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## Article

# Exploring Solar Drying of Henna (*Lawsonia inermis*) and Its Effect on the Bioactivities

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**Abstract:** This study has two main focuses: the first focus is to explore the design and the behavior of a locally built solar dryer during processing henna leaves. This was performed through the determination of the most important parameters that have effect on the drying process. Appropriately, it was found that the outlet air temperature of the solar collector depends entirely on the solar radiation, Moreover, the inlet temperature of the air to the drying chamber has also an important impact on the moisture content of the product. It was found that most of the studies dealing with solar drying do not take into consideration the quality of the final product. Accordingly, the second objective of this study is to present the effect of drying process on the quality of the drier samples. For this purpose, the bioactivity of the samples, in terms of determination of the total phenol content, total flavonoid content, and antioxidant activities was performed. Three different samples: fresh sample, samples dried with the presented solar dryer and open solar dried samples are tested. It was found that samples dried with the solar dryer presented the best results.

**Keywords:** open sun drying; convective solar drying; henna leaves; operating conditions; bioactivities; final product quality

## Introduction

In different countries of the Mediterranean region, solar drying is still considered as a practical method that can be used for the preservation of products such as fruits, vegetables and grains. Even though, the direct effect on the physical and chemical properties, the application of this process for aromatic and medicinal herbs is progressively attracting the attention of researchers and scientists. Nurhaslina et al. (2022) presented an extensive review related to the effect of drying process on essential oil yield and bioactive compounds of several medicinal herbs. The following were particularly reviewed: antioxidant, aromatic, antimicrobial compounds and color. The authors described the most important drying methods, such as sun drying, shade drying and convective drying, keeping the objective of making the final product as a marketable product. They pointed that temperature and drying time are two important parameters that play essential role on the quality of the final product. They presented the antibacterial activity of several herbal extract bioactive compounds, such as Citronella and Curcuma. Moreover, the authors showed that for leaves freeze drying is the most favorable drying method, with a less degree solar drying is also favorable, however sun drying is the less favorable drying method. The authors focused in their review on giving general

information without going into details with the technical side, such as the design of the dryer. Pise et al. (2024) presented important information about aromatic plants and the classification of herbs, spices and medicinal plants, which was based on its usage, chemical composition or the time life of the product. The authors mentioned the importance of having controlled drying conditions, in particular the temperature of the air, and its effect on the quality of the final product. Ndukwu et al. (2021) discussed the solar drying experience in Africa including Mediterranean countries such as Algeria, Egypt, Tunisia, and Morocco. In this review, the authors focused on the technical side of the process with the description of several solar dryers that has been used. The authors made a classification of the published research papers based on the country and the tested plant. Indeed, in Algeria researchers investigated recently drying of henna, spearmint leaves, colocynth gourd and bay leave. In Morocco, they focused on eucalyptus, rosemary leave, some kind of flowers, citrus leave and other plants. The authors of this review discussed via the published studies the impact of solar drying on essential oils of the plants, as well as the physical appearance and microbial activities. It was clearly stated that drying conditions (i.e. temperature of the heated air, its velocity and humidity) have a direct effect on the quality of the final product. Finally, the authors and based on the published papers, gave an idea about the optimum drying conditions that were used for the different tested plants.

Henna is one of the valuable traditional plants that is regularly consumed in Algeria. Labed et al. (2016) stated that the average production of henna in Algeria in the last decade was around 3000 tons per year. As described by Bennaceur et al. (2021), this plant, and due to its richness with natural components, is widely used in cosmetic and pharmaceutical fields. It is also used to prevent from several skin and eye diseases. The authors stated that the most important component that we can find in henna leave extracts is: Lawsone. Hosseiny et al. (2023) found that the harvesting date and the climate conditions is this period, as well as the genotype are two important factors that influence the concentration of the bioactive compounds and minerals in henna, pointing in particular: lawsone, flavonoid and phenol. It was concluded in this research that the best results were obtained when henna harvesting was performed in periods with low humidity and low rainfalls, around the month of July. In addition to the benefits of henna mentioned by Bennaceur et al. (2021), Semwal et al. (2014) declared that henna is used against fungal pathogens and even in digestive disorder treatments. In this review, the authors discussed the phytochemistry and the different phenolic compounds in particular lawsone, which has been used in several pharmacological industries.

Bennaceur et al. (2021) explored the effect of drying on the lawsone component. It was found that adding ultrasound to the traditional convective drying helped to reduce drying time and consequently the increase of the extraction yield of lawsone. Labed et al. (2016, 2015) studied solar drying of henna using a forced convective air dryer. The study focused on the design and the behavior of the solar dryer by presenting the variation of the drying kinetics and drying curves with the operating conditions (i.e. temperature, humidity and velocity of the heated air), which vary with the weather conditions. Unfortunately, there was no evaluation of the final quality of the product, which is an important part that should be taken into consideration. Hassanain (2010) tested three different solar drying methods: unglazed transpired dryer, direct open sun drying and in a shaded house. Similarly, to the previous studies, the author focused on the effect of the weather conditions of three medicinal plants including henna. It was found that the unglazed transpired dryer gave better results as more oil was extracted from these plants. Boubeghal and Chaker (2018) investigated the drying process of henna in the south region of Algeria. The used dryer was an indirect natural convection dryer, usually the drying time for this kind of dryers is longer than the ones using forced convection. For this case, between 20 and 22 hours were necessary to dry the product. In this study and similarly to the other presented publications, the authors focused on the design of the dryer and the effect of the weather conditions on the variation of the moisture of henna without studying the effect of this process on the physio-chemical properties of the product and its final quality.

The aim of the research is to study the behavior of henna leaves during drying forced convection drying. The evaluation of the performances of this process was determined through the variation of the moisture content of the henna leaves, as well as the efficiency of the solar collector. The second

axe of this study focused on the quality of the final product in terms of assessing the effect of open solar drying and the convective solar drying on bioactivities (total polyphenol content, total flavonoid content, antioxidant activity).

## Material and methods

### Convective solar drier

A solar drier that uses forced convection was utilized for this study. The main components of this dryer are a flat air collector and a drying chamber. The dryer was built with the objective of using inexpensive and available local materials. Indeed, the frame of the flat collector and the drying chamber was wood. The dimensions of the solar collector are:  $1.24\text{m} \times 0.9\text{m}$ . The collector is installed with an inclination of  $31^\circ$  from the horizontal. A sheet of iron is used as an absorber and polystyrene is used as insulator. The external dimensions of the drying chamber are:  $0.94\text{m} \times 0.63\text{m} \times 0.61\text{m}$ , however the internal dimensions are:  $0.48\text{m} \times 0.52\text{m} \times 0.47\text{m}$ . Similarly, to the solar collector, the insulation is insured using polystyrene. The drying system contains a thermostat that adjusts the temperature of the heated air using an auxiliary source of heat in case the temperature of the air coming from the solar collector does not reach the required temperature. Moreover, the velocity of the air is measured using an anemometer and controlled using a suction system and velocity variator. Figure 1 shows a schema of the solar dryer used in this study.

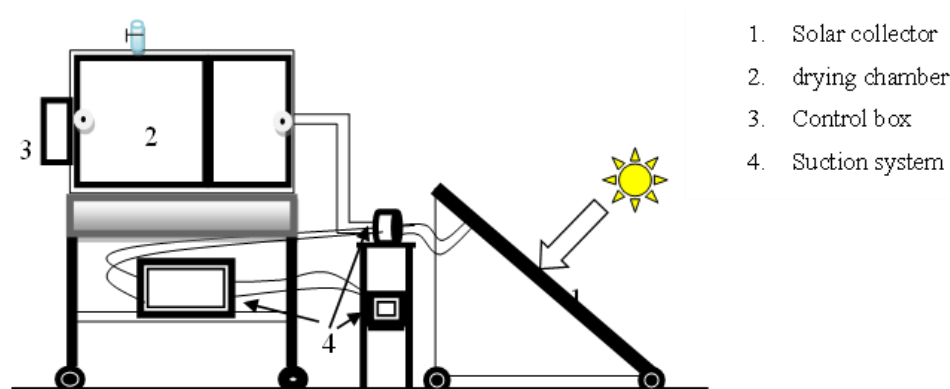
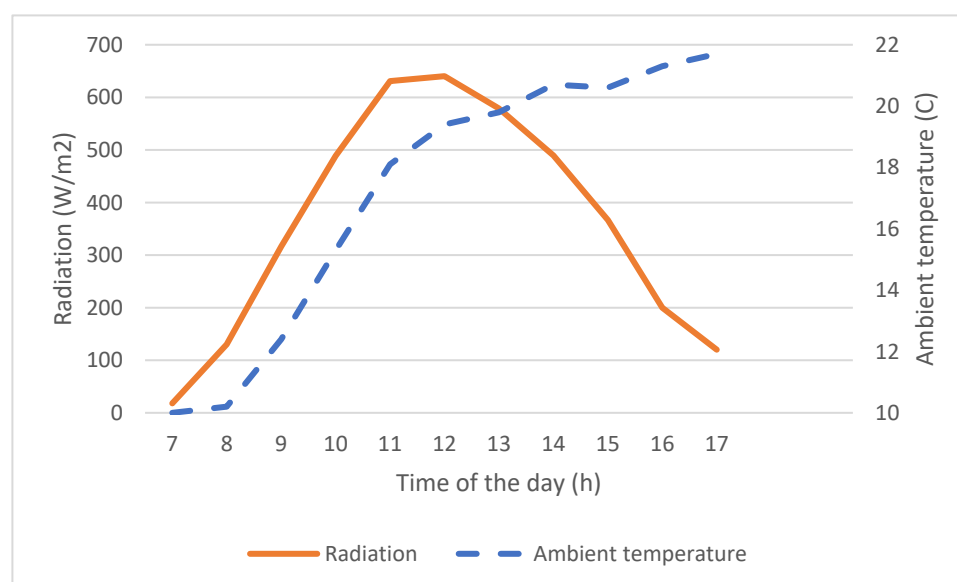


Figure 1. Schema giving the different components of the solar dryer.

The experiment is performed in the month of November. Solar drying is directly affected by the weather conditions, in particular the ambient temperature of the air and the radiation. Figure 2 shows the variation of these two parameters in the region of Béchar, where the experiment is executed.



**Figure 2.** Variation of the weather conditions at the time of the experiment.

Indeed, Figure 2 shows that the solar radiation starts increasing from 7 a.m. and reaches the maximum between 11 a.m. and 1 p.m. then it decreases until reaching low radiations, about 100 W/m<sup>2</sup>, at 5 p.m. As mentioned by (Bennamoun and Li, 2018), solar radiation has a direct effect on the solar collector and its outlet temperature, this effect will be discussed later in this study. Moreover, Figure 2 shows the variation of the ambient temperature and its continuous increase from 7 a.m. to 5 p.m. this increase is less important after 1 p.m. The ambient temperature changed from 10°C to 22°C.

#### *Preparation of the samples*

*Lawsonia inermis* leaves were collected in November 2021 from cultivated farm from area of Béchar, located in South-West of Algerian desert (Latitude: 31°37'00" North Longitude: 2°13'00" West).

The process began by cleaning the leaves from impurities. After that, the leaves have been spread in a thin layer and exposed to two different types of drying open solar drying (with no control of the weather conditions, in particular the temperature) and convective solar drying under constant temperature of 50°C and air velocity of about 3.4 m/s. After the drying process, the samples were delivered to Mentouri University, Constantine to begin the study of the bioactive properties and its extraction.

The initial moisture of the leaves was determined using the standard gravimetric method at 105 °C for 24 h, the measurements were taken until there was no changes in the mass of the samples. Each measurement was performed in triplicate (Bennamoun et al., 2020).

#### *Preparation of the leave extracts*

The plant extracts were prepared according to method described by Boutaghane et al. (2013). In which the leaves of *Lawsonia inermis* from each sample (250 g) were grinded individually to semi-powder. Each dried sample was macerated in EtOH-H<sub>2</sub>O (80% Ethanol, 20% Distillated water). This extract was concentrated to dryness for 3 days at room temperature. The filtrate was separated from the solid residue by filtering through hydrophilic cotton.

Resulting residue underwent a second extraction using a varied volume of the same solution to enhance the extract yield. Recover the two filtrates by evaporating the solvent using a rotary evaporator set at 40°C. Subsequently, keep the resulting dry extracts stored in the host until they are finally dried and ready to be used. The recovered extracts are listed below: Fresh leaves (FL), Open sun drying (OSD); Convective solar drying (CSD)

The extraction yield was calculated according to equation below:

$$Y\% (Yield) = \frac{P1 - P2}{P3} \times 100 \quad (1)$$

P1: the weight of the dishes after drying which contains the crude extract

P2: the weight of empty dishes in grams

P3: the weight of the dried extracts

#### *Determination of the bioactivity*

##### Total Phenol Content (TPC)

The total phenol content of the extracts was assessed according to Muller et al. (2010) method, using a spectrophotometer and Folin Ciocalteu reagent. First, 20 µl of the samples were added to 100 µl of Folin Ciocalteu reagent diluted to (1:10) and 75 µl of sodium carbonate (Na<sub>2</sub>CO<sub>3</sub>) (7.5%). The resulting mixtures were incubated in dark condition at room temperature for 2h before the absorption was measured at 765 nm.

To calculate sample concentration, the gallic acid ( $y = 0.0034x + 0.22$ ,  $R^2 = 0.9624$ ) calibration curve was used as the standard. The values are expressed in grams of gallic acid equivalent /100 g extract.



### Total Flavonoid Content (TFC)

The estimation of total flavonoid content (TFC) is based on the formation of a complex between  $Al^{+3}$  and flavonoids **extract** (Topçu et al., 2007). A volume of 50  $\mu$ L of each diluted extract was added to 130  $\mu$ L of methanol, 10  $\mu$ L of potassium acetate, and 10  $\mu$ L of aluminum nitrate. The reaction mixture was incubated for 40 min in the dark at room temperature. Before the absorbance of the reaction mixture was read at 415 nm. Quercetin is used as a calibration curve ( $y = 0.0026x + 0.1824$ ,  $R^2 = 0.9738$ ). Flavonoid content was expressed as  $\mu$ g of Quercetin equivalent (QE)/mg of extract.

### Antioxidant activities

The extracts were assessed for their antioxidant activity by using three different methodologies which are: Free radical scavenging (DPPH), Radical cation trapping (ABTS) and reducing power (FRAP).

#### DPPH radical scavenging activity

One of the most popular techniques to evaluate the antioxidant capacity is the method employing 2,2'-diphenyl-1-picrylhydrazyl DPPH. The free radical scavenging activity was evaluated spectrophotometrically using the method described by Bendjabeur et al. (2018) with slight modification. Briefly, 40  $\mu$ L of sample solution was mixed with 160  $\mu$ L of DPPH solution. The reaction mixture was incubated for 30 min in the dark at room temperature. The absorbance was measured at 517 nm. Trolox and ascorbic acid were utilized as standards. The radical scavenging activity was calculated using as follows:

$$I\% = \frac{\text{Control Absorbance} - \text{Sample Absorbance}}{\text{Control Absorbance}} \times 100 \quad (2)$$

The results are expressed as  $IC_{50}$  value ( $\mu$ g/ml).

#### ABTS radical scavenging assay

The ABTS scavenging activity was measured using the method outlined in the protocol described by Re et al. 1999 with slight modifications. Different extract concentrations were mixed with ABTS solution: mix 160  $\mu$ L (ABTS+) with 40  $\mu$ L (extract), incubating 10 min in the dark, and measure the absorbance at 734 nm. The results are presented with different  $IC_{50}$  concentrations. The radical scavenging activity was calculated using Equation (2)

#### Ferric reducing ability assay

Reducing power activity is measured using the Oyaizu (1986) explained method, with slight modification. Substances with the ability to undergo reduction react with potassium ferricyanide ( $Fe^{3+}$ ) to create potassium ferrocyanide ( $Fe^{2+}$ ). This compound then reacts with ferric chloride, ultimately forming a cyan-blue ferrous complex with maximum absorption at 700 nm which is measured using microplate spectrophotometer, and results are expressed by  $A_{0.50}$ , which corresponds to the concentration producing 0.50 absorbance. The absorbance was spectrophotometrically measured at 700nm. Butylatedhydroxytoluene (BHT) was used as a positive reference compound. The results were given as  $A_{0.50}$ , which corresponds to the concentration producing 0.50 absorbance. The same Equation (2) was also used for the calculation.

### Statistical analysis

Results are reported as mean value  $\pm$  SD of three measurements ( $p < 0.05$ ); the  $IC_{50}$  and  $A_{0.50}$  values were calculated by linear regression analysis.

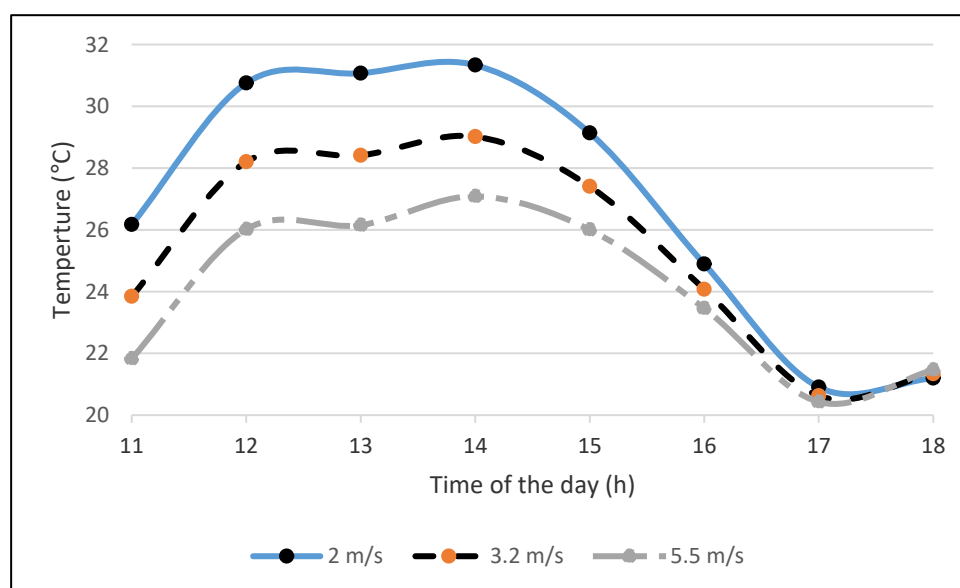
### Results and discussion

As it was confirmed in the introduction, most of the published studies dealing with solar drying of herbs and medicinal plants do not take in consideration the quality of the final product, which is crucial to determine the appropriate drying conditions. Accordingly, two main axes are presented in

this study: results related to the solar dryer and its design and then results related to the quality of the final product in terms of the antioxidant properties of the final product.

### Solar drying results

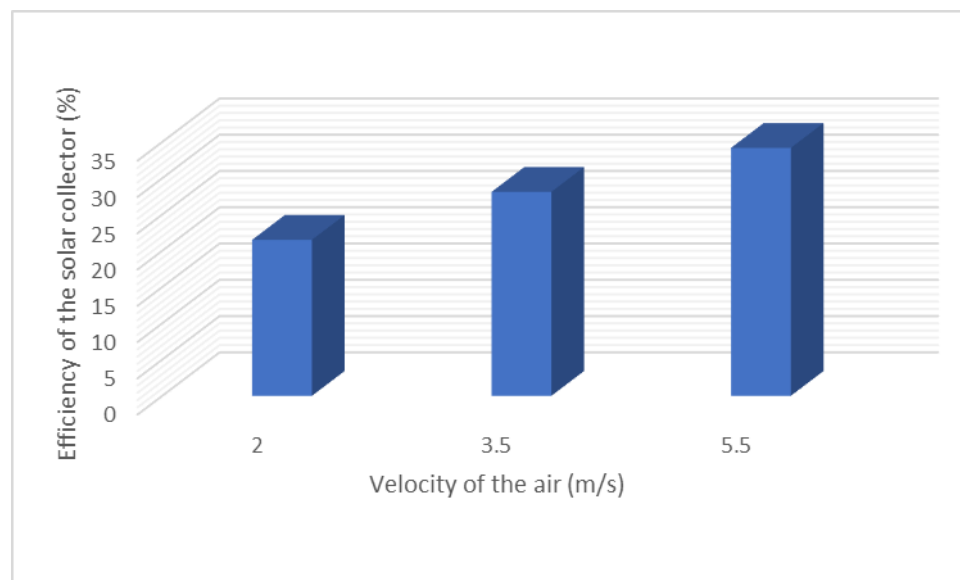
Figure 3 represents the effect of the air velocity on the outlet temperature of the solar collector. The general shape of the three curves of the figure is quite similar to the radiation presented in Figure 2, as the temperature of the collector increases with the increase of the radiation and decreases with its decrease. This allows saying that radiation could be the most important parameter that has influence on the temperature of the collector. It is important to note that the maximum values of the exit temperature of the solar collector was obtained between 12 p.m. and 14 p.m., compared to the maximum radiations shown in Figure 2 (between 11 a.m. and 1 p.m.), it can be deduced that there is a reaction time of 1 hour. Moreover, it was found that increasing the velocity of the air led to the decrease of the exit temperature of the solar collector, which means that the air velocity plays a role of dissipation of the thermal energy (temperature of the collector). These results are comparable with simulation and modeling results obtained by Bennamoun and Belhamri (2006), where they find the same effect of the air velocity on the outlet temperature of a solar collector. The authors (Bennamoun and Belhamri, 2006) have also find that there is a response time of about 1 hour between, as similar to our experimental study, the maximum value of the radiation was obtained around 12 p.m. and, however the maximum value of the outlet temperature was reached around 1 p.m. Accordingly, increasing the residence time of the ambient air using fins (Kharrafi et al., 2023). Bennamoun and Belhamri (2003) presented a detailed simulation and mathematical modeling of such convective solar drying system. The mathematical modeling was performed based on heat and mass transfer applied for each component of the drying system which are mainly: the solar collector and the drying chamber. The effect of the surface of the solar collector was also studied. The results confirm the effect of the radiation on the outlet temperature of the collector, as mentioned previously. Bennamoun et Belhamri (2003) found that increasing the surface leads to the increase of the outlet temperature. It is important to note that in this simulation, the study started at 5 a.m. and that the heat used from this time to 8 a.m. was used to heat-up the collector. Indeed, it was faster to heat-up a surface of 1 m<sup>2</sup> than 4 m<sup>2</sup>, which means that in this lap of time the outlet temperature increased with the surface decrease then the tendence changes and the outlet temperature increases with the increase of the collector surface.



**Figure 3.** Effect of the controlled air velocity on the exit temperature from the solar air collector.

Figure 3 shows the variation of the average solar collector thermal efficiency with the velocity of the air entering the solar collector controlled using the suction system and velocity variator. We refer

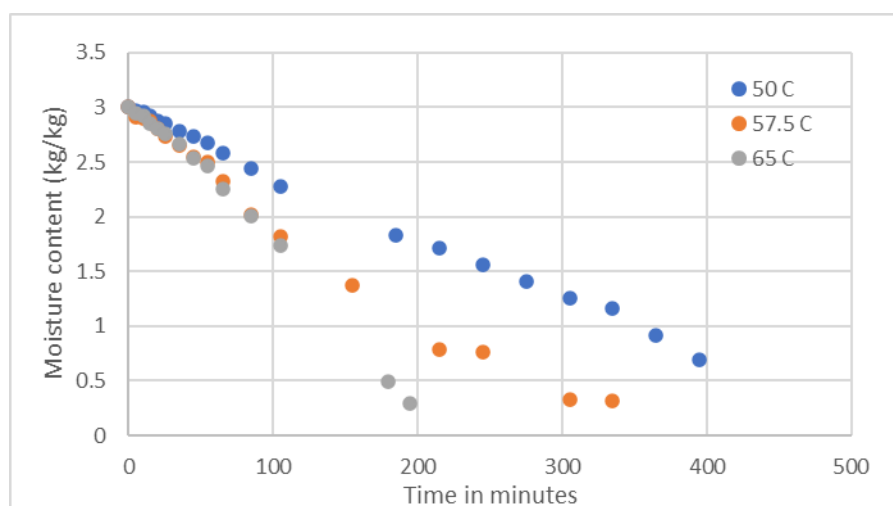
the reader to the study published by Amer et al. (2010). The instant thermal efficiency of the solar collector is also directly affected by the solar radiation with an increase with its increase and vis-versa. Figure 3 shows that even increasing the air velocity leads to the decrease of the outlet temperature, the average thermal efficiency of the solar collector increases with the air velocity increase, as it is function of the flow rate of the air and the solar radiation. This observation is in agreement with the experimental results obtained by Tarminzi et al. (2021).



**Figure 3.** Effect of the air velocity on the thermal efficiency of the solar collector.

Figure 4 shows the variation of the moisture content with time and the effect of the drying air temperature for henna leaves. The curves of this figure show that the moisture content of the leaves decreases with time until reaching the equilibrium moisture content. The drying time varies from 200 minutes to 400 minutes depending on the applied temperature. It is important to note that and as shown in Figure 3, the maximum outlet temperature from the solar collector, entering the drying chamber, did not reach the required drying temperature, which was 50 °C, 57.5 °C or 65 °C. This situation means that the auxiliary source of heat was working during the whole process, giving a similar situation of drying at constant drying temperature. Figure 4 shows the importance of the temperature parameter and its important influence of the drying time. Similar modeling and simulation results were obtained by Bennamoun et al. (2003 and 2006), as it was found that increasing the drying temperature allows reducing drying time. Furthermore, drying without an auxiliary source of energy or controlled temperature may show a completely different tendency of the drying curves with no regularities in the curve. As an example of this behavior, the study published by Babar et al. (2020), which was applied for agricultural products. Moreover, it was found in this study that the acquired energy or heat coming from the solar collector will serve in decreasing the moisture content of the product, but also in increasing its temperature until reaching the temperature of the drying air at the end of the process. Similar results were also obtained by Bennamoun and Belhamri (2003) at the completion of their modeling and simulation work applied for agriculture product.





**Figure 4.** Effect of the drying temperature on the drying curve of henna leaves.

### Quality of the final product

#### Total bioactive content (TPC and TFC)

In the present study, the yield of extraction, TPC and TFC were influenced by the method of drying of *Lawsonia inermis* samples (Table 1). The yield of *Lawsonia inermis* ethanolic extracts showed that the highest yield was obtained by the CSD sample.

The highest amounts of TPC ( $867.82 \pm 6.39 \mu\text{g GAE/mg E}$ ) and TFC ( $407.47 \pm 33.53 \mu\text{g QE/mg E}$ ) were recorded with CSD. Whereas the lowest amounts were found in FL sample.

Phenolic compounds are generally polar and ethanol solvent appears to play a significant role in their extraction. The solvent is available to penetrate the plant matrix, allowing bioactive compounds to dissolve and, in turn, increasing the recovery yield (Naseem et al., 2022).

Furthermore, ethanol was used as a solvent because it has the advantage of being a non-polluting, cheaper and non-toxic polar solvent (polarity of 5.2) compared to other solvents such as methanol (Jokić et al., 2010). Also, it is known for its greater efficiency in extracting maximum quantities of bioactive components (Bourgou et al., 2016).

Previous investigations on Tunisian *L. inermis* revealed that the butanolic fraction of leaves has strong antioxidant activities, and these activities were attributed to phenolic glycosides including 1,2,4 trihydroxynaphthalene-1-O- $\beta$ -d-glucopyranoside (Hsouna et al., 2011).

Phenolic and flavonoids are bioactive compounds that shown considerable antioxidant and protective characteristics (Naseem et al., 2020). In the current study, the phenolic levels of *Lawsonia inermis* extracts ranged from  $429.77 \pm 22.76$  to  $867.82 \pm 6.39 \mu\text{g GAE/mg E}$ . Such results were higher than those obtained by Elansary et al., 2020, who studied methanolic extracts of *Lawsonia inermis* from natural populations growing in Northern of Saudi Arabia and found phenolic content of  $81 \pm 13.2 \text{ GAE/g E}$ .

Regarding the flavonoid contents, our findings were higher than the Tunisian ( $20.5 \pm 1.4 \text{ mg QE / g MS}$ ) (Enneb et al., 2015).

Among the drying methods the convective solar drying allows the preservation or even the increase of phenolic and flavonoids compounds.

The drying operation conditions applied shows that high temperatures  $55^\circ\text{C}$  are conducive to rapid heating, which inactivates oxidative enzymes and contributes to better preservation of polyphenols and flavonoids (Casto et al., 2023). On the other hand, the maximum TPC was obtained with mid-range temperatures and air velocities.

**Table 1.** Extraction yield, TPC and TFC of *Lawsonia inermis* ethanolic extracts.

Sample	Extraction yield %	TPC	TFC
		(µg GAE/mg E)	(µg QE/mg E)
FL	11.64	429.77 ± 22.76	171.56 ±8.88
OSD	13.92	617.72 ± 5.56	307.46 ±15.39
CSD	28.6	867.82 ± 6.39	407.47 ±33.53

Antioxidants activities

Antioxidant activity cannot be unequivocally determined with only one method, as plant raw materials have complex chemical composition, and the components may interact with each other Wojdylo et al. (2020). For these reasons, the antioxidant capacity of fresh and dried *Lawsonia inermis* was determined using three assays based on different mechanisms. Scavenging activity, ABTS radical action, and ferric reducing capacity. The findings from these tests are shown in Table 2. The antioxidant activity of different extracts was evaluated using the  $A_{0.5}$  and  $IC_{50}$  indices. The results showing that all sample extracts displayed potent antioxidant activity with regards to all tested methods.

Table 2 shows significant effects of different drying on antioxidant activities on dried extracts of *Lawsonia inermis*. For DPPH and ABTS radical scavenging activities, OSD had the highest to trap DPPH with  $IC_{50}$  value ( $35.86\pm0.42$  µg/ml). Whereas CSD exerted the most potent scavenging activity for ABTS with  $IC_{50}$  values of  $9.54\pm0.51$  µg/ml. The strong antiradical potential of OSD and CSD could be linked to their TPC an TFC.

Our results were coherent with those of literature that indicated different ranges of DPPH, ABTS radicals scavenging effects. In DPPH scavenging activity, Ling et al. (2013) indicated  $IC_{50}$  value  $1.26\pm5.55$  mg/ml of ethanolic extract from Malaysian plant.

Again, the strongest activity showed by ABTS method was observed in CSD sample ( $9.54\pm0.51$  µg/ml) but lower activity was observed in FL. the study by Al-Snafi (2019) on the ethanolic extract of *Lawsonia inermis* Linn showed a very low  $IC_{50}$  of  $6.9\pm0.1$  mg/l compared to our results.

In the other hand, the antioxidant capacity determined by FRAP assay increased in OSD and CSD depending on the drying method.

Interestingly, our results showed that antioxidants in *Lawsonia inermis* extracts were well preserved during convective solar drying at 50 °C and Open solar drying, thus indicating that mild heat treatment processes were promising drying methods to obtain antioxidants from *Lawsonia inermis* plant.

As reported in the literature (Haminiuk et al., 2012; Oancea, 2021), phenolics are the major contributors to the antioxidant potential of plants. A strong correlation between TPC and antioxidant capacities was reported (Sirichai et al., 2022), concurring with our results and indicating that TPC and antioxidant activities of open solar drying and convective solar drying followed the same trend. In various scientific findings, milder heat conditions during processing have demonstrated mitigative effects on the structural degradation of bioactive components. Interestingly, our results showed that antioxidants in henna plant were well preserved during hot-air drying at 50 °C, thus indicating that mild heat treatment processes were promising drying methods to obtain antioxidants from henna plant.

**Table 2.** Antioxidant activity of *Lawsonia inermis* drying. Values are the mean ± standard deviation (n = 3).

Sample	DPPH	ABTS	Ferric reducing
	$IC_{50}$ (µg/ml)	$IC_{50}$ (µg/ml)	$A_{0.5}$ (µg/ml)
FL	56.92±0.12	10.92±0.12	183.64±6.70

OSD	35.86±0.42	10.75±1.43	83.47±3.58
CSD	43.21±0.87	9.54±0.51	93.07±2.43
BHT	12.99±0.41	1.29±0.30	
Ascorbic acid	4.39±0.01	3.04±0.05	3.62±0.29

Conclusions

Herbal medicine is a true legacy of human knowledge and use. In recent years, their importance in the field of public health has expanded dramatically due to the therapies they provide.

The two primary causes of this growing interest are the desire for a better pharmaceutical with a gentler therapy that has no side effects and the fact that medicinal plants are a limitless source of bioactive natural chemicals and molecules.

Our research aims to optimize the effects of various drying techniques, such as open solar drying, and convective solar drying, in relation to fresh plants.

Solar drying technology offers an alternative that can process the plants, vegetables and fruits in clean, hygienic, and sanitary conditions to national and international standard.

Convective drying was applied to obtain dehydrated henna. The obtained results in this study indicates that the use of convective solar drying average temperature of 50°C and air velocities and a thinner thickness are related to low drying times and are favourable conditions to avoid higher antioxidant capacity determined by the DPPH, ABTS and FRAP methods.

CSD is one of the drying techniques that preserves or even improves antioxidant activity extracts from most studied species, as indicated by a decrease in IC<sub>50</sub>.

When compared to fresh extract, the spectrophotometric analysis of polyphenols and flavonoids showed the abundance of henna in various drying modes.

Reduced IC<sub>50</sub> values indicate that even higher antioxidant activity tested extracts can be preserved by solar drying and convective solar drying. We may infer from the various outcomes that the processes solar drying appear to be the most preserving of the phenolic components and have the highest antioxidant activity of all the extracts that were studied.

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