

# Capacitive Imaging for Skin Characterizations and Solvent Penetration Measurements

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## Abstract

Capacitive contact imaging has shown potential in measuring skin properties including hydration, micro relief analysis, as well as solvent penetration measurements. Through calibration we can also measure the absolute permittivity of the skin, and from absolute permittivity we then work out the absolute solvent content in skin. In this paper, we present our latest study of capacitive contact imaging for skin characterizations and vivo skin solvent penetration. We will show

how to use capacitive occlusion measurements to characterize skin damages, and  
how to calculate the absolute water content and solvent content in skin.

## **Keywords**

Capacitive imaging, skin hydration, solvent penetration, water content, solvent  
content.

## **1. Introduction**

The Epsilon permittivity imaging system which is based on Fujitsu fingerprint sensors (Fujitsu Ltd), has shown potentials in skin hydration imaging, skin characterization analysis, and skin solvent penetration measurements (Leveque et al., 2003; Batisse et al., 2006; Xiao et al., 2007; Bevilacqua et al., 2008; Singh et al., 2008). It is based on capacitance measurement principles, and the measurement results depend on the sample's dielectric permittivity, because of calibration, we can get the absolute dielectric permittivity so that we can work out the absolute water hydration. Our latest studies showed that Epsilon permittivity imaging system can be used in many solvents, due to their high dielectric permittivity. This makes the technique very useful for in-vivo trans-dermal drug delivery studies (Xiao et al., 2012a; Xiao et al., 2012b).

## 2. Materials and Methods

### 2.1 Epsilon permittivity imaging system

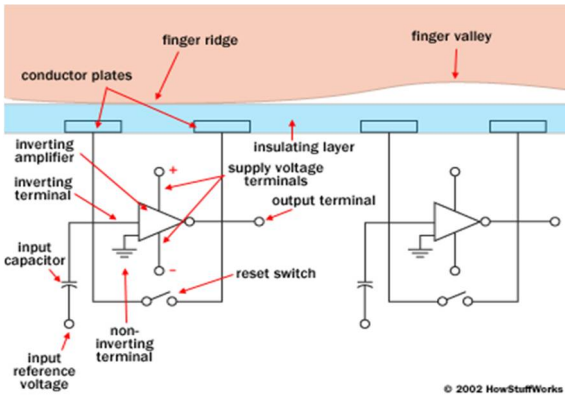
The Epsilon permittivity imaging system, based on Fujitsu fingerprint sensors (Fujitsu Ltd), has 256x300 pixels with 50 $\mu$ m spatial resolution, as shown in **Fig. 1** (Xiao et al., 2012b). Each pixel is equivalent of a capacitive sensor, which measures the dielectric constant or permittivity of the sample. Epsilon has an 8-bit grey-scale capacitance resolution per pixel (0 - 255).



A) Epsilon and the in-vivo stand



B) Epsilon and the in-vitro stand



C) Capacitive fingerprint sensor measurement principle [8]

**Fig. 1.** Epsilon Permittivity Imaging System: A) Epsilon and the *in-vivo* stand, B) Epsilon and the *in-vitro* stand, and C) its measurement principle.

2.2 Epsilon calibration

The Epsilon differs from other similar systems in its calibrated, linear response to near-surface dielectric permittivity, see **Fig. 2**. The linear response is important because hydration is linearly related to permittivity. The calibration ensures consistency from instrument to instrument and from time to time. With calibrated Epsilon imaging systems we can measure the absolute dielectric permittivity of the material. A RGB colour scheme is used to represent the dielectric constant values, the brighter the colour, the higher the dielectric constant, and darker the colour the lower the dielectric constant.

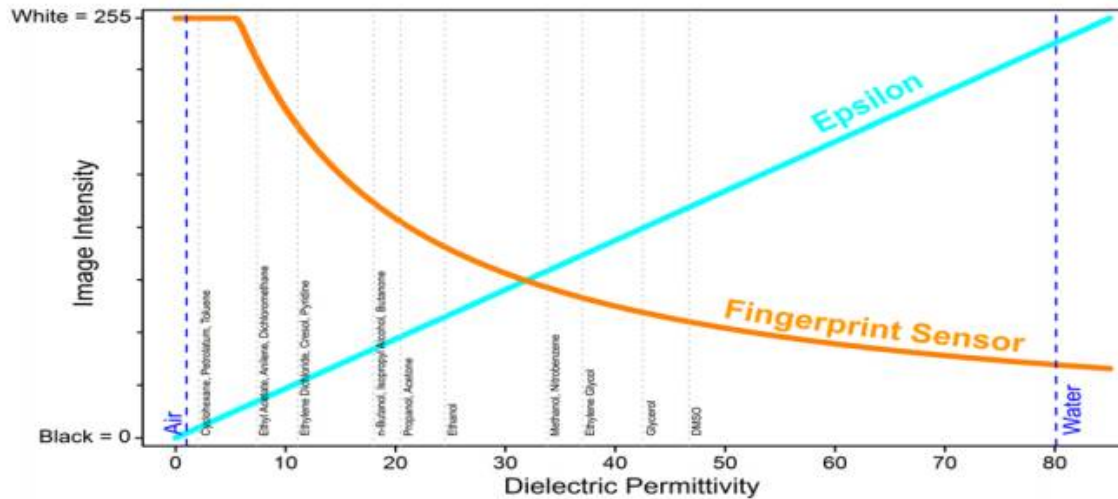


Fig. 2. The Epsilon Calibration curve and linear response to dielectric permittivity

### 2.3 Skin Hydration and Solvent Concentration Calculations

If we assume measured skin dielectric constant value ( $\epsilon_m$ ) is linearly dependent on that of dry skin ( $\epsilon_{dry}$ ) and water ( $\epsilon_{water}$ ), i.e.

$$\epsilon_m = \epsilon_{dry} \times \left(1 - \frac{H}{100}\right) + \epsilon_{water} \times \frac{H}{100} \quad (1)$$

where H is water concentration in skin as percentage of volume ratio, then we can work out the water content using following equation [6]

$$H = \frac{\epsilon_m - \epsilon_{dry}}{\epsilon_{water} - \epsilon_{dry}} \times 100 \quad (2)$$

For the solvent penetration through skin, measured dielectric constant value ( $\epsilon_m$ ) is a combination of pure solvent ( $\epsilon_{sol}$ ) and skin ( $\epsilon_{skin}$ ), similar to Eq.2, we also can calculate the solvent concentration (C, volume ratio percentage),

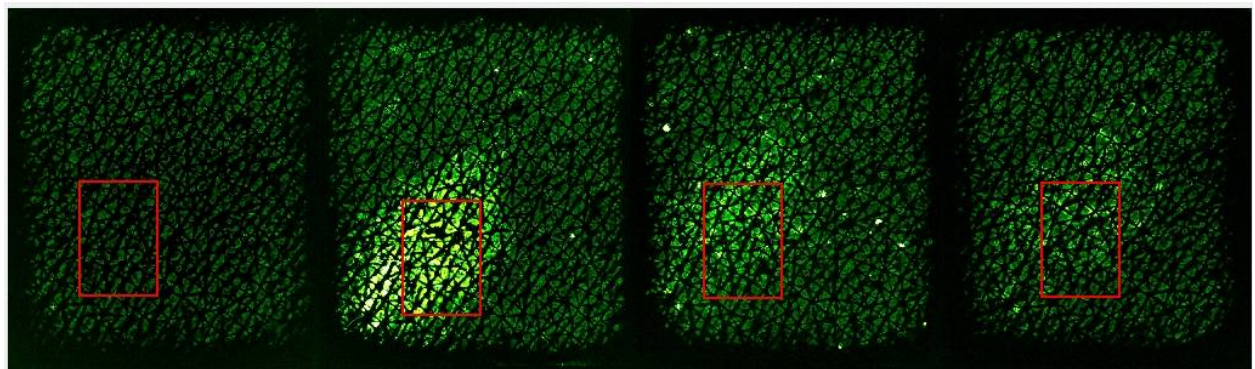
$$C = \frac{\epsilon_m - \epsilon_{skin}}{\epsilon_{sol} - \epsilon_{skin}} \times 100 \quad (3)$$

80 In the paper, the dielectric constants we used for water is  $\epsilon_{\text{water}} = 80.1$ , for dry skin  
 81 is  $\epsilon_{\text{dry}} = 1$ , for pure alcohol is  $\epsilon_{\text{sol}} = 24.3$ , for pure ethylene glycol is  $\epsilon_{\text{sol}} = 55.81$ , and  
 82 for pure glycerol is  $\epsilon_{\text{sol}} = 38.91$ .

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#### 84 2.4 Re-positioning Using Normalized Cross-Correlation Algorithm

85 Re-positioning is very important in image processing. In many occasions, we would  
 86 prefer to compare the results from exactly the same skin site. To select the same  
 87 skin site manually is cumbersome and inaccurate. We have developed an efficient  
 88 repositioning technique based on normalized cross-correlation algorithm. As  
 89 illustrated in **Fig.3.**, users can select the region of interests (RoI) in the first image,  
 90 the re-positioning technique will automatically find the exactly same position in the  
 91 subsequent images.



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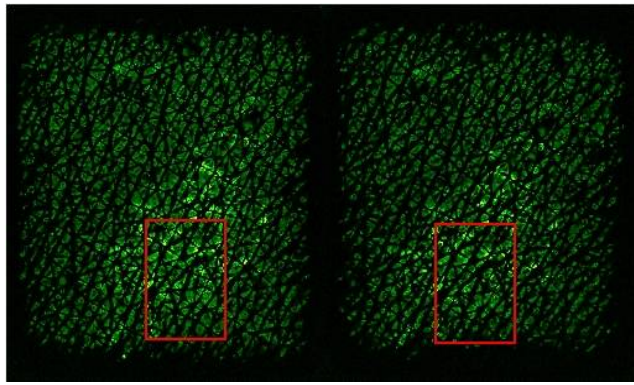
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Normal Skin

After 0 min

20min

30 min



40min

50min

**Fig.3.** Capacitive contact images of skin before and after Glycerol application. Red squares show the repositioning of the RoI in Normal Skin image (before application) and in the consequent images (after application).

If we use matrix  $\mathbf{R}$  to represent the region of interests, and matrix  $\mathbf{T}$  to represent the target images, we first normalized the  $\mathbf{R}$  using following formula,

$$R_N = \frac{R - \min}{\max - \min} \times 255 \quad (4)$$

Where **max** and **min** are the maximum value and minimum value of  $\mathbf{R}$ , and  $\mathbf{R}_N$  is the normalized  $\mathbf{R}$ . Similarly we can also normalize  $\mathbf{T}$  to get  $\mathbf{T}_N$ . Then we can calculate the cross-correlation of  $\mathbf{R}_N$  and  $\mathbf{T}_N$  as,

$$Corr = \sum_i \left( \frac{\sum_j (R_N(j) - RM) \times (T_N(j-i) - TM)}{\sqrt{\sum_j (R_N(j) - RM)^2} \times \sqrt{\sum_j (T_N(j-1) - TM)^2}} \right) \quad (5)$$

Where **i** and **j** are the matrix indexes, **RM** is the mean of  $\mathbf{R}_N$  and **TM** is the mean of  $\mathbf{T}_N$ . From calculation result **Corr**, we can find out at which position that two matrix

are best correlated, and that will be the position of the original region of interests **R** in target image **T**.

## 2.5 Experimental Procedures

All the measurements were taken under normal ambient laboratory conditions, of  $21 \pm 1$  °C and  $45 \pm 5$  % RH. The volar forearm skin sites from healthy volunteers (aged 20 – 40) were chosen for this study. The volunteers were acclimatized in the laboratory for 20 min prior to the experiments.

For skin characterization, three types of skin damages were studied: intensive washes, tape stripping and sodium lauryl sulphate (SLS) irritation. For intensive washes, room temperature running water and washing-up liquid were used to rub the skin site gently for 3 min with a finger. After wash, the skin site was carefully patted dry with a tissue before the measurements. Tape stripping was performed 20 times per site by the use of standard stripping tape. SLS irritation was achieved by applying 2% SLS water solution (v/v, volume ratio) on skin. Capacitive contact imaging measurements were performed both before and after. The measurements were done by using capacitive imaging sensor to occlude the skin test sites for a period of one minute, during which skin capacitance images were recorded



continuously at the rate of one image per second. The average water contents of the images were then calculated at different times during occlusion using **Eq.2**.

For skin solvent penetration measurements, three solvents were studied, pure ethanol, pure ethylene glycol, and pure glycerol. A small quantity of solvent, held in a plastic well, was applied on skin site for 5 minutes, the skin site was then carefully wiped clean to make sure no solvent residue left on skin surface. The capacitive imaging measurements were performed both before and after the application. Again, the measurements were done by using capacitive imaging sensor to occlude the skin test sites for a period of one minute, during which skin capacitance images were recorded continuously. The solvent concentrations in skin were calculated by using **Eq.4**, and the re-positioning technique was used to make sure the calculations are done on exactly the same skin site.

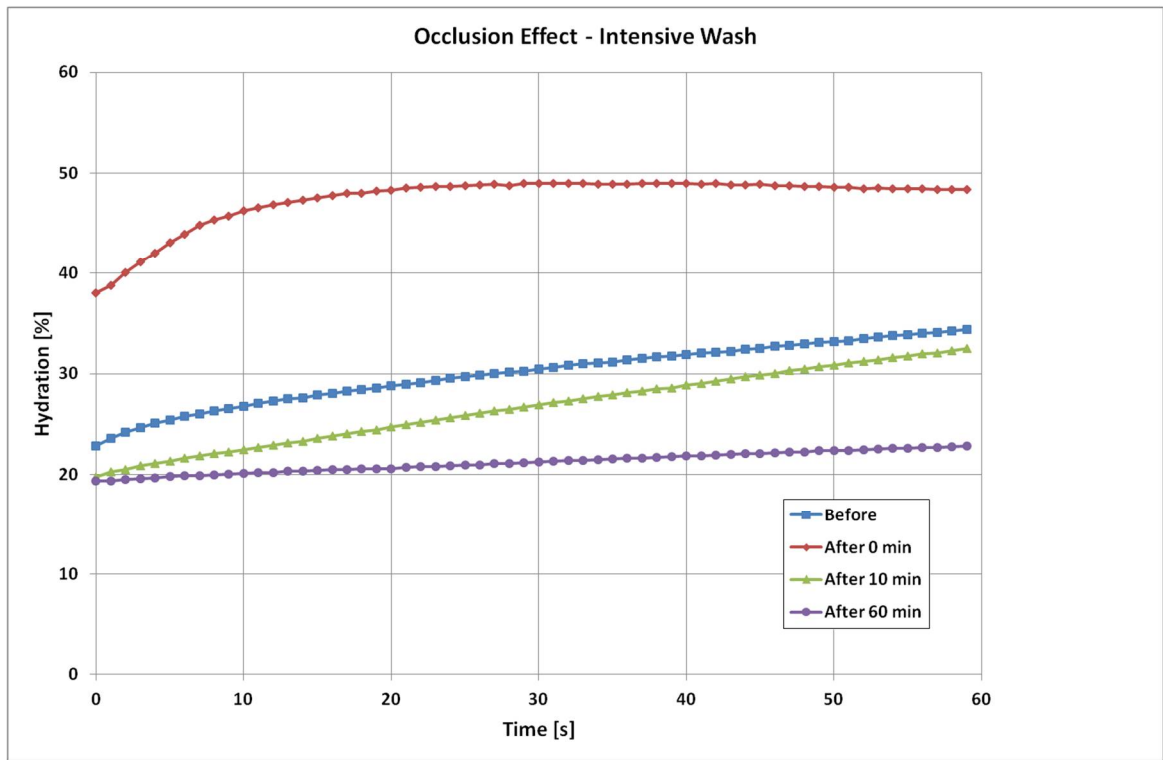
## **2. Results and Discussions**

### **2.1 Skin Damage Characterizations**

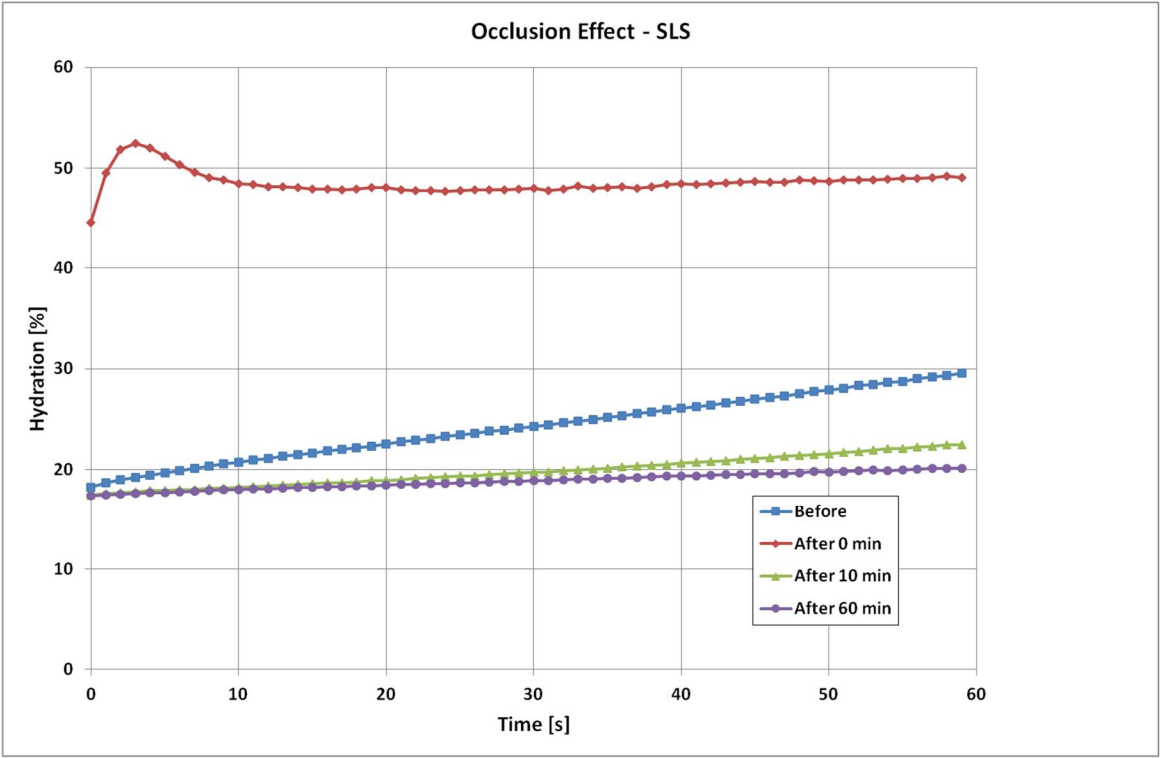
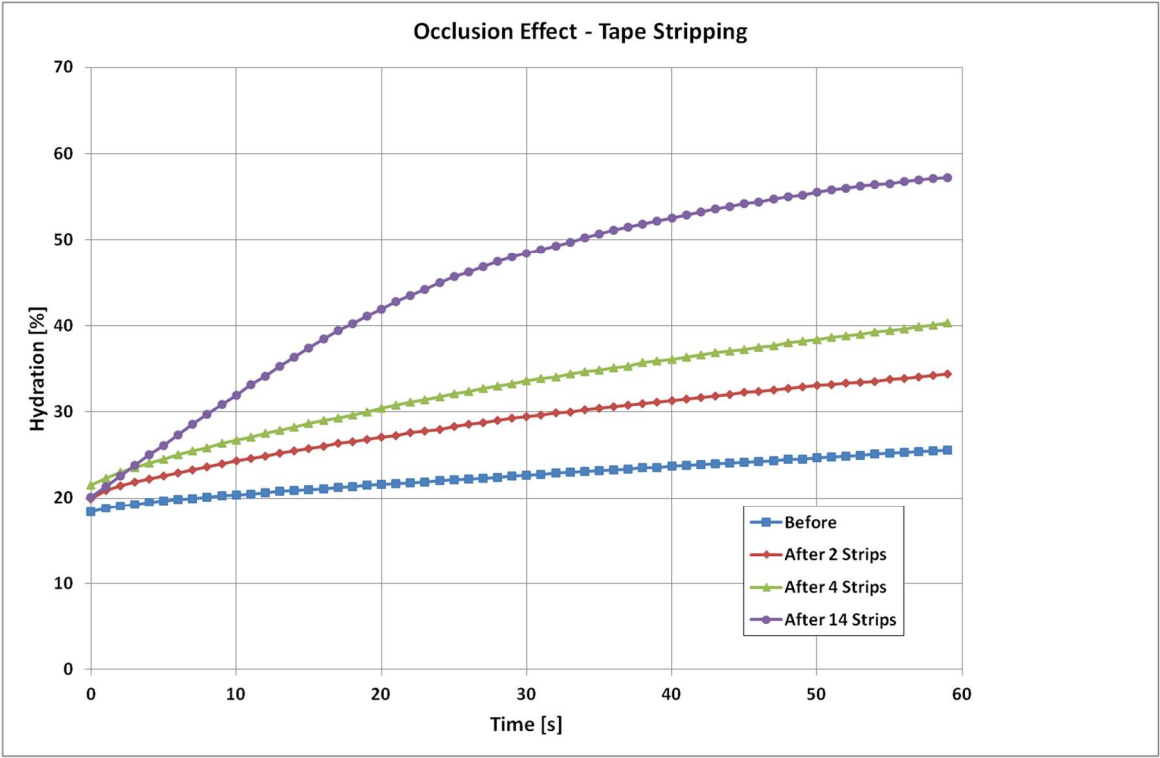
**Fig. 4** shows the time dependent skin hydration occlusion curves before and after the intensive wash, tape stripping and SLS irritation. The results show that the levels and the shapes of occlusion curves changed after the skin damage, from the levels

and the shapes we can understand how badly the skin is damaged, and how long does it take to recover. It also is interesting to point out that for different types of skin damages, the time dependent occlusion curves are also different. This suggests that it is possible to use these time dependent occlusion curves to characterize skin, i.e. not only to show how badly the skin is damaged, but also to show what types of damages.

(A)



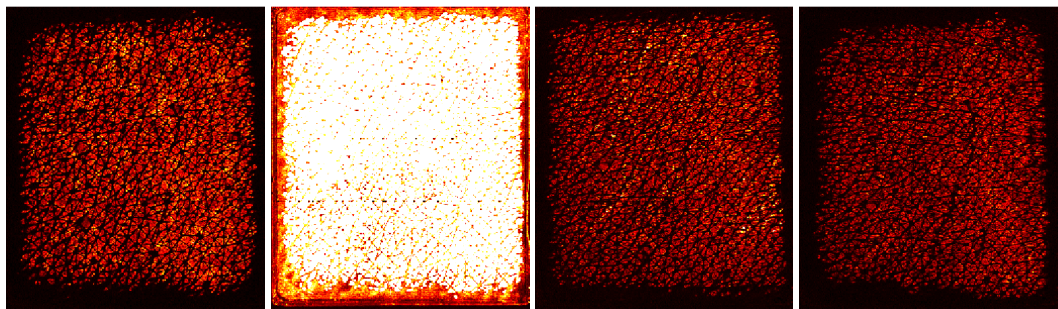
(B)



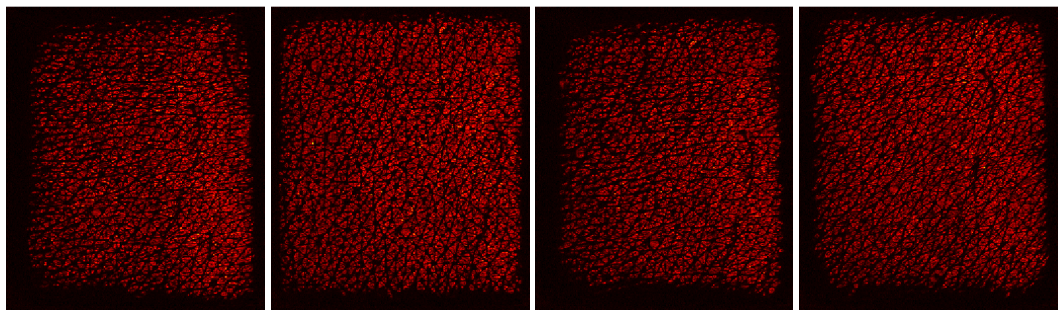
**Fig.4.** The time dependent skin hydration occlusion curves before and after the intensive wash (A), tape stripping (B) and SLS irritation (C).

**Fig. 5** shows corresponding capacitive contact images before and after intensive washes, tape stripping and SLS irritation measurements. The skin images are generally getting brighter after damage, which indicates higher water content in skin. In both intensive washes and SLS irritation, the darker recovery skin images indicate there is a drying effect after the damage.

**(A)**



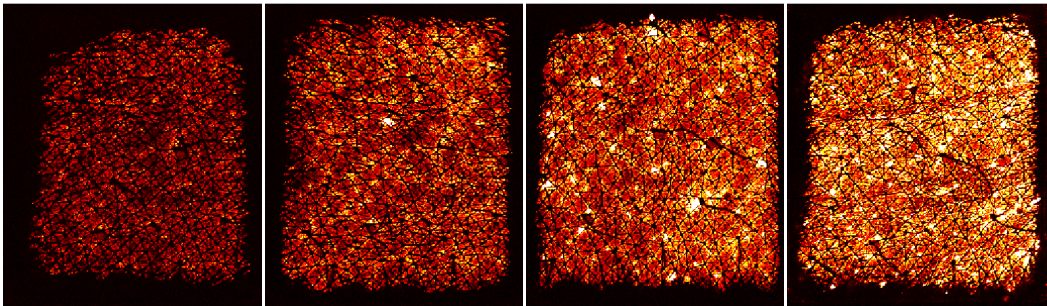
Before                      After (Wash)                      10 min                      20 min



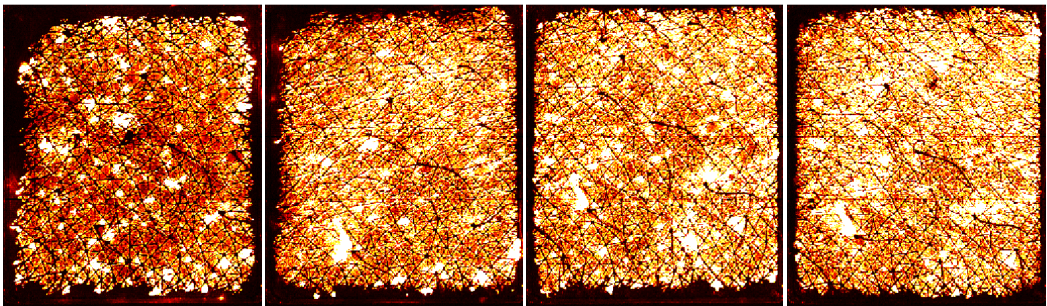
30 min                      40 min                      50 min                      60min

**(B)**



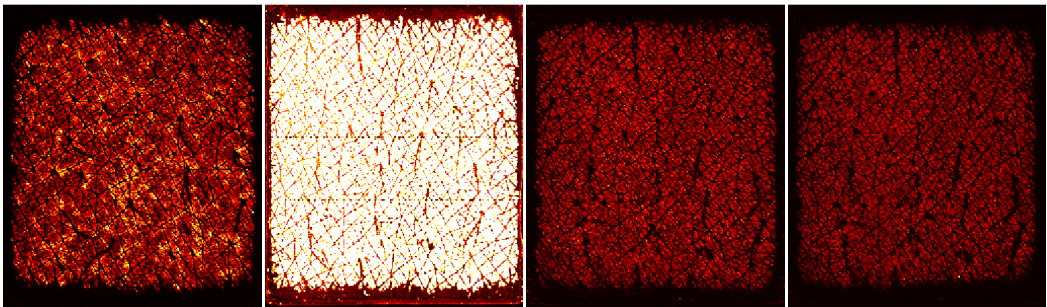


Before                      After (2strips)                      4 strips                      6 strips

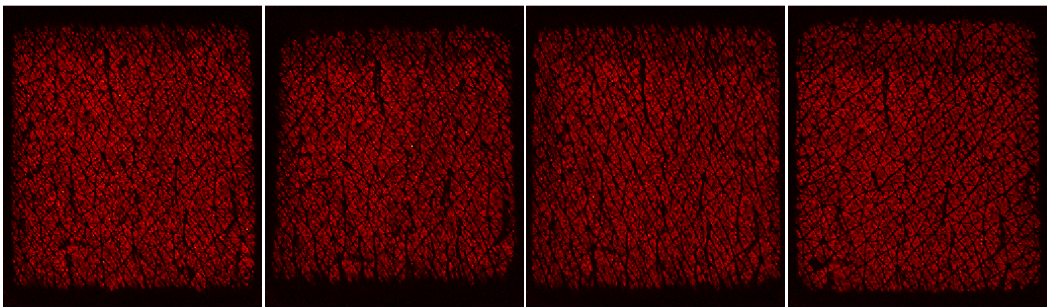


8 strips                      10 strips                      12 strips                      14 strips

**(C)**



Before                      After (SLS)                      10 min                      20 min



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30 min40 min50 min60min

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**Fig. 5.** Typical capacitive contact images of before and after intensive washes (A),

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tape stripping (B) and SLS irritation (C).

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2.2 Skin Solvent Penetrations

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**Fig. 6** shows the capacitive images before and after the three solvent applications.

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The image results show that after 5 minutes application, Ethylene Glycol penetrates

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most into the skin, while Ethanol penetrates the least. This is understandable, as

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Ethanol is very volatile, and evaporated quickly after the application. Ethylene Glycol

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also disappears quicker from the skin surface. As Ethylene Glycol is non-volatile, we

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assume it goes into deeper skin. Glycerol, however, tends to stuck within skin for a

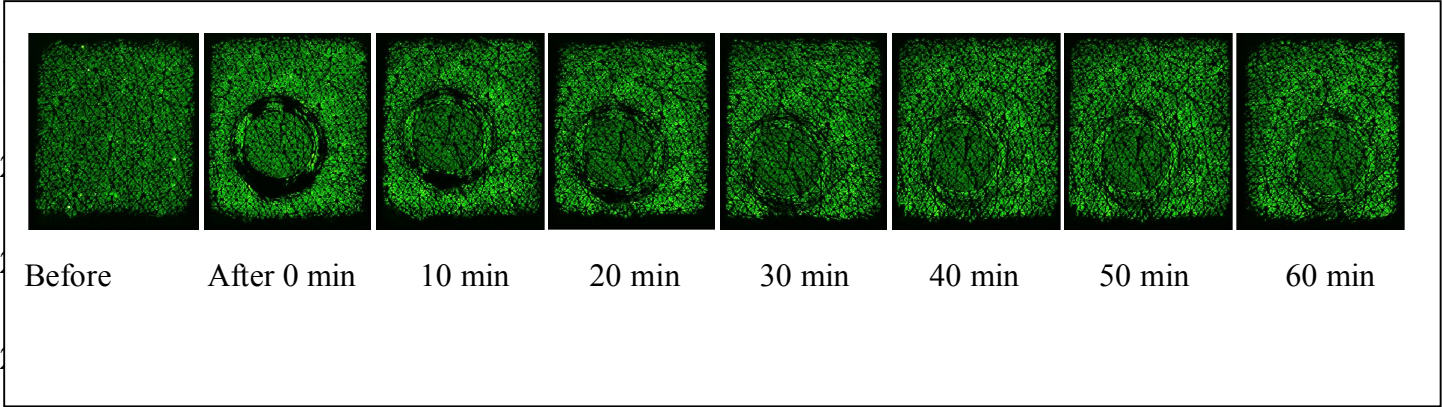
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long period of time.

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**(A) Ethanol**

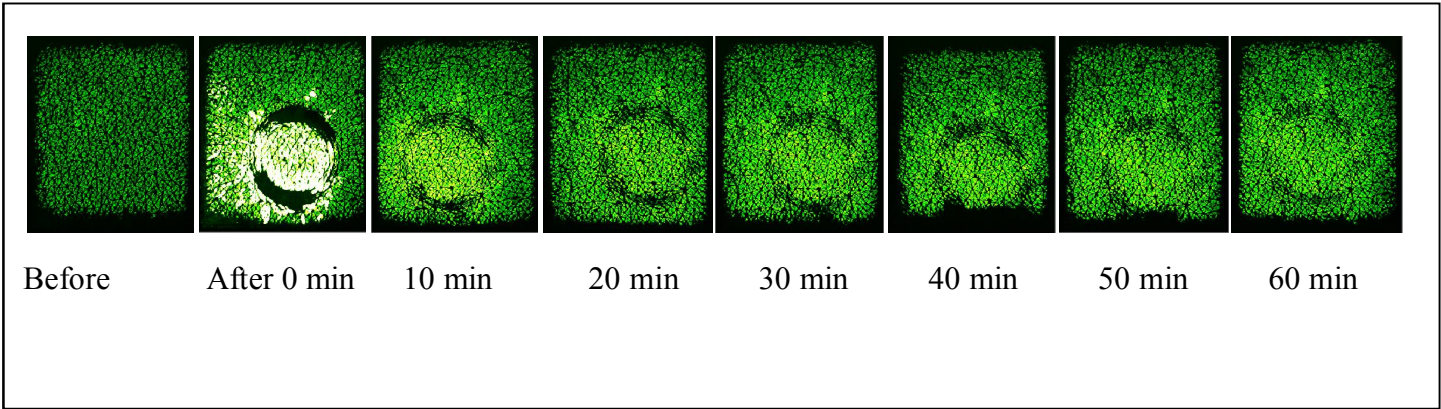


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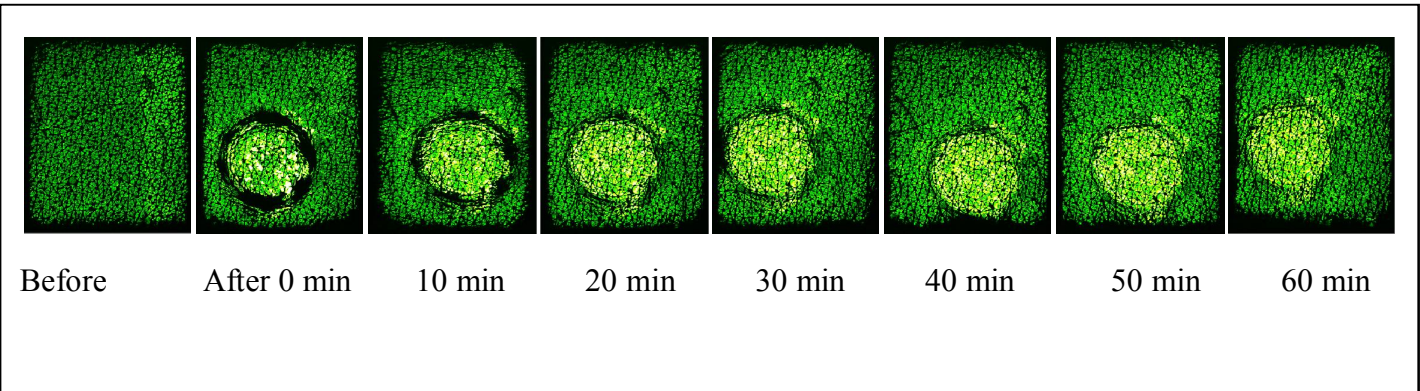
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205    **(B) Ethylene Glycol**



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211    **(C) Glycerol**



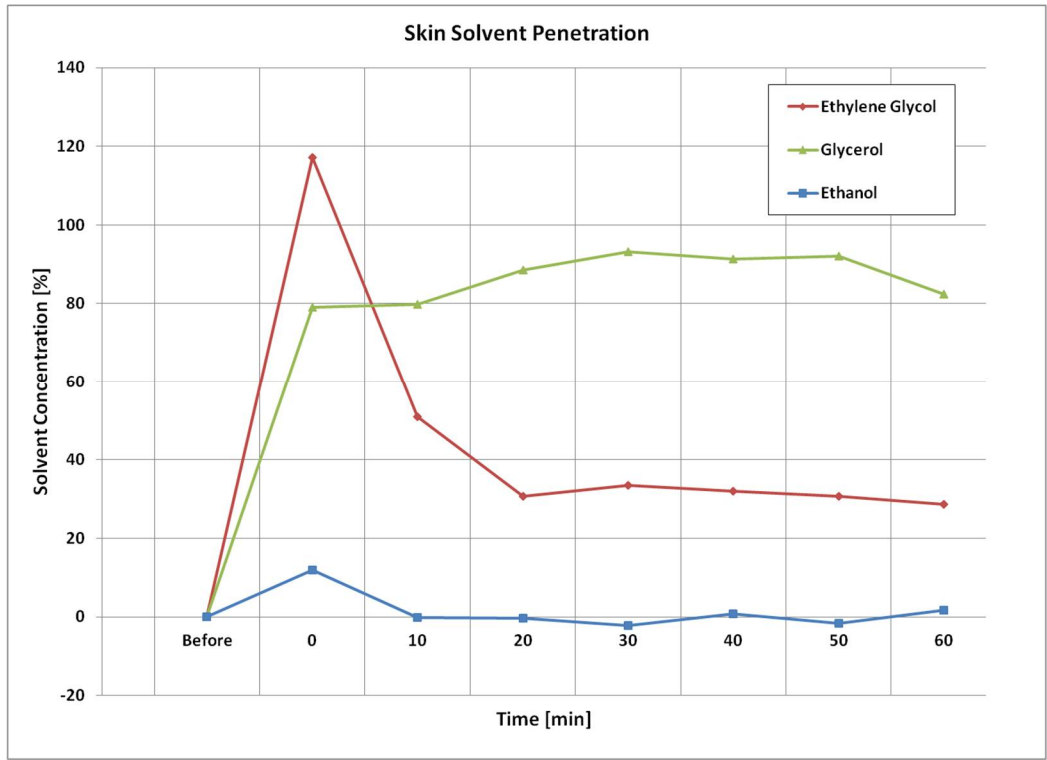
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217    **Fig. 6.** Capacitive contact images of skin before and after solvent applications, (A)  
218    Ethanol, (B) Ethylene Glycol and (C) Glycerol.

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220    **Fig. 6** shows the corresponding calculated solvent concentration as volume ratio  
221    percentage before after solvent applications using Eq.4. The calculated results

agree well with the image results. The slight negative undershoots in Ethanol results are likely due to the skin drying because of Ethanol application.

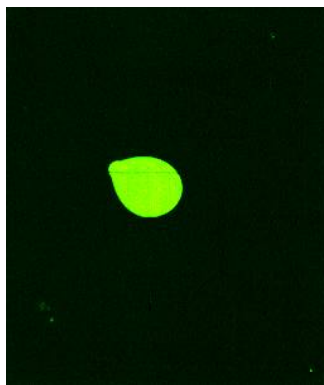


**Fig. 6.** The calculated solvent concentration as volume ratio percentage before after solvent applications, (A) Ethanol, (B) Ethylene Glycol and (C) Glycerol.



The table below shows the dielectric permittivity of alcohol, ethylene glycol, glycerol, and the concentration of these 3 solvents in the skin from base site to 60 minutes (base site to t6, every 10 minutes).

Figure 6 shows a 5min Glycerol application measurement. Similarly, we can use  $C=(\epsilon - \epsilon_{\text{skin}})/(\epsilon_{\text{Glycerol}} - \epsilon_{\text{skin}})$  to work out the Glycerol concentration. By using re-positioning technique (RoI + cross correlation), we can select exactly the same skin region we want to analyse in each image, see following red squares. The results show that Glycerol in skin has reached to ~38% (v/v) after just 5min Glycerol application. Then as Glycerol gradually diffuse into to skin, the concentration reduced to 16% (20 min), 9% (30min), 10% (40min) and 6% (50 min).



Glycerol

#### 4 Conclusions

250 The result shows that, because of volatilizing, the permittivity of alcohol decreased,  
251 after  $t_0$  ( $t_1$ - $t_6$ ), and the capacitive images become darker than base site, the  
252 permittivity of both ethylene glycol and glycerol much more increased than base site  
253 and diffusion under the surface, the capacitive images of these 2 solvents become  
254 brighter than base site, but we can notice the different between these 2 solvents,  
255 ethylene glycol is absorbed faster than glycerol.

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