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Article

The Orbital Approach to Understanding the Quantum Geometry of Solids Involves Correlated Electron Pairs Oscillating in Resonant Quantum States, Which Leads to the Phenomenon of Superconductivity

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Abstract

Developing the Bardeen-Cooper-Schrieffer (BCS) theory of superconductivity was one of the most difficult problems in 20th-century physics. According to the BCS theory, superconductivity is entirely controlled by the pairing energy scale, which creates the superconducting energy gap. Today, the wider applicability of the BCS theory has been increasingly questioned. Einstein attempted to understand superconductivity using the idea that "supercurrents are carried through closed molecular chains where electrons undergo continuous cyclic exchanges." Although no additional justification for the existence of such conduction paths was provided, this approach was based on the belief that superconductivity is strongly connected to the specific chemistry of a material and depends on the presence of a state that links the outer electrons of an atom or molecule with those of its neighbors. Fermions may be susceptible to further capture by bosons, resulting in mixed configurations. This work develops a superconducting pairing mechanism based on the correlation of the same type of fermion exchanging a boson, creating oscillating, resonant quantum states. Our use of the concept of resonating valence bonds (RVB) confirms that electrons from neighboring atoms can form valence bonds to create pairs. "The specific chemistry" is represented by the formation of molecular structures in solids. We emphasize that this non-classical analysis could serve as a generalized or universal model of correlated pair formation.

Keywords: resonating valence bonds; electron pairs; and molecular structures in solids

1. Introduction

Despite decades of intense theoretical and experimental efforts, a microscopic theory of unconventional superconductivity remains elusive. The Bardeen–Cooper–Schrieffer (BCS) theory of superconductivity is based on the Fermi liquid model, which describes electrons as non-interacting, ballistic quasiparticles while disregarding the Coulomb interaction. In BCS theory, superconductivity is primarily determined by the pairing energy scale, which gives rise to the superconducting energy gap. Another group of theories has evolved to explain superconductivity, particularly in the field of unconventional superconductors.



For example, the concept of resonating valence bonds (RVB) suggests that electrons from neighboring atoms can create valence bonds to form pairs, which can be doped to achieve a stable superconducting ground state ^{1,2}. The RVB state is a mixture of singlet pairings of electrons at different sites arranged in a specific way. It is a distinct singlet, lacks obvious long-range order, and is fluid, exhibiting quantum transport of spin excitations. The quantum spins continuously alter their singlet partners ¹. Strongly interacting chains of fermions are expected to display two types of collective excitations: spinons, which carry only spin, and holons, which have only charge ³. The low-energy excitations from the RVB states comprise these fractional quasiparticles (spinons and holons), which facilitate spin-charge separation ^{3,4}.

Quantum spin fluctuations in a low-dimensional or frustrated magnet disrupt magnetic ordering while preserving spin correlations 4 . Such fluctuations have been a central topic in magnetism due to their relevance to high-temperature superconductivity and topological states. However, despite numerous efforts, utilizing such spin states has proven to be quite challenging. Based on experimental and theoretical evidence, we know that the physical carrier of the superconductivity phenomenon is a pair of electrons (2e), and the isotopic effect $T_c = 1/M^{\alpha}$, with $\alpha \square \frac{1}{2}$. To justify the experimental transport of 2e, Cooper tried to convert the Coulomb repulsion into attraction by using the electron-phonon interaction as a foundation.

The electron-phonon interaction produces several effects. For instance, the scattering of electrons through the emission and absorption of phonons contributes to the resistance of electric current. The polaronic effect involves a renormalization of the charge and mass of electrons as they move through the lattice of a metal. Screening via a dielectric function of the Coulomb interaction with another electron yields a repulsive positive potential, as they are charges of the same type. It is possible that Cooper based his argument on this last effect to justify the existence of correlated electron pairs mediated by electron-phonon interactions and to explain the superconductivity observed in BCS theory.

Recently, the orbital correlation effect in p- and d-electron-based solid systems has been described as a means to realize unusual forms of charge-density-wave order $^{5-10}$. The causes of "Cooper pairs" formation in unconventional superconductors are often elusive due to the rarity of experimental signatures that link to a specific pairing mechanism. The typical mechanism of exchange interaction between localized electrons involves the virtual hopping of electrons to neighboring sites and back, which facilitates coupling between spin orientations. When hopping occurs, the energy decreases 11,12 .

In 2018, Riera et al introduced the Theory of correlated pairs of electrons oscillating in resonant quantum states to reach the critical temperature in a metal ¹³. Coherent, resonant excitation drives coherent oscillations in a solid-state quantum emitter, which have been described and proposed as a means to generate entangled photon pairs ¹⁴. Recently, the Correlated Electron Pairs at Oscillating Resonant Quantum States (CEP-ORQS) has been theoretically proven to explain the superconductivity observed in polyaromatic compounds, such as benzene, and the nitrogenous bases of DNA ^{15,16}.

Neither the BCS nor the RVB theory fully explains the phenomenon of superconductivity. The BCS mechanism relies on the attraction between electrons with opposite momentum and spin, resulting from their interaction with the vibrations of the crystal lattice, known as phonons. In contrast, the RVB theory treats spin and charge as separate entities, allowing the spectrum of electronic excitations to be presented as two distinct branches: charge without spin and spin without charge. In this work, we employ elements from both approaches to clarify the CEP-ORQS, which leads to the emergence of superconducting states. We utilize the concept of orbital approximation to describe the formation of molecular clusters within solid crystals, where an external agent (such as pressure or temperature) induces hybridization between neighboring orbitals, resulting in a readjustment of the electronic configuration of those free electrons. Due to Coulomb repulsion, two electrons tend to repel each other, and by the Pauli principle, they cannot occupy the same space. As a result, an electron is compelled to seek permissible space within its corresponding orbital. The jump

of this electron to a lower level not only alters its spin and sign but also emits a quantum of vibrational energy that is captured by a nearby electron, forcing it to change position. This constitutes a half-oscillation. The same phenomenon occurs during a second oscillation, forming CEP-ORQS.

For a better understanding, the paper will address the topics as follows: Section I discusses the orbital effects and the quantum geometry in solids that contribute to "molecular cluster formations". Section II explains how elements from the BCS and RVB theories converge in the formation of CEP-ORQS. Section III expands the physical-mathematical model for the pair correlation. Finally, in section IV, we discuss our results.

SECTION I. The orbital effects in quantum geometry of solids

Three characteristics—charge, spin, and orbital—define electrons in a solid ^{17,18}. Recent studies have described the connection between orbital structure, order, and electron hopping that leads to exchange in solids. Two types of electron states exist: itinerant electrons, which align with the molecular orbital description in chemistry and are effectively represented by band theory, and localized electrons, which arise from substantial electron-electron interactions in solids (Heitler-London description) ^{12,19}. Changes in electron behavior—whether in the typical free-electron-like or band description—lead to localization at specific sites, thereby altering the properties of the systems ²⁰. Regions of allowed states create energy bands, which are typically separated by forbidden areas known as energy gaps.

The weak coupling approximation, which treats a periodic potential of the lattice as a perturbation, and the tight-binding approximation are the two primary methods for describing energy bands ^{21,22}. When there is one electron per site, the material in band theory exhibits partially filled bands and behaves as a metal. Conversely, when interactions are strong enough, some bands become filled, while the upper bands, separated from the occupied ones by an energy gap, remain empty, indicating a band insulator or semiconductor ^{19,20}.

In the tight-binding approximation, the bands form through the hopping of electrons between sites. Metal-insulator transitions, and vice versa, can occur due to structural changes that introduce lattice distortions, opening a gap precisely at the Fermi surface ²³. With itinerant electrons, the metallic system displays ferromagnetic coupling, occupying different orbitals. The band of metals is typically not completely filled. The electron occupation number can change, or an electron hopping parameter could affect the system ^{17,20} (Figure 1).

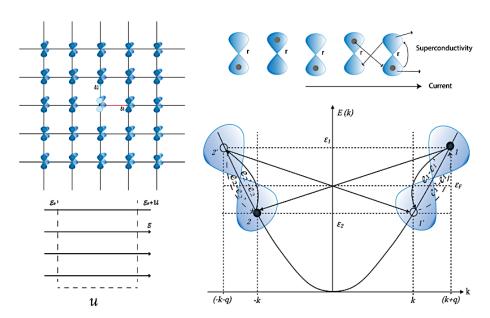


Figure 1. Orbitals directed toward neighbors in a solid may be delocalized like molecular orbitals. The index r denotes the coordinate of each atom in the crystal lattice such that the atomic orbital \emptyset_r has its center at r. δ are

the relative vectors that connect a given atom in the lattice with its first neighbors. Mobile electrons hop from site to site, orienting their spin and forming a state with opposite spins (S = 1), whereas the spin at localized orbitals is only ½. The lowest energy state is known as the ground state, characterized by a total spin of zero. The independent-particle Hamiltonian governs four excited states for the system. They all have the same excitation energy, E. In the absence of interactions (U=0), the eigenstates of several particles are simply slater determinants of the eigenstates of one particle. On the other hand, given the translation symmetry, the eigenstates of a particle are also eigenstates of the translation operator. On the right, orbital selectivity modifies the band structure and influences the effective interactions, favoring the pairing states.

The spin-orbit coupling mixes up and down spin states, producing fluctuations in molecular geometry inside the solids ¹⁸.

SECTION II. The convergence between the BCS and RVB can be used to explain the ORQS

In natural systems, orbital degrees of freedom are essential. Certain energy levels become degenerate in isolated atoms that exhibit complete spherical symmetry. However, this degeneracy is disrupted in solids due to environmental interactions. Furthermore, space orbitals have unique directional properties 24 . The occupancy of specific orbitals determines the final geometry of the solid 25 , primarily leading to variations in octahedral geometry due to lower symmetry 2627 . The crystal field splitting results from the combination of d-p orbital hybridization and Coulomb repulsion 23 . Atomic radii, the degree of covalency, and other important factors can influence crystal orbital distribution and configuration 28 .

The state of the system, where each electronic state has its distortion, is known as the vibronic state ²⁹. This cooperative lattice distortion results in a specific orbital being occupied at each center ³⁰. Temperature and pressure exhibit different types of structural transitions in solids ^{31,32}. The Pauli exclusion principle for parallel spins prevents this process, but antiparallel spins permit it, resulting in an energy gain ³³. Consequently, effective electron hopping occurs between the half-filled orbitals with antiparallel spins at their centers ^{12,19}. This is well-explained by Goodenough-Kanamori rules ^{34–36}

The Fourier transform of the electron-electron Coulomb interaction must be modified by the electronic dielectric function in the manner of Thomas-Fermi screening to account for the effect of other electrons in screening the interaction between a given pair:

$$\frac{4\pi e^2}{k^2} \to \frac{4\pi e^2}{k^2 \varepsilon_{cl}} = \frac{4\pi e^2}{k^2 + k_0^2} \tag{1}$$

However, the ions also screen interactions, and we should not have used the electronic dielectric constant ε_{el} , but rather the total dielectric constant ε , which considers both electrons and phonons. We recommend replacing (1) with:

$$\frac{4\pi e^2}{k^2} \to \frac{4\pi e^2}{k^2 \varepsilon} = \frac{4\pi e^2}{k^2 + k_0^2} \left[1 + \frac{\omega^2(k)}{\omega^2 - \omega^2(k)} \right] \tag{2}$$

Therefore, the effect of the ions is enhanced by a correction factor that depends on the frequency and the wave vector.

$$\frac{1}{\varepsilon} = \left(\frac{1}{1 + \frac{k_0^2}{k^2}}\right) \left[1 + \frac{\omega^2}{\omega^2 - \omega^2(k)}\right] \tag{3}$$

BCS Theory's criterion for forming pairs is based on the same Cooper principles outlined in Equation (2.6) from Bardeen et al. (1957) ^{37,38}. An effective potential between the Coulomb repulsion and the negative electron-phonon interaction should be about 2e. However, no experimental evidence supports this hypothesis, as an effective potential between the Coulomb repulsion and the electron-phonon interaction does not guarantee the formation of pairs. Nevertheless, through shielding, Cooper established a negative effective potential between the electron-electron and electron-phonon interactions. In the first hypothesis, a negative potential causes the electrons to collapse due to the infinite approach. To achieve an effective negative potential, the electron-phonon interaction must exceed the Coulomb repulsion, or the energy difference between the correlated electrons must be less than the phonon energy, which is mathematically feasible but not physically realizable. One straightforward consequence is that the BCS work must articulate the commutation rules governing the creation and annihilation operators of Cooper pairs.

When computing commutators in the BCS framework, the mixed rules of fermions and bosons are derived solely from the electron creation and annihilation operators. To address this, we recommend considering that if \hat{b}_q^+ and \hat{b}_q^- are the creation and annihilation operators of a phonon (the theory can be generalized to bosons) in the state q, and \hat{C}_k^+ and \hat{C}_k^- are the creation and annihilation operators of an electron (the theory can be generalized to fermions of the same type) in the state k. Then, the following commutation rules must be satisfied:

$$\begin{bmatrix} \hat{b}_{q}^{+}, \hat{b}_{q}^{-} \end{bmatrix} = \hat{b}_{q}^{+} \hat{b}_{q'}^{-} - \hat{b}_{q'}^{-} \hat{b}_{q}^{+} = \delta_{qq'}
\begin{bmatrix} \hat{C}_{k}^{+}, \hat{C}_{k'}^{-} \end{bmatrix}_{\perp} = \hat{C}_{k}^{+} \hat{C}_{k'}^{-} + \hat{C}_{k'}^{-} \hat{C}_{k}^{+} = \delta_{kk'}$$
(4)

Where the sign + under the bracket symbol indicates the anti-commutator.

II.1. Rules for correlation

Different conditions exist for two fermions to be correlated through phonon exchange:

- The energy difference between the two fermions must equal the phonon's energy
- Fermions should possess opposite momentum, but vary in the phonon's momentum.
- The ORQS of the fermion pair is established when a phonon is exchanged via emission and absorption process.
- The total momentum or sum of the pair of fermions is $\pm q$ the momentum of the phonon, and the difference is $\pm (2k+q)$.

The system must satisfy specific critical states for these conditions to be fulfilled.

Figure 2 presents a diagram of the conduction band of a metal at the critical temperature T_C . The vertical axis represents energy $E\left(k\right)$, which is proportional to the temperature in the region $0 \le T \le T_c$. The horizontal axis represents the momentum k. Two electrons are initially in positions 1 and 2 with momentum k+q and -k, respectively. There are holes at positions 1' and 2' with momentum -k-q and k. The energy difference between the states (2 and 2') and $(1 \ and \ 1')$

is the energy of the phonon $\hbar\omega_q$. These states are symmetrically located above and below the Fermi level, indicated by the horizontal line. It coincides with the Fermi energy E_F .

SECTION III. The physical-mathematical model for the correlation of electron pairs in Oscillatory Resonant Quantum States.

Electrons are correlated in pairs that oscillate in ORQS by exchanging a phonon with energy $\hbar\omega_q$ and momentum q. Two interactions are involved: the Coulomb repulsion and the electron-phonon-electron interaction. Here, electron 1 (above the Fermi surface) with momentum k+q and kinetic energy $E\left(k+q\right)$ correlates with electron 2 (below E_F) with momentum -k and kinetic energy $E\left(-k\right)$. Moreover, the energy difference between them is exactly the energy of the phonon. Figure 2

The energy band satisfies the following relations:

$$E(k+q) = E(-k-q) \quad and \quad E(k) = E(-k)$$
 (5)

Also,

$$E(k+q)-E(k)=\hbar\omega_q$$
 and $E(-k-q)-E(-k)=\hbar\omega_q$ (6)

Therefore,

$$\frac{\hbar^2 \left(k+q\right)^2}{2m} + \frac{\hbar^2 k^2}{2m} = \hbar \omega_q \tag{7}$$

And

$$\frac{\hbar^2 \left(k \cdot q\right)}{m} + \frac{\hbar^2 q^2}{2m} = \hbar \omega_q = \hbar c q \quad (8)$$

Where $k \cdot q = |k| |q| \cos \alpha (k, q)$, $k_F = k + \frac{q}{2}$, and $c = \frac{\hbar k_F}{m}$.

However, for the ORQS, electrons tend to have equal and opposite momentum:

$$\alpha(k,q) = \begin{cases} 0 & \text{si } q < 0 \\ \pi & \text{si } q > 0 \end{cases} \tag{9}$$

Then,

$$q\left(\frac{\hbar^2}{m}k + \frac{\hbar^2 q^2}{2m}\right) = \hbar cq \Rightarrow \frac{\hbar^2}{m}\left(k + \frac{q}{2}\right) = \hbar c \quad (10)$$

The total initial momentum for the ORQS 1 and 2 is: k+q-k=q. One phonon is absorbed by the electron in state 2. The total final momentum is k-k-q=-q when the electron in the state 2'emits one phonon. The variation in the momentum of the electrons in the PRQS 1 and 2 is:

$$k + q + k = 2k + q \tag{11}$$

$$-(k+q)-k = -2k-q$$
 (12)

As we can see from (11) and (12), electrons oscillate in the ORQS, trying to match their individual momentum. They reverse the sign of the momentum between pairs of states. The sum and total difference of the electron momentum in the ORQS are equal but opposite in sign, which justifies the

existence of these states. At each temperature in the range, electrons can only occupy the permitted states.

When the Coulomb repulsion (described by the potential $V^{e_1e_2}$ and screened by the dielectric constant of Thomas-Fermi) is connected, electron 1 emits a phonon with momentum q to electron

2 and now occupies the position of hole 1' with momentum k and kinetic energy $\varepsilon(k)$.

Meanwhile, electron 2 absorbs the phonon emitted and occupies hole 2's position with momentum

-k-q and kinetic energy $\varepsilon(-k-q)$. This first process defines half oscillation. The opposite

happens in the second half of the oscillation. The 1' and 2' electrons exchange a phonon momentum -q and return to their original positions in 1 and 2. Note that, in the last exchange, the first electron absorbs a phonon (given by the potential $V^{e_1fe_2}$). The second electron emits a phonon (given by the potential $V^{e_1fe_2}$). These indicate that the potentials $V^{e_1fe_2}$ and $V^{e_1fe_2}$ have opposite signs and are responsible for the complete static oscillation of the electrons, that is, in the absence of an external field.

At the moment when $T = T_C$ the Hamiltonian of the system becomes

$$H_T = H_p + H_{\text{int}} \tag{13}$$

Therefore, we can say that:

$$H_{\rm int} = V^{e_1 e_2} \pm V^{efe} \tag{14}$$

Where the positive sign corresponds to $V^{\it efe}=V^{\it e_1fe_2}$ for the first half oscillation and the negative sign corresponds to $V^{\it efe}=-V^{\it e_1fe_2}$ for the second half oscillation. We can also say that $V^{\it e_1fe_2}=-V^{\it e_1fe_2}$.

III.1. The wave function of the Electrons Pair Correlated in the RQS in Every Oscillation

When $T=T_C$ we can write the wave functions of the individual electrons 1 and 2 as $\left|k+q\right>$ and $\left|-k\right>$, and for the phonon $\left|\pm q\right>$. The correlated pair in the RQS has $\left|k+q;-k;q\right>$ a wave function. The same happens with the individual holes 1' and 2', whose wave functions are $\left|-k-q\right>$ and $\left|k\right>$. For the pair of holes in the RQS, we have $\left|-k-q;k;-q\right>$ (Figure 2).

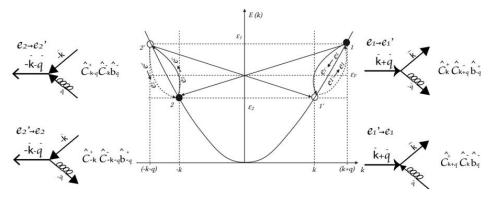


Figure 2: The central part represents a diagram of energy E(K) versus momentum K of a metal conduction band that becomes a superconductor upon reaching the critical temperature. In the initial

state, electrons 1 and 2 are correlated and oscillate in the resonant quantum states in two oscillations, exchanging a phonon. On the left side, the oscillation process of the electron in 2 is represented, which absorbs the phonon emitted by 1 passing to 2′, and then, this one emits the phonon 1′ to 2 passing again to 2′, for an arbitrary angle of the momentum of the electron and the phonon. On the right, the oscillation between 1′ and 1 is represented.

III.2. Internal State Characterization of the Electronic Pair in the RQS

For the calculation of the initial state of the oscillations in the RQS, the pair formed by electrons 1 and 2 with a wave function $|k+q;-k;q\rangle$, we apply $H_{\rm int}$:

$$H_{\text{int}} \left| k + q; -k; q \right\rangle = \left(V^{e_1 e_2} + V^{e_1 f e_2} \right) \left| k + q; -k; q \right\rangle = \left(E^{e_1 e_2} + E^{e_1 f e_2} \right) \left| -k - q; k; -q \right\rangle \tag{15}$$

Where $E^{e_1e_2}$ $E^{e_1e_2}$ is the energy corresponding to the Coulombian repulsion operator between electrons 1 and 2. $E^{e_1fe_2}$ is the energy that results from the exchange of the phonon emitted by electron 1 and absorbed by electron 2. Note that the wave function $|k+q;-k;q\rangle$ is not a self-function of the Hamiltonian $H_{\rm int}$, so we apply $H_{\rm int}$ again to obtain the initial wave function $|k+q;-k;q\rangle$:

$$H_{\text{int}}H_{\text{int}}|k+q;-k;q\rangle = \left(V^{e_{1}\cdot e_{2}\cdot} + V^{e_{1}\cdot fe_{2}\cdot}\right)\left(V^{e_{1}e_{2}} + V^{e_{1}\cdot fe_{2}}\right)|k+q;-k;q\rangle$$

$$= \left(E^{e_{1}e_{2}} + E^{e_{1}fe_{2}}\right)\left(V^{e_{1}\cdot e_{2}\cdot} + V^{e_{1}\cdot fe_{2}\cdot}\right)|-k-q;k;-q\rangle =$$

$$= \left(E^{e_{1}e_{2}} + E^{e_{1}fe_{2}}\right)\left(E^{e_{1}\cdot e_{2}\cdot} + E^{e_{1}\cdot fe_{2}\cdot}\right)|k+q;-k;q\rangle$$
(16)

In the last equation (16), we considered that $V^{e_1\cdot e_2\cdot}=V^{e_1e_2}$ and $V^{e_1\cdot fe_2\cdot}=-V^{e_1fe_2}$. If $E^{e_1e_2}=E^{e_1\cdot e_2\cdot}$ and $E^{e_1fe_2}=-E^{e_1\cdot fe_2\cdot}$ then, we obtain:

$$H_{\text{int}}^{2} | k+q; -k; q \rangle = (V^{e_{1}e_{2}} - V^{e_{1}fe_{2}}) (V^{e_{1}e_{2}} + V^{e_{1}fe_{2}}) | k+q; -k; q \rangle$$

$$= (E^{e_{1}e_{2}} - E^{e_{1}fe_{2}}) (E^{e_{1}e_{2}} + E^{e_{1}fe_{2}}) | k+q; -k; q \rangle = (E^{e_{1}e_{2}} - E^{e_{1}fe_{2}}) | k+q; -k; q \rangle$$
(17)

The interaction Hamiltonian $H_{\rm int}$ does not have self-states of the CEP-ORQS because it only characterizes half of the oscillation of the pair. $H_{\rm int}^2$ is the operator that describes the internal state of the CEP-ORQS for a complete oscillation.

It is crucial to interpret the square of the Hamiltonian as the square of the binding energy Δ^2 between the CEP-RQS in the range of $0 \le T \le T_C$. The participating energies Δ^2 , $E^{e_1e_22}$ and $E^{e_1fe_22}$ are functions that depend only on the temperature and the T_C $\left(0 \le T \le T_C\right)$.

The dispersion relation of the CEP-ORQS in $\left(0 \le T \le T_C\right)$ is:

$$\Delta^2 = E^{e_1 e_2 2} - E^{e_1 f e_2 2} \tag{18}$$

Now, we must find the explicit dependence of the energies Δ , $E^{e_1e_2}$ and $E^{e_1fe_2}$ on the temperature and the T_C . For that, we will analyze the instant at which $T=T_C$ is reached. When T_C is equivalent to the room temperature, electrons are in individual states. They are correlated at 0K. When $T=T_C$ the electron-electron interaction is the only participant and the total energy of the electrons depends on it. If we lower the temperature until the T_C , the electron-phonon-electron interaction is connected, the total energy will rely on T for each T_C and so:

$$\Delta = \Delta(T; T_C), E^{e_1 e_2} = E^{e_2 e_1}(T_C) \text{ and } E^{e_1 f e_2} = -E^{e_1, f e_2}(T; T_C)$$
 (19)

If we consider that $E^{e_1e_2}$ has a linear temperature dependence, then, we can write:

$$E^{e_1 e_2} = \rho k_B T_C$$
 and $E^{e_1 e_2 2} = (\rho k_B T_C)^2$ (20)

Where ρ is the parameter that characterizes the qualitative nature of the material to achieve the oscillations.

However, the temperature dependence is not linear $E^{e_1fe_2}$. It is more complicated because we need a potential that involves, simultaneously, the residual Coulomb repulsion and the interactions of both electrons with the phonon. That is why we propose the dependence as follows:

$$E^{e_1 f e_2} = \rho k_B \sqrt{T_C T}$$
 and $E^{e_1 f e_2 2} = (\rho k_B)^2 T T_C$ (21)

Now, the dispersion law that gives us the binding energy of the CEP-ORQS is:

$$\Delta^{2}\left(T;T_{C}\right) = \left(\rho k_{B}T_{C}\right)^{2} - \left(\rho k_{B}\right)^{2}TT_{C} \tag{22}$$

Or

$$\Delta(T; T_C) = \rho k_B T_C \left(1 - \frac{T}{T_C} \right)^{1/2} \tag{23}$$

This last equation coincides with that reported in equation (3.31) of the BCS article ³⁷ and Buckingham's suggestions.

In equation (3.31) of BCS, the parameter is considered constant and equal to 3.2 for all materials. In our work, the parameter characterizes each material to undergo CEP-ORQS. If, in the previous equation, we take $T \to 0K$, we obtain the parameter from the magnitudes that are measured experimentally:

$$T \to 0K, \quad \rho = \frac{\Delta(0; T_C)}{k_B T_C}$$
 (24)

It remains constant yet varies for each material. We refer to it as the correlation parameter (see Table $1 \text{ in } ^{13}$.

In the same way, we can calculate the highest binding energy for a metal, which represents the energy accumulated when $T \to 0K$:

$$\Delta(0;T_C) = \rho k_B T_C \tag{25}$$

This means that as the temperature decreases from T_C to zero, energy is accumulated as the binding energy of the pair.

In conclusion, the equation for calculating the binding energy of the pair as a function of temperature and critical temperature is:

$$\Delta(T;T_C) = \Delta(0;T_C) \left[1 - \frac{T}{T_C} \right]^{\frac{1}{2}}$$
 (26)

SECTION IV. DISCUSSION

Families of strongly correlated electron materials are well established. These compounds exhibit fascinating and technologically appealing electronic properties ^{2,39}. They also possess complex phase diagrams, including both conventional and exotic phases that reveal new physics. The study of strongly correlated electron systems is currently one of the most active and exciting fields in condensed matter physics as well as in other disciplines ^{40,41}.

In superconducting states, two electrons with opposite momentum attract each other, forming a bound pair. The pairing mechanism involves coupling electrons and phonons, which are quantum representations of lattice vibrations. Cooper is widely recognized as the father of two-electron correlation. He demonstrated that two electrons could pair up by considering interactions with phonons to explain superconductivity in metals. This phenomenon is also referred to as superconductivity at low critical temperatures.

Today, new compounds have emerged whose superconductivity cannot be explained by BCS ^{42,43}, and they are referred to as "non-conventional" superconductors ⁴⁴. Quantum spin liquids are systems that avoid magnetic long-range order down to absolute zero and exhibit a variety of remarkable phenomena due to an intrinsic tendency to form massive quantum superpositions of local, product-like state wavefunctions ^{45,46}. In general, the interplay between covalent bond formation, spin-orbit coupling, and intra-atomic exchange interaction is essential for analyzing the particular physical properties seen transition metal oxides ^{47–53}. One of the most important examples of this system is the RVB state.

The RVB model explains a superposition of fluctuating singlet pairs. RVB correlations and excitations are limited to nearest neighbors. Such valence bonds can be formed between nearest-neighbor spins and between spins beyond nearest neighbors. In this paper, we propose a different interpretation of these correlations and excitations. We argue that the next-nearest-neighbor couplings and multiple-spin exchange are favored by specific geometries adopted by a material under specific conditions. For instance, doping the pure La₂Cu₀₄ in the RVB state is compensated not by a displacement of the pseudo-Fermi surface as in a conventional insulator, but by the creation of boson "hole" excitations, as long as the system remains in the RVB state. Therefore, the RVB state represents the equilibrium condition. The holes can experience Bose condensation and form a superconductor ⁵⁴.

The electronic (spin, charge, orbital) and structural degrees of freedom contribute to superconductivity. Orbital selectivity alters the band structure and affects the effective interactions, promoting pairing states 55,56 . Strong orbital selectivity can lead to gap anisotropy. Additionally, hybridization with neighboring orbitals supplies the holes with kinetic energy. Specifically, the d_Z^2 orbitals exhibit strong interlayer hopping and can form a localized spin-singlet dimer with a pair of electrons 27 .

In a single atom, the double occupation of an orbital is associated with a Coulomb repulsion term $U_{11}=U_{22}=U$. The kinetic energy or tunneling between orbitals of different atoms gives rise to an antiferromagnetic coupling 20 . A crystal is a periodic arrangement of infinite atoms, so the number of atomic orbitals $\{\varphi r\}$ is infinite. As in the case of the molecule, to build a theory of low energies, we will be left with a single φ orbital per atom.

Additionally, a specific type of orbital ordering is linked to dimensionality reduction (orbital-dependent valence bond condensation) ⁵⁷. Two and one-dimensional systems frequently exhibit unusual magnetic properties and lead to novel types of phase transitions ^{57–59}. One-dimensional metallic systems can alternate between short and long bonds, showing significant and minimal intersite hopping ⁶⁰, which creates a gap at the Fermi surface and results in a decrease in the energies of the occupied electron states. The unusual pattern of short–intermediate–long–intermediate bond alternation in solid lattices produces higher-order interaction structures and imparts extraordinary properties ^{61,62}. In such dimers, the metal-metal distance is generally shorter than the corresponding distance in the original material ⁶³. The transition from a regular structure to a dimer structure under pressure has also been recorded ⁶⁴.

Dimer clusters naturally occur in some crystal structures 65 . Orbitals directed toward neighbors may behave as delocalized, similar to molecular orbitals, while the electrons in other orbitals remain localized. In transition metals with a honeycomb lattice, electrons hopping between p orbitals create structures that remain as hexagons, a state akin to that of a benzene molecule, with the same molecular orbitals. In fact, as in benzene, six of these molecular orbitals form bands in a solid rather than levels.

Conversely, several polyaromatic hydrocarbons, such as benzene, have been identified as topological superconductors $^{15,65-67}$. The (4n+2) π -electrons rule can be split into two pairs of (2n+1) each, namely, electrons and holes with spin up and down, and can be understood as the winding number in global next-nearest-neighbor hopping 67 . Intrinsic topological superconductors do not rely on specific external perturbations 65 . Using the Valence-shell electron-pair repulsion (VSEPR) theory, aromaticity and superconductivity in benzene have been explained in terms of the CEP-ORQS model 15,16 . VSEPR theory can describe the spatial arrangement of almost all electrons. Similar to RVB, VSEPR considers orbital interactions and correlations.

2. Conclusions

Systems exhibit orbital-selective behavior, with different orbitals responding uniquely to external perturbations. In conducting solids, electrons are localized in non-hopping "fixed" orbitals, one at each lobe. Meanwhile, mobile electrons in orbitals hop from site to site, aligning their spins parallel to the spins of the localized electrons at each site. Reorganizing these spin-½ objects generates magnetic ordering and properties. For superconductivity to occur, there must be orbitals with free electrons (or lone pairs of electrons), available spaces in empty orbitals (orbital holes), and hybridized orbitals. Some systems naturally possess these characteristics (for instance, some polyaromatic hydrocarbons), while others require external modifications, such as changes in temperature (to reach the Tc) or pressure. One could hypothesize that all models of the so-called unconventional or high-temperature superconductors may contain (or be doped with) at least molecular crystals containing O₂, S, or N₂ molecules as structural units, as well as carbon-based materials. These elements have

partially occupied p orbitals with free electrons and holes, which create vacancies and allow for the freedom of movement.

Furthermore, the partial occupation is significantly easier to achieve for the d shell of transition metal ions. They have outer s valence electrons that shield the d electrons by forming bonds with their neighbors, which leaves the d electrons weakly hybridized. This enables an efficient atomic-like intra-shell exchange interaction within a crystal.

Orbital-dependent directional hopping is proposed as a universal resource to explain superconductivity in solids. This work offers a physical-mathematical model for correlating two electrons by exchanging a boson, which operates in every equally charged fermion. The model elucidates the self-doped orbital-selective transition across valence transition-driven superconductivity. Two fermions with the same charge can only be attracted in the presence of bosons. Fermions must oscillate in specific states where total energy is conserved, and the dispersion law is quadratic. This occurs only when a certain critical state is reached. Three operators of simple particles form the operators of creation and annihilation of pairs: two fermions and one boson. The energy difference of the correlated fermion pair is equal to the energy of the boson, and the difference in the fermions' momentum is less than that of the boson. There is always a parameter that characterizes each particular superconductor: ρ . It represents the qualitative nature of the material to achieve oscillations.

In the BCS formalism, reducing the electronic kinetic energy at the superconducting transition would require a Bose condensation of the superconducting carriers, similar to aromaticity. However, the quantum statistics of Cooper pairs differ from those of Bose-like particles. Therefore, BCS theory and the reduction of electronic kinetic energy in the superconducting state are mutually exclusive. Einstein attempted to understand superconductivity by positing that "supercurrents are carried through closed molecular chains where electrons undergo continuous cyclic exchanges." Although no further justification for the existence of such conduction paths was provided, this approach was based on the view that superconductivity is deeply rooted in the specific chemistry of a given material, and on the existence of a state that connects the outer electrons of an atom or molecule with those of its neighbors. This work develops a superconducting pairing mechanism based on the correlation of the same type of fermion when they exchange a boson, creating the CEP-ORQS. "The specific chemistry" is given by forming molecular structures in solids.

The electronic structure is central to understanding the physical and chemical properties of different compounds. In superconductivity, electrons cannot condense into a single state due to Pauli's principle; they are compelled to oscillate.

"Nature loves to oscillate"---Theodore Holmes Bullock

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