

1 Article

2 Pyrochlore-Group Minerals in the Granite-Hosted 3 Katugin Rare-Metal Deposit, Transbaikalia, Russia

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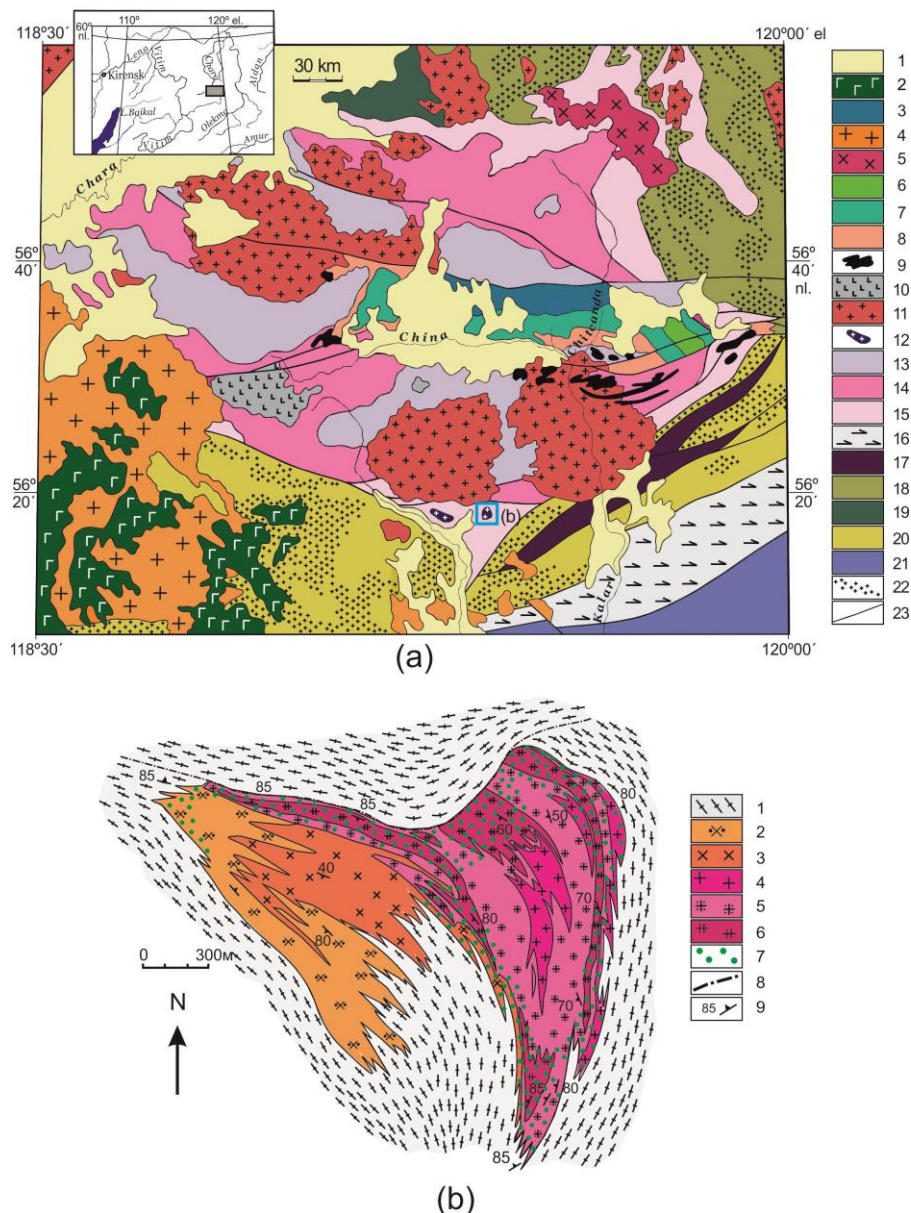
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16 **Abstract:** Pyrochlore group minerals are the main raw phases in granitic rocks of the Katugin
17 complex-ore deposit that stores Nb, Ta, Y, REE, U, Th, Zr, and cryolite. They are of three main
18 generations: primary magmatic (I), early postmagmatic (II), and supergene (III) pyrochlores. The
19 primary magmatic phase (generation I) is fluornatropyrochlore with high concentrations of Na₂O
20 (to 10.5 wt.%), F (to 5.4 wt.%) and REE₂O₃ (to 17.1 wt.%) but low CaO (0.6-4.3 wt.%), UO₂ (to 2.6
21 wt.%), ThO₂ (to 1.8 wt.%), and PbO (to 1.4 wt.%). Pyrochlore of this type is very rare in nature and
22 limited to a few occurrences, such as rare-metal deposits of Nechalacho in syenite and nepheline
23 syenite (Canada) and Mariupol in nepheline syenite (Ukraine). It may have crystallized
24 synchronously with or slightly later than melanocratic minerals (aegirine, biotite, and arfvedsonite)
25 at the late magmatic stage when Fe from the melt became bound making impossible the formation
26 of columbite. Second generation pyrochlore formed at the early postmagmatic stage of the Katugin
27 deposit. It differs from that of first generation in lower Na₂O concentrations (2.8 wt.%), relatively
28 low F (4 wt.%), and less occupancy of the A and Y sites at similar contents of other components.
29 Generation III pyrochlore is a product of supergene alteration processes. It is compositionally
30 heterogeneous and contains K, Ba, Pb, Fe, and significant Si concentrations but low Na and F. Its
31 compositions mostly fall within the field of hydro- and kenopyrochlore.

32 **Keywords:** pyrochlore-group minerals, fluornatropyrochlore, alkaline granites, Katugin rare-metal
33 deposit, East Transbaikalia

35 1. Introduction

36 Pyrochlore-group minerals are broadly occurring in nature and are of important economic
37 interest due to their ability to host Ta, Nb, U, and REE. They are common accessory phases in
38 carbonatite and are often found in such lithologies as nepheline syenite, alkaline gabbro and
39 granitoids, pegmatite, and in albite and greysen granites [1-6, etc.]. Pyrochlore minerals have diverse
40 compositions with the general formula A_{2-m}B₂O_{6-w}Y_{1-n}, where A represents large cations (Na, Ca, U,
41 Th, REE, Y, Sr, Ba, Pb, Sn, and H₂O) and site vacancies, the B-site is occupied by smaller cations (Nb,
42 Ti, and Ta), and Y is occupied by F, OH⁻, O²⁻ and H₂O or site vacancies [7-9]. The symbols *m*, *w*, and *n*
43 represent parameters indicating incomplete occupancy of the A, X, and Y.



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Figure 1. a: Simplified geological map of the region of the Katugin granite complex, after [11,12]. 1 = Quaternary sediments; 2 = flood basalts (N₂-Q); 3 = Jurassic coal-bearing clastic sediments; 4 = granite, granodiorite, granosyenite and monzonite of the Ingamakit complex (PZ₃); 5 = nepheline syenite, granosyenite and monzonite of the Khanin complex (PZ₃); 6 = Ordovician variegated sediments; 7 = Cambrian variegated sediments; 8 = Vendian variegated sediments; 9 = gabbro-dolerite, gabbro and dolerite porphyrite of the Doros complex; 10 = layered plutons of the Chinei complex; 11 = granite of the Kodar complex; 12 = rare-metal granite of the Katugin complex; 13-15 = carbonate and clastic sediments of the Udokan supergroup: Kemen group (13), Chinei group (14), Kodar group (15); 16 = anorthosite of the Kalar complex; 17 = low-grade metamorphic volcanic-sedimentary rocks of the Subgan complex; 18 = tonalite-trondjemite orthogneiss of the Olekma complex; 19 = Chara sequence (garnet-biotite and garnet-hypersthene-biotite plagiogneiss, mafic schists, quartzite and magnetite quartzite); 20 = Kalar sequence (garnet-biotite plagiogneiss with layers and lenses of two-pyroxene mafic granulite, clac-silicate rocks, quartzite, and magnetite quartzite); 21 = metamorphic and igneous complexes of the Selenga-Stanovik superterrane in the Central Asian orogenic belt; 22 = zones of widespread Precambrian granitic rocks; 23 = faults.

b: Simplified geology of the Katugin granite intrusion, after [13,14]. 1 = gneiss, migmatite, blastomylonite of the Kodar group, Udokan group; 2-6 = Katugin granites of different types: biotite (2), biotite-riebeckite (3), biotite-arfvedsonite (4), arfvedsonite (5), aegirine-arfvedsonite and aegirine (6); 7 = ore-bearing granite; 8 = faults; 9 = bedding elements.

64 The Ca/Na ratios vary considerably. Fluorcalciopyrochlore $(Ca,Na)_2Ca_2O_6F$ is the most
65 widespread phase in the group [3,8,10]. Findings of pyrochlore with prominent Na enrichment over
66 Ca and other cations at the A-site are very few and limited to several rare-metal deposits hosted by
67 the Nechalacho syenite and nepheline syenite (Canada) [15,16], the Mariupol nepheline syenite
68 (Ukraine) [17], the Strange Lake granite (Canada) [18], and the Halzdan-Buregteg alkaline granite
69 (West Mongolia) [19]. Fluornatropyrochlore $(Na,Pb,Ca,REE,U)_2Nb_2O_6F$ was reported as a mineral
70 species from the Boziguoer granites (China) only in 2013 [20].

71 This paper focuses on high-Na pyrochlores from the Katugin rare-metal deposit hosted by
72 alkaline granites. In previous publications, pyrochlores were described in brief among other
73 rock-forming, ore and accessory minerals [13,21,22], mostly based on monofraction wet chemistry
74 data and few EPMA determinations.

75 2. Geological setting

76 The Katugin deposit extremely rich in REE, Nb, and Ta is located in the Aldan shield, at the
77 southern margin of the Chara-Olekma terrane near the border with terranes of the Central Asian
78 Orogenic Belt marked by the Stanovik suture [23]. The ores are hosted by two relatively small (3 and
79 18 km²) A-type alkaline granite intrusions of the Katugin complex [11,13] (Fig.1a), which were earlier
80 interpreted as alkaline metasomatites [21,, 24-26]. The Western intrusion is 80% buried under a
81 moraine and remains quite poorly documented. The heart-shaped Eastern intrusion (Fig. 1b) is
82 composed of three types of granites corresponding to three phases of magmatism [11,13]: phases I,
83 II, and III, respectively, of biotite (Bt) and biotite-riebeckite (Bt-Rbk), biotite-arfvedsonite (Bt-Arf),
84 and arfvedsonite, aegirine-arfvedsonite and aegirine (Arf, Aeg-Arf, Aeg) granites. Phase III granites
85 have diverse mineralogy and lack distinct boundaries. They are joined into one phase because the
86 Aeg, Aeg-Arf, and Arf varieties may grade from one to another in a single sample. All types are
87 medium- and fine-grained, often gneissic, leucocratic and mesocratic quartz-albite-microcline
88 granites with variable amounts of melanocratic minerals; some arfvedsonite and aegirine granites
89 appear as isolated pegmatitic veins.

90 Biotite and biotite-riebeckite granites (phase I) have moderate or high Al contents and quite
91 high CaO, but contain the smallest amount of F among all granites we sampled. Biotite-arfvedsonite
92 granites (phase II) are supersaturated with respect to alkalis, rich in CaO and Y, but poor in Na₂O
93 and F. The ore-bearing granites of phase III are likewise supersaturated in alkalis; they have high
94 concentrations of Na₂O and F but very low CaO, and are mostly rich in Zr, Hf, Nb, and Ta [11].

95 The Katugin granites host Zr, Ta-Nb-REE, and F-Al mineralization [21,22]. Zircons are present
96 either as sporadic grains or as large clusters (to 20 vol.%) in all rocks of the intrusion. Fluoraluminate
97 minerals include abundant cryolite (a large body on the intrusion margin and ≤ 3 cm globules in Aeg
98 granites) and smaller percentages of other phases (weberite, chiolite, prosopite, pachtolite,
99 thomsenolite, and Ba-fluoraluminates). Ta-Nb-REE phases are mainly pyrochlore, columbite-(Fe),
100 monazite-(Ce), fluorides (fluorzerite-(Ce), gagarinite-(Y), tveitite-(Y), yttrifluorite) and
101 fluorcarbonates (bastnäsite-(Ce) and parizite-(Ce)) or less often monazite-(Ce). REE fluorides and
102 fluorcarbonates often form aggregates unevenly distributed in the rocks [27,28]. Other minerals are
103 ilmenite, magnetite, and sulfides (sphalerite, galena, pyrrhotite or pyrite). The mineralization is
104 roughly coeval to the host intrusion (2055 \pm 7 Ma against 2066 \pm 6 Ma) [23,29].

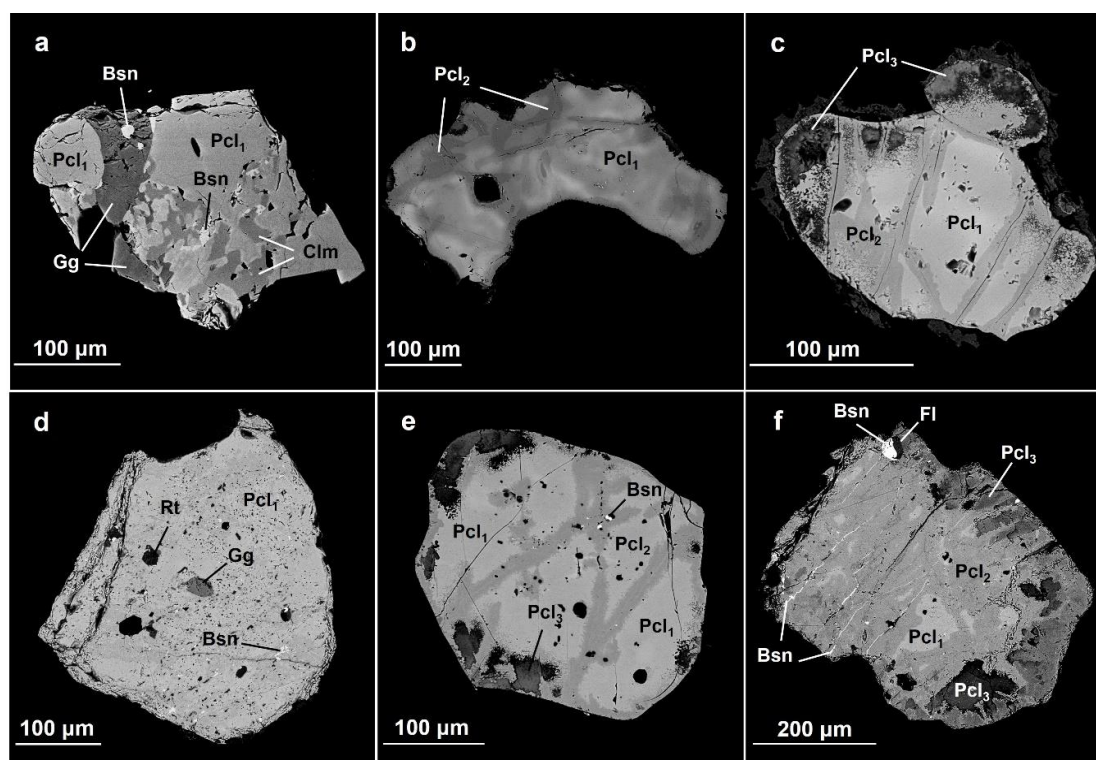
105 3. Materials and Methods

106 The analytical work was performed at the Analytical Center for Multielement and Isotope
107 Studies of the V.S. Sobolev Institute of Geology and Mineralogy (Novosibirsk, Russia) and at the
108 Analytical Center for Mineralogical, Geochemical, and Isotope Studies of the Geological Institute
109 (Ulan-Ude, Russia). We analyzed pyrochlore monofractions from different types of the Katugin
110 granites: biotite, biotite-riebeckite, biotite-arfvedsonite, arfvedsonite, aegirine-arfvedsonite, and
111 aegirine granites, as well as aegirine granite enriched in zircon, pyrochlore, and REE phases (the
112 ore-rich aegirine granite). The mineral phases were identified and analyzed using a *Carl Zeiss*
113 *LEO-1430VP* scanning electron microscope equipped with an Oxford Instruments *INCAEnergy 350*

114 energy-dispersive microanalysis system (EDS method, Analysts E.A. Khromova and S.V. Kanakin,
 115 Geological Institute, Ulan-Ude). The electron beam was operated at an accelerating voltage of 20 kV,
 116 a beam current of >0.5 nA, X-ray generation region of 2-3 μm , and a count time of 50 s. Most of
 117 analyses were applied to small zones of preparations in order to reduce the effect of grain surface
 118 microtopography. Some phases were identified using point by point analyses when necessary.
 119 Element concentrations in pyrochlores were measured on a Jeol *JXA-8100* electron-microprobe
 120 analyzer (WDS method) at the Institute of Geology and Mineralogy, Novosibirsk (analyst E.N.
 121 Nigmatulina). The operation conditions were: 2 μm beam diameter, 30 nA beam current, 20 kV
 122 accelerating voltage, and 10 s count time on peak per element. The results were calibrated using the
 123 following standards: albite (Na, Al, Si); diopside (Ca); F-phlogopite (F, K, and Mg); LiNbO_3 (Nb);
 124 rutile (Ti); Ta_2O_5 (Ta); zircon (Zr); UO_2 (U); ThO_2 (Th); SrSi glass (Sr); PbTe (Pb); BaSi glass (Ba);
 125 KLaMoO_4 (La); LiCeWO_4 (Ce); CsPrMoO_4 (Pr); RbNdWO_4 (Nd); LiSmMoO_4 (Sm); LiGdMoO_4 (Gd);
 126 LiDyWO_4 (Dy); and $\text{Y}_3\text{Al}_5\text{O}_{12}$ (Y). The PAP program was applied for matrix correction. The error was
 127 less than 2 rel.% for major elements.

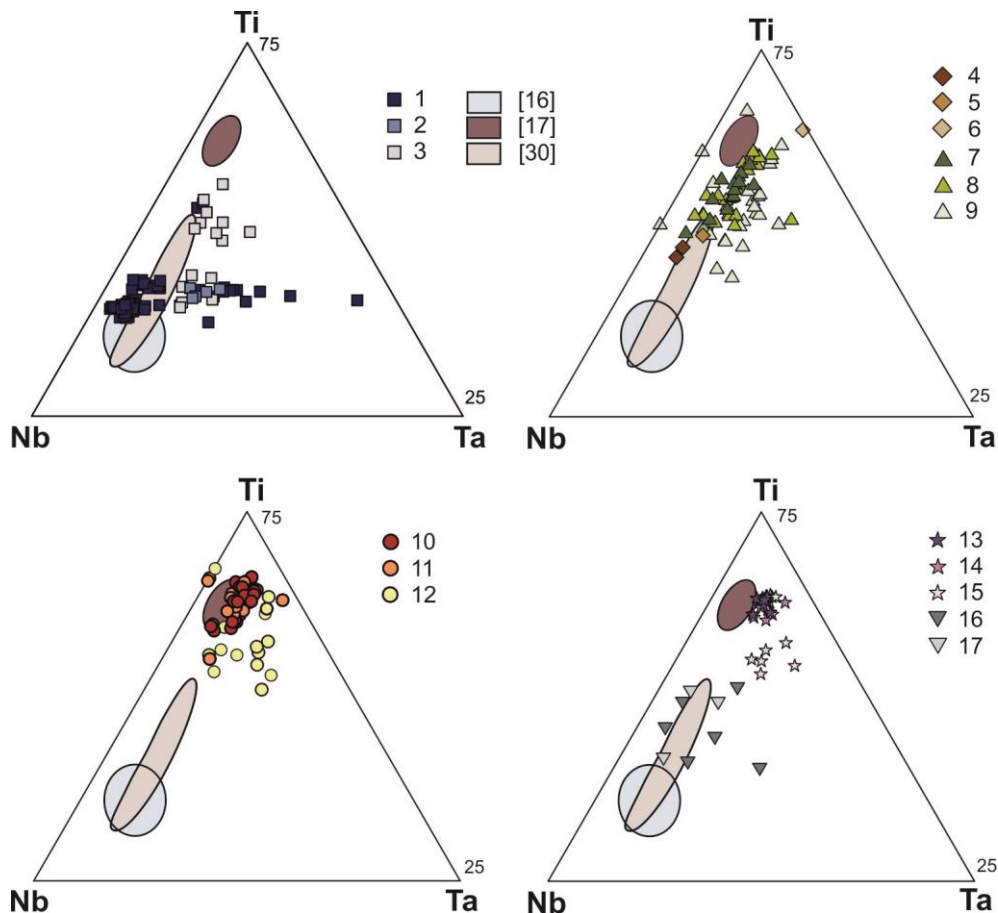
128 4. Results

129 Pyrochlore-group minerals are main ore phases of the Katugin rare-metal deposit common to
 130 different granitic rocks of the complex, except for Bt and Bt-Rbk granites of phase I which bear
 131 columbite-(Fe) as a predominant opaque mineral while pyrochlores are few. Pyrochlore is especially
 132 abundant in phase III Aeg and Aeg-Arf granites where it occurs as round or rarely octahedral
 133 euhedral grains ($\leq 5 \text{ mm}$) or as clusters and lenses, including in ore-rich zones. The pyrochlore grains
 134 are honey-yellow to almost black but most often brown, with signatures of replacement by later
 135 generations (Fig. 2). There are three main generations of pyrochlore identified according to
 136 crystallization sequence and chemistry. Early (I generation) pyrochlores are isolated crystals or
 137 aggregates. Second-generation pyrochlores, looking darker-shaded in BSE images, follow cracks or
 138 replace earlier pyrochlore in grain rims (Fig. 2); some occur in crystalline inclusions in zircon from
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141 **Figure 2.** Pyrochlore-group minerals from the Katugin granites of different types (BSE image). a, b:
 142 Bt-Arf granite; c: Bt-Rbk granite; d: Aeg-Arf granite; e, f: Aeg granite. Mineral names are abbreviated
 143 as Bsn = bastnäsite, Clm = columbite, Fl = fluorite, Gg = gagarinite-(Y), Pcl₁ = first-generation
 144 pyrochlore, Pcl₂ = second-generation pyrochlore, Pcl₃ = third-generation pyrochlore, Rt = rutile.



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146 **Figure 3.** Ternary classification diagrams (B-site) for pyrochlore-group minerals from the Katugin
 147 granites: 1-3 = first- (1), second- (2), and third- (3) generation pyrochlores from Bt-Arf granite; 4-6 =
 148 first- (4), second- (5), and third- (6) generation pyrochlores from Arf granite; 7-9 = first- (7), second-
 149 (8), and third- (9) generation pyrochlores from Arf-Aeg granite; 10-12 = first- (10), second- (11), and
 150 third- (12) generation pyrochlores from Aeg granite; 13-15 = first- (13), second- (14), and third- (15)
 151 generation pyrochlores from ore-rich Aeg granite; 16-17 = first- (16) and second- (17) generation
 152 pyrochlores from Bt granite. [16], [17], [30] = fields of fluornatropyrochlores from literature [16,17,30],
 153 respectively.

154 Aeg-Arf granites. Late (III generation) pyrochlore appears in most strongly altered and deformed
 155 grain parts. In Bt-Arf granites, early pyrochlore is rarely replaced by later phases but rather by
 156 aggregates of columbite, gagarinite-(Y), and bastnäsité-(Ce) (Fig. 2 a). Pyrochlore from Aeg, Aeg-Arf,
 157 and Arf granites (phase III) often encloses numerous veinlets or microcrystals (10 μm on average) of
 158 other phases (bastnäsité-(Ce), gagarinite-(Y), yttrifluorite, and occasionally rutile) produced by
 159 alteration (Fig.2 f). Columbite-(Fe) replaces pyrochlore in some grains. Some pyrochlores enclose
 160 cryolite, quartz, albite, feldspar, aegirine, zircon, and magnetite.

161 4.1. First-generation pyrochlore

162 Among early pyrochlores, the (#Nb) pyrochlore-group end-members predominates over those
 163 of microlite (#Ta) or betafite (#Ti) groups in all analyses (Fig. 3; Tables 1-3). The B-site is occupied by
 164 73-90 at.% of Nb, 7.5-23 at.% of Ti, and 0-15 at.% of Ta. The B-site is occupied by 73-90 at.% of Nb,
 165 7.5-23 at.% of Ti, and 0-15 at.% of Ta. The A-site accommodates large cations of Na, REE+Y, and Ca
 166 (Fig. 4), as well as U, Th and Pb (no more than 6 at.% in total); other cations occur in minor amounts.
 167 REE are mainly Ce while HREE and Y are low. Abundant F occupies the Y-site (Fig. 5).

168 First-generation pyrochlore from Bt-Arf granites differs from those in other granite types and
 169 contains as much as 6.4 to 10.7 wt.% Na_2O (Fig.5; Table 1), which exceeds previously reported values
 170 from different occurrences worldwide (within 9.18 wt.% Na_2O) [12]. It is also rich in niobium (up to
 171

172 **Table 1.** Chemistry (wt.%) of pyrochlore-group minerals from the Katugin biotite-arfvedsonite granite (WDS analyses).

	I						II		III		
	1	2	3	4	5	6	n=12	sd	n=8	min	max
SiO ₂	bd	bd	bd	bd	bd	bd	0.02	0.06	4.47	1.49	7.40
TiO ₂	2.84	2.79	3.05	2.67	2.85	3.05	4.20	1.19	3.66	2.64	4.56
Al ₂ O ₃	bd	bd	bd	bd	bd	bd	bd	-	0.17	bd	0.30
CaO	1.68	1.66	3.23	3.47	2.78	3.19	3.53	1.26	2.44	0.19	3.52
Na ₂ O	10.46	9.18	8.60	9.95	7.22	6.77	2.25	1.24	1.11	3.07	1.26
K ₂ O	bd	bd	bd	bd	bd	bd	bd	0.02	0.54	0.06	1.31
Ce ₂ O ₃	8.3	8.72	6.15	5.97	7.81	7.92	7.45	1.23	4.36	1.84	6.55
La ₂ O ₃	2	2.14	1.24	1.42	2.18	2.08	2.00	0.56	0.86	0.34	1.75
Pr ₂ O ₃	1.18	1.06	0.90	0.77	0.88	0.85	0.78	0.11	0.38	bd	0.68
Nd ₂ O ₃	3.52	3.51	3.2	2.78	2.56	2.53	2.39	0.45	1.11	0.53	1.91
Sm ₂ O ₃	0.75	0.83	1.02	0.83	0.50	0.47	0.49	0.13	0.26	bd	0.46
Gd ₂ O ₃	0.45	0.30	0.82	0.81	0.38	0.28	0.29	0.15	0.15	bd	0.28
Dy ₂ O ₃	0.25	0.24	0.60	0.65	0.24	0.17	0.14	0.08	bd	bd	bd
Y ₂ O ₃	0.49	0.53	0.90	1.29	bd	0.56	0.19	0.24	0.09	bd	0.30
ThO ₂	1.30	1.51	0.84	0.69	0.81	0.82	1.35	0.69	1.26	0.73	1.63
UO ₂	0.37	0.39	0.70	0.66	1.49	1.56	1.76	0.18	2.19	1.68	3.06
PbO	bd	bd	bd	bd	0.11	0.11	0.54	0.75	1.28	bd	2.31
Nb ₂ O ₅	57.99	59.23	60.42	59.21	47.22	51.14	55.71	1.67	50.06	46.13	55.38
Ta ₂ O ₅	2.64	2.06	2.20	2.27	15.23	11.51	4.54	1.42	5.03	4.39	6.13
SrO	bd	bd	bd	bd	bd	bd	0.09	0.17	0.96	0.28	1.55
BaO	bd	bd	bd	bd	bd	bd	0.15	0.23	1.90	bd	2.94
F	4.62	4.73	4.69	4.72	4.31	4.61	4.26	0.46	2.28	0.63	3.91
Total	98.84	98.88	98.55	98.17	96.57	97.61	92.13		84.87		
O ₂ =F	1.95	1.99	1.97	1.99	1.82	1.94	1.79		0.96		
Total	96.9	96.89	96.58	96.18	94.76	95.67	90.33		83.91*		
ΣREE₂O₃	16.45	16.79	13.93	13.23	14.55	14.30	13.53		7.12		
Na	1.395	1.210	1.104	1.313	1.013	0.920	0.295		0.136		
Ca	0.124	0.121	0.229	0.253	0.216	0.239	0.256		0.165		
K	-	-	-	-	-	-	-		0.044		
LREE	0.394	0.401	0.300	0.290	0.367	0.353	0.323		0.161		
HREE	0.016	0.012	0.031	0.033	0.015	0.010	0.009		0.003		
Y	0.018	0.019	0.032	0.047	-	0.021	0.007		0.003		
Th	0.020	0.023	0.013	0.011	0.013	0.013	0.021		0.018		
U	0.006	0.006	0.010	0.010	0.024	0.024	0.027		0.031		
Pb	-	-	-	-	0.002	0.002	0.010		0.022		
Sr	-	-	-	-	-	-	0.003		0.035		
Ba	-	-	-	-	-	-	0.004		0.047		
ΣA	1.973	1.792	1.718	1.956	1.650	1.583	0.954		0.62		
Al	-	-	-	-	-	-	-		0.013		
Si	-	-	-	-	-	-	0.001		0.283		
Ti	0.147	0.143	0.152	0.137	0.155	0.161	0.213		0.174		
Nb	1.804	1.819	1.809	1.821	1.545	1.620	1.702		1.433		
Ta	0.049	0.038	0.040	0.042	0.300	0.219	0.084		0.087		
ΣB	2.000	2.000	2.000	2.000	2.000	2.000	2.000		2.000		
F	1.006	1.017	0.981	1.017	0.988	1.021	0.911		0.456		
#Nb	90.18	90.97	90.43	91.07	77.26	81.00	85.15		84.61		
#Ti	7.35	7.13	7.59	6.83	7.76	8.04	10.67		10.27		
#Ta	2.47	1.90	1.98	2.10	14.99	10.97	4.178		5.12		

Note: 1-4 = pyrochlores without visible zoning; 5, 6 = rims in zoned pyrochlores. *n* is number of analyses; sd is standard deviation; numerals I, II, III correspond to pyrochlore generations, bd is below detection limit. For compositionally variable third-generation pyrochlores, minimum (min) and maximum (max) contents are given instead of standard deviations. Formula based on 2 cations at B-site. #Nb, #Ti, #Ta are pyrochlore, betafite, and microlite end-member components (mol.%), respectively. * - Sum also includes 0.07 wt.% Fe₂O₃ and 0.19 ZrO₂.

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181**Table 2.** Chemistry (wt.%) of pyrochlore-group minerals from the Katugin biotite and biotite-riebeckite granites (1) and arfvedsonite aegirine-arfvedsonite granites (2).

	1				2						
	I		II		I		II		III		
	n=7	sd	n=3	sd	n=11	sd	n=21	sd	n=23	min	max
SiO ₂	bd	-	1.51	0.23	bd	-	0.02	0.07	5.47	2.34	9.24
TiO ₂	4.33	0.87	4.3	0.97	5.92	0.91	6.27	0.82	5.04	2.81	6.54
Al ₂ O ₃	bd	-	bd	-	bd	-	bd	-	0.23	bd	0.64
Fe ₂ O ₃	bd	-	bd	-	bd	0.01	bd	0.01	0.17	bd	2.05
CaO	3.71	1.05	2.85	0.44	3.56	1.28	2.52	1.69	1.34	0.39	3.06
Na ₂ O	6.61	0.63	2.66	1.56	6.91	0.96	3.00	0.50	1.90	0.20	3.83
K ₂ O	bd	-	bd	-	bd	-	bd	0.02	0.45	0.15	1.17
Ce ₂ O ₃	8.55	1.32	8.80	0.93	6.85	0.77	7.11	0.66	5.49	3.10	7.92
La ₂ O ₃	2.19	0.35	2.15	0.45	1.44	0.64	1.43	0.33	0.78	0.26	1.59
Pr ₂ O ₃	0.36	0.59	bd	-	0.73	0.11	0.77	0.18	0.50	0.24	0.77
Nd ₂ O ₃	3.29	0.31	3.06	0.28	2.58	0.42	2.55	0.38	1.35	0.61	2.32
Sm ₂ O ₃	bd	-	bd	-	0.69	0.15	0.76	0.18	0.43	0.16	0.69
Gd ₂ O ₃	bd	-	bd	-	0.58	0.20	0.70	0.32	0.32	bd	0.66
Dy ₂ O ₃	bd	-	bd	-	0.40	0.26	0.58	0.43	0.43	bd	1.18
Y ₂ O ₃	bd	-	bd	-	1.10	0.93	1.81	1.52	0.71	bd	1.99
ThO ₂	bd	-	bd	-	1.00	0.43	1.03	0.45	1.05	0.20	2.06
UO ₂	1.64	1.18	2.16	0.56	1.86	0.06	1.90	0.28	2.61	1.53	5.50
PbO	bd	-	bd	-	0.94	0.48	0.57	0.45	1.04	0.25	2.52
Nb ₂ O ₅	58.08	2.56	55.06	1.15	54.26	1.13	53.21	0.84	47.56	41.76	54.36
Ta ₂ O ₅	4.38	2.53	2.96	1.08	3.27	0.59	3.75	0.51	4.74	3.24	6.83
SrO	bd	-	bd	-	0.13	0.27	0.29	0.30	0.42	bd	1.40
BaO	bd	-	bd	-	bd	-	0.06	0.15	0.81	bd	2.89
ZrO ₂	bd	-	bd	-	0.02	0.05	0.01	0.04	0.37	bd	1.53
F	4.88	0.51	2.98	1.78	4.55	0.53	3.79	1.17	1.85	bd	3.27
Total	98.10		88.51		96.79		92.13		85.05		
O ₂ =F	2.05		1.25		1.91		1.60		0.78		
Total	96.05		87.25		94.88		90.53		84.27		
ΣREE ₂ O ₃	14.43		14.02		13.28		13.90		9.230		
Na	0.837		0.339		0.897		0.390		0.226		
Ca	0.260		0.201		0.256		0.181		0.088		
K	-		-		-		-		0.035		
LREE	0.342		0.336		0.299		0.307		0.190		
HREE	-		-		0.022		0.028		0.015		
Y	-		-		0.039		0.065		0.023		
Th	-		-		0.015		0.016		0.015		
U	0.024		0.032		0.028		0.028		0.036		
Pb	-		-		0.017		0.010		0.017		
Sr	-		-		0.005		0.011		0.015		
Ba	-		-		-		0.002		0.019		
ΣA	1.463		0.907		1.577		1.038		0.679		
Fe ³⁺	-		-		-		-		0.008		
Al	-		-		-		-		0.017		
Si	-		0.099		-		0.001		0.334		
Ti	0.231		0.213		0.298		0.316		0.237		
Nb	1.591		1.635		1.642		1.613		1.315		
Ta	0.073		0.053		0.059		0.068		0.079		
Zr	-		-		0.001		-		0.011		
ΣB	2.000		2.000		2.000		2.000		2.000		
F	1.005		0.619		0.963		0.804		0.357		
#Nb	90.18	90.97	90.43	91.07	77.26	81.00	85.15		84.61		
#Nb	85.49		86.02		82.13		80.75		80.65		
#Ti	10.64		11.20		14.90		15.83		14.52		

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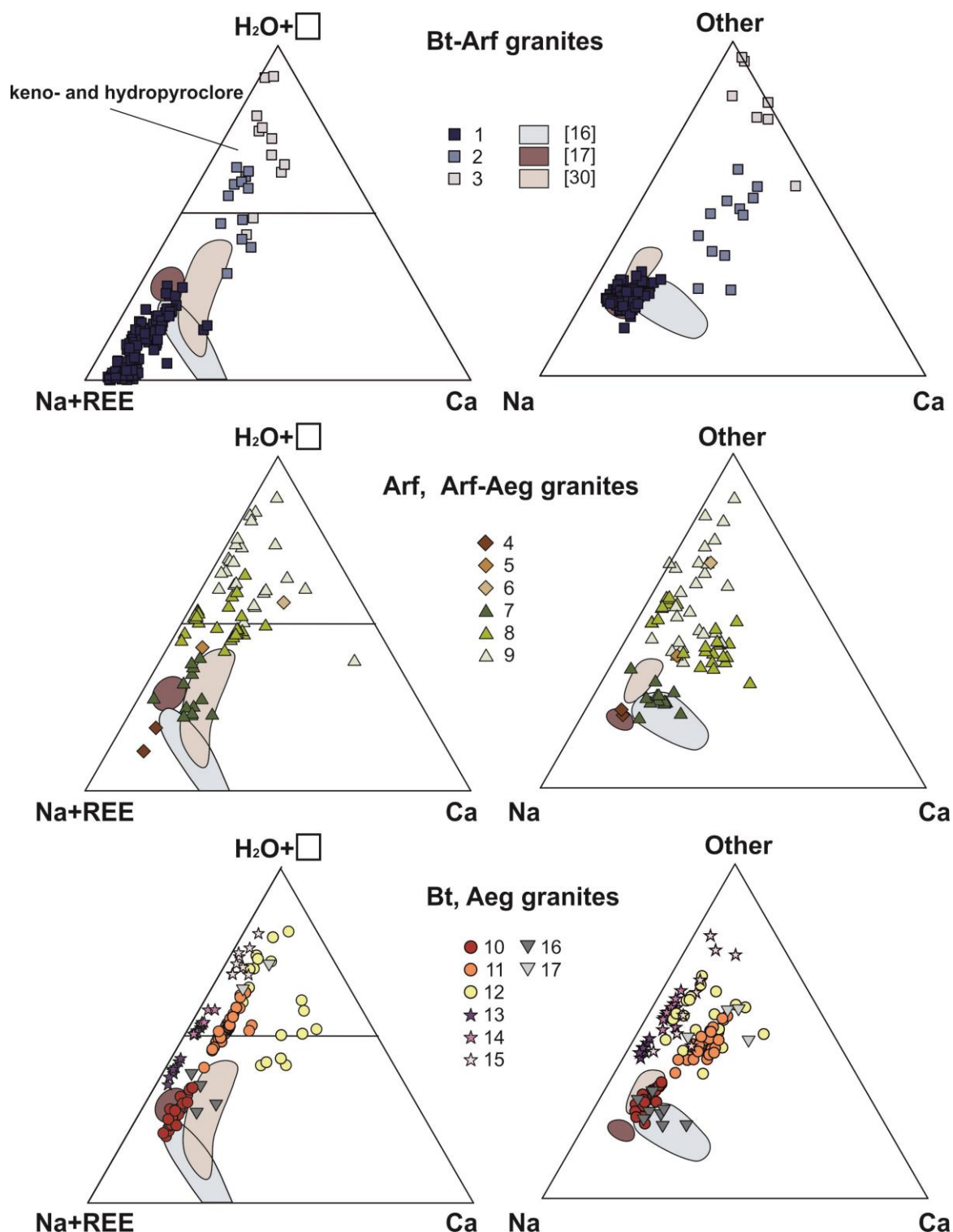
Note: 1 = WDS analyses; 2 = EDS analyses. Other symbols as in Table 1.

183
184**Table 3.** Chemistry (wt.%) of pyrochlore-group minerals from the Katugin aegirine granites (WDS analyses).

	1						2			
	I		II		III		I	II	III	
	n=42	sd	n=46	sd	n=23	min	max	n=8	n=14	n=9
SiO ₂	0.04	0.19	0.08	0.27	5.07	1.23	9.05	bd	bd	5.50
TiO ₂	7.57	0.87	7.49	0.73	6.47	4.52	10.01	7.76	7.51	5.94
Al ₂ O ₃	bd	-	bd	-	0.18	bd	0.38	bd	bd	0.35
Fe ₂ O ₃	bd	-	bd	-	1.18	bd	2.89	bd	bd	0.05
MnO	bd	-	bd	-	1.24	bd	2.09	bd	bd	bd
CaO	2.68	0.21	2.61	0.20	1.56	bd	2.98	0.69	0.68	0.81
Na ₂ O	6.67	0.63	2.68	0.55	2.03	0.63	3.96	4.93	2.83	1.47
K ₂ O	bd	-	bd	-	0.46	bd	1.67	bd	bd	0.72
Ce ₂ O ₃	8.36	0.56	8.26	0.45	4.1	bd	7.67	7.42	7.18	5.60
La ₂ O ₃	2.22	0.67	2.18	0.61	0.68	bd	1.45	1.48	1.44	0.73
Pr ₂ O ₃	0.93	0.08	0.91	0.07	0.45	0.15	0.82	0.82	0.83	0.52
Nd ₂ O ₃	2.89	0.34	2.80	0.28	1.23	0.47	2.14	2.88	2.76	1.42
Sm ₂ O ₃	0.75	0.05	0.74	0.05	0.34	0.12	0.57	1.01	0.99	0.48
Gd ₂ O ₃	0.54	0.05	0.51	0.07	0.26	bd	0.46	1.13	1.08	0.55
Dy ₂ O ₃	0.35	0.04	0.35	0.06	0.21	bd	0.40	1.11	1.05	0.72
Y ₂ O ₃	0.88	0.18	0.82	0.12	0.06	bd	0.50	3.84	3.70	0.96
ThO ₂	1.12	0.11	1.13	0.10	1.39	0.80	1.91	0.75	0.70	0.93
UO ₂	1.85	0.11	1.84	0.24	2.46	bd	5.62	1.77	1.75	2.78
PbO	0.57	0.10	0.60	0.14	0.98	bd	3.52	0.18	0.23	0.86
Nb ₂ O ₅	54.21	2.76	53.31	2.48	50.66	40.22	63.26	53.22	51.75	47.97
Ta ₂ O ₅	3.08	0.47	2.77	0.05	4.23	bd	10.91	3.53	3.56	4.96
SrO	0.04	0.10	0.05	0.11	0.02	bd	0.17	bd	0.03	0.02
ZrO ₂	bd	-	bd	-	bd	bd	bd	bd	bd	0.31
F	4.78	0.54	3.74	1.45	1.00	0.55	3.55	4.94	4.00	1.44
Total	99.54		92.88		85.86			97.45	92.06	85.10
O ₂ =F	2.01		1.58		0.42			2.08	1.69	0.61
Total	97.53		91.30		84.51			95.37	90.37	84.49
ΣREE ₂ O ₃	16.05		15.75		7.18			15.84	15.33	10.01
Na	0.833		0.339		0.231			0.620	0.365	0.170
Ca	0.185		0.183		0.098			0.048	0.049	0.052
K	-		-		0.034			-	-	0.055
Mn	-		-		0.122			-	-	-
LREE	0.355		0.354		0.143			0.320	0.319	0.190
HREE	0.019		0.018		0.009			0.047	0.046	0.025
Y	0.030		0.028		0.002			0.132	0.131	0.031
Th	0.016		0.017		0.019			0.011	0.011	0.013
U	0.027		0.027		0.032			0.026	0.026	0.037
Pb	0.010		0.011		0.015			0.003	0.004	0.014
Sr	0.002		0.002		0.001			-	0.001	0.001
ΣA	1.477		0.979		0.706			1.207	0.952	0.586
Fe ³⁺	-		-		0.051			-	-	0.002
Al	-		-		0.012			-	-	0.025
Si	0.003		0.005		0.289			-	-	0.329
Ti	0.366		0.369		0.277			0.378	0.376	0.268
Nb	1.577		1.577		1.305			1.560	1.559	1.295
Ta	0.054		0.049		0.066			0.062	0.065	0.082
ΣB	2.000		2.000		2.000			2.000	2.000	2.000
F	0.973		0.774		0.181			1.014	0.844	0.272
#Nb	78.96		79.05		79.05			77.98	77.96	78.86
#Ti	18.34		18.48		18.48			18.90	18.81	16.23
#Ta	2.70		2.47		2.47			3.11	3.23	4.91

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Note: 1 = pyrochlore-group minerals from common (1) and ore-rich (2) aegirine granites. Other symbols as in Table 1.



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Figure 4. Ternary classification diagrams (A-site) for pyrochlore-group minerals from the Katugin granites. Legend same as in Fig. 3. In the Ca-Na+REE-H₂O+□ diagrams, Ca stands also for other divalent cations (Mn, Pb, Sr, Ba); Na+REE stands also for Y, U, Th, and K.

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62 wt.% Nb₂O₅) but has low TiO₂ of 2.4 to 3.8 wt.% (Fig.3). Some pyrochlore grains in these granites are zoned, possibly according to growth (Fig.2 b), with Ta₂O₅-rich (to 15 wt.%) lighter-color zones. Ta enrichment may result from accumulation in evolving granitic melt [7]. The Bt-Arf granite-hosted pyrochlore shows the most complete cation occupancy of the A-site (1.55-2 apfu), with high REE (16.7 wt.% LREE₂O₃ and 1.6 wt.% HREE₂O₃) and 1.3 wt.% Y₂O₃ but low UO₂ (0.3-0.9 wt.%) and PbO (to 0.21 wt.%). The contents of CaO vary from 1.4 to 3.5 wt.%, with the Na/Ca ratio from 3.7 to 13 (7.1 on average). Fluorine reaches 5.34 wt.% or 1.15 apfu. The LREE/HREE and Y/Ce ratios vary from 8 to

198 37 and 0.04 to 0.34, with the averages 18.2 and 0.16, respectively. The general formula of this
199 pyrochlore can be expressed as $(\text{Na}_{1.2}\text{Ca}_{0.18}\text{LREE}_{0.36}\text{HREE}_{0.04}\text{Y}_{0.03}\text{U}_{0.02}\text{Th}_{0.02})_{1.85}(\text{Nb}_{1.8}\text{Ti}_{0.15}\text{Ta}_{0.05})_2\text{O}_{5.94}\text{F}_{1.06}$.

200 First-generation pyrochlores from Bt, Bt-Rbk granites of phase I and Arf, Aeg-Arf and Aeg
201 granites of phase III are compositionally similar (Table 2-3) and differ from that in the ore-rich Aeg
202 granite. Pyrochlore from the ore-rich Aeg granite contains less CaO (0.63-0.85 wt.%) and relatively
203 high HREE₂O₃ (2.4 wt.%) and Y₂O₃ (4 wt.%) (Table 3); it has the highest ratios of Na/Ca (10.2-15.2)
204 and Y/Ce (0.72-0.81) but lowest LREE/HREE (6.5-7.7) compared to the counterparts from other
205 granite types (Fig. 5).

206 The betafite component (#Ti) with 3.75 to 8 wt.% TiO₂ increases progressively from 9.8 % in Bt
207 and Bt-Rbk granites to 19.9 % in Aeg granite while pyrochlore (#Nb) decreases correspondingly
208 (Fig.3); the microlite component (#Ta) varies from 2 to 4.4 wt.% Ta₂O₅. The A occupancy in
209 pyrochlores from Bt, Bt-Rbk, Arf, Aeg-Arf, and Aeg granites varies from 1 to 1.55 apfu while Na
210 does not exceed 0.95 apfu (8.1 wt.% Na₂O) being on average 0.83 apfu (6.5 wt.% Na₂O). CaO in
211 pyrochlores from Bt, Bt-Rbk and Aeg-Arf granites may have a large range, including within the
212 same sample (2 to 5.5 wt.%; Na/Ca = 2-7.2), but is quite stable (2.3-2.8 wt.%; Na/Ca = 3.7-5) in
213 pyrochlore from all aegirine granites (Fig. 5). The latter also has high REE₂O₃: 15-16 wt.%
214 (LREE/HREE = 16-21; Y/Ce = 0.08-0.23) against 10-14.5 wt.% REE₂O₃ (LREE/HREE = 10-19; Y/Ce =
215 0.1-0.25) in pyrochlores from other granite types. UO₂, ThO₂ and PbO do not exceed 2.6 wt.%, 1.8
216 wt.%, and 1.4 wt.%, respectively. F is 3.6 to 5.5 wt.% (4.6 wt.% on average).

217 4.2. Second-generation pyrochlore

218 Later pyrochlore differs from that of first generation in low contents of Na₂O (1-4.5 wt.%, 2.8
219 wt.% on average), lower F (2-5 wt.%, 4 wt.% on average), and less occupancy of A- and Y-sites
220 (Tables 1-3; Figs. 4-5). Other cations have concentrations commensurate with those in early
221 pyrochlore.

222 4.3. Third-generation pyrochlore

223 The composition of third-generation pyrochlore differs in large ranges of both major and trace
224 elements (Table. 1-3, Fig. 3-5) and analytical totals notably below 100%, as well as in the presence of
225 K₂O (up to 2.3 wt.%), BaO (to 3 wt.%), PbO (to 2.5 wt.%), Fe₂O₃ (to 0.6 wt.%) and significant SiO₂ (to
226 9.2 wt.%). It also contains more ThO₂ (to 2.1 wt.%) and UO₂ (to 5.6 wt.%) and markedly lower
227 concentrations of REE₂O₃, CaO, Na₂O, Nb₂O₅, Ta₂O₅, and F relative to pyrochlores of earlier
228 generations. Most of these pyrochlore samples fall in the field of hydro- or kenopyrochlore (Fig. 4).

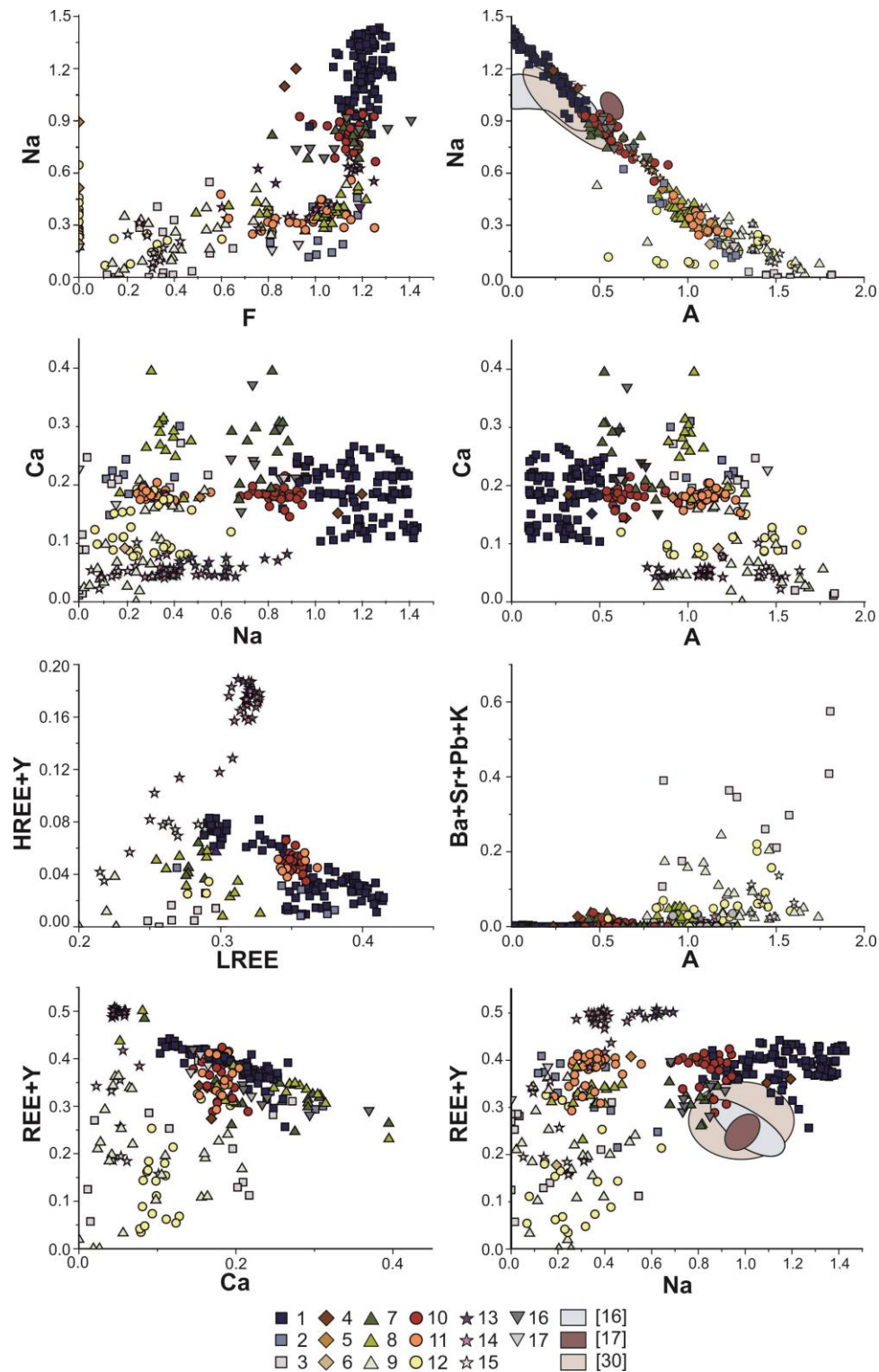
229 5. Discussion

230 5.1. Comparison with fluornatropyrochlore from other occurrences

231 Pyrochlores of different generations from the Katugin granites differ notably in their chemistry
232 and show quite large ranges of some element concentrations. First-generation pyrochlore, with high
233 Na and F concentrations and low Ca, is of special interest and can be classified as
234 fluornatropyrochlore. The latter has been rarely found in natural occurrence and became a
235 registered mineral species in 2013 only [20]. In this respect, it is pertinent to compare the Katugin
236 pyrochlore with its counterparts known from other places.

237 Fluornatropyrochlore from the Boziguoer rare-metal granites in China (the first finding of this
238 mineral) differs from the samples we analyzed in higher concentrations of PbO (to 16.17 wt.%) and
239 UO₂ (5.8 - 7.5 wt.%) [20,30] and much lower Na₂O (5.5-8.5 wt.%) (Fig. 4-5) at quite high Na/Ca ratios
240 of 4.1-6.6. The F contents in the Boziguoer pyrochlore range within 4.6-7.1 wt.% or 1.25 to 1.63 apfu
241 [30]. The 1.25 to 1.63 apfu exceeding the common value of 1 apfu by a factor of ~1.5 requires
242 additional explanation, as the overestimation may lead to notable structure distortion.

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Figure 5. Binary diagrams (apfu) for pyrochlore-group minerals from the Katugin granites. A = vacancy or H₂O at the A-site. The HREE+Y/LREE diagram includes only WDS data. 1-3 = I-, II-, and III-generation pyrochlores from Bt-Arf granite; 4-6 = I-, II-, and III-generation pyrochlores from Arf granite; 7-9 = I-, II-, and III-generation pyrochlores from Arf-Aeg granite; 10-12 = I-, II-, and III-generation pyrochlores from Aeg granite; 13-15 = I-, II-, and III-generation pyrochlores from ore-bearing Aeg granite; 16-17 = I- and II-generation pyrochlores from Bt granite. [16], [17], [30] = fields of fluornatropyrochlores from literature [16,17,30], respectively.

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Fluornatropyrochlores with the highest Na contents were reported from the Nechalacho [25,16] and the Mariupol Massif [17] syenites and nepheline syenites but the pyrochlores from both

254 occurrences have Na₂O lower than in those from the Katugin Bt-Arf granites (7-9.18 wt.% Na₂O
255 against an average of 9.4 wt.%, respectively). On the other hand, they have UO₂ (within 1.1 wt.%),
256 ThO₂ (≤1.6 wt.%) and PbO (≤0.5 wt.%) as low as in the Katugin pyrochlores.

257 All other published fluornatropyrochlore compositions with high Na/Ca ratios have
258 predominant components with 0.78 to 0.93 #Nb (Fig.3), but their REE₂O₃ contents (mainly LREE) are
259 much lower (to 13.4 wt.%) than in the Katugin pyrochlores (Fig.4). Low Y/Ce ratios (0.12-0.2) in all
260 Katugin pyrochlores we analyzed, except for those from ore-bearing Aeg granite (Y/Ce to 0.8), are
261 similar to the ratios reported for their counterparts from other occurrences.

262 5.2. Isomorphism in the Katugin pyrochlores

263 Judging by their chemistry, pyrochlore-group minerals are prone to extensive iso- and
264 hetero-valent isomorphic substitutions at all sites, which follow hardly decipherable scenarios.
265 Incorporation of HFSE (REE³⁺, U⁴⁺ and Th⁴⁺) in early pyrochlore likely compensates for Na excess
266 with respect to the ideal neutral formula NaCaNb₂O₆F, at complete X occupancy. Therefore, the
267 substitution is expected to proceed as 2Ca²⁺ → Na⁺+REE³⁺; 3Ca²⁺ → 2Na⁺+U⁴⁺(Th⁴⁺). However, the
268 correlation restricted to Ca vs. REE+Y ($r = -0.87$ to -0.97) (Fig.5) suggests rather the scenario 3Ca²⁺ →
269 2REE³⁺. On the other hand, Na correlates inversely with the A-site vacancy ($r = -0.85$ to -0.99) (Fig.5),
270 and its poor correlation with F is inconsistent with the Na⁺+F → A+Y substitution, where A and Y are
271 the respective site vacancies. The Na-A correlation may result from incorporation of the H₃O⁺ group
272 at A as Na⁺ → H₃O⁺, or by some more intricate substitution. The correlation of Na and F is better in
273 second-generation pyrochlore (Na⁺+F → A+Y), where H₃O⁺ can also substitute for Na⁺.

274 Significant amounts of silica found in third-generation pyrochlore may occur as small
275 inclusions of amorphous SiO₂. However, since high-Si pyrochlores were reported previously [1, etc.],
276 we infer that silica may be isomorphic impurity rather than being SiO₂ mineral inclusions.

277 5.3. Genesis of different pyrochlore generations in the Katugin rare-metal deposit

278 It was previously found out that REE phases in the Katugin alkaline granites crystallized after
279 main rock-forming minerals (feldspar and quartz) from residual melt rich in Fe, Na, HFSE, Zn, Sn
280 and volatiles (F, CO₂, P, S, and H₂O) [27]. First-generation pyrochlore is the earliest REE phase.
281 Ultimately high contents of F and Na indicate their high activity during crystallization of pyrochlore
282 which was synchronous with or slightly postdating the formation of melanocratic phases (aegirine,
283 biotite, and arfvedsonite). Fe extracted from the melt became bound in the melanocratic minerals,
284 which hindered the crystallization of columbite [2].

285 Second-generation pyrochlore likely formed at the early postmagmatic stage in the evolution of
286 the Katugin complex, judging by the presence of generation I and II pyrochlores enclosed in zircon
287 which was previously interpreted as magmatic or early postmagmatic [23,31].

288 Third-generation pyrochlore may have replaced the earlier phases during supergene alteration
289 processes after the formation of the Katugin ore deposit, whereby pyrochlore gained K, Fe, Mn, Si,
290 Th, and U but lost REE, Ca, Na, Nb, and Ta.

291 The compositional variations of pyrochlore-group minerals in granitic systems remain poorly
292 investigated because of their rare occurrence. Minerals of this group were studied in carbonatites,
293 where many authors [1,5,32,33] distinguished similar formation and transformation stages in the
294 course of magmatic, postmagmatic, and supergene evolution.

295 Although being different in high Ca contents (about 1 apfu in magmatic pyrochlore),
296 carbonatite-hosted pyrochlores show composition trends similar to those revealed in their Katugin
297 granite-hosted analogs: progressive decrease in Na and F from magmatic to postmagmatic varieties,
298 at invariable Ca and slightly variable relative amounts of cations at the B-site. The Katugin late
299 pyrochlore shares chemical similarity with the pyrochlore variety classified as supergene [33].

300 6. Conclusions

- 301 1. The Katugin granites store pyrochlore of three generations which formed, respectively, during
302 late magmatic and early postmagmatic evolution of the complex and later supergene alteration.

- 303 2. Primary magmatic pyrochlore is fluornatropyrochlore with high Na/Ca ratios, which is of
 304 extremely rare natural occurrence (only few sites in the world are known). Some pyrochlore
 305 samples from the Katugin deposit contain as much as 10.7 wt.% Na₂O exceeding all previously
 306 reported values.
- 307 3. Early postmagmatic pyrochlore has lower concentrations of Na and F than the primary phase,
 308 while the contents of other elements are commensurate.
- 309 4. Supergene postmagmatic pyrochlore differs in markedly lower REE, Ca, Nb, and Ta and zero
 310 Na and F, as well as in the presence of Si, K, Ba, Pb and Fe.
- 311 5. The Katugin pyrochlores contain more REE₂O₃ than their high-Na counterparts from other
 312 known occurrences, but the Y/Ce ratios are as low as elsewhere. They are compositionally
 313 similar especially to fluornatropyrochlore hosted by syenite and nepheline syenite at the
 314 deposits of Nechalacho (Canada) and Mariupol Massif (Ukraine).

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 317 E.P.B.; writing—review and editing, A.E.S., E.V.S. and V.B.S.; visualization, A.E.S.

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