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## Article

# Synthesis of Oxymethyl Derivatives of 1,2,4-Triazole-3-Carboxamides and Their Biological Activities

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**Abstract:** Ribavirin and its analogues exhibit an in vitro antiproliferative effect in cancer cells. In this work we studied the biological activities of a number of oxymethyl derivatives of ribavirin's aglycon – 1,2,4-triazole-3-carboxamide. Oxymethyl derivatives of 1,2,4-triazole-3-carboxamide with substitutions at the fifth or first position of the triazole ring, were synthesized and their antiproliferative and antimicrobial effects were assessed. For both series the presence of an antiproliferative effect was investigated and in the case of 1-oxymethyl derivatives was shown an antimicrobial potential against a Gram-positive bacteria *Micrococcus luteus* and Gram-negative bacterium *Pseudomonas aeruginosa*. The obtained results showed that the n-decyloxymethyl derivatives induced leukemia cell death at low micromolar concentrations. We confirmed that n-decyloxymethyl derivatives of ribavirin inhibited cell cycle progression and induced accumulation of leukemia cells in subG1-phase. The molecular docking results suggest that oxymethyl derivatives may act by inhibiting translation initiation due to interfering with eIF4E assembly. The outcome results relived that active derivatives (1- or 5-n-decyloxymethyl-1,2,4-triazole-3-carboxamides) can be considered as a lead compound for anticancer treatments.

**Keywords:** 1,2,4-triazole-3-carboxamides; ribavirin; acute lymphoblastic leukemia; chronic myeloid leukemia; cancer treatment; antimicrobial effect

## 1. Introduction

Synthetic analogues of natural nucleosides have wide spectrum of applications as antiviral, anticancer and antibacterial compounds. An example is ribavirin (1-( $\beta$ -D-ribofuranosyl)-1,2,4-triazole-3-carboxamide, **1a**), which is not only used as an antiviral drug, but also shows significant potential in a therapy of cancer including blood cancer [1,2], as well as a moderate antimicrobial effect [3]. Thus, the multivalency of ribavirin applications makes it an interesting parent structure for new drug candidate design.

However, numerous studies have shown that ribavirin has teratogenic and genotoxic effects, which significantly limit its therapeutic application [4]. This feature is common to most of the other nucleoside analogues, probably due to their involvement into the basic metabolic pathways. It is therefore of interest to search for alternatives of compound **1a** that have meaningful structural differences from nucleosides. For some ribavirin analogues with different fragments at the 1- and/or

5-positions of triazole ring show specific activity in the cells of several cancer type. For example, several triazoles with biphenyl group in 1-position are cytotoxic in breast cancer cells in micromolar concentrations [5]. Several hybrid molecules containing secoestroides and triazole fragments demonstrate cytotoxicity in cervical cancer and breast cancer cell lines [6–8]. However, despite significant progress in bioorganic chemistry of nucleoside analogues, identification of the most promising ways of their modification to maintain high efficacy and simultaneously reduce the significant side effects is highly in demand. Therefore, the traditional approach in the synthesis of new agents, followed by the identification of their biological effects, remains relevant to the solution of this problem.

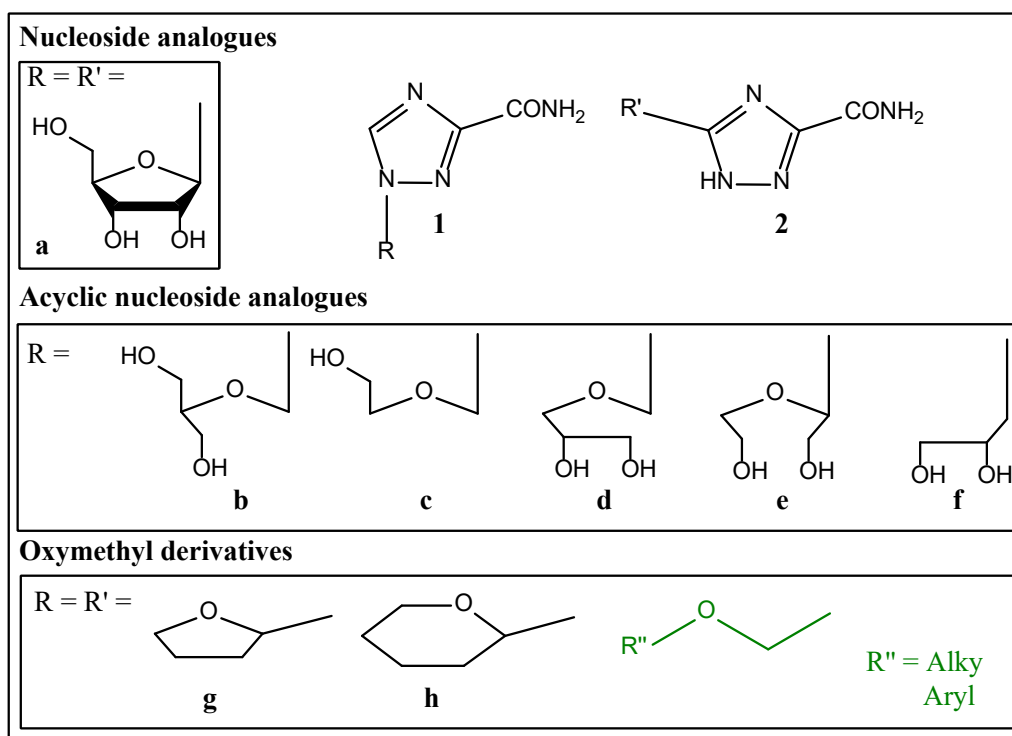
Recently we showed antiproliferative effects of the ribavirin aglycon in acute lymphoblastic leukemia and chronic myeloid leukemia cell lines [9]. We obtained ribavirin analogues by a replacement of ribavirin ribose fragment with tetrahydropyran and tetrahydrofuran groups in 5- and 1-positions of the triazole ring. We showed the accumulation of cancer cells treated with 1,2,4-triazole-3-carboxamides in the G1 phase of the cell cycle, and the induction of caspase-3 cleavage resulting in apoptosis in leukemia cells [9]. Therefore, we assume that the ribavirin analogs with non-sugar fragment in 1- or 5-position may act like nucleoside analogs.

The present study was undertaken to prepare a series of oxymethyl derivatives of 1,2,4-triazole-3-carboxamide as well as evaluation of its anticancer actions in leukemia cell lines and antimicrobial activities in Gram-positive and Gram-negative bacteria.

## 2. Results and Discussions

Biologically active nucleosides analogues were obtained by replacing the carbohydrate fragment with its acyclic analogue, mostly with the exclusion of some hydroxyl groups. In the case of ribavirin analogues obtained using this approach, there should be distinguished several derivatives with acyclic carbohydrate fragment and substitution at the position 1 – 1,2,4-triazole-3-carboxamides **1b-f**. Another approach to the nucleoside analogues modification was the replacement of the N-glycoside bond with a C-glycoside for easier exclusion of biolabile bond from the structure of the analogue without change in the main pharmacophore fragments. The ribavirin C-nucleoside analogue **2a** was obtained by this method applied to modification **1a** [10,11]. The authors noted the importance of the hydroxyethoxymethyl fragment presence in the structure of the molecule for an acyclic analogue to retain antiviral activity. Later, the other 1-hydroxyethoxymethyl derivatives of 5-substituted 1,2,4-triazole-3-carboxamides were synthesized, some of which showed antiviral activity against hepatitis C virus and anticancer effect on cell models [9,12,13]. Thus, the reduction of the carbohydrate to a hydroxyethoxymethyl moiety does not prevent molecular recognition of ribavirin analogues by a significant number of enzymes.

In our previous work on modification of the carbohydrate moiety, we have shown that derivatives of 1,2,4-triazole-3-carboxamide **1g-h** and **2g-h** substituted at the position 5 as well as at the position 1 of the triazole ring with 2-tetrahydrofuranyl or 2-tetrahydropyranyl groups, analogues of the carbohydrate backbone lacking hydroxyl groups, inhibit proliferation of chronic myeloid (K562) and acute lymphoblastic (CCRF-SB) leukemia cells [9]. However, the detailed mechanism of the biological activity of compounds **1g-h** and **2g-h** remains unclear. Therefore, we assumed that the ribavirin analogs with non-sugar fragment in 1- or 5-position may in some aspects act like a nucleoside analogs, in particular reveal anticancer and antimicrobial effects.

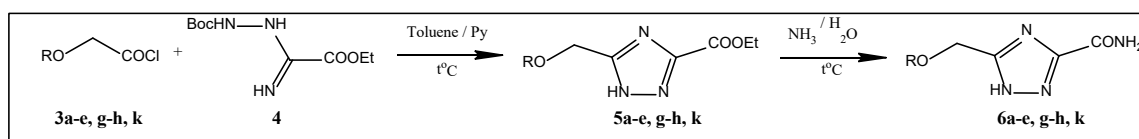


**Figure 1.** Ribavirin **1a** and its analogues.

### 2.1. Synthesis

Two series of compounds were synthesized: 5-oxymethyl-1,2,4-triazole-3-carboxamides **6a-e**, **g-h**, **k** and 1-oxymethyl-1,2,4-triazole-3-carboxamides **1c** and **11a-k**. The methods of their synthesis differed for each series: the introduction of 5-oxymethyl fragment was carried out by cyclization of triazole fragment, while the introduction of 1-oxymethyl fragment was carried out by alkylation of methyl 1,2,4-triazole-3-carboxylate.

5-oxymethyl analogues of ribavirin **6a-e**, **g-h**, **k** were synthesized by the previously described method [15] consisting in ammonolysis of ethyl esters of 5-oxymethyl-1,2,4-triazole-3-carboxylic acids **5a-e**, **g-h**, **k** obtained by treatment of  $\beta$ -N-t-butyloxycarbonyloxalamidrazone **11** with oxyacetic acids chlorohydrates followed by one-pot cyclization of intermediates (Figure 2).



**Figure 2.** 5-oxymethyl-1,2,4-triazole-3-carboxamides preparation.

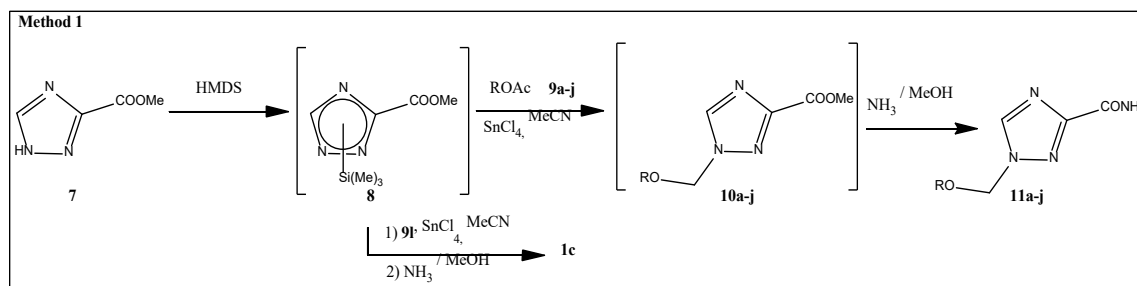
**Table 1.** Synthesized 5-oxymethyl-1,2,4-triazole-3-carboxamides **6**.

S. No.	R	Yield
<b>6a</b>	Me	60%
<b>6b</b>	Et	53%
<b>6c</b>	n-Pr	62%
<b>6d</b>	i-Pr	24%

<b>6e</b>	n-Bu	33%
<b>6g</b>	n-C <sub>10</sub> H <sub>21</sub>	43%
<b>6h</b>	Bn	68%
<b>6k</b>	Ph	76%

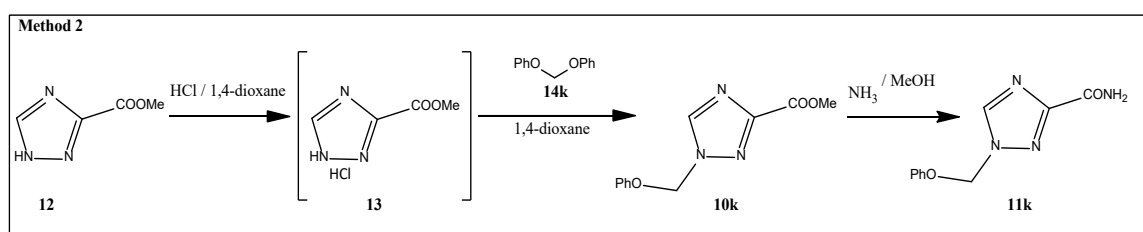
In the case of oxymethyl substituents introduction into the 1,2,4-triazole ring, which is necessary step in compounds **11** synthesis, an alkylation can occur at any nitrogen of the triazole ring, leading to a formation of three regioisomers [16,17]. According to literature sources, the method of introducing an oxymethyl substituent via triazole carboxylic acid esters N-silyl derivatives by oxymethylacetates is considered as the most regioselective, for example, the only product of such alkylation for methyl 1,2,4-triazole-3-carboxylate is methyl 1-alkoxymethyl 1,2,4-triazole-3-carboxylate [17].

In our study, methyl 1-oxymethyl-1,2,4-triazole-3-carboxylates **10a-j** were prepared in two steps: first, we obtained the silyl derivatives of methyl 1,2,4-triazole-3-carboxylate **7** by its treatment with 1,1,1,3,3,3-hexamethyldisilazane (HMDS) followed by an addition of oxymethylacetates **9a-j** in the presence of Lewis acid – tin tetrachloride (Figure 3). Compounds **9a-j** were synthesized from dialkoxymethanes **14a-j** [18] which, in turn, were obtained by a known method [19]. In the case of 1-([2-hydroxyethoxy]methyl)-1,2,4-triazole-3-carboxamide **1c**, compound **8** was treated with [2-(acetyloxy)ethoxy]methyl acetate **9l** [20] obtained from 1,3-dioxalane, the acetate protecting group of the ethyloxymethyl moiety was removed by ammonolysis. Methyl 1-methoxymethyl-1,2,4-triazole-3-carboxylate **10a** was isolated by column chromatography resulting 38.5% yield. The esters **10b-j** were used at the next stage without further purification. The amides **11a-j** were obtained by ammonolysis of the esters **10a-j** (Figure 3) and were purified by recrystallization from an ethanol-ethyl acetate mixture in yields ranging from 23 to 91%.



**Figure 3.** Introduction of a 1-alkoxymethyl moiety.

1-(Phenoxymethyl)-1,2,4-triazole-3-carboxamide **11k** was prepared using diphenoxymethane **14k** synthesized according to a procedure described in the literature [21]. Methyl 1,2,4-triazole-3-carboxylate hydrochloride **13** was treated with **14k** to give the ester 1-(phenoxymethyl)-1,2,4-triazole-3-carboxylate **10k** followed by its ammonolysis (Figure 4). Amide **11k** was purified in a similar to the previous amides **16a-j** manner resulting 52% yield.

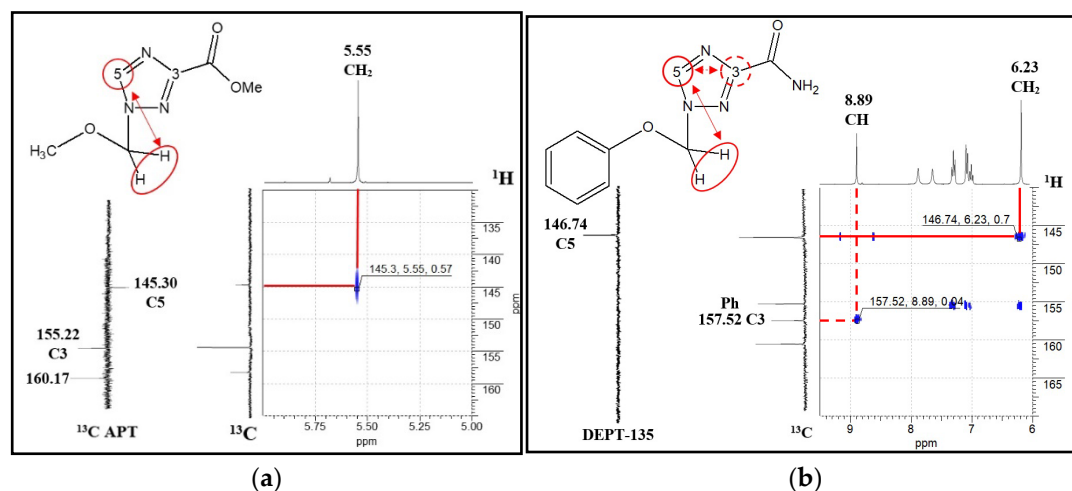


**Figure 4.** Introduction of a 1-phenoxymethyl moiety.

The structures of the obtained compounds were established using a set of physicochemical methods: <sup>1</sup>H and <sup>13</sup>C NMR, HRMS. A combination of APT and <sup>1</sup>H-<sup>13</sup>C HMBC NMR experiments was used to establish the position of the oxymethyl substituent in the case of compound **10a**. We identified

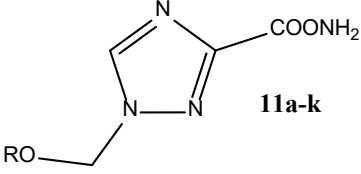


structures **11b-j** and **1c** as the position 1 isomers based on the similarity of their NMR characteristics to those of **11a** (**11a** was obtained by ammonolysis of **10a**). In the case of **11k**, the position of the phenoxyethyl radical was established by DEPT-135 and  $^1\text{H}$ - $^{13}\text{C}$  HMBC NMR.



**Figure 5.** APT, DEPT-135 and  $^1\text{H}$ - $^{13}\text{C}$  HMBC NMR spectra fragments: (a) methyl 1-(methoxymethyl)-1,2,4-triazole-3-carboxylate and (b) 1-(phenoxyethyl)-1,2,4-triazole-3-carboxamide.

**Table 2.** Synthesized 1-oxymethyl-1,2,4-triazole-3-carboxamides **11**.

		
S. No.	R	Yield
<b>11a</b>	Me	87%
<b>11b</b>	Et	78%
<b>11c</b>	n-Pr	78%
<b>11d</b>	i-Pr	91%
<b>11e</b>	n-Bu	81%
<b>11f</b>	t-Bu	49%
<b>11g</b>	n-C <sub>10</sub> H <sub>21</sub>	78%
<b>11h</b>	Bn	89%
<b>11i</b>	cyclo-C <sub>5</sub> H <sub>10</sub>	82%
<b>11j</b>	cyclo-C <sub>6</sub> H <sub>12</sub>	53%
<b>11k</b>	Ph	52%
<b>1c</b>	HO(CH <sub>2</sub> ) <sub>2</sub>	83%

## 2.2. In Vitro Stu

### 2.2.1. Anti-Cancer Activity In Vitro

The in vitro cytotoxic activities of the synthesized compounds were evaluated on CCRF-SB and K562 cells using MTT assay. Compounds **11g** and **6g** showed the highest cytotoxic activity in the leukemia cell lines after 24 h exposition, as shown in Table 3. CC<sub>50</sub> values for **11g** were calculated as 13.6±0.3  $\mu\text{M}$  in the K562 cells and 112±19  $\mu\text{M}$  in the CCRF-SB cells, respectively. CC<sub>50</sub> for **6g** were 391±15  $\mu\text{M}$  in K562 cell line.

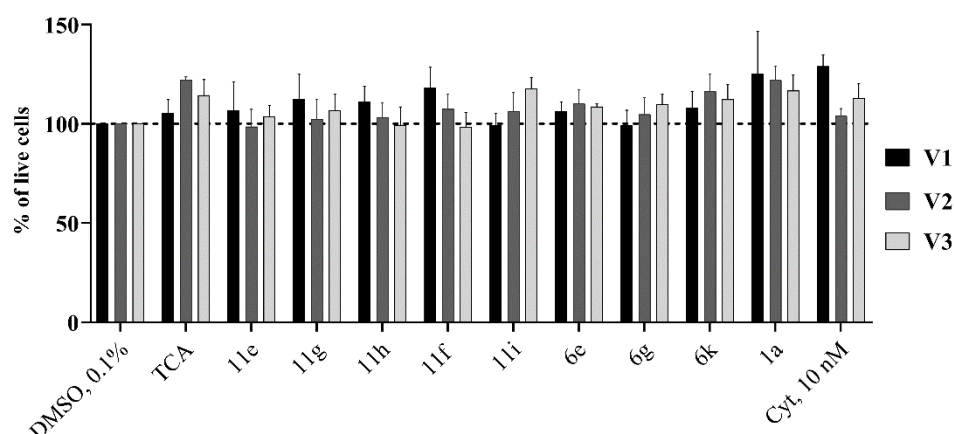
**Table 3.** CC<sub>50</sub> values for compounds after incubation with cells for 24 and 72h (MTT assay, n = 3).

\*Data was previously obtain and published in [9].

For other compounds CC<sub>50</sub> values were not determined. However, a number of compounds showed a dose-response cytostatic effect on leukemia cells at 72 h exposure assuming that newly synthesized 1,2,4-triazole-3-carboxamide derivatives may possess antiproliferative effect associated with low toxicity. MTT assay after 72 h revealed the cytotoxic effects of **11e**, **6g**, and **6k** in acute lymphoblastic leukemia cell line, and cytotoxic effects of **11i**, **11h**, **11f**, **6e** and **6k** in chronic myeloid leukemia cells (data not shown). Consequently, compounds **11e**, **11g**, **11i**, **11h**, **11f**, **6e**, **6g** and **6k** were selected for study of its antiproliferative activity.

To evaluate the effect of compounds on non-transformed cells, human peripheral blood mononuclear cells (PBMC) were isolated from the whole blood of 3 healthy volunteers. Then PBMCs were incubated 72 h with 1,2,4-triazole-3-carboxamide derivatives in the highest concentrations (500  $\mu$ M). The normal cells were less sensitive to compounds than cancer cells (Figure 6), highlighting the selectivity of action of novel compounds on malignant cells.

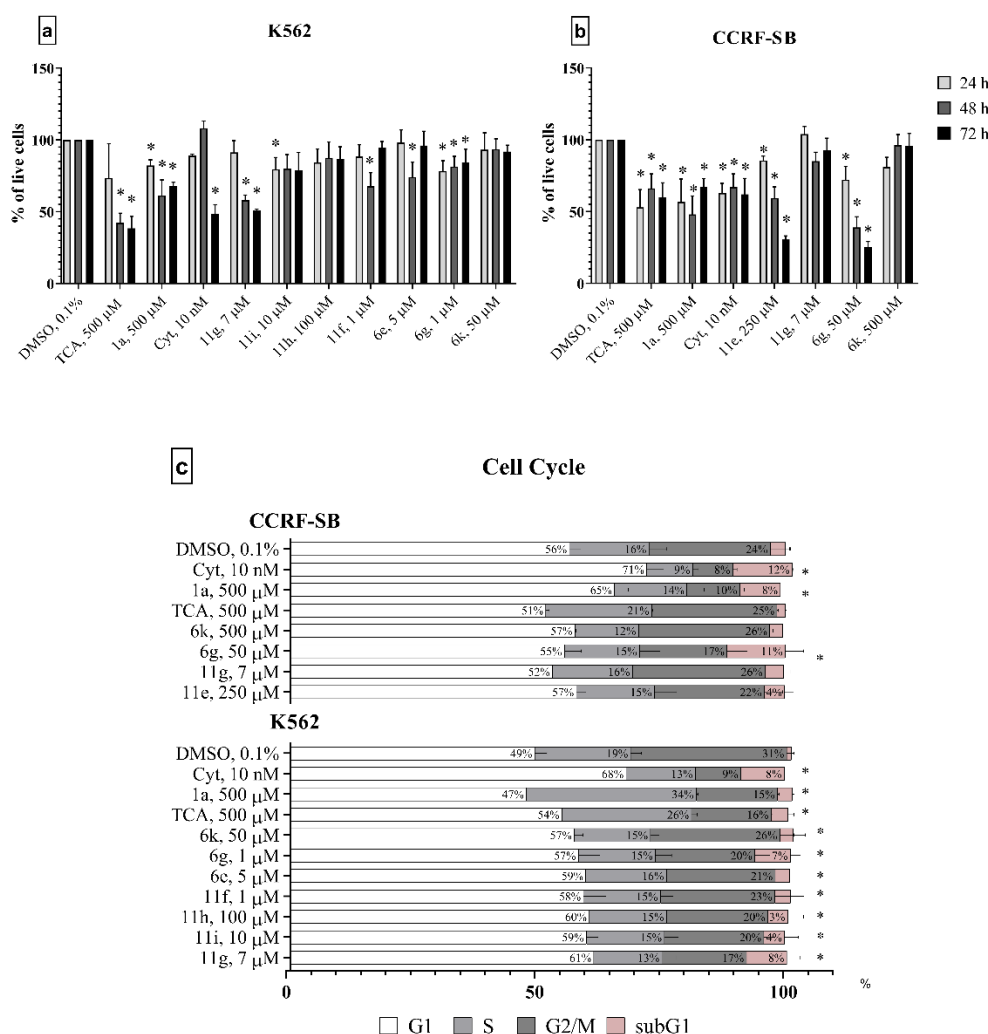
## Human PBMCs proliferation



**Figure 6.** Effect of compounds on human PBMCs proliferation. Cells were treated with the DMSO, Cyt, 1a or its derivatives for 72 h and then were counted using trypan blue exclusion test. V - volunteer. All data are expressed as percent of DMSO treated control. Significant differences were analyzed by the one-way ANOVA test. \*- significant differences from the control ( $p < 0.05$ ).

We conducted in vitro trypan blue exclusion assay to test the cytostatic (antiproliferative) activity of newly synthesized 1,2,4-triazole-3-carboxamide derivatives. The cells were incubated with the active compounds for 24-72 h at a concentrations equal to the calculated  $CC_{20}$  value. Compound **11g** significantly reduced cell proliferation in chronic myeloid leukemia cells and caused minimal cell death in acute lymphoblastic leukemia cell line. Compounds **6g** and **11e** showed dose-dependent antiproliferative action in CCRF-SB cell line (Figure 7a, b).





**Figure 7. (a-b)** Antiproliferative effects of compounds in CCRF-SB and K562 cancer cells. The cells were cultured with the solvent (DMSO), cytarabine (Cyt), or ribavirin (**1a**), or its derivatives. Cells were stained with trypan blue and counted after 24, 48 and 72 hours of the treatment. **(c)** Effect of selected compounds on cell cycle progression in K562 and CCRF-CEM cells after 72 h of incubation with the DMSO, Cyt, **1a** or its derivatives. Cells were fixed with ethanol and then stained with propidium iodide and analyzed by flow cytometry. All data are expressed as percent of DMSO treated control. Significant differences were analyzed by the one-way ANOVA test. \*- significant differences from the control ( $p < 0.05$ ).

To investigate the mechanism underlying the cell growth inhibition induced by 1,2,4-triazole-3-carboxamide derivatives, the cell cycle profile was analyzed by flow cytometry with PI staining. The K562 and CCRF-SB cells were exposed to compounds for 72h. In Figure 6c it is demonstrated that the compound **6g** caused an increase in the cell population in the G0 phase, indicating cell death in CCRF-SB culture. The population of the G2/M and S phases of CCRF-SB cells reduced after the treatment with 7  $\mu$ M **6g** compared with control. At the same time, all compounds increased accumulation of cells in G1 phase and caused a decrease in the percentage of cells in the G2/M phase in K562 cell line. The treatment of K562 cells with 7  $\mu$ M **11g** significantly reduced the fraction of cells in the S and G2/M phases and increased the proportion of cells in the G1 phase. Furthermore, **6g** and **11g** significantly increased accumulation of cells in the subG1 phase corresponding to apoptotic cells by 7 and 8 times respectively in K562 cells. Compounds **6g** and **11g** ability to induce cell death is concordant with cytotoxicity determined by the MTT assay.

2.2.2. Antimicrobial Effects Studies

The multivalency of the biological effects of ribavirin – the parent structure of the 1- or 5-oxymethyl-1,2,4-triazole-3-carboxamides – prompted us to study their antimicrobial properties. The antimicrobial potential of compounds **6a-e**, **g-h**, **k** and **11a-k** was investigated in comparison with that of reference molecules **1a** and **1c** against the following series of microorganisms: *Micrococcus luteus* ATCC 9341, *Staphylococcus aureus* INA 00985, *Pseudomonas aeruginosa* ATCC 27853, *Candida albicans* ATCC 14053 on agarose nutrient medium at concentrations of 25 mM (Table 4).

**Table 4.** 1-Oxymethyl-1,2,4-triazole-3-carboxamides antimicrobial effects.

S. No.	Zone of growth inhibition, mm			
	<i>S. aureus</i>	<i>M. luteus</i>	<i>P. aeruginosa</i>	<i>C. albicans</i>
1-oxymethyl-1,2,4-triazole-3-carboxamides				
<b>1a</b>	-	-	25±1	30±1
<b>11a</b>	-	-	-	-
<b>11b</b>	-	-	-	-
<b>11c</b>	-	-	12±1	-
<b>11d</b>	-	-	-	-
<b>11e</b>	-	-	-	-
<b>11f</b>	-	-	-	-
<b>11g</b>	-	-	-	-
<b>11h</b>	-	-	-	-
<b>11i</b>	-	12±1	-	-
<b>11j</b>	-	12±1	-	-
<b>11k</b>	-	-	-	-
<b>1c</b>	-	12±1	-	-

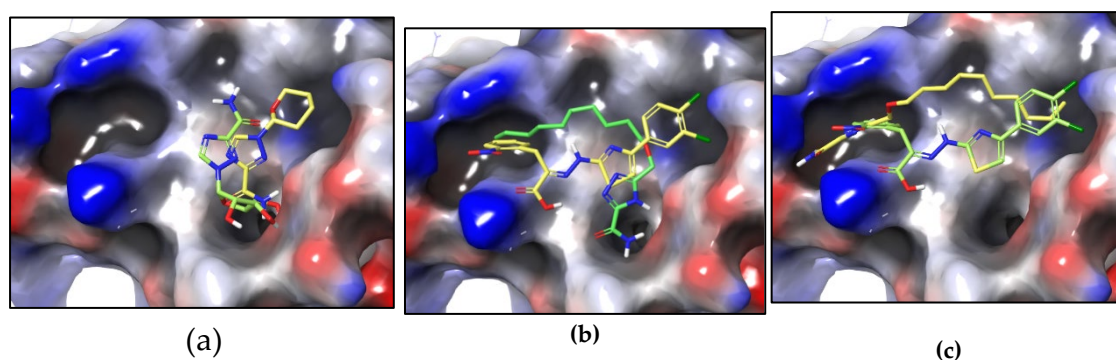
Compounds **6** have not showed any suppression of microorganism growth. Compounds **11i, j** and **1c** showed bacteriostatic activity against the Gram-positive organism *M. luteus*, but not against *S. aureus*, in contrast to ribavirin **1a**, which showed no antimicrobial activity against such organisms. In the case of the Gram-negative microorganism *P. aeruginosa*, moderate activity compared to that of **1a** was observed for compound **11c**. In relation to *C. albicans*, the studied compounds **1c, 11** showed no activity, ribavirin **1a** showed the highest activity.

2.3. Molecular Docking

Due to the revealed effects of the synthesized compounds **6a-e**, **g-h**, **k** and **11a-k**, **1c** towards acute lymphoblastic leukemia and chronic myeloid leukemia cell lines, we became interested in trying to assume the mechanism underlying the action. As known, ribavirin undergoes phosphorylation in cells to form ribavirin 5-monophosphate (RMP) [22]. A number of cellular targets have been shown for RMP, including inosine-5'-phosphate dehydrogenase (IMPDH) and eukaryotic translation initiation factor 4E (eIF4F) [23–25]. Inhibition of IMPDH occurs due to insertion into the inosine monophosphate binding site, and eIF4E is presumably according to various sources either due to insertion into the 5'-cap mRNA binding site or by interfering with the assembly of protein subunits of the factor [26–30]. Oxymethyl derivatives of TCA **6a-e,g,h,k** and **11a-k** do not have a hydroxyl substituent and therefore cannot be phosphorylated. Therefore, the main opportunity for them to participate in these biochemical pathways remains to block the interaction of the eIF4E and eIF4G subunits of factor 4E by binding to at least one of them [31]. An example of such an effect of low molecular weight compounds is a inhibitor of this interaction, 4EGI-1 [32]. 4EGI-1 disrupts the eIF4E/eIF4G association in vitro and in vivo, and reduces viability of a wide range of cancer cells such as breast cancer and multiple myeloma [33]. 4EGI-1 inhibits tumor growth in in vivo models of acute myeloid leukemia and chronic lymphocytic leukemia. [34–37]. Therefore, we used a region of the protein surface characteristic of 4EGI-1 binding as a target for modeling. The structure of eIF4E

protein (PDB: 4TPW) and the structures of low molecular weight ligands **1g**, **h** and **2g**, **h**, **6**, **11** optimized with OPLS3e force field were used for molecular docking which was performed in Schrodinger Maestro.

According to the simulation results, several newly synthesized compounds **6** and **11** as well as ribavirin **1a** and its C-nucleoside analogue **2a** demonstrate a preferential localization in the binding site similar to the known inhibitor 4EGI-1. Compound **1g** in simulation binds mirrorly to ribavirin **1a** binding. **1g** and **1a** occupies the same cavities in the binding site (Figure 8a). Compound **6g** with its carboxamide fragment falls into the binding region of ribose fragment **1a**, similarly to **1g**, **h** and **2g**, **h**, and its lipophilic tail binds in the cavity characteristic for 4EGI-1 inhibitor formed by Asp 51, Asn 59 and Lys 49. It is in accordance with the results of higher cytotoxicity of **6g** compared to compounds **1g**, **h** and **2g**, **h** (Figure 8b). Compound **11g** demonstrating higher toxicity in cancer cells *in vitro*, is located at the binding site in a manner similar to the known 4EGI-1 inhibitor. The lipophilic tail of **11g** enters the hydrophobic cavity of the binding site formed by amino acid residues Phe 47, Tyr 91, Ile 79 and 63, while the formation of hydrogen bonds occurs via triazole fragment in the cavity formed by Asp 51, Asn 59 and Lys 49, similarly to 4EGI-1 (Figure 8c).



**Figure 8.** Binding of test compounds to protein eIF4E: (a) ribavirin **1a** – green and **1g** – yellow; (b) known inhibitor 4EGI-1 – yellow and **6g** – green; (c) known inhibitor 4EGI-1 – green and **11g** – yellow.

Thus, as a result of the modeling, it was possible to trace the correlation between the pattern of location of hydroxymethyl analogues of ribavirin on the surface of eIF4E and their *in vitro* toxicity to cancer cells. This suggests that a possible mechanism of action of the synthesized compounds may be associated with inhibition of RNA translation due to disruption of the assembly of the eIF4F complex.

### 3. Materials and Methods

#### 3.1. Synthetic Section

##### 3.1.1. 5-(n-Propoxymethyl)-1,2,4-Triazole-3-Carboxamide **6c**

1.88 g (11.02 mmol) of n-propyloxyacetic acid chloroanhydride was added dropwise to a suspension of 1.19 g (5.15 mmol) of  $\beta$ -N-(t-butyloxycarbonyl)ethyloxalamidrazone **4** in anhydrous pyridine while cooling the reaction mixture to 0°C. The reaction mass was brought to boiling point and stirred for 20 h. After the reaction was completed (control was carried out using TLC), the solvent was removed on a vacuum rotary evaporator. The residue was dissolved in 1 M aqueous HCl solution and extracted 3 times with ethyl acetate in equal portions. The organic phases were combined and dried using Na<sub>2</sub>SO<sub>4</sub>, the solvent was removed on a vacuum rotary evaporator. The crude product **5c** was isolated by flash chromatography on silica gel using chloroform/methanol system (with a methanol gradient from 0 to 7%) as an eluent. Crude **5c** was dissolved in 2 ml of a 10 M ammonia methanol solution and heated to boiling under reflux for 12 hours, the solvent was removed using a

vacuum rotary evaporator. Residue was suspended in anhydrous acetone, filtered and dried in a desiccator under reduced pressure above NaOH for 12 hours to yield 0.23 g (24%) **6c** as white crystals.

$R_f$  = 0.61 (1% CH<sub>3</sub>OH in CHCl<sub>3</sub>), mp 110-111°C. <sup>1</sup>H NMR spectrum (DMSO-d<sub>6</sub>)  $\delta$ : 0.83 (t, 3H, J=7.09, CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>); 1.50 (se, 2H, J=7.09, CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>); 3.40 (t, 2H, J=6.09, CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>); 4.51 (s, 2H, OCH<sub>2</sub>); 7.70 and 8.01 (2s, 2H, NH<sub>2</sub>). <sup>13</sup>C NMR spectrum (DMSO-d<sub>6</sub>)  $\delta$ : 10.54; 22.38; 63.66; 72.13; 153.22; 156.84; 159.39. HRMS: for C<sub>7</sub>H<sub>12</sub>N<sub>4</sub>O<sub>2</sub> m/z [M+H]<sup>+</sup> calculated: 185.0960; found: 185.0981; LC 5-(n-propoxymethyl)-1,2,4-triazole-3-carboxamide content: spectrophotometric detection 235 nm no less than 98%.

### 3.1.2. Methyl 1-(Methoxymethyl)-1,2,4-Triazole-3-Carboxylate (10a)

0.5 g (3.9 mmol) of methyl 1,2,4-triazole-3-carboxylate was suspended in 4 ml (19 mmol) of HMDS and stirred under reflux for 1 h. After cooling the excess of HMDS was removed using a rotary evaporator. 5 ml of anhydrous acetonitrile, 1.70 ml (18 mmol) of **9a**, 0.45 ml (3.9 mmol) of SnCl<sub>4</sub> were added to the residue and the reaction was stirred under reflux until the starting ester was no longer detectable by TLC. The reaction mass was poured into 10 ml of saturated sodium bicarbonate solution and the precipitates formed were filtered off. The filtrate was extracted with chloroform (4x10 ml), the combined chloroform extracts were washed with water (10 ml) and dried over CaCl<sub>2</sub>. Volatile components were evaporated. 0.25 g (38.5%) of the product **10a** was isolated by column chromatography on silica gel, eluent: toluene-acetone, modified with 1% triethylamine (acetone gradient from 5 to 7%), as transparent oil.

$R_f$  = 0.26 (30% acetone in toluene). <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>)  $\delta$ : 3.40 (s, 3H, OCH<sub>3</sub>); 4.99 (s, 3H, COOCH<sub>3</sub>); 5.53 (s, 2H, OCH<sub>2</sub>); 8.34 (s, 1H, CH). <sup>13</sup>C NMR spectrum (CDCl<sub>3</sub>)  $\delta$ : 52.86; 57.66; 80.39; 145.89; 154.89; 159.93. For C<sub>6</sub>H<sub>9</sub>N<sub>3</sub>O<sub>3</sub> m/z [M+H]<sup>+</sup> calculated: 172.1; found: 172.0.

### 3.1.3. 1-(Methoxymethyl)-1,2,4-Triazole-3-Carboxamide (11a)

0.2 g (1.2 mmol) of methyl 1-(methoxymethyl)-1,2,4-triazole-3-carboxylate was dissolved in 1.5 ml of 10 M ammonia solution in methanol and stirred at room temperature to conversion of the starting material (control by TLC). Volatile components were removed on a rotary evaporator, 0.15g (87%) of the product **11a** was isolated by recrystallization from a solvent mixture: ethanol-ethyl acetate as white crystals.

$R_f$  = 0.53 (1% CH<sub>3</sub>OH in CHCl<sub>3</sub>), mp 146-147°C. <sup>1</sup>H NMR spectrum (DMSO-d<sub>6</sub>)  $\delta$ : 3.29 (s, 3H, OCH<sub>3</sub>); 5.51 (s, 2H, OCH<sub>2</sub>); 7.51 and 7.67 (2s, 2H, NH<sub>2</sub>); 8.80 (s, 1H, CH). <sup>13</sup>C NMR spectrum (DMSO-d<sub>6</sub>)  $\delta$ : 56.58; 79.16; 146.21; 157.48; 160.41. HRMS: for C<sub>5</sub>H<sub>8</sub>N<sub>4</sub>O<sub>2</sub> m/z [M+H]<sup>+</sup> calculated: 157.0726; found: 157.0733; LC 1-(methoxymethyl)-1,2,4-triazole-3-carboxamide content: spectrophotometric detection 235 nm no less than 98%.

### 3.1.4. General Procedure for the Preparation of 1-Substituted of 1,2,4-Triazole-3-Carboxamides 11b-j, 1c

Methyl 1,2,4-triazole-3-carboxylate was suspended in 5 eq. HMDS and stirred under reflux for 1 hour in an anhydrous atmosphere. After cooling the excess of HMDS was removed using a rotary evaporator. Anhydrous acetonitrile, 5 eq. **9a**, 1 eq. SnCl<sub>4</sub> were added to the residue and the reaction was stirred under reflux until the starting ester was no longer detectable by TLC. The reaction mass was poured into saturated sodium bicarbonate solution and the precipitates formed were filtered off. The filtrate was extracted with chloroform, the combined chloroform extracts were washed with water (10 ml) and dried over CaCl<sub>2</sub>. Volatile components were evaporated. The product was isolated by column chromatography on silica gel, eluent: toluene-acetone, modified with 1% triethylamine (acetone gradient from 5 to 7%).

#### *1-(ethoxymethyl)-1,2,4-triazole-3-carboxamide (11b).*

From 0.5 g (2.5 mmol) of methyl 1,2,4-triazole-3-carboxylate, 0.38 mg (78%) of product **11b** was obtained as white crystals.

$R_f = 0.52$  (1%  $\text{CH}_3\text{OH}$  in  $\text{CHCl}_3$ ), mp  $127^\circ\text{C}$ .  $^1\text{H}$  NMR spectrum ( $\text{DMSO}-d_6$ )  $\delta$ : 1.08 (t,  $J=7.03$ , 2H,  $\text{CH}_3\text{CH}_2$ ); 3.29 (s, 3H,  $\text{OCH}_3$ ); 3.54 (q,  $J=7.03$ , 2H,  $\text{CH}_3\text{CH}_2$ ); 5.55 (s, 2H,  $\text{OCH}_2$ ); 7.61 and 7.75 (2s, 2H,  $\text{NH}_2$ ); 8.79 (s, 1H, CH).  $^{13}\text{C}$  NMR spectrum ( $\text{DMSO}-d_6$ )  $\delta$ : 14.57; 64.50; 77.64; 146.01; 157.37; 160.37. HRMS: for  $\text{C}_6\text{H}_{10}\text{N}_4\text{O}_2$   $m/z$   $[\text{M}+\text{H}]^+$  calculated: 171.0882; found: 171.0893; LC 1-(ethoxymethyl)-1,2,4-triazole-3-carboxamide content: spectrophotometric detection 235 nm no less than 98%.

*1-(n-propyloxymethyl)-1,2,4-triazole-3-carboxamide (11c).*

From 1 g (7.8 mmol) of methyl 1,2,4-triazole-3-carboxylate, 0.36 mg (78%) of product **11c** was obtained as white crystals.

$R_f = 0.69$  (1%  $\text{CH}_3\text{OH}$  in  $\text{CHCl}_3$ ), mp  $125-126^\circ\text{C}$ .  $^1\text{H}$  NMR ( $\text{DMSO}-d_6$ )  $\delta$ : 0.80 (t, 3H,  $J = 7.41$ ,  $\text{OCH}_2\text{CH}_2\text{CH}_3$ ); 1.41-1.53 (m, 2H,  $\text{OCH}_2\text{CH}_2\text{CH}_3$ ); 3.44 (t, 2H,  $J = 6.60$ ,  $\text{OCH}_2\text{CH}_2\text{CH}_3$ ); 5.55 (s, 2H,  $\text{OCH}_2$ ); 7.57 and 7.79 (2s, 2H,  $\text{NH}_2$ ); 8.79 (s, 1H, CH).  $^{13}\text{C}$  NMR ( $\text{DMSO}-d_6$ )  $\delta$ : 10.19; 22.00; 70.61; 77.88; 145.99; 157.35; 160.35. HRMS: for  $\text{C}_5\text{H}_8\text{N}_4\text{O}_2$   $m/z$   $[\text{M}+\text{H}]^+$  calculated: 185.1038; found: 185.1048. LC 1-(n-propyloxymethyl)-1,2,4-triazole-3-carboxamide content: spectrophotometric detection 235 nm no less than 97%.

*1-(isopropyloxymethyl)-1,2,4-triazole-3-carboxamide (11d).*

From 1 g (7.8 mmol) of methyl 1,2,4-triazole-3-carboxylate, 0.42 mg (91%) of product **11d** was obtained as white crystals.

$R_f = 0.65$  (1%  $\text{CH}_3\text{OH}$  in  $\text{CHCl}_3$ ), mp  $145^\circ\text{C}$ .  $^1\text{H}$  NMR ( $\text{DMSO}-d_6$ )  $\delta$ : 1.06 (d, 3H,  $J = 6.12$ ,  $\text{OCHCH}_3$ ); 3.77-3.81 (m, 1H,  $J = 6.11$ ,  $\text{OCH}$ ); 5.56 (s, 2H,  $\text{OCH}_2$ ); 7.57 and 7.79 (2s, 2H,  $\text{NH}_2$ ); 8.79 (s, 1H, CH).  $^{13}\text{C}$  NMR ( $\text{DMSO}-d_6$ )  $\delta$ : 21.94; 70.36; 75.75; 145.98; 157.33; 160.47. HRMS: for  $\text{C}_5\text{H}_8\text{N}_4\text{O}_2$   $m/z$   $[\text{M}+\text{H}]^+$  calculated: 185.1039; found: 185.1058. LC 1-(isopropyloxymethyl)-1,2,4-triazole-3-carboxamide content: spectrophotometric detection 235 nm no less than 98%.

*1-(n-butyloxymethyl)-1,2,4-triazole-3-carboxamide (11e).*

From 1 g (7.8 mmol) of methyl 1,2,4-triazole-3-carboxylate, 0.53 mg (81%) of product **11e** was obtained as white crystals.

$R_f = 0.5$  (1%  $\text{CH}_3\text{OH}$  in  $\text{CHCl}_3$ ), mp  $123-126^\circ\text{C}$ .  $^1\text{H}$  NMR spectrum (300 MHz,  $\text{DMSO}-d_6$ )  $\delta$ : 0.80 (t,  $J=7.41$ , 2H,  $\text{CH}_3\text{CH}_2$ ); 1.47 (q,  $J=6.85$ , 2H,  $\text{CH}_3\text{CH}_2$ ); 3.44 (t,  $J=6.60$ , 2H,  $\text{CH}_2\text{CH}_2$ ); 5.55 (s, 2H,  $\text{OCH}_2$ ); 7.58 and 7.79 (2s, 2H,  $\text{NH}_2$ ); 8.79 (s, 1H, CH).  $^{13}\text{C}$  NMR spectrum (75 MHz,  $\text{DMSO}-d_6$ )  $\delta$ : 10.57; 22.00; 70.61; 77.88; 145.99; 157.35; 160.35. HRMS: for  $\text{C}_8\text{H}_{14}\text{N}_4\text{O}_2$   $m/z$   $[\text{M}+\text{H}]^+$  calculated: 199.1195; found: 199.1205. LC 1-(n-butyloxymethyl)-1,2,4-triazole-3-carboxamide content: spectrophotometric detection 235 nm no less than 97%.

*1-(tert-butoxymethyl)-1,2,4-triazole-3-carboxamide (11f).*

From 1 g (7.8 mmol) of methyl 1,2,4-triazole-3-carboxylate, 0.39 mg (49%) of product **11f** was obtained as white crystals.

$R_f = 0.65$  (1%  $\text{CH}_3\text{OH}$  in  $\text{CHCl}_3$ ), mp  $194-195^\circ\text{C}$ .  $^1\text{H}$  NMR ( $\text{DMSO}-d_6$ )  $\delta$ : 1.18 (s, 9H,  $\text{O}(\text{CH}_3)_3$ ); 5.57 (s, 2H,  $\text{OCH}_2$ ); 7.55 and 7.75 (2s, 2H,  $\text{NH}_2$ ); 8.76 (s, 1H, CH).  $^{13}\text{C}$  NMR ( $\text{DMSO}-d_6$ )  $\delta$ : 27.28; 72.94; 73.51; 132.74; 157.63; 159.84. HRMS: for  $\text{C}_8\text{H}_{14}\text{N}_4\text{O}_2$   $m/z$   $[\text{M}+\text{H}]^+$  calculated: 199.1195; found: 199.1208. LC 1-(tert-butoxymethyl)-1,2,4-triazole-3-carboxamide content: spectrophotometric detection 235 nm no less than 96%.

*1-(n-decyloxymethyl)-1,2,4-triazole-3-carboxamide (11g).*

From 1 g (7.8 mmol) of methyl 1,2,4-triazole-3-carboxylate, 0.36 mg (78%) of product **11g** was obtained as white crystals.

$R_f = 0.60$  (1%  $\text{CH}_3\text{OH}$  in  $\text{CHCl}_3$ ), mp  $122-124^\circ\text{C}$ .  $^1\text{H}$  NMR ( $\text{DMSO}-d_6$ )  $\delta$ : 0.84 (t, 3H,  $J = 6.83$ ,  $\text{O}(\text{CH}_2)_9\text{CH}_3$ ); 1.20 (s, 14H,  $\text{OCH}_2\text{CH}_2(\text{CH}_2)_7\text{CH}_3$ ); 1.42-1.46 (m, 2H,  $\text{OCH}_2\text{CH}_2(\text{CH}_2)_7\text{CH}_3$ ); 0.37 (t, 2H,  $J = 6.50$ ,  $\text{OCH}_2\text{CH}_2(\text{CH}_2)_7\text{CH}_3$ ); 7.57 and 7.77 (2s, 2H,  $\text{NH}_2$ ); 8.78 (s, 1H, CH).  $^{13}\text{C}$  NMR ( $\text{DMSO}-d_6$ )  $\delta$ : 22.00; 28.59; 28.84; 68.97; 145.98; 160.34. HRMS: for  $\text{C}_{14}\text{H}_{26}\text{N}_4\text{O}_2$   $m/z$   $[\text{M}+\text{H}]^+$  calculated: 283.2134; found: 283.2150. LC 1-n-decyloxymethyl-1,2,4-triazole-3-carboxylic acid amide content: spectrophotometric detection 235 nm no less than 96%.

*1-(benzyloxymethyl)-1,2,4-triazole-3-carboxamide (11h).*



From 1 g (7.8 mmol) of methyl 1,2,4-triazole-3-carboxylate, 0.63 mg (89%) of product **11h** was obtained as white crystals.

$R_f$  = 0.65 (1% CH<sub>3</sub>OH in CHCl<sub>3</sub>), mp 168-169°C. <sup>1</sup>H NMR (DMSO-d<sub>6</sub>)  $\delta$ : 4.60 (s, 2H, CH<sub>2</sub>C<sub>6</sub>H<sub>5</sub>); 5.67 (s, 2H, OCH<sub>2</sub>); 7.26-7.37 (m, 2H, C<sub>6</sub>H<sub>5</sub>); 7.60 and 7.82 (2s, 2H, NH<sub>2</sub>); 8.83 (s, 1H, CH). <sup>13</sup>C NMR (DMSO-d<sub>6</sub>)  $\delta$ : 70.69; 77.47; 127.65; 128.28; 136.87; 146.19; 157.46; 160.38. HRMS: for C<sub>11</sub>H<sub>12</sub>N<sub>4</sub>O<sub>2</sub> m/z [M+H]<sup>+</sup> calculated: 233.1039; found: 233.1089. LC 1-(benzyloxymethyl)-1,2,4-triazole-3-carboxamide content: spectrophotometric detection 235 nm no less than 97%.

*1-(cyclopentyloxymethyl)-1,2,4-triazole-3-carboxamide (11i).*

From 1 g (7.8 mmol) of methyl 1,2,4-triazole-3-carboxylate, 0.31 mg (53%) of product **11i** was obtained as white crystals.

$R_f$  = 0.66 (1% CH<sub>3</sub>OH in CHCl<sub>3</sub>), mp 153-154°C. <sup>1</sup>H NMR (DMSO-d<sub>6</sub>)  $\delta$ : 1.46-1.66 (m, 8H, OC<sub>5</sub>H<sub>9</sub>); 4.08 (s, 1H, OCH); 5.54 (s, 2H, OCH<sub>2</sub>); 7.57 and 7.79 (2s, 2H, NH<sub>2</sub>); 8.79 (s, 1H, CH). <sup>13</sup>C NMR (DMSO-d<sub>6</sub>)  $\delta$ : 22.90; 31.78; 76.42; 79.97; 145.99; 157.31; 160.40. HRMS: for C<sub>9</sub>H<sub>10</sub>N<sub>4</sub>O<sub>2</sub> m/z [M+H]<sup>+</sup> calculated: 211.1195; found: 211.1208. LC 1-(cyclopentyloxymethyl)-1,2,4-triazole-3-carboxamide content: spectrophotometric detection 235 nm no less than 98%.

*1-(cyclohexyloxymethyl)-1,2,4-triazole-3-carboxamide (11j).*

From 1 g (7.8 mmol) of methyl 1,2,4-triazole-3-carboxylate, 0.58 mg (82%) of product **11j** was obtained as white crystals.

$R_f$  = 0.46 (1% CH<sub>3</sub>OH in CHCl<sub>3</sub>), mp 155-156°C. <sup>1</sup>H NMR (DMSO-d<sub>6</sub>)  $\delta$ : 1.70-1.17 (m, 10H, C<sub>5</sub>H<sub>10</sub>); 3.49-3.51 (m, 1H, OCH); 5.59 (s, 2H, OCH<sub>2</sub>); 7.62 and 7.85 (2s, 2H, NH<sub>2</sub>); 8.81 (s, 1H, CH). <sup>13</sup>C NMR (DMSO-d<sub>6</sub>)  $\delta$ : 23.26; 25.06; 31.61; 75.64; 145.97; 157.33; 160.48. HRMS: for C<sub>10</sub>H<sub>16</sub>N<sub>4</sub>O<sub>2</sub> m/z [M+H]<sup>+</sup> calculated: 225.1352; found: 225.1380. LC 1-(cyclohexyloxymethyl)-1,2,4-triazole-3-carboxamide content: spectrophotometric detection 235 nm no less than 96%.

*1-([2-hydroxyethoxy]methyl)-1,2,4-triazole-3-carboxamide (1c).*

From 0.2 g (0.99 mmol) of methyl 1,2,4-triazole-3-carboxylate, 0.58 mg (83%) of product **1c** was obtained as white crystals.

$R_f$  = 0.35 (5% CH<sub>3</sub>OH in CHCl<sub>3</sub>), mp 154-156°C. <sup>1</sup>H NMR spectrum (DMSO-d<sub>6</sub>)  $\delta$ : 3.44-3.55 (m, 4H, -OCH<sub>2</sub>CH<sub>2</sub>O-); 5.59 (s, 2H, OCH<sub>2</sub>); 7.57 and 7.79 (2s, 2H, NH<sub>2</sub>); 8.79 (s, 1H, CH). <sup>13</sup>C NMR spectrum (DMSO-d<sub>6</sub>)  $\delta$ : 59.79; 70.98; 78.09; 146.01; 158.36; 160.39. HRMS: for C<sub>6</sub>H<sub>10</sub>N<sub>4</sub>O<sub>3</sub> m/z [M+H]<sup>+</sup> calculated: 187.0831; found: 187.0838; LC 1-([2-hydroxyethoxy]methyl)-1,2,4-triazole-3-carboxamide content: spectrophotometric detection 235 nm no less than 97%.

### 3.1.5. Methyl 1-(Phenoxymethyl)-1,2,4-Triazole-3-Carboxylate (10k)

2 g (16 mmol) methyl 1,2,4-triazole-3-carboxylate was suspended in 5 ml of a 3.4 M hydrogen chloride solution in 1,4-dioxane and stirred under reflux for 1 hour. The excess of 1,4-dioxane was removed using a rotary evaporator. 3.2 ml (16 mmol) of diphenoxymethane **14k**, 5 ml of 1,4-dioxane were added to the residue and the reaction was stirred under reflux until the starting ester was no longer detectable by TLC. Volatile components were evaporated. 1.2 g (32%) of the product **10k** was isolated by column chromatography on silica gel, eluent: toluene-acetone, modified with 1% triethylamine (acetone gradient from 5 to 7%), as transparent oil.

$R_f$  = 0.62 (30% acetone in toluene). <sup>1</sup>H NMR spectrum (CDCl<sub>3</sub>)  $\delta$ : 3.99 (s, 3H, COOCH<sub>3</sub>); 6.46 (s, 2H, OCH<sub>2</sub>); 6.85-7.34 (m, 5H, Ph); 8.07 (s, 1H, CH). <sup>13</sup>C NMR spectrum (75 MHz, CDCl<sub>3</sub>)  $\delta$ : 53.33; 81.77; 117.63; 122.93; 130.34; 148.57; 156.56; 157.06; 160.41. For C<sub>6</sub>H<sub>9</sub>N<sub>3</sub>O<sub>3</sub> m/z [M+H]<sup>+</sup> calculated: 234.2; found: 234.1.

### 3.1.6. 1-(Phenoxymethyl)-1,2,4-Triazole-3-Carboxamide (11k)

Prepared as **11a** from 0.74 g (3.18 mmol) of methyl 1-(phenoxymethyl)-1,2,4-triazole-3-carboxylate **10k** in 2 ml of methanolic ammonia. The yield was 0.31 g (52%) as white crystals.

$R_f$  = 0.75 (1% CH<sub>3</sub>OH in CHCl<sub>3</sub>), mp 188-192°C. <sup>1</sup>H NMR spectrum (300 MHz, DMSO-d<sub>6</sub>)  $\delta$ : 6.23 (s, 2H, OCH<sub>2</sub>); 7.03-7.35 (m, 5H, Ph); 7.67 and 7.90 (2s, 2H, NH<sub>2</sub>); 8.79 (s, 1H, CH). <sup>13</sup>C NMR spectrum (75 MHz, DMSO-d<sub>6</sub>)  $\delta$ : 75.02; 116.03; 122.63; 129.82; 146.75; 155.74; 160.26. HRMS: for C<sub>10</sub>H<sub>10</sub>N<sub>4</sub>O<sub>2</sub> m/z



[M+H]<sup>+</sup> calculated: 219.0882; found: 219.0896; 1-(Phenoxymethyl)-1,2,4-triazole-3-carboxamide content: spectrophotometric detection 235 nm no less than 95%.

### 3.2. Antiproliferative Assays

#### 3.2.1. Cell Cultures

Acute lymphoblastic leukemia (CCRF-SB) and chronic myeloid leukemia K562 cell lines were obtained from Bioresource collection of cell lines of N.N. Blokhin National Medical Research Center of Oncology. Cells were cultured in RPMI-1640 media ("Paneco", Russia) supplemented with 10% fetal bovine serum ("Biowest", France), 2 mM L-glutamine, 5 ME/ml penicillin and 5 µg/ml streptomycin ("Paneco", Russia) at 37°C and 5% CO<sub>2</sub>.

#### 3.2.2. MTT Assay

Cells were seeded in 96-well plates (15 000 cells/well) and treated with various concentrations (5 nM - 1 mM) of ribavirin or its derivatives or 0.1% solvent DMSO for 24 h. Cell viability was determined using the MTT assay. Cells were incubated at 37°C for 3 h with a solution of 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide ("Paneco", Russia) in PBS, final concentration 0.25 mg/ml in well. The supernatant was discarded and formazan was dissolved in 100 µl of DMSO. The absorbance values were measured at 570 nm on a Microplate Photometer Multiskan FC ("Thermo Fisher Scientific", USA). The percentage of viable cells were calculated as a percentage of solvent treated control. Each concentration was tested in three technical and three biological replicates.

#### 3.2.3. Cell Proliferation

Cells were seeded in 24-well plates (30 000 cells/well), treated with ribavirin or its derivatives or 0.1% solvent (DMSO) and incubated for 24 h 48 or 72 h. Cytarabine (Cyt, "SelleckChem", USA) used as a positive control (at 10 nM). Then cells stained with 0.4% trypan blue in PBS (pH 7.4) solution (1:1 v/v) and immediately counted using TC20 automatic cell counter ("Bio-Rad", USA). Each point was tested in two technical and three biological replicates.

#### 3.2.4. Cell Cycle

Cells were cultured in 24-well plates (30 000 cells/well) and treated with 0.1% DMSO (solvent control), 10 nM Cyt (positive control), ribavirin or its derivatives for 72 h. Then cells were fixed in 70% ethanol for 2 h at 4°C. Cells were then washed twice with cold PBS, pH 7.4, then staining with 500 µL cold propidium iodide (PI) solution (50 µg/mL PI, 1% Triton X-100 and 100 µg/mL RNase A in PBS). Cell cycle distribution of cells in samples were analyzed using FACSCalibur Flow Cytometer ("BD Biosciences", San Jose, CA, USA). Each point was tested in two technical and three biological replicates.

#### 3.2.5. Human Peripheral Blood Mononuclear Cells (PBMCs) Isolation and Culture

Peripheral blood samples were collected from 3 healthy volunteers (21-28-year-old, non-smoking). Monocytes were isolated by centrifugation with Ficoll-Isopaque ("Paneco", Russia) and then cultured in the RPMI-1640 media ("Paneco", Russia) supplemented with 20% FBS ("Biowest", France), 2 mM L-glutamine, 0.5 ME/ml penicillin and 0.5 µg/ml streptomycin ("Paneco", Russia), 10 mg/L phytohaemagglutinin ("Paneco", Russia). Cells were incubated with 0.1% DMSO (solvent control), 10 nM Cyt or 500 µM of **1a** or its derivatives for 72 h at 37°C and 5% CO<sub>2</sub>. Each point was tested in three technical replicates.

### 3.3. Antimicrobial Assays

Antimicrobial activity was determined with standard method of agar wells measuring the diameter of the inhibition zones. The following microorganisms from the collection of cultures of the

Gause Institute of New Antibiotics were used as test cultures: *Staphylococcus aureus* INA 00985, *Micrococcus luteus* ATCC 9341, *Pseudomonas aeruginosa* ATCC 27853, *Candida albicans* ATCC 14053. The cultures grown at 35°C on the following media: Mueller-Hinton agar (*Staphylococcus aureus* INA 00985, *Micrococcus luteus* ATCC 9341, *Pseudomonas aeruginosa* ATCC 27853), Sabouraud agar (*Candida albicans* ATCC 14053) for 24 h before assay preparation. Preparation of inoculum: the cell density of the bacterial suspension in sterile saline was 0.5 McFarland standard, completely suspend by shaking on a vortex mixer by 10-15 sec and applied to petri dishes with Mueller-Hinton agar and Mueller-Hinton agar with 2% glucose for *Candida albicans*. Plates incubated at 35°C. Growth inhibition zones size were measured after 24 h of incubation.

### 3.4. Statistical Analysis

All data were calculated as the mean  $\pm$  standard error of mean (S.E.M.). The data were analyzed using GraphPad v8.2.1 software (San Diego, CA, USA). The treatment effects in each experiment were compared by one-way Student's t-test. Differences between groups were considered significant at  $p < 0.05$ . All in vitro experiments were repeated three times in 2-3 technical replications.

## 4. Conclusions

In the present work, we synthesized two series of fully deoxy acyclic analogues of ribavirin – 5-oxymethyl **6** and 1-oxymethyl **11** derivatives of 1,2,4-triazole-3-carboxamide, and compared their anticancer and antimicrobial properties. Derivatives of series **6** apparently lose even the weak antimicrobial potential characteristic of ribavirin **1a**, while 1-oxymethyls of series **11** show antimicrobial activity against gram-positive bacteria.

Novel derivatives of 1,2,4-triazole-3-carboxamide **6g** and **11g** exhibited high cytostatic effect and antiproliferative activities in leukemia cell lines. The effect of the new compounds was comparable to ribavirin or Cyt (in the K562 line), and revealed specific cytotoxicity to leukemia cells compared to PBMC. Thus it was shown that compounds with n-decyloxymethyl radicals, regardless of the triazole ring substitution position, exhibit anticancer activity. Molecular docking results suggest that cell cycle arrest and suppression of cell proliferation may be mediated by inhibition of eIF4E like in case of ribavirin.

These results implies that 1-oxymethyl-1,2,4-triazole-3-carboxamides have potential to further development and application as anticancer agents.

**Supplementary Materials:** The following supporting information can be downloaded at the website of this paper posted on Preprints.org.

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**Data Availability Statement:** We encourage all authors of articles published in MDPI journals to share their research data. In this section, please provide details regarding where data supporting reported results can be found, including links to publicly archived datasets analyzed or generated during the study. Where no new data were created, or where data is unavailable due to privacy or ethical restrictions, a statement is still required. Suggested Data Availability Statements are available in section “MDPI Research Data Policies” at <https://www.mdpi.com/ethics>.

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