the spin-up direction, which contribute to their strong light absorption properties. As illustrated in Figure 6a,e, both phases

display pronounced absorption peaks within the 0–200 nm range. This behavior is primarily attributed to their similarly large indirect band gaps in the spin-up channel, resulting in a low absorption coefficient in the visible light spectrum [63]. The half-metallicity of ScSi_2N_4 monolayers induces spin-polarized semiconducting behavior in the majority-spin channel, enabling optical transitions between the valence and conduction bands across this spectral range. These transitions correspond to fundamental bandgap excitations typical of semiconductors and insulators, or may involve intra-band excitations within the material's molecular framework [64].

Moreover, the inherently wide bandgap of these materials enhances their resistance to strong light irradiation, making them highly suitable for ultraviolet (UV) applications. As shown in Figure 6, both $\alpha_1\text{-ScSi}_2\text{N}_4$ and $\alpha_2\text{-ScSi}_2\text{N}_4$ monolayers exhibit remarkable optical absorption in the deep ultraviolet (DUV, <200 nm) region, with absorption coefficients exceeding $3.2 \times 10^5 \text{ cm}^{-1}$, peaking in the 120–200 nm range. This material exhibits a pronounced absorption peak in the deep ultraviolet region, with an optical absorption capability comparable to that of other reported two-dimensional materials. For instance, monolayer MoS_2 [65] exhibits an absorption edge around 450 nm, with maximum UV absorption coefficients below $1.0 \times 10^5 \text{ cm}^{-1}$. The absorption intensity of α_1 - and $\alpha_2\text{-ScSi}_2\text{N}_4$ is also comparable to that of WSi_2N_4 , which has a reported maximum absorption coefficient of $3.61 \times 10^5 \text{ cm}^{-1}$ [66].

Therefore, with their excellent deep-ultraviolet absorption capability, along with superior structural and thermal stability, α_1 - and $\alpha_2\text{-ScSi}_2\text{N}_4$ are promising candidates for next-generation high-performance ultraviolet optoelectronic devices, exhibiting performance comparable to some conventional two-dimensional materials in this field.

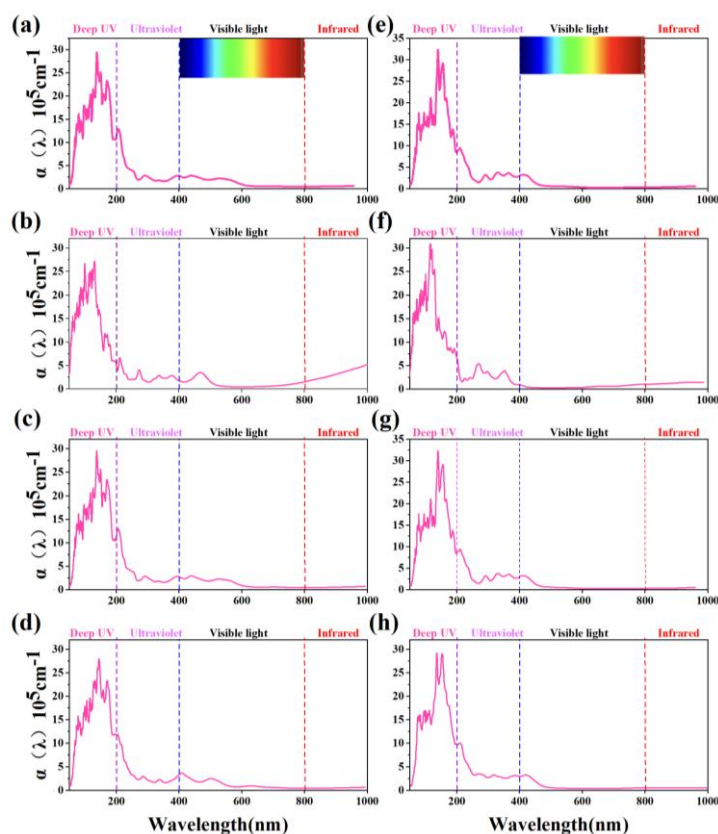


Figure 6. Absorption spectra of (a-d, the left side row) $\alpha_1\text{-ScSi}_2\text{N}_4$ and (e-h, the right side row) $\alpha_2\text{-ScSi}_2\text{N}_4$ under different conditions. (a, e) Intrinsic monolayers, (b, f) under 10% biaxial compressive strain, (c, g) under an electric field of 5 V/A, (d, h) adsorbing an H atom.

3.5. Regulation of Electronic and Optical Properties

Half-metals are a unique class of materials whose electronic structure exhibits characteristics of both metals and semiconductors. This distinctive electronic structure endows half-metals with numerous advantages, granting them broad application potential in fields such as electronics, spintronics, thermoelectric materials, and optical devices. To investigate the stability of half-metals, we employed methods such as strain, atomic adsorption, and external electric fields to modulate their half-metallic properties. These approaches effectively alter the band structure of the materials, enabling precise control over their electronic and optical properties and opening avenues for optimizing material performance.

3.5.1. Strain Effects

Strain engineering is a powerful technique for modulating the electronic [67], magnetic [68] and optical properties [66] of two-dimensional materials. Biaxial strain can be experimentally realized through thermal expansion mismatch [69]. In our calculations, we applied biaxial strain ranging from -10% to 10% to adjust the interatomic distances and analyzed its effects on the spin-up valence band maximum (VBM) and conduction band minimum (CBM) of α_1 -ScSi₂N₄ and α_2 -ScSi₂N₄, as shown in Figure 7a. For α_1 -ScSi₂N₄, a 10% compressive strain induces a transition to metallic behavior by causing the spin-up band to cross the Fermi level at the M point (see Figure S4, the blue circle indicates the half-metal to metal transition). Under tensile strain, the lattice expands, which can widen previously overlapping bands, leading to modifications in electronic band distribution. However, even under tensile strain, α_1 -ScSi₂N₄ retains its half-metallic nature (see Figure S4). In α_2 -ScSi₂N₄, a compressive strain in the range of 6%-10% induces a transition to metallicity, with the spin-up band crossing the Fermi level at the K point. (see Figure S5, the blue circle indicates the half-metal to metal transition) As tensile strain increases, α_2 -ScSi₂N₄ also retains its half-metallic properties, as confirmed in Figure S5.

Additionally, applying a 10% biaxial compressive strain modifies the optical properties of both α_1 -ScSi₂N₄ and α_2 -ScSi₂N₄ (as depicted in Figure 6b,f); their primary absorption peaks remain and sharpen in the deep ultraviolet region. Interestingly, an additional peak emerges in the visible range of α_1 -ScSi₂N₄ under 10% compressive strain. These results indicate that strain can serve as an effective tuning parameter for regulating the electronic properties of ScSi₂N₄ monolayers, which is crucial for their application in high-performance electronic and optoelectronic devices.

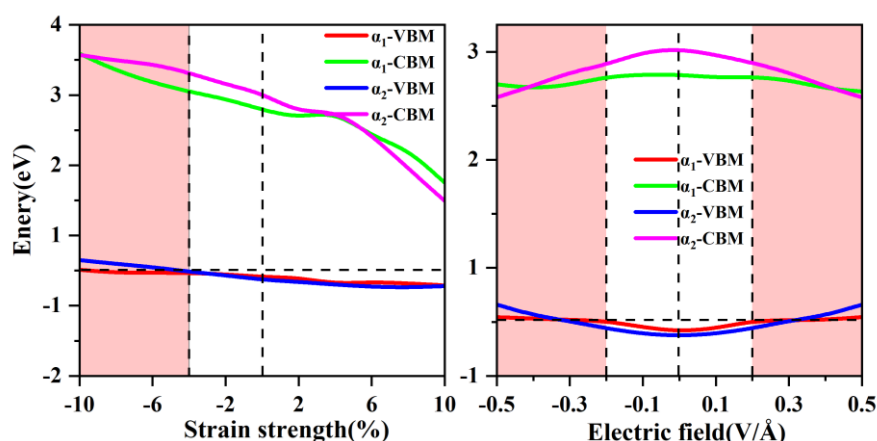


Figure 7. (a) The spin-up VBM and CBM of α_1 and α_2 -ScSi₂N₄ vary with biaxial strain. (b) The band gap of α_1 and α_2 -ScSi₂N₄ varies with increasing electric field strength.

3.5.2. Atoms Adsorption

Atomic or molecular adsorption is an effective strategy for modifying the physical properties of materials [63,64,70]. We examined the effects of adsorption on the physical properties of a ScSi₂N₄

monolayer based on a $2 \times 2 \times 1$ supercell. During the adsorption of H, O, and F atoms on the ScSi_2N_4 supercell, four different adsorption sites were selected for structural optimization: the top of the N atom, the top of the Si atom, the Si-N bridge site, and the ring center position [71], as shown in Figure S6. By calculating the adsorption energy (E_a), the most stable adsorption positions of atoms on the ScSi_2N_4 surface were determined.

$$E_a = E_{\text{Total}} - E_{\text{ScSi}_2\text{N}_4} - E_{\text{Atom}} \quad (1)$$

Where E_{Total} represents the total energy of the adsorption system, $E_{\text{ScSi}_2\text{N}_4}$ is the energy of the pristine $\alpha_1\text{-ScSi}_2\text{N}_4$ or $\alpha_2\text{-ScSi}_2\text{N}_4$ monolayer, and E_{Atom} is the energy of the isolated H, O, or F atom. Tables S3 and S4 list the total energies and adsorption energies of the ScSi_2N_4 monolayer at different adsorption sites. The results indicate that in the α_1 system, H, O, and F atoms preferentially adsorb at the top of the N atom (a-site), with H-N, O-N, and F-N bond lengths of 1.10 Å, 1.60 Å, and 1.50 Å, respectively, representing the most stable configuration. In the $\alpha_2\text{-ScSi}_2\text{N}_4$ system, H and O atoms tend to adsorb at the top of the N atom (a-site), with H-N and O-N bond lengths of 1.10 Å and 1.60 Å, respectively, while the F atom prefers to adsorb at the top of the Si atom (b-site), with an F-Si bond length of 1.60 Å, forming the most stable configuration. Figure S7 presents the AIMD simulation results of H, O, and F atom adsorption on the $\alpha_1\text{-ScSi}_2\text{N}_4$ and $\alpha_2\text{-ScSi}_2\text{N}_4$ monolayers. No bond breaking, phase transitions, or significant structural distortions were observed during the simulation, confirming the excellent thermal stability of the atomic adsorption system at room temperature.

Subsequently, the electronic properties of hydrogen, oxygen, and fluorine atoms at their most stable adsorption sites were investigated. The electronic structures of the optimal adsorption sites were calculated, and the results are shown in Figure 8. It reveals that after the adsorption of H and F atoms, the half-metallic characteristics of the $\alpha_1\text{-ScSi}_2\text{N}_4$ monolayer disappear, transitioning to metallic properties. For the $\alpha_2\text{-ScSi}_2\text{N}_4$ monolayer, the electronic properties change from half-metallic to metallic only upon H atom adsorption.

Further calculations of the optical properties of the H atom adsorption system exhibits an absorption peak in the near-ultraviolet region, along with weak absorption peaks in the visible light region (blue and green light) (shown in Figure 6d,h). Tables S5 and S6 list total and partial magnetic moments of H, O, and F atoms adsorbed at the most stable site of $\alpha_1\text{-ScSi}_2\text{N}_4$ and $\alpha_2\text{-ScSi}_2\text{N}_4$ under adsorption conditions. The data show that the N atom remains the primary contributor to the total magnetic moment.

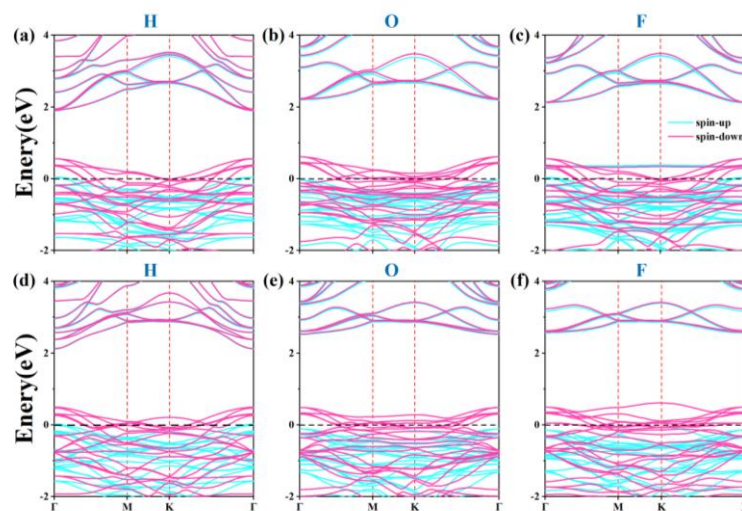


Figure 8. Energy bands of (a-c) $\alpha_1\text{-ScSi}_2\text{N}_4$ and (d-f) $\alpha_2\text{-ScSi}_2\text{N}_4$ after adsorption of H, O, and F atoms.

3.5.3. Electric Field Control

The electronic properties of 2D materials are also highly responsive to external electric fields [72–74]. We investigated the effects of applying an electric field along the z-direction, varying its strength from -0.5 to 0.5 V/Å. Electronic structure calculations (Figure 8b) indicate that increasing the electric field strength gradually reduces the spin-up band gap. When the electric field strength is -0.2 to -0.5 and 0.2 to 0.5, the spin-up conduction band at the Γ point moves down and crosses the Fermi level, which enhances the metallicity of α_1 -ScSi₂N₄. (As shown in Figure S8, the blue circle indicates the transition from half-metal to metal). In α_2 -ScSi₂N₄, the system maintains half-metallicity for field strengths in the range of -0.1 to -0.2 V/Å and 0.1 to 0.2 V/Å. However, for stronger fields (-0.3 to -0.5 V/Å or 0.3 to 0.5 V/Å), the spin-up valence band crosses the Fermi level at the K point, inducing a transition to metallic behavior. (As shown in Figure S9, the blue circle indicates the transition from half-metal to metal)

Additionally, the optical properties of α_1 -ScSi₂N₄ remain robust under an applied electric field, with its deep ultraviolet absorption peak intensifying as the field strength increases. Furthermore, a new absorption peak emerges in the visible region (Figure 6c). The optical response of α_2 -ScSi₂N₄ follows a similar trend, with increasing deep ultraviolet absorption intensity and the emergence of weak visible light absorption peaks (Figure 6g).

3.5.4. Mechanism of Regulating Electronic and Optical Properties

Half-metallic properties and optical properties of α_1 -ScSi₂N₄ and α_2 -ScSi₂N₄ are modulated under strain, atomic adsorption, and external electric fields. We thought that the mechanisms were different from the diverse regulating methods. The underlying mechanism for strain-induced property modulation is likely symmetry breaking, which alters electronic band structures and facilitates phase transitions. The primary mechanism driving the adsorption-induced modifications is mainly caused by charge transfer, which alters the electronic band structure and magnetic properties of the system. The dominant mechanism underlying electric-field-induced transitions is energy level shifting, where the applied field modulates charge distributions and band alignments, leading to phase transitions between half-metallic and metallic states. These findings highlight the potential of electric fields as a precise tuning parameter for controlling the electronic and optical properties of ScSi₂N₄ monolayers, paving the way for their application in reconfigurable spintronic and optoelectronic devices.

4. Conclusions

In this work, we calculated two novel 2D materials, α_1 -ScSi₂N₄ and α_2 -ScSi₂N₄ monolayers, and systematically investigated their structural stability, electronic properties, magnetic origin, and tunability under external perturbations. Phonon dispersion and ab initio molecular dynamics calculations confirm that both structures are dynamically and thermally stable. Our results reveal that α_1 -ScSi₂N₄ and α_2 -ScSi₂N₄ exhibit intrinsic half-metallicity, where the half-metallic behavior is predominantly governed by N-p orbitals rather than the transition metal Sc, highlighting an unconventional origin of magnetism in these materials. Furthermore, we explore the tunability of their electronic and optical properties under strain, atomic adsorption, and external electric fields. In the α_1 -ScSi₂N₄ system, a half-metal-to-metal transition occurs under a compressive strain of 10%, while α_2 -ScSi₂N₄ exhibits a similar transition within a 6%-10% strain range. Atomic adsorption studies show that H and F adsorption eliminate the half-metallicity of α_1 -ScSi₂N₄, leading to a metallic state, whereas in α_2 -ScSi₂N₄, only H adsorption induces a transition to metallic behavior. Under an applied electric field, α_1 -ScSi₂N₄ transforms from a half-metal to metal at -0.2 to -0.5 V/Å and 0.2 to 0.5 V/Å, while when subjected to an electric field strength of -0.3 to -0.5 V/Å or 0.3 to 0.5 V/Å, α_2 -ScSi₂N₄ exhibits metallic behavior. These tunability mechanisms (symmetry breaking, charge transfer, and energy level shifting) provide new insights into electronic phase transitions in 2D half-metals. Additionally, optical absorption calculations reveal that both α_1 -ScSi₂N₄ and α_2 -ScSi₂N₄ exhibit strong

absorption peaks in the deep ultraviolet region, suggesting potential applications in photodetectors. Our findings not only expand the family of 2D half-metals with unconventional magnetic origins but also establish ScSi₂N₄ monolayers as promising candidates for reconfigurable spintronic, electronic, and optoelectronic devices.

Data availability: The data that support the findings of this study are available from the corresponding author upon reasonable request.

Conflicts of interest: The authors declare no competing financial interest.

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