

Review

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Review

Cryogenic Technologies for Biomethane Production: A Critical Review of Processes, Performance and Prospects

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Abstract

Cryogenic upgrading represents a promising route for the production of high-purity biomethane, aligning with current decarbonization goals and the increasing demand for renewable gases. This review provides a critical assessment of cryogenic technologies applied to biogas purification, focusing on process fundamentals, technological configurations, energy and separation performance, and their industrial integration potential. The analysis covers standalone cryogenic systems as well as hybrid configurations combining cryogenic separation with membrane or chemical pre-treatment to enhance efficiency and reduce operating costs. A comparative evaluation of key performance indicators—including methane recovery, specific energy demand, product purity, and technology readiness level—is presented, along with a discussion of representative industrial applications. In addition, recent techno-economic studies are examined to contextualize cryogenic upgrading within the broader landscape of CO₂ separation technologies. Environmental trade-offs, investment thresholds, and sensitivity to gas prices and CO₂ taxation are also discussed. The review identifies existing technical and economic barriers, outlines research and innovation priorities, and highlights the relevance of process integration with natural gas networks. Overall, cryogenic upgrading is confirmed as a technically viable and environmentally competitive solution for biomethane production, particularly in contexts requiring liquefied biomethane or CO₂ recovery. Strategic deployment and regulatory support will be key to accelerating its industrial adoption.

Keywords: biogas upgrading; carbon dioxide separation; gas purification; gas refining; hybrid separation systems; renewable fuel; cryogenic separation.

1. Introduction

The global energy landscape is at a critical crossroads, driven by the urgent need to reconcile the ever-growing demand for energy with pressing environmental imperatives. Rising population and industrial expansion continue to fuel global energy consumption, with fossil fuels like crude oil and natural gas still dominating the mix. After the temporary downturn caused by the COVID-19 pandemic, global oil demand has continued its strong recovery. In 2024, it reached approximately 103.9 million barrels per day, and is projected to rise to 105 million b/d in 2025 [1]. Similarly, global natural gas demand grew by 2.7% in 2024, corresponding to an increase of around 115 billion cubic metres, and is expected to maintain a similar growth rate in 2025 [2]. Yet this reliance on fossil energy is increasingly unsustainable, given its major contribution to greenhouse gas emissions and climate change. This situation has catalyzed a fundamental transformation of the energy paradigm. Global attention has shifted toward clean, low-carbon energy sources, and within this context, biomethane has emerged as a renewable vector of particular promise. Derived from biogas—produced via anaerobic digestion of organic residues including sewage sludge, agricultural waste, and energy crops [3]—biomethane offers both environmental and energy security advantages. The anaerobic process not only diverts waste from landfills but simultaneously transforms it into usable energy,

aligning with circular economy principles. The substrates for anaerobic digestion are diverse and locally available, including livestock manure, agro-industrial residues, source-separated organic municipal solid waste and wastewater sludge. This flexibility supports decentralized energy models and allows tailoring the process to local biomass availability.

Biomethane's versatility further reinforces its role in decarbonization. It can serve diverse end-uses: industrial cogeneration, heating, domestic energy supply, and transportation fuel, effectively replacing natural gas [4]. Additionally, biomethane utilization directly contributes to the reduction of GHG emissions. The typical production pathway includes feedstock pretreatment, anaerobic digestion, and subsequent upgrading to produce grid-quality or liquefied biomethane, depending on the intended application. However, realizing this potential hinges on addressing the critical technical challenge of biogas purification. Biogas typically contains a mixture of methane (CH_4), carbon dioxide (CO_2), nitrogen, hydrogen sulfide, oxygen, and water vapor, which must be efficiently removed to obtain grid-quality biomethane [5]. Cryogenic upgrading technologies, based on extremely low-temperature processing, stand out for their ability to produce high-purity biomethane, often exceeding 99% CH_4 . [6]. These processes enable effective methane- CO_2 separation and often facilitate CO_2 recovery as a valuable byproduct [7].

Recent studies highlight the competitive advantages of cryogenic upgrading when integrated with liquefaction of natural gas (bio-LNG) routes. Compared to conventional methods such as membrane separation or chemical scrubbing, cryogenic systems can simultaneously deliver high methane purity and energy efficiency, while producing a liquefied product with increased energy density. For long-distance transport, bio-LNG (with ~ 21 MJ/L) is significantly more efficient than compressed biomethane [8]. Moreover, the potential for recovering liquid CO_2 as a co-product enhances the environmental and economic performance of these systems [9,10].

The growing interest in biogas upgrading is also reflected in the diversification of technologies adopted across Europe. While membrane separation currently dominates the market, cryogenic methods are gaining momentum due to their integration potential and purity levels [11]. In parallel, favorable policies—such as the EU's target to achieve 15% renewable fuels in transport by 2030—continue to push for biomethane adoption at scale [12]. In 2023, combined biogas and biomethane production in Europe reached 22 billion cubic metres (bcm), equivalent to approximately 234 TWh, representing about 7% of the EU's natural gas consumption. Biomethane production alone grew to 4.9 bcm, marking a 21% year-on-year increase, and installed capacity rose to 6.4 bcm/year by Q1 2024, the highest annual increase recorded to date [13].

Beyond technical merit, cryogenic upgrading technologies are increasingly evaluated from a sustainability and techno-economic perspective. Recent assessments have compared the energy and exergy efficiencies of different cryogenic routes, such as distillation, anti-sublimation, and controlled freeze zone (CFZ) systems, in both standalone and integrated bio-LNG configurations. Reported efficiencies for optimized configurations exceed 95%, with payback periods under 2.5 years in favorable economic scenarios [14–16].

Against this backdrop, the present article aims to provide a critical and comprehensive review of cryogenic upgrading as a cornerstone technology in the biomethane value chain. The underlying principles, key configurations, and recent technological advances in cryogenic separation are analyzed in detail. Special attention is paid to comparing cryogenic options with conventional upgrading methods in terms of energy consumption, methane purity, carbon capture potential, and integration feasibility with downstream processes such as liquefaction. Furthermore, the review identifies knowledge gaps related to the comparative performance of emerging cryogenic techniques (e.g., CFZ vs. anti-sublimation) and their scalability. It also examines the challenges associated with high energy demands in refrigeration cycles, the behavior of contaminants under cryogenic conditions, and prospects for optimizing process efficiency through heat integration or hybridization. In summary, this review positions cryogenic upgrading as a key enabler in the large-scale deployment of biomethane, contributing not only to energy diversification but also to climate

mitigation. It aims to support informed decision-making by policymakers, industry actors, and researchers seeking high-performance solutions in the transition toward a low-carbon energy system.

2. Principles and Configurations of Cryogenic Biogas Upgrading

Cryogenic upgrading of biogas is founded on the principle of differential condensation behavior of its constituents under extremely low temperatures. Unlike conventional separation techniques that rely on chemical affinity or physical differences such as molecular size or permeability, cryogenic methods exploit the distinct phase-change thresholds of each compound present in biogas when progressively cooled. The aim is to isolate methane—biogas's energy-rich component—by cooling the gas mixture to specific temperatures where impurities condense or solidify while methane remains in the vapor phase or is selectively liquefied [17]. The most common contaminants in biogas—such as carbon dioxide, water vapor, hydrogen sulfide, and siloxanes—exhibit varying condensation and solidification points. This variation enables a staged cooling strategy for their selective removal. For instance, water vapor tends to condense at relatively higher temperatures, allowing for early separation. Impurities like hydrogen sulfide and siloxanes can be extracted in subsequent steps as their respective liquefaction or freezing points are reached. Carbon dioxide, which is present in significant quantities (up to 50%), typically requires lower temperatures to undergo phase transition and is often the most technically challenging component to separate.

The efficiency of the process depends critically on maintaining precise temperature gradients across multiple stages of compression and refrigeration [18]. Advanced temperature regulation ensures that each target compound transitions phase at its specific thermal threshold, thereby maximizing methane purity while reducing energy demand [19]. In standard configurations, the process begins at a working pressure of about 10 bar. Sequential cooling steps are then applied: at approximately -25°C , most of the water vapor, hydrogen sulfide, and volatile siloxanes are removed; at around -55°C , carbon dioxide begins to condense, and final CO_2 purification is typically achieved at -85°C when solidification is complete [20]. The solid or liquid CO_2 recovered through this process may be valorized as a commercial by-product, enhancing the economic feasibility of the system [21]. Alternatively, some systems employ a configuration involving initial gas drying and multistage compression up to 80 bar, which allows operation at moderately higher temperatures (-45°C to -55°C). However, this approach requires intermediate cooling stages during compression, which increases process complexity and energy consumption [17].

Among the multiple cryogenic technologies under development or commercial use, two broad categories dominate. Systems based on cryogenic distillation facilitate separation through differential boiling points, allowing methane and carbon dioxide to be isolated as liquid phases under well-controlled thermal conditions [22]. Meanwhile, anti-sublimation technologies rely on the direct solidification of CO_2 onto cold surfaces, bypassing the liquid phase and offering potential benefits in compactness and reduced equipment footprint [23]. According to thermodynamic analyses, anti-sublimation configurations can operate efficiently under sub-atmospheric temperatures, offering CO_2 capture with high selectivity and minimal methane loss [10]. These systems are especially attractive for distributed biomethane applications where on-site CO_2 valorization is feasible, and transport of compressed gas is limited. Cryogenic packed beds have also been proposed as an alternative configuration. In these systems, the gas passes through a bed of cryogenically cooled material, where specific components are trapped via deposition [24]. Computational models have been developed to optimize key design parameters such as bed geometry, flow rate, and heat transfer conditions, aiming to maximize methane recovery while ensuring stable operation. Recent studies have shown that careful control of temperature gradients and packing structure is essential to avoid blockages caused by frost accumulation, and to improve the efficiency of contaminant capture [25]. Simulation tools also support the evaluation of system behavior during transient conditions, contributing to the development of robust and energy-efficient configurations [26].

3. Cryogenic Biogas Upgrading Technology

Cryogenic biogas upgrading capitalizes on the distinct phase-change characteristics of each gas constituent at cryogenic temperatures. Instead of relying on chemical reactions or membrane-based selectivity, this approach exploits the physical transitions—condensation or solidification—that occur when the biogas stream is cooled in a controlled sequence [27]. By progressively lowering the temperature under defined pressure conditions, impurities such as CO₂, H₂O, H₂S, and siloxanes can be effectively removed, resulting in a methane-enriched stream that meets biomethane quality standards.

A widely implemented operational regime maintains the system at approximately 10 bar, under which cooling stages are configured to selectively condense and separate unwanted compounds at their characteristic temperature thresholds [28]. Water vapor, hydrogen sulfide, and siloxanes are typically eliminated at around -25 °C, while carbon dioxide requires lower temperatures—liquefying near -55 °C and ultimately solidifying at -85 °C to achieve complete separation [29]. As an alternative, some configurations apply initial drying followed by multi-stage compression up to 80 bar. This allows operation at moderately higher cryogenic temperatures, between -45 °C and -55 °C, but requires intermediate cooling during compression steps, increasing system complexity and energy demands [30]. To improve process flexibility and economic performance, numerous modifications and design alternatives have been proposed in recent years, ranging from novel cycle architectures to hybrid and intensified systems [17,31,32]. These technological developments are examined in detail throughout the following subsections.

3.1. Cryogenic Distillation Process

Cryogenic distillation is a thermally-driven technology used to separate carbon dioxide from biogas by exploiting the specific phase-change behavior of CO₂ at low temperatures and elevated pressures. The process enables CO₂ to be recovered in liquid form while producing biomethane of high purity. Traditionally, this method is associated with high energy demands, which can negatively impact its economic competitiveness compared to other upgrading techniques [33]. To address these limitations, recent developments have focused on integrating process intensification techniques and hybrid systems that significantly improve energy performance. Reported results indicate that, when properly designed, hybrid cryogenic distillation schemes can achieve up to 70% energy savings compared to baseline configurations [34].

Although most cryogenic distillation systems have been validated under natural gas purification scenarios, several authors have outlined their potential applicability to biogas upgrading [35,36]. Figure 1 shows a representative setup in which the gas stream is pre-treated to remove trace contaminants and then cooled in two stages before entering a distillation column. Inside the column, separation is based on boiling point differences: methane, as the lightest compound, is collected at the top via a partial condenser, while CO₂ is recovered as a bottom product in liquid form. A fraction of the CO₂ stream is recycled to assist in the reboiler duty of the pre-vaporization unit, whereas the rest is extracted for commercial use, thereby enhancing the system's economic viability.

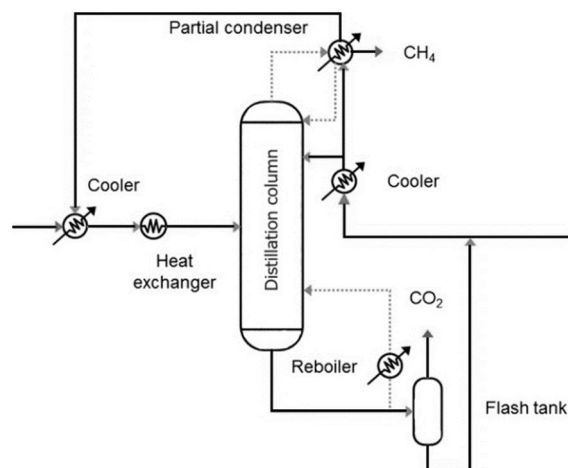


Figure 1. Schematic representation of a cryogenic distillation system for biogas upgrading. Pre-cooled biogas enters the distillation column, where high-purity methane is recovered overhead and liquid CO₂ is collected at the bottom. Part of the CO₂ is recycled to support reboiler vaporization.

In the work by Yousef et al. [37], cryogenic distillation was optimized to avoid CO₂ freezing within the column, a common issue that may disrupt process performance. They investigated the influence of feed pressure, CO₂ concentration, temperature, reflux ratio, and feed location within the column. Simulation results demonstrated that a CO₂ purity of 99.5 mol% could be attained at a specific energy consumption of 0.31 kWh/Nm³ of raw biogas, while methane purity reached up to 97 mol% without CO₂ solidification. Notably, the tendency for CO₂ freezing was found to remain constant across a wide range of inlet CO₂ concentrations, suggesting robustness in process operation. In a complementary study [38], a novel design combining a four-stage compression train, a distillation column, and an integrated flash separator was proposed to optimize energy recovery and separation efficiency. This configuration yielded a record CH₄ purity of 97.2 mol%, outperforming previous single-column systems. At the same time, a valuable CO₂ co-product of more than 99% purity was generated. The associated energy penalty was just 0.38 kWh/Nm³ of clean gas, positioning the system among the most efficient reported to date in terms of methane recovery, CO₂ valorization, and overall process economics.

Moreover, recent numerical studies have explored the feasibility of process intensification in cryogenic distillation by adopting column internals that improve mass and heat transfer performance. For instance, the use of structured packing and enhanced heat integration has been shown to reduce condenser and reboiler loads, especially when CO₂-rich feeds are treated at optimized pressure levels [39]. Other experiments have demonstrated that cascading refrigeration stages and adjusting the CO₂ reflux ratio can further suppress freezing risk while maintaining operational efficiency [40]. These innovations underscore the ongoing evolution of cryogenic distillation from an energy-intensive method to a viable and scalable solution for biogas upgrading.

3.2. Dual-Pressure Cryogenic Distillation Process

The process of removing carbon dioxide from gas through a dual-pressure low-temperature (or cryogenic) distillation process represents a significant advancement in gas purification, especially in fields containing acid gas, for example, natural gas that is typically laden with acidic compounds like CO₂ and H₂S. These fields, accounting for about 40% of the globe's natural gas reserves, span continents from Europe and America to Asia. Yet, to harness these resources effectively, eliminating acidic components is paramount [36]. The process proposed by Pellegrini et al. [41] tackles these hurdles with an innovative approach, employing two distillation stages at varying pressures alongside an intermediary heating phase (Figure 2). Initially, at high pressure, the feed mixture

undergoes distillation, facilitating CO_2 separation in liquid form. Subsequently, at lower pressure, further methane purification is achieved.

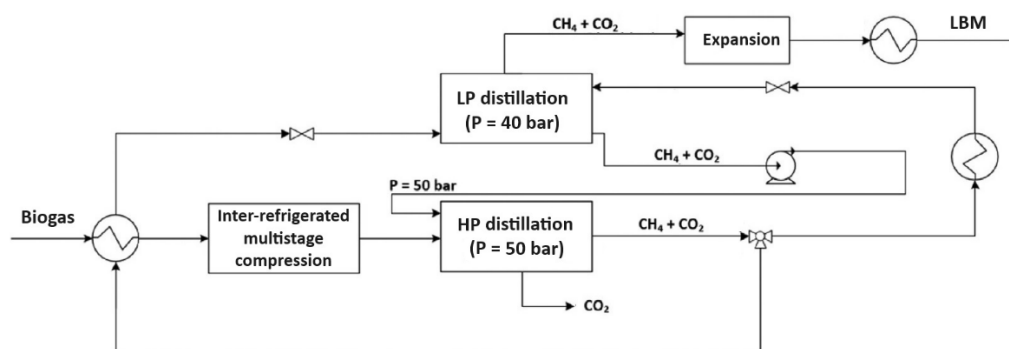


Figure 2. Schematic representation of a dual-pressure cryogenic distillation process for biogas upgrading. The system integrates high- and low-pressure distillation columns with intermediate heat exchange and expansion to produce liquid biomethane (LBM) and recover high-purity CO_2 .

More specifically, purification occurs within two distillation units: the first operates at high pressure (50 bar), while the second operates below methane's critical pressure (40 bar). The high-pressure section functions akin to the stripping section of a conventional distillation column, where liquid reflux is facilitated by recycling the flow from the low-pressure section's bottom. Similarly, the low-pressure section acts as the enrichment section: it features a partial condenser at the top, and the feed gas flow originates from the high-pressure section's output. The gas stream exiting the top of the low-pressure section meets the required methane purity specifications. Liquid biomethane (LBM) is subsequently generated via an appropriate liquefaction process. The high-pressure distillation section's bottom product comprises highly pure carbon dioxide. The incoming biogas feed is pre-cooled in a primary heat exchanger utilizing the available cooling capacity from an intermediate process stream, necessitating preheating before entering the low-pressure section. After pre-cooling, the biogas is compressed to 50 bar and further cooled to its dew point at that pressure before entering the high-pressure distillation section. Compression follows pre-cooling to reduce the compression power by lowering the feed stream's temperature.

The high-pressure section effectively eliminates input CO_2 : the bottom stream is high-pressure liquid CO_2 , while the top product stream is methane-rich (approximately 6.5 mol% CO_2). [41]. Since the high-pressure section operates above methane's critical pressure, obtaining pure methane through distillation in a single 50-bar unit is infeasible. Hence, final purification occurs in the low-pressure section at 40 bar. Outputs from the low-pressure section include an upper methane gas flow and a lower methane-rich liquid flow pumped back to the high-pressure section. The feed flow enters the high-pressure section on the fourth tray from the top, while liquid reflux from the low-pressure section's bottom is pumped and introduced into the first tray from the top. The upper gas flow from the high-pressure section undergoes splitting into two streams. Before entering the low-pressure section's bottom, a portion of the high-pressure section's upper product flow is heated and expanded to the low-pressure section's operating pressure, ensuring a temperature higher than its dew point. This prevents solid phase formation during expansion, with heat sourced from the input raw biogas stream pre-cooled prior to compression. The remaining upper product flow from the high-pressure section is cooled to 50 bar (below CO_2 's solubility limit) and expanded to the low-pressure section's operating pressure, yielding a liquid flow at its bubble point, fed into the low-pressure section one theoretical plate above the gas feed flow [42].

A pivotal advantage of this method lies in its adeptness at handling sour gas, even with substantial CO_2 and other impurities like higher hydrocarbons and hydrogen sulfide. This adaptability stems from meticulously tailored pressure and temperature conditions at each distillation phase, preventing solid CO_2 formation and ensuring efficient acid component separation.

Moreover, the process exhibits versatility, proving equally effective for both conventional natural gas and biogas derived from organic waste. This adaptability renders it a multipurpose solution for gas purification across diverse industrial and environmental settings [41]. Recent literature supports the industrial relevance and scalability of this approach. According to Pellegrini et al. [19], dual-pressure cryogenic distillation demonstrates strong operational efficiency: it utilizes only 14 % of the produced biomethane as internal energy—significantly lower than anti-sublimation (~22 %) and MEA (monoethanolamine) scrubbing (~29 %) schemes. Furthermore, most of this energy (61 %) is used for cooling, with no requirement for high-temperature reboiler heat. In terms of process stages, about 38 % of energy is consumed in compression, ~32 % in upgrading, and ~30 % in liquefaction, while CO₂ pressurization is effectively zero since CO₂ is handled as a pressurized liquid.

3.3. Anti-Sublimation Process

The production of liquefied biomethane via the anti-sublimation process involves the utilization of heat exchange surfaces to refine biogas, operating within the solid-vapor equilibrium region at atmospheric pressure [43]. In this process, CO₂ freezes directly within the gas stream, resulting in methane enrichment. This innovative approach, patented by Clodic et al. [44], is based on a specially designed cold chamber where CO₂ solidifies upon contact with cryogenic refrigerants. Its modular configuration allows for the decoupling of vaporization and solidification stages in separate chambers, thus enhancing operational flexibility. The process configuration is illustrated in Figure 3.

In the anti-sublimation process, purification is achieved through dry ice formation within a sealed cryogenic system. Two dynamic heat exchangers (HE in Figure 3) alternate between purification and regeneration phases, enabling continuous operation. Figure 3 depicts solid lines for the operational phase and dashed lines for the regeneration cycle. The incoming raw biogas, typically at ambient temperature and pressure, is first routed through a pre-cooling exchanger that recovers cold from the liquid CO₂ stream in the recovery loop. This pre-cooled gas then passes through the recovery exchanger, where residual dry ice from the previous cycle contributes additional cooling. By melting this layer of CO₂, temperatures as low as -51 °C are attained, and the resulting CO₂ is collected in liquid form. The liquid CO₂ is then pressurized to 50 bar and reheated in the recovery exchanger. To reach the anti-sublimation threshold, further cooling is required to bring the gas temperature down to approximately -160 °C, enabling the direct solidification of CO₂ from the gas phase. This is achieved with an external refrigeration system where nitrogen acts as the final cooling agent [16]. Inside the anti-sublimation exchanger, a stable layer of solid CO₂ accumulates, reducing CO₂ concentration in the methane-rich stream to below 50 ppm at atmospheric pressure. The purified gas is subsequently liquefied in a dedicated cryogenic condenser for downstream storage and use.

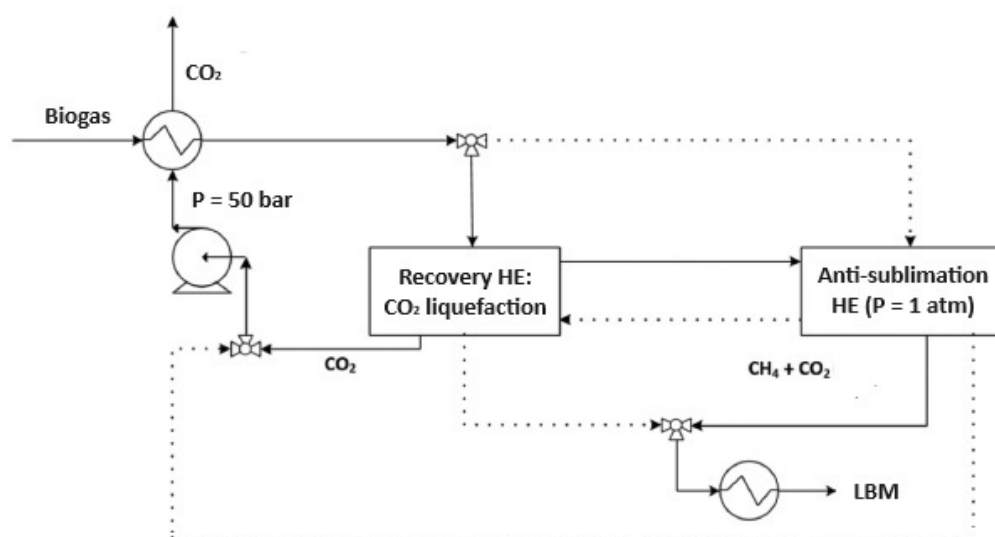


Figure 3. Schematic representation of biogas purification via the anti-sublimation process. The system integrates a CO₂ liquefaction stage (Recovery Heat Exchanger, HE) and an anti-sublimation unit operated at atmospheric pressure, enabling CH₄ enrichment and Liquid Biomethane (LBM) production through sequential cooling and solid CO₂ separation.

The anti-sublimation process requires precise control of pressure and temperature, guided by the CH₄-CO₂ phase diagram, to ensure efficient CO₂ solidification. Simulation studies conducted in Aspen Plus indicate that by reducing feed gas temperature to the CO₂ sublimation point, extraction purities up to 90% can be achieved [45]. Energy consumption ranges from 0.286 to 1.3 kWh/kgCO₂ removed, depending on operational conditions and the degree of thermal integration. Under optimal configurations, the lowest consumption case also corresponds to a specific cost of approximately 0.1 €/kg of CO₂ [16,45].

Additional insights from Naquash et al. [10] further support the practical implementation of this process. In comparative analyses, anti-sublimation exhibited the highest refrigeration demand among cryogenic systems, with over 99% of total energy consumption allocated to cooling duties, largely due to the necessity of reaching extreme cryogenic temperatures. Despite this, it enables CH₄ recoveries of over 98% and achieves CO₂ concentrations well below 100 ppm, making it suitable for high-purity biomethane production. However, the lack of process flexibility under variable flow rates and the need for robust frost management strategies were identified as key limitations. The design's integrated heat exchange between incoming gas and recovered CO₂ stream plays a pivotal role in energy efficiency, particularly in reducing the refrigeration load of the final purification stage. By leveraging the cold potential of sublimated or liquefied CO₂, the system reduces external cooling demands during startup and regeneration cycles. This integrated thermodynamic behavior illustrates the technological ingenuity of anti-sublimation and underscores its relevance for both bio-LNG production and high-value CO₂ recovery.

3.4. Controlled Freeze Zone Process

Controlled Freeze Zone, or CFZ technology, is a cryogenic one-step distillation process developed by Exxon Mobil over four decades ago to efficiently separate CO₂ and other acid gases from natural gas streams [46]. Originally demonstrated in a pilot unit with a processing capacity of 0.02 MNm³/day and CO₂ concentrations ranging from 15% to 65%, this technology achieved outlet gas CO₂ levels as low as 100 ppm. Following initial success, it was scaled up to accommodate natural gas flows up to 30 MNm³/day while maintaining output specifications. The key innovation of CFZ lies in its integration of vapor-liquid-solid equilibrium within a single distillation column. The column is structured into three main zones: an upper rectifying section (packed bed), a central CFZ chamber, and a lower stripping section. The central freezing zone contains a spray subzone for distributing liquid CO₂, a freezing zone where solid CO₂ forms, and a melting zone where the solids are recovered as liquid. Pre-treated and dehydrated natural gas enters the upper section. As methane rises and vaporizes, a CO₂-enriched liquid is sprayed into the CFZ chamber, where it solidifies under cryogenic conditions. The solid CO₂ accumulates, melts, and is withdrawn as a high-purity liquid at the bottom, while purified CH₄ exits at the top. The process is illustrated in Figure 4, which depicts the structural layout of the integrated CFZ column.

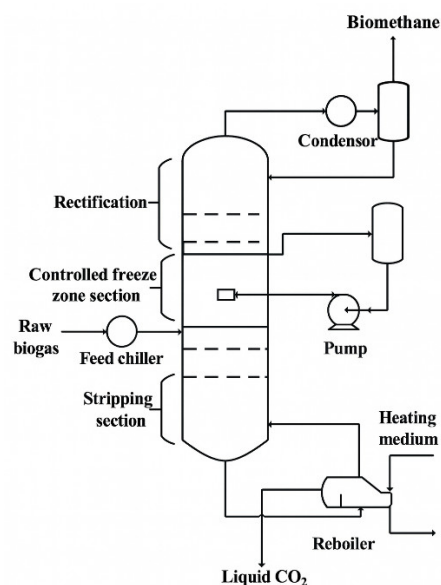


Figure 4. Purification of biogas through controlled freeze zone technology. The CFZ system integrates rectifying and stripping packed sections with a central chamber where CO_2 is frozen and melted in sequence. Liquid CO_2 is sprayed into the freezing zone, forming dry ice that is later melted and removed as a pure liquid stream, while high-purity methane exits from the top of the column.

According to Exxon Mobil, this configuration reduces capital costs by 10–30% compared to conventional absorption systems [47], with added operational benefits such as simplified process flow and reduced refrigerant demand. However, its performance diminishes in gas mixtures with high ethane or heavier hydrocarbons, which tend to co-solidify and hinder selectivity, thereby increasing separation complexity and cost. A comparative techno-economic and performance analysis between CFZ and dual-pressure cryogenic distillation (Pellegrini process) was conducted on the Natuna and Bojonegoro gas fields in Indonesia [48]. At Natuna, CFZ achieved a CH_4 purity of 95.62% and CO_2 recovery of 99.04%, outperforming Pellegrini's scheme in CO_2 separation but slightly lower in methane purity (91.98% CH_4 , 97.86% CO_2 recovery). At Bojonegoro, however, Pellegrini's approach surpassed CFZ in both parameters, reaching 99.91% CH_4 purity and 99.99% CO_2 recovery, compared to CFZ's 97.99% and 98.45%, respectively. Despite these mixed results in purity, CFZ demonstrated a lower CO_2 capture cost, at 7.9 USD/tonne, compared to 10.26 USD/tonne for Pellegrini distillation. This cost advantage is attributed to the integration of freezing and melting within a single column, reduced reboiler duty, and minimized refrigerant inventory. According to additional insights from Spitoni et al. [31], CFZ technology exhibits superior thermodynamic integration and a compact footprint, which enhances modularity and scalability—particularly in offshore or decentralized applications. Moreover, the direct collection of liquid CO_2 facilitates its downstream valorization for industrial use.

CFZ remains a promising option for high-volume natural gas upgrading, especially where cryogenic utilities are already present or CO_2 recovery for reuse is a priority. Nevertheless, its implementation in biogas treatment remains limited, primarily due to the lower scale and compositional variability of typical biogas streams. However, its technical underpinnings continue to inform the development of hybrid cryogenic–adsorption systems being considered in recent research for biomethane production [49].

3.5. Ryan-Holmes Process

The Ryan-Holmes process [50] represents a cryogenic biogas upgrading route based on extractive distillation, which facilitates carbon dioxide removal by modifying the phase behavior of

the CH₄–CO₂ mixture. This is achieved by incorporating an entrainer that elevates the critical locus and displaces the CO₂ freezing boundary toward lower pressures and temperatures, thus minimizing the risk of CO₂ solidification [19]. The process typically employs heavier hydrocarbons, such as n-butane, as entraining agents. These components interact with CO₂ to depress its freezing point and promote efficient separation from methane. The overall system comprises five principal units: (i) biogas compression, (ii) extractive distillation, (iii) entrainer recovery and regeneration, (iv) biomethane liquefaction, and (v) CO₂ condensation and pumping.

Initially, the raw biogas feed is compressed to approximately 40 bar to prepare it for extractive distillation. Within the distillation column, n-butane is injected as a solvent. Its presence shifts the critical region of the gas mixture, enabling lower-temperature operation without encountering the risk of CO₂ freeze-out. The biogas feed enters the column at an upper tray, optimizing the contact with the descending entrainer and maximizing separation efficiency. The overhead product from this column is primarily composed of methane and undergoes further cooling and depressurization via a liquid expander and condensation stage, ultimately yielding liquefied biomethane at near-atmospheric pressure. Conversely, the bottom stream from the extractive distillation unit—rich in CO₂ and n-butane—is first depressurized to 30 bar and then introduced into a regeneration column. In the regeneration section, thermal input enables the selective desorption and recovery of CO₂ in liquid form. The n-butane, now separated, is condensed, cooled, and recycled back to the extractive distillation stage, thereby closing the entrainer loop and minimizing solvent losses. The extractive distillation column itself operates under a temperature gradient that decreases from bottom to top—heated at the reboiler and cooled at the condenser—allowing stable phase separation throughout the column height. In contrast, the regeneration column maintains relatively higher temperature levels to ensure effective desorption and recovery of the entrainer. The Ryan-Holmes process is distinguished by its relatively high energy consumption, largely attributable to the reboiler duty required for entrainer regeneration and the condenser duty associated with methane product recovery. Nevertheless, its ability to circumvent CO₂ solidification without operating at extreme pressures or employing anti-sublimation steps makes it a promising alternative for biogas upgrading.

Originally developed for sour gas treatment in enhanced oil recovery operations, the Ryan-Holmes process was specifically designed to address two critical technical challenges: the formation of a CO₂–C₂H₆ azeotrope and the high freezing point of CO₂, both of which compromise the performance of conventional cryogenic distillation. By incorporating an extractive recycle loop, this process suppresses CO₂ solid formation and breaks the azeotropic barrier, enabling efficient separation of methane, CO₂, and light hydrocarbons [51].

Numerous large-scale natural gas processing facilities have implemented this technology successfully, including the Seminole Unit operated by Amerada Hess, the Willard Unit by ARCO, the GMK South Field by Mobil Oil Corporation, and the Wasson Denver Unit by Shell Oil. These applications confirm the maturity and industrial viability of the process under high CO₂ concentration conditions. In comparison with amine scrubbing systems, the Ryan-Holmes process presents multiple advantages: lower dehydration requirements, elimination of corrosive solvents, the use of natural gas liquids as solvents, and the production of high-pressure CO₂ streams that are readily suitable for geological reinjection. Additionally, the process achieves high recovery rates of heavier hydrocarbons—such as ethane and propane—providing an important revenue stream, particularly in scenarios where methane content in the feed gas is reduced [51,52].

Beyond its traditional configuration, the Ryan-Holmes process has also been explored in hybrid schemes combined with membrane technologies to improve cost-effectiveness and operational flexibility. Membrane units may be installed upstream for bulk CO₂ removal prior to cryogenic distillation, reducing the load on the distillation system and optimizing overall energy use. Alternatively, membranes can be integrated downstream of the CO₂ stripper, particularly in the overhead stream rich in CO₂ and ethane, to further purify the CO₂ product and break azeotropic mixtures. This hybrid approach has been demonstrated to lower utility costs by up to 20% and capital expenditure by approximately 25% compared to stand-alone cryogenic systems [51,53].

3.6. Hybrid Cryogenic Processes

Building upon the foundational concepts discussed in the Ryan-Holmes process and its membrane-integrated variants, hybrid cryogenic systems have emerged as a promising direction to overcome some of the limitations associated with conventional cryogenic separation. These integrated configurations aim to improve energy efficiency, process flexibility, and cost-effectiveness by combining cryogenic methods with other separation technologies, such as membranes, adsorption, or absorption, to create synergistic effects. Cryogenic technologies alone may face several barriers to effective application, including high energy consumption, production costs, and the availability of cooling energy sources [54]. In this context, hybrid cryogenic processes, such as those integrating cryogenic adsorption with membrane separation, have been proposed to mitigate these drawbacks and enable more sustainable biogas upgrading or CO₂ capture routes [55]. For instance, an experimental study assessed the energy-saving opportunities of a cryogenic-membrane hybrid process designed for biogas refining applications [56]. In this system, membrane modules were evaluated under a range of operating temperatures, revealing excellent energy performance—0.8 MJ per kilogram of CH₄—and high product purity, reaching 98% for methane and 99.8% for CO₂, when operated at temperatures between -30 and 40 °C. In another work based on simulation, a hybrid system combining vacuum cryogenic adsorption with subsequent CO₂ liquefaction was examined [57]. The results showed that a total energy input of 1.4 GJ per tonne of CO₂ was required to achieve a product stream with 98.2% CO₂ purity, indicating the potential of such configurations under moderate thermal demand. From an economic perspective, a techno-economic assessment evaluated the cost implications of a hybrid membrane-cryogenic process for CO₂ capture, reporting that the avoided CO₂ cost associated with an 85% capture ratio was approximately 9% higher than that of a traditional amine absorption process [58]. Although this figure reflects a relative disadvantage in terms of direct cost, the hybrid system demonstrated favorable performance when factors such as solvent regeneration and environmental compatibility were considered. A comparative study further analyzed cryogenic hybrid processes for biogas upgrading, showing that while conventional membrane-integrated systems achieved lower overall costs, the cryogenic hybrid alternatives offered notable advantages in terms of plant-level benefits and system robustness under variable feed conditions [59]. These findings underline the potential of hybridization as a design principle in advanced CO₂ separation systems.

Beyond membrane integration, cryogenic hydrate-based processes also present a viable option for high-purity CO₂ recovery. These systems rely on hydrate formation at low temperatures and high pressures, following a preliminary cryogenic separation stage that reduces the CO₂ content of the feed stream [60]. The sequential operation allows efficient impurity removal and CO₂ capture through solid hydrate formation. Similarly, cryogenic absorption processes using ammonia (NH₃) have gained attention as an alternative for biogas upgrading. NH₃-based absorption offers several advantages, such as low regeneration energy requirements, compatibility with low-temperature operation, and the commercial availability of NH₃ as a working fluid [61]. However, the practical implementation of these emerging hybrid processes still requires further optimization to enhance selectivity, scalability, and cost competitiveness under real operating conditions.

4. Energy Performance of Cryogenic Biomethane Upgrading Technologies

An integral aspect of cryogenic upgrading for biomethane production involves conducting a thorough energy analysis [62]. This analysis is essential for assessing the efficiency and viability of the process, focusing on various facets of energy utilization. Central to the energy analysis is the examination of energy consumption across the biomethane production process. This encompasses evaluating the energy inputs required for cryogenic cooling, compression, and liquefaction, as well as any ancillary processes involved in pretreatment and post-treatment stages [63]. Understanding energy consumption profiles facilitates the identification of optimization opportunities to minimize energy wastage and enhance overall process efficiency. Moreover, the energy analysis entails

exploring energy efficiency metrics to assess the effectiveness of energy utilization within the cryogenic upgrading system [64]. This involves evaluating the ratio of useful output energy (biomethane production) to input energy (electricity, refrigeration, etc.). Quantifying energy efficiency metrics enables the identification of areas for improvement and the implementation of strategies to enhance overall system performance. Additionally, energy recovery mechanisms play a crucial role in the energy analysis of cryogenic biomethane upgrading. Strategies such as heat integration, waste heat recovery, and the utilization of renewable energy sources can significantly reduce the net energy demand of the process [65]. By maximizing energy utilization and harnessing energy from various sources, the overall sustainability and economic viability of cryogenic biomethane production can be enhanced.

A comparative analysis of energy consumption among three prominent cryogenic biogas upgrading technologies, dual-pressure low-temperature distillation, Ryan-Holmes process and antisublimation process [17] revealed significant insights. In the dual-pressure low-temperature distillation process, compression energy accounts for 38% of the total energy consumed, primarily due to the necessity of high pressures (40–50 bar). Additionally, approximately 62% of the total energy is attributed to the cryogenic duty essential for low-temperature distillation, which is crucial for producing liquefied biomethane. Contrastingly, the Ryan-Holmes process, featuring extractive distillation with a regenerative step, allocates 32% of the total energy to upgrading and 12% to heating load. In the antisublimation process, the upgrading step emerges as the most energy-intensive, consuming about 74% of the total energy. This intensity arises from the solid-to-vapor phase change involved in removing CO₂ from CH₄ during upgrading, consuming nearly all the energy for cooling duty. Moreover, a comparison of upgrading technologies concerning biomethane purity versus energy consumption highlights the challenges faced by cryogenic distillation in achieving high biomethane purity due to CO₂ freezing, particularly at low temperatures within the distillation column. Dual-pressure low-temperature distillation configurations demonstrate enhanced purity levels, reaching up to 99%. Conversely, single-column distillation processes offer slightly lower purity but are preferable when CO₂ purity is not a constraint, presenting an energy-efficient option. The antisublimation process, which leverages CO₂ freezing, has been evaluated against conventional methods, showing promising results in terms of biomethane purity, albeit with higher specific energy consumption due to the energy-intensive solidification of CO₂ and compression requirements. These findings underscore ongoing efforts to optimize energy utilization in cryogenic biogas upgrading processes.

Additional insights from the comprehensive assessment by Naquash et al. [10] reinforce the differences in energy profiles and reveal new benchmarks for evaluating process performance. Table 1 summarizes the comparative results reported in this study. The dynamic simulations varied out by these authors show that dual-pressure cryogenic distillation can achieve specific energy consumption values as low as 0.30 kWh/kgCH₄, with refrigeration and compression duties accounting for 61% and 39% of total energy use, respectively. Notably, the elimination of reboiler requirements through full cryogenic operation contributes to lowering overall utility demand. The Ryan-Holmes process, although advantageous for some gas compositions, generally presents higher energy intensities, ranging between 0.44 and 0.55 kWh/kgCH₄, primarily due to the energy needs for both the refrigerant loop and chemical solvent regeneration. In contrast, the antisublimation system displays the highest energy demand, exceeding 0.75 kWh/kgCH₄ in most configurations, with more than 99% of energy allocated to cooling to reach solid–gas equilibrium temperatures. Despite this, its superior separation selectivity allows CH₄ recoveries above 98% and CO₂ concentrations below 100 ppm. Naquash et al. [10] also identify that operating pressure strongly influences refrigeration needs: increasing pressure to 50 bar can cut condenser duty by up to 50%, though it may raise compression or even melting zone loads if thermal recovery is not implemented. Importantly, they demonstrate that heat integration strategies—particularly pinch-based exchanger network design combined with internal Joule–Thomson expansion of CO₂-rich side streams—can lower net cooling energy demand by up to 47%

in dual-pressure schemes. Such internal cold recovery is not always feasible in antisublimation setups, where temperature control and frost management limit integration potential.

These numerical insights stress that while anti-sublimation excels in separation purity, it does so at the cost of higher energy intensity. Dual-pressure distillation offers a better balance between purity and energy efficiency, and Ryan-Holmes provides operational flexibility but with moderate energy penalties. The choice of configuration ultimately depends on the specific boundary conditions of each case—such as feed gas composition, biomethane purity requirements, availability of cryogenic utilities, and integration with liquefaction chains or CO₂ valorization routes. In any scenario, improving system-wide energy efficiency remains the most consistent strategy to enhance the techno-economic feasibility of cryogenic upgrading.

Table 1. Comparative energy indicators for cryogenic biogas upgrading technologies [10,17].

Technology	CH ₄ Purity (%)	Specific Energy Consumption (kWh/kgCH ₄)	Production Cost (€/kg CH ₄)	Main Energy Loads (% of total)	Notable Limitations	Features /
Dual-pressure cryogenic distillation	up to 99.0	0.30–0.34	0.36–0.44	Refrigeration (61%), Compression (39%)	No reboiler; energy/purity enables cold integration	good balance;
Ryan-Holmes process	97.0–98.5	0.44–0.55	≈0.45	Refrigeration (~32%), Heating (~12%)	High flexibility; solvent regeneration adds energy burden	
Anti-sublimation	>98.0	>0.75	>0.50	Refrigeration (>99%)	Highest purity; energy demand due to CO ₂ freezing	high

5. Techno-Economic Assessment of Cryogenic Biomethane Upgrading

Economic analysis also plays a pivotal role in determining the feasibility and economic viability of implementing cryogenic upgrading processes for biomethane production. This section focuses on evaluating various economic factors crucial for assessing the profitability and competitiveness of cryogenic biomethane production technologies. The economic analysis encompasses a thorough examination of capital investment requirements, operational expenditures, and potential revenue streams associated with cryogenic biomethane production facilities [66]. Initial capital investment considerations include equipment procurement, infrastructure development, and installation costs essential for establishing cryogenic upgrading plants. Operational costs constitute a significant aspect of the economic analysis, encompassing ongoing expenses such as energy consumption, maintenance, and labor costs associated with the operation of cryogenic biomethane production facilities. Understanding these operational expenses is essential for estimating the overall cost of biomethane production and determining its economic viability. In addition to costs, revenue generation opportunities are evaluated, primarily through the sale of biomethane and potential by-products. Revenue streams may include income from biomethane sales, carbon credit trading, or other renewable energy incentives available in the market [67]. Assessing market demand, pricing dynamics, and regulatory frameworks is crucial for accurately forecasting revenue and assessing the economic feasibility of cryogenic biomethane production projects.

Cost-benefit analysis and financial modeling techniques are employed to evaluate the financial performance metrics of cryogenic biomethane production projects. Parameters such as net present

value (NPV), return on investment (ROI), and internal rate of return (IRR) are analyzed to gauge the project's economic attractiveness and profitability over its lifecycle [68]. Furthermore, sensitivity analysis is conducted to assess the impact of key variables, such as feedstock costs, energy prices, and market conditions, on the financial performance of cryogenic biomethane production projects. Identifying sensitivities and potential risks enables stakeholders to develop mitigation strategies and make informed decisions regarding investment in cryogenic biomethane production technologies [69].

A recent study on the economic evaluation of cryogenic technologies [17] yields significant insights into investment and production costs. Firstly, comparing biogas upgrading through single-column and dual-column cryogenic distillation processes at different capacities reveals that higher capacity leads to decreased investment costs while maintaining constant feed composition and operating conditions. Additionally, transitioning from a single-column to a dual-column distillation process results in a slight increase in investment costs, with percentages varying depending on plant capacities, at 1.30% for 1400 m³/h plants and 2.94% for 600 m³/h plants. Furthermore, the cost analysis between cryogenic distillation and anti-sublimation technologies demonstrates that cryogenic distillation incurs higher production costs due to the need for elevated pressures for biogas, consequently increasing the cost of LBM production. In contrast, the desublimation process operates below the CO₂ triple point, avoiding the need for high pressures, resulting in lower production costs (0.2 €/kg) compared to cryogenic distillation (0.45 €/kg). According to the authors, the economic trade-offs associated with cryogenic technologies for biogas upgrading highlight the importance of considering both investment and production costs in technology selection and process optimization.

Additional insights [14] complement this economic outlook by providing comparative cost indicators under standardized performance assumptions. Their techno-economic simulations indicate that dual-pressure cryogenic distillation (DPCD) has a lower methane loss and specific energy consumption per kg of CH₄ recovered than both single-column cryogenic distillation and anti-sublimation systems, which directly affects production cost. In terms of capital investment, DPCD configurations exhibit marginally higher equipment costs due to the inclusion of a second distillation column and advanced exchanger networks, but this is offset by improved energy efficiency and lower operational costs. In contrast, anti-sublimation systems, while simpler in design, show higher total cost of ownership due to the large cooling demands and the need for precision frost management infrastructure. Naquash et al. [10] report specific production costs for cryogenic systems ranging between 0.36–0.44 €/kg of CH₄ depending on feed composition and heat integration level, while anti-sublimation systems, despite achieving high product purity, often surpass 0.50 €/kg CH₄ due to elevated refrigeration requirements and limited thermal recovery. Their analysis further emphasizes that economies of scale and integration with bio-LNG value chains are decisive for improving cost-effectiveness, especially in decentralized or modular applications.

6. Overcoming Bottlenecks and Emerging Pathways for Cryogenic Biomethane Production

Cryogenic technologies for biomethane upgrading are emerging as high-purity, low-emission alternatives to conventional methods. However, their broader implementation is still constrained by a complex mix of technical, economic, and systemic challenges. Among the most critical is the significant energy demand associated with cooling, phase change, and separation at very low temperatures. Maintaining cryogenic conditions requires robust refrigeration systems and multistage compression, which together can represent up to 90% of the total energy input in some configurations. Reducing these demands through energy integration, internal cold recovery, and advanced thermal management is essential for enhancing sustainability and lifecycle efficiency [70].

Equally important are the capital and operational expenditures (CAPEX/OPEX), which are often higher than those of chemical or membrane-based systems. Cryogenic upgrading requires specialized components—such as cryogenic heat exchangers, distillation columns, and vacuum-insulated pipelines—that significantly increase equipment and installation costs. Operational

expenses, particularly electricity for refrigeration and compression, represent a recurring burden. Strategies such as heat pump coupling, use of renewable electricity, and optimization of pressure and temperature levels are being explored to reduce the overall cost per unit of biomethane produced [71].

Scaling cryogenic systems poses another hurdle. While pilot-scale and small demonstration plants (typically <1000 m³/h) have shown technical feasibility, there is a lack of standardized engineering solutions for medium and large-scale units that can operate continuously and reliably under fluctuating feedstock and ambient conditions. Modular design principles and dynamic simulation tools are being developed to guide the scale-up process, ensuring process stability, control precision, and rapid start-up and shut-down capabilities under load variation [72]. The variability in biogas composition, both in terms of CH₄/CO₂ ratio and presence of trace components (e.g., NH₃, H₂S, siloxanes), further complicates process control. These variations affect condensation temperatures, freezing risks in distillation columns, and the selectivity of sublimation mechanisms. Advanced process analytical technologies (PAT) and real-time gas monitoring tools are being proposed to support adaptive process control, enabling cryogenic systems to self-regulate according to the fluctuating properties of the biogas stream [73].

In terms of future trends, innovation in core cryogenic components is key. Notable developments include magnetocaloric refrigeration systems, mixed refrigerant cycles tailored to biogas upgrading, and compact multi-stream plate-fin heat exchangers that enhance thermal performance while minimizing pressure drop. In parallel, digital twins and AI-driven process optimization are being introduced to monitor performance, detect anomalies, and optimize energy and product recovery across multi-criteria targets [74]. The integration of cryogenic upgrading with renewable energy systems (solar PV, wind, hydro) is gaining traction, particularly in remote or off-grid installations. When co-located with renewable electricity generation, these systems can operate flexibly during surplus power periods, turning electricity into storable, high-purity methane. This concept aligns with the EU's "smart sector integration" strategy and supports grid balancing through Power-to-Gas-to-Power loops [75].

Moreover, policy support and financial instruments play a crucial role in fostering adoption. Feed-in tariffs, renewable gas guarantees of origin, carbon pricing mechanisms, and inclusion of liquefied biomethane in national energy mixes all contribute to improving project bankability. The current revision of the Renewable Energy Directive III (RED III) directive and the introduction of CO₂ performance standards for heavy-duty vehicles are expected to stimulate demand for high-purity biomethane produced via cryogenic pathways. From a circular economy perspective, cryogenic upgrading enables valorization of industrial off-gases (e.g., from pulp and paper, breweries, wastewater treatment) by integrating biogas upgrading with CO₂ purification and liquefaction for reuse. The high pressure and purity of recovered CO₂ from cryogenic systems make it suitable for food-grade, chemical synthesis, and carbon capture and utilization (CCU) applications. Notably, cryogenic separation technologies offer several distinct advantages that make them highly attractive for advanced biogas upgrading applications. They enable the production of methane (CH₄) with purity levels exceeding 99%, suitable for direct injection into natural gas distribution grids or for conversion into bio-LNG, meeting stringent quality standards for both uses. Furthermore, these systems facilitate the co-production of high-quality carbon dioxide (CO₂) at pipeline pressure, which reduces or eliminates the need for energy-intensive post-treatment and compression stages. An additional benefit lies in the minimal use of chemical absorbents or solvents, which not only decreases operational complexity and cost but also avoids the generation of hazardous secondary waste streams, aligning with environmental regulations. Moreover, cryogenic upgrading is naturally compatible with bio-LNG value chains, enabling direct liquefaction of the upgraded methane and thus enhancing energy efficiency and system integration. These features position cryogenic technologies as a strategic component within circular economy frameworks and decarbonized energy systems, as emphasized in recent market analyses forecasting their growing role in renewable gas infrastructure [76].

Nevertheless, some systemic challenges remain unresolved. For instance, anti-sublimation technologies, while highly selective, demand precise frost management and pose scale-up difficulties due to the formation and accumulation of solid CO₂ [43,77]. Likewise, cryogenic distillation systems must carefully avoid CO₂ freeze-out in condenser zones, which may require reboiler adjustments or external heat pulses—complicating thermal balance and increasing CAPEX [28,78]. Furthermore, controlled freeze zone and Ryan-Holmes processes, although promising, are not yet fully commercialized for biogas. Their integration into modular, decentralized plants requires tailored designs, robust automation, and hybridization with upstream pre-treatment and downstream liquefaction units. These systems can benefit from ongoing innovations in cryogenic insulation materials, vacuum super-jacketing, and magnetic valve control for ultra-low temperature operations.

To overcome these technical constraints, recent studies recommend integrating cryogenic upgrading directly with biomethane liquefaction, creating compact plants capable of producing bio-LNG for off-grid transport or marine use. This co-location allows cold recovery, improving energy efficiency by 15–25% depending on feed composition and operating pressure. Complementary integration with LNG regasification terminals is also being explored, allowing cascaded cold energy reuse and minimizing thermal losses during conversion stages. Advanced modeling tools—such as multi-objective genetic algorithms, pinch analysis, and exergy-based cost allocation—are being used to fine-tune process parameters and identify optimal trade-offs between energy efficiency, CAPEX, CH₄ recovery, and CO₂ quality. These models are particularly relevant when designing hybrid systems that combine cryogenic distillation with membrane or pressure swing adsorption (PSA) pre-separation steps. In parallel, life cycle assessment (LCA) and techno-economic analysis (TEA) studies are indispensable to quantify the environmental and economic impacts of each configuration under real-world constraints. Parameters such as global warming potential (GWP), energy return on investment (EROI), and levelized cost of methane (LCOM) are used to benchmark cryogenic technologies against alternatives [79]. Finally, from a policy perspective, targeted incentives for small-scale and modular systems, as well as recognition of liquefied biomethane in national fuel mandates, can significantly accelerate the deployment of cryogenic upgrading technologies. Pilot programs, preferential grid access, and exemptions from fossil gas taxation are potential mechanisms to foster innovation and market entry.

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Abbreviations

The following abbreviations are used in this manuscript:

bcm	billion cubic metres
CAPEX	Capital Expenditure
CFZ	Controlled Freeze Zone
DPCD	Dual-Pressure Cryogenic Distillation
GHG	Greenhouse Gases
HE	Heat Exchanger
LBM	Liquid Biomethane
LNG	Liquefied Natural Gas
OPEX	Operational Expenditure

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