

Communication

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Communication

Synthesis of (S)-4-Benzyl-3-Butyl-1-(2-Cycloheptylethyl)imidazolidine

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Abstract

LiAlH_4 reduction of *tert*-butyl (S)-butyl(1-((2-cycloheptylethyl)amino)-1-oxo-3-phenylpropan-2-yl)carbamate (**1**) gave imidazolidine **2**, while treatment with LDA furnished the β -elimination product, cinnamamide **3**. Both products were fully characterized. Reductive cyclization of *N*-alkylated-*N*-Boc-protected amino acid amides with LiAlH_4 may be a viable synthetic method for trisubstituted chiral imidazolidines.

Keywords: LiAlH_4 ; amide reduction; imidazolidine; β -elimination; cinnamamide

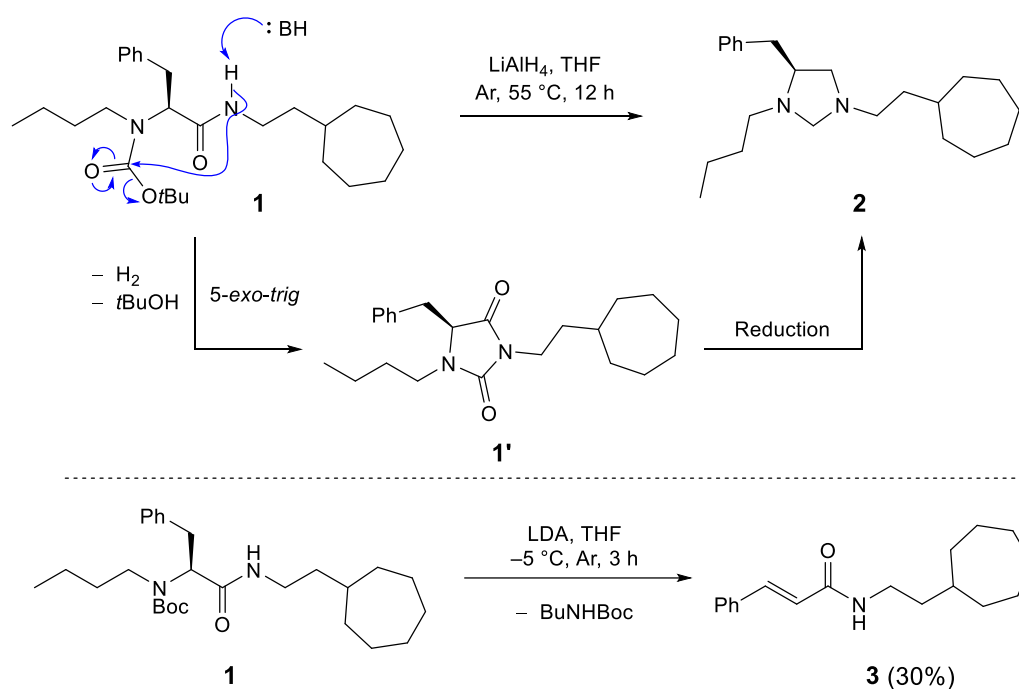
1. Introduction

Five-membered cyclic amins, imidazolidines, are structural motifs found in various natural products [1–5] and synthetic compounds, exhibiting diverse biological activities [6–9], including fungicidal, antiparasitic, antibacterial, antiamebic, antiviral, and anticancer effects [10]. Furthermore, imidazolidines, along with diarylprolinol silyl ethers [11], are workhorses of covalent enamine and iminium ion organocatalysis [12–16].

Following disconnection at the amination carbon of imidazolidines, the starting compounds are (substituted) 1,2-diamines and aldehydes or ketones, which is also the most frequently used method for their synthesis [17]. For a detailed account of synthetic methods for assembling imidazolidines, see the review by Abolfazl Olyaei and Mahdih Sadeghpour and the references cited therein [10]. In this communication, we report the synthesis of an L-phenylalanine-derived imidazolidine **2** prepared from *N*-alkylated-*N*-Boc-protected amino acid amide **1** by reductive cyclization with LiAlH_4 .

2. Results and Discussion

tert-Butyl (S)-butyl(1-((2-cycloheptylethyl)amino)-1-oxo-3-phenylpropan-2-yl)carbamate (**1**), prepared in three steps from L-phenylalanine according to literature procedures [18,19], was treated with excess LiAlH_4 at 55 °C for 12 hours. Purification by column chromatography afforded (S)-4-benzyl-3-butyl-1-(2-cycloheptylethyl)imidazolidine (**2**) in 40% yield (Scheme 1). The formation of imidazolidine **2** can be tentatively explained by initial LiAlH_4 -mediated deprotonation of amide **1**, followed by favorable 5-*exo-trig* cyclization to hydantoin derivative **1'**, which is then reduced to amination **2**. Complete reduction of hydantoin derivatives to cyclic amins with LiAlH_4 is documented in the literature [6,20]. An attempt to cyclize amide **1** to hydantoin **1'** using lithium diisopropylamide (LDA) as a strong base did not yield hydantoin **1'**; instead, the E2 elimination product **3** was isolated in 30% yield (Scheme 1). To better understand the sequence of events in the formation of cyclic amination **2** (cyclization vs. reduction), further mechanistic investigation is required.



Scheme 1. Synthesis of cyclic aminal **2** from *N*-alkylated-*N*-Boc-protected amino acid amide **1**, along with the proposed mechanism for its formation, and synthesis of cinnamamide **3** from **1** using LDA.

The structures of imidazolidine **2** and cinnamamide **3** were confirmed by ^1H and ^{13}C NMR, DEPT 135, HSQC, IR, and high-resolution mass spectrometry. The *trans* configuration around the $\text{C}=\text{C}$ bond of cinnamamide **3** is supported by the coupling constant ($^3J = 15.6$ Hz) in the proton spectrum. Similarly, the proton and carbon spectra of imidazolidine **2** are consistent with those of the closely related imidazoline prepared from L-tryptophan [21]. The most prominent signals in the proton spectrum of **2** are the geminal aminal protons at 3.25 ppm and 3.73 ppm, with a 2J coupling constant of 5.5 Hz, and the corresponding aminal carbon at 77.40 ppm in the ^{13}C NMR spectrum (Figure 1) [21,22].

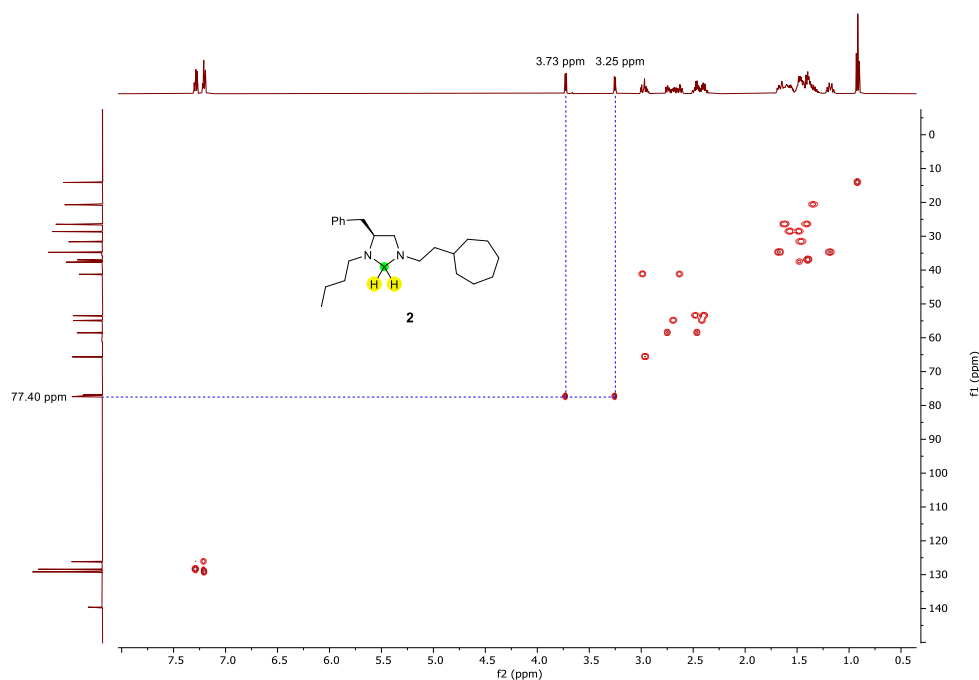


Figure 1. HSQC spectrum of imidazolidine **2** in CDCl_3 .

3. Materials and Methods

Solvents for extractions and chromatography were technical grade and distilled before use. Extracts were dried over technical grade anhydrous Na_2SO_4 . Melting points were determined with an SRS OptiMelt MPA100 Automated Melting Point System (Stanford Research Systems, Sunnyvale, CA, USA). NMR spectra were recorded on a Bruker UltraShield 500 Plus spectrometer (Bruker, Billerica, MA, USA) at 500 MHz for ^1H and 126 MHz for ^{13}C , using CDCl_3 as the solvent and TMS as the internal standard. Mass spectra were recorded on an Agilent 6224 Accurate Mass TOF LC/MS (Agilent Technologies, Santa Clara, CA, USA), and IR spectra were recorded on a PerkinElmer Spectrum BX FTIR spectrophotometer (PerkinElmer, Waltham, MA, USA). Column chromatography was performed on silica gel (Silica gel 60, particle size 0.035–0.070 mm; Sigma-Aldrich, St. Louis, MI, USA). All commercially available chemicals were purchased from Sigma-Aldrich (St. Louis, MI, USA). The starting compound, *tert*-butyl (S)-butyl(1-((2-cycloheptylethyl)amino)-1-oxo-3-phenylpropan-2-yl)carbamate (**1**), was prepared according to literature procedures [19].

3.1. Synthesis of (S)-4-benzyl-3-butyl-1-(2-cycloheptylethyl)imidazolidine (**2**)

To *tert*-butyl (S)-butyl(1-((2-cycloheptylethyl)amino)-1-oxo-3-phenylpropan-2-yl)carbamate (**1**) (294 mg, 0.660 mmol) under argon, LiAlH_4 (3 mL, 2.4 M in THF, 7.20 mmol) was added. The reaction mixture was stirred under argon at 55 °C for 12 h. The mixture was cooled in an ice bath and quenched by careful addition of saturated aqueous NaCl. The resulting semi-solid was extracted with Et_2O (5×30 mL). The combined organic phases were dried over anhydrous Na_2SO_4 , filtered, and the volatile components were evaporated *in vacuo*. The residue was purified by column chromatography (Silica gel 60, $\text{CH}_2\text{Cl}_2/\text{MeOH}$ = 50:1). Fractions containing pure product **2** were combined, and the volatile components were evaporated *in vacuo*. Yield: 92 mg (0.269 mmol, 40%) of colorless oil. $[\alpha]_{\text{D}}^{25} = +58.5$ (0.93 mg/mL, CH_2Cl_2). EI-HRMS: m/z = 343.3101 (MH^+); $\text{C}_{23}\text{H}_{39}\text{N}_2$ requires: m/z = 343.3108 (MH^+); ν_{max} 3062, 3026, 2919, 2853, 2798, 1603, 1495, 1454, 1356, 1237, 1186, 1081, 1031, 902, 742, 698 cm^{-1} . ^1H -NMR (500 MHz, CDCl_3): δ (ppm) 0.92 (*t*, J =7.3, 3H), 1.12 – 1.24 (*m*, 2H), 1.28 – 1.73 (*m*, 17H), 2.34 – 2.52 (*m*, 4H), 2.58 – 2.79 (*m*, 3H), 2.91 – 3.02 (*m*, 2H), 3.25 (*d*, J =5.5 Hz, 1H), 3.73 (*d*, J =5.5 Hz, 1H), 7.16 – 7.25 (*m*, 3H), 7.25 – 7.32 (*m*, 2H). ^{13}C -NMR (126 MHz, CDCl_3): δ (ppm) 14.12, 20.69, 26.46, 26.47, 28.60, 31.63, 34.76, 36.97, 37.66, 41.24, 53.46, 54.92, 58.53, 65.65, 77.40, 126.15, 128.39, 129.16, 139.61.

3.2. Synthesis of N-(2-cycloheptylethyl)cinnamamide (**3**)

To a solution of *tert*-butyl (S)-butyl(1-((2-cycloheptylethyl)amino)-1-oxo-3-phenylpropan-2-yl)carbamate (**1**) (338 mg, 0.760 mmol) in anhydrous THF (3 mL) under argon at –5 °C, LDA (1.0 mL, 2 M in THF, 2.0 mmol) was added over 10 minutes. The reaction mixture was stirred under argon at –5 °C for 3 hours. The mixture was quenched by addition of saturated aqueous NaCl (5 mL). The resulting mixture was extracted with EtOAc (3×15 mL). The combined organic phases were dried over anhydrous Na_2SO_4 , filtered, and the volatile components were evaporated *in vacuo*. The residue was purified by column chromatography (Silica gel 60, EtOAc /petroleum ether 1:2). Fractions containing pure product **3** were combined, and the volatile components were evaporated *in vacuo*. Yield: 63 mg (0.232 mmol, 30%) of white solid; m.p. = 86–89 °C. EI-HRMS: m/z = 272.2009 (MH^+); $\text{C}_{18}\text{H}_{26}\text{NO}$ requires: m/z = 272.2009 (MH^+); ν_{max} 3253, 3083, 2918, 2851, 1666, 1651, 1607, 1552, 1495, 1450, 1344, 1284, 1223, 1174, 1070, 981, 871, 842, 762, 679 cm^{-1} . ^1H -NMR (500 MHz, CDCl_3): δ (ppm) 1.12 – 1.27 (*m*, 2H), 1.36 – 1.66 (*m*, 11H), 1.67 – 1.75 (*m*, 2H), 3.39 (*dt*, J =5.7, 7.5 Hz, 2H), 5.86 (*t*, J =5.8 Hz, 1H), 6.42 (*d*, J =15.6 Hz, 1H), 7.29 – 7.38 (*m*, 3H), 7.44 – 7.51 (*m*, 2H), 7.62 (*d*, J =15.6 Hz, 1H). ^{13}C -NMR (126 MHz, CDCl_3): δ (ppm) 26.42, 28.58, 34.54, 37.02, 37.84, 38.21, 121.00, 127.86, 128.89, 129.67, 135.02, 140.83, 166.02.

4. Conclusions

Reduction of *tert*-butyl (S)-butyl(1-((2-cycloheptylethyl)amino)-1-oxo-3-phenylpropan-2-yl)carbamate (**1**) with excess LiAlH_4 yielded imidazolidine **2**. Treatment of **1** with LDA gave the β -

elimination product, cinnamamide **3**, rather than the hydantoin cyclization product. The formation of imidazolidine **2** and its previously reported tryptophan analogue [21] suggests that reductive cyclization of *N*-alkylated-*N*-Boc-protected amino acid amides with LiAlH₄ may be a viable synthetic method for trisubstituted chiral imidazolidines. Further studies of the reaction scope and mechanism of the reported transformation are necessary.

Supplementary Materials: The following supporting information can be downloaded at the website of this paper posted on Preprints.org: copies of ¹H- and ¹³C-NMR spectra; copies of DEPT 135 and HSQC spectra, copies of IR and HRMS spectra.

Author Contributions: Conceptualization, L.C., U.G., J.S., and B.Š.; methodology, L.C. and U.G.; software, L.C., M.S., U.G., J.S., and B.Š.; validation, L.C., N.P., M.S., U.G., J.S., F.P., and B.Š.; formal analysis, U.G., M.S., and L.C.; investigation, M.S., L.C., and U.G.; resources, L.C., U.G., and J.S.; data curation, L.C., N.P., M.S., U.G., J.S., and B.Š.; writing—original draft preparation, L.C., U.G., J.S., and B.Š.; writing—review and editing, L.C., N.P., U.G., J.S., F.P., and B.Š.; visualization, L.C., M.S., U.G., B.Š., and J.S.; supervision, U.G.; project administration, U.G. and J.S.; funding acquisition, U.G. and J.S. All authors have read and agreed to the published version of the manuscript.

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