

Supporting Material

On the fragmentation of Ni(II) β -diketonate-diamine complexes as molecular precursors for NiO films: a theoretical and experimental investigation

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§ S1 Optimized structures of species detected in mass spectrometry experiments

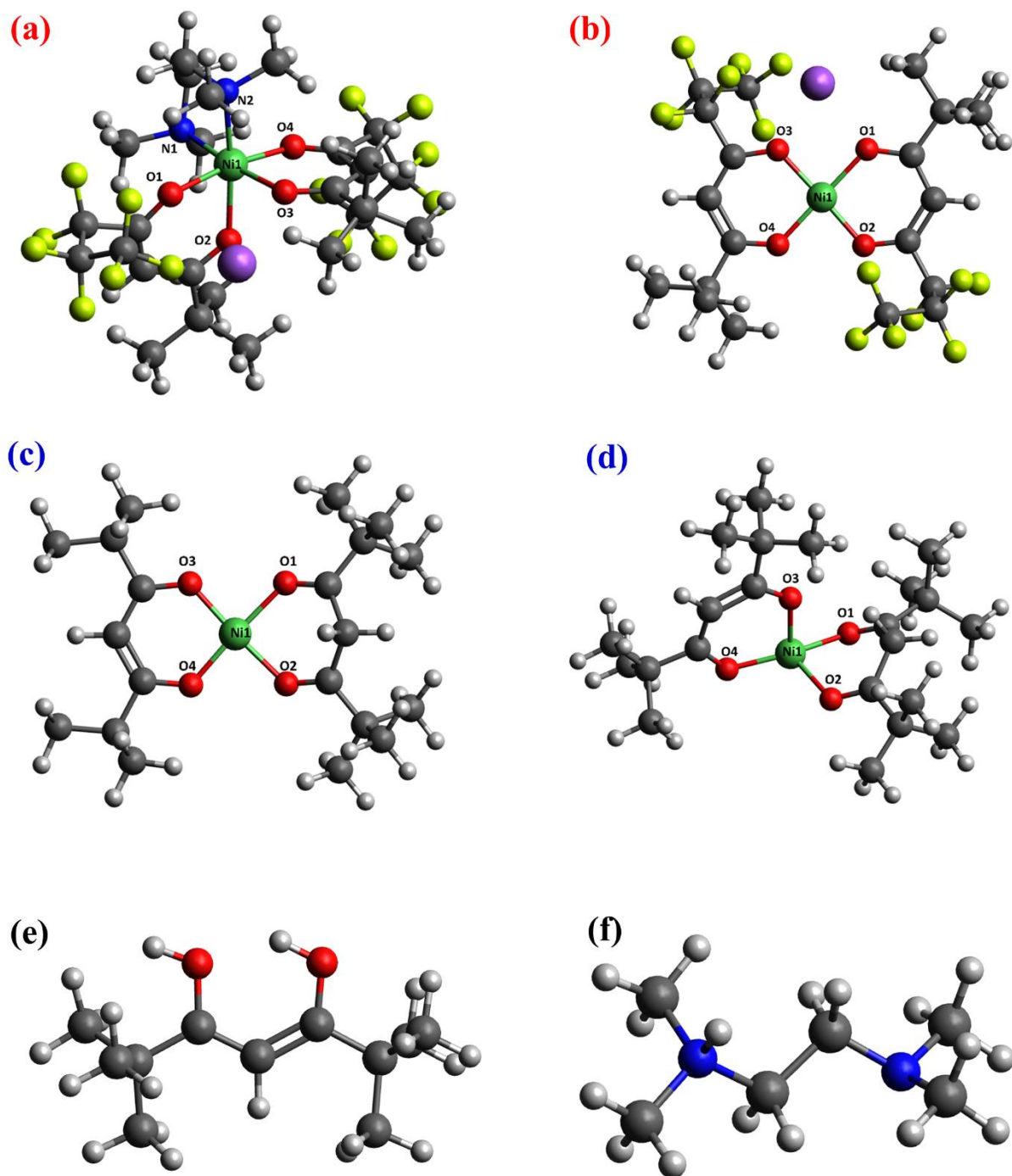


Figure S1. Graphical representation of the optimized structures of: (a) $[M+Na]^+ = [Ni(fod)_2TMEDA+Na]^+$, (triplet state); (b) $[M-TMEDA+Na]^+ = [Ni(fod)_2+Na]^+$, obtained by loss of TMEDA and addition of Na^+ from complex **2**; (c) and (d) $[M-TMEDA+H]^+ = [Ni(thd)_2+H]^+$ pseudo-planar geometry (singlet state) and distorted tetrahedral geometry (triplet state), respectively. The second one is $0.32 \text{ kcal}\cdot\text{mol}^{-1}$ lower in energy with respect to (d); (e) $[HL+H]^+ = [thd+H]^+$ obtained by addition of H^+ to thd ligand; (f) $[TMEDA+H]^+$ obtained by addition of H^+ to neutral TMEDA. Color codes: purple = Na; green = Ni; yellow = F; blue = N; red = O; grey = C; white = H.

Bond Lengths (Å)	[M+Na] ⁺ (a)	[M-TMEDA+Na] ⁺ (b)	[M-TMEDA+H] ⁺ (c)	[M-TMEDA+H] ⁺ (d)
<i>Ni-O1</i>	2.080	1.876	1.897	1.986
<i>Ni-O2</i>	2.112	1.822	1.897	1.981
<i>Ni-O3</i>	2.098	1.847	1.793	1.901
<i>Ni-O4</i>	1.989	1.837	1.793	1.842
<i>Ni-N1</i>	2.152	-	-	-
<i>Ni-N2</i>	2.126	-	-	-
<i>Na-O1</i>	2.502	2.389		
<i>Na-O2</i>	2.532	-		
<i>Na-O3</i>	2.215	2.240		
<i>Na-F</i>	2.441	2.542; 2.470		

Table S1. Ni-O and Ni-N bond distances of the species reported in Figure S1, i.e: (a) [M+Na]⁺ = [Ni(fod)₂TMEDA+Na]⁺, (triplet state); (b) [M-TMEDA+Na]⁺ = [Ni(fod)₂+Na]⁺, obtained from complex **2** by TMEDA loss and Na⁺ addition; (c)-(d), [M-TMEDA+H]⁺ = [Ni(thd)₂+H]⁺ pseudo-planar geometry (singlet state) and distorted tetrahedral geometry (triplet state), respectively.

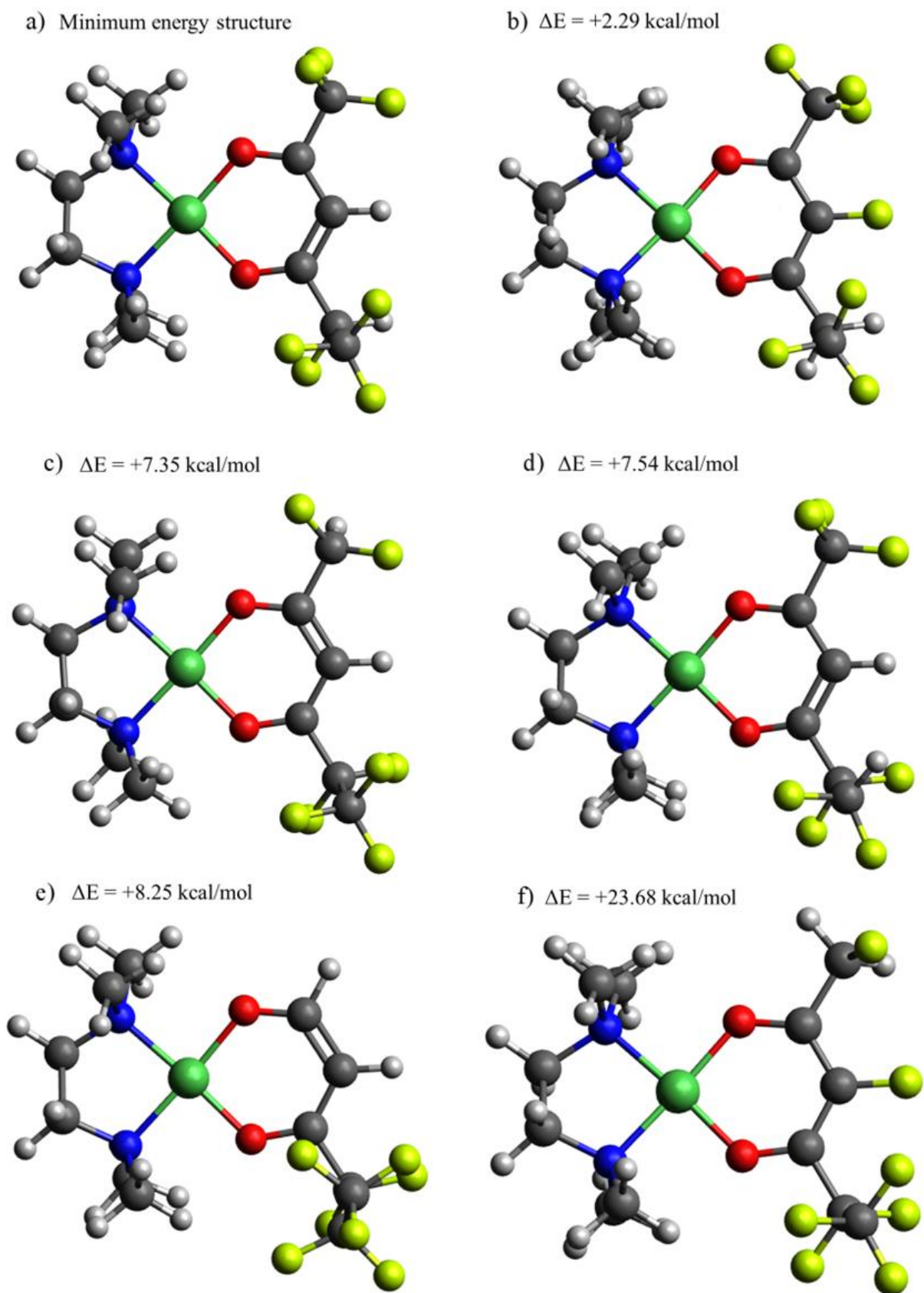


Figure S2. Graphical representation of six possible structures of fragment ion $[\text{Ni}(\text{fod})\text{TMEDA}]^+-(\text{CH}_3)_2\text{C}=\text{CH}_2$, derived from an additional rearrangement of fod ligand and detected in MS^2 spectra of precursor 2. Energy differences (relative energies) with respect to the minimum energy structure (a) are reported on the top of each model geometry. Color codes as in Figure S1.

§ S2. NBO analysis

Fragment	NBO type	Bond	Occupancy	% Ni	% X
Ni(tfa)TMEDA⁺					
	BD (σ)	<i>Ni-O2</i>	1.909	11	89
	BD (σ)	<i>Ni-O1</i>	1.904	11	89
	BD (σ)	<i>Ni-N1</i>	1.866	12	88
	BD (σ)	<i>Ni-N2</i>	1.868	12	88
	BD* (σ)	<i>Ni-O2</i>	0.100	89	11
	BD* (σ)	<i>Ni-O1</i>	0.102	89	11
	BD* (σ)	<i>Ni-N1</i>	0.128	88	12
	BD* (σ)	<i>Ni-N2</i>	0.128	88	12
Ni(fod)TMEDA⁺					
	BD (σ)	<i>Ni-O2</i>	1.903	11	89
	BD (σ)	<i>Ni-O1</i>	1.897	11	89
	BD (σ)	<i>Ni-N1</i>	1.861	11	89
	BD (σ)	<i>Ni-N2</i>	1.863	11	89
	BD* (σ)	<i>Ni-O2</i>	0.111	89	11
	BD* (σ)	<i>Ni-O1</i>	0.112	89	11
	BD* (σ)	<i>Ni-N1</i>	0.132	89	11
	BD* (σ)	<i>Ni-N2</i>	0.133	89	11
Ni(thd)TMEDA⁺					
	BD (σ)	<i>Ni-O2</i>	1.896	11	89
	BD (σ)	<i>Ni-O1</i>	1.896	11	89
	BD (σ)	<i>Ni-N1</i>	1.866	10	90
	BD (σ)	<i>Ni-N2</i>	1.866	10	90
	BD* (σ)	<i>Ni-O2</i>	0.110	89	11
	BD* (σ)	<i>Ni-O1</i>	0.110	89	11
	BD* (σ)	<i>Ni-N1</i>	0.137	90	10
	BD* (σ)	<i>Ni-N2</i>	0.137	90	10

Table S2. Natural bond orbital (NBO) analysis for [Ni(L)TMEDA]⁺, where L = tfa, fod, thd, reporting: NBO type, atoms on which NBOs are localized, NBO occupancy, and localization (%) of NBOs on the involved atoms. Legend: BD = bonding character; BD* = anti-bonding character; X = O or N. Atoms labels as in Figure 3.

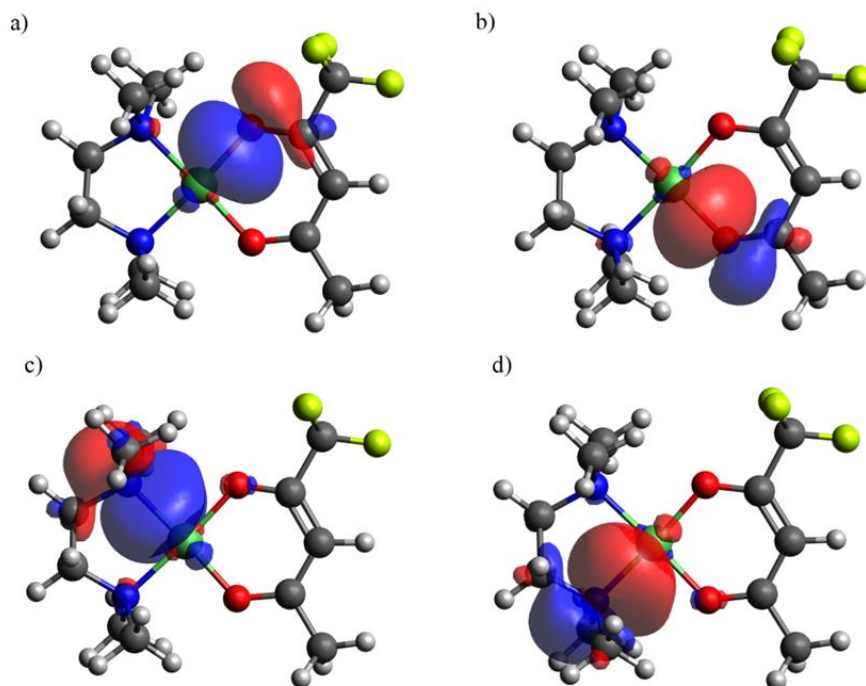


Figure S3. Graphical representation of natural bond orbitals of $[\text{Ni}(\text{tfa})\text{TMEDA}]^+$ localized on the following bonds: a) Ni-O1 (BD); b) Ni-O2 (BD); c) Ni-N1 (BD); d) Ni-N2 (BD) (see also Table S2). Atom color codes: green = Ni; yellow = F; blue = N; red = O; grey = C; white = H.

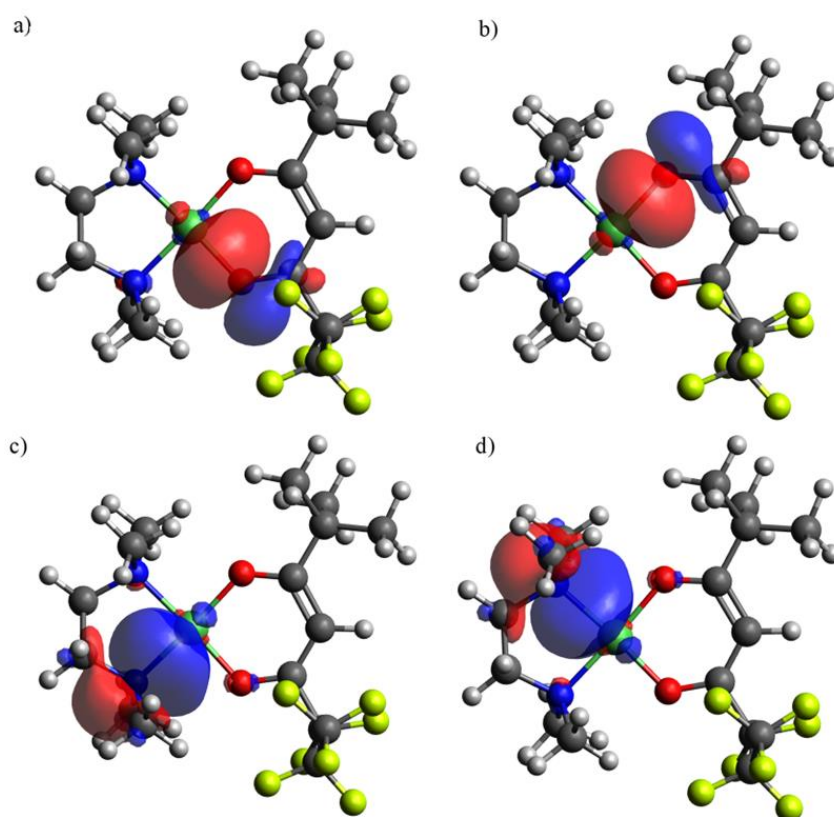


Figure S4. Graphical representation of natural bond orbitals of $[\text{Ni}(\text{fod})\text{TMEDA}]^+$ localized on the following bonds: a) Ni-O1 (BD); b) Ni-O2 (BD); c) Ni-N1 (BD); d) Ni-N2 (BD) (see also Table S2). Atom color codes: green = Ni; yellow = F; blue = N; red = O; grey = C; white = H.

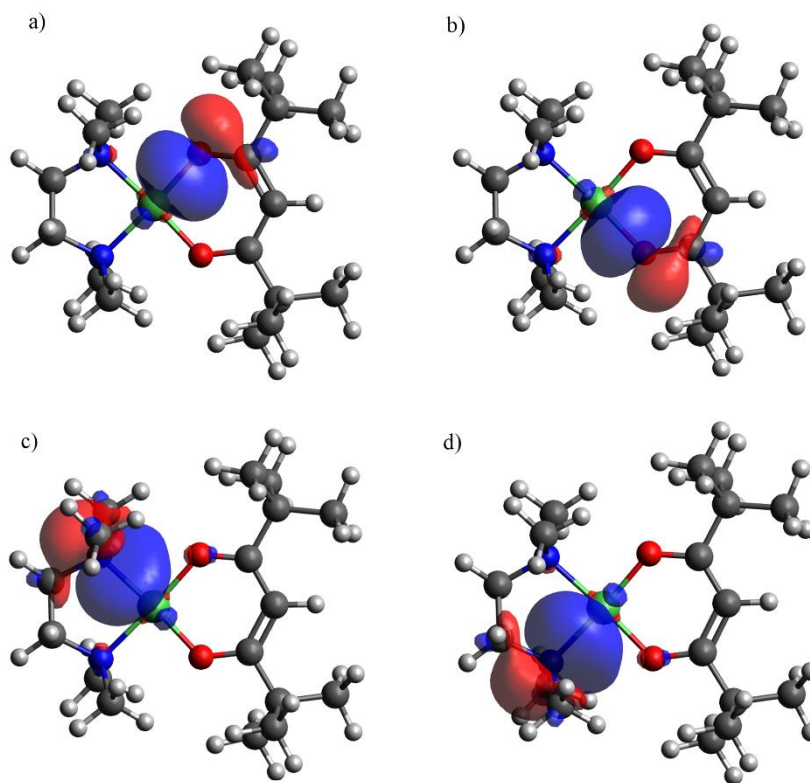


Figure S5. Graphical representation of natural bond orbitals of $[\text{Ni}(\text{thd})\text{TMEDA}]^+$ localized on the following bonds: a) Ni-O1 (BD); b) Ni-O2 (BD); c) Ni-N1 (BD); d) Ni-N2 (BD) (see also Table S2). Atom color codes: green = Ni; blue = N; red = O; grey = C; white = H.

Fragment	NBO type	Bond	Occupancy	% Y	% X
Ni(tfa)TMEDA⁺ -NH(CH₃)₂					
	BD (π)	<i>CHT-CH2T</i>	1.985	52	48
	BD (π)	<i>CHT-CH2T</i>	1.663	49	51
	BD (σ)	<i>Ni-N1</i>	1.878	26	74
	BD* (π)	<i>CHT-CH2T</i>	0.011	48	52
	BD* (π)	<i>CHT-CH2T</i>	0.187	51	49
Ni(fod)TMEDA⁺ -NH(CH₃)₂	BD* (σ)	<i>Ni-N1</i>	0.527	74	26
	Ni(fod)TMEDA⁺ -NH(CH₃)₂				
	BD (π)	<i>CHT-CH2T</i>	1.981	52	48
	LP (π)	<i>CH2T</i>	0.959	-	-
	BD (σ)	<i>CHT-N1</i>	1.749	22	78
Ni(thd)TMEDA⁺ -NH(CH₃)₂	BD* (π)	<i>CHT-CH2T</i>	0.010	48	52
	BD* (σ)	<i>CHT-N1</i>	0.676	78	22
	Ni(thd)TMEDA⁺ -NH(CH₃)₂				
	BD (π)	<i>CHT-CH2T</i>	1.985	52	48
	BD (π)	<i>CHT-CH2T</i>	1.691	49	51
Ni(thd)TMEDA⁺ -NH(CH₃)₂	BD (σ)	<i>Ni-N1</i>	1.871	24	76
	BD* (π)	<i>CHT-CH2T</i>	0.011	48	52
	BD* (π)	<i>CHT-CH2T</i>	0.198	51	49
	BD* (σ)	<i>Ni-N1</i>	0.548	76	24

Table S3. NBO analysis for [Ni(L)TMEDA]⁺-NH(CH₃)₂, where L = tfa, fod, thd. The table reports: NBO type, atoms on which NBOs are localized, NBOs occupancy and the percentage localization of NBOs on the involved atoms. Legend: BD = bonding; BD* = anti-bonding; LP = lone-pair; Y = Ni or CHT; X = N1 or CH2T. Atoms labels as in Figure 4.

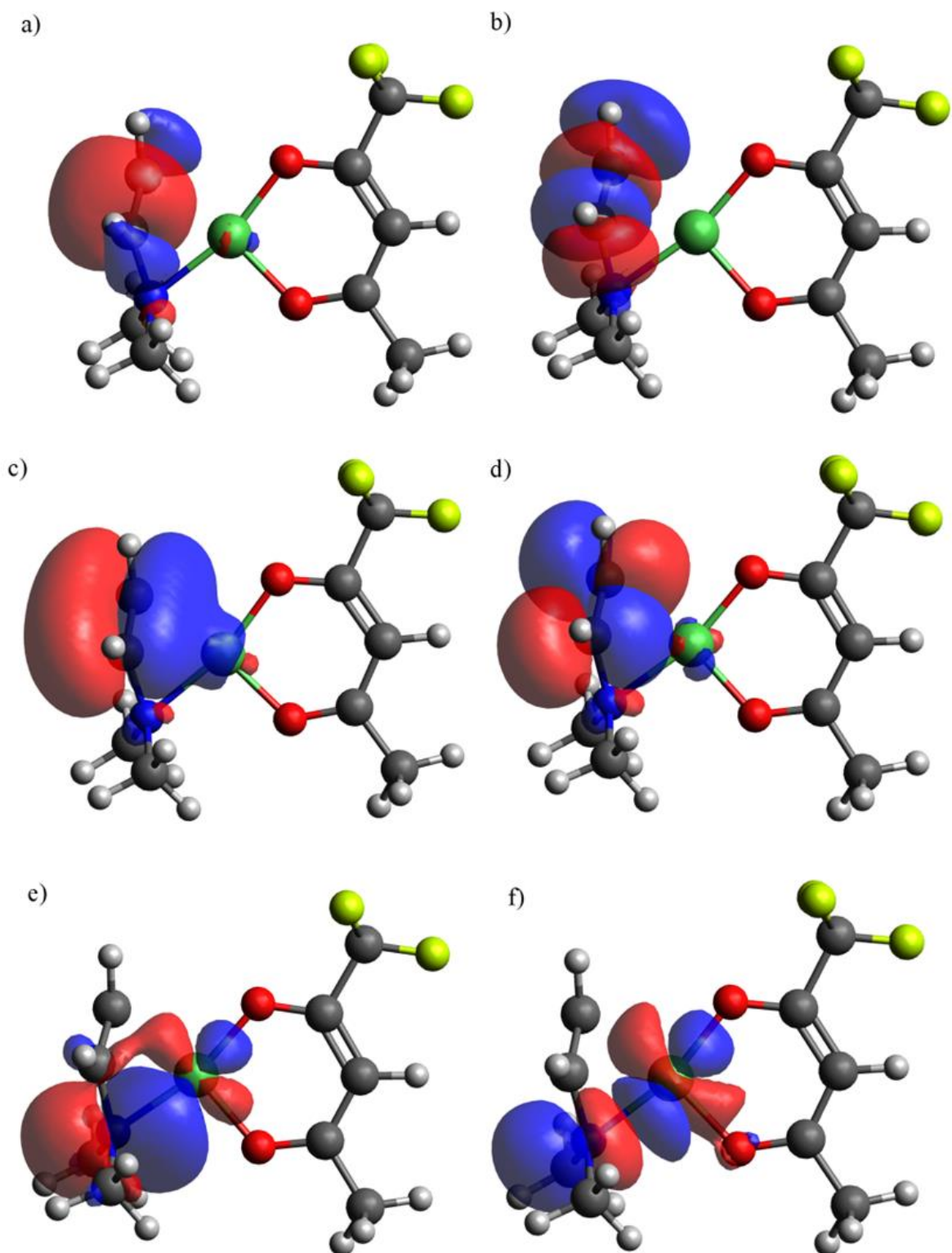


Figure S6. Graphical representation of natural bond orbitals of Ni(tfa)TMEDA⁺-NH(CH₃)₂ localized on the following bonds: a) CHT-CH₂T (BD); b) CHT-CH₂T (BD*); c) CHT-CH₂T (BD); d) CHT-CH₂T (BD*); e) Ni-N1 (BD); f) Ni-N1 (BD*) (see also Table S3). Atom color codes: green = Ni; yellow = F; blue = N; red = O; grey = C; white = H.

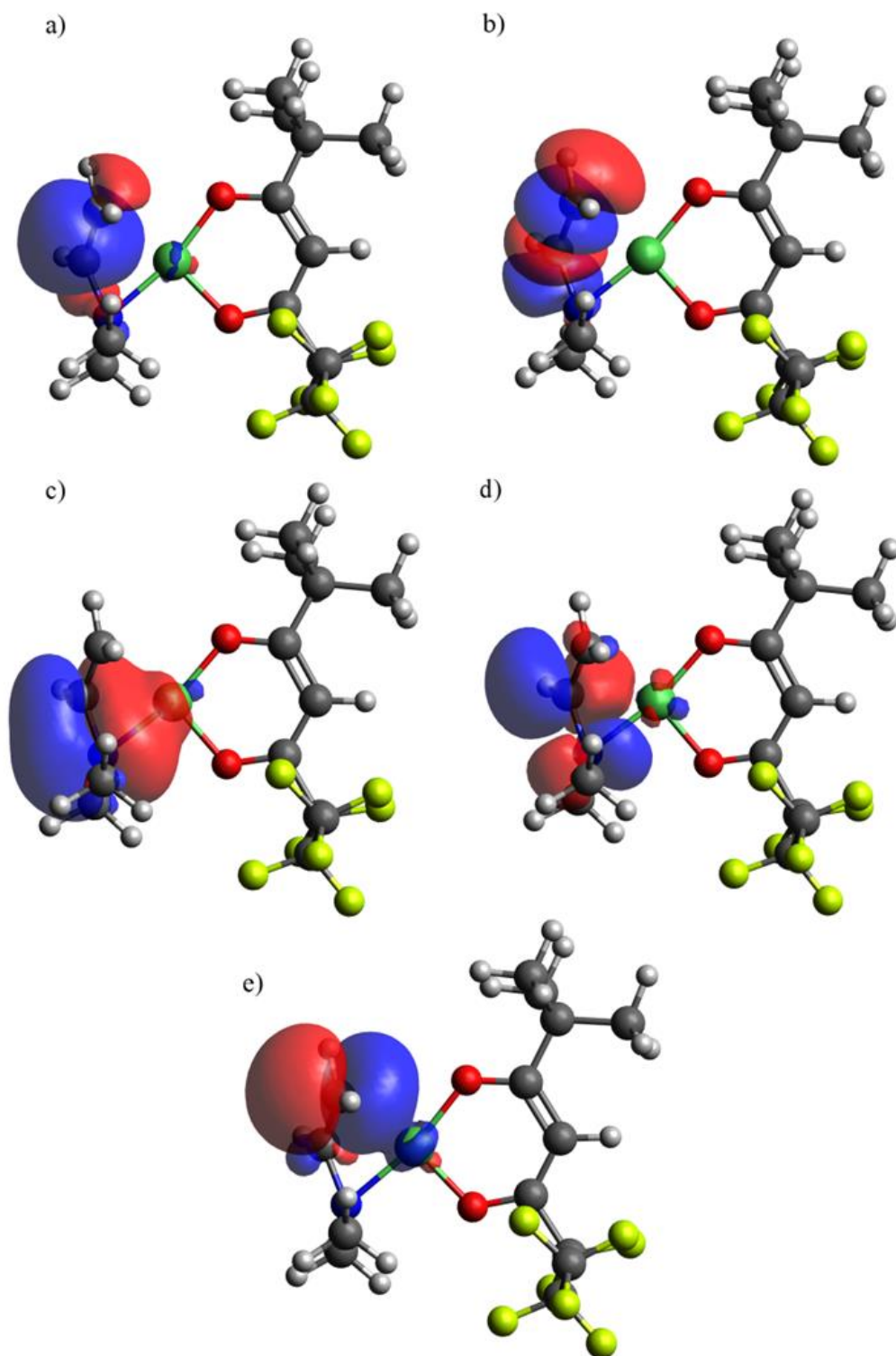


Figure S7. Graphical representation of natural bond orbitals of $\text{Ni}(\text{fod})\text{TMEDA}^+\text{-NH}(\text{CH}_3)_2$ localized on the following bonds: a) CHT-CH₂T (BD); b) CHT-CH₂T (BD*); c) Ni-N1 (BD); d) Ni-N1 (BD*); e) CH₂T (LP) (see also Table S3). Atom color codes: green = Ni; yellow = F; blue = N; red = O; grey = C; white = H.

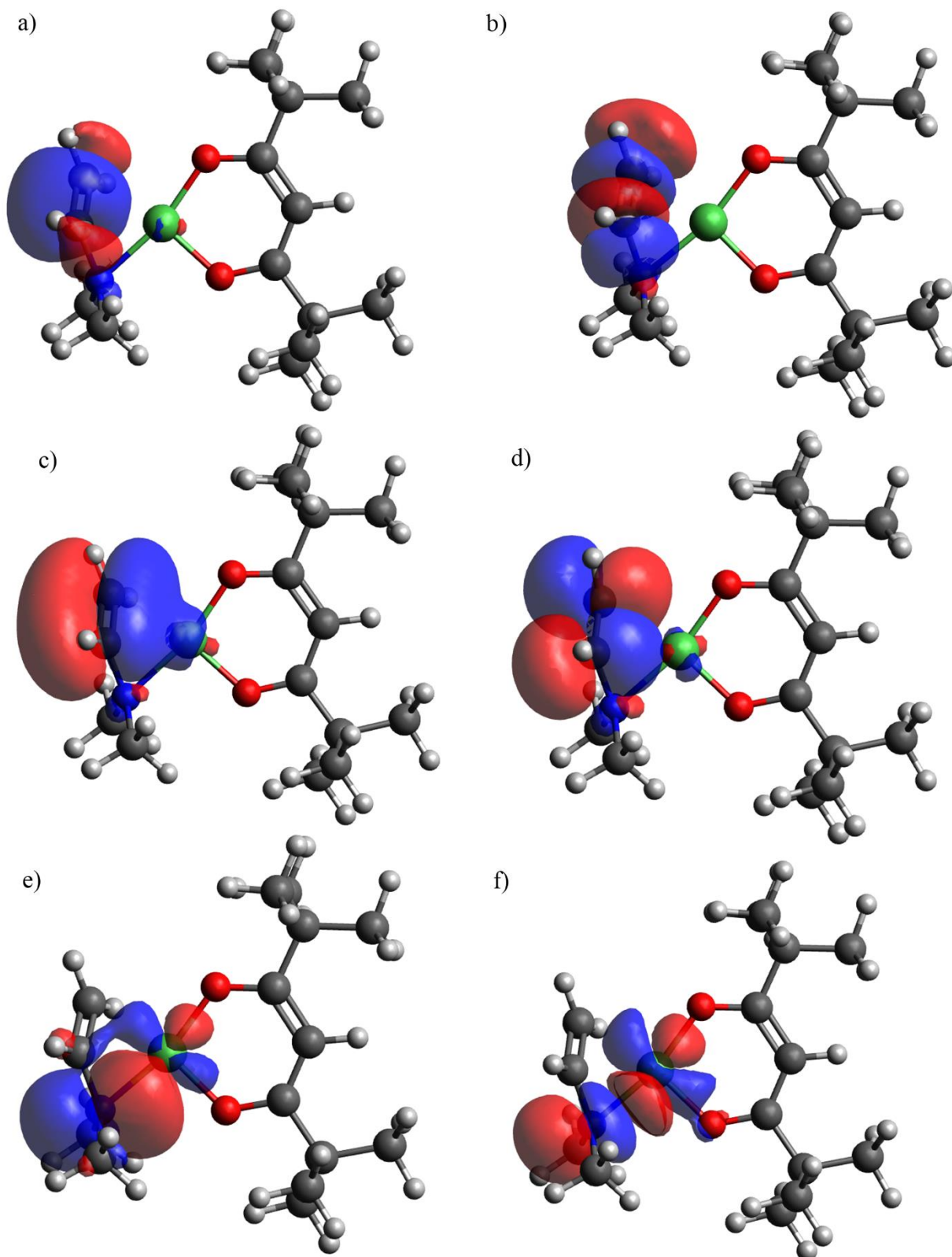


Figure S8. Graphical representation of natural bond orbitals of Ni(thd)TMEDA⁺-NH(CH₃)₂ localized on the following bonds: a) CHT-CH₂T (BD); b) CHT-CH₂T (BD*); c) CHT-CH₂T (BD); d) CHT-CH₂T (BD*); e) Ni-N1 (BD); f) Ni-N1 (BD*) (see also Table S3). Atom color codes: green = Ni; blue = N; red = O; grey = C; white = H.

Fragment	NBO type	Bond	Occupancy	% Y	% X
Ni(tfa)TMEDA⁺ -CH₃CH₂N(CH₃)₂					
	BD (π)	<i>CHT-CH2T</i>	1.987	52	48
	LP (π)	<i>CHT</i>	0.916	-	-
	LP (π)	<i>CH2T</i>	0.914	-	-
	BD (σ)	<i>Ni-N1</i>	1.836	14	86
	BD* (π)	<i>CHT-CH2T</i>	0.010	48	52
	BD* (σ)	<i>Ni-N1</i>	0.197	86	14
Ni(fod)TMEDA⁺ - CH₃CH₂N(CH₃)₂					
	BD (π)	<i>CHT-CH2T</i>	1.987	52	48
	LP (π)	<i>CHT</i>	0.920	-	-
	LP (π)	<i>CH2T</i>	0.914	-	-
	BD (σ)	<i>Ni-N1</i>	1.838	14	86
	BD* (π)	<i>CHT-CH2T</i>	0.010	48	52
	BD* (σ)	<i>Ni-N1</i>	0.197	86	14

Table S4. NBO analysis for [Ni(L)TMEDA]⁺-CH₃CH₂N(CH₃)₂, where L = tfa and fod. The table shows the NBO type, the atoms on which NBOs are localized, NBO occupancy and the percentage localization of NBOs on the involved atoms. Legend: BD = bonding; BD* = anti-bonding; LP = lone-pair; Y = Ni or CHT; X = N1 or CH2T. Atoms labels as in Figure 4.

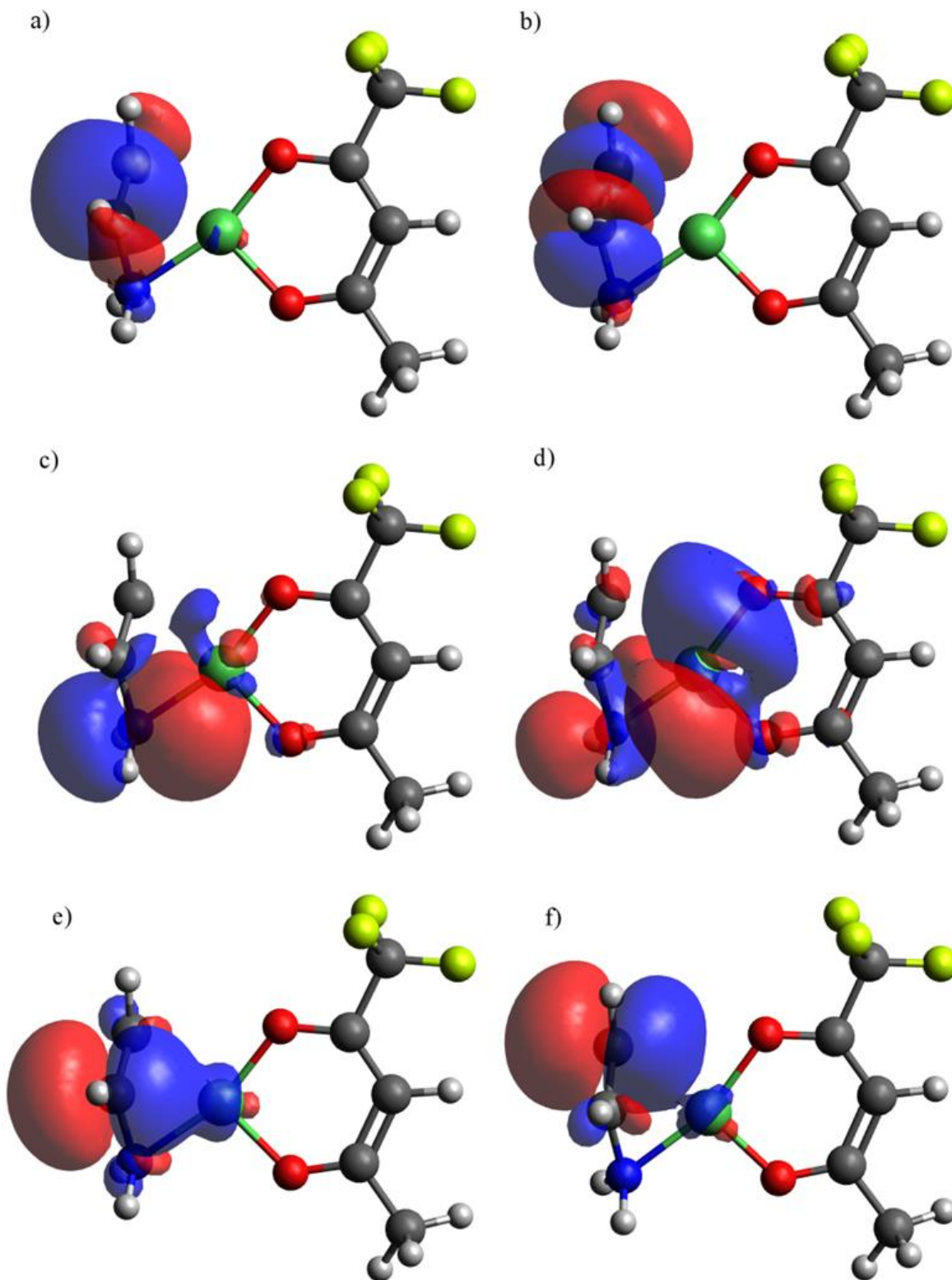


Figure S9. Graphical representation of natural bond orbitals of Ni(tfa)TMEDA⁺-CH₃CH₂N(CH₃)₂ localized on the following bonds: a) CHT-CH₂T (BD); b) CHT-CH₂T (BD*); c) Ni-N1 (BD); d) Ni-N1 (BD*); e) CHT (LP); f) CH₂T (LP) (see also Table S4). Atom color codes: green = Ni; yellow = F; blue = N; red = O; grey = C; white = H.

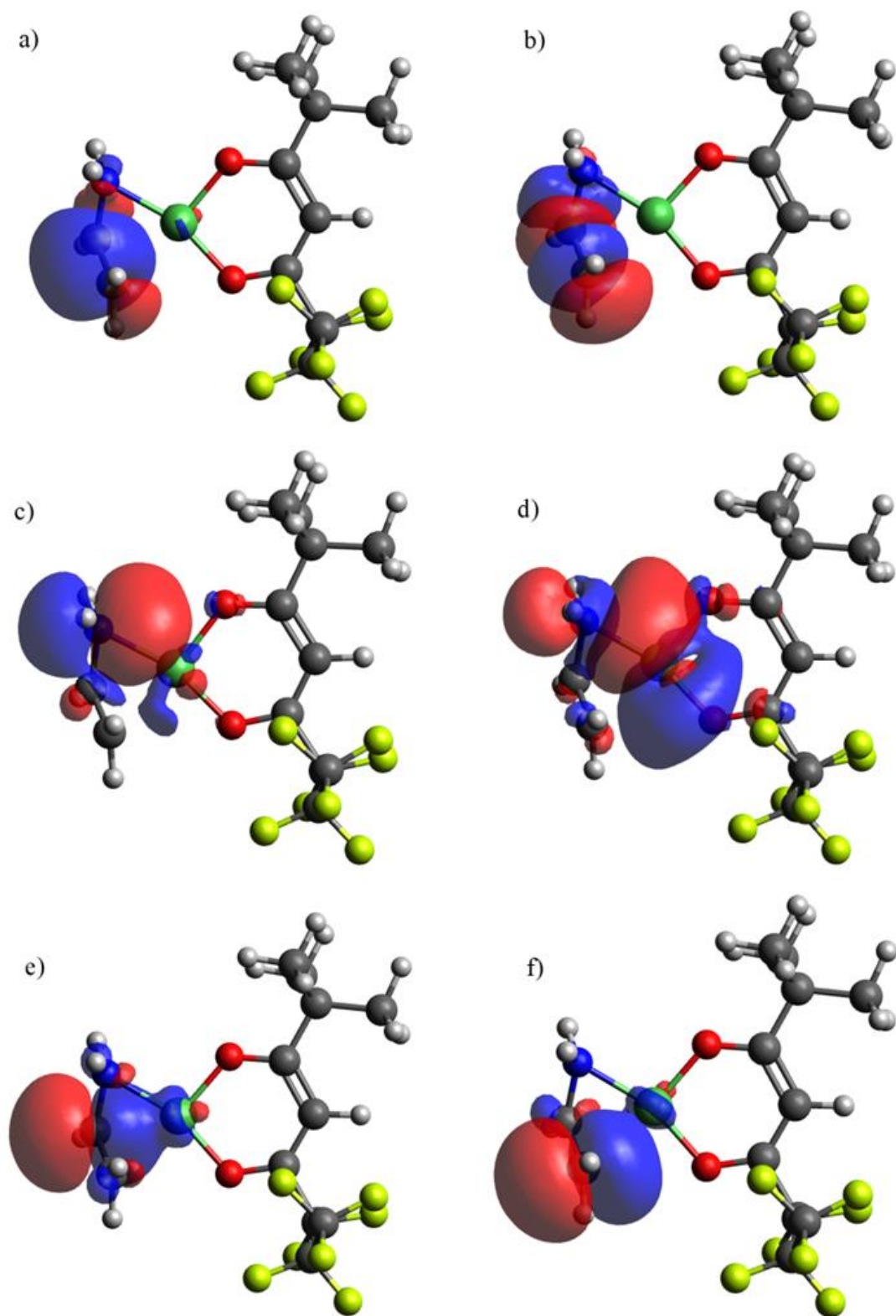


Figure S10. Graphical representation of natural bond orbitals of Ni(fod)TMEDA⁺-CH₃CH₂N(CH₃)₂ localized on the following bonds: a) CHT-CH₂T (BD); b) CHT-CH₂T (BD*); c) Ni-N1 (BD); d) Ni-N1 (BD*); e) CHT (LP); f) CH₂T (LP) (see also Table S4). Atom color codes: green = Ni; yellow = F; blue = N; red = O; grey = C; white = H.

Atom	Ni(tfa) ₂ TMEDA	Ni(fod) ₂ TMEDA	Ni(thd) ₂ TMEDA
Ni	+1.33	+1.33	+1.33
O1	-1.19	-1.18	-1.21
O2	-1.20	-1.19	-1.21
O3	-1.20	-1.20	-1.19
O4	-1.19	-1.18	-1.21
N1	-1.03	-1.03	-1.04
N2	-1.05	-1.05	-1.04
C5	+0.81	+0.81	+0.78
C6	+0.82	+0.77	+0.77
C7	+0.86	+0.84	+0.78
C8	+0.86	+0.85	+0.78
CH1	-0.01	-0.03	-0.07
CH2	-0.03	0.00	-0.05
CHT	+0.27	+0.25	+0.28
CH ₂ T	+0.31	+0.28	+0.28

Table S5. Bader charges for [Ni(L)₂TMEDA], where L = tfa, fod, thd. Atom labels as in Figure 3.

Atom	[Ni(tfa)TMEDA] ⁺	[Ni(fod)TMEDA] ⁺	[Ni(thd)TMEDA] ⁺
Ni	+0.99	+1.00	+1.00
O1	-1.14	-1.16	-1.16
O2	-1.14	-1.17	-1.17
N1	-1.03	-1.01	-1.01
N2	-1.01	-1.03	-1.03
C5	+0.71	+0.67	+0.67
C7	+0.72	+0.65	+0.65
CH1	+0.10	+0.08	+0.08
CHT	+0.28	+0.27	+0.27
CH ₂ T	+0.28	+0.27	+0.27

Table S6. Bader charges for [Ni(L)TMEDA]⁺, where L = tfa, fod, thd. Atom labels as in Figure 3.

Atom	[Ni(tfa)TMEDA] ⁺ - NH(CH ₃) ₂	[Ni(fod)TMEDA] ⁺ - NH(CH ₃) ₂	[Ni(thd)TMEDA] ⁺ - NH(CH ₃) ₂
Ni	+0.95	+0.95	+0.95
O1	-1.12	-1.13	-1.13
O2	-1.17	-1.12	-1.12
N	-1.09	-1.10	-1.10
C5	+0.70	+0.67	+0.67
C7	+0.72	+0.71	+0.71
CH1	+0.09	+0.09	+0.09
CHT	+0.22	+0.21	+0.21
CH ₂ T	-0.05	-0.08	-0.08

Table S7. Bader charges for [Ni(L)TMEDA]⁺ -NH(CH₃)₂, where L = tfa, fod, thd. Atom labels as in Figure 4. N indicates the nitrogen atom bound to Ni.

Atom	[Ni(tfa)TMEDA] ⁺ - CH ₃ CH ₂ N(CH ₃) ₂	[Ni(fod)TMEDA] ⁺ - CH ₃ CH ₂ N(CH ₃) ₂	[Ni(fod)TMEDA] ⁺ - (CH ₃) ₂ C=CH ₂
Ni	+0.97	+0.97	+1.00
O1	-1.11	-1.09	-1.12
O2	-1.12	-1.13	-1.14
N	-1.15	-1.14	-1.01
C5	+0.68	+0.68	+0.76
C7	+0.72	+0.69	+0.77
CH1	+0.09	+0.08	+0.07
CHT	+0.21	+0.19	+0.27
CH ₂ T	-0.06	-0.05	+0.24

Table S8. Bader charges for [Ni(L)TMEDA]⁺ - CH₃CH₂N(CH₃)₂, where L = tfa, fod, and for [Ni(fod)TMEDA]⁺ - (CH₃)₂C=CH₂. Atom labels as in Figure 4. In each case, N indicates the nitrogen atom(s) bound to Ni (for [Ni(fod)TMEDA]⁺ - (CH₃)₂C=CH₂, both N atoms are ligated to Ni and bear identical Bader charges).